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ORIGINAL ARTICLE

A morphologically transformable hypoxia-induced radical anion for tumor-specific photothermal therapy



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Amino acid

Abstract Tumor microenvironment activatable therapeutic agents and their effective tumor accumulation are significant for selective tumor treatment. Herein, we provide an unadulterated nanomaterial combining the above advantages. We synthesize a perylene diimide (PDI) molecule substituted by glutamic acid (Glu), which can self-assemble into small spherical nanoparticles (PDI-SG) in aqueous solution. PDI-SG can not only be transformed into nanofibers at low pH conditions but also be reduced to PDI radical anion (PDI^{•−}), which exhibits strong near-infrared absorption and excellent photothermal performance. More importantly, PDI-SG can also be reduced to PDI^{•−} in hypoxic tumors to ablate the tumors and minimize the damage to normal tissues. The morphological transformation from small nanoparticles to nanofibers makes for better tumor accumulation and retention. This work sheds light on the design of tumor microenvironment activatable therapeutics with precise structures for high-performance tumor therapy.

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1. Introduction

The research for safe and effective strategies to treat tumors remains a challenge. Hypoxia, reducibility, and low pH are important features of the tumor microenvironment (TME)^{1–7}. The complex TME may confine the therapeutic outcome of some tumor therapies but can also be employed to design tumor-specific drug delivery systems^{8–13}. In addition, both the accumulation and retention ability of nanomaterials in tumor tissues play crucial roles in the therapeutic outcome and are closely connected with their dimensions and morphologies^{14,15}. Small nanoparticles are conducive to rapid accumulation at the tumor sites but easily escape from the tumor tissues. In contrast, nanofibers can prolong the retention time due to the restriction on diffusion in tumor tissues¹⁶. Therefore, the development of responsive nanostructures with transformable morphology is a promising strategy^{17–21}.

Photothermal therapy (PTT) as an emerging treatment for tumor ablation has many prominent advantages, such as low invasiveness, high specificity, and simple operation^{22–27}. An ideal photothermal agent (PTA) should switch on the photothermal activity only in tumor tissues to eliminate the damage to normal tissues^{28,29}. Relative to the most frequently used PTAs, radicals have been gradually developed for photothermal conversions^{30–36}. Perylene diimides (PDIs), as a class of important organic fluorescent dyes, possess many beneficial features, such as high fluorescence intensity, excellent thermal stability, and easy functionalization^{37–40}. It has been reported that PDIs can be reduced by some reductants such as $\text{Na}_2\text{S}_2\text{O}_4$ ^{41,42} and KO_2 ⁴³ to generate PDI radical anions. The near-infrared (NIR) absorption and the excellent photothermal conversion performance of PDI radical anions have attracted extensive attention from researchers⁴². However, most reported PDI radical anions were utilized to inhibit the growth of *Escherichia coli* because the facultative anaerobic bacteria can provide a highly reductive environment^{44–47}. Recently, the application scope of PDI

radical anions as PTAs has been extended to ablation of solid tumors by Wang et al.⁴⁸. They have demonstrated that PDI supramolecules could also be reduced to PDI radical anions in the tumors.

In this work, a glutamic acid (Glu) functionalized PDI (PDI-Glu) was synthesized. PDI-SG can be obtained by self-assembly of PDI-Glu in the aqueous solution of sodium hydroxide and reduced to PDI radical anion ($\text{PDI}^{\cdot-}$) by $\text{Na}_2\text{S}_2\text{O}_4$. The introduced Glu not only optimizes the solubility of PDI but also interferes with the $\pi-\pi$ stacking of PDI-Glu to maintain the stability of $\text{PDI}^{\cdot-}$ as an isolating group⁴⁴. Furthermore, PDI-SG can transform from spherical nanoparticles to short nanofibers in the TME and continue to grow into elongated nanofibers in tumor cells, which could achieve high tumor accumulation and long retention simultaneously (Scheme 1). PDI-SG can be reduced to $\text{PDI}^{\cdot-}$ in hypoxic tumors but not in normal tissues. The hypoxia-induced $\text{PDI}^{\cdot-}$ exhibits a potent photothermal effect, high specificity, and significant inhibition capability toward tumors.

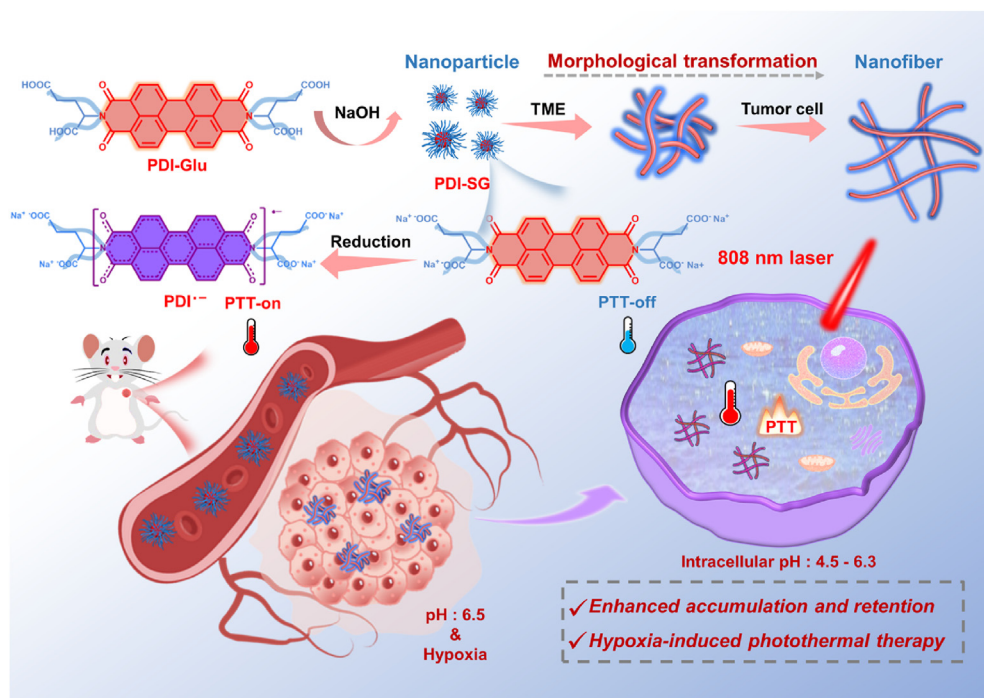
2. Materials and methods

2.1. EPR spectra of PDI-SG and $\text{PDI}^{\cdot-}$

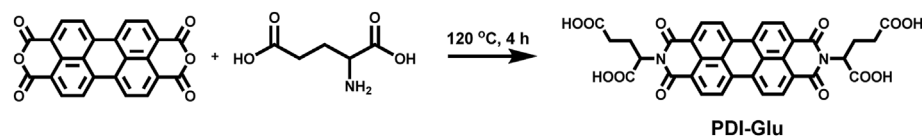
First, PDI-SG solution was added into an electron paramagnetic resonance (EPR) sample tube, and then the EPR signal of PDI-SG was characterized after the air inside the sample tube was replaced by inert gas. $\text{Na}_2\text{S}_2\text{O}_4$ was added into the sample tube to obtain $\text{PDI}^{\cdot-}$, and the concentration of $\text{Na}_2\text{S}_2\text{O}_4$ was 10 times that of PDI-SG. After PDI-SG solution and $\text{Na}_2\text{S}_2\text{O}_4$ were mixed for 30 min, the EPR signal of $\text{PDI}^{\cdot-}$ was measured.

2.2. Morphological transformation of PDI-SG

PDI-SG solutions were diluted with water and buffer solutions at pH 6.5 and 5.0, respectively. The three solutions were left at room



Scheme 1 Schematic representation of the preparation and the morphological transformation of PDI-SG as well as the hypoxia-induced generation of $\text{PDI}^{\cdot-}$ for tumor-specific PTT.



Scheme 2 The synthetic route of PDI-Glu.

temperature for 12 h and then dripped onto silicon wafers, air-dried at room temperature, and observed by scanning electron microscopy (SEM).

2.3. The photothermal effect of $\text{PDI}^{\cdot-}$

PDI-SG can be easily reduced to $\text{PDI}^{\cdot-}$ by $\text{Na}_2\text{S}_2\text{O}_4$, and the concentration of $\text{Na}_2\text{S}_2\text{O}_4$ is ten times that of PDI-SG. After PDI-SG solution and $\text{Na}_2\text{S}_2\text{O}_4$ were mixed for 30 min, $\text{PDI}^{\cdot-}$ from various concentrations (0.10–0.20 mmol/L) of PDI-SG solution was subjected to 808 nm laser irradiation. For PDI-SG at a fixed concentration of 0.15 mmol/L, the temperature of $\text{PDI}^{\cdot-}$ under irradiation (0.8–1.2 W/cm²) was also recorded.

3. Results and discussion

3.1. Preparation and characterization of PDI-SG

PDI-Glu was synthesized in one step from perylene-3,4,9,10-tetracarboxylic acid dianhydride and L-Glu with the presence of imidazole (Scheme 2)⁴⁹. The structure of PDI-Glu was validated by ¹H nuclear magnetic resonance spectroscopy (Supporting Information Fig. S1). PDI-Glu can self-assemble in the aqueous solution of sodium hydroxide to obtain PDI-SG. First, the chirality of PDI-SG was studied by circular dichroism (CD) spectroscopy. As shown in Supporting Information Fig. S2, significant cotton effects of PDI-SG can be observed. As shown in Fig. 1A, PDI-Glu

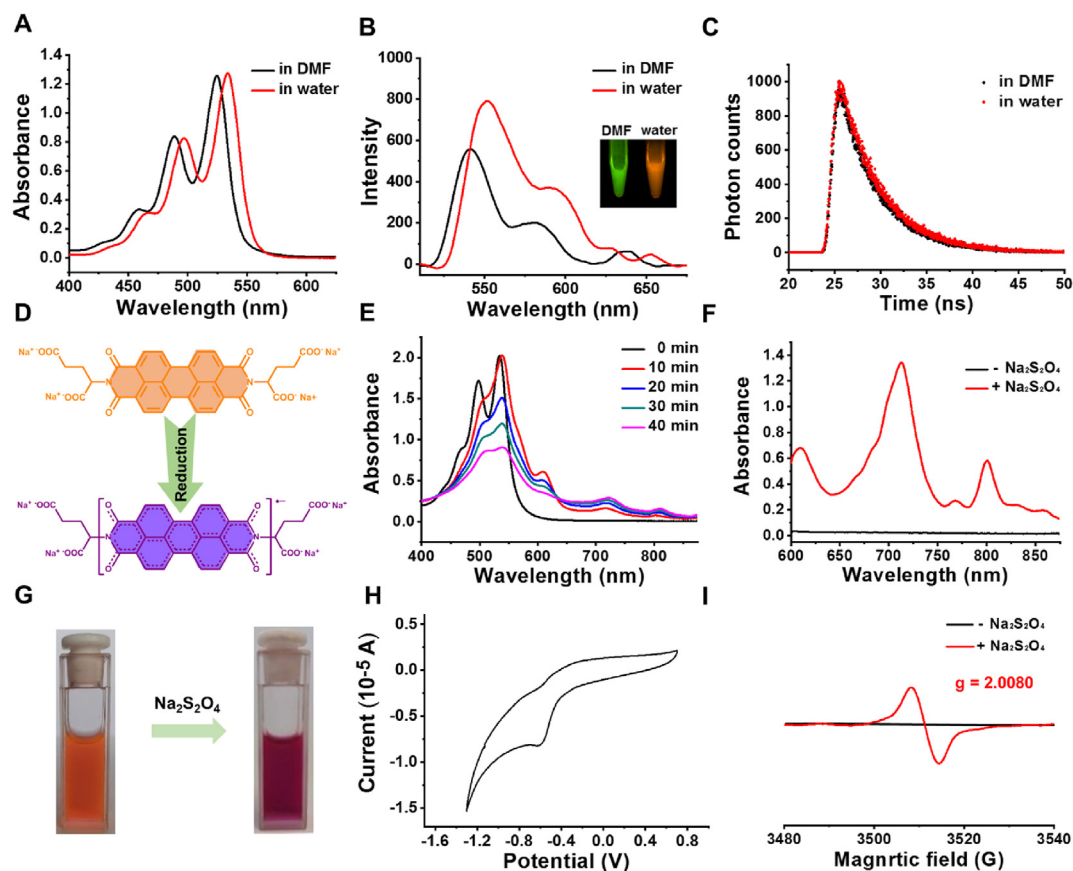


Figure 1 (A) Absorption spectra of PDI-Glu (20 μmol/L) in DMF and PDI-SG (20 μmol/L) in water. (B) Fluorescence spectra and fluorescence images of PDI-Glu (1 μmol/L) in DMF and PDI-SG (1 μmol/L) in water. (C) Fluorescence lifetimes of PDI-Glu in DMF and PDI-SG in water. (D) Changes of the chemical structure of PDI-SG to $\text{PDI}^{\cdot-}$. (E) Changes of the absorption spectra of PDI-SG after the addition of $\text{Na}_2\text{S}_2\text{O}_4$ (20 μmol/L) for different times. (F) The amplified absorption spectra (600–900 nm) of PDI-SG before and after the addition of $\text{Na}_2\text{S}_2\text{O}_4$ for 1 h. (G) Colour change of PDI-SG solution after the addition of $\text{Na}_2\text{S}_2\text{O}_4$ under inert conditions. (H) CV of the aqueous solution of PDI-SG. (I) EPR spectra of PDI-SG before and after the addition of $\text{Na}_2\text{S}_2\text{O}_4$.

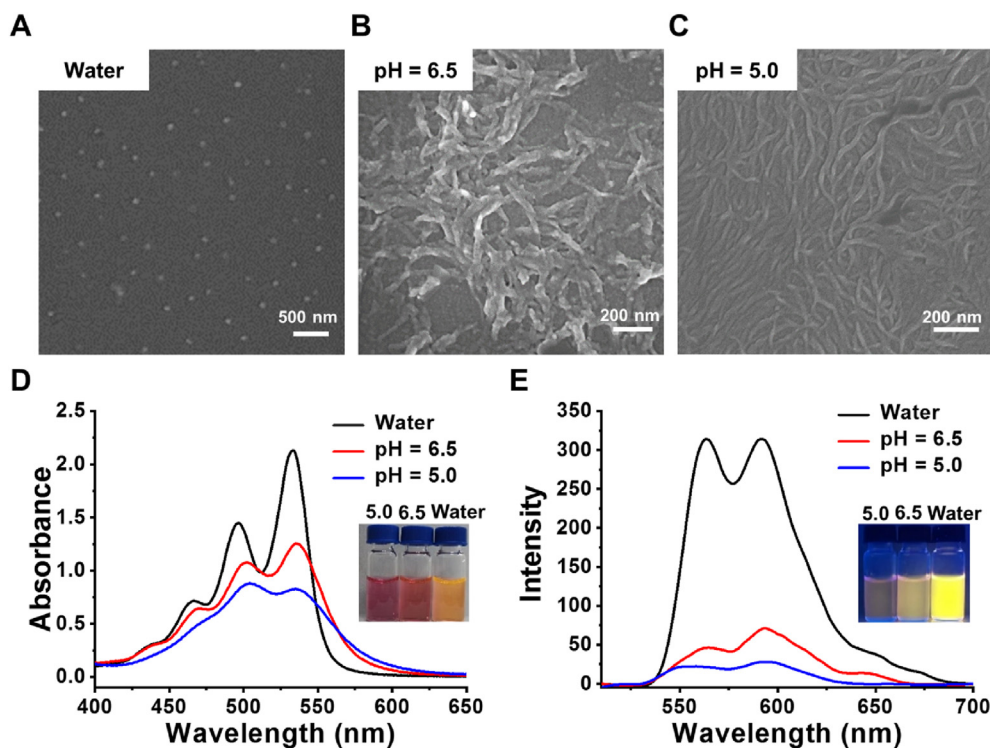


Figure 2 SEM images of PDI-SG (A) in water, at (B) pH 6.5 and (C) pH 5.0. (D) Absorption and (E) fluorescence spectra of PDI-SG under different conditions.

in *N,N*-dimethylformamide (DMF) has three absorption peaks at 460, 490, and 525 nm. Compared with the absorption spectrum of PDI-Glu in DMF, that of PDI-SG in water exhibits an obvious red shift of about 10 nm, which might be due to the aggregation of molecules. What is unexpected is that PDI-SG in water displays enhanced fluorescence intensity (Fig. 1B). Meanwhile, the green fluorescence of PDI-Glu in DMF changes to the orange fluorescence of PDI-SG in water under the excitation of the same 365 nm light (inset in Fig. 1B). As shown in Supporting Information Figs. S3 and S4, the fluorescence spectra of PDI-SG differ substantially between the lower concentrations and the higher concentrations. The luminescence lifetime of PDI-SG in water (4.34 ns) is slightly longer than that of PDI-Glu in DMF (4.13 ns) (Fig. 1C). These results confirm the formation of nanoscale aggregates. The size distribution and stability of the PDI-SG were investigated. As shown in Supporting Information Fig. S5, PDI-SG has a size of 63.5 nm and exhibits excellent stability in water and PBS.

3.2. Formation and characterization of $\text{PDI}^{\cdot-}$

PDI-SG can be easily reduced to $\text{PDI}^{\cdot-}$ by $\text{Na}_2\text{S}_2\text{O}_4$ (Fig. 1D), and the formation of $\text{PDI}^{\cdot-}$ is time- and $\text{Na}_2\text{S}_2\text{O}_4$ concentration-dependent (Fig. 1E and Supporting Information Fig. S6). As shown in the absorption spectra in Fig. 1E and F, new absorption peaks appear at 609, 715, and 804 nm and exhibit increasing absorbance with the decrease of the absorbance of PDI-SG at 466, 497, and 534 nm. In a visual sense, the orange color of PDI-SG in water turned purple because of the generation of $\text{PDI}^{\cdot-}$ (Fig. 1G). The purple color of the solution turned pink when exposed to the air due to the quenching of $\text{PDI}^{\cdot-}$ by oxygen (Supporting Information Fig. S7). The fluorescence of PDI-SG was quenched when it was reduced to $\text{PDI}^{\cdot-}$, but it was partially

recovered when the solution was exposed to the air (Supporting Information Fig. S8). The zeta potentials of PDI-SG and $\text{PDI}^{\cdot-}$ are -13 and -46.5 eV, respectively (Supporting Information Fig. S9). As shown in Fig. 1H, the reduction peak of PDI-SG assigned to the one-electron reduction process of $\text{PDI}/\text{PDI}^{\cdot-}$ is -0.63 V in the cyclic voltammogram (CV). To directly prove that PDI-SG could be reduced to $\text{PDI}^{\cdot-}$ by $\text{Na}_2\text{S}_2\text{O}_4$, EPR spectroscopy of PDI-SG before and after the addition of $\text{Na}_2\text{S}_2\text{O}_4$ was studied. As shown in Fig. 1I, the EPR signal of PDI-SG after reduction was observed with a g-factor of 2.0080, suggesting the formation of $\text{PDI}^{\cdot-}$. To our delight, the absorbance of $\text{PDI}^{\cdot-}$ at 808 nm remained almost unchanged after storage at 37°C for 6 h, which demonstrated that $\text{PDI}^{\cdot-}$ was highly stable under inert conditions (Supporting Information Fig. S10).

3.3. Morphological transformation of PDI-SG

Both TME and tumor cells have an acidic environment, and the pH of the lysosomes is lower than the extracellular pH of tumor cells. The acidic environment of TME and lysosomes was simulated by using buffer solutions to investigate the pH-triggered morphological transformations of PDI-SG. As observed in Fig. 2A, SEM results show that PDI-SG exists in the form of small spherical nanoparticles with sizes of around 60 nm, which is beneficial to the accumulation of tumors. The spherical nanoparticles can transform into short nanofibers at pH 6.5 (Fig. 2B) and continue to grow into elongated nanofibers at pH 5.0 (Fig. 2C). This transition will prolong the retention time of PDI-SG in the tumors. UV-Vis absorption spectroscopy was also utilized to examine the aggregation behaviors of PDI-SG under acidic conditions. As shown in Fig. 2D, PDI-SG possesses characteristic absorption peaks at 534 and 497 nm, which are put down

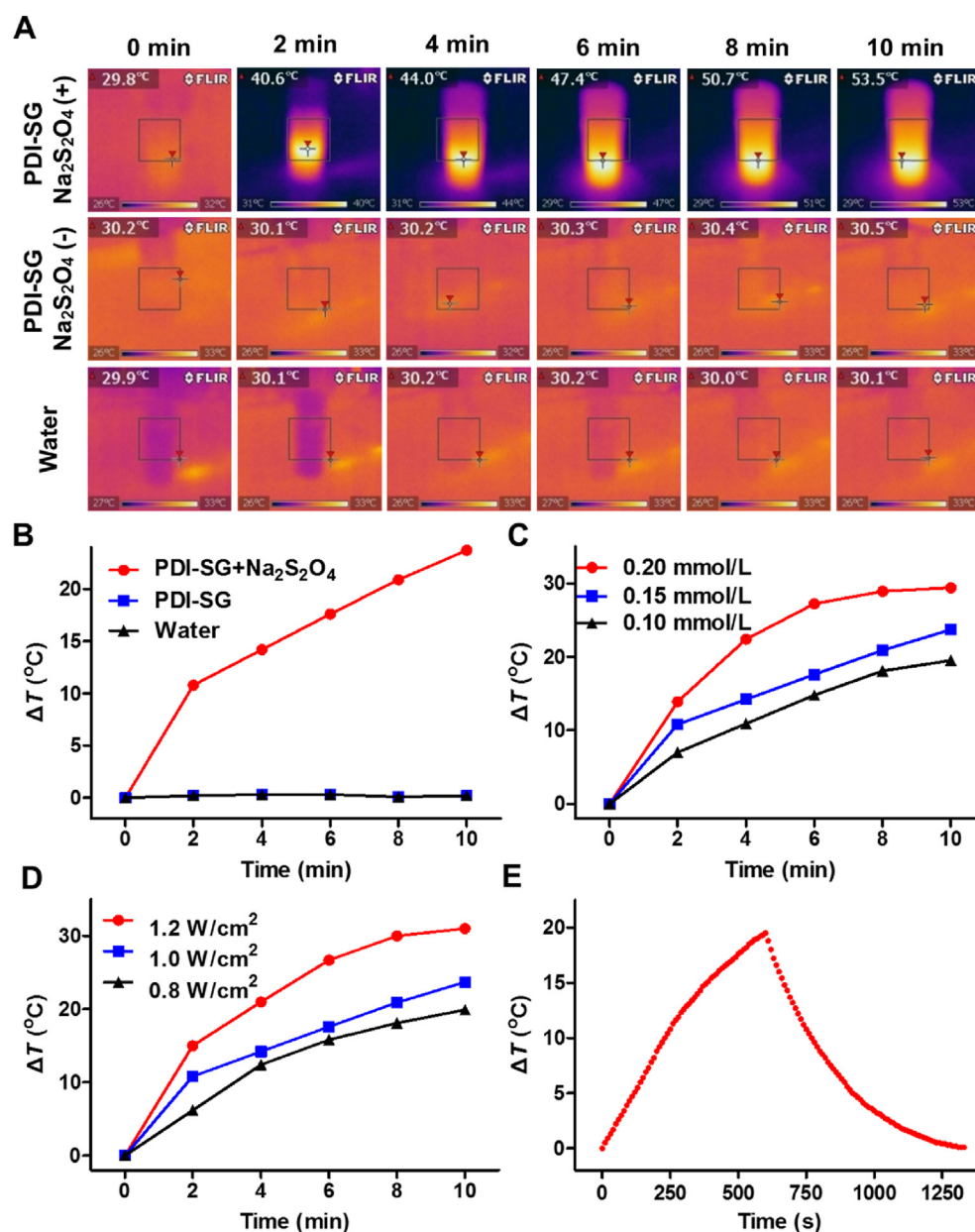


Figure 3 (A) The infrared thermal images of PDI-SG treated with or without Na₂S₂O₄ and water under irradiation (1.0 W/cm²). The interval between infrared thermal images was 2 min. (B) Photothermal heating curves of PDI^{•−} from PDI-SG (0.15 mmol/L) under irradiation with water and PDI-SG as the references. Photothermal properties of (C) PDI^{•−} from various concentrations (0.10–0.20 mmol/L) of PDI-SG subjected to irradiation (1.0 W/cm²) and (D) PDI^{•−} from a fixed concentration of PDI-SG (0.15 mmol/L) under irradiation of varying laser power densities (0.8–1.2 W/cm²). (E) Photothermal heating curve of PDI^{•−} from 0.10 mmol/L of PDI-SG under the irradiation followed by natural cooling.

to the $0 \rightarrow 0$ and $0 \rightarrow 1$ bands in the $\pi \rightarrow \pi^*$ vibronic transition, respectively. The absorbance of the two characteristic peaks of PDI-SG decreased sharply at pH 6.5, and the value ratio of the two characteristic peaks (1.16) at pH 6.5 is lower than that of PDI-SG itself (1.47), which implies the occurrence of acid-induced morphological transformation behaviour^{45,50}. The value ratio of the two characteristic peaks drops to 0.94 at pH 5.0, which indicates that the morphological transformation of PDI-SG is further enhanced by increased acidity. The sharply decreased fluorescence intensity of PDI-SG in Fig. 2E also confirms the changes in the aggregation states of fluorophores under acidic conditions.

3.4. In vitro photothermal effect of PDI^{•−}

First, the PDI-SG solution (0.15 mmol/L) and that treated with Na₂S₂O₄ were irradiated by an 808 nm laser. As observed in Fig. 3A and B, there is almost no temperature increase in the PDI-SG solution without Na₂S₂O₄ or in water, while the temperature of the PDI-SG solution after the addition of Na₂S₂O₄ substantially increases by 23.7 °C. When different concentrations of PDI-SG treated with Na₂S₂O₄ are irradiated by an 808 nm laser, the temperature rises more rapidly with increasing concentrations of PDI-SG (Fig. 3C and Supporting Information Fig. S11). Similarly,

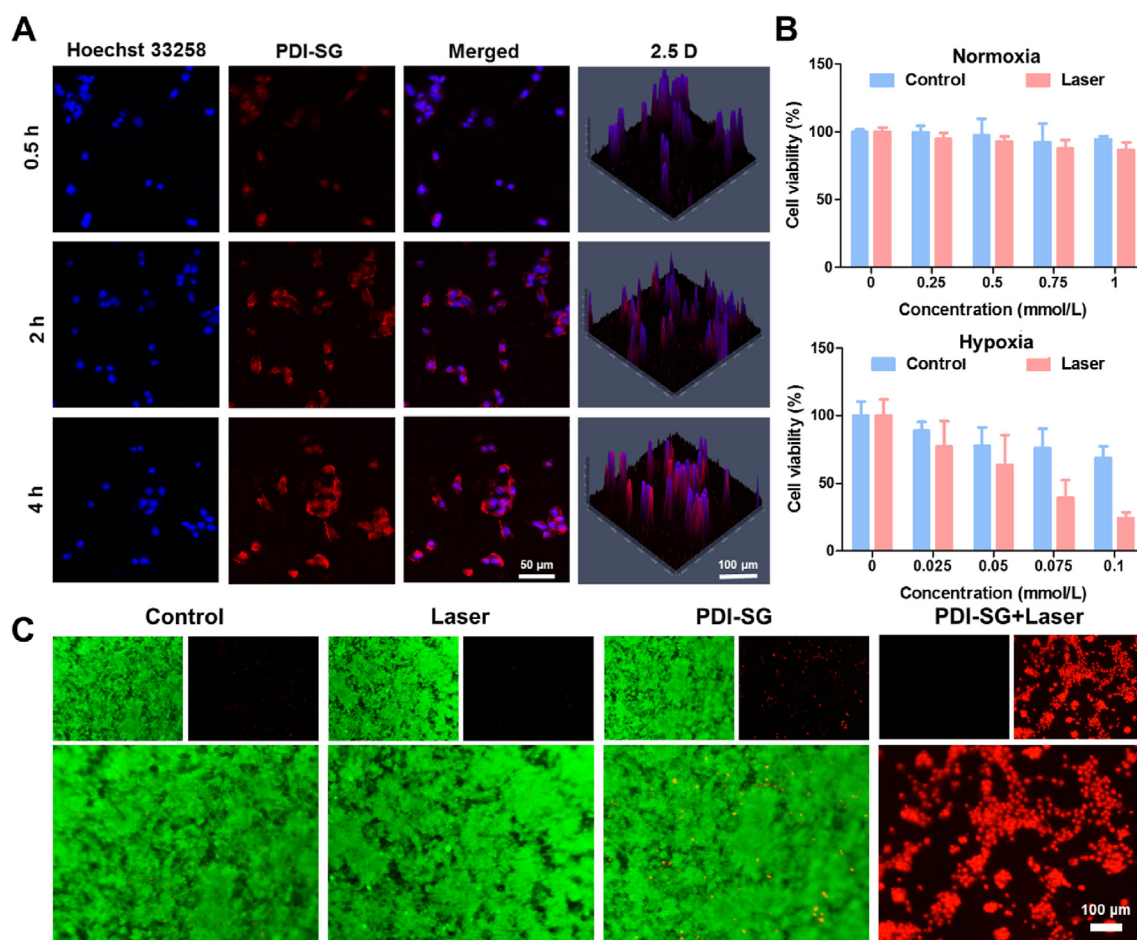


Figure 4 (A) CLSM images of 4T1 cells incubated with PDI-SG at 37 °C for 0.5, 2, and 4 h. (B) Cytotoxicity of PDI-SG toward 4T1 cells under normoxia and hypoxia conditions without or with irradiation (1.0 W/cm², 10 min). (C) Live/dead staining assays of 4T1 cells incubated with PDI-SG or culture medium under hypoxic conditions with or without irradiation.

PDI-SG at a constant concentration (0.15 mmol/L) was treated with Na₂S₂O₄ and irradiation. The higher the laser power densities are, the higher the temperature can be reached (Fig. 3D and Supporting Information Fig. S12). The photothermal conversion efficiency of PDI⁻ was determined to be 20.1% (Fig. 3E and Supporting Information Fig. S13).

3.5. Internalization of PDI-SG and cytotoxicity of PDI⁻

To evaluate the photothermal antitumor effect of PDI⁻, the internalization of PDI-SG by mouse breast cancer (4T1) and human cervical carcinoma (HeLa) cells was first investigated. The enhancement of fluorescence intensity is time-dependent from 0.5 to 4 h in the confocal laser scanning microscopy (CLSM) images (Fig. 4A and Supporting Information Fig. S14). The efficient internalization of PDI-SG is a prerequisite for the PTT effect of PDI⁻.

The cytotoxicity of PDI-SG toward 4T1 and HeLa cells was further studied through 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assays. For 4T1 and HeLa cells in normoxia conditions, no apparent cytotoxicity was observed when the concentration of PDI-SG was up to 1 mmol/L (Fig. 4B and Supporting Information Fig. S15). However, PDI-SG exhibited a potent killing effect on tumor cells under hypoxia conditions at a low concentration of 0.1 mmol/L. At the same time, similar results

were obtained from the bright field images of cells (Supporting Information Fig. S16). Under hypoxic conditions, the cells only treated with irradiation remained intact, but the morphologies of the cells incubated with increasing concentrations of PDI-SG were destroyed gradually after irradiation. Nearly all the cells were dead in spherical shapes when the concentration of PDI-SG reached 0.1 mmol/L. The PTT effect of PDI-SG after reduction was further demonstrated by live/dead cell-staining assays toward 4T1 and HeLa cells under hypoxic conditions. It is apparent that only the cells treated by PDI-SG with 808 nm laser irradiation under hypoxic conditions are in red fluorescence, and the proportion of the red fluorescence increases with the concentrations of PDI-SG (Fig. 4C and Supporting Information Fig. S17). However, the green fluorescence of live cells dominates the field of vision in the cells treated with only PDI-SG or irradiation (Fig. 4C and Fig. S17). Flow cytometry results (Supporting Information Fig. S18) show that the cells in the PDI-SG + Laser group die mainly through necrosis.

3.6. In vivo photothermal therapeutic effect of PDI⁻ after intratumoral injection

4T1 tumor-bearing BALB/c mice were selected to study the production of PDI⁻ in the tumors. All experimental procedures

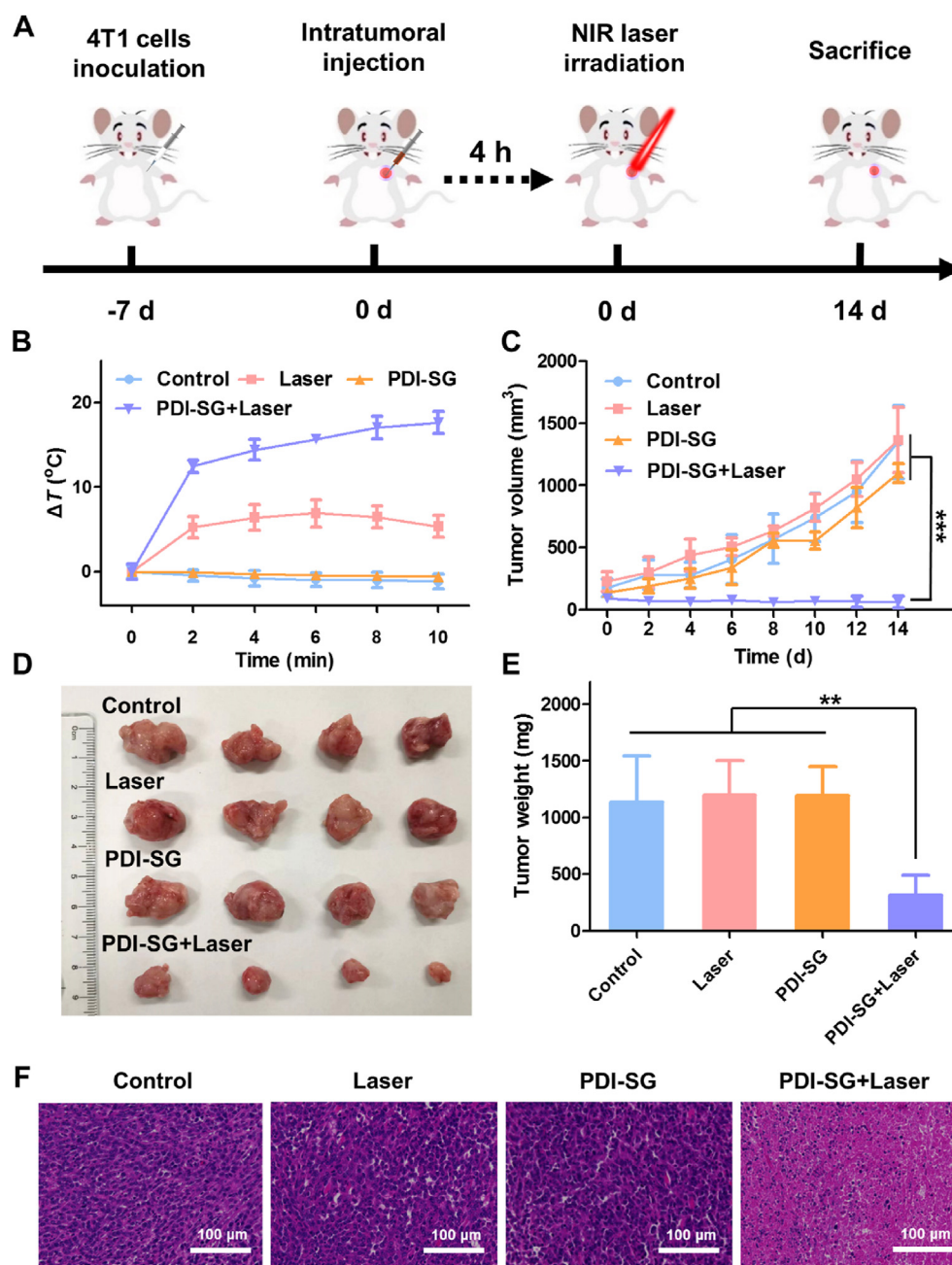


Figure 5 (A) Schematic representation of the animal experiments for evaluation of the photothermal therapeutic effect of $\text{PDI}^{\cdot-}$ after intratumoral injection. (B) Temperature changes of the tumor with different treatments *versus* time. (C) Changes in the average tumor volumes of the mice after different therapies. (D) Photos of the excised tumor. (E) Average weights of the tumor collected from the mice. (F) H&E staining of the tumor tissues from the mice sacrificed after 14 days post treatments. Data are presented as mean \pm SD ($n = 4$). ** $P < 0.01$, *** $P < 0.001$.

were executed according to the protocols approved by the Animal Ethics Committee of Changchun Institute of Applied Chemistry, Chinese Academy of Sciences (Approved No. 2022-0006). The robust fluorescence brightness of PDI-SG will be quenched when PDI-SG is reduced to $\text{PDI}^{\cdot-}$. As shown in Supporting Information Fig. S19, PDI-SG disperses throughout the tumor after intratumoral injection. The fluorescence signal of the tumor decreased sharply after 1 h of injection and completely disappeared 12 h later, which should be attributed to the generation of $\text{PDI}^{\cdot-}$ and the morphological transformation of PDI-SG.

After intratumoral injection of PDI-SG for 2, 3, 4, and 5 h, the tumors were irradiated, respectively (Supporting Information Fig. S20). After the injection of PDI-SG for 4 h, the temperature increase of the tumors under laser irradiation reached the maximum (Supporting Information Fig. S21A). The temperature changes of the tumors indicate that PDI-SG could be reduced to $\text{PDI}^{\cdot-}$ in the TME, and $\text{PDI}^{\cdot-}$ has good photothermal performance *in vivo*. Therefore, 4 h post-injection should be selected as the optimum treatment time for the photothermal therapeutic experiments.

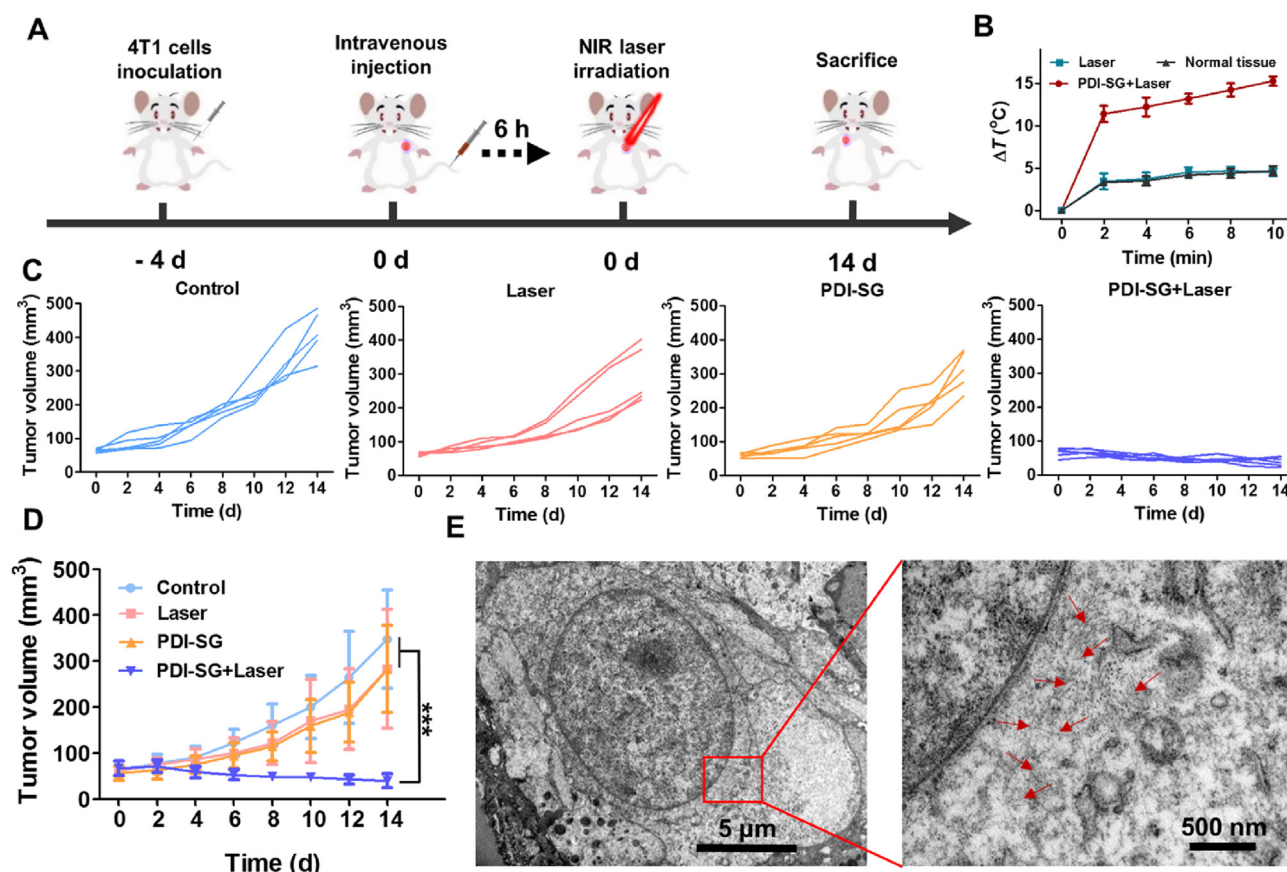


Figure 6 (A) Schematic representation of the animal experiments for evaluation of the photothermal therapeutic effect of PDI[−] after intravenous injection. (B) Temperature changes of the tumor with different treatments *versus* time. (C) Growth curves of the tumor volumes of each mouse. (D) Changes in the average tumor volumes. (E) TEM images of the tumor tissue slices of the mouse after intravenously injected with PDI-SG for 12 h. Data are presented as mean \pm SD ($n = 5$). *** $P < 0.001$.

The normal tissues in the symmetrical positions of the tumors were subcutaneously injected with PDI-SG and irradiated after injection for 4 h. As shown in Fig. S21B, the temperatures of the normal tissues only increase by 3.7 $^{\circ}\text{C}$, much lower than the temperature changes of the tumors (16.3 $^{\circ}\text{C}$). These results demonstrate that PDI-SG could be reduced to PDI[−] in the hypoxic TME but not in the normal tissues. The specificity of the hypoxia-induced photothermal properties prevents the normal tissues around the tumors from being damaged during irradiation.

The 4T1 tumor-bearing BALB/c mice were randomly divided into four groups, including Control, Laser, PDI-SG, and PDI-SG + Laser. As illustrated in Fig. 5A, the mice in PDI-SG and PDI-SG + Laser groups were intratumorally administered with PDI-SG (1 mmol/L, 50 μL), and the mice in Laser and PDI-SG + Laser groups were irradiated after injection for 4 h. The temperature changes of the tumors in the PDI-SG + Laser group were superior compared with those in other groups, demonstrating the specific and excellent photothermal effect of PDI[−] *in vivo* (Fig. 5B and Supporting Information Fig. S22). As described in Fig. 5C, remarkable tumor inhibition can be observed in the PDI-SG + Laser group. On the contrary, the tumor volumes of the mice in other groups are on the rise within 14 days, indicating that laser irradiation or PDI-SG alone negligibly inhibits tumor growth. 14 days after various treatments, the mice were sacrificed, and the tumors were excised and photographed. Compared with the other three groups, the mice in the PDI-SG + Laser group had the

smallest tumors (Fig. 5D). A similar result can be obtained from the tumor weights (Fig. 5E). Thereafter, staining was also performed to study the tumors. Moreover, there is a prominent reduction of tumor cells in the hematoxylin and eosin (H&E) staining image of the PDI-SG + Laser group (Fig. 5F), which further evidence that PDI-SG combined with 808 nm laser irradiation has a strong inhibitory effect on tumor cells. The body weights of the mice maintained stable throughout the PTT period (Supporting Information Fig. S23). Furthermore, the H&E staining images of the major organs (Supporting Information Fig. S24) and the results of the complete blood panels of the mice in the PDI-SG + Laser group (Supporting Information Fig. S25) show negligible differences from those in other groups, which proves that PDI-SG combined with 808 nm laser irradiation has no obvious systemic toxicity.

3.7. *In vivo* photothermal therapeutic effect of PDI[−] after intravenous injection

The photothermal performance of PDI-SG *in vivo* after intravenous injection was further studied. After PDI-SG was intravenously injected for 4, 6, 8, and 12 h, respectively, the increase of tumor temperatures after 808 nm laser irradiation was monitored. As displayed in Supporting Information Figs. S26 and S27A, the temperatures of the tumors after 6 h of injection reach a maximum after laser irradiation, indicating that the concentration of the

generated $\text{PDI}^{\cdot-}$ is the highest at that time. After 8 or 12 h, the temperature elevations are nearly the same, implying that $\text{PDI}^{\cdot-}$ remains stable in the tumor regions.

The normal tissues on the opposite side of the tumors were also irradiated after intravenous injection of PDI-SG for 6 h to assess the biocompatibility of the intravenously injected PDI-SG during laser irradiation. As shown in Fig. S27B, normal tissues produce only negligible temperature changes compared to the tumors. That is, the photothermal activity of PDI-SG is turned off during blood circulation, and $\text{PDI}^{\cdot-}$ is generated only in the tumors, which greatly enhances the safety of the PTA.

Inspired by the morphological transformation and the tumor-specific photothermal activity of PDI-SG, we explored the anti-tumor efficiency of PDI-SG after intravenous injection into the 4T1 tumor-bearing BALB/c mice (Fig. 6A). The tumors of the mice in Laser and PDI-SG + Laser groups were irradiated (0.8 W/cm^2 , 10 min) after intravenous administration of PDI-SG for 6 h. As illustrated in Fig. 6B, the temperatures of the tumor tissues of the mice in the PDI-SG + Laser group are much higher than those in the Laser group. Ultimately, the mice in the PDI-SG + Laser group had much smaller tumors than those in other groups (Fig. 6C and D), indicating that PDI-SG exerted excellent photothermal antitumor effect after being reduced to $\text{PDI}^{\cdot-}$. The photograph and weighing of tumors further demonstrated the superior photothermal suppression effect of PDI-SG toward tumors under laser irradiation (Supporting Information Figs. S28 and S29). Moreover, the morphological transformation of PDI-SG *in vivo* was visually verified by bio-TEM. In the TEM images of the tumor slices of the PDI-SG-treated mouse after intravenous injection, nanofibers are distinctly observed inside the tumor cells (Fig. 6E), providing more convincing evidence for the advantage of PDI-SG in tumor treatment. During the whole treatment period, there were negligible changes in the body weights of the mice (Supporting Information Fig. S30). The haematological data of the mice in the treatment groups are similar to those in the control group (Supporting Information Fig. S31). In addition, the H&E staining results of tumors and major organs further demonstrate that PDI-SG exerts a potent photothermal effect only in the tumors without causing obvious systemic toxicity (Supporting Information Figs. S32 and S33). The biodistribution of PDI-SG in major organs was studied after the mice were intravenously injected with PDI-SG for 48 h. As shown in Supporting Information Fig. S34, relatively stronger fluorescence can be detected in the kidneys.

4. Conclusions

In summary, an activatable PTA was developed for the selective treatment of tumors. The aggregates of the Glu-substituted PDI (PDI-SG) have been successfully obtained, and they could transform from spherical nanoparticles to nanofibers in the tumors for enhanced tumor retention. In addition, PDI-SG could be reduced to $\text{PDI}^{\cdot-}$ with robust NIR absorption under hypoxic conditions. The hypoxia-induced $\text{PDI}^{\cdot-}$ exhibits excellent PTT efficiency and biological safety toward the 4T1 tumor-bearing mice, and at the same time, it can minimize the damage to the normal tissues surrounding the tumors during laser irradiation. The construction of hypoxia-induced radical anions and morphologically transformable nanomaterials sheds light on the development of hypoxia-specific phototherapeutic agents for tumors or other diseases.

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Author contributions

Hongyu Wang: Writing – original draft, Methodology, Investigation, Formal analysis. Dengyuan Hao: Methodology, Investigation. Qihang Wu: Methodology, Investigation. Tingting Sun: Writing – review & editing, Funding acquisition, Conceptualization. Zhigang Xie: Writing – review & editing, Supervision, Project administration, Funding acquisition, Conceptualization.

Conflicts of interest

The authors declare no conflicts of interest.

Appendix A. Supporting information

Supporting information to this article can be found online at <https://doi.org/10.1016/j.apsb.2024.09.017>.

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