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OPEN Unusual Flavones from *Primula* macrocalyx as Inhibitors of OAT1 and OAT3 and as Antifungal Agents against Candida rugosa

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A bioactivity guided program exploring the interaction of phytochemicals in the entire plant Primula macrocalyx with the organic anion transporters (OAT1 and OAT3) and microorganisms led to the elucidation of ten known flavones (1-4, 6-10, 12) and two previously undescribed flavones (5, 11). The structures of the compounds were determined by extensive analysis of spectroscopic data, as well as by comparison with data from previous reports. Two known flavones (9, 12) are reported for the first time from the family Primulaceae. All compounds were evaluated for inhibition of OAT1 and OAT3. Six flavones (2, 3, 6-8, 12) showed potent inhibitory activity on OAT1, while seven flavones (2, 3, 6-9, 12) showed marked inhibitory activity on OAT3, with IC₅₀ \leq 10.0 μ M. Antimicrobial activities of crude fractions against sixteen microorganisms were tested to give a target yeast strain Candida rugosa for further evaluation of MICs on the isolates. Three flavones (7, 8, 12) showed marked antifungal activity with MIC $< 2.0 \mu$ M. To our knowledge, this study is the first to evaluate these flavones as inhibitors of the OAT1 and OAT3, and as antifungal agents.

Primula macrocalyx Bunge (Primulaceae) is a perennial herbaceous plant, which has been widely used in folk medicine as an expectorant, diuretic, sedative, spasmolytic, and sudorific to treat a variety of maladies such as vitamin deficiency, colds, fever, headache, insomnia, paralysis, scurvy, tuberculosis, heart disease, rheumatism, and kidney diseases^{1,2}. The dosage forms involving *P. macrocalyx* are diverse, including tinctures, decoctions, powders and teas². Previous phytochemical investigations on P. macrocalyx have led to the isolation of flavones^{2,3}, triterpene glycosides and saponins⁴⁻⁶, bisbibenzyl compounds¹⁻³, salicylates and their derivatives². The content of free and total fatty acids, mainly palmitic, octadecatetraenoic, linoleic, and linolenic, from the aerial part of P. macrocalyx were determined by GC and GC-MS². While P. macrocalyx is rich in triterpene glycosides, the content of these compounds is dependant on the locality. Similarly, the content of ascorbic acid and flavonols in this plant decreased with increasing elevation above sea level⁸. Moreover, the plants of the genus *Primula* are considered promising as an accessible raw plant source of triterpene saponins in Russia9. Modern pharmacologic research has shown that riccardin C is a potent inhibitor of NO synthesis¹⁰ and the related bisbibenzyl compounds having cytotoxic, antibacterial, and fungicidal activity were inhibitors of 5-lipoxygenase¹. These chemical compositions may contribute to the medicinal properties mentioned above.

The organic anion transporters (OATs in humans or Oats in rodents) play key roles in the distribution and excretion of drugs¹¹. Specifically, organic anion transporter 1 (OAT1) and 3 (OAT3), which are highly expressed in the kidney, play an important part in the renal elimination of a range of substrate molecules 12,13. Moreover, both OAT1 and OAT3 are considered to be therapeutic targets for hypertension¹⁴. Research in mice suggests that Oat3 may mediate blood pressure regulation, so Oat3 inhibitors might be considered as potential antihypertensive agents¹⁵. The tincture of *P. macrocalyx* roots is widely used as a diuretic, and the tea of its flowers is drunk for

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R_1	R_2	R_3	R_4	R_5	R_6	R_7	R_8	R_9
Н	Н	Н	Н	Н	OCH ₃	Н	Н	Н
Н	Н	Н	Н	Н	Н	Н	Н	Н
Н	Н	Н	Н	OCH ₃	Н	Н	Н	Н
Н	Н	Н	Н	OCH ₃	Н	Н	OCH	₃ H
Н	Н	Н	Н	OCH ₃	Н	Н	OH	Н
Н	Н	Н	Н	Н	ОН	OCH ₃	OCH	₃ H
Н	Н	OCH ₃	OCH ₃	OCH ₃	Н	Н	Н	OCH ₃
Н	Н	OCH_3	OCH ₃	OCH_3	OCH	₃ H	Н	OCH_3
Н	Н	Н	Н	Н	ОН	Н	Н	Н
Н	Н	Н	Н	OH	Н	Н	Н	Н
OCH_3	Н	OCH ₃	OCH ₃	OCH ₃	Н	Н	Н	OCH_3
Н	Η	OCH ₃	OCH_3	OCH ₃	Н	Н	Н	Н
	H H H H H H H	H H H H H H H H H H H H H H H H H H H	H H H H H H H H H H H H H H H H OCH ₃ H H OCH ₃ H OCH ₃	H H H H H H H H H H H H H H H H H H H	H H H H H H H H H H OCH ₃ H H H H H H H H OCH ₃ OCH ₃ OCH ₃ H H OCH ₃ OCH ₃ OCH ₃ H H H H H H H H H H OH OCH ₃ OCH ₃ OCH ₃ OCH ₃	H H H H H OCH ₃ H H H H H H H H H H OCH ₃ H H OCH ₃ OCH ₃ OCH ₃ OCH ₃ OCH ₃ OCH ₄ OCH ₃ OCH ₃ OCH ₄ OCH ₃ OCH ₃ OCH ₃ OCH ₄ OCH ₃ OCH ₃ OCH ₄ OCH ₄ OCH ₃ OCH ₃ OCH ₄	H H H H H OCH ₃ H OCH ₃ H H H H H H OCH ₃ H H H H H OCH ₃ H H H H H H OCH ₃ H H H H H H OCH ₃ OCH ₃ H H H H OCH ₃ OCH ₃ OCH ₃ H H H H OCH ₃ OCH ₃ OCH ₃ H H H H H H H OH H H H H H H OH H	H H H H H OCH ₃ H H H H H H H H H H H H H H H H H H H

Figure 1. Structures of compounds 1-12.

kidney disease in folk medicine², making the interaction between OAT1/3 and *P. macrocalyx* an attractive target for further investigation.

Recent years have seen a resurgence of interest in antimicrobial agents from plants due to their ethnomedicinal uses and low toxicity and side effects. Particularly, developing countries rely on plants for the treatment of infectious and non-infectious diseases¹⁶. *P. macrocalyx* powder is in ethnomedicinal use for the treatment of tuberculosis¹. Herein, we screened four fractions (*n*-hexane-soluble, dichloromethane-soluble, *n*-butanol-soluble and water-soluble) of the methanol extract of *P. macrocalyx* on sixteen kinds of microorganisms as part of an ongoing search for new antimicrobial chemotypes.

In our preliminary studies, the dichloromethane soluble fraction of a methanol extract of entire plant of *P. macrocalyx* elicited marked inhibition of OAT1 and OAT3 *in vitro*, and potent antifungal activity against yeast strain *Candida rugosa*.

In the present study, a bioactivity guided fractionation was performed on the methanol extract of *P. macrocalyx* collected in Armenia, followed by structure determination of the isolated compounds based on LC-MS and NMR, leading to the elucidation of twelve flavones (1–12), including two previously undescribed compounds (5, 11). To our knowledge, this study is the first to evaluate these flavones as inhibitors of the OAT1 and OAT3. These data may allow an initial elucidation of the structure activity relationships within this group, and may also provide a rational basis for the therapeutic applications of *P. macrocalyx* in traditional medicine. Additionally, the isolated antifungal agents could play a complementary role in the chemotherapy of fungal infections.

Results

Samples comprising the whole plant of P. macrocalyx were extracted with methanol. The methanol-free extract was subjected to standard solvent partition, bioassay as well as a combination of different chromatographic techniques to afford twelve flavones (1-12), including two previously undescribed flavones (5, 11) (Fig. 1).

Compound 5 was obtained as a yellow amorphous powder. It showed two quasimolecular ions at m/z 269.0805 $[M+H]^+$ (calcd. for $C_{16}H_{12}O_4$ 269.0808) and 291.0625 $[M+Na]^+$ (calcd. for $C_{16}H_{12}O_4$ Na 291.0633) corresponding to the molecular formula $C_{16}H_{12}O_4$ in the HRESIMS. It was ascribed as having a flavone skeleton ^{17,18} bearing methoxy and hydroxy substituents as shown by the ¹H and ¹³C NMR spectroscopic analysis (Table 1). In the COSY spectrum (Fig. 2), the correlations: δ_H 6.96 (H-4') and δ_H 7.08 (H-3'), δ_H 7.70 (H-8) and δ_H 7.82 (H-7) and δ_H 7.49 (H-6), and δ_H 7.49 (H-6) and δ_H 8.04 (H-5), were observed, which established two spin systems of 5. Among them, the characteristic coupling pattern observed at δ_H 6.96 (1 H, dd, J = 8.9, 3.1 Hz, H-4'), 7.08 (1 H, d, J = 8.9 Hz, H-3'), and 7.31 (1 H, d, J = 3.1 Hz, H-6') in the ¹H NMR spectrum indicated the presence of a B-ring with a typical ABX system. The methoxy protons at δ_H 3.83 showed HMBC correlations (Fig. 2) with the carbon at δ_C 150.8, and NOESY correlations (Fig. 2) with δ_H 6.93 (H-3) and δ_H 7.08 (H-3'), establishing the location of this substituent group at C-2'. In addition, the downfield hydroxyl was located at C-5' in the B-ring, supported by HMBC correlations between δ_H 9.41 and C-4' (δ_C 119.3), C-6' (δ_C 115.0) and C-5' (δ_C 151.1), and NOESY correlations of δ_H 9.41 with δ_H 7.31 and δ_H 6.96. If the hydroxyl group had been located at C-4' rather than C-5', H-3' would be flanked by two oxygen containing substituents and expected farther upfield around

	5 (DMSO-d ₆)		11 (Acetone-d ₆)		11 (DMSO-d ₆)		11 (CDCl ₃)	
no.	$\delta_{\rm C}$, type	$\delta_{ m H}$	$\delta_{\rm C}$, type	$\delta_{ m H}$	$\delta_{\rm C}$, type	$\delta_{ m H}$	$\delta_{\rm C}$, type	$\delta_{ m H}$
2	160.5, C		159.5, C		158.5, C		158.8, C	
3	111.5, CH	6.93, s	115.6, CH	6.06, s	114.4, CH	6.07, s	115.5, CH	6.36, s
4	177.1, C		177.4, C		176.3, C		178.1, C	
5	124.7, CH	8.04, dd (7.9, 1.0)	140.9, C		138.7, C		140.8, C	
6	125.4, CH	7.49, dt (7.9, 1.0)	146.4, C		145.0, C		145.4, C	
7	134.3, CH	7.82, dt (7.9, 1.0)	104.6, CH	7.18, s	103.7, CH	7.21, s	104.3, CH	6.89, s
8	118.4, CH	7.70, d (7.9)	150.4, C		149.1, C		149.0, C	
9	155.8, C		143.2, C		141.2, C		143.2, C	
10	123.1, C		120.5, C		118.7, C		119.8, C	
1'	120.3, C		112.3, C		110.5, C		111.4, C	
2'	150.8, C		159.5, C		158.0, C		158.7, C	
3'	114.0, CH	7.08, d (8.9)	105.1, CH	6.78, d (8.6)	104.3, CH	6.79, d (8.5)	104.1, CH	6.60, d (8.4)
4'	119.3, CH	6.96, dd (8.9, 3.1)	133.1, CH	7.46, t (8.6)	132.4, CH	7.47, t (8.5)	132.0, CH	7.35, t (8.4)
5'	151.1, C		105.1, CH	6.78, d (8.6)	104.3, CH	6.79, d (8.5)	104.1, CH	6.60, d (8.4)
6'	115.0, CH	7.31, d (3.1)	159.5, C		158.0, C		158.7, C	
5-OCH ₃			61.6, CH ₃	3.78, s	61.1, CH ₃	3.68, s	62.0, CH ₃	3.90, s
6-OCH ₃			57.1, CH ₃	3.91, s	56.5, CH ₃	3.88, s	57.1, CH ₃	3.90, s
8-OCH ₃			57.5, CH ₃	3.92, s	56.9, CH ₃	3.89, s	57.7, CH ₃	3.92, s
2'-OCH ₃	56.3, CH ₃	3.83, s	56.5, CH ₃	3.80, s	56.1, CH ₃	3.74, s	56.0, CH ₃	3.76, s
6'-OCH ₃			56.5, CH ₃	3.80, s	56.1, CH ₃	3.74, s	56.0, CH ₃	3.76, s
5'-OH		9.41, s						

Table 1. 13 C (150 MHz) and 1 H (600 MHz) NMR Data of Compounds **5** and **11** (δ in ppm, J in Hz).

 $\delta_{\rm H}$ 6.7 PPM¹⁹. Additionally, this proton would be expected to show NOESY correlations with both the methoxy protons and the proton of the hydroxy group. This is clearly not the case.

In the region of $\delta_{\rm H}$ 7.49–8.04, four adjacent aromatic protons of the A-ring were confirmed by $^1{\rm H}$ COSY correlations described above and a characteristic coupling pattern located at $\delta_{\rm H}$ 7.49 (1 H, dt, J = 7.9, 1.0 Hz, H-6), 7.70 (1 H, d, J = 7.9 Hz, H-8), 7.82 (1 H, dt, J = 7.9, 1.0 Hz, H-7), and 8.04 (1 H, dd, J = 7.9, 1.0 Hz, H-5), which indicated the presence of an unsubstituted flavone A-ring. HMBC correlations from H-5 ($\delta_{\rm H}$ 8.04) to C-4 ($\delta_{\rm C}$ 177.1), C-9 ($\delta_{\rm C}$ 155.8) and C-7 ($\delta_{\rm C}$ 134.3) were observed, establishing that these protons are situated at C-6, C-8, C-7, and C-5 in the A-ring, respectively. These results support assignment of the structure of 5 as 2'-methoxy-5'-hydroxyflavone.

Compound 11 was obtained as a pale yellow amorphous powder. Its molecular formula, $C_{20}H_{20}O_7$, was determined by HRESIMS m/z 373.1275 [M+H]⁺ (calcd $C_{20}H_{21}O_7$ 373.1282). On the basis of 1D and 2D NMR spectra recorded in acetone- d_6 (Table 1), 11 was also found to have a flavone skeleton bearing five methoxy groups. The HMBC correlations (Fig. 2) from the aromatic proton at δ_H 7.18 to carbons bearing methoxy groups at δ_C 140.9, δ_C 146.4, and δ_C 150.4, and NOESY correlations (Fig. 2) between this aromatic singlet (δ_H 7.18) and the methoxy group protons at δ_H 3.91 and δ_H 3.92, and between one methoxy signal at δ_H 3.78 and another one at δ_H 3.91, indicated the presence of a 5,6,8-trimethoxy substituted A-ring. The characteristic proton singlet observed at δ_H 6.06 (H-3) showed HMBC interactions with C-1′ (δ_C 112.4), C-10 (δ_C 120.5), C-5 (δ_C 140.9 (w)), C-2 (δ_C 159.5) and C-4 (δ_C 177.4), also supported the position of substituents in the A-ring. The B-ring substitution of 11 was the same as that of the known 5,6,2′,6′-tetramethoxyflavone (7)²0. From these data, 11 was characterized as 5,6,8,2′,6′-pentamethoxyflavone (11). The NMR data of 11 recorded in DMSO- d_6 and CDCl₃ are also provided in Table 1.

The remaining ten known flavones were identified by interpretation of their 1D and 2D NMR, HRESIMS and UV spectral data and comparison with data available in the literature, including 3'-methoxyflavone ($\mathbf{1}$)²¹, flavone ($\mathbf{2}$)²², 2'-methoxyflavone ($\mathbf{3}$)²¹, 2',5'-dimethoxyflavone ($\mathbf{4}$)²⁰, 3'-hydroxy-4',5'-dimethoxyflavone ($\mathbf{6}$)²⁰, 5,6,2',6'-tetramethoxyflavone ($\mathbf{7}$)²⁰, 5,6,2',3',6'-pentamethoxyflavone ($\mathbf{8}$)²³, 3'-hydroxyflavone ($\mathbf{9}$)²⁴, 2'-hydroxyflavone ($\mathbf{10}$)²⁴, 5,6,2'-trimethoxyflavone ($\mathbf{12}$)²⁵. Their ¹H and ¹³C NMR data are shown in Tables 1–3 in supplementary information.

All compounds (purity > 95%) isolated from the active fractions of *P. macrocalyx* were tested for their inhibitory activity on OAT1 and OAT3. When evaluated as inhibitors of OAT1 and OAT3 (Table 2), these flavones showed a concentration dependent inhibition of the transporters with IC_{50} 's ranging from 3 to $50 \,\mu\text{M}$. The observed activities were consistent with the activity originally observed in the crude extract.

Sixteen microorganisms, including eight bacteria, Staphylococcus aureus, Staphylococcus epidermis, Streptococcus mutans, Pseudomonas fluorescens, Enterococcus hirae, Moraxella (Branhamella) catarrhalis, Pseudomonas aeruginosa, Bacillus subtilis subsp. Spizizenii, and eight fungi, Candida albicans, Aspergillus niger, Saccharomyces kudriavzevii, Penicillium chrysogenum, Candida parapsilosis, Candida rugosa, Candida tropicalis and Rhizopus stolonifer, were used to evaluate antimicrobial activities of the four crude fractions from P. macrocalyx, at a concentration of 1 mg/mL in DMSO, using an agar well diffusion assay. Results showed the dichloromethane fraction had marked antifungal activity against Candida rugosa with a 20 mm diameter zone of inhibition

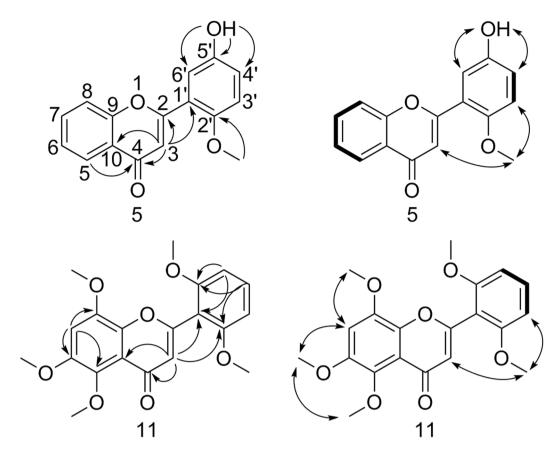


Figure 2. Key ${}^{1}H \rightarrow {}^{13}C$ HMBC (arrow), ${}^{1}H \rightarrow {}^{1}H$ NOESY (double arrow) and ${}^{1}H \rightarrow {}^{1}H$ COSY (solid bond) correlations of compounds 5 and 11.

		IC ₅₀ (µM) ^a	IC ₅₀ (μM) ^b	MICs (μM) ^c
No.	Name	OAT1	OAT3	C. rugosa
1	3'-methoxyflavone	_	_	_
2	Flavone	9.4	4.6	500
3	2'-methoxyflavone	2.9	6.3	_
4	2',5'-dimethoxyflavone	32.9	12.1	_
5	2'-methoxy-5'-hydroxyflavone	26.9	27.3	_
6	3'-hydroxy-4',5'-dimethoxyflavone	3.2	5.1	_
7	5,6,2',6'-tetramethoxyflavone	8.4	7.3	0.4
8	5,6,2',3',6'-pentamethoxyflavone	10.0	7.4	1.2
9	3'-hydroxyflavone	11.4	8.0	_
10	2'-hydroxyflavone	_	18.8	_
11	5,6,8,2',6'-pentamethoxyflavone	_	10.9	_
12	5,6,2'-trimethoxyflavone	9.6	2.9	2.0
	Probenecid	29.4	8.7	_
	Amphotericin B	_	_	0.4

Table 2. IC₅₀ values for OAT1 and OAT3 inhibition and MIC values against *C. rugosa* of isolated compounds. ^aCompounds **1, 10, 11** showed <70% inhibition of 6-CF uptake up by OAT1 to 50 μ M. ^bCompound **1** showed <70% inhibition of 6-CF uptake up by OAT3 to 50 μ M. ^cCompounds **1, 3–6,** and **9–11** did not inhibit *C. rugosa* up to 500 μ M.

(positive control, amphotericin B: 11 mm). Bioactivity guided isolation was performed to isolate the active antifungal compounds. Minimal inhibitory concentration (MIC) determinations are summarized in Table 2.

Discussion

Twelve flavones (1–12) have been isolated and identified from the active dichloromethane soluble fraction of the entire plant of *P. macrocalyx*. Among them, 5 and 11 are newly described. Compounds 1–3, 6, 7, 10 are reported

for the first time from this species while 4 and 8 have been reported previously from this plant², and 6 was previously reported from *Primula veris*²⁰. Compounds 9 and 12 are herein reported for the first time from family Primulaceae. Additionally, 9 is described for the first time as a natural product, having been previously described as a product of chemical synthesis or biotransformation^{26,27}. Previous reports and our present research on the phytochemistry of this species indicate that flavones are characteristic chemical constituents in *P. macrocalyx*. The unusual substitution patterns of the newly described constituents may prove useful in further chemotaxonomic studies of *P. macrocalyx* or the Primulaceae in general.

Evaluation of the isolated compounds as OAT inhibitors showed that six flavones (2, 3, 6–8, 12) showed good concentration-dependent inhibition on OAT1 mediated 6-CF uptake and seven flavones (2, 3, 6–9, 12) showed good concentration-dependent inhibition on 6-CF uptake mediated by OAT3 with $IC_{50} \leq 10\,\mu\text{M}$. Among these compounds, 2'-methoxyflavone (3) was the most potent inhibitor of OAT1, with IC_{50} value of $2.9\,\mu\text{M}$, while 3'-hydroxy-4',5'-dimethoxyflavone (6) was of comparable potency with an IC_{50} of $3.2\,\mu\text{M}$. Comparing 2 with 3 and 10, the inhibitory activity on OAT1 increased with the presence of methoxy group and decreased with the presence of hydroxyl group at the 2' position in the B-ring. Additional B ring substitution appeared to reduce potency. Compound 12, 5.6.2'-trimethoxyflavone, was the most potent inhibitor of OAT3, with an IC_{50} of $2.9\,\mu\text{M}$. The parent molecule, unsubstituted flavone (2) was also a potent inhibitor on OAT3, with an IC_{50} of $4.6\,\mu\text{M}$. Comparing 2 with 3 and 10, the addition of a methoxy or hydroxy group at the 2' position in the B-ring reduced the inhibitory activity on OAT3. Again, additional substitution on the B ring appeared to reduce potency.

Inhibitory activity on OAT1/3 of these flavones is highly dependent on the number and nature of the substituents attached to the flavone ring system, as well as the substitution pattern. In our previous study, 2'-hydroxy-6, 7,8-trimethoxyisoflavone and 7-methoxyflavone showed inhibitory activity against OAT1 with IC₅₀ values of 9.1 and $3.6\,\mu\text{M}$, respectively, and 7.2'-dihydroxy-6,8-dimethoxyisoflavone, 2'-hydroxy-6,7,8-trimethoxyisoflavone, 6,2'-dihydroxy-7,8-dimethoxyisoflavone, and 7-methoxyflavone showed moderate inhibitory activity against OAT3 with IC₅₀ values of 5.6, 8.7, 17.9, and $5.8 \mu M$, respectively²⁸. Morin and luteolin were reported to be potent OAT1 inhibitors (IC₅₀ values of <0.3 and 0.47 μ M, respectively) in a para-aminohippuric acid (PAH) uptake assay in LLC-PK1 cells, while the tested flavonoid glycosides had weak inhibitory effects on OAT1²⁹. Apigenin has been reported to inhibit OAT1 (IC₅₀ value of 0.737 µM) on the uptake of acyclovir in MDCK cells³⁰. Wogonin has been reported to inhibit OAT1 (IC $_{50}$ value of $5.4 \mu M$), and baicalein and wogonin have been reported to inhibit OAT3 (IC₅₀ values of 2.4 and 1.3 μ M, respectively) in an assay similar to that used in our study³¹. In addition, flavone was identified by UPLC-ESI-MS/MS as the one main active component of Gnaphalium pennsylvanicum extract which showed activity in reducing serum uric acid through OAT132. We have also recently reported the OAT3 inhibitory activity of the more complex dimeric biflavonoids, amentoflavone, cupressuflavone and podocarpusflavone A from Juniperus oblonga³³, with IC₅₀ values of 2.0 μ M, > 50 μ M and 3.8 μ M, respectively. Orally administered amentoflavone markedly altered the pharmacokinetic parameters of furosemide, a substrate of Oat3, in rats. Comparison of potency data among these studies is complicated by the use of the number of different bioassays commonly used in this area.

The unusual substitution patterns seen in compounds 1–12 may help to define the flavone chemotype as a pharmacophore for OAT inhibitors, providing the basis for future structure activity relationship studies. Identification of these active compounds may also help establish the chemical basis for use of this herb in the treatment of kidney disease.

Evaluation of these flavones as antifungal agents against C. rugosa showed that compounds 7, 8, and 12 possessed marked antifungal activity against this organism with MIC values of $0.4\,\mu$ M, $1.2\,\mu$ M, and $2.0\,\mu$ M, respectively. These three flavones share the common structural features of methoxylation at positions 5 and 6 of the A ring and position 2' of the B ring. While 11 which is inactive also shares these features, it is also methoxylated at C-8, suggesting that substitution at C-8 greatly reduces or eliminates activity. Although flavone (2) shows minimal activity against C. rugosa (MIC = $500\,\mu$ M), none of the other compounds lacking the methoxylation at 5, 6, and 2' showed any antifungal activity. Most known flavones are oxygenated at positions 5 and 7 on the A ring but are not noted for having antifungal activity, suggesting that the unusual 5,6-disubstitution pattern found in our isolates may be required for this activity. Further study is needed to define optimal substitution on the B ring and to explore the effect of substituents other than methoxy groups on the activity of these compounds. The pressing need for new antifungal chemotypes, especially for non-albicans Candida infections including the emerging "superbug", C. auris, highlights the importance of the discovery and potential development of a new pharmacophore such as this.

Methods

General experimental procedures. UV spectra were recorded on a HITACHI U-3900 spectrophotometer (Tokyo, Japan). UPLC-QQQ-MS data were recorded on Agilent 1260–6420 series spectrometer (Agilent Technologies, Santa Clara, CA, USA). HRESIMS spectra were performed on a MicroTOF-QII mass spectrometer (Bruker Daltonics Inc, Billerica, MA, USA) in positive or negative ion mode. All 1D and 2D NMR spectra were acquired using Avance III spectrometers (Bruker-Biospin, Billerica MA, USA) operating at 400 MHz or 600 MHz (1 H) and 100 MHz or 150 MHz (1 C) at ambient temperature. Residual solvent resonances were used as internal standard. Chemical shifts are expressed in PPM. Open column chromatography was carried out using Sephadex LH-20 (GE Healthcare Bio-Science AB, Uppsala, Sweden). Thin-layer chromatography (TLC) was performed on precoated layers of Silica gel GF254 (Qingdao Marine Chemical, Inc., Qingdao, China). Flash chromatography was performed on a CombiFlash Rf+instrument (Teledyne-ISCO, Lincoln, NE, USA) with prepacked silica flash column (Santai Technologies, Inc., Jiangsu, China) in various sizes. Analytical and semi-preparative HPLC were performed on an Agilent 1260 VL quad gradient system (G1311C pump, G1329B autosampler, G1316A thermostatted column compartment and G1315D photodiode array detector, Agilent Technologies, Santa Clara CA, USA), using either a Thermo Hypersil Gold C_{18} (150 × 2.1 mm, 3 μ m), an Agilent Poroshell 120 EC- C_{18} analytical

chromatographic column (150×2.1 mm, 2.7μ m), or a Thermo Hypersil Gold C_{18} semipreparative chromatographic column (250×10 mm, 5μ m). Preparative HPLC separations were performed on an Agilent 1260VL quad gradient system (G1311C pump, and G1315D photodiode array detector) equipped with a Rheodyne 7725i manual injection valve (IDEX Health & Science, Middleboro MA, USA) and Shimadzu CTO-20A column oven (Shimadzu Scientific Instruments, Kyoto, Japan), using a Thermo Hypersil Gold C_{18} chromatographic column (250×21.2 mm, 5μ m). The determination of fluorescence was performed on an Infinite M200 plate reader (Tecan, Mannedorf, Switzerland) with excitation and emission wavelengths at 485 and 528 nm, respectively. Microbial culture density was measured on a MAPADA UV-1800 spectrophotometer (MAPADA, Shanghai, China). Incubation and biological operation were performed in electro-thermal constant temperature incubators and super clean bench (Shanghai Zhicheng Analysis Instrument Manufacturing Co., Ltd., Shanghai, China), respectively. 96 well cell culture plates (Costar 3599) was purchased from Corning (NY, USA). Optical density reader for microorganism was recorded on a Multiskan plate reader (Thermo Fisher Scientific Oy Microplate Instrumentation, Vantaa, Finland).

All solvents used were HPLC grade (Tianjin Concord Technologies; Tianjin Guangfu Technology Development Co., Ltd., Tianjin, China). 6-Carboxyfluorescein (6-CF) was purchased from Aladdin (Shanghai, China). Probenecid and hygromycin B were purchased from Solarbio (Beijing, China). Poly-D-Lysine was purchased from Sigma-Aldrich Chemical Co. (St. Louis, MO, USA). Dulbecco's modified Eagle's medium (DMEM), trypsin and fetal bovine serum (FBS) were purchased from Gibco (Gaithersburg, MD, USA). Bicinchoninic acid (BCA) protein assay kit was purchased from Cwbio (Beijing, China).

All antibiotics, such as penicillin, ampicilin, ciprofloxacin, cephalexin, vancomycin amphotericin B, and fluconazole, were purchased from Sigma-Aldrich Chemical Co. (St. Louis, MO, USA). Five media, such as nutrient ager (AOBOX, Beijing, China), brain heart infusion broth and potato dextrose broth (Solarbio, Beijing, China), YM medium (Hopebio, Qingdao, China), and agar (Guangfu-chem, Tianjin, China), were used. Microorganisms were obtained from the American Type Culture Collection (ATCC, Manassas, Virginia, USA).

Plant material. The whole plants *Primula macrocalyx* Bunge (Primulaceae) used in this study were collected in Armenia, Vayots Dzor Province, near Yeghegnadzor, between Aghnjadzor and the Selim Pass (Alt: 2100 m; GPS: 39.9300 N, 45.2425 E) in May 2006. Voucher specimens (G. Fayvush 10–2006) have been deposited in the herbaria of the Institute of Botany, Armenian National Academy of Sciences (ERE), and the New York Botanical Garden (NY).

Extraction and isolation. Plant samples comprising the whole plant of *P. macrocalyx* were freed of extraneous matter, air dried in the shade and then ground to a coarse powder. A 1 kg portion of this sample was extracted three times with methanol (6 L, 1 day each) at room temperature to give the methanol extract (108 g) on removal of the solvent *in vacuo*.

The sample was dissolved in methanol and water (300 mL, 9:1 v/v), and extracted with n-hexane (300 mL \times 3) to get an n-hexane fraction. The residual methanolic phase was freed of methanol in vacuo, suspended in water (300 mL), extracted with dichloromethane (300 mL \times 3) followed by water-saturated n-butanol (300 mL \times 3), to obtain a dichloromethane fraction, an n-butanol fraction and an aqueous fraction. The four fractions were each freed of solvents in vacuo and subjected to preliminary evaluation of inhibitory activity on OAT1 and OAT3 and antimicrobial activities against sixteen microorganisms. The dichloromethane fraction elicited strong inhibition of OAT1 and OAT3 and strong antifungal activity against C. vugosa, in vitro.

The active dichloromethane soluble fraction (3.5 g) was then fractionated by chromatography on silica gel (40–63 μ m, 60 Å, 200 g, 204 × 60 mm) eluted with a step-gradient of dichloromethane, ethyl acetate (100:0, 98:2, 96:4, 94:6, 92:8, 90:10, 70:30, 50:50, 0:100, 780 mL each) to yield fractions 1–30 (each 260 mL). Thirty fractions from this separation were evaluated for bioactivity on OAT1 and OAT3 and antifungal activity against *C. rugosa*. HPLC comparison of the active fractions using an Agilent Poroshell 120 EC-C₁₈ column (150 × 2.1 mm, 2.7 μ m) eluted with ACN: H₂O (1:1) containing 0.1% formic acid at 0.2 mL/min at 40 °C (fractions 1–15) or ACN: H₂O (2:3) containing 0.1% formic acid at 0.2 mL/min at 40 °C (fractions 16–30) allowed combination to form 7 pools, Fractions A-G.

Active fractions 4–5 from the silica gel column were pooled to form Fraction A, which consisted of one major peak (1) with a retention time of 7.45 min. Fraction A (216 mg) was purified by recrystallization using dichloromethane-methanol mixtures (7:3, 10 mL) three times to afford compound 1 (10 mg).

Active fractions 6–7 from the silica gel column were pooled to form Fraction B, which consisted of two main peaks with retention times of 6.78 min (2) and 7.25 min (3). Fraction B (372 mg) was chromatographed on a flash silica column $(40-63\,\mu\text{m}, 60\,\text{Å}, 4\,\text{g}, 106\times12\,\text{mm})$ eluted with a linear gradient of n-hexane and dichloromethane (75:25–50:50, 240 mL) and purified further on Sephadex LH-20 (12 g, $305\times17\,\text{mm}$) eluted with dichloromethane: methanol (50:50, 200 mL) to yield compound 3 (12 mg) and 2 (9 mg).

Active fractions 12 from the silica gel column formed Fraction C, which consisted of one main peak with a retention time of 7.69 min (4). Fraction C (403 mg) was purified by gel filtration on Sephadex LH-20 (12 g, 305×17 mm) eluted with dichloromethane: methanol (50:50, 200 mL), then methanol (100 mL), affording compound 4 (34 mg).

Active fractions 17–20 from the silica gel column were pooled to form Fraction D, which consisted of two main peaks with retention times of 10.13 min (7) and 12.68 min (12). Active fraction 21 formed Fraction E with two main components eluting at 5.06 min (5) and 6.22 min (6). Fractions D and E were combined (780 mg) and separated by repeated preparative HPLC (Thermo Hypersil Gold C_{18} , 250×21.2 mm, 5μ m) eluted with a gradient of 22–28% aqueous MeCN containing 0.1% formic acid (1200 mL, at 10 mL/min, 40 °C) and further purified by semi-preparative HPLC (Thermo Hypersil Gold C_{18} , 250×10 mm, 5μ m, 4 mL/min; 40 °C) eluted isocratically

with 22% (400 mL) or 24% (480 mL) aqueous MeCN containing 0.1% formic acid to afford compound 5 (17 mg), 6 (20 mg), 7 (10 mg) and 12 (15 mg).

Active fractions 22–24 from the silica gel column were pooled to form Fraction F, which consisted of four main peaks with retention times of 5.63 min (9), 7.16 min (10), 8.84 (11), and 10.18 min (8). Fraction F (330 mg) was subjected to preparative HPLC (Thermo Hypersil Gold C_{18} , 250×21.2 mm, $5\,\mu$ m, 8 mL/min, $40\,^{\circ}$ C) eluted with 22–25% aqueous MeCN with 0.1% formic acid (360 mL) to give 11 (5 mg) and 8 (12 mg), and purified further by semi-preparative HPLC (Thermo Hypersil Gold C_{18} , 250×10 mm, $5\,\mu$ m, 4 mL/min; $40\,^{\circ}$ C) eluted with 22% aqueous MeCN with 0.1% formic acid (240 mL) to afford 9 (8 mg) and 10 (17 mg).

Active fraction 25 from the silica gel column afforded additional quantities of 8.

Antifungal activity was limited to fractions 17-25 from the silica gel column (Fractions D-G).

2'-methoxy-5'-hydroxyflavone (5). Yellow amorphous powder; UV (MeOH) λ_{max} (log ε) 212 sh (3.90), 245 (3.75), 295 (3.63), 352 (3.39) nm; HRESIMS m/z 269.0805 [M+H]+ (calcd. for $C_{16}H_{13}O_4$ 269.0808), m/z 291.0625 [M+Na]+ (calcd. for $C_{16}H_{12}O_4$ Na 291.0633); ¹H NMR (DMSO- d_6 , 600 MHz) and ¹³C NMR (DMSO- d_6 , 150 MHz) data, see Table 1.

5,7,8,2',6'-pentamethoxyflavone (11). Pale yellow amorphous powder; UV (MeOH) λ_{max} (log ε) 227 (4.05), 261 (3.91), 335 (3.39) nm; HRESIMS m/z 373.1275 [M+H]⁺ (calcd. for $C_{20}H_{21}O_7$ 373.1282); ¹H NMR (Acetone- d_6 , DMSO- d_6 , and CDCl₃, 600 MHz) and ¹³C NMR (Acetone- d_6 , DMSO- d_6 and CDCl₃, 150 MHz) data, see Table 1.

Cell culture. Human embryonic kidney 293 (HEK293) cell lines stably overexpressing OAT1 and OAT3 were established and identified as previously described¹⁴. Cell lines stably expressing HEK-OAT1 and HEK-OAT3 were obtained by hygromycin B (75 μ g/mL) selection, and further characterized by both mRNA expression of transporters and their uptake of 6-CF, a fluorescent substrate for both OAT1 and OAT3³⁴. The cells were cultivated in DMEM supplemented with 10% FBS, 1% penicillin/streptomycin, and 75 μ g/mL hygromycin B at 37 °C with 5% CO₂.

Uptake assay. The interactions of 12 flavones with the uptake of 6-CF in HEK-OAT1 and HEK-OAT3 cells were evaluated. This cell uptake assay was performed as previously described 14,35 . A density of 5×10^4 cells were seeded per well in 96-well culture plates precoated with poly-D-lysine. Approximately 85% confluency of cells was obtained after growing 24 h. The cells were washed twice and preincubated for 5 min with preheated (37 °C) uptake buffer (135 mM NaCl, 5 mM KCl, 2.5 mM CaCl₂, 1.2 mM MgCl₂, 0.8 mM MgSO₄, 28 mM glucose, and 13 mM Hepes, pH 7.2) for the following uptake experiments. The uptake buffer containing 4μ M 6-CF in the presence or absence of test compounds and probenecid (a classic inhibitor of OAT1 and OAT3, used as a positive control) was incubated for 5 min to allow uptake. Uptake was terminated by adding 100 µL ice-cold uptake buffer, and quickly washing the cells in each well three times with ice-cold phosphate-buffered saline (PBS). The cells were lysed with $100 \,\mu\text{L}$ of $20 \,\text{mM}$ Tris-HCl containing 0.2% TritonX-100. A $50 \,\mu\text{L}$ aliquot of lysate was used to quantify fluorescence using a Tecan Infinite M200 plate reader with excitation and emission wavelengths at 485 and 528 nm, respectively. The protein content of the cell lysate was quantified using a BCA Protein Assay Kit. The intensity of fluorescence was standardized against total protein content, and measured in triplicate. The stock solutions of tested compounds were dissolved in DMSO with a final concentration of 50 mM and dilutions were made using uptake buffer. In our initial screening, the inhibition of the 12 compounds was evaluated at a concentration of $50 \,\mu\text{M}$ on OAT1 and OAT3. As shown, the majority of compounds showed marked inhibitory effects on OAT1 and OAT3. An inhibitor is defined as a compound that results in > 70% inhibition of 6-CF uptake. Concentration-dependent inhibition experiments were carried out on all selected inhibitors to determine IC₅₀ (50% inhibitory concentration) values on OAT1 and OAT3.

Statistical analysis. IC $_{50}$ values were summarized in Table 2 were estimated by non-linear regression analysis and expressed as mean \pm standard error of mean. Statistical analysis was performed using GraphPad Prism version 7.0. For the uptake experiments, data were analyzed with a two-tailed unpaired Student's t-test.

Well diffusion antimicrobial assay. Antimicrobial activities of four crude fractions (n-hexane fraction, dichloromethane fraction, n-butanol fraction and aqueous fraction) against sixteen bacteria and fungi were performed based on the method reported previously^{36,37} with slight modification using agar plates.

Briefly, nutrient ager (NA), brain heart infusion agar (BHIA) and potato dextrose agar (PDA), YM agar (YMA) were mixed in appropriate proportions with ultrapure water and autoclaved for 20 min at 121 °C. Approximately 15 mL aliquots of each medium were dispensed into petri plates and allowed to solidify, then incubated for 24 h to ensure sterility prior to use. Microbial suspensions (0.5 mL) at a density of 5×10^6 cfu/mL were inoculated onto the sterile media and allowed to stand for 10-20 min before wells were cut into the agar using a sterile pipet tip. A $10\,\mu$ L aliquot of a DMSO solution of plant extracts (1 mg/mL) or fractions (2, 0.2, 0.02 mg/mL) were pipetted into each well. Similarly, $10\,\mu$ L portions of DMSO and a solution of an appropriate antibiotic standard were applied to each plate as negative and positive controls, respectively. All treated plates were incubated for 24 h in incubators at appropriate temperatures. Zones of inhibition were measured in millimeters. The assays were performed in triplicate. Proper media, temperature and antibiotic standards for each microorganism are shown in Table 3.

Broth microdilution antifungal assay. Antifungal activity against *C. rugosa* (ATCC 10571) was measured as previously described using a broth microdilution method $^{36-38}$ with minor modifications. The assay was performed in sterile 96-well microtiter plates. Briefly, YM broth was prepared and autoclaved for 20 min at 121 °C, then allowed to cool to room temperature (25 °C). *C. rugosa* was dispersed in YM broth to an optical density of 1 UA at 600 nm (approximately 1×10^7 cfu/mL) using uninoculated broth as the blank. A 198 μ L aliquot of the resulting fungal suspension was dispensed into each well of a 96-well plate. Stock solutions of the twelve

ATCC No.	Species	Media	Temp/°C	Antibiotic			
Bacteria							
ATCC 12600	Staphylococcus aureus	Nutrient Agar	37	Penicillin			
ATCC 14990	Staphylococcus epidermis	Nutrient Agar	37	Ampicillin			
ATCC 25175	Streptococcus mutans	Brain Heart Infussion Agar	37	Ampicillin			
ATCC 13525	Pseudomonas fluorescens	Nutrient Agar	26	Ciprofloxacin			
ATCC 8043	Enterococcus hirae	Brain Heart Infussion Agar	37	Ampicillin/Cephalexin			
ATCC 25238	Moraxella (Branhamella) catarrhalis	Brain Heart Infussion Agar	37	Cephalexin			
ATCC 15692	Pseudomonas aeruginosa	Nutrient Agar	37	Ciprofloxacin			
ATCC 6633	Bacillus subtilis subsp. spizizenii	Brain Heart Infusion Agar	30	Vancomycin			
Fungi							
ATCC 76615	Candida albicans	YM Medium Agar	24	Amphotericin B			
ATCC 16888	Aspergillus niger	Potato Dextrose Agar	30	Amphotericin B			
ATCC 2601	Saccharomyces kudriavzevii	YM Medium Agar	30	Fluconazole			
ATCC 10106	Penicillium chrysogenum	Potato Dextrose Agar	25	Amphotericin B			
ATCC 22019	Candida parapsilosis	YM Medium Agar	24	Amphotericin B			
ATCC 10571	Candida rugosa	YM Medium Agar/Broth		Amphotericin B			
ATCC 750	Candida tropicalis	YM Medium Agar	25	Amphotericin B			
ATCC 34103	Rhizopus stolonifer	Potato Dextrose Agar	25	Amphotericin B			

Table 3. Microorganisms, media, incubation temperature, and antibiotic controls.

compounds were prepared in DMSO at a concentration of $50\,\mathrm{mM}$. The stock solutions were diluted to a range of final concentrations ($500-0.14\,\mu\mathrm{M}$) in YM broth such that the final DMSO concentration was no higher than 1%, v/v. A $2\,\mu\mathrm{L}$ aliquot of the resulting test samples was transferred to each well of the plate. Plates were incubated at $25\,^\circ\mathrm{C}$ for 24h prior to measurement of the optical density of each well at $600\,\mathrm{nm}$. Minimal inhibitory concentrations (MICs), defined as the lowest concentration required to visibly inhibit the fungal growth compared to the untreated control, were measured in triplicate. Amphotericin B was used as positive control.

Data Availability

The datasets generated in this study are available from the corresponding author on reasonable request.

References

- 1. Kosenkova, Y. S. et al. Riccardin C, a bisbibenzyl compound from Primula macrocalyx. Chem. Nat. Compd. 43, 712–713 (2007).
- 2. Kosenkova, Y. S. et al. Fatty-acid composition and secondary metabolites from slightly polar extracts of the aerial part of *Primula macrocalyx*. Chem. Nat. Compd. 44, 564–568 (2008).
- 3. Kosenkova, Y. S. et al. Seasonal dynamics of the accumulation of riccardin C in *Primula macrocalyx* Bge. *Khim. Interesakh Ustoich.* Razvit. 17, 507–511 (2009).
- Zakharov, A. M., Zakharova, O. I., Pakalns, D. & Sokol'skii, I. N. Triterpenoid glycoside from Primula macrocalyx. Khim. Prir. Soedin. 7, 672 (1971).
- Murav'ev, I. A. & Michnik, O. V. Chromatographic study of saponins from Primula macrocalyx Berg. Khromatogr. Elektroforeticheskie Metody Issled. Biol. Aktiv. Soedin. 73 (1976).
- Calis, I., Yuruker, A., Ruegger, H., Wright, A. D. & Sticher, O. Triterpene saponins from Primula veris subsp. macrocalyx and Primula elatior subsp. meyeri. J. Nat. Prod. 55, 1299–1306 (1992).
- 7. Zakharov, A. M., Pakaln, D. A., Zakharova, O. I. & Boryaev, K. I. Triterpenoid glycoside content of Primulaceae species of flora from Central Asia and the Caucasus. *Rastit. Resur.* 10, 375–379 (1974).
- 8. Gontar, E. M. Vitamin content in some species of the genus *Primula* of the Altai range. *Ekol. -Morfol. Biokhim. Osob. Polez. Rast. Dikorastushchei Flory Sib.* 262–264 (1970).
- 9. Latypova, G. M., Bubenchikova, V. N. & Kataev, V. A. The level of ursolic acid in the plants of the genus *Primula*. Farmatsiya (Moscow) 4, 21–24 (2015).
- Kosenkova, Y. S., Polovinka, M. P., Komarova, N. I., Korchagina, D. V. & Salakhutdinov, N. F. Preparation of ethers and esters of riccardin C, potential inhibitors of NO synthase. Khim. Interesakh Ustoich. Razvit. 17, 495–506 (2009).
- 11. Sekine, T., Cha, S. H. & Endou, H. The multispecific organic anion transporter (OAT) family. Pfluegers Arch. 440, 337-350 (2000).
- 12. Hagos, Y. & Wolff, N. A. Assessment of the role of renal organic anion transporters in drug-induced nephrotoxicity. *Toxins* 2, 2055–2082 (2010).
- 13. Nigam, S. K. et al. The organic anion transporter (OAT) family: a systems biology perspective. Physiol. Rev. 95, 83–123 (2015).
- 14. Lu, H. *et al.* Interactions of 172 plant extracts with human organic anion transporter 1 (SLC22A6) and 3 (SLC22A8): a study on herb-drug interactions. *PeerJ* 5, e3333 (2017).
- Vallon, V. et al. Organic anion transporter 3 contributes to the regulation of blood pressure. J. Am. Soc. Nephrol. 19, 1732–1740 (2008).
- 16. Khan, M. S. A. & Ahmad, I. Biofilm inhibition by *Cymbopogon citratus* and *Syzygium aromaticum* essential oils in the strains of *Candida albicans. J. Ethnopharmacol.* **140**, 416–423 (2012).
- 17. Kuwabara, H., Mouri, K., Otsuka, H., Kasai, R. & Yamasaki, K. Tricin from a *Malagasy Connaraceous* Plant with Potent Antihistaminic Activity. *J. Nat. Prod.* 66, 1273–1275 (2003).
- Iinuma, M., Matsuura, S. & Kusuda, K. Carbon-13 nuclear magnetic resonance (nmr) spectral studies on polysubstituted flavonoids.
 I. Carbon-13-nmr spectra of flavones. Chem. Pharm. Bull. 28, 708–716 (1980).
- 19. Park, Y. et al. NMR data of flavone derivatives and their anti-oxidative activity. Bull. Korean Chem. Soc. 27, 1537-1541 (2006).

- Budzianowski, J., Morozowska, M. & Wesolowska, M. Lipophilic flavones of Primula veris L. from field cultivation and in vitro cultures. Phytochemistry 66, 1033–1039 (2005).
- 21. Tanemossu, S. A. F. et al. Rare biscoumarin derivatives and flavonoids from Hypericum riparium. Phytochemistry 105, 171–177 (2014).
- 22. Chuang, D. W. et al. Synthesis of Flavones and γ -Benzopyranones Using Mild Sonogashira Coupling and 18-Crown-6 Ether Mediated 6-endo Cyclization. Eur. J. Org. Chem. 4533–4540 (2012).
- 23. Heneka, B. et al. A furanocoumarin and polymethoxylated flavonoids from the Yucatec Mayan plant Casimiroa tetrameria. Phytochemistry 66, 649-652 (2005).
- 24. Park, Y. et al. Spectral assignments and reference data: ¹H and ¹³C-NMR data of hydroxyflavone derivatives. Magn. Reson. Chem. 45, 674–679 (2007).
- 25. Wollenweber, E., Iinuma, M., Tanaka, T. & Mizuno, M. 5-Hydroxy-6,2'-dimethoxyflavone from *Primula denticulata*. *Phytochemistry* 29, 633–637 (1990).
- 26. Niraula, N. P., Bhattarai, S., Lee, N. R., Sohng, J. K. & Oh, T. J. Biotransformation of flavone by CYP105P2 from Streptomyces peucetius. J. Microbiol. Biotechnol. 22, 1059–1065 (2012).
- 27. Chun, H. K. et al. Biotransformation of flavone and flavanone by Streptomyces lividans cells carrying shuffled biphenyl dioxygenase genes. J. Mol. Catal. B: Enzym. 21, 113–121 (2003).
- Wang, X. et al. Isoflavones from Camphorosma lessingii Inhibit the Organic Anion Transporters OAT1 and OAT3. Planta Med. 85, 225–230 (2019).
- An, G., Wang, X. & Morris, M. E. Flavonoids are inhibitors of human organic anion transporter 1 (OAT1)-mediated transport. *Drug Metab Dispos* 42, 1357–1366 (2014).
- 30. Wu, T. et al. Apigenin, a novel candidate involving herb-drug interaction (HDI), interacts with organic anion transporter 1 (OAT1). Pharmacological Reports 69, 1254–1262 (2017).
- 31. Xu, F. et al. The inhibitory effects of the bioactive components isolated from Scutellaria baicalensis on the cellular uptake mediated by the essential solute carrier transporters. Journal of pharmaceutical sciences 102, 4205–4211 (2013).
- 32. Jiang, Y. et al. Caffeoylquinic acid derivatives rich extract from Gnaphalium pensylvanicum Willd. ameliorates hyperuricemia and acute gouty arthritis in animal model. BMC complementary and alternative medicine 17, 320–330 (2017).
- 33. Qiao, Y. et al. Biflavonoids from Juniperus oblonga inhibit organic anion transporter 3. Biochem. Biophys. Res. Commun. 509, 931–936 (2019).
- 34. Ma, L. et al. Interaction of five anthraquinones from rhubarb with human organic anion transporter 1 (SLC22A6) and 3 (SLC22A8) and drug-drug interaction in rats. J. Ethnopharmacol. 153, 864–871 (2014).
- 35. Wang, X. et al. Identification of Natural Products as Inhibitors of Human Organic Anion Transporters (OAT1 and OAT3) and Their Protective Effect on Mercury-Induced Toxicity. Toxicol Sci. 161, 321–334 (2018).
- 36. Saranya, D., Sekar, J. & Raj, G. A. Phytochemical analyses, antibacterial and antifungal activity of leaves from *Abutilon indicum* (L.) Sweet. *J. Chem. Pharm. Res.* 7, 524–530 (2015).
- 37. Chen, X. et al. Efficacy of a combination of nisin and p-Anisaldehyde against *Listeria* monocytogenes. Food Control 66, 100–106 (2016)
- 38. De Zoysa, G. H., Glossop, H. D. & Sarojini, V. Unexplored antifungal activity of linear battacin lipopeptides against planktonic and mature biofilms of *C. albicans. Eur. J. Med. Chem.* **146**, 344–353 (2018).

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Author Contributions

X.L. conceived and designed experiments, performed experiments, analyzed the data, wrote the paper, prepared figures and tables, reviewed drafts of the paper. X.W. and C.Y.L. contributed reagents/materials/analysis tools, and helped analyze the data. M.K., G.F. and D.A. contributed in collecting and identifying the plant materials. Y.C.Z. conceived and designed experiments, performed experiments, analyzed data, and reviewed drafts of the paper. R.P.B. conceived and designed experiments, performed experiments, analyzed data, and reviewed drafts of the paper.

Additional Information

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