


Article

Facile Three-Component Synthesis, Insecticidal and Antifungal Evaluation of Novel Dihydropyridine Derivatives

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Abstract: In an attempt to find the neonicotinoid insecticides, twenty novel dihydropyridine derivatives were designed, “green” synthesized via one pot facile three-component reaction and evaluated for their bioactivities against *Tetranychus cinnabarinus*, *Myzus persicae*, *Brevicoryne brassicae*, *Fusarium oxysporum* f. sp. *vasinfectum*, *Magnaporthe oryzae*, *Sclerotinia sclerotiorum* and *Botrytis cinerea*. All of the tested compounds showed potent insecticidal activity, and some were much better in comparison with imidacloprid (IMI). Especially, compounds **3d** (LC₅₀: 0.011 mM) and **5c** (LC₅₀: 0.025 mM) were 12.2- and 5.4-fold more active than IMI (LC₅₀: 0.135 mM) against *T. cinnabarinus*, respectively. Moreover, out of all the derivatives, compound **3d** (LC₅₀: 0.0015 mM) exhibited the strongest insecticidal activity against *B. brassicae* and compound **3i** (LC₅₀: 0.0007 mM) displayed the strongest insecticidal activity against *M. persicae*. Surprisingly, when the concentration of compound **4** was 50 mg/L, the inhibition rate against *F. oxysporum* and *S. sclerotiorum* reached 45.00% and 65.83%, respectively. The present work indicated that novel dihydropyridine derivatives could be used as potential lead compounds for developing neonicotinoid insecticides and agricultural fungicides.

Keywords: neonicotinoid; insecticide; fungicide; dihydropyridine derivatives; three-component reaction

1. Introduction

Neonicotinoid insecticides are one of the most important chemical classes of insecticides introduced to global markets due to their broad spectrum of biological activities, favorable safety profile and unique mechanism of action [1], which have been registered globally in more than 120 countries for more than 25% of global insecticide market [2,3]. However, the great success of commercialization and the widespread and frequent use of these insecticides have inevitably led to the occurrence of resistance and cross-resistance [1,4–6]. In addition, reports of the toxicity of neonicotinoid insecticides on honey bees have raised concerns about whether the ecological balance has been destroyed [7,8]. Therefore, it is of great significance to seek a novel structure that is more efficient and less toxic as a potential candidate for future pest control.

Representative generation of commercial neonicotinoid insecticides is shown in Figure 1. The common molecular structural features of neonicotinoids consist of four sections: (i) aromatic heterocycle, (ii) flexible linkage, (iii) hydroheterocycle or guanidine/amidine and (iv) electron-withdrawing segment [9,10]. Through our previous work on pesticides [11–14] and other reported structure activity relationships for neonicotinoid insecticides [10], it has been found that pesticides containing dihydropyridine or dihydropyran rings have relatively low non-target organisms and environmental risks, high target specificity and a wide range of biological uses [15,16]. Furthermore, in recent years, multi-component reactions (MCRs) have become powerful tools for the synthesis of target molecules in organic chemistry, which are efficient, convenient, economical, practical and avoid purification and measurement of intermediate structures during the synthesis process [17,18]. On this basis, minimizing the amount of toxic waste and by-products and performing the reaction in the absence of non-environmental organic solvents is one of the goals that participated with green chemistry and green pesticides, as they were usually used in large quantities [19]. Thus, a series of derivatives with dihydropyridine as the core structure were designed and “green” synthesized by one-pot and three-component reactions and evaluated for their activity against *T. cinnabarinus*, *M. persicae*, *B. brassicae*, and four kinds of phytopathogens (Figure 2).

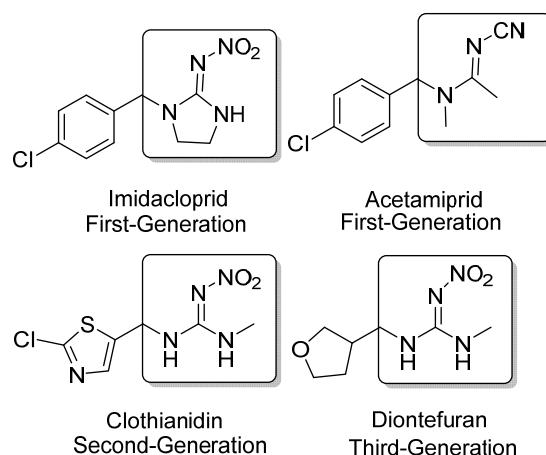


Figure 1. Representative generation of commercial neonicotinoid insecticides.

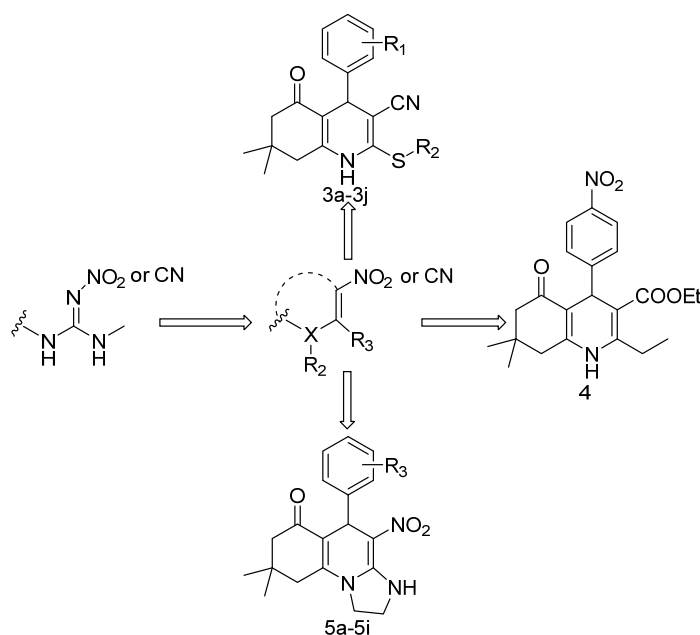
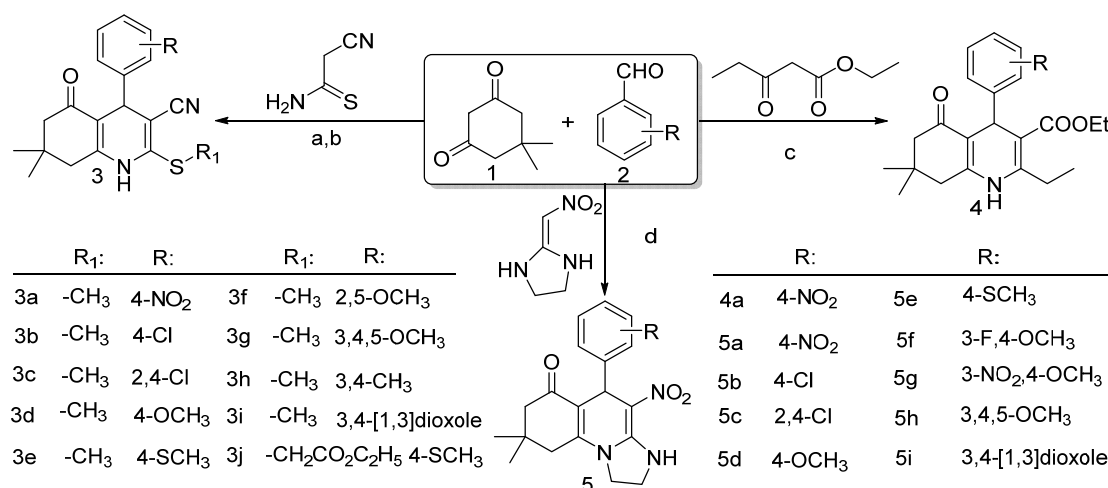


Figure 2. The molecular design of target compounds.

2. Results and Discussion

2.1. Chemistry

MCRs provide an efficient, economical, and rapid approach towards the efficient synthesis of diverse compounds and compound libraries. The combination of MCR-type chemistry-planning with evolutionary in vitro prediction of drug potential or biological properties is a new and powerful tool in drug discovery [20]. The derivatives of **3a–3j**, **4a** and **5a–5i** were prepared according to the synthetic Scheme 1. Compounds **3a–3j** were produced by the three-component one-pot reaction of dimedone (**1**), aromatic aldehydes (**2**), and 2-cyanothioacetamide in the presence of morpholine under ethanol conditions at reflux temperature, and then further replaced by iodoalkane. Compound **4a** was produced by the one-pot three-component reaction of dimedone (**1**), aromatic aldehydes (**2**), and ethyl propionate under acetic acid conditions at room temperature. The reaction conditions of the **5a–5i** were substantially the same as those of **3a–3j** except that 2-cyanothioacetamide was replaced with and 2-(nitromethylene) imidazolidine. The structures of synthesized **3a–3j**, **4** and **5a–5i** were characterized by melting point, ¹H-NMR, ¹³C-NMR and MS. The ¹H and ¹³C NMR spectra of representative compounds can be found in the supplementary materials. The obtained results showed consistency with the expected structures and formulas of the targeted products. According to the reaction route, the reaction could occur under the conditions of ethanol and acetic acid as a solvent, in line with the importance of economical and green transformations in green chemistry.



Scheme 1. General synthetic procedure for target compounds **3a–3j**, **4** and **5a–5i**. Reagents and conditions: (a) Et₃N, EtOH, reflux; (b) CH₃I or CH₂ICO₂C₂H₅, EtOH, reflux; (c) HOAc, rt; (d) Et₃N, EtOH, reflux.

2.2. Evaluation of Insecticidal Activity

Based on the methodology in Scheme 1, with twenty derivatives **3a–3j**, **4a** and **5a–5i** in hand, we examined their acaricidal activities against *T. cinnabarinus* and the results were summarized in Table 1. Next, we selected the five most active compounds (**3c**, **3d**, **3i**, **5c** and **5e**) to evaluate their insecticidal activities against *M. persicae* and *B. brassicae* and the results were shown in Tables 2 and 3. IMI were tested under the same conditions as a comparison compound.

As indicated in Table 1, all new compounds exhibited potent acaricidal activity against *T. cinnabarinus*, with LC₅₀ values ranging from 0.011 to 0.523 mM. Almost all of the derivatives showed significant acaricidal activities against *T. cinnabarinus*, and even half of the compounds were more than that of imidacloprid (0.135 mM). In particular, compounds **3c**, **3d**, **3i**, **5c** and **5e** showed pronounced acaricidal activities with respective LC₅₀ values of 0.057, 0.011, 0.033, 0.025 and 0.064 mM, respectively, higher than that of imidacloprid (IMI) (0.135 mM). The activities of these compounds in

Table 1 varied drastically, depending upon the types and patterns of substitution on the phenyl ring and dihydropyridinecore.

For the effect of substituents at phenyl ring in the series of **3a–3i** compounds, it was observed that compounds with electron-donating is favorable for high activities from data analysis present, the introduction of electron-donating groups, like **3d** (0.011 mM), **3e** (0.282 mM), **3f** (0.122 mM), **3g** (0.076 mM) and **3i** (0.033 mM), resulted in higher acaricidal potency than the corresponding analog with electron-withdrawing group, such as **3a** (0.473 mM) and **3b** (0.523 mM). Unexpectedly, the activity was found to increase rapidly when chlorine group was simultaneously introduced at the 2,4-position of the phenyl ring (**3c**, 0.057 mM), and its activity was nearly 10 times higher than that of the **3b** (0.523 mM), which introduced the chlorine group only at the 4-position the phenyl ring. Furthermore, the insecticidal activity of compound **3j** (0.079 mM) with an ethyl acetate group quadrupled compared to **3e** (0.282 mM) containing a methylthio group. In order to explore the effect of ester group on acaricidal activity, we replaced cyano group (**3a**, 0.473 mM) with ethyl formate (**4a**, 0.204 mM) and found that the activity was doubled. Of course, this may be due to the replacement of thiomethyl with ethyl. To identify more potent acaricidal activity, the effect of different phenyl groups on the insecticidal activity of the **5a–5j** derivatives was investigated. However, the structure activity relationship of **5a–5i** derivatives was completely different from that of **3a–3j**. Remarkably, compounds **5a**, **5c** and **5e** exhibited significant acaricidal activity against *T. cinnabarinus*, with LC₅₀ values of 0.096, 0.025 and 0.064 mM, respectively. This indicated that the change in activity of the compound **5** series depended not only on the type of electronic effect of the substituent group on the benzene ring, but also on the number and position of the substituents. By summarizing the structure activity relationship of all compounds, the results clearly underlined that the acaricidal difference could be ascribed to combination of factors, like nature of the substitutes (which may depend on electronic characteristics of substitutes, the position of substitutes, and other factors) or by a different interaction at the site.

Table 1. Insecticidal activities of compounds **3a–3j**, **4a** and **5a–5i** and imidacloprid (IMI) against *T. Cinnabarinus*.

Compd.	Mortality (%) ^{a,b}					
	250 mg/L	100 mg/L	50 mg/L	10 mg/L	1 mg/L	LC ₅₀ (mM) ^c
3a	63	37	30	23	10	0.473
3b	63	37	23	13	7	0.523
3c	100	60	53	37	17	0.057
3d	100	97	93	57	30	0.011
3e	60	50	40	27	10	0.282
3f	90	58	45	21	8	0.122
3g	97	87	43	17	10	0.076
3h	90	50	33	23	17	0.159
3i	97	70	63	50	20	0.033
3j	93	80	37	27	7	0.079
4a	90	40	27	17	7	0.204
5a	87	60	47	30	13	0.096
5b	90	53	43	20	13	0.132
5c	100	67	57	43	33	0.025
5d	80	43	40	23	20	0.191
5e	97	80	60	20	12	0.064
5f	80	57	53	30	10	0.107
5g	70	30	20	17	3	0.404
5h	100	33	23	20	13	0.184
5i	90	47	30	23	10	0.173
IMI	90	60	33	30	27	0.135

^a Temperature: 25 ± 2 °C; room humidity (RH): 65–80%; photoperiod: light/dark = 12/12 h. ^b Experimental size: 10 insects per group, three groups. ^c LC₅₀ calculations were determined by Probit analysis using a maximum quasi-likelihood curve fitting algorithm.

Subsequently, **3c**, **3d**, **3i**, **5c** and **5e** with the most acaricidal activities against *T. cinnabarinus* among of tested compounds, were selected to evaluate the insecticidal activity against *M. persicae* and *B. brassicae*. As to the insecticidal activities, from Tables 2 and 3, it was surprising that all of the target compounds showed strong insecticidal activities against *M. persicae* and *B. oleracea*, and the activity of **3i** (0.0007 and 0.0025 mM) was even better than or equal to that of imidacloprid (0.0010 and 0.0006 mM). This result indicated that the series of compounds not only have good acaricidal activities, but also have extremely strong aphicidal activities. These encouraging results would prompt us to study the dihydropyridine derivatives as insecticidal agent in future.

Table 2. Insecticidal activities of compounds **3c**, **3d**, **3i**, **5c**, **5e** and imidacloprid (IMI) against *M. persicae*.

Compd.	Mortality (%) ^{a,b}								
	24 h				48 h				LC ₅₀ (mM) ^c
	50 mg/L	10 mg/L	1 mg/L	0.1 mg/L	50 mg/L	10 mg/L	1 mg/L	0.1 mg/L	
3c	63	39	30	17	87	43	37	30	0.0077
3d	53	27	17	13	93	63	40	33	0.0033
3i	57	30	20	10	90	83	70	37	0.0007
5c	40	33	20	10	70	50	40	27	0.0111
5e	40	33	20	7	87	57	43	23	0.0053
IMI	40	23	6	3	100	87	70	37	0.0010

^a Temperature: 25 ± 2 °C; RH: 65–80%; photoperiod: light/dark = 12/12 h. ^b Experimental size: 10 insects per group, three groups. ^c LC₅₀ calculations were determined by Probit analysis using a maximum quasi-likelihood curve fitting algorithm.

Table 3. Insecticidal activities of compounds **3c**, **3d**, **3i**, **5c**, **5e** and imidacloprid (IMI) against *B. brassicae*.

Compd.	Mortality (%) ^{a,b}								
	24 h				48 h				LC ₅₀ (mM) ^c
	50 mg/L	10 mg/L	1 mg/L	0.1 mg/L	50 mg/L	10 mg/L	1 mg/L	0.1 mg/L	
3c	37	30	20	13	70	63	40	20	0.0087
3d	67	40	33	17	90	70	60	33	0.0015
3i	60	43	30	17	83	67	57	27	0.0025
5c	30	20	13	7	60	50	33	13	0.0291
5e	53	37	23	13	67	57	37	23	0.0125
IMI	93	77	43	10	97	90	77	43	0.0006

^a Temperature: 25 ± 2 °C; RH: 65–80%; photoperiod: light/dark = 12/12 h. ^b Experimental size: 10 insects per group, three groups. ^c LC₅₀ calculations were determined by Probit analysis using a maximum quasi-likelihood curve fitting algorithm.

2.3. Evaluation of Antifungal Activity

Plant fungal diseases are increasingly becoming a food security threat, and fungicides are widely used to control the development of phytopathogenic fungi. Inspired by the excellent insecticidal activities of this series of compounds, all compounds were evaluated for fungicidal activities against phytopathogenic fungi and it would inspire us to find a wider range of biologically active uses. Surprisingly, as shown in Table 4, although compound **4a** exhibited moderate insecticidal activity, it showed significant antifungal activity against *F. oxysporum* and *S. sclerotiorum* in vitro, and the inhibition rate reached 45.00% and 65.85% at 50 mg/L, respectively. The conversion of cyano group (**3a**) to ester groups (**4a**) could significantly increase the activities against phytopathogenic fungi, which provides a reference for our search to discovery the promising candidates with insecticidal and antifungal activities.

Table 4. Antifungal activities of compounds **3a–j**, **4**, **5a–i** and azoxystrobin against *F. oxysporum*, *M. oryzae*, *S. sclerotiorum*, *B. cinerea* at 50 mg/L.

Compd.	Inhibition (%)			
	<i>F. oxysporum</i>	<i>M. oryzae</i>	<i>S. sclerotiorum</i>	<i>B. cinerea</i>
3a	5.00	8.75	0.00	6.25
3b	23.33	12.50	7.08	5.00
3c	17.92	4.17	24.17	11.25
3d	5.42	11.67	0.00	0.00
3e	0.00	3.75	0.00	0.00
3f	11.67	11.25	6.25	0.00
3g	7.87	10.56	7.32	0.00
3h	0.00	0.00	8.33	0.00
3i	12.45	4.56	7.52	6.27
3j	5.42	10.83	10.83	6.25
4	45.00	21.67	65.83	18.33
5a	0.00	6.45	0.00	7.78
5b	5.42	0.00	0.00	0.00
5c	5.83	0.00	0.00	32.50
5d	0.00	9.58	11.67	0.00
5e	0.00	7.56	9.24	0.00
5f	2.92	16.25	6.25	0.00
5g	0.00	5.83	6.25	0.00
5h	0.00	9.58	10.83	0.00
5i	7.08	5.42	0.00	0.00
azoxystrobin	52.50	88.48	74.58	42.08

3. Experimental Section

3.1. Chemicals and Instruments

All reactions were performed with commercially available reagents without further purification. All reactions were monitored by thin-layer chromatography (TLC) and preparative thin-layer chromatography (PTLC) were performed with silica gel plates using silica gel 60 GF254 (Qingdao Haiyang Chemical Co., Ltd., Qingdao, China). Melting points were determined in an open capillary using WRS-2U melting point apparatus (Shanghai Precision Instrument Co., Ltd., Shanghai, China) and are uncorrected. Mass spectra were recorded on a Bruker Daltonics APEXII49e spectrometer (Bruker Daltonics Inc., Billerica, MA, US.) with ESI source as ionization. ¹H and ¹³C-NMR spectra were recorded at 400 and 100 MHz on a Bruker AM-400 (Bruker Company, Billerica, MA, US.) spectrometer using TMS as reference. The commercial insecticide imidacloprid and fungicides azoxystrobin (analytical grade, 98% purity) (Jiangsu Bailing Agrochemical Co., Ltd., Jiangying, China) was used as a positive control in vitro experiment. *Tetranychus Cinnabarinus*, *Myzus persicae*, *Brevicoryne brassicae*, *Fusarium oxysporum* f. sp. *vasinfectum*, *Magnaporthe oryzae*, *Sclerotinia sclerotiorum* and *Botrytis cinerea* were obtained from the Institute of Plant Protection, Gansu Academy of Agricultural Science, Lanzhou, China.

3.2. Synthesis

3.2.1. General Synthetic Procedure for Target Compounds **3a–3j**

A 0.5 L round-bottom flask fitted with an overhead stirrer was charged with the corresponding aromatic aldehyde (0.1 mol), cyanothioacetamide **10** (10.0 g, 0.10 mol) and EtOH (100 mL). Triethylamine (0.8–1.0 mL) was added, and the mixture was stirred for 1 h at 20 °C (yellow/orange crystalline may precipitate from the solution). Then dimedone (15.0 g, 0.104 mol) and *N*-methylmorpholine (16.5 mL, 0.15 mol) were added, and the solution was refluxed for 2–4 h. The mixture of salt was added

iodomethane (0.1 mol) or ethyl iodoacetate (0.1 mol) in 80% ethanol (15 mL) and boiled for 2 min and then filtered through paper (Scheme 1). The solid was recrystallized from EtOH to afford the pure products [21].

3a: Yellow solid; yield 56%; m.p. 192–194 °C; $^1\text{H-NMR}$ (400 MHz, DMSO- d_6) δ : 9.75 (s, 1H, br, NH), 8.19 (d, 2H, $J = 8$ Hz, Ar-H), 7.44 (d, 2H, $J = 8$ Hz, Ar-H), 4.76 (s, 1H, CH), 2.54 (s, 3H, -SCH₃), 2.42 (d, 2H, $J = 16$ Hz, CH₂), 2.21 (d, 1H, $J = 16$ Hz, CH), 2.09–1.99 (m, 1H, CH), 1.03 (s, 3H, -CH₃), 0.90 (s, 3H, -CH₃). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6) δ 194.1, 149.5, 148.3, 144.9, 144.4, 127.0, 127.0, 123.5, 123.5, 118.6, 111.9, 104.5, 51.2, 39.6, 32.8, 31.3, 29.7, 26.7, 17.1. HRMS: calcd. for C₁₉H₁₉N₃O₃S [2M + Na]: 761.2192, found: 761.2195.

3b: Yellow solid; yield 53%; m.p. 201–203 °C; $^1\text{H-NMR}$ (400 MHz, DMSO- d_6) δ : 9.66 (s, 1H, br, NH), 7.36 (d, 2H, $J = 8$ Hz, Ar-H), 7.16 (d, 2H, $J = 8$ Hz, Ar-H), 4.49 (s, 1H, -CH), 2.52 (s, 3H, SCH₃), 2.43–2.46 (m, 2H, CH₂), 2.20 (d, 1H, $J = 16$ Hz, CH), 2.01 (t, 1H, $J = 16$ Hz, CH), 1.02 (s, 3H, CH₃), 0.89 (s, 3H, CH₃). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6) δ 194.6, 152.1, 145.5, 140.2, 131.3, 131.4, 131.4, 128.8, 128.8, 118.6, 111.9, 104.5, 51.2, 39.6, 32.8, 31.3, 29.7, 26.7, 17.1. HRMS: calcd. for C₁₉H₁₉ClN₂OS [2M + Na]: 739.1711, found: 739.1720.

3c: Yellow solid; yield 48%; m.p. 243–245 °C; $^1\text{H-NMR}$ (400 MHz, DMSO- d_6) δ : 9.67 (s, 1H, br, NH), 7.54 (s, 1H, Ar-H), 7.39 (dd, 1H, $J = 8$ Hz, $J = 4$ Hz, Ar-H), 7.24 (d, 1H, $J = 8$ Hz, Ar-H), 4.99 (s, 1H, CH), 2.50 (s, 3H, -SCH₃), 2.44–2.40 (m, 2H, CH₂), 2.18 (d, 1H, $J = 16$ Hz, CH), 1.97–2.09 (m, 1H, CH), 1.03 (s, 3H, CH₃), 0.93 (s, 3H, -CH₃). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6) δ 194.6, 152.1, 145.5, 140.2, 131.3, 131.4, 131.4, 129.5, 127.5, 118.6, 111.9, 104.5, 51.2, 39.6, 32.8, 31.3, 29.7, 26.7, 17.1. MS-ESI m/z : 807.0 [2M+Na] HRMS: calcd. for C₁₉H₁₉Cl₂N₂O₃S [2M + Na]: 807.0931, found: 807.0935.

3d: Yellow solid; yield 52%; m.p. 189–191 °C; $^1\text{H-NMR}$ (400 MHz, DMSO- d_6) δ : 9.59 (s, 1H, br, NH), 7.05 (d, 2H, $J = 8$ Hz, Ar-H), 6.85 (d, 2H, $J = 8$ Hz, Ar-H), 4.40 (s, 1H, CH), 3.71 (s, 3H, -OCH₃), 2.50 (s, 3H, -SCH₃), 2.38–2.45 (m, 2H, CH₂), 2.19 (d, 1H, $J = 16$ Hz, CH), 1.99–2.09 (m, 1H, CH), 1.02 (s, 3H, -CH₃), 0.90 (s, 3H, -CH₃). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6) δ 194.5, 157.8, 148.5, 146.2, 136.2, 133.3, 133.3, 118.6, 118.4, 118.4, 111.9, 104.5, 59.2, 52.7, 39.6, 32.8, 31.3, 29.7, 26.7, 17.1. MS-ESI m/z : 731.1 [2M + Na] HRMS: calcd. for C₂₀H₂₂N₂O₂S [2M + Na]: 731.2702, found: 731.2710.

3e: Yellow solid; yield 55%; m.p. 197–199 °C; $^1\text{H-NMR}$ (400 MHz, DMSO- d_6) δ : 9.63 (s, 1H, br, NH), 7.18 (d, 1H, $J = 12$ Hz, Ar-H), 7.08 (t, 2H, $J = 8$ Hz, Ar-H), 4.42 (s, 1H, CH), 2.44 (m, 5H, CH₂, SCH₃), 2.50 (s, 3H, SCH₃), 2.20 (d, 1H, $J = 16$ Hz, CH), 2.09–2.00 (m, 1H, CH), 1.03 (s, 3H, CH₃), 0.90 (s, 3H, CH₃). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6) δ 194.5, 148.5, 146.2, 142.8, 140.3, 130.3, 130.3, 128.9, 128.9, 118.6, 111.9, 104.5, 52.7, 39.6, 32.8, 31.3, 29.7, 26.7, 17.1, 14.8. MS-ESI m/z : 763.0 [2M + Na] HRMS: calcd. for C₂₀H₂₂N₂O₅S [2M + Na]: 763.2245, found: 763.2251.

3f: Yellow solid; yield 48%; m.p. 195–199 °C; $^1\text{H-NMR}$ (400 MHz, DMSO- d_6) δ : 9.54 (s, 1H, br, NH), 6.89 (d, 1H, $J = 8$ Hz, Ar-H), 6.73 (dd, 1H, $J = 8$ Hz, $J = 4$ Hz, Ar-H), 6.52 (d, 1H, $J = 4$ Hz, Ar-H), 4.80 (s, 1H, CH), 3.73 (s, 3H, OCH₃), 3.64 (s, 3H, OCH₃), 2.46 (s, 3H, -SCH₃), 1.04 (s, 3H, CH₃), 2.44–2.40 (m, 2H, CH₂), 2.19 (d, 1H, $J = 16$ Hz, CH), 2.09–1.98 (m, 1H, CH), 0.96 (s, 3H, CH₃). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6) δ 194.5, 152.8, 151.2, 150.5, 146.2, 123.3, 118.6, 116.5, 112.8, 111.9, 106.5, 56.6, 55.9, 52.7, 39.6, 32.8, 29.7, 26.7, 25.3 17.1. MS-ESI m/z : 792.3 [2M + Na] HRMS: calcd. for C₂₁H₂₄N₂O₃S [2M + Na]: 791.2913, found: 791.2918.

3g: Light yellow solid; yield 55%; m.p. 221–223 °C; $^1\text{H-NMR}$ (400 MHz, DMSO- d_6) δ : 9.66 (s, 1H, br, NH), 6.40 (s, 1H, Ar-H), 4.42 (s, 1H, CH), 3.71 (s, 6H, OCH₃), 3.67 (s, 3H, OCH₃), 2.50 (s, 3H, SCH₃) 2.45–2.41 (m, 2H, CH₂), 2.23 (d, 1H, $J = 16$ Hz, CH), 2.09–1.99 (m, 1H, CH), 1.05 (s, 3H, CH₃), 0.98 (s, 3H, CH₃). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6) δ 194.5, 152.8, 152.8, 150.5, 138.5, 137.2, 118.6, 111.9, 106.5, 106.5, 104.5, 61.6, 56.8, 56.8, 52.7, 39.6, 32.8, 31.9, 29.7, 26.7, 25.3 17.1. HRMS: calcd. for C₂₂H₂₆N₂O₄S [2M + Na]: 851.3124, found: 851.3129.

3h: Light yellow solid; yield 58%; m.p. 201–203 °C; ¹H-NMR (400 MHz, DMSO-*d*₆) δ: 9.77 (s, 1H, br, NH), 7.18 (d, 1H, *J* = 12 Hz, Ar-H), 7.08 (t, 2H, *J* = 8 Hz, Ar-H), 4.43 (s, 1H, CH), 4.04 (q, 2H, *J* = 8 Hz, CH₂), 3.84 (dd, 2H, *J* = 16 Hz, *J* = 44 Hz, CH₂), 2.37–2.41 (m, 5H, CH₂, SCH₃), 2.22–2.18 (d, 1H, *J* = 16 Hz, CH), 2.00–2.09 (m, 1H, CH), 1.13 (t, 3H, *J* = 4 Hz, CH₃), 1.02 (s, 3H, CH₃), 0.91 (s, 3H, CH₃). ¹³C-NMR (100 MHz, DMSO-*d*₆) δ 194.5, 150.5, 146.2, 140.2, 137.7, 134.2, 131.2, 130.8, 127.6, 118.6, 111.9, 106.5, 54.6, 39.6, 32.8, 31.6, 29.7, 26.7, 19.1, 18.8, 17.1. HRMS: calcd. for C₂₁H₂₄N₂O₅ [2M + Na]: 727.3116, found: 727.3120.

3i: Light yellow solid; yield 51%; m.p. 195–197 °C; ¹H-NMR (400 MHz, DMSO-*d*₆) δ: 9.60 (s, 1H, br, NH), 6.82 (d, 1H, *J* = 8 Hz, Ar-H), 6.06 (t, 2H, *J* = 8 Hz, Ar-H), 5.07 (s, 2H, CH₂), 4.39 (s, 1H, CH), 2.50 (s, 3H, -SCH₃), 2.44 (d, 2H, *J* = 4 Hz, CH₂), 2.19 (d, 1H, *J* = 16 Hz, CH), 2.02–2.09 (m, 1H, CH) 1.02 (s, 3H, CH₃), 0.91 (s, 3H, CH₃). ¹³C-NMR (100 MHz, DMSO-*d*₆) δ 194.5, 150.5, 148.7, 145.8, 144.4, 135.5, 122.6, 118.6, 115.9, 112.3, 111.9, 106.5, 101.5, 54.6, 39.6, 32.8, 31.6, 29.7, 26.7, 17.1. HRMS: calcd. for C₂₀H₂₀N₂O₃S [2M + Na]: 759.2287, found: 759.2290.

3j: Yellow solid; yield 57%; m.p. 213–215 °C; ¹H-NMR (400 MHz, DMSO-*d*₆) δ: 9.57 (s, 1H, br, NH), 7.04 (d, 1H, *J* = 8 Hz, Ar-H), 6.84 (m, 2H, Ar-H), 4.36 (s, 1H, CH), 2.50 (s, 3H, SCH₃), 2.43–2.46 (m, 2H, CH₂), 2.17–2.15 (m, 6H, CH₃), 2.09–1.99 (m, 2H, CH₂), 1.03 (s, 3H, CH₃), 0.91 (s, 3H, CH₃). ¹³C-NMR (100 MHz, DMSO-*d*₆) δ 194.5, 169.9, 150.5, 145.8, 139.8, 139.2, 130.5, 130.5, 128.6, 128.6, 118.6, 111.9, 106.5, 62.8, 54.6, 39.6, 34.2, 32.8, 31.6, 29.7, 26.7, 14.8, 14.1. HRMS: calcd. for C₂₃H₂₆N₂O₃S₂ [2M + Na]: 907.2667, found: 907.2672.

3.2.2. General Synthetic Procedure for Target Compounds 4

In a dry 50 mL flask, arylaldehyde (1 mmol), ethyl propionylacetate (1 mmol), dimedone (1 mmol) and excessive ammonium acetate and HOAc (10 mL) were mixed and then stirred at room temperature for 8–10 h. After completion of the reaction, as indicated by thin layer chromatography (TLC), the reaction mixture was poured into water, then the solid product was collected and purified by flash column chromatography (Scheme 1) [22].

4: Yellow solid; yield 56%; m.p. 223–225 °C; ¹H-NMR (400 MHz, DMSO-*d*₆) δ: 9.22 (s, 1H, br, NH), 8.09 (d, 2H, *J* = 8 Hz, Ar-H), 7.40 (d, 2H, *J* = 8 Hz, Ar-H), 4.97 (s, 1H, CH), 3.94 (q, 2H, *J* = 8 Hz, CH₂), 2.76–2.67 (m, 2H, CH₂), 2.32–2.18 (m, 2H, CH₂), 2.09–1.95 (m, 2H, CH₂), 1.13–1.09 (m, 6H, CH₃), 1.01 (s, 3H, CH₃), 0.80 (s, 3H, CH₃). ¹³C-NMR (100 MHz, DMSO-*d*₆) δ 194.5, 169.9, 152.6, 150.5, 149.5, 145.8, 127.8, 127.8, 124.6, 124.6, 111.9, 102.5, 62.8, 54.6, 42.4, 40.3, 32.8, 29.7, 26.7, 23.8, 14.2, 12.9. HRMS: calcd. for C₂₂H₂₆N₂O₅ [2M + Na]: 819.3581, found: 819.3583.

3.2.3. General Synthetic Procedure for Target Compounds 5a–5i

2-(nitromethylene) imidazolidine (0.5 mmol), aldehydes (0.5 mmol), dimedone (0.5 mmol) and EtOH (10 mL) and Et₃N (0.25 mmol) were added into a 25 mL flask and the mixture was stirred for the appropriate reaction time at 80 °C in an oil bath until the 2-(nitromethylene) imidazolidine was completely consumed. The solid mixture was washed with EtOH (2 × 5 mL). The crude residue was recrystallized from EtOH to afford the pure products (Scheme 1) [23].

5a: Dark yellow solid; yield 56%; m.p. 264–267 °C; ¹H-NMR (400 MHz, DMSO-*d*₆) δ 9.52 (s, 1H, br, NH), 8.14–7.97 (m, 2H, Ar-H), 7.63–7.42 (m, 2H, Ar-H), 5.16 (s, 1H, CH), 4.26–4.14 (m, 1H, CH), 4.05 (q, *J* = 9.5 Hz, 1H, CH), 3.91–3.77 (m, 2H, CH₂), 2.62 (d, *J* = 17.7 Hz, 1H, CH₂), 2.55 (s, 1H, CH₂), 2.21 (d, *J* = 16.1 Hz, 1H, CH₂), 2.00 (d, *J* = 16.1 Hz, 1H, CH₂), 1.05 (s, 3H, CH₃), 0.84 (s, 3H, CH₃). ¹³C-NMR (100 MHz, DMSO-*d*₆) δ 194.7, 160.8, 148.3, 146.5, 144.9, 134.2, 131.5, 128.89, 126.7, 111.9, 94.0, 52.2, 50.5, 39.9, 40.9, 33.1, 30.1, 29.7, 26.7. HRMS: calcd. for C₁₉H₂₀N₄O₅ [M + H]⁺: 385.1512, found: 385.1518.

5b: Yellow solid; yield 56%; m.p. 295–296 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.45 (s, 1H, br, NH), 7.34–7.14 (m, 4H, Ar-H), 5.50 (s, 1H, CH), 4.17 (dd, $J = 10.0, 6.7$ Hz, 1H, CH), 4.03 (q, $J = 9.5$ Hz, 1H, CH_2), 3.87–3.78 (m, 2H, CH_2), 2.65 (d, $J = 20.0$ Hz, 1H, CH_2), 2.58 (s, 1H, CH_2), 2.20 (d, $J = 16.1$ Hz, 1H, CH_2), 2.00 (d, $J = 16.3$ Hz, 1H, CH_2), 1.05 (s, 3H, CH_3), 0.85 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 193.7, 151.9, 150.0, 140.5, 134.3, 134.2, 131.5, 128.89, 126.7, 112.1, 106.7, 45.2, 43.9, 39.9, 38.8, 37.4, 32.1, 29.7, 26.7. HRMS: calcd. for $\text{C}_{19}\text{H}_{20}\text{ClN}_3\text{O}_3$ $[\text{M} + \text{Na}]^+$: 396.1091, found: 396.1095.

5c: Yellow solid; yield 56%; m.p. 284–287 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.47 (s, 1H, br, NH), 7.48–7.33 (m, 2H, Ar-H), 7.26 (dd, $J = 8.4, 2.2$ Hz, 1H, Ar-H), 5.28 (s, 1H, CH), 4.19 (t, $J = 9.0$ Hz, 1H, CH), 4.08 (s, 1H, CH_2), 3.84 (s, 2H, CH_2), 2.60 (d, $J = 17.6$ Hz, 1H, CH_2), 2.46 (s, 1H, CH_2), 2.18 (d, $J = 16.1$ Hz, 1H, CH_2), 1.98 (s, 1H, CH_2), 1.05 (s, 3H, CH_3), 0.87 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 193.7, 151.9, 150.0, 140.5, 134.3, 134.2, 131.5, 128.89, 126.7, 112.1, 106.7, 45.2, 43.9, 39.9, 38.8, 37.4, 32.1, 29.7, 26.7. HRMS: calcd. for $\text{C}_{19}\text{H}_{19}\text{Cl}_2\text{N}_3\text{O}_3$ $[\text{M} + \text{Na}]^+$: 430.0701, found: 430.0731.

5d: Yellow solid; yield 56%; m.p. 310–311 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.38 (s, 1H, br, NH), 7.22–7.05 (m, 2H, Ar-H), 6.86–6.63 (m, 2H, Ar-H), 5.02 (s, 1H, CH), 4.17 (s, 1H, CH_2), 4.02 (d, $J = 9.6$ Hz, 1H, CH_2), 3.83 (t, $J = 9.9$ Hz, 2H, CH_2), 3.68 (s, 3H, CH_3), 2.65 (d, $J = 21.3$ Hz, 1H, CH_2), 2.58 (s, 1H, CH_2), 2.19 (d, $J = 16.1$ Hz, 1H, CH_2), 1.99 (d, $J = 16.1$ Hz, 1H, CH_2), 1.05 (s, 3H, CH_3), 0.86 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 198.5, 162.7, 156.8, 153.9, 141.7, 134.2, 134.0, 119.1, 119.1, 118.2, 94.0, 60.1, 54.7, 50.0, 48.6, 43.4, 41.4, 37.0, 34.6, 31.4. HRMS: calcd. for $\text{C}_{20}\text{H}_{23}\text{N}_3\text{O}_4$ $[\text{M} + \text{H}]^+$: 370.1767, found: 370.1771.

5e: Yellow solid; yield 56%; m.p. 279–281 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.42 (s, 1H, br, NH), 7.19–7.13 (m, 2H, Ar-H), 7.09 (d, $J = 8.3$ Hz, 2H, Ar-H), 5.03 (s, 1H, CH), 4.19 (q, $J = 8.7$ Hz, 1H, CH_2), 4.03 (q, $J = 9.6$ Hz, 1H, CH_2), 3.84 (d, $J = 9.9$ Hz, 2H, CH_2), 2.58 (d, $J = 17.0$ Hz, 1H, CH_2), 2.53 (s, 1H, CH_2), 2.41 (s, 3H, CH_3), 2.20 (d, $J = 16.0$ Hz, 1H, CH_2), 2.00 (d, $J = 16.1$ Hz, 1H, CH_2), 1.05 (s, 3H, CH_3), 0.86 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 193.7, 152.1, 149.7, 141.6, 135.8, 128.9, 128.9, 126.0, 126.0, 114.0, 107.7, 49.9, 45.3, 43.9, 38.7, 37.2, 32.3, 29.8, 26.6, 15.3. HRMS: calcd. for $\text{C}_{20}\text{H}_{24}\text{N}_3\text{O}_3\text{S}$ $[\text{M} + \text{H}]^+$: 386.1538, found: 386.1540.

5f: Yellow solid; yield 56%; m.p. 278–279 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.42 (s, 1H, br, NH), 6.97 (d, $J = 5.2$ Hz, 3H, Ar-H), 5.02 (s, 1H, CH), 4.25–4.16 (m, 1H, CH_2), 4.02 (q, $J = 9.6$ Hz, 1H, CH_2), 3.83 (t, $J = 8.6$ Hz, 2H, CH_2), 3.77 (s, 3H, CH_3), 2.60 (d, $J = 17.7$ Hz, 1H, CH_2), 2.54 (s, 1H, CH_2), 2.18 (s, 1H, CH_2), 2.03 (s, 1H, CH_2), 1.05 (s, 3H, CH_3), 0.87 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 193.8, 152.0, 149.5, 145.7, 124.3, 124.2, 115.7, 115.5, 113.7, 113.4, 107.5, 56.4, 45.3, 43.9, 40.5, 38.7, 36.9, 32.3, 29.8, 26.6. HRMS: calcd. for $\text{C}_{20}\text{H}_{22}\text{FN}_3\text{O}_4$ $[\text{M} + \text{H}]^+$: 388.1673, found: 388.1675.

5g: Yellow solid; yield 56%; m.p. 279–280 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.47 (s, 1H, br, NH), 7.62 (d, $J = 2.3$ Hz, 1H, Ar-H), 7.51 (dd, $J = 8.7, 2.3$ Hz, 1H, Ar-H), 7.19 (d, $J = 8.7$ Hz, 1H, Ar-H), 5.04 (s, 1H, CH), 4.19 (td, $J = 9.5, 6.9$ Hz, 1H, CH), 4.08–3.94 (m, 1H, CH_2), 3.85 (d, $J = 10.8$ Hz, 5H, CH_3, CH_2), 2.61 (d, $J = 17.7$ Hz, 1H, CH_2), 2.55 (s, 1H, CH_2), 2.21 (d, $J = 16.1$ Hz, 2H, CH_2), 2.01 (d, $J = 16.1$ Hz, 2H, CH_2), 1.06 (s, 3H, CH_3), 0.88 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 193.8, 151.8, 150.8, 150.0, 139.1, 137.3, 134.4, 124.1, 113.9, 113.0, 107.2, 57.0, 49.8, 45.3, 43.9, 38.7, 37.2, 32.3, 29.8, 26.6. HRMS: calcd. for $\text{C}_{20}\text{H}_{22}\text{N}_4\text{O}_6$ $[\text{M} + \text{H}]^+$: 415.1618, found: 415.1621.

5h: Yellow solid; yield 56%; m.p. 240–242 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.40 (s, 1H, br, NH), 6.46 (s, 2H, Ar-H), 5.09 (s, 1H, CH), 4.28–4.13 (m, 1H, CH), 4.02 (q, $J = 9.6$ Hz, 1H, CH), 3.83 (dd, $J = 10.2, 7.2$ Hz, 2H, CH_2), 3.70 (s, 6H, CH_3, CH_3), 3.60 (s, 3H, CH_3), 2.63 (d, $J = 17.7$ Hz, 1H, CH_2), 2.58 (s, 1H, CH_2), 2.23 (d, $J = 16.2$ Hz, 1H, CH_2), 2.05 (d, $J = 16.1$ Hz, 1H, CH_2), 1.07 (s, 3H, CH_3), 0.94 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 198.6, 157.5, 156.8, 154.6, 157.5, 144.9, 141.3, 118.6, 112.3, 110.4, 110.4, 94.0, 65.1, 61.0, 61.0, 54.7, 48.6, 43.5, 42.0, 37.0, 34.8, 31.2. HRMS: calcd. for $\text{C}_{22}\text{H}_{27}\text{N}_3\text{O}_6$ $[\text{M} + \text{H}]^+$: 430.1978, found: 430.1985.

5i: Yellow solid; yield 56%; m.p. 345–356 °C; $^1\text{H-NMR}$ (400 MHz, $\text{DMSO-}d_6$) δ 9.13 (s, 1H, br, NH), 6.74–6.65 (m, 3H, Ar-H), 5.91 (d, $J = 3.8$ Hz, 2H, CH_2), 4.25 (d, $J = 10.8$ Hz, 1H, CH), 3.75–3.57 (m, 4H, CH_2), 2.59 (d, $J = 17.4$ Hz, 1H, CH_2), 2.44 (s, 1H, CH_2), 2.11 (d, 1H, CH_2), 1.92 (d, $J = 14.2$ Hz, 1H, CH_2), 1.04 (s, 3H, CH_3), 1.01 (s, 3H, CH_3). $^{13}\text{C-NMR}$ (100 MHz, $\text{DMSO-}d_6$) δ 204.8, 157.6, 146.7, 145.3, 139.1, 135.5, 121.4, 109.4, 108.6, 107.8, 100.9, 85.6, 54.3, 46.3, 43.3, 36.6, 35.3, 33.3, 32.3, 28.1. HRMS: calcd. for $\text{C}_{20}\text{H}_{23}\text{N}_3\text{O}_6$ $[\text{M} + \text{H}]^+$: 384.1559, found: 384.1565.

3.3. Insecticidal Activity

Bioassays on representative test organisms reared in the laboratory were carried out at 25 ± 2 °C apropos to statistical standards. Series concentrations of 250.0, 100.0, 50.0, 10.0, 1.0 and 0.1 mg/L for bioassays were obtained by dissolving all the synthesized dihydropyridine analogues in acetone and diluted with water containing Tween-20 (0.1 mg/L). The control imidacloprid was tested under the same experimental conditions.

3.3.1. Acaricidal Assay against *T. Cinnabarinus*.

Slide immersion method recommended by FAO [13] was employed to evaluate the acaricidal activity of all the synthesized agents. All the test compounds were prepared in acetone at a concentration of 250 mg/L and diluted to the required concentration with distilled water containing TW-80. Using a small brush, thirty adult spider mites were fixed dorsally to a strip of double-sided tape attached to the slide. The slide was immersed diluted solution of the test compounds and shaken for 3 s. The treated slides with the mites were kept at 25 ± 2 °C in a covered dish with wet filter paper after the excessive solution was removed. After 24 h treatment, the number of demised mites was recorded. Each treatment was repeated with triplicate experiments and each replicate involved 30 adult mites. Control groups were tested with only acetone.

3.3.2. Insecticidal Assay Against *M. Persicae* and *B. Brassicae*.

The insecticidal activities of five compounds **3c**, **3d**, **3i**, **5c**, **5e** and imidacloprid against *M. persicae* and *B. brassicae* were evaluated according to the reported procedure [24].

3.4. Antifungal Activity

The effects of **3a–j**, **4a**, **5a–i** and azoxystrobin on the mycelial growth against *F. oxysporum*, *M. oryzae*, *S. sclerotiorum* and *B. cinerea* were assessed using Poison Food Technique in solid media [25]. After completely covered the Petri dishes of the fungal, the mycelial growth diameters were measured and inhibition percentages relative to the control with DMSO were calculated using the formula from Agarwal: $I(\%) = [(C - d) - (T - d)] / [(C - d)] \times 100$, where d is diameter of the cut fungus (5 mm), I is the inhibition (%), and C and T are the average colony diameters of the mycelium of the control and treatment, respectively.

4. Conclusions

In summary, a series of novel dihydropyridine derivatives were designed, “green” synthesized via one pot facile three-component reaction and their structures were characterized by melting point, $^1\text{H-NMR}$, $^{13}\text{C-NMR}$ and EI-MS. The bioactivities were evaluated against *T. cinnabarinus*, *M. persicae*, *B. brassicae*, *F. oxysporum*, *M. oryzae*, *S. sclerotiorum* and *B. cinereal*. In particular, Compound **3d** (LC_{50} : 0.011 and 0.0015 mM) exhibited the strongest insecticidal activity against *T. cinnabarinus* and *B. brassicae* in all of the derivatives we prepared. Compound **3i** (LC_{50} : 0.0007 mM) exhibited the strongest insecticidal activity against *M. persicae*, and, surprisingly, when the concentration of compound **4a** was 50 mg/L, the inhibition rate against *F. oxysporum* and *S. sclerotiorum* reached 45.00% and 65.83%. SARs clearly indicated that variations of R groups in the position of benzene ring markedly affected the insecticidal activity. When cyano group was replaced by ethyl acetate at the 3-position of

dihydropyridine core, the biological activity spectrum was markedly affected. These results provide a reference for searching for neonicotinoid insecticides and agricultural fungicide candidates in the future.

Supplementary Materials: The representative compounds ^1H and ^{13}C NMR spectra are available online.

Author Contributions: G.-Z.Y. and P.-L.C. performed the chemical synthesis; X.-D.Y. and J.-K.Z. carried out the insecticidal and antifungal activity evaluation; X.-F.S. contributed to the practical aspects of the research work and Y.-Q.L. and J.Z. supervised the research and prepared the manuscript.

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Sample Availability: Samples of the compounds **3a–3j**, **4a** and **5a–5i** are available from the authors.



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