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catena-Poly[[*(2-aminopyrimidine-κN¹)-(thiocyanato-κS)mercury(II)-μ-thiocyanato-κ²S:N*]]

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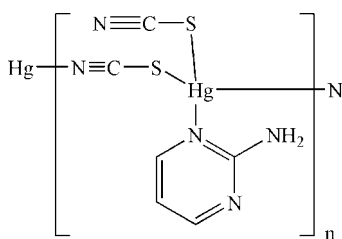
Received 8 April 2012; accepted 17 April 2012

Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.015$ Å; R factor = 0.049; wR factor = 0.088; data-to-parameter ratio = 16.8.

In the title coordination polymer, $[\text{Hg}(\text{NCS})_2(\text{C}_4\text{H}_5\text{N}_3)]_n$, the Hg^{II} atom is four-coordinated by one aromatic N atom from a 2-aminopyrimidine ligand, one S atom from a terminal thiocyanate ligand, and one S atom and one N atom from a bridging thiocyanate ligand. The crystal structure features polymeric chains running along the b axis which are stabilized by $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds.

Related literature

For related structures with aminopyridine as a ligand, see: Albada *et al.* (2002); Castillo *et al.* (2011); Cheng *et al.* (2009); Cui *et al.* (2011); Gao & Ng (2010); Lee *et al.* (2003); Li *et al.* (2006); Lin & Zeng (2007); Masaki *et al.* (2002); Qu *et al.* (2008); Zhu *et al.* (2002, 2003).



Experimental

Crystal data

 $[\text{Hg}(\text{NCS})_2(\text{C}_4\text{H}_5\text{N}_3)]_n$
 $M_r = 411.88$ Monoclinic, $C2/c$ $a = 25.819$ (2) Å $b = 6.0060$ (4) Å $c = 20.1176$ (15) Å $\beta = 136.222$ (4)° $V = 2158.4$ (3) Å³ $Z = 8$ Mo $K\alpha$ radiation $\mu = 14.62$ mm⁻¹ $T = 298$ K

0.25 × 0.22 × 0.11 mm

Data collection

Bruker APEXII CCD area-detector diffractometer

Absorption correction: numerical (*SADABS*; Bruker, 2001) $T_{\text{min}} = 0.039$, $T_{\text{max}} = 0.222$

7675 measured reflections

2131 independent reflections

1671 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.078$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.049$ $wR(F^2) = 0.088$ $S = 1.10$

2131 reflections

127 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.72$ e Å⁻³ $\Delta\rho_{\text{min}} = -1.23$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1A}\cdots\text{N5}^i$	0.86	2.21	3.052 (14)	167
$\text{N1}-\text{H1B}\cdots\text{N3}^{ii}$	0.86	2.15	3.004 (13)	176

Symmetry codes: (i) $x, y + 1, z$; (ii) $-x, -y + 1, -z$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5873).

References

- Albada, G. A., Mutikainen, I., Turpeinen, U. & Reedijk, J. (2002). *Acta Cryst. E* **58**, m55–m57.
- Bruker (2001). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Castillo, A. S., Calahorra, A. J., Lazarte, D. C., Seco, J. M. & Dieguez, A. R. (2011). *Polyhedron*, **30**, 1295–1298.
- Cheng, X.-L., Gao, S. & Ng, S. W. (2009). *Acta Cryst. E* **65**, m1634.
- Cui, L. N., Hu, K. Y., Jin, Q. H., Li, Z. F., Wu, J. Q. & Zhang, C. L. (2011). *Polyhedron*, **30**, 2253–2259.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Gao, S. & Ng, S. W. (2010). *Acta Cryst. E* **66**, m1279.
- Lee, J. H. P., Lewis, B. D., Mendes, J. M., Turnbull, M. M. & Awwadi, F. F. (2003). *J. Coord. Chem.* **56**, 1425–1442.
- Li, Y., Zhang, X.-Q., Zhang, X.-C., Wang, X.-J. & Fang, R.-Q. (2006). *Acta Cryst. E* **62**, m1159–m1161.
- Lin, Z.-D. & Zeng, W. (2007). *Acta Cryst. E* **63**, m1597.
- Masaki, M. E., Prince, B. J. & Turnbull, M. M. (2002). *J. Coord. Chem.* **55**, 1337–1351.
- Qu, Y., Zhang, S. M., Wu, X. Z., Zhang, H. & Lin, Z. D. (2008). *Acta Cryst. E* **64**, m732.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
- Zhu, H. L., Ren, C. X. & Chen, X. M. (2002). *J. Coord. Chem.* **55**, 667–673.
- Zhu, H.-L., Yang, S., Ma, J.-L., Qiu, X.-Y., Sun, L. & Shao, S.-C. (2003). *Acta Cryst. E* **59**, m1046–m1047.

supplementary materials

Acta Cryst. (2012). E68, m646 [doi:10.1107/S1600536812016790]

catena-Poly[[2-aminopyrimidine- κN^1](thiocyanato- κS)mercury(II)]- μ -thiocyanato- $\kappa^2 S:N$]

Fatemeh Hoseinzadeh, Sadif A. Shirvan and Sara Haydari Dezfuli

Comment

Numerous complexes with 2-aminopyrimidine as a ligand have been prepared, such as that of cobalt (Li *et al.*, 2006), manganese (Lee *et al.*, 2003), nickel (Masaki *et al.*, 2002), zinc (Gao *et al.*, 2010; Qu *et al.*, 2008; Lin & Zeng, 2007), cadmium (Castillo *et al.*, 2011; Cheng *et al.*, 2009), silver (Zhu *et al.*, 2003; Cui *et al.*, 2011) and copper (Zhu *et al.*, 2002; Albada *et al.*, 2002). Here, we report the synthesis and structure of the title compound.

In the title coordination polymer, (Fig. 1), the Hg^{II} atom is four-coordinated in a butterfly configuration by one N atom from one 2-aminopyrimidine, one S atom from one terminal SCN ligand, one S atom from one bridging SCN and one N atom from one bridging SCN ligand.

In the crystal structure, intermolecular N—H \cdots N hydrogen bonds (Table 2, Fig. 2) stabilize the structure.

Experimental

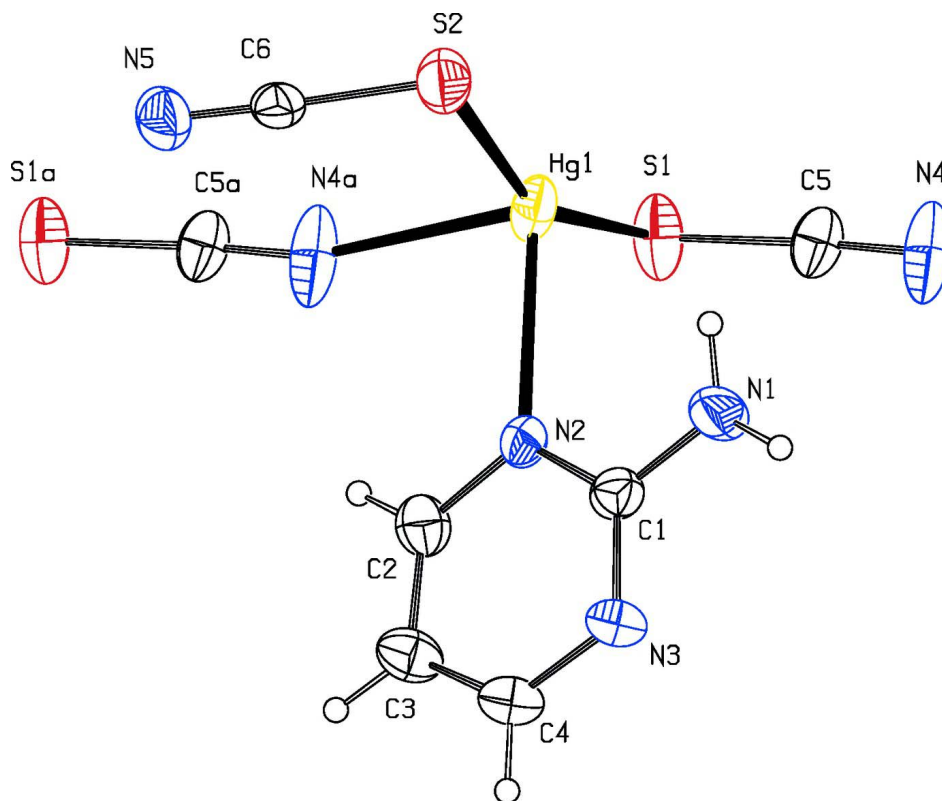
A solution of 2-aminopyrimidine (0.19 g, 2.0 mmol) in methanol (20 ml) was added to a solution of Hg(SCN)₂ (0.43 g, 1.0 mmol) in methanol (20 ml) and the resulting colorless solution was stirred for 20 min at 313 K. This solution was left to evaporate slowly at room temperature. After one week, colourless prismatic crystals of the title compound were isolated (yield; 0.33 g, 72.8%).

Refinement

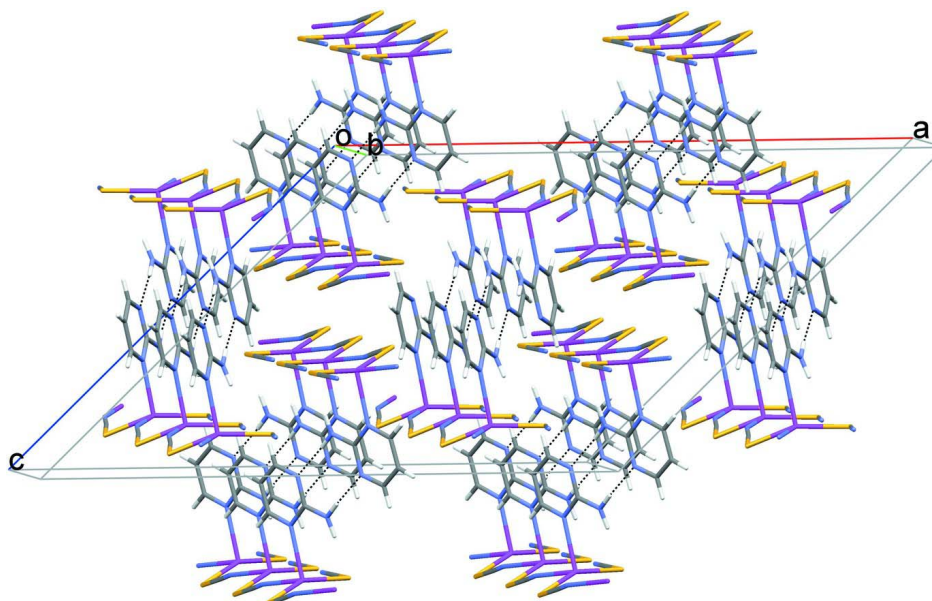
All H atoms were positioned geometrically, with C—H = 0.93 Å and N—H = 0.86 Å and constrained to ride on their parent atoms with $U_{\text{iso}}(\text{H})=1.2U_{\text{eq}}(\text{C},\text{N})$.

Computing details

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT* (Bruker, 2007); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

**Figure 1**

The molecular structure of the title molecule, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. [Symmetry codes: (a) $x, -1 + y, z$].

**Figure 2**

Unit-cell packing diagram for title molecule. Hydrogen bonds are shown as dashed lines.

catena-Poly[[2-(2-aminopyrimidine- κ N¹)(thiocyanato- κ S)mercury(II)]- μ -thiocyanato- κ S:N]

Crystal data

[Hg(NCS)₂(C₄H₅N₃)]
 $M_r = 411.88$
 Monoclinic, $C2/c$
 Hall symbol: $-C\ 2yc$
 $a = 25.819\ (2)\ \text{\AA}$
 $b = 6.0060\ (4)\ \text{\AA}$
 $c = 20.1176\ (15)\ \text{\AA}$
 $\beta = 136.222\ (4)^\circ$
 $V = 2158.4\ (3)\ \text{\AA}^3$
 $Z = 8$

$F(000) = 1504$
 $D_x = 2.535\ \text{Mg m}^{-3}$
 Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$
 Cell parameters from 7675 reflections
 $\theta = 2.9\text{--}26.0^\circ$
 $\mu = 14.62\ \text{mm}^{-1}$
 $T = 298\ \text{K}$
 Prism, colorless
 $0.25 \times 0.22 \times 0.11\ \text{mm}$

Data collection

Bruker APEXII CCD area-detector
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 ω scans
 Absorption correction: numerical
 (SADABS; Bruker, 2001)
 $T_{\min} = 0.039$, $T_{\max} = 0.222$

7675 measured reflections
 2131 independent reflections
 1671 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.078$
 $\theta_{\max} = 26.0^\circ$, $\theta_{\min} = 2.9^\circ$
 $h = -31 \rightarrow 27$
 $k = -7 \rightarrow 7$
 $l = -23 \rightarrow 24$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.088$
 $S = 1.10$
 2131 reflections
 127 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0349P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.008$
 $\Delta\rho_{\max} = 0.72\ \text{e \AA}^{-3}$
 $\Delta\rho_{\min} = -1.23\ \text{e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Hg1	0.16617 (2)	0.10942 (6)	0.33894 (3)	0.05561 (16)
S1	0.1305 (2)	0.3052 (4)	0.4049 (2)	0.0707 (8)
S2	0.25302 (16)	-0.0298 (4)	0.3378 (2)	0.0641 (7)
C1	0.0285 (5)	0.2478 (15)	0.1021 (6)	0.049 (2)
C2	0.0007 (6)	-0.0719 (16)	0.1318 (7)	0.055 (2)

H2	0.0126	-0.1780	0.1749	0.067*
C3	-0.0648 (5)	-0.0935 (19)	0.0381 (7)	0.058 (3)
H3	-0.0961	-0.2152	0.0153	0.069*
C4	-0.0821 (5)	0.0738 (17)	-0.0211 (7)	0.054 (3)
H4	-0.1278	0.0675	-0.0855	0.065*
C5	0.1323 (6)	0.5655 (15)	0.3790 (8)	0.055 (3)
C6	0.2379 (5)	-0.3009 (18)	0.3346 (6)	0.047 (2)
N1	0.0752 (5)	0.4154 (14)	0.1322 (6)	0.073 (3)
H1A	0.1176	0.4234	0.1916	0.088*
H1B	0.0629	0.5155	0.0923	0.088*
N2	0.0491 (4)	0.0940 (11)	0.1659 (5)	0.0437 (16)
N3	-0.0368 (4)	0.2472 (13)	0.0082 (5)	0.051 (2)
N4	0.1294 (7)	0.7461 (14)	0.3608 (9)	0.090 (4)
N5	0.2301 (5)	-0.4879 (17)	0.3329 (7)	0.070 (3)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Hg1	0.0759 (3)	0.03563 (19)	0.0604 (2)	0.0129 (2)	0.0510 (2)	0.0059 (2)
S1	0.123 (3)	0.0400 (12)	0.097 (2)	-0.0033 (15)	0.096 (2)	-0.0014 (14)
S2	0.0652 (17)	0.0482 (13)	0.0875 (19)	-0.0009 (13)	0.0581 (16)	-0.0002 (13)
C1	0.047 (5)	0.049 (5)	0.041 (5)	0.004 (4)	0.029 (5)	-0.001 (4)
C2	0.062 (6)	0.050 (6)	0.063 (6)	-0.005 (5)	0.048 (6)	0.002 (5)
C3	0.047 (5)	0.073 (7)	0.058 (6)	-0.013 (5)	0.039 (5)	-0.005 (6)
C4	0.039 (5)	0.072 (7)	0.045 (5)	-0.006 (5)	0.028 (4)	-0.004 (5)
C5	0.080 (7)	0.045 (6)	0.074 (7)	0.010 (5)	0.067 (6)	0.013 (5)
C6	0.033 (5)	0.050 (5)	0.049 (5)	0.001 (4)	0.026 (5)	-0.005 (4)
N1	0.062 (5)	0.058 (5)	0.043 (4)	-0.017 (4)	0.018 (4)	0.007 (4)
N2	0.047 (4)	0.036 (4)	0.042 (4)	0.004 (3)	0.030 (3)	0.006 (3)
N3	0.037 (4)	0.062 (5)	0.042 (4)	-0.007 (4)	0.024 (4)	-0.005 (4)
N4	0.163 (11)	0.036 (5)	0.164 (11)	0.016 (6)	0.148 (10)	0.014 (6)
N5	0.063 (6)	0.051 (5)	0.089 (7)	-0.004 (5)	0.053 (6)	-0.016 (5)

Geometric parameters (Å, °)

Hg1—S1	2.399 (5)	N3—C4	1.343 (16)
Hg1—S2	2.409 (5)	N4—C5	1.130 (13)
Hg1—N2	2.464 (7)	N5—C6	1.137 (15)
Hg1—N4 ⁱ	2.542 (13)	N1—H1B	0.8600
S1—C5	1.659 (11)	N1—H1A	0.8600
S2—C6	1.665 (11)	C2—C3	1.352 (15)
N1—C1	1.334 (16)	C3—C4	1.364 (16)
N2—C1	1.341 (12)	C2—H2	0.9300
N2—C2	1.337 (16)	C3—H3	0.9300
N3—C1	1.346 (12)	C4—H4	0.9300
Hg1...N5 ⁱⁱ	2.980 (13)	C2...S1 ⁱⁱⁱ	3.679 (16)
Hg1...H1A	2.9200	C3...S1 ⁱⁱⁱ	3.569 (15)
S1...N4 ⁱ	3.468 (10)	C4...S1 ⁱⁱⁱ	3.630 (15)
S1...C2 ⁱⁱⁱ	3.679 (16)	C4...N5 ^{vii}	3.44 (2)

S1...C3 ⁱⁱⁱ	3.569 (15)	C4...C1 ^{vi}	3.400 (19)
S1...C4 ⁱⁱⁱ	3.630 (15)	C5...C6 ⁱⁱ	3.52 (2)
S2...N5 ⁱⁱ	3.297 (11)	C5...N5 ⁱⁱ	3.27 (2)
S2...H3 ^{iv}	3.1900	C6...C5 ⁱ	3.52 (2)
N1...N5 ⁱⁱ	3.052 (14)	C1...H1B ^v	3.0900
N1...N3 ^v	3.004 (13)	C4...H1B ^v	3.0700
N3...N1 ^v	3.004 (13)	C6...H1A ⁱ	2.7900
N4...C6 ⁱⁱ	3.22 (3)	C6...H3 ^{iv}	3.0200
N4...S1 ⁱⁱ	3.468 (10)	C6...H4 ^{iv}	3.0200
N4...C2 ⁱⁱ	3.370 (16)	H1A...Hg1	2.9200
N5...N1 ⁱ	3.052 (14)	H1A...N5 ⁱⁱ	2.2100
N5...C4 ^{iv}	3.44 (2)	H1A...C6 ⁱⁱ	2.7900
N5...Hg1 ⁱ	2.980 (13)	H1B...N3 ^v	2.1500
N5...S2 ⁱ	3.297 (11)	H1B...C1 ^v	3.0900
N5...C5 ⁱ	3.27 (2)	H1B...C4 ^v	3.0700
N3...H1B ^v	2.1500	H2...N4 ⁱ	2.6500
N4...H2 ⁱⁱ	2.6500	H3...S2 ^{vii}	3.1900
N5...H1A ⁱ	2.2100	H3...C6 ^{vii}	3.0200
N5...H4 ^{iv}	2.7700	H4...N5 ^{vii}	2.7700
C1...C4 ^{vi}	3.400 (19)	H4...C6 ^{vii}	3.0200
S1—Hg1—S2	155.11 (12)	C1—N1—H1B	120.00
S1—Hg1—N2	103.3 (3)	N1—C1—N2	118.8 (8)
S1—Hg1—N4 ⁱ	89.1 (4)	N1—C1—N3	116.4 (9)
S2—Hg1—N2	100.0 (3)	N2—C1—N3	124.8 (10)
S2—Hg1—N4 ⁱ	99.8 (5)	N2—C2—C3	123.5 (10)
N2—Hg1—N4 ⁱ	89.0 (4)	C2—C3—C4	116.0 (11)
Hg1—S1—C5	100.3 (6)	N3—C4—C3	123.8 (10)
Hg1—S2—C6	98.4 (6)	S1—C5—N4	175 (2)
Hg1—N2—C1	124.1 (7)	S2—C6—N5	177.0 (15)
Hg1—N2—C2	119.5 (6)	N2—C2—H2	118.00
C1—N2—C2	116.4 (8)	C3—C2—H2	118.00
C1—N3—C4	115.4 (9)	C2—C3—H3	122.00
Hg1 ⁱⁱ —N4—C5	159.2 (18)	C4—C3—H3	122.00
H1A—N1—H1B	120.00	N3—C4—H4	118.00
C1—N1—H1A	120.00	C3—C4—H4	118.00
S2—Hg1—S1—C5	79.5 (5)	S2—Hg1—N4 ⁱ —C5 ⁱ	18 (3)
N2—Hg1—S1—C5	−80.0 (5)	N2—Hg1—N4 ⁱ —C5 ⁱ	118 (3)
N4 ⁱ —Hg1—S1—C5	−168.8 (6)	Hg1—N2—C1—N1	3.2 (18)
S1—Hg1—S2—C6	118.0 (4)	C2—N2—C1—N1	−179.2 (13)
N2—Hg1—S2—C6	−82.3 (4)	Hg1—N2—C1—N3	−176.8 (10)
N4 ⁱ —Hg1—S2—C6	8.4 (4)	C2—N2—C1—N3	1 (2)
S1—Hg1—N2—C1	91.2 (11)	C1—N2—C2—C3	2 (2)
S1—Hg1—N2—C2	−86.3 (10)	Hg1—N2—C2—C3	180.0 (12)
S2—Hg1—N2—C1	−80.2 (10)	C4—N3—C1—N2	−2 (2)
S2—Hg1—N2—C2	102.3 (11)	C4—N3—C1—N1	178.3 (13)
N4 ⁱ —Hg1—N2—C1	−179.9 (11)	C1—N3—C4—C3	0 (2)

N4 ⁱ —Hg1—N2—C2	2.5 (11)	N2—C2—C3—C4	-4 (2)
S1—Hg1—N4 ⁱ —C5 ⁱ	-139 (3)	C2—C3—C4—N3	3 (2)

Symmetry codes: (i) $x, y-1, z$; (ii) $x, y+1, z$; (iii) $-x, y, -z+1/2$; (iv) $x+1/2, -y-1/2, z+1/2$; (v) $-x, -y+1, -z$; (vi) $-x, -y, -z$; (vii) $x-1/2, -y-1/2, z-1/2$.

Hydrogen-bond geometry (Å, °)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1A \cdots N5 ⁱⁱ	0.86	2.21	3.052 (14)	167
N1—H1B \cdots N3 ^v	0.86	2.15	3.004 (13)	176

Symmetry codes: (ii) $x, y+1, z$; (v) $-x, -y+1, -z$.