



Supporting Information

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A Multi-Scale Structural Engineering Strategy for High-Performance MXene Hydrogel Supercapacitor Electrode

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Supplementary Materials for

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Calculation formula of Specific capacitance ($C_{specific}$) Areal capacitance (C_{areal}), Energy density (E), and Power density (P)

$$C_{specific} = (\int IdV)/(s\Delta Vm) \text{ (F g}^{-1}\text{) (3-electrode configuration);} \quad (1)$$

$$C_{areal,CV} = (\int IdV)/(s\Delta VA) \text{ (F cm}^{-2}\text{) (2-electrode configuration);} \quad (2)$$

$$C_{areal,GCD} = (i\Delta t)/(A\Delta V) \text{ (F cm}^{-2}\text{);} \quad (3)$$

$$E = C_{areal,GCD}\Delta V^2/7.2 \text{ (mWh cm}^{-2}\text{);} \quad (4)$$

$$P = 3.6E/\Delta t \text{ (mW cm}^{-3}\text{).} \quad (5)$$

Where m is the solid content of MXene hydrogel electrode, I is current, s is the scan rate, ΔV is the potential window, A is the area of MSC electrodes, i is the applied current, and Δt is the discharge time.

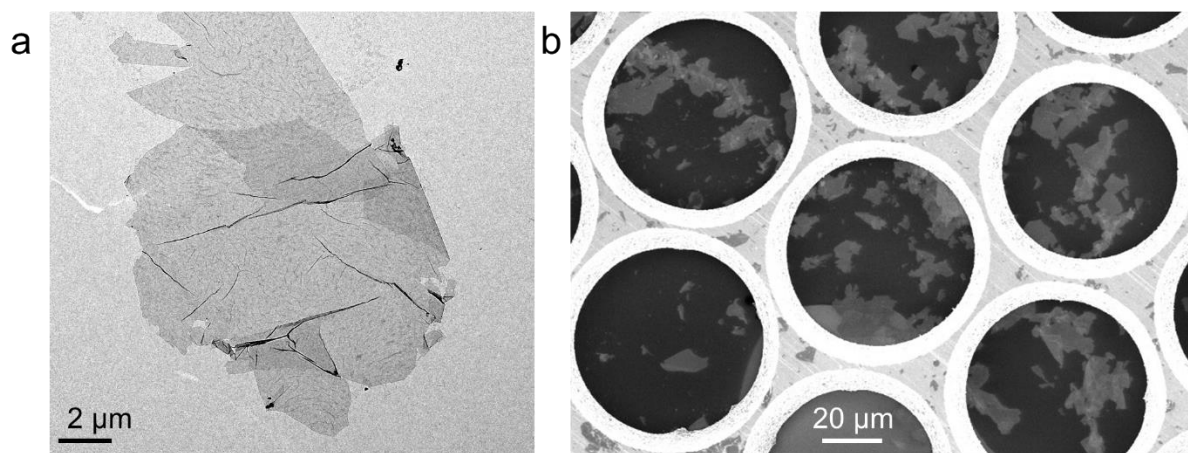


Figure S1. (a) The TEM and (b) SEM image of large MXene flakes for statistical purposes.

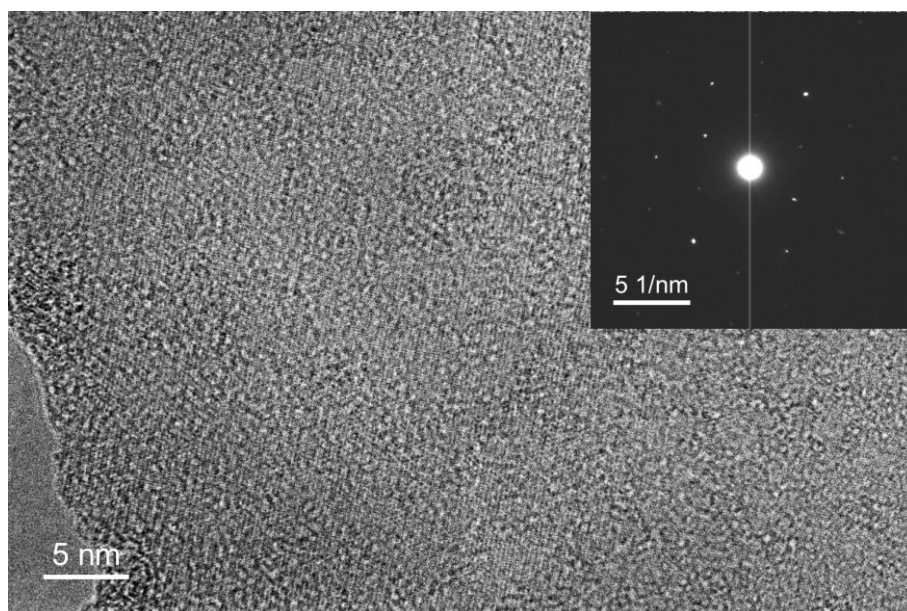


Figure S2. The HRTEM and (inset) SAED image of MXene flakes.

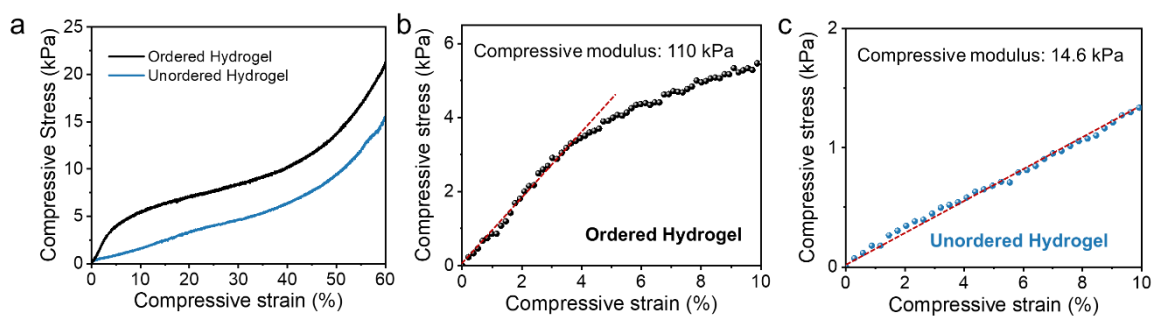


Figure S3. (a) The compressive stress-strain curves and compressive modulus of (b) ordered and (c) unordered MXene hydrogel.

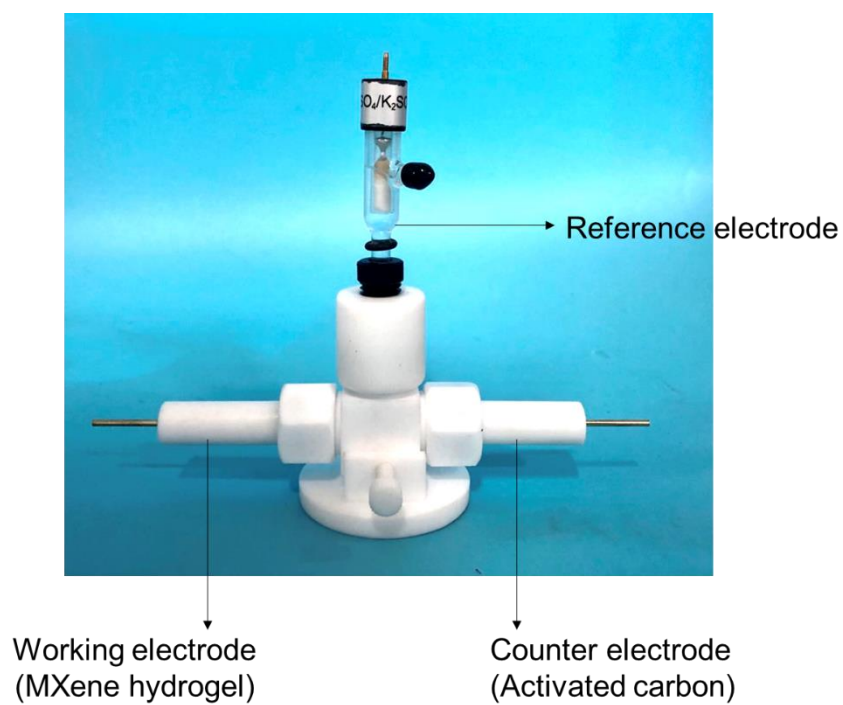


Figure S4. Swagelok-type cell with three-electrode setup.

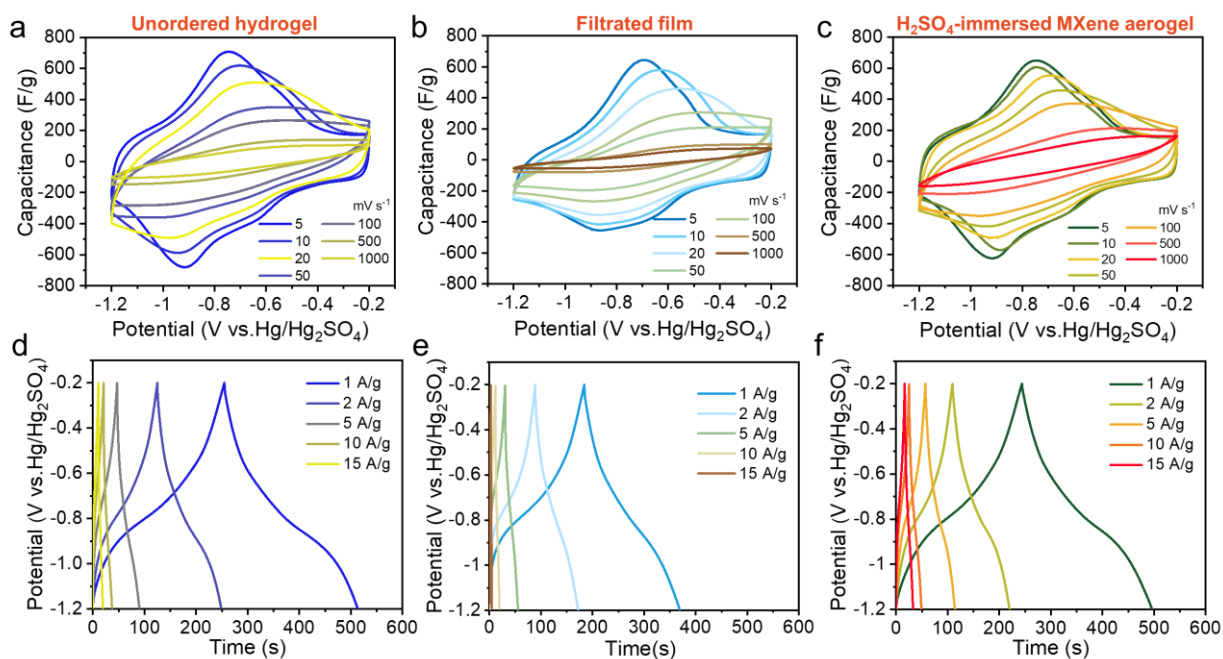


Figure S5. (a, b, and c) CV curves and (d, e, and f) GCD profiles of unordered hydrogel, filtrated film, and H₂SO₄-immersed MXene aerogel, respectively.

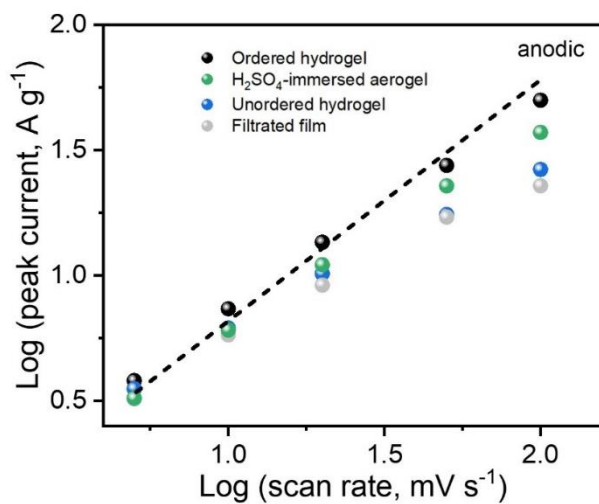


Figure S6. Plot of the anodic peak current against the scan rate for the four MXene samples.

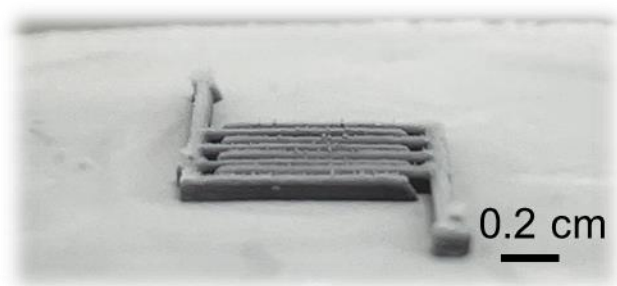


Figure S7. Unidirectional-freezing MXene MSC.

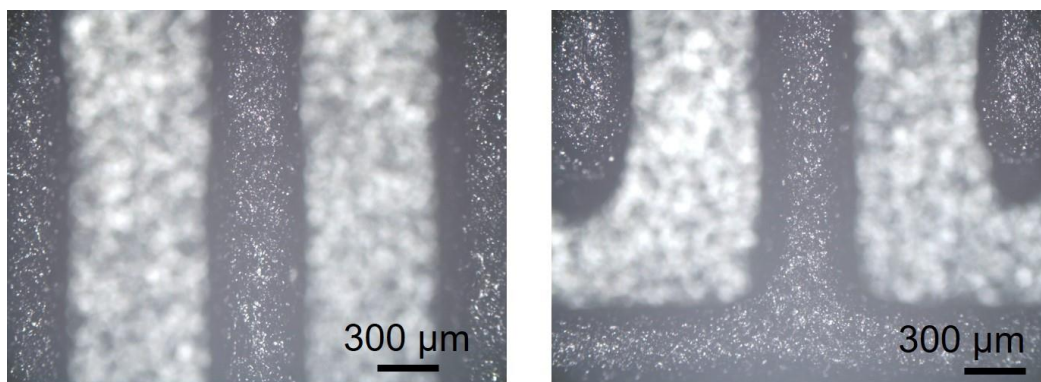


Figure S8. The optical microscope photographs of 3D-printed MXene MSC.

Table S1. The comparison of MXene-based MSCs electrochemical performance.

Materials	Methods	Electrolyte	Areal capacitance (mF cm ⁻²)	Rate performance (mF cm ⁻²)	Energy density (μWh cm ⁻²)	Power density (mW cm ⁻²)	Ref.
Ti ₃ C ₂ T _x	3D-printing	PVA/H ₂ SO ₄	1035 (2 mV s ⁻¹)	4.68 (500 mV s ⁻¹)	51.7	5.7	[1]
Ti ₃ C ₂ T _x	3D-printing	PVA/H ₂ SO ₄	2100 (1.7 mA cm ⁻²)	1000 (42.7 mA cm ⁻²)	24.4	0.64	[2]
MXene-AgNW-MnONW-C60	3D-printing	PVA/KOH	216.2 (10 mV s ⁻¹)	72.8 (2000 mV s ⁻¹)	19.2	58.3	[3]
MXene sediment	Screen-printing	PVA/H ₂ SO ₄	158 (0.08 mA cm ⁻²)	127 (2.4 mA cm ⁻²)	1.64	0.778	[4]
Ti ₃ C ₂ T _x	Laser-scribing	PVA/H ₂ SO ₄	25 (20 mV s ⁻¹)	10 (1000 mV s ⁻¹)	0.77	46.6	[5]
Ti ₃ C ₂ T _x	Stamping	PVA/H ₂ SO ₄	61 (0.025 mA cm ⁻²)	50 (0.8 mA cm ⁻²)	0.63	0.33	[6]
Ti ₃ C ₂ T _x	Extrusion-printing	PVA/H ₂ SO ₄	43 (0.005 mA cm ⁻²)	33 (0.05 mA cm ⁻²)	0.32	0.0114	[7]
Ti ₃ C ₂ T _x /RuO ₂	Laser-engraving	1 M H ₂ SO ₄	60 (5 mV s ⁻¹)	50 (20 mV s ⁻¹)	19	1.5	[8]
Ti ₃ C ₂ T _x /Be ²⁺	Template-depositing	ZnSO ₄ /Gelatin	77.2 (5 mV s ⁻¹)	31 (100 mV s ⁻¹)	3.86	0.12	[9]
Ti ₃ C ₂ T _x /CNF	Template-depositing	PVA/H ₂ SO ₄	25.3 (2 mV s ⁻¹)	13 (200 mV s ⁻¹)	0.08	0.145	[10]
Ti ₃ C ₂ T _x /CNT	Extrusion-printing	PVA/H ₃ PO ₄	30.8 (0.025 mA cm ⁻²)	20 (0.4 mA cm ⁻²)	8.37	17.3	[11]
Ti₃C₂T_x	3D-printing and UFAT method	PVA/H₂SO₄	2004 (1.2 mA cm⁻²)	1202 (60 mA cm⁻²)	100	0.38	This work

Movie S1. 3D printing all-MXene MSC.

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