



Since January 2020 Elsevier has created a COVID-19 resource centre with free information in English and Mandarin on the novel coronavirus COVID-19. The COVID-19 resource centre is hosted on Elsevier Connect, the company's public news and information website.

Elsevier hereby grants permission to make all its COVID-19-related research that is available on the COVID-19 resource centre - including this research content - immediately available in PubMed Central and other publicly funded repositories, such as the WHO COVID database with rights for unrestricted research re-use and analyses in any form or by any means with acknowledgement of the original source. These permissions are granted for free by Elsevier for as long as the COVID-19 resource centre remains active.



Review

Synthesis of β -amino- α -trifluoromethyl alcohols and their applications in organic synthesis

Grzegorz Mlostoń^{a,*}, Emilia Obijalska^a, Heinz Heimgartner^{b,**}

^a Department of Organic and Applied Chemistry, University of Łódź, Tamka 12, PL-91-403, Łódź, Poland

^b Institute of Organic Chemistry, University of Zurich, Winterthurerstrasse 190, CH-8057 Zurich, Switzerland

ARTICLE INFO

Article history:

Received 17 April 2010

Received in revised form 21 May 2010

Accepted 29 May 2010

Available online 4 June 2010

Keywords:

Organic synthesis

Amino alcohols

Fluorinated compounds

Trifluoromethyl alcohols

Stereocontrolled processes

Biological activity

ABSTRACT

A comprehensive overview on methods applied for syntheses of β -amino- α -trifluoromethyl alcohols, including stereocontrolled variants, is presented. In addition, reported cases of the exploration of β -amino- α -trifluoromethyl alcohols for the preparation of trifluoromethylated peptidomimetics and other biologically active, fluorinated compounds are discussed. Attractive opportunities for their applications as organocatalysts are also presented.

© 2010 Elsevier B.V. All rights reserved.

Contents

1. Introduction	829
2. Synthetic approaches to β -amino- α -trifluoromethyl alcohols	830
2.1. Reduction of cyanohydrines derived from trifluoromethylketones and fluorol	830
2.2. Henry reaction and subsequent reduction of β -nitro alcohols	830
2.3. Reduction of α -aminoalkyl trifluoromethyl ketones	831
2.4. Ring opening of trifluoromethyloxiranes with <i>N</i> -nucleophiles	833
2.5. Nucleophilic trifluoromethylation of α -amino aldehydes and ketones	834
2.6. Nucleophilic trifluoromethylation of α -iminocarbonyl derivatives	836
2.7. Petasis reaction of 2-hydroxy-3,3,3-trifluoropropanal	836
2.8. Miscellaneous reactions	836
3. Applications of β -amino- α -trifluoromethyl alcohols in organic synthesis	838
3.1. Transformations of β -amino- α -trifluoromethyl alcohols into trifluoromethylated carboxylic acids and heterocycles	838
3.2. Preparation of peptidomimetics and other biologically active compounds	840
3.3. Application in stereocontrolled synthesis	841
4. Conclusions	842
Acknowledgements	842
References	842

1. Introduction

Amino alcohols form a group of difunctional compounds with great importance as building blocks in organic synthesis and their diverse applications as pharmaceuticals and catalysts are widely known [1]. Of special interest are β -amino alcohols [1f], and many of them display significant biological activity (e.g. ephedrine, quini-

* Corresponding author. Tel.: +48 42 635 57 61; fax: +48 42 665 51 62.

** Corresponding author. Tel.: +41 44 635 42 82; fax: +41 44 635 68 12.

E-mail addresses: gmloston@uni.lodz.pl (G. Mlostoń), heimgart@oci.uzh.ch (H. Heimgartner).

dine, terbutaline, etc.). On the other hand, some of them are known as efficient catalysts for stereocontrolled syntheses (e.g. 3-amino-borneols, phenylalaninols, prolinols, etc.) [1a–e,h] and as building blocks for the preparation of optically active heterocycles [1g].

In the recent three decades, a rapid development of fluorinated organic compounds was observed [2a], and special attention is focused on the preparation and practical exploration of trifluoromethylated systems. The incorporation of the CF₃ group results in relevant modification of physico-chemical and biological properties of organic compounds [2b–f]. The current development of methods for enantioselective trifluoromethylations opens an attractive access to enantiopure compounds, which are of particular interest for further applications [3].

It seems obvious that β-amino-α-trifluoromethyl alcohols, which combine the important structural features of the compounds mentioned above, are especially attractive as unique building blocks, catalysts, and potentially, as pharmaceuticals. In spite of the fact that there are numerous papers related to the chemistry of β-amino-α-trifluoromethyl alcohols, their chemistry has never been summarized in the form of a review. Therefore, the aim of the present paper is the presentation of reported methods, which were developed for their preparation as well as their most relevant applications in the field of modern organic synthesis.

2. Synthetic approaches to β-amino-α-trifluoromethyl alcohols

2.1. Reduction of cyanohydrines derived from trifluoromethylketones and fluoral

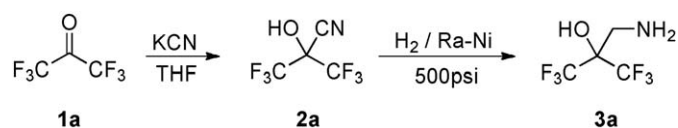
The conversion of hexafluoroacetone (**1a**) into its cyanohydrine **2a** was achieved by treatment of **1a** with potassium or sodium cyanide in THF and subsequent acidification of the solution containing the obtained salt [4] (Scheme 1).

The same method was used for the formation of corresponding cyanohydrines starting with other trifluoromethyl ketones [4d,5] or fluoral [6]. In recent papers, analogous conversions were reported to occur efficiently by using trimethylsilyl cyanide (Me₃SiCN, TMS-CN) instead of the poisonous alkali metal cyanides [7].

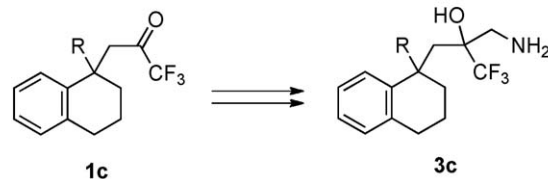
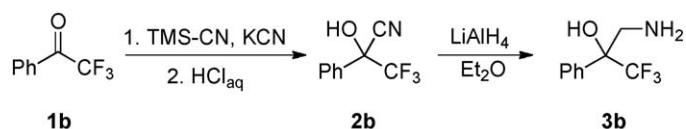
The reduction of the cyano group in cyanohydrines of type **2** offers a convenient route to the desired β-amino-α-trifluoromethyl alcohols. In the case of **2a**, hydrogenolysis over Raney-Ni under high-pressure led to β-amino alcohol **3a** in 75% yield [4e] (Scheme 1).

Cyanohydrine **2b**, derived from α,α,α-trifluoroacetophenone (**1b**) by treatment with TMS-CN, was converted to **3b** by reduction with LiAlH₄ in ether [7a] (Scheme 2). This two-step protocol was successfully applied for the conversion of trifluoromethyl ketones of type **1c** into the corresponding racemic 1-amino-2-(trifluoromethyl)propan-2-ol **3c** [7d] (Scheme 2). The partially deuterated derivatives (e.g. R = CH₂D, CHD₂) were converted into various amides, which were subsequently tested as glucocorticoid agonists.

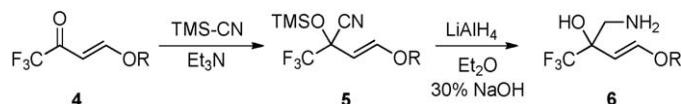
Cyanosilylation of the α,β-unsaturated trifluoromethyl ketone **4** (R = Et) gave the desired 1,2-adduct **5** in high yield, in the presence of catalytic amounts of Et₃N, whereas under acidic conditions the 1,4-adduct was obtained [7b,c] (Scheme 3). Attempted diastereoselective cyanosilylations of ketones of type **4** bearing an enantiopure residue R (e.g. (–)-menthyl) resulted in



Scheme 1.



Scheme 2.



Scheme 3.

poor diastereoselectivity [7c]. The silylated cyanohydrine **5** (R = Et), upon treatment with LiAlH₄ in ether, yielded the β-amino alcohol **6** (R = Et).

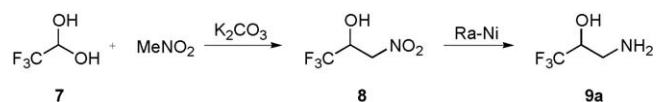
2.2. Henry reaction and subsequent reduction of β-nitro alcohols

The base-catalyzed aldol-type addition of α-CH nitroalkanes with carbonyl compounds is known as Henry reaction [8]. Depending on reaction conditions and the substitution pattern, the primary β-nitro alcohol can be isolated or transferred into the α,β-unsaturated nitro compound. The initially formed β-nitro alcohols are convenient precursors for the preparation of β-amino alcohols.

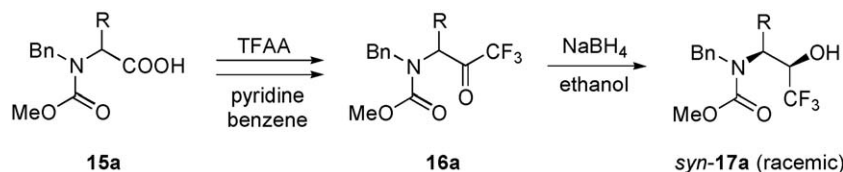
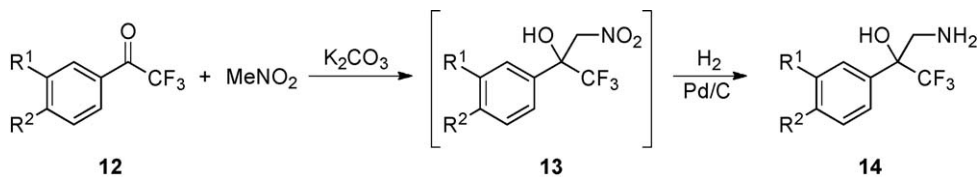
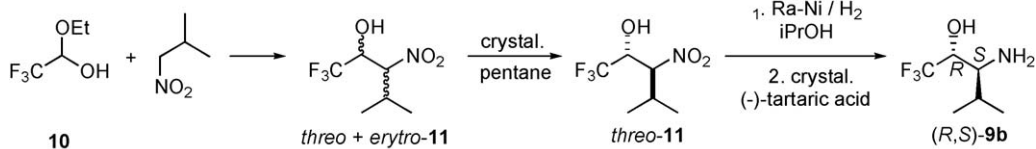
As a result of the electron-withdrawing properties of the trifluoromethyl group, trifluoromethylketones and fluoral are suitable reaction partners for nitroalkanes. In a pioneering work with fluorinated aldehydes, fluoral hydrate (**7**) was reacted with nitromethane in the presence of K₂CO₃ to give the trifluoromethyl β-nitro alcohol **8** in fair yield [9] (Scheme 4). The subsequent reduction with Raney-Ni led to the β-amino alcohol **9a**. In contrast to chlorinated analogues, no reductive replacement of F-atoms by H-atoms was observed. More complex nitroalkanes were used in the reaction with **7** under the same conditions leading, after reduction, to β-amino alcohols of type **9** bearing amino acid side chain residues [10].

In the described reactions, fluoral hydrate can be replaced by its ethyl hemiacetal **10** [11]. In the case of 2-methyl-1-nitropropane, a mixture of four stereoisomeric nitroalcohols (*threo* + *erythro*-**11**) was formed [12]. After fractional crystallization from pentane, the racemic *threo*-**11** was isolated and reduced with Raney-Ni/H₂ in isopropanol. Finally, the resolution of the racemic *threo*-**9b** was achieved by crystallization with (–)-tartaric acid in ethanol (Scheme 5).

The free trifluoroacetone underwent smoothly the reaction with excess nitromethane in the presence of catalytic amounts of Et₂NH or K₂CO₃ yielding 1-nitro-2-trifluoromethylpropan-2-ol [13]. The latter was converted into the corresponding β-amino alcohol upon reduction with Raney-Ni/H₂ at 5 bar [14].



Scheme 4.



A series of substituted α,α,α -trifluoroacetophenones **12** was used in the Henry reaction with nitromethane (cat. K_2CO_3), and the in situ generated β -nitro alcohols **13** were reduced with H_2 over Pd/charcoal under high-pressure to afford the desired β -amino alcohols **14** in good yields [15] (Scheme 6).

In a recent paper, a chiral version of the Henry reaction was reported, in which α,α,α -trifluoroacetophenones and aliphatic trifluoromethyl ketones reacted with nitromethane in the presence of a chiral lanthanide(III)-‘(S)-binolam’ complex to give optically active products with *ee*-values of 67–98% [16]. Selected enantiomerically enriched β -nitro alcohols (80–98% *ee*) were reduced with $NiCl_2/NaBH_4$ in methanol. The obtained β -amino alcohols showed *ee*-values of 72–97%. In the case of the optically active form of **3b** (94.5% *ee*), single crystals suitable for the X-ray crystal-structure determination were isolated and the absolute (*S*)-(+)-configuration was established for this enantiomer.

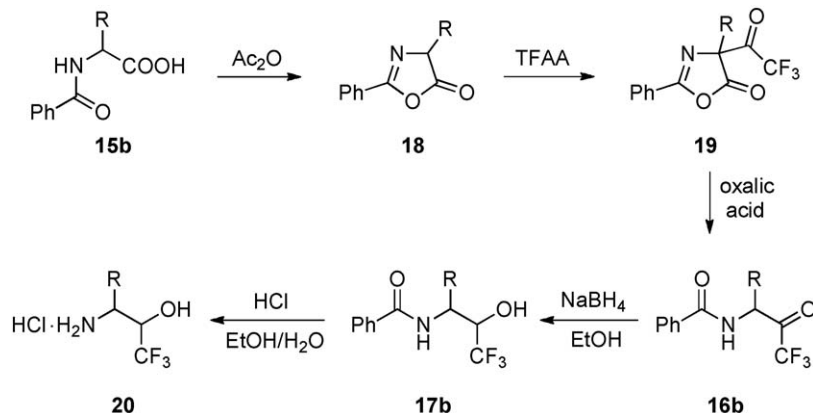
2.3. Reduction of α -aminoalkyl trifluoromethyl ketones

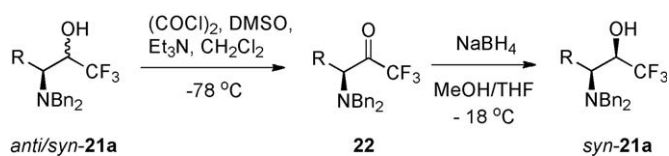
A general method for the preparation of α -amino alcohols is the reduction of the corresponding α -amino ketones [17]. Therefore, the reduction of α -aminoalkyl trifluoromethyl ketones is a

convenient approach to β -amino- α -trifluoromethyl alcohols. A general method for the synthesis of the trifluoromethyl ketones is the Dakin–West reaction [18], in which an α -amino acid derivative, by treatment with trifluoroacetic acid anhydride (TFAA), is converted in a multi-step reaction into the desired α -aminoalkyl trifluoromethyl ketone. The alkyl residue is determined by the amino acid used. For example, a series of *N*-benzyl-*N*-(methoxycarbonyl)- α -amino acids **15a**, was transformed to the trifluoromethyl ketones **16a** [19] (Scheme 7).

Further conversion of **16a** into the amino alcohol *syn*-**17a** occurred diastereoselectively by the reduction with $NaBH_4$ in ethanol. In some instances, however, the reduction was accompanied by the formation of the corresponding 1,3-oxazolidin-2-one. The same conversion was achieved by treatment of *syn*-**17a** with NaH in THF.

A modified version of the presented transformation started with an *N*-benzoyl α -amino acid **15b**, which in the presence of acetic acid anhydride (Ac_2O) underwent cyclization to give the 2,4-disubstituted 1,3-oxazol-5(4*H*)-one **18**. The latter compound was trifluoroacetylated at C(4) and subsequent treatment of the resulting **19** with oxalic acid resulted in the formation of *N*-benzoyl- α -aminoalkyl trifluoromethyl ketone **16b**, which was





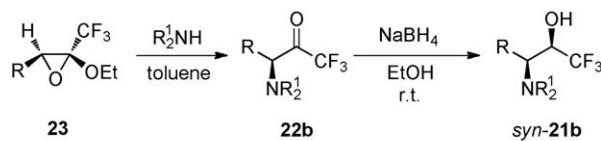
Scheme 9.

finally reduced to give the β -*N*-benzoylamino alcohol **17b** [20] (Scheme 8). Based on the ^{19}F NMR spectra, mixtures of two diastereoisomers of **17b** were obtained. Without separation, these mixtures were hydrolyzed with HCl in water/ethanol to give the desired amino alcohols as hydrochlorides **20**.

An efficient method for the preparation of β -amino- α -trifluoromethyl alcohols is the nucleophilic trifluoromethylation of α -amino aldehydes with (trifluoromethyl)trimethyl silane (TMS-CF_3 , see Section 2.5). The addition occurred diastereoselectively in favour of the formation of the *anti*-isomer. In order to obtain the *syn*-isomer, a two-step procedure was developed [21]. For example, the *N,N*-dibenzyl derivative *anti*-**21a** with $\text{R} = \text{Me}$ was oxidized under Swern conditions to give the corresponding ketone **22**, which then was diastereoselectively converted to *syn*-**21a** (Scheme 9).

In a recent report, the reduction of ethyl 2-dibenzylamino-4,4,4-trifluoro-3-oxobutanoate with KBH_4 was described to yield the corresponding 2-amino-3-hydroxybutanoate in a diastereoselective manner [22].

Another convenient approach to α -aminoalkyl trifluoromethyl ketones was developed via the regioselective nucleophilic ring opening of 2-ethoxy-2-trifluoromethyloxiranes with secondary amines [23] (Scheme 10). In the studied series of aminoketones **22b**, the reduction with NaBH_4 occurred with very high stereoselectivity in favour of the *syn*-isomers. However, replacement of the secondary amines by dimethylaluminium benzylamide (Me_2AlNHBn) and subsequent reduction of the crude aminoketone at -78°C resulted in the formation of *anti*-**21c** as the major product [24]. The authors claimed that in this case an aluminium complex is formed as the intermediate and it governs the inversed stereochemical course of the reaction. On the one hand, the reaction of racemic epoxide **23** with enantiopure dimethylaluminium (*R*)- α -methylbenzylamide led to 1:1 mixtures of the

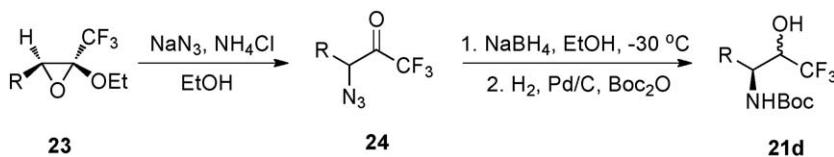


Scheme 10.

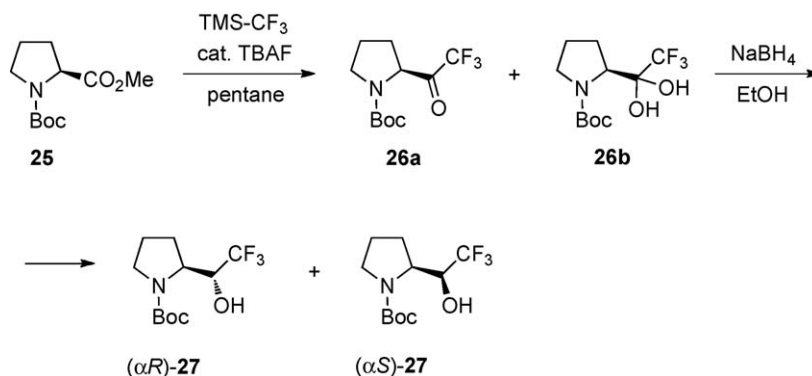
diastereoisomeric products [24]. The analogous reaction carried out with enantiomerically enriched oxirane **23** ($\text{R} = \text{Ph}$) afforded optically active *N*-benzylated amino alcohols of type **21**, which were smoothly debenzylated (H_2 , $\text{Pd}(\text{OH})_2/\text{C}$) yielding the optically active 3-amino-1,1,1-trifluoro-3-phenylpropan-2-ol [24].

In another study, the ring opening of trifluoromethyloxiranes **23** was performed with sodium azide in ethanol in the presence of ammonium chloride [25] (Scheme 11). Treatment of the crude product with NaBH_4 and subsequent hydrogenolysis of the azido group, followed by Boc-protection, led to a mixture of diastereoisomers of **21d** in favour of the *syn*-isomer. The reversed sequence of the reduction steps, *i.e.*, reduction of the azido group and Boc-protection, followed by reduction of the keto function with LiAlH_4 , or NaBH_4 yielded a mixture of **21d**, in which the *anti*-isomer was predominant [25].

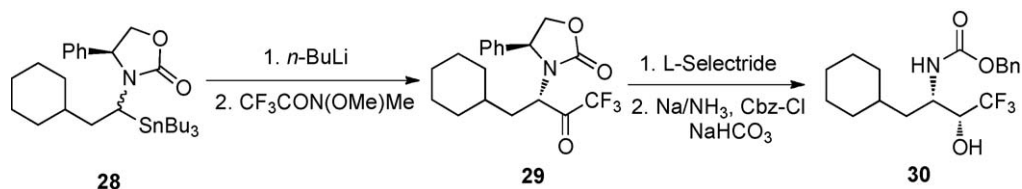
Amino alcohols derived from proline, *i.e.*, prolinols, are of particular interest [26]. However, only little is known about fluorinated representatives. In a recent paper, an elegant approach to trifluoromethyl derivatives was reported [27]. In this case, the trifluoromethyl ketone **26a**, required for the next step, was obtained from *N*-Boc-proline methyl ester (**25**) via substitution of MeO by CF_3 in the reaction with (trifluoromethyl)trimethylsilane (CF_3SiMe_3 , TMS-CF_3) [28] (Scheme 12). The mixture of ketone **26a** and its hydrate **26b** was reduced with NaBH_4 in ethanol at room temperature to give an 85:15 mixture of two diastereoisomers **27**. Carrying out the reduction at 0°C , the diastereoselectivity increased to 96:4.



Scheme 11.



Scheme 12.



Scheme 13.



Scheme 14.

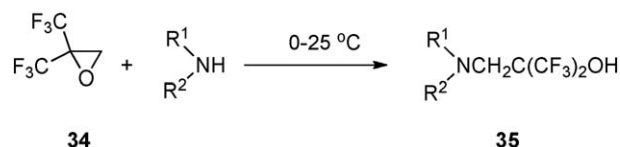
Another approach to proline-based trifluoromethyl ketones, which potentially can be used for the preparation of the corresponding amino alcohols, is the reaction of *N*-protected proline with TFAA in pyridine in the presence of 4-(dimethylamino)pyridine (DMAP, 'Steglich Base'), and subsequent hydrolysis with 10% HCl solution [29].

The diastereoisomeric mixture of the organostannane **28** was transmetalated with *n*-BuLi leading to the lithium compound, which in solution existed predominantly as the 'trans' isomer [30] (Scheme 13). Treatment of this intermediate with the Weinreb-amide of trifluoroacetic acid led to the protected α -aminoalkyl trifluoromethyl ketone **29** diastereoselectively with retention of the configuration. Subsequent diastereoselective reduction of the keto group with *L*-selectride, followed by reductive deprotection and in situ carboxybenzylation of the amino group, gave the 2-protected β -amino- α -trifluoromethyl alcohol **30** with >95% *de*.

2.4. Ring opening of trifluoromethyloxiranes with *N*-nucleophiles

It is well established that oxiranes easily undergo ring-opening reactions with primary and secondary amines [31]. The reactions of trifluoromethyloxirane (**31**) with ammonia or diethylamine yield the trifluoromethyl amino alcohols **9** and 3-diethylamino-1,1,1-trifluoropropan-2-ol, respectively [32]. The ring cleavage occurred regioselectively. Secondary amines of type **32** smoothly react with commercially available trifluoromethyloxirane (**31**) yielding the desired amino alcohols **33** [33]. The ring opening of the oxirane proceeds in the presence of catalytic amounts of ytterbium(III) triflate in acetonitrile with complete regioselectivity (Scheme 14). The biologically most active examples are also prepared in the enantiomerically pure (*R*)-(+)-form.

It was also reported that **31** and ammonium hydroxide at ambient temperature undergo a threefold reaction to give

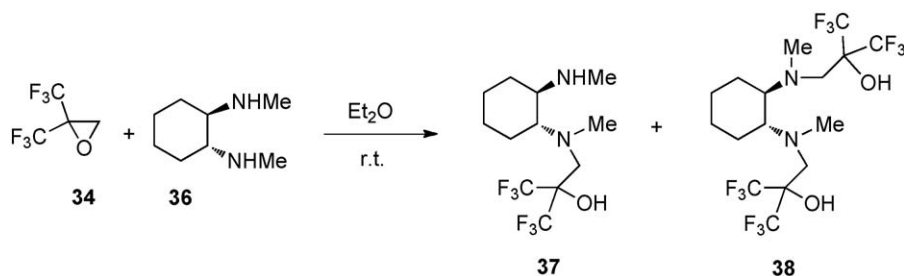


Scheme 15.

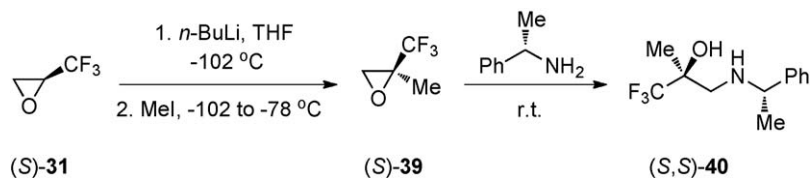
tris(3,3,3-trifluoro-2-hydroxypropyl)amine [34]. Under similar conditions, the reaction of 2,2-bis(trifluoromethyl)oxirane (**34**) and ammonia or ammonium hydroxide yield β -amino- α , α -bis(trifluoromethyl)ethanol as the major product [35]. As a minor product, the secondary amine with two 2-hydroxy-3,3,3,3'-hexafluoroisopropyl residues is formed (ratio 5:1). The latter is the only product when the reaction is carried out at 60 °C. The reactions of **34** with dimethylamine, aniline and 4-fluoroaniline, respectively, occurred already at ambient temperature in the absence of any solvent, and lead to the corresponding *N*-substituted β -amino- α , α -bis(trifluoromethyl)ethanols **35** in 66–83% yield [35] (Scheme 15).

In situ generated **34** (2 mol.-equiv.) reacts at room temperature with 1 mol.-equiv. of (*R,R*)-*N,N'*-dimethylcyclohexane-1,2-diamine (**36**) yielding a mixture of the mono-substituted product **37** (major) and the disubstituted **38** (minor) one. Separation of both amino alcohols was achieved by fractional sublimation followed by recrystallization. Restricted rotation (fluxionality) in **38**, resulting from the intramolecular O–H...N hydrogen bonding, was observed in both ¹H and ¹⁹F NMR spectra at room temperature [36] (Scheme 16).

The optically active oxirane (*S*)-**31** is an attractive building block for the preparation of chiral organofluorine compounds. As an example, initial methylation of enantiomerically enriched (*S*)-**31** (75% *ee*) by treatment with *n*-BuLi and methyl iodide followed



Scheme 16.



Scheme 17.

by aminolysis of the in situ generated (*S*)-**39** with (*S*)- α -methylbenzylamine led to a 9:1 mixture of diastereoisomers with (*S,S*)-**40** as the major product [37] (Scheme 17). An analogous procedure was reported as an excellent method for the preparation of a series of optically active (>90% *ee*) amino alcohols derived from piperidine, starting with (*S*)-**39** (optical purity >99% *ee*) [38].

Similarly, the in situ obtained (*S*)-**39** was trapped with benzophenone and the modified oxirane was cleaved with (*R*)- α -methylbenzylamine [39]. The expected structure of the major diastereoisomer was confirmed by X-ray crystallography.

The oxirane (*S*)-**31** with high *ee*-value (96%) was prepared from 1,1,1-trifluoro-3-bromopropanone in a two-step procedure via enantioselective borane-mediated reduction with β -chlorodiisopinocampheylborane (DIP-Cl) [40]. The reactions with potassium bis(trimethylsilyl)amide or diethylamine afforded the optically active amino alcohol (*S*)-**9** (95% *ee*) and (*S*)-3-(diethylamino)-1,1,1-trifluoropropan-2-ol (96% *ee*) [40]. An alternative approach to (*S*)-**9** was the ring opening of (*S*)-**31** with sodium azide in ethanol and subsequent reduction with hydrogen over Pd/C. Also in this case, both reaction steps occurred with high stereoselectivity.

The enantiomerically pure oxirane (*S*)-**31** reacts easily with *N*-methylallylamine (**41**) in acetonitrile solution without a catalyst to give stereoselectively the 'allylic' amino alcohol (*S*)-**42** in 90% yield (Scheme 18) [41].

The presence of the *p*-tolylsulfinyl methyl auxiliary attached to C(2) of trifluoromethyloxirane governs efficiently the regio- and stereoselective opening of the three-membered ring. The oxiranes are available by treatment of the corresponding trifluoromethylketones with diazomethane at room temperature [42]. Depending on the solvent, the ratio of oxirane and the isomeric methyl vinyl ether varied from 1:1 (in benzene) to 63:1 (in ethanol) and, in all cases, the (*S,R*)-diastereoisomeric oxirane was formed in high excess (>97:3).

The reactions of oxirane (*S,R*)-**43** with benzyl- or dibenzylamine were carried out at room temperature in THF and the corresponding amino alcohols (*S,R*)-**44** were obtained in high yields [42] (Scheme 19). Upon treatment with typical agents, the sulfinyl auxiliary was removed and diverse derivatives of the amino trifluoromethyl alcohols **45–47** were obtained.

Analogous conversions with higher substituted sulfinylmethyl residues (CHMe, CHPr, CHPh instead of CH₂) were reported by the same research group [43] and reviewed in a recent survey [44]. In addition to conversions presented in Scheme 19, (*S,S,R*)-**48** yielded, after heating in boiling xylene, the optically active vinyl-substituted amino alcohol (*S*)-**49** in 82% yield [43] (Scheme 20).

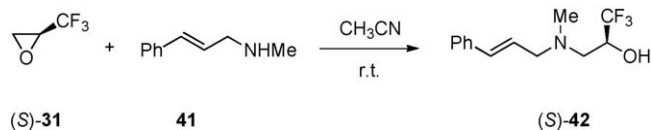
2.5. Nucleophilic trifluoromethylation of α -amino aldehydes and ketones

The nucleophilic addition of (trifluoromethyl)trimethylsilane (TMS-CF₃, Ruppert–Prakash reagent [45]) to a carbonyl group followed by desilylation of the initial adduct is a powerful and clean method for the synthesis of α -trifluoromethyl alcohols. In a typical procedure, anhydrous ethers are used as solvents, and the fluoride ion is used as a catalyst. However, recent studies showed that the reaction can be performed without fluoride in DMF/K₂CO₃ [46a] or in DMSO only [46b].

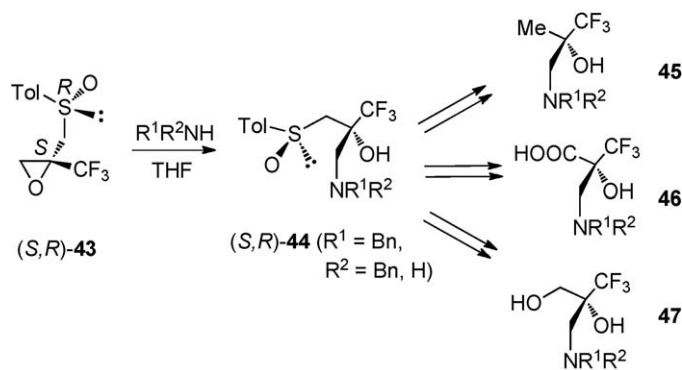
The reaction of TMS-CF₃ with α -amino aldehydes offers an attractive approach to β -amino- α -trifluoromethyl alcohols. For example, the *N*-protected α -amino aldehydes **51**, which were prepared by reduction of the Weinreb-amides **50**, were treated with TMS-CF₃ in the presence of tetrabutylammonium fluoride (TBAF) and, after acidic hydrolysis, amino alcohols **52** were obtained in 40–50% yield [47] (Scheme 21). The same reaction sequence was applied in the case of Boc-protected *L*-phenylalanine [11a]. The obtained trifluoromethylated amino alcohol was identified as a mixture of the two diastereoisomers.

A two-step procedure for the synthesis of another series of *N*-protected α -amino aldehydes is outlined in Scheme 22 [48]. The *N*-protected α -amino acid ester **53** was reduced to the alcohol and the latter was selectively oxidized to **54**. The addition of TMS-CF₃ was reported to occur diastereoselectively and the amino alcohol **55** was finally obtained, after reduction/deprotection, as a 4.5:1 mixture of the diastereoisomers in favour of **55**.

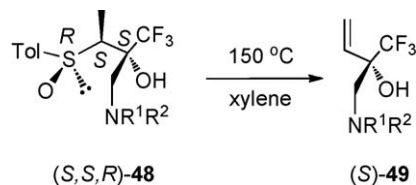
A series of enantiomerically pure *N,N*-dibenzyl- α -amino aldehydes **56** (R = alkyl or aryl) was used for the reaction with TMS-CF₃ to examine the influence of the substituent R on the stereochemical outcome of the reaction [21] (Scheme 23). Whereas in the case of alkyl groups the ratio of *anti/syn*-**21a** was between



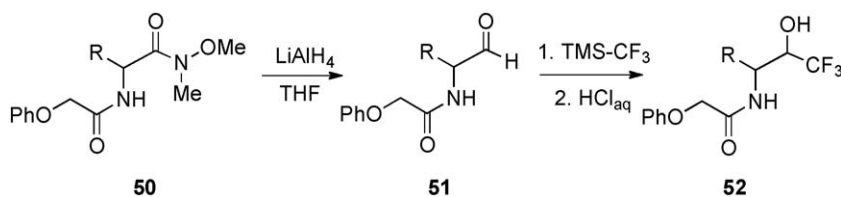
Scheme 18.



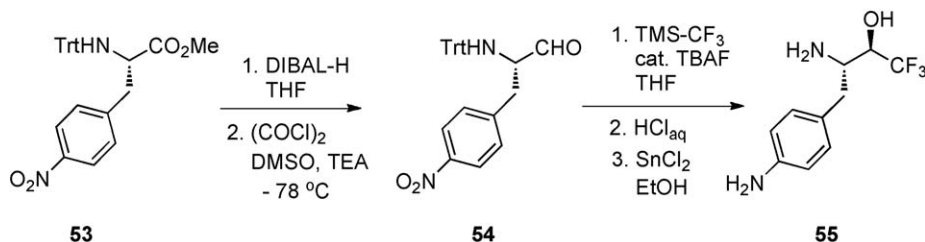
Scheme 19.



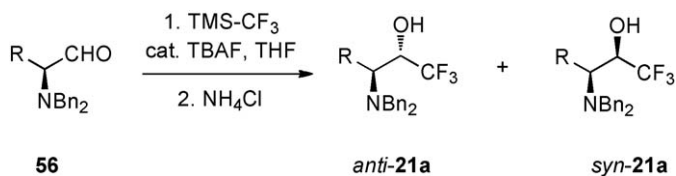
Scheme 20.



Scheme 21.



Scheme 22.



Scheme 23.

84:16 and 62:38, the phenylglycinal derivative **56** (R = Ph) gave a 46:54 mixture of the *anti/syn*-isomers.

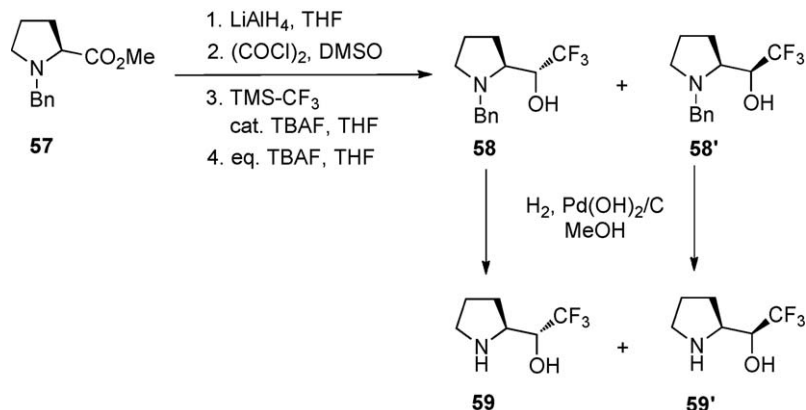
The pure *anti-21a* was modified by the replacement of the *N*-benzyl groups by Me₂, Et₂, -(CH₂)₄- and -(CH₂)₅-, respectively, in order to obtain a series of compounds which were tested as catalysts of enantioselective Reformatzky reactions [49]. Starting with L-isoleucine, the *N,N*-dimethylisoleucine methyl ester was prepared in a two-step procedure. Subsequent reduction and Swern oxidation, followed by treatment with TMS-CF₃ and desilylation afforded the *anti*- and *syn*-isomers of type **21a** in 37 and 7% yield, respectively.

The analogous procedure starting with methyl *N*-benzyl L-prolinate (**57**) yielded a mixture of comparable amounts of the isomeric amino alcohols **58/58'**, which, after separation, were debenzylated to give **59** and **59'** [49] (Scheme 24). Finally, these amino alcohols were converted into the *N*-methyl derivatives.

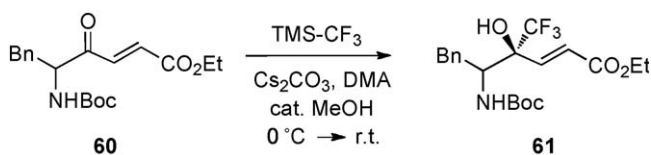
The *N*-Boc-protected proline reacts under typical conditions (TBAF, dry THF, 0 °C to room temp.) with TMS-CF₃ yielding a mixture (ratio 1:1.2) of the corresponding diastereoisomeric amino alcohols **27** (Scheme 12), analogous to **58** and **58'**, which by treatment with trifluoroacetic acid in CH₂Cl₂ were converted into **59** and **59'** [50].

In contrast to α-amino aldehydes, reactions of TMS-CF₃ with α-amino ketones were only scarcely studied. In a very recent report, compound **60**, derived from 5-amino-4-oxohex-2-enoate, was described to react with TMS-CF₃ in *N,N*-dimethyl acetamide (DMA) in the presence of alkali metal carbonates. Under these conditions the desilylation occurred spontaneously, and the desired trifluoromethylated product **61** was obtained in fair yield [51] (Scheme 25). It is worth mentioning that the reaction proceeded chemoselectively at the keto group. Starting with enantiomerically pure **60**, a mixture of *syn* and *anti*-diastereoisomers was isolated, and the *de* values for **61** varied between 61 and 83% depending on the reaction conditions. The presence of a chiral additive (BINOL) did not effect the *de* value.

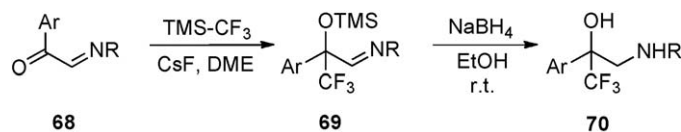
Ketones **62** (R¹ = Bz or Boc), derived from proline, reacted with TMS-CF₃ in DMSO in the presence of CsF at room temperature yielding, after desilylation with TBAF in THF, mixtures of diastereoisomers of the corresponding trifluoromethylated prolinol **63** [52] (Scheme 26). The ratios of the diastereoisomers were determined by ¹⁹F NMR spectroscopy to ca. 1:1.



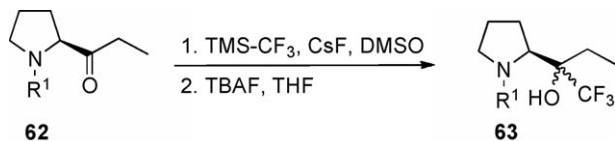
Scheme 24.



Scheme 25.



Scheme 28.



Scheme 26.

2.6. Nucleophilic trifluoromethylation of α -iminocarbonyl derivatives

Reactions of TMS-CF₃ with α -imino ketones offer a straightforward access to the corresponding trifluoromethylated amino alcohols. However, similarly to α -amino ketones, this class of substrates was explored to a limited extent, only. Monoimines **64**, derived from camphorquinone, are easily available and stable under standard conditions. Nucleophilic trifluoromethylations of **64** occur smoothly at room temperature in dimethoxyethane (DME) in a chemo- and diastereoselective manner. The ¹⁹F NMR spectra of the crude product confirmed the formation of a single diastereoisomer **65** in all cases studied [53] (Scheme 27). Desilylation leading to the imino alcohols **66** was achieved by treatment with NaBH₄ in boiling ethanol. Under these conditions, the C=N bond was not reduced and only the reduction with DIBAL-H led to the *exo,exo*-isomer of the α -trifluoromethylated amino alcohols **67** in a diastereoselective manner. The studies were carried out with monoimines **64** bearing an alkyl or aryl residue at the *N*-atom.

Imino ketones of type **68**, which are quite labile compounds obtained from 2-aryl-2-oxoethanol hydrates (arylglyoxal hydrates) and primary amines, reacted easily with TMS-CF₃ with complete chemoselectivity to afford the *O*-silylated α -trifluoromethyl alcohols **69** in high yields [54] (Scheme 28). Typically, the reaction was performed in DME/CsF, but comparable yields were obtained in DMF/K₂CO₃ or in DMSO without any additive. In contrast to adducts **65**, treatment of **69** with NaBH₄ in ethanol at room temperature led directly to the amino alcohol **70**.

Compounds **71**, *i.e.* α -imino lactones, contain the structure fragment of an α -imino carbonyl derivative. They reacted with TMS-CF₃ in THF at 0 °C in the presence of an activator, and the corresponding products **72** were obtained exclusively [55] (Scheme 29). The latter were transformed to amino alcohols **73** by treatment with LiBH₄ in THF. Subsequent hydrogenolysis, Boc-protection, and acidic deprotection led to the hydrochlorides of the *anti*-diastereoisomers **74** in a diastereoselective manner. In an alternative approach, imino lactones **71** were stereoselectively hydrogenated and *N*-benzylated to give the corresponding *cis*-disubstituted α -amino lactones **75**. Trifluoromethylation followed

by treatment with NaBH₄ in methanol/water gave the diastereoisomers **76**, which finally were converted to *syn*-**74**.

The enantiomerically pure β -amino alcohol **77** derived from (+)-pulegon is a useful auxiliary for a highly diastereoselective multi-step procedure applied for the preparation of α -hydroxy aldehydes [56]. Condensation of **77** with glyoxal derivatives (R = alkyl, aryl) led to the intermediate products **78**, which, upon treatment with TMS-CF₃ and TBAF, were converted into trifluoromethyl alcohols **79** with diastereoselectivities between 11:1 and >50:1 [57] (Scheme 30).

Selective reduction with AlH₃ in THF resulted in the cleavage of the C,O bond of the aminoacetal to give **80**. Finally, removal of the auxiliary by treatment with pyridinium chlorochromate (PCC) followed by NaOH afforded the *N*-benzylamino alcohols **81** in excellent yields [57].

2.7. Pétasis reaction of 2-hydroxy-3,3,3-trifluoropropanal

An attractive multi-step approach to diverse β -amino- α -trifluoromethyl alcohols offers the Pétasis reaction [58] with 2-hydroxy-3,3,3-trifluoropropanal (**83**), boronic acids and secondary amines [59] (Scheme 31). The key compound **83** was synthesised via nucleophilic addition of TMS-CF₃ to cinnamic aldehyde, subsequent ozonolysis and typical workup with dimethyl sulfide. Remarkably, the Pétasis reaction leading to **84** occurred with high diastereoselectivity.

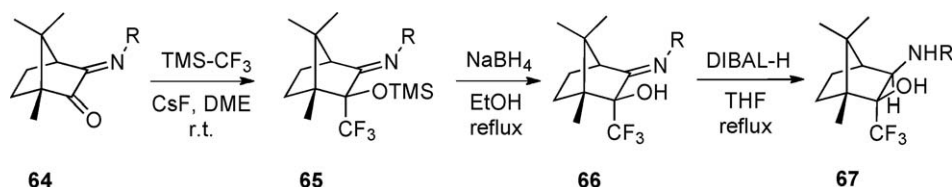
In the extension of the study, enantiomerically enriched (*R*)- and (*S*)-**83** were prepared and used for the reaction with dibenzylamine and 2-bromo-2-phenylvinyl boronic acid. The best result with almost complete diastereoselectivity was obtained in a 1:4 mixture of methanol and dichloromethane without chromatographic purification of the intermediate hydroxyaldehyde **83**, leading to the product **84** (R¹ = R² = Bn, R³ = Ph(Br)C=CH₂) with *ee*-values up to 92%.

2.8. Miscellaneous reactions

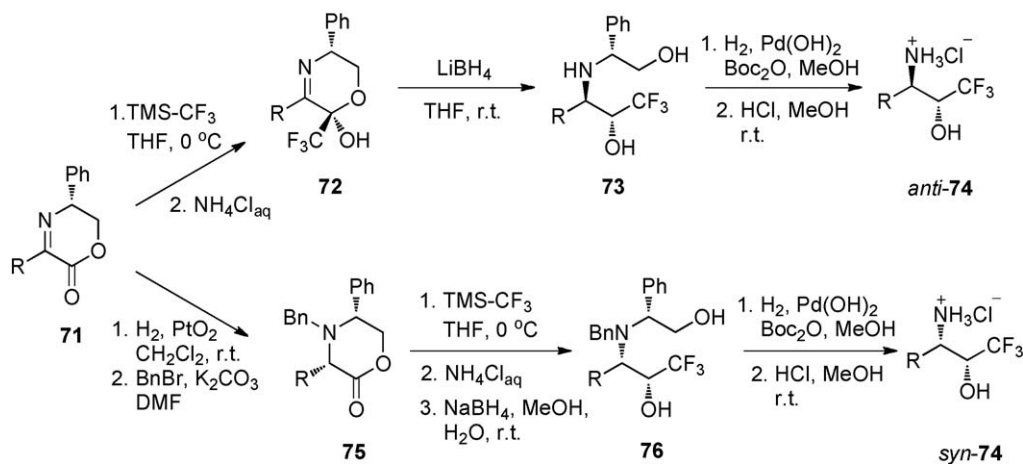
Some less general methods for the preparation of β -amino- α -trifluoromethyl alcohols from CF₃-containing starting materials are also known. In one case, ethyl trifluoroacetyl acetate (**85**) was converted into the *O*-protected 4,4,4-trifluoro-3-hydroxybutanoic acid **86** [60] (Scheme 32). This intermediate was reacted with diphenyl phosphorazidate (DPPA) in the presence of Et₃N and benzyl alcohol yielding the protected amino alcohol **87**.

The presented protocol was used for the conversion of 2-alkyl-4,4,4-trifluoro-3-hydroxybutanoates, which are accessible via LDA-mediated aldol addition of a carboxylic acid with fluoral.

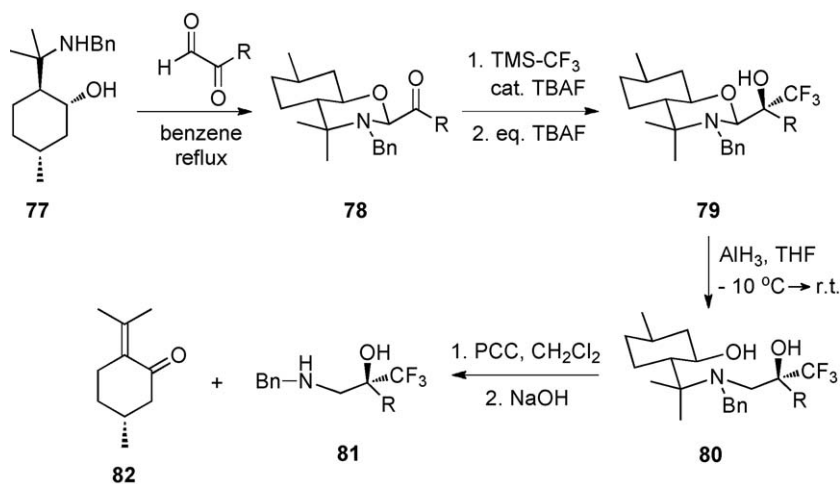
In the second case, the conversion of the unsaturated ether **88** to the amino alcohol **91** was achieved via initial Sharpless



Scheme 27.



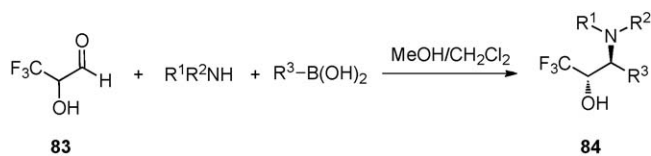
Scheme 29.



Scheme 30.

dihydroxylation to give **89**, transformation to the azide **90**, reduction of the azido group, and in situ Boc-protection of the amino group [61] (Scheme 33). Compound **91** was further transformed into the amino alcohol **92** and finally to the trifluoromethylated α -amino acid **93**.

Heating of bis(trifluoromethyl)carbinols **94a–b** in a two-phase system of toluene and 20% aqueous NaOH, in the presence of tetrabutylammonium fluoride (TBAF) as the catalyst, was reported

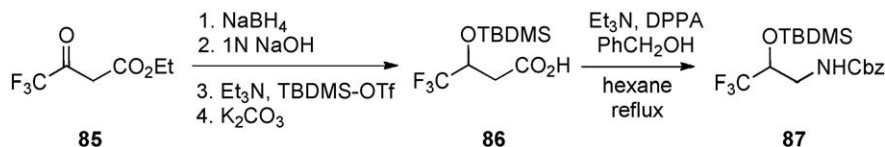


Scheme 31.

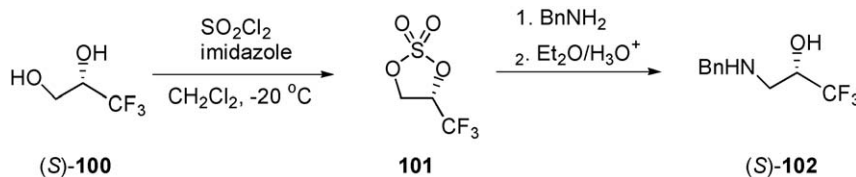
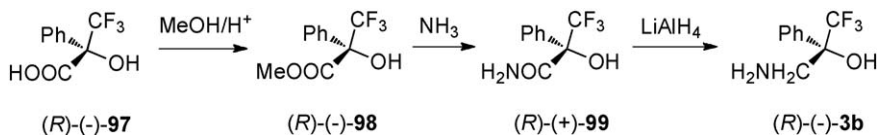
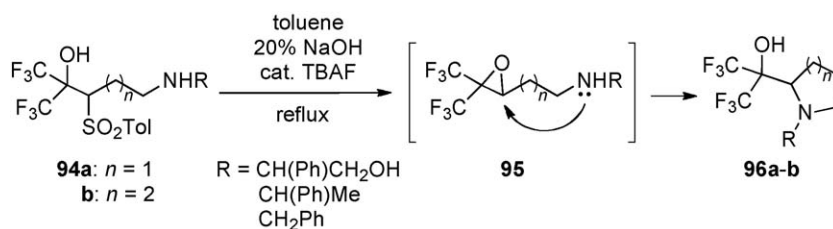
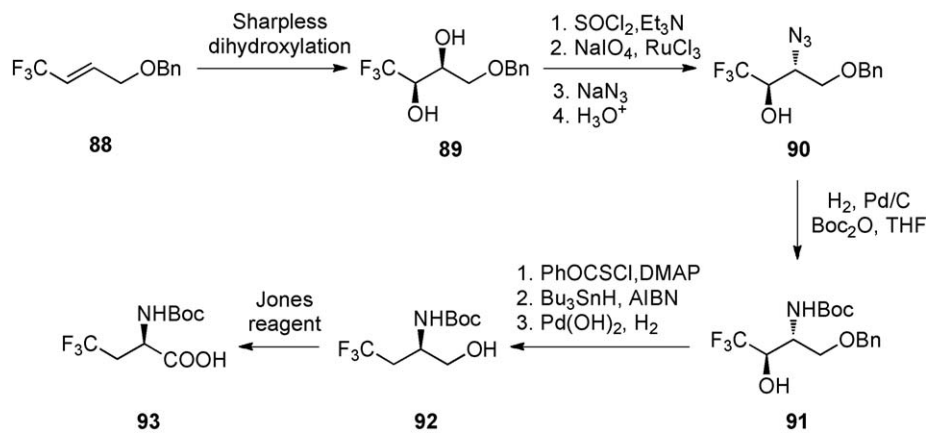
to yield pyrrolidines **96a** ($n = 1$) and piperidines **96b** ($n = 2$) in satisfactory to good yields (50–90%) [62] (Scheme 34). The reaction course was explained via initial formation of oxiranes **95** and subsequent cyclization by intramolecular nucleophilic substitution. The analogous protocol applied for the attempted synthesis of the corresponding seven-membered azepine derived amino alcohol ($n = 3$) failed. In an additional experiment, the appearance of the intermediate oxirane derivative **95** was convincingly evidenced. Thus, the reaction follows a two-step mechanism via oxirane formation/oxirane ring-opening sequence.

Enantiomerically enriched (*R*)-(-)-3-amino-1,1,1-trifluoro-2-phenylpropan-2-ol (**3b**) was prepared via a 3-step synthesis starting with (*R*)-(-)-2-hydroxy-3,3,3-trifluoro-2-phenylpropanoic acid (**97**) (Scheme 35), and its optical purity was proved by NMR spectroscopy with a chiral europium complex [63].

The optically active diol (*S*)-**100** can be prepared from 3,3,3-trifluoropropene by osmium-catalyzed asymmetric dihydroxylation



Scheme 32.



in the presence of phthalazine and pyrimidine ligands [64]. Subsequent reaction of (*S*)-**100** with sulfonyl chloride in the presence of imidazole leads to 1,3,2-dioxathiolane **101**, which upon treatment with benzylamine and in situ performed acidic hydrolysis provides enantioselectively amino alcohol (*S*)-**102** in 75% yield (Scheme 36). The same method was applied to produce the corresponding CCl_3 -analogue.

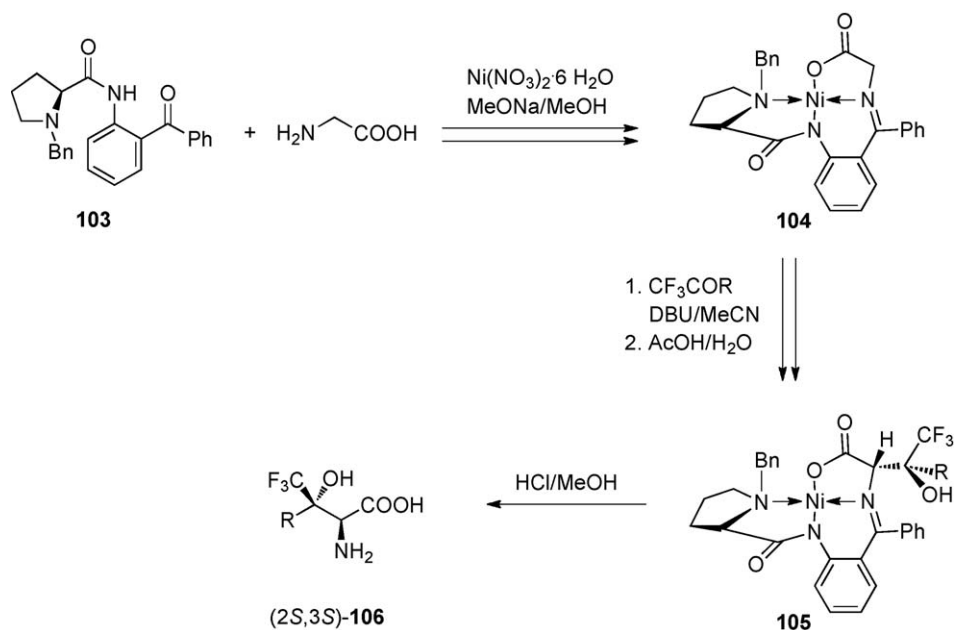
A highly diastereoselective asymmetric aldol reaction was elaborated for the synthesis of trifluoromethylated threonine and analogous amino acids [65]. The monochiral Schiff base–Ni(II) complex **104**, which is conveniently accessible from (*S*)-*ortho*-[*N*-(*N*-benzylpropyl)amino]benzophenone (**103**) and glycine, reacts with trifluoromethylketones in the presence of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) to yield the aldol adduct **105** with high to excellent diastereoselectivity (*de* 95 to >98%). Hydrolysis leads to the amino acids (2*S*,3*S*)-**106** and the chiral auxiliary **103** in very good chemical yields (Scheme 37).

3. Applications of β -amino- α -trifluoromethyl alcohols in organic synthesis

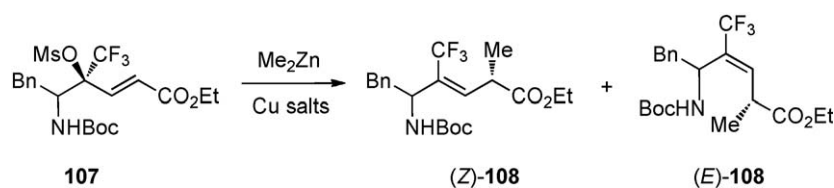
3.1. Transformations of β -amino- α -trifluoromethyl alcohols into trifluoromethylated carboxylic acids and heterocycles

Diverse transformations of amino alcohols described in Section 2 are known and they lead to either linear or cyclic products. Thus, in the case of the protected amino alcohol **91**, dehydroxylation by the Barton–McCombie method and debenzoylation afforded amino alcohol **92**, which, after oxidation with Jones reagent, gave the enantiomerically pure α -amino acid derivative **93** [61] (Scheme 33).

Mesylation of **61** to give **107** and subsequent treatment with dimethylzinc or methyl magnesium chloride in the presence of copper(I) or copper(II) salts led to the α -methyl- β,γ -unsaturated esters (*Z*)- and (*E*)-**108** [51] (Scheme 38). The major product (*Z*)-



Scheme 37.



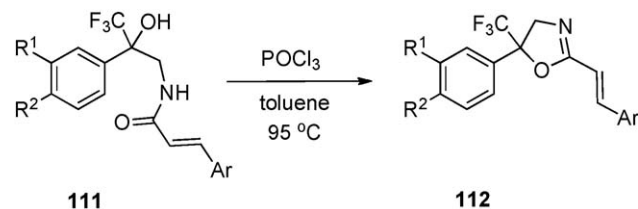
Scheme 38.

108 was formed with lower diastereoselectivity (20–86% *de*) than the minor (*E*)-configured product (88–99% *de*). Analogous alkylations were carried out with other zinc-organic compounds.

Compound **109**, i.e. ethyl 2-dibenzylamino-4,4,4-trifluoro-3-hydroxybutanoate, after tosylation and base-induced elimination was converted into (*Z*)-but-2-enoate **110** [22] (Scheme 39). Debenzylation of **109** followed by twofold benzoylation and elimination according to Scheme 39 afforded the corresponding *N*-benzoyl derivative of type **110**, which was used for the cyclopropanation with diazomethane. On the other hand, the attempted cyclopropanation with dimethyl sulfoxonium methylide unexpectedly gave a 4,5-dihydro-4-(2,2,2-trifluoroethyl)-1,3-oxazole-4-carboxylate [22].

A series of 4,5-dihydro-1,3-oxazoles of type **112** was obtained from amino alcohols **14** (Scheme 6) via cinnamoylation to **111** and subsequent treatment with POCl_3 in toluene at 95 °C [15] (Scheme 40).

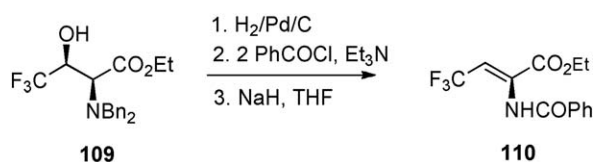
Numerous conversions of α -trifluoromethylated β -amino alcohols into the corresponding 1,3-oxazolidin-2-ones **113** were reported. They were performed via carbonylation with phosgene



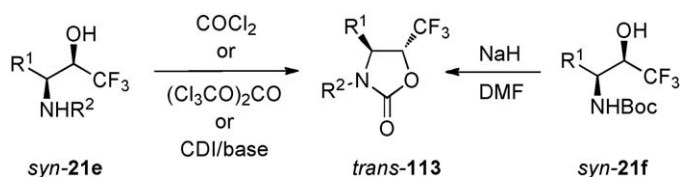
Scheme 40.

[24,66], triphosgene [21], and carbonyldiimidazole [51], respectively, or via base-induced 1,5-cyclization of the corresponding *N*-Boc-protected derivatives [19,25,55] (Scheme 41). Irrespective of the applied method, the *anti*-amino alcohol led to the *cis*-4,5-disubstituted product and the *syn*-isomer to the *trans*-disubstituted oxazolidinone.

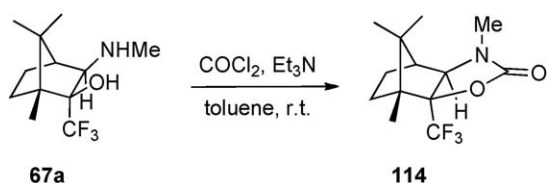
The carbonylation of the trifluoromethylated amino alcohol **67a** with the bornane skeleton by treatment with phosgene, leading to the oxazolidinone **114**, proved the *exo* orientation of both the hydroxyl and amino groups [53] (Scheme 42).



Scheme 39.



Scheme 41.



Scheme 42.

Cyclization of amino alcohols **70** (Scheme 28) with formaldehyde, thionylchloride or dichlorophenyl phosphane led to the formation of trifluoromethylated derivatives of 1,3-oxazolidine **115**, 1,2,3-oxathiazolidine **116**, and 1,3,2-oxazaphospholidine **117**, respectively [66] (Scheme 43). On the other hand, treatment of **70** with Ph_3PCl_2 in boiling acetonitrile in the presence of triethylamine gave the corresponding aziridines **118**. Finally, the reaction of **70** with oxalyl chloride afforded the 1,4-oxazidine-2,3-dione **119**.

The unsaturated amino alcohols **6a** were hydrolyzed and the in situ formed γ -amino- β -hydroxyalkyl trifluoromethyl carbonyl compounds **120** underwent spontaneously cyclization and dehydration leading to 3-trifluoromethyl pyrroles **121** [7e] (Scheme 44).

3.2. Preparation of peptidomimetics and other biologically active compounds

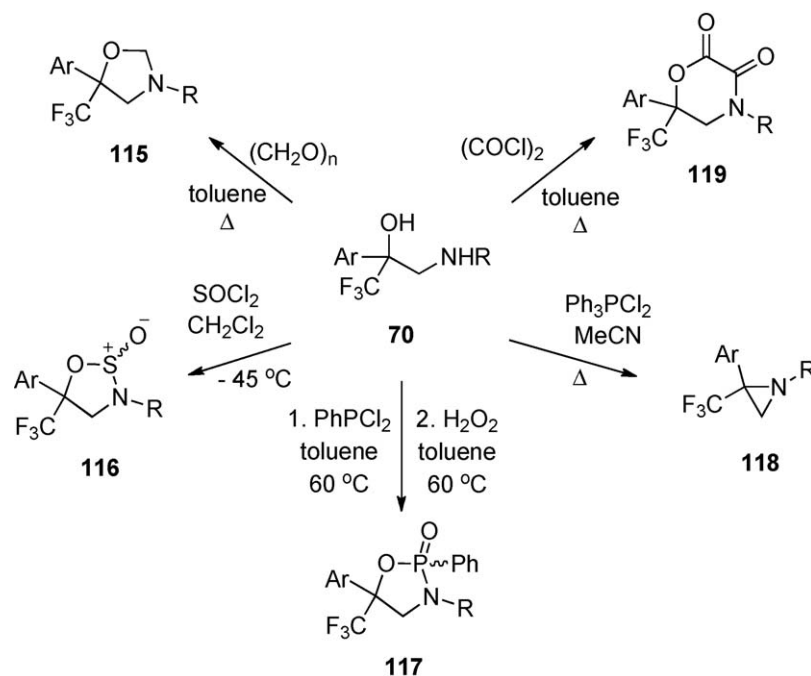
It is a well-established fact that the trifluoromethyl group can significantly enhance the biological activity of organic compounds and, therefore, trifluoromethylated products are of high interest in

life sciences [67]. In general, the trifluoromethyl group is recognized as an important pharmacophore displaying a key role in modulation of biological activity of the studied compounds, e.g. in binding ability and agonist activity of a glucocorticoid receptor ligand [68]. With regard to the electrostatic properties, the trifluoromethyl group is evaluated as an equivalent of the phenyl group [69]. Therefore, trifluoromethylated β -amino alcohols can be considered as compounds, which structurally mimic α -hydroxy- β -phenylethylamine, well known as β -agonist and/or β -antagonist.

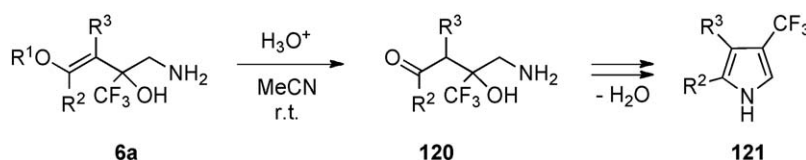
For these reasons, β -amino- α -trifluoromethyl alcohols are very attractive building blocks for the synthesis of biologically active compounds. To the best of our knowledge, there are only few examples known, e.g. 3-amino-1,1,1-trifluoro-2-phenylpropan-2-ol, which showed biological activity without chemical modification [7a]. Another example offers a series of *N*-[3-(1,1,2,2-tetrafluoroethoxy)benzyl]-*N*-(3-phenoxyphenyl)trifluoro-3-aminopropan-2-ols of type **33** (Scheme 14), which potently and reversibly inhibit cholesteryl ester transfer protein (CETP) [33]. In most cases, the amino alcohols were converted into amides and eventually oxidized to the trifluoromethylated ketones.

A large series of carboxamides and a sulfonamide of type **122** and **123**, respectively, derived from amino alcohols **3c** (Scheme 2), were tested as non-steroidal glucocorticoid agonists [7d]. In some cases, aryl analogues of type **124** were also investigated.

A mixture of the diastereoisomeric amino alcohols **59** and **59'** (Scheme 24) was treated with a sulfonyl chloride derived from isatin, and the diastereoisomeric sulfonamides **125** obtained thereby were evaluated for their caspase-inhibition potency.

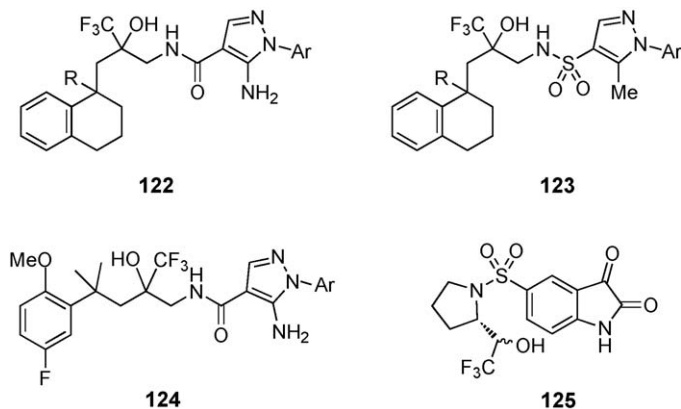


Scheme 43.



Scheme 44.

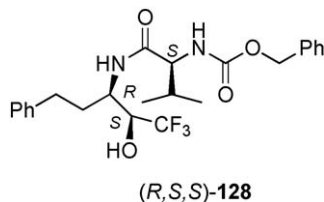
Unfortunately, the results obtained in this case were not encouraging [50].



The unsaturated protected amino alcohol **126** derived from **6** (Scheme 3) was hydrolyzed to the corresponding aldehyde, which then was oxidized and deprotected yielding finally 3-trifluoromethyl-4-amino-3-hydroxybutanoic acid (**127**) as a racemate [7c] (Scheme 45).

Both, amino alcohols **6** and carboxylic acid **127**, were resolved to give the pure enantiomers. The carboxylic acids **127** are recognized as β -hydroxy- β -trifluoromethyl derivatives of γ -aminobutyric acid (GABA).

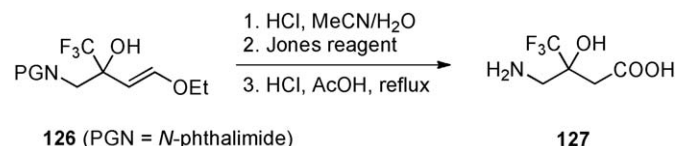
The *syn/anti*-amino alcohols **21d** ($R = \text{PhCH}_2\text{CH}_2$) (Scheme 11) were separated and, after deprotection, converted into mixtures of diastereoisomeric dipeptide-like products **128** [25]. Chromatographic separation led to all four diastereoisomers in pure form, which were tested as inhibitors of human leucocyte elastase.



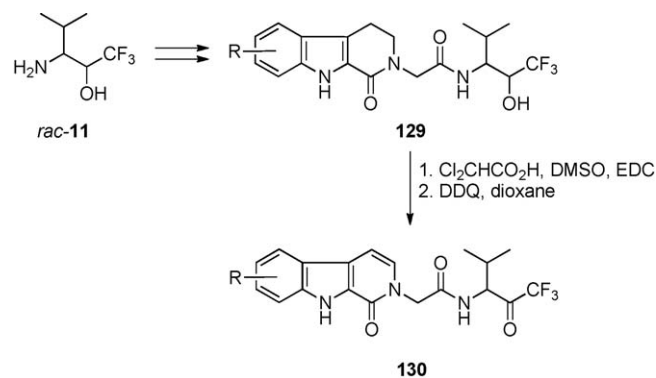
Trifluoromethyl ketones of type **130** containing the β -carbolinone skeleton were tested as potential non-peptidic inhibitors of human leucocyte elastase [70]. The key step of their synthesis was the oxidation of the secondary alcohol fragment in **129** originating from the starting trifluoromethylated amino alcohol (Scheme 46). This oxidation was performed using a modified Pfitzner-Moffatt protocol. Finally, dehydrogenation with dichlorodicyanobenzoquinone (DDQ) led to the target molecules.

An analogous strategy with oxidation of a α -trifluoromethyl alcohol fragment leading to the trifluoroacetyl group was described in numerous reports on preparations of trifluoroacetyl peptidomimetics. They were tested as renin inhibitors [60], inhibitors of human leucocyte elastase [11a,12], metallo- β -lactamase inhibitors [47b], serine protease inhibitors [10], inhibitors of the human cytomegalovirus protease [11b], human plasma kallikrein [48], and SARS-CoV 3CL protease [11d].

In another paper, peptidyl trifluoromethyl ketones were investigated as inhibitors of porcine pancreatic elastase, human



Scheme 45.

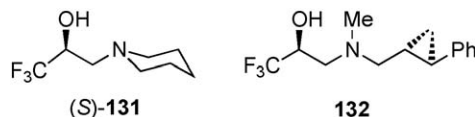


Scheme 46.

neutrophil elastase, and rat and human neutrophil cathepsin G [20]. The oxidations of the α -trifluoromethyl alcohol residue to the trifluoroacetyl function were carried out by using different protocols, such as Dess–Martin periodinane oxidation, Swern and Pfitzner-Moffatt reactions as well as potassium permanganate oxidation under basic aqueous conditions. The incorporation of the α -trifluoromethyl- β -amino alcohol into the peptidomimetics was also performed by using solid-phase methodology [11b,c].

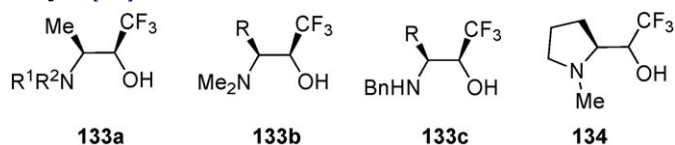
3.3. Application in stereocontrolled synthesis

The enantiomerically pure β -amino- α -trifluoromethyl alcohols of type (*S*)-**131** turned out to be more efficient ligands for the enantioselective addition of diethylzinc to benzaldehyde than the non-fluorinated analogue [38,71]. The *ee*-values increased up to >90% (ca. 45% with the non-fluorinated analogue) with increasing concentration of the ligand.



By treatment with bis(iodomethyl)zinc (ICH_2)₂Zn in dichloromethane solution (-5°C , 1 h) (Simmons–Smith reaction), the ‘allylic’ amino alcohol (*S*)-**42b** (Scheme 18) easily undergoes cyclopropanation to give **132** in 86% yield and with very high diastereoselectivity (>99%). Acceleration of the reaction rate results from the electron-withdrawing effect of the trifluoromethylated amino alcohol auxiliary [41].

A series of differently substituted α -trifluoromethyl- β -amino alcohols **133–134** was applied as chiral ligands in enantioselective additions of a Reformatzky reagent to benzaldehyde and in diethylzinc additions to *N*-(diphenylphosphinoyl)imine of benzaldehyde [49].



In the first reaction, the *ee*-values varied between 39 and 86%, and the best result (chemical yield 89%) was obtained with **133a** ($R^1 = R^2 = \text{Et}$). The reactions with the two diastereoisomers of **134** led to enantiomeric products with comparable yields and *ee*-values. In the diethylzinc addition to the imine bond, the results were even better, e.g. with **133c** ($R = t\text{-Bu}$), *ee*-values of 95% were obtained.

Complexes of amino alcohol **3a** (Scheme 1) with Cu^{2+} , Ni^{2+} , and Co^{2+} were obtained and their stoichiometry was determined as 2:1 [4e].

The chiral fluorinated diamino-diol **38** (Scheme 16) reacts smoothly as a proligand with $Zr(CH_2Ph)_4$ and $Ti(OiPr)_4$ forming the corresponding (dialkoxy)zirconium and (dialkoxy)titanium complexes. The cationic zirconium complex was tested as a new, efficient catalyst for the regioselective polymerization of 1-hexene [36].

4. Conclusions

The results summarized in the present review evidence that β -amino- α -trifluoromethyl alcohols form an important subclass of vicinal amino alcohols. The presence of the trifluoromethyl group enables their exploration in the synthesis of more complex molecules with special properties. On the other hand, both, the amino and the hydroxy group can easily be transformed to other functional groups. Of special importance is the fact that many of the trifluoromethylated alcohols can be prepared in enantio- and diastereomerically pure form. For that reason, they are promising substances for stereocontrolled reactions. However, their exploration in this part of organic synthesis is limited to a few cases only. Similarly, syntheses of new heterocycles via selective reactions of the OH or NH_2 group are almost unknown.

The incorporation of β -amino- α -trifluoromethyl alcohols into the backbone of peptidomimetics, and eventually further oxidation of the secondary alcohol function to the trifluoroacetyl group, results in a remarkable increase of the biological activities of the obtained products.

The presented review supplements the survey on the synthesis and biological activity of structurally similar β -amino- β -fluoroalkyl alcohols published in 1999 [72].

Acknowledgments

The authors thank the Polish Ministry for Science and Higher Education for a grant (Grant # N N204 130335) (G.M. and E.O.), and financial support by F. Hoffmann-La Roche AG, Basel, is gratefully acknowledged (H.H.).

References

- [1] (a) B.F. Kukharev, V.K. Stankevich, G.R. Klimenko, Russ. Chem. Rev. (Engl. Transl.) 64 (1995) 523–540; (b) D.J. Ager, I. Prakash, D.R. Schaad, Chem. Rev. 96 (1996) 835–875; (c) M. Periasamy, Pure Appl. Chem. 68 (1996) 663–666; (d) C.H. Senenayake, Aldrichim. Acta 31 (1998) 3–15; (e) F. Fache, E. Schulz, M.L. Tommasino, M. Lemaire, Chem. Rev. 100 (2000) 2159–2232; (f) S.C. Bergmeier, Tetrahedron 56 (2000) 2561–2576; (g) E. Anakabe, D. Badia, L. Carillo, E. Reyes, J.L. Vicario, in: O.A. Attanasi, D. Spinelli (Eds.), Targets in Heterocyclic Systems-Chemistry and Properties, vol. 6, Italian Society of Chemistry, Rome, Italy, 2002, pp. 270–311; (h) M. Zaidlewicz, W. Sokol, A. Wolan, J. Cytarska, A. Tafelska-Kaczmarek, A. Dzielendziak, A. Prewysz-Kwinto, Pure Appl. Chem. 75 (2003) 1349–1355; (i) S.M. Lait, D.A. Rankic, B.A. Keay, Chem. Rev. 107 (2007) 767–796.
- [2] (a) Z.-P. Liu (Guest Ed.), Curr. Org. Chem. 14 (2010) 888–999; (b) G.A. Olah, R.D. Chambers, G.K.S. Prakash (Eds.), Synthetic Fluorine Chemistry, J. Wiley & Sons, New York, 1992; (c) R.D. Chambers, Fluorine in Organic Chemistry, Blackwell, Oxford, 2004; (d) P. Kirsch, Modern Fluoroorganic Chemistry, Wiley-VCH, Weinheim, 2004; (e) J.-P. Bégué, D. Bonnet-Delpon, Bioorganic and Medicinal Chemistry of Fluorine, J. Wiley & Sons, Hoboken, NJ, 2008; (f) S.E. Lopez (Guest Ed.), Curr. Org. Synth., 7, in press.
- [3] (a) H. Kawai, A. Kusuda, S. Nakamura, M. Shiro, N. Shibata, Angew. Chem. Int. Ed. 48 (2009) 6324–6327; (b) J.-A. Ma, D. Cahard, Chem. Rev. 108 (2008) PR1–PR43; (c) N. Shibata, S. Mizuta, H. Kawai, Tetrahedron: Asymmetry 19 (2008) 2633–2644; (d) T. Billard, B.R. Langlois, Eur. J. Org. Chem. (2007) 891–897.
- [4] (a) R.A. Darrall, F. Smith, M. Stacey, J.C. Tatlow, J. Chem. Soc. (1951) 2329–2332; (b) T. Mill, J.O. Rodin, R.M. Silverstein, C. Woolf, J. Org. Chem. 29 (1964) 3715–3716; (c) Y.V. Zeifman, S.A. Postovoi, A.F. Aerov, E.I. Mysov, J. Fluorine Chem. 65 (1993) 189–194; (d) L. Sutin, S. Andersson, L. Bergquist, V.M. Castro, E. Danielsson, S. James, M. Hendriksson, L. Johansson, C. Kaiser, K. Flyren, M. Williams, Bioorg. Med. Chem. Lett. 17 (2007) 4837–4840; (e) I.-S. Chang, C.J. Wills, Can. J. Chem. 55 (1977) 2465–2472.
- [5] (a) J. Burdon, V.C.R. McLaughlin, J. Tatlow, J. Chem. Soc. (1960) 3184–3188; (b) S. Usui, S. Tsuboya, Y. Umezawa, K. Hazama, M. Okamura, Bull. Chem. Soc. Jpn. 82 (2009) 254–260.
- [6] T. Kitazume, J. Fluorine Chem. 35 (1987) 287–294.
- [7] (a) I. Choudhury-Mukherjee, H.A. Schenck, S. Cechova, T.N. Pajewski, J. Kapur, J. Ellena, D.S. Cafiso, M.L. Brown, J. Med. Chem. 46 (2003) 2494–2501; (b) I.S. Kruchok, I.I. Gerus, V.P. Kukhar, Tetrahedron 56 (2000) 6533–6539; (c) E.N. Shaitanova, I.I. Gerus, M.Y. Belik, V.P. Kukhar, Tetrahedron: Asymmetry 18 (2007) 192–198; (d) M. Clackers, D.M. Coe, D.A. Demaine, G.W. Hardy, D. Humphreys, G.G.A. Inglis, M.J. Jonston, H.D. Jones, D. House, R. Loiseau, D.J. Minick, P.A. Skone, I. Uings, I.M. Mc Lay, S.J.F. Macdonald, Bioorg. Med. Chem. Lett. 17 (2007) 4737–4745; (e) E.N. Shaitanova, I.I. Gerus, V.P. Kukhar, Tetrahedron Lett. 49 (2008) 1184–1187.
- [8] (a) H.B. Hass, E.F. Riley, Chem. Rev. 32 (1943) 373–430; (b) G. Rosini, Compr. Org. Synth. 2 (1991) 321–340.
- [9] D.J. Cook, O.R. Pierce, E.T. McBee, J. Am. Chem. Soc. 76 (1954) 83–87.
- [10] B. Imperiali, R.H. Abeles, Tetrahedron Lett. 27 (1986) 135–138.
- [11] (a) J.W. Skiles, V. Fuchs, C. Miao, R. Sorcek, K.G. Grotzinger, S.C. Mauldin, J. Vitous, P.W. Mui, S. Jacober, G. Chow, M. Matteo, M. Skoog, S.M. Weldon, G. Possanza, J. Keirns, G. Letts, A.R. Rosenthal, J. Med. Chem. 35 (1992) 641–662; (b) W. Ogilvie, M. Bailey, M.A. Poupard, A. Abraham, A. Bhavsar, P. Bonneau, J. Bordeleau, Y. Bousquet, C. Chabot, J.-S. Duceppe, G. Fazal, S. Goulet, Ch. Grand-Maitre, I. Guse, T. Halmos, P. Lavallee, M. Leach, E. Malenfant, J. O'Meara, R. Plante, C. Plouffe, M. Poirier, F. Soucy, Ch. Yoakim, R. Déziel, J. Med. Chem. 40 (1997) 4113–4135; (c) M.-A. Poupard, G. Fazal, S. Goulet, L.T. Mar, J. Org. Chem. 64 (1999) 1356–1361; (d) Y.-M. Shao, W.B. Yang, T.-H. Kuo, K.-C. Tsai, C.-H. Lin, A.-S. Yang, P.-H. Liang, C.-H. Wong, Bioorg. Med. Chem. 16 (2008) 4652–4660.
- [12] C.A. Veale, P.R. Bernstein, C.M. Bohnert, F.J. Brown, C. Bryant, J.R. Damewood Jr., R. Earley, S.W. Feeney, P.D. Edwards, B. Gomes, J.M. Hulsizer, B.J. Kosmider, R.D. Krell, G. Moore, T.W. Salcedo, A. Shaw, D.S. Silberstein, G.B. Steelman, M. Stein, A. Strimpler, R.M. Thomas, E.P. Vacek, J.C. Williams, D.J. Wolanin, S. Woolson, J. Med. Chem. 40 (1997) 3173–3181.
- [13] (a) E.T. McBee, C.E. Hathaway, C.W. Roberts, J. Am. Soc. 78 (1956) 4053–4057; (b) H. Schechter, D.E. Ley, E.B. Roberson Jr., J. Am. Chem. Soc. 78 (1956) 4984–4991.
- [14] K.-R. Gassen, W. Kirmse, Chem. Ber. 119 (1986) 2233–2248.
- [15] L. Poszawác, G. Simig, J. Heterocycl. Chem. 37 (2000) 343–348.
- [16] F. Tur, J.M. Saá, Org. Lett. 9 (2007) 5079–5082.
- [17] M. Tramontani, Synthesis (1982) 605–641.
- [18] G.L. Buchanan, Chem. Soc. Rev. 17 (1988) 91–109.
- [19] M. Kawase, M. Hirabayashi, H. Kumakura, S. Saito, K. Yamamoto, Chem. Pharm. Bull. 48 (2000) 114–119.
- [20] N.P. Peet, J.P. Burkhart, M.R. Angelastro, E.L. Giroux, S. Mehdi, P. Bey, M. Kolb, B. Neises, D. Schirlin, J. Med. Chem. 33 (1990) 394–407.
- [21] J.M. Andrés, R. Pedrosa, A. Pérez-Encabo, Eur. J. Org. Chem. (2004) 1558–1566.
- [22] W. Wan, Y. Gao, W. Jiang, J. Hao, J. Fluorine Chem. 129 (2008) 510–514.
- [23] J.P. Bégué, D. Bonnet-Delpon, H. Sdassi, Tetrahedron Lett. 33 (1992) 1882–1892.
- [24] A. Abouabdellah, J.-P. Bégué, D. Bonnet-Delpon, A. Kornilov, I. Rodrigues, C. Richard, J. Org. Chem. 63 (1998) 6529–6534.
- [25] J.-P. Bégué, D. Bonnet-Delpon, N. Fischer-Durand, Tetrahedron: Asymmetry 5 (1994) 1099–1110.
- [26] (a) D. Enders, M. Klatt, Synthesis (1996) 1403–1418; (b) C. Palomo, A. Mielgo, Angew. Chem. Int. Ed. 45 (2006) 7876–7880.
- [27] K. Funabiki, A. Shibata, H. Iwata, K. Hatano, Y. Kubota, K. Komura, M. Ebihara, M. Matusi, J. Org. Chem. 73 (2008) 4694–4697.
- [28] J. Wiedemann, T. Heiner, G. Mostoń, G.K.S. Prakash, G.A. Olah, Angew. Chem. Int. Ed. 37 (1998) 820–821.
- [29] (a) M. Kawase, H. Miyamae, M. Narita, T. Kurihara, Tetrahedron Lett. 34 (1993) 859–862; (b) M. Kawase, H. Miyamae, T. Kurihara, Chem. Pharm. Bull. 46 (1998) 749–756.
- [30] T. Tomoyasu, K. Tomooka, T. Nakai, Synlett (1998) 1147–1149.
- [31] (a) A.K. Yudin (Ed.), Aziridines and Epoxides in Organic Synthesis, Wiley-VCH, Weinheim, 2006; (b) G. Dake, Oxiranes and Oxirenes, in: A.R. Katritzky, C.A. Ramsden, E.F.V. Scriven, R.J.K. Taylor (Eds.), Comprehensive Heterocyclic Chemistry III, vol. 1, Elsevier, 2008, pp. 174–217 (Chapter 1.03).
- [32] E.T. McBee, C.E. Hathaway, C.W. Roberts, J. Am. Chem. Soc. 78 (1956) 3851–3854.
- [33] E.J. Reinhard, J.L. Wang, R.C. Durley, Y.M. Fobian, M.L. Grapperhaus, B.S. Hickory, M.A. Massa, M.B. Norton, M.A. Promo, M.B. Tollefson, W.F. Vernier, D.T. Connolly, B.J. Witherbee, M.A. Melton, K.J. Regina, M.E. Smith, J.A. Sikorski, J. Med. Chem. 46 (2003) 2152–2168.
- [34] M.G. Voronkov, V.P. Baryshok, S.N. Tandura, V.Y. Vitkovskii, V.M. D'yakov, V.A. Pestunovich, Zh. Obsch. Khim. 48 (1978) 2238–2245 (J. Gen. Chem. USSR, 48, 1978, 2033).
- [35] V.A. Petrov, Synthesis (2002) 2225–2231.
- [36] E. Kirillov, L. Lavanant, Ch. Thomas, T. Roisnel, Y. Chi, J.-F. Carpentier, Chem. Eur. J. 13 (2007) 923–935.
- [37] Y. Yamauchi, T. Kawate, T. Katagiri, K. Uneyama, Tetrahedron 59 (2003) 9839–9847.
- [38] A. Harada, Y. Fujiwara, T. Katagiri, Tetrahedron: Asymmetry 19 (2008) 1210–1214.

- [39] Y. Yamauchi, T. Katagiri, K. Uneyama, *Org. Lett.* 4 (2002) 173–176.
- [40] P.V. Ramachandran, B. Gong, H.C. Brown, *J. Org. Chem.* 60 (1995) 41–46.
- [41] T. Katagiri, N. Iguchi, T. Kawate, S. Takahashi, K. Uneyama, *Tetrahedron: Asymmetry* 17 (2006) 1157–1160.
- [42] P. Bravo, A. Farina, M. Frigerio, S.V. Meille, F. Viani, *Tetrahedron: Asymmetry* 5 (1994) 987–1004.
- [43] A. Arnone, P. Bravo, M. Frigerio, F. Viani, V.A. Soloshonok, *Tetrahedron* 54 (1998) 11841–11860.
- [44] A.E. Sorochinsky, V.A. Soloshonok, *J. Fluorine Chem.* 131 (2010) 127–139.
- [45] (a) G.K.S. Prakash, A.K. Yudin, *Chem. Rev.* 97 (1997) 757–786;
(b) G.K.S. Prakash, M. Mandal, *J. Fluorine Chem.* 112 (2001) 123–131;
(c) G.A. Olah, G.K.S. Prakash, Q. Wang, X.-Y. Li, in: L.A. Paquette (Ed.), *Encyclopedia of Reagents in Organic Synthesis*, vol. 7, J. Wiley & Sons, New York, 1995, pp. 5165–5166;
(d) R.S. Bastos, *Synlett* (2008) 1425–1426.
- [46] (a) G.K.S. Prakash, C. Panja, H. Yaghoo, V. Surampudi, R. Kultyshev, M. Mandal, G. Rasul, T. Mathew, G.A. Olah, *J. Org. Chem.* 71 (2006) 112–114;
(b) K. Iwanami, T. Oriyama, *Synlett* (2006) 112–114.
- [47] (a) M.W. Walter, A. Felici, M. Galleni, R.P. Soto, R.M. Adlington, J.E. Baldwin, J.-M. Frère, M. Gololobov, C.J. Schofield, *Bioorg. Med. Chem. Lett.* 6 (1996) 2455–2458;
(b) M.W. Walter, R.M. Adlington, J.E. Baldwin, C.J. Schofield, *Tetrahedron* 53 (1997) 7275–7290.
- [48] G.S. Garrett, S.J. McPhail, K. Tornheim, P.E. Correa, J.M. McIver, *Bioorg. Med. Chem. Lett.* 9 (1999) 301–306.
- [49] X.-H. Xu, X.-L. Qiu, F.-L. Qing, *Tetrahedron* 64 (2008) 7353–7361.
- [50] A.K. Podichetty, S. Wagner, A. Faust, M. Schäfers, O. Schober, K. Kopka, G. Haufe, *Future Med. Chem.* 1 (2009) 969–989.
- [51] K. Kobayashi, T. Narumi, S. Oishi, H. Ohno, N. Fujii, *J. Org. Chem.* 74 (2009) 4626–4629.
- [52] G. Mlostoń, K. Gębicki, B. Kołodziej, XXII Conference on Advances in Organic Synthesis, Karpacz, Poland, 2009, Book of Abstracts, P07.
- [53] E. Obijalska, G. Mlostoń, A. Linden, H. Heimgartner, *Tetrahedron: Asymmetry* 19 (2008) 1676–1683.
- [54] G. Mlostoń, E. Obijalska, A. Tafelska-Kaczmarek, M. Zaidlewicz, *J. Fluorine Chem.*, submitted for publication.
- [55] S. Fustero, L. Albert, J.L. Aceña, J.F. Sanz-Cervera, A. Asensio, *Org. Lett.* 10 (2008) 605–608.
- [56] (a) X.-C. He, E.L. Eliel, *Tetrahedron* 43 (1987) 4979–4987;
(b) E.L. Eliel, X.-C. He, *J. Org. Chem.* 55 (1990) 2114–2119.
- [57] R. Pedrosa, S. Sayalero, M. Vicente, A. Maestro, *J. Org. Chem.* 71 (2006) 2177–2180.
- [58] (a) N.A. Petasis, I. Akritopoulou, *Tetrahedron Lett.* 34 (1993) 583–586;
(b) N.A. Petasis, I.A. Zavialov, *J. Am. Chem. Soc.* 119 (1997) 445–446;
(c) N.A. Petasis, I.A. Zavialov, *J. Am. Chem.* 120 (1998) 11798–11799.
- [59] G.K.S. Prakash, M. Mandal, S. Schweizer, N.A. Petasis, G.A. Olah, *Org. Lett.* 2 (2000) 3173–3176.
- [60] D.V. Patel, K. Rielly-Gauvin, D.E. Ryono, *Tetrahedron Lett.* 29 (1988) 4665–4668.
- [61] Z.-X. Jiang, Y.-Y. Qin, F.L. Qing, *J. Org. Chem.* 68 (2003) 7544–7547.
- [62] T. Taguchi, Y. Suda, M. Hamochi, Y. Fujino, Y. Itaka, *Chem. Lett.* (1991) 1425–1428.
- [63] C. Mioskowski, G. Solladié, *Tetrahedron* 29 (1973) 3669–3674 (cf. data given in [16]).
- [64] K.P.M. Vanhessche, K.B. Sharpless, *Chem. Eur. J.* 4 (1997) 517–522.
- [65] (a) V.A. Soloshonok, D.V. Avilov, V.P. Kukhar, *Tetrahedron: Asymmetry* 7 (1996) 1547–1550;
(b) V.A. Soloshonok, V.P. Kukhar, S.V. Galushko, N.Y. Svistunova, D.V. Avilov, N.A. Kuz'mina, N.I. Raevski, Y.T. Struchkov, A.P. Pysarevski, Y.N. Belkova, *J. Chem. Soc., Perkin Trans. 1* (1993) 3143–3155.
- [66] E. Obijalska, G. Mlostoń, H. Heimgartner, *Helv. Chim. Acta*, submitted for publication.
- [67] P. Maienfisch, R.G. Hall, *Chimia* 58 (2004) 93–99.
- [68] R. Betageri, Y. Zhang, R.M. Zindell, D. Kuzmich, T.M. Kirrane, J. Bentzien, M. Cardozo, A.J. Capolino, T.N. Fadra, R.M. Nelson, Z. Paw, D.-T. Shih, Ch.-K. Shih, L. Zuvella-Jelaska, G. Nabozny, D.S. Thomson, *Bioorg. Med. Chem. Lett.* 15 (2005) 4761–4769.
- [69] T. Katagiri, K. Uneyama, *Functional Group Transformations at α -Carbon to Trifluoromethyl Group, in Fluorine Containing Synthons*, in: V.A. Soloshonok (Ed.), *ACS Symposium Series*, 2005, vol. 911, Chapter 18, pp. 318–341.
- [70] C.A. Veale, J.R. Damewood Jr., G.B. Steelman, C. Bryant, B. Gomes, J. Williams, *J. Med. Chem.* 38 (1995) 86–97.
- [71] T. Katagiri, Y. Fujiwara, S. Takahashi, N. Ozaki, K. Uneyama, *Chem. Commun.* (2002) 986–987.
- [72] P. Bravo, M. Crucianelli, T. Ono, M. Zanda, *J. Fluorine Chem.* 97 (1999) 27–49.