



Article Anti-Inflammatory Dysidazirine Carboxylic Acid from the Marine Cyanobacterium *Caldora* sp. Collected from the Reefs of Fort Lauderdale, Florida[†]

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- + Dedicated to William Fenical, University of California San Diego, for his pioneering work on bioactive natural products.

Abstract: Dysidazirine carboxylic acid (1) was isolated from the lipophilic extract of a collection of the benthic marine cyanobacterium Caldora sp. from reefs near Fort Lauderdale, Florida. The planar structure of this new compound was determined by spectroscopic methods and comparisons between HRMS and NMR data with its reported methyl ester. The absolute configuration of the single chiral center was determined by the conversion of 1 to the methyl ester and the comparison of its specific rotation data with the two known methyl ester isomers, 2 and 3. Molecular sequencing with 16S rDNA indicated that this cyanobacterium differs from Caldora penicillata (Oscillatoriales) and represents a previously undocumented and novel Caldora species. Dysidazirine (2) showed weak cytotoxicity against HCT116 colorectal cancer cells (IC₅₀ 9.1 μ M), while dysidazirine carboxylic acid (1) was non-cytotoxic. Similar cell viability patterns were observed in RAW264.7 cells with dysidazirine only (2), displaying cytotoxicity at the highest concentration tested (50 μ M). The noncytotoxic dysidazirine carboxylic acid (1) demonstrated anti-inflammatory activity in RAW264.7 cells stimulated with LPS. After 24 h, 1 inhibited the production of NO by almost 50% at 50 μ M, without inducing cytotoxicity. Compound 1 rapidly decreased gene expression of the pro-inflammatory gene *iNOS* after 3 h post-LPS treatment and in a dose-dependent manner (IC₅₀ ~1 μ M); the downregulation of iNOS persisted at least until 12 h.

Keywords: marine cyanobacteria; *Caldora*; Oscillatoriales; marine natural products; azirine natural product; cytotoxicity; anti-inflammatory activity; iNOs

1. Introduction

Azirine natural products are rare in nature. Azirinomycin is the first example of a natural product containing an unstable strained azirine ring. It was discovered in 1971 from a cultured broth of a strain of *Streptomyces aureus* from a soil sample [1]. Azirinomycin has a simple structure of 3-methyl-2*H*-azirine-2-carboxylic acid. Azirinomycin and its semi-synthetic methyl ester exhibit broad spectrum in vitro antibiotic activity against both Gram-positive and Gram-negative bacteria [2]. Molinski and Ireland reported the isolation of cytotoxic dysidazirine (2), an azacyclopropene methyl ester, from the marine sponge *Dysidea fragilis* collected in Fiji, and this was the first example of a strained 2H-azirine from a marine source [3]. In 1995, Faulkner's group reported the isolation of four azacyclopropane methyl esters (**3–6**) related to dysidazirine (**2**) from the same marine sponge *Dysidea fragilis*



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). collected in Pohnpei, the Federated States of Micronesia [4] (Figure 1). Compound (2), (4*E*)-*R*-dysidazirine, was previously shown to be active against *Pseudomonas aeruginosa, Candida albicans*, and *Saccharomyces cerevisiae* [3,5,6]. It was also shown to possess activity against liquid tumor cells, with an IC₅₀ of 0.88 μ M against the L1210 (mouse lymphocytic leukemia) cell line. Activity against cancer cell lines derived from solid tumors has not been reported for this compound; however, halogenated analogues have shown activity against the human colon cancer cell line HCT116 [3,5,6]. Dysidazirine (2) has received attention from chemists and inspired the synthesis of new antibacterial and antifungal derivatives [7,8]. Structure activity studies of synthetic dysidazirine analogues indicated that antifungal activity was dependent on the C4-C5 unsaturation but was not strongly dependent on the configuration at C2 or minor changes in chain lengths. Branching of the terminal end abolished antifungal activity [7].



Figure 1. Chemical structures of azirine natural products.

Here, we report the first isolation of 2H-azirine free carboxylic acid (1), a possible biosynthetic intermediate of dysidazirine (2), from a cyanobacterium *Caldora* sp. collected from a reef near Fort Lauderdale, Florida. The existence of biosynthetic pathways to produce rare azirines in the terrestrial bacterium *Streptomyces aureus*, the marine sponge *Dysidea fragilis*, and marine cyanobacterium *Caldora* sp. suggest a possible microbial origin for the compounds in the sponge.

2. Results and Discussion

2.1. Isolation and Structure Elucidation

The sample of the marine cyanobacterium *Caldora* sp. was collected in shallow water (7–10 m depth) while scuba diving near Fort Lauderdale, Florida, in July 2017, and was repeatedly present at this location near a coral reef monitoring site during the summer months in 2017–2018, resulting in several recollections. The samples were transported live in seawater in a cooler to the Smithsonian Marine Station at Fort Pierce, Florida, where they were drained, frozen, and freeze-dried. Freeze-dried material was extracted with

a mixture of EtOAc-MeOH (1:1) to afford a lipophilic extract, which was subsequently partitioned between EtOAc and H_2O . The EtOAc-soluble portion was fractionated by SiO_2 column chromatography, followed by reversed-phase C18 column chromatography, and then further purified by reversed-phase HPLC to give the new compound, dysidazirine carboxylic acid (1) (Figure 1).

Dysidazirine carboxylic acid (1) was obtained as a colorless solid. The molecular formula $C_{18}H_{31}NO_2$ was determined from HRESIMS data. The appearance of a strong negative ion at m/z 292.2276 for [M-H]⁻ and the presence of a single carbonyl signal at 176.2 ppm in the ¹³C spectrum suggested the compound to be an acid. To confirm the presence of a free carboxylic acid, 1 was reacted with trimethylsilyldiazomethane, yielding the methyl ester (2). HRESIMS data gave the expected molecular formula of $C_{19}H_{33}NO_2$ for the methyl ester (2).

Following the interpretation of DQF ¹H-¹H COSY, edited HSQC and HMBC experiments (Table 1) (Supplementary Materials S2–S6), the ¹H and ¹³C NMR signals of **1** were assignable to one terminal methyl group (C-18, $\delta_{\rm H}$ 0.89, $\delta_{\rm C}$ 14.4), twelve consecutive methylene groups, three conjugated olefinic carbons (C-3 $\delta_{\rm C}$ 157.1; C-4 $\delta_{\rm H}$ 6.61, $\delta_{\rm C}$ 114.1; C-5 $\delta_{\rm H}$ 6.75, $\delta_{\rm C}$ 158.4), one methine carbon as a singlet (C-2, $\delta_{\rm H}$ 2.49, $\delta_{\rm C}$ 29.4), and one carbonyl carbon C-1 ($\delta_{\rm C}$ 176.2). The COSY spectrum connected the alkyl chain to the double bond via single allylic methylene group (C-6, $\delta_{\rm H}$ 2.38, $\delta_{\rm C}$ 34.1). The two coupled proton signals C-4 at 6.61 ppm (d, *J* = 15.1 Hz) and C-5 at 6.75 ppm (dd, *J* = 15.1, 6.9 Hz) were assignable to a *trans*-disubstituted olefin. HMBC correlations indicated H-2 ($\delta_{\rm H}$ 2.49) to C-1 carbonyl group (δ_{C} 176.2), C-3 (δ_{C} 157.1) and C-4 (δ_{C} 114.1), and this information connected the acid terminal end to the alkyl chain via the double bond. Dereplication using the molecular formula of the methyl ester 2 in the marine natural product database, together with the 1 H and ¹³C NMR spectral data analysis, identified the prepared methyl ester **2** of the isolate as one of the three previously reported dysidazirines (2–4) [3,4]. This information, together with the ¹H-¹H COSY, HMBC, and HSQC NMR data analysis, particularly the coupling constant of J = 15.1 Hz for 4-H and 5-H of the free acid (1) and the methyl ester (2), eliminated the (Z) geometric isomer (4). This limited the two enantiomers to (4E)-R-dysidazirine (2) and (4E)-S-dysidazirine (3). Comparison of the observed negative optical rotation data of the prepared methyl ester $[\alpha]^{25}D - 166.7$ (c 0.06, CDCl₃) with the reported optical data of the two enantiomers identified the prepared methyl ester as (4E)-R-dysidazirine (2); literature value $[\alpha]D - 165$ (c 0.5, MeOH), first reported as dysidazirine [3]. In contrast, the reported optical rotation for (4*E*)-*S*-dysidazirine (3) was positive ($[\alpha]D + 47.2$ (*c* 1.08, CDCl₃)) [4]. With only one stereocenter, this confirmed the structure of the isolated precursor acid as (4*E*)-*R*-dysidazirine carboxylic acid (1).

Table 1.	NMR	spectroscopic	data	for	(4 <i>E</i>)- <i>R</i> -dysidazirine	carboxylic	acid	(1)	and	(4E)-R-
dysidazirir	ne (2).									

C No	δ_{C} Mult.	$\delta_{ m H}$ (J in Hz)	COSY	HMBC ^{<i>a</i>}	δ_{C} Mult.	$\delta_{ m H}$ (J in Hz)
		1 ^b	2 ^{<i>c</i>,<i>d</i>}			
1	176.2, C			2	172.2, C	
2	29.4, CH	2.49, s		4	28.3, CH	2.45, s
3	157.1, C			2, 4, 5	156.6, C	
4	114.1, CH	6.61, d (15.1)	5,6	2,5	112.9, CH	6.47, d (15.8)
5	158.4, CH	6.75, dt (15.1, 6.9)	4,6	6,7	155.8, CH	6.64, dt (15.8, 6.8)
6	34.1, CH ₂	2.38, dt (6.9, 7.4)	5,7	4, 5, 7	33.2, CH ₂	2.28, dt (6.8, 7.5)
7	29.0, CH ₂	1.52, m	6, 8	5,6	29.1, CH ₂	1.44, m
8-15	30.3–30.8, CH ₂	1.35–1.24, m			29.2–29.7, CH ₂	1.1–1.3, m
16	33.0, CH ₂	1.32, m		18	33.2, CH ₂	1.17, m
17	23.7, CH ₂	1.24, m	18	18	22.6, CH ₂	1.17, m
18	14.4, CH ₃	0.89, t (6.8)	17	16, 17	14.1, CH ₃	0.79, t (6.8)
19					52.2, CH ₃	3.34, s

^{*a*} HMBC correlations, optimized for $^{2/3}J_{CH} = 8$ Hz, are from proton(s) stated to the indicated carbon. ^{*b*} In CD₃OD.

^c Proton data in CD₃OD with a few drops of CDCl₃ to improve solubility. ^d Carbon data in CDCl₃.

2.2. Identification of the Cyanobacterium

DNA extraction, sequencing, and phylogenetic analyses were conducted, as previously described [9]. Phylogenetic reconstruction based on 16S rDNA identified our *Caldora* sample, FTL6 collected in Fort Lauderdale on 16 October 2017, as an undocumented and novel sequence for this genus (Figure 2). Our tree showed three sequences that branch prior to the clade delimited as *Caldora penicillata* by Engene et al. [10], which receives overall low bootstrap support (56%), possibly due to sequences swapping position within this clade during tree reconstruction (see Supplementary Materials Figure S7 for the poor topology within this clade caused by the overall variability of 16S rDNA sequences). However, the *Caldora* genus was strongly supported, with FTL6 consistently branching first (99% node bootstrap support) (Figure 2).



Figure 2. 16S rDNA maximum-likelihood tree showing the early branching phylogenetic position of the sample FTL6 within *Caldora* spp. The numerous sequences found in the *C. penicillata* clade, as delimited by Engene et al. [10], are summarized as a triangle for figure clarity (see Supplementary Materials Figure S7 for additional details).

Considering its unique and distinctive morphology (Figure 3), the novel chemistry of *Caldora* sp. FTL6, and its consistent phylogenetic branching prior to *C. penicillata*, we consider this specimen as a novel species within the genus *Caldora*. For further confirmation, future endeavors should reassess genus diversity with more variable markers than 16S rDNA (e.g., such as the protein encoding gene *tufA*) to improve topological features of the tree (Figure 2).



Figure 3. Cont.



Figure 3. Underwater photographs of *Caldora* sp. in comparison to *Caldora penicillata* on the reef near Fort Lauderdale, FL. (**A**,**B**) Macroscopic photographs of tufts of *Caldora* sp. Note the distinct morphology of *Caldora* sp., as the filaments are intertwined giving it a "stringy" appearance. (**C**) *Caldora penicillata* [10]. (**D**) The two species can also grow together, and the white box shows *Caldora* sp. as it overgrows *Caldora penicillata*.

2.3. Biological Activity

Carboxylic acid 1 and its methyl ester 2 were tested against human colon cancer HCT116 cells to probe for antiproliferative activity. The cell viability assay indicated modest antiproliferative activity of 2 (IC₅₀ 9.1 μ M) and 8.8-fold reduced activity of 1 (79.7 μ M) (Figure 4A). Compounds 1 and 2 were then tested for potential anti-inflammatory activity by measuring nitric oxide (NO) production in RAW264.7 cells stimulated by LPS. Cells were pretreated with compounds 1 and 2 separately for 1 h and then challenged with LPS. After 24 h, acid 1 inhibited the production of NO by almost 50% at 50 μ M, without inducing cytotoxicity at that concentration, while methyl ester 2 was cytotoxic at that concentration, leading to apparent NO inhibition (Figure 4B,C). The increased cytotoxicity of 2 versus 1 was consistent with our data in HCT116 cells. Therefore, we focused on compound 1 for further mechanistic studies related to NO inhibition and anti-inflammatory activity at non-cytotoxic concentrations. To determine if NO inhibition might be a result of the transcriptional regulation of inducible nitric oxide synthase (iNOS), the NF- κ B target gene encoding the synthesis of NO, we measured the transcript levels of this pro-inflammatory gene. Compound 1 rapidly decreased gene expression after 3 h post-LPS treatment and in a dose-dependent manner; the downregulation of iNOS persisted at least until 12 h (Figure 4D). Compound 1 at 3 h and 12 h showed a similar response, with the target gene showing a 50% reduction near 1 μ M and roughly 75% reduction at 10 and 50 μ M. Therefore, early transcriptional changes were even more sensitive to compound **1** than functionally-linked NO levels after 24 h.



Figure 4. Bioactivity of (4*E*)-*R*-dysidazirine carboxylic acid (**1**) and methyl ester **2**. (**A**) Antiproliferative activity against human colon cancer cells (HCT116) measured by MTT assay at 48 h. Gatorbulin-1 was used as a positive control, tested at the same time (IC₅₀ 0.80 μ M) [**11**]. (**B**) Anti-inflammatory activity of pretreatment for 1 h with **1**, **2** (50, 10, 1 and 0.1 μ M), or vehicle control (0.5% DMSO) by measuring the production of nitric oxide (NO) in murine macrophages (RAW264.7) 24 h after LPS stimulation. (**C**) Cell viability of RAW264.7 cells using the MTT assay at 24 h under the same conditions used in the NO assay. (*D*) *iNOS* target gene expression in RAW264.7 after pretreatment for 1 h with **1** or vehicle control (0.5% DMSO prior to LPS addition for 3 and 12 h). RNA was isolated, reverse-transcribed, and subjected to qPCR, using β -actin as the endogenous control. The values were normalized to vehicle control treated with LPS for each time point. Non-stimulated cells (no LPS) were tested simultaneously (**B**–**D**). Error bars indicate the mean \pm SD of three replicates for graphs B, C, and D. Statistical analysis was performed using multiple comparison t-tests (* *p* < 0.05, ** *p* < 0.001, *** *p* < 0.001, *** *p* < 0.001, compared to LPS treatment alone).

3. Materials and Methods

3.1. General Experimental Procedures

Optical rotations were recorded on a Rudolph Research Analytical Autopol III automatic polarimeter. UV spectrophotometric data were acquired on a Shimadzu PharmaSpec UV–visible spectrophotometer. NMR data were collected on a JEOL ECA-600 spectrometer operating at 600.17 MHz for ¹H and 150.9 MHz for ¹³C. The edited-HSQC experiment was optimized for $J_{CH} = 140$ Hz and the HMBC spectrum was optimized for ${}^{2/3}J_{CH} = 8$ Hz. ¹H NMR chemical shifts (referenced to residual CHCl₃ observed at δ 7.25 and residual CH₃OH observed at δ 3.30) were assigned using a combination of data from 2D DQF COSY and multiplicity-edited HSQC experiments. Similarly, ¹³C NMR chemical shifts (referenced to residual CHCl₃ observed at δ 77.0 and residual CH₃OH observed at δ 49.0) were assigned based on the multiplicity-edited HSQC experiments. The HRMS data were obtained using an Agilent 6210 LC-TOF mass spectrometer equipped with an APCI/ESI multimode ion source detector at the Mass Spectrometer Facility at the University of California, Riverside, California. Silica gel 60 (EMD Chemicals, Inc., Port Wentworth, GA, USA, 230–400 mesh) was used for column chromatography. All solvents used were of HPLC grade (Fisher Scientific, Waltham, MA, USA).

3.2. Collection, Extraction and Isolation

The first sample of *Caldora* sp. for this study was collected in July 2017 from Fort Lauderdale, Florida. The sample was transported in seawater in a cooler to the Smithsonian Marine Station at Ft. Pierce, Florida, where it was drained, frozen in a plastic bag, and then freeze-dried. The freeze-dried material (48.5 g) was repeatedly extracted in a glass beaker with EtOAc-MeOH (1:1) to give 4.93 g of the lipophilic extract. This extract was partitioned between EtOAc and H₂O. The EtOAc-soluble fraction (1.04 g) was chromatographed on a column of SiO₂ (25 g) using a step gradient system of hexanes-EtOAc, EtOAc, EtOAc-MeOH, and EtOAc-MeOH to give four sub-fractions. The second sub-fraction that was eluted with EtOAc (0.080 g) was further chromatographed on a column of C₁₈ (4 g) using a step gradient system of MeOH-H₂O, MeOH-H₂O, and MeOH to give 3 sub-fractions. The second sub-fraction (0.021 g) was further purified by reversed-phase HPLC (semi-prep 250 × 10 mm, 5 μ m, RP-18, flow 3.0 mL/min) using MeOH-12% H₂O to give dysidazirine carboxylic acid (1) (3.0 mg, $t_{\rm R}$ = 19.8 min, yield, 0.006% dry wt).

Dysidazirine carboxylic acid (1): colorless, solid; $[\alpha]^{25}$ D 145.2 (*c* 0.12, CDCl₃); UV(MeOH) λ_{max} (log ε) 222 (4.11) nm; ¹H and ¹³C NMR data (see Table 1) assignments were made by the interpretation of 2D DQF COSY, edited-HSQC, and HMBC data (Supplementary Materials S2–S6); HRESI/TOFMS *m*/*z* 292.2276 [M-H]⁻ (calcd for C₁₈H₃₀NO₂, 292.2278).

For the methylation of dysidazirine carboxylic acid (1), compound 1 (1.0 mg) was dissolved in 1.0 mL of MeOH and cooled to 4 °C. The cooled solution was treated with 50 μ L of 2.0 M trimethylsilyl diazomethane in diethyl ether. Evaporation of MeOH and excess trimethylsilyl diazomethane gave an oily product. This product was purified by reversed-phase HPLC (semi-prep 250 × 10 mm, 5 μ m, RP-18, flow 3.0 mL/min) using MeOH-5% H₂O to give methyl ester (4*E*)-*R*-dysidazirine **2** (0.9 mg, *t*_R = 14.4 min).

(4*E*)-*R*-dysidazirine (**2**): colorless, solid; $[\alpha]^{25}$ D -166.7 (*c* 0.06, CDCl₃), [lit. $[\alpha]$ D -165 (*c* 0.5, MeOH)[4]; UV (MeOH) λ_{max} (log ε) 222 (4.11) nm; ¹H and ¹³C NMR data, see Table 1; HRESI/TOFMS *m*/*z* 308.2613 [M-H]⁻ (calcd for C₁₉H₃₄NO₂, 308.2590).

3.3. Molecular Identification

One of multiple recollections over a 2-year period (2017–2018), the sample used for sequencing of the 16S rRNA gene (FTL6) was collected in Fort Lauderdale on 16 October 2017 in the same coral monitoring site as the original collection (see above). A voucher specimen is maintained at the Smithsonian Marine Station, Fort Pierce, FL. DNA extraction, sequencing, and phylogenetics were conducted as previously described [9].

3.4. Cancer Cell Viability Assay (HCT116)

Human colon cancer cells HCT116 were used as a representative model system to probe for antiproliferative activity. HCT116 cells were cultured in Dulbecco's modified Eagle's medium (DMEM), supplemented with 10% fetal bovine serum at 37 °C humidified air and 5% CO₂. Cells were seeded (8000 cells/well) in 96-well plates, allowed to attach overnight and treated with (4*E*)-*R*-dysidazirine carboxylic acid (1), methyl ester (2), and the solvent control (0.5% DMSO). Cell viability was measured after 48 h, following treatment with MTT dye using the manufacturer's protocol (Promega, Madison, WI, USA). IC₅₀ values were calculated from variable slope fitting for a dose response curve using GraphPad Prism software. Gatorbulin-1 was tested at the same time (IC₅₀ 0.80 μ M) [11] and served as a positive control.

3.5. NO Assay and Cell Viability Assay (RAW264.7 Cells)

RAW264.7 cells were cultured in Dulbecco's modified Eagle's medium (DMEM), supplemented with 10% fetal bovine serum at 37 °C humidified air and 5% CO₂. Cells were seeded (2×10^4 /well) in 96-well plates and allowed to attach for 24 h before being treated with the compounds (**1** and **2**) or the solvent control (0.5% DMSO) for 1 h, followed by the addition of LPS at 1 µg/mL. Non-stimulated cells (no LPS) were tested simultaneously. The production of NO in the cell supernatant was measured after 24 h by measuring the nitrite concentration, which is an oxidative product of NO. A total of 50 µL of the supernatant was mixed with the Griess reagent using the manufacturer's protocol (Promega) and the absorbance was measured at 540 nm. The nitrite concentration was derived from a calibration curve generated from a fresh nitrite standard solution. Cell viability was measured under the same seeding conditions and time points, using MTT dye following the manufacturer's protocol (Promega).

3.6. iNOS mRNA Measurement in RAW264.7 Cells

RAW264.7 cells were seeded $(2.5 \times 10^5$ /well) in 6-well plates and treated with different concentrations of 1 or vehicle control for 1 h prior to treatment with LPS at 1 µg/mL and then incubated for 3 and 12 h. Non-stimulated cells (no LPS) were tested simultaneously. Total RNA was extracted using the RNAeasy Mini Kit (QIAGEN, Germantown, MD, USA). RNA was reverse-transcribed to cDNA using SuperScript II Reverse Transcriptase (Invitrogen, Carlsbad, CA, USA) and oligo (dT) (Invitrogen), starting with 500 ng of total RNA. The cDNA samples were used as templates for TaqMan gene expression assays (Applied Biosystems, Waltham, MA, USA). The qPCR analysis was performed in triplicate at 25 µL total volume (12.5 µL of TaqMan 2× universal master mix, 1.25 µL of a 20× TaqMan gene expression assay probe, 1 µL of cDNA, and 10.25 µL of RNase-free sterile water) to detect the expression of *iNOS* (Mm00440502_m1, Applied Biosystems) and β -actin (internal standard, 4352663, Applied Biosystem). The qPCR method was as follows: 50 °C for 2 min, 95 °C for 10 min, and 40 cycles of 95 °C for 15 s and 60 °C for 1 min.

Supplementary Materials: The following are available online: ¹H, ¹³C, and 2D ¹H-¹H COSY, HSQC and HMBC NMR spectra in CD₃OD for dysidazirine carboxylic acid (**1**), and ¹H and ¹³C NMR spectra for (4*E*)-*R*-dysidazirine (**2**) in CDCl₃-CD₃OD and CDCl₃, respectively. Fully developed 16S rDNA maximum likelihood tree showing *Caldora penicillata* sequences summarized as a triangle in Figure 2.

Author Contributions: H.L., V.J.P.: project conception. H.L., V.J.P., T.S.: experimental design. S.P.G., S.K., R.R., L.A.H.d.S.: performing experiments. T.S., L.A.H.d.S.: molecular sequencing and phylogenetics. S.P.G., S.K., R.R., T.S., H.L., V.J.P.: data analysis. S.P.G., S.K., R.R., H.L., T.S., V.J.P.: draft writing. All authors have read and agreed to the published version of the manuscript.

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Sample Availability: Samples of the compounds are available from the corresponding author.

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