

SCIENTIFIC REPORTS

OPEN

Imaging in Biologically-Relevant Environments with AFM Using Stiff qPlus Sensors

Korbinian Pürckhauer¹, Alfred J. Weymouth¹, Katharina Pfeffer¹, Lars Kullmann², Estefania Mulvihill³, Michael P. Krahn^{2,4}, Daniel J. Müller³ & Franz J. Giessibl¹

High-resolution imaging of soft biological samples with atomic force microscopy (AFM) is challenging because they must be imaged with small forces to prevent deformation. Typically, AFM of those samples is performed with soft silicon cantilevers ($k \approx 0.1\text{--}10\text{ N/m}$) and optical detection in a liquid environment. We set up a new microscope that uses a stiff qPlus sensor ($k \geq 1\text{ kN/m}$). Several complex biologically-relevant solutions are non-transparent, and even change their optical properties over time, such as the cell culture medium we used. While this would be problematic for AFM setups with optical detection, it is no problem for our qPlus setup which uses electrical detection. The high stiffness of the qPlus sensor allows us to use small amplitudes in frequency-modulation mode and obtain high Q factors even in liquid. The samples are immersed in solution in a liquid cell and long tips are used, with only the tip apex submerged. We discuss the noise terms and compare the minimal detectable signal to that of soft cantilevers. Atomic resolution of muscovite mica was achieved in various liquids: H_2O , Tris buffer and a cell culture medium. We show images of lipid membranes in which the individual head groups are resolved.

Atomic force microscopy (AFM) imaging at the atomic scale in biologically-relevant conditions is challenging because of the complex interaction between tip and sample. Biological samples need to be imaged with minimal interaction forces because if the force is larger than about 100 pN the sensitive sample might get damaged^{1,2}. Typically, AFM imaging of these sample systems is performed in tapping mode^{3–5} in liquid environments by scanning the sample with a soft silicon cantilever (stiffness $k \approx 0.1\text{--}10\text{ N/m}$) that is completely immersed in liquid⁶. In tapping mode, soft cantilevers are usually used to scan biological samples as one obtains a large signal and reversible or irreversible deformations of the sample are reduced^{7–9}. In this work we demonstrate that frequency modulation AFM (FM-AFM) with stiffer sensors ($k \geq 1\text{ kN/m}$), a technique which was used to obtain submolecular resolution on organic molecules in ultra-high vacuum (UHV) at low temperature¹⁰ and even room temperature¹¹, facilitate imaging also of biological samples with high spatial resolution. Frequency modulation AFM (FM-AFM)^{12–15} is a technique in which the sensor is driven to oscillate at a given amplitude and the measured variable is the frequency shift Δf which is itself a measure of the force gradient k_{ts} between tip and sample. The key question in optimizing imaging is not the stiffness of the sensor but the detectable force gradient that is limited by instrumental noise.

In FM-AFM, noise is characterized by the minimal detectable force gradient $\delta\langle k_{ts} \rangle_{\min}$ ¹⁶. It is a function of the quality factor of the cantilever, Q , which relates the energy stored in the cantilever to the energy lost per cycle. Operation in liquid lowers the Q value due to additional damping. Soft cantilevers completely immersed in liquid (see Fig. 1a) have a very low Q value, around 1–30^{17,18}. A solution to maintain high Q values is to use a stiffer sensor like the self-sensing qPlus sensors^{19,20} ($k \geq 1\text{ kN/m}$), shown in Fig. 1b).

In most cases, an AFM sensor is excited mechanically (also called acoustic excitation) with a vibrating piezo stack. The frequency response of the oscillation with respect to the excitation is a single excitation peak at the resonance frequency and a smooth monotonic phase curve. The monotonic phase response is exploited when performing FM-AFM as it allows a Phase Locked Loop (PLL) to follow the current resonance frequency. It has been observed that driving a soft cantilever mechanically in liquid causes many unwanted new peaks near the

¹University of Regensburg, Institute of Experimental and Applied Physics, Regensburg, 93053, Germany. ²University of Regensburg, Institute for Molecular and Cellular Anatomy, Regensburg, 93053, Germany. ³Eidgenössische Technische Hochschule (ETH) Zürich, Department of Biosystems Science and Engineering, Basel, 4058, Switzerland.

⁴University Hospital of Münster, Internal Medicine D, Münster, 48149, Germany. Correspondence and requests for materials should be addressed to K. Pürckhauer (email: korbinian.puerckhauer@ur.de)

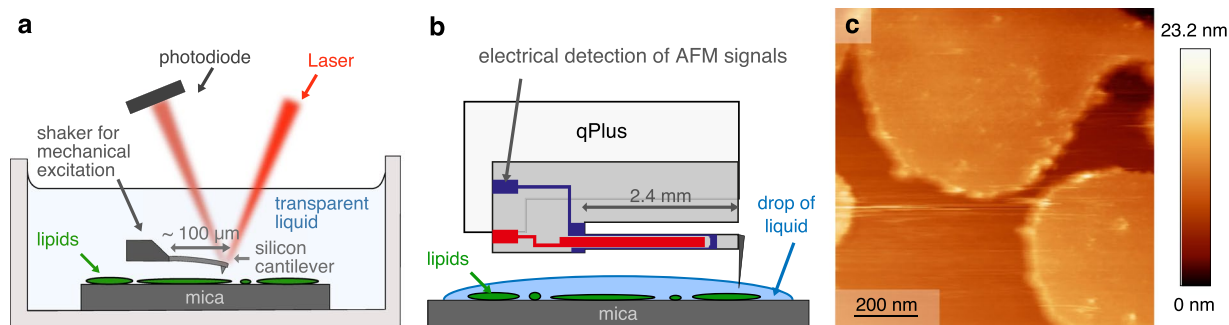


Figure 1. Comparison of AFM setups for imaging biological samples with silicon cantilevers and qPlus AFMs. **(a)** Schematic drawing of a typical silicon cantilever AFM setup which is completely immersed in a transparent liquid. **(b)** Schematic drawing of the qPlus AFM setup for measurements in a drop of liquid. **(c)** FM-AFM image of lipid membrane islands (extracted from *Halobacterium salinarum*) performed in z-feedback with $f_0 = 18.92$ kHz, $k = 3500$ N/m, $A = 150$ pm and $\Delta f = +20$ Hz. Buffer solution: 300 mM KCl and 20 mM Tris-HCl (pH 7.5).

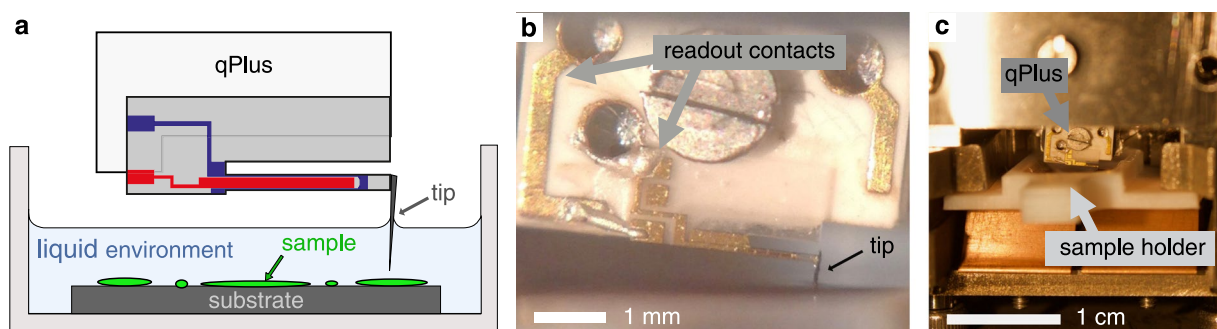


Figure 2. qPlus AFM setup for liquid measurements. **(a)** Schematic drawing of the qPlus AFM setup for measurements in a liquid bath. The sample is immersed in liquid and only the tip apex is submerged. **(b)** View on a qPlus sensor (equipped with a long sapphire tip) oscillating in liquid. **(c)** Image of the AFM head including the sample holder with integrated liquid cell.

resonance frequency, an effect referred to as the “forest of peaks”²¹. In addition, the phase response is no longer monotonic, which prevents stable amplitude feedback in FM-AFM. To avoid the “forest of peaks” effect and a non-monotonic phase either a direct excitation method, such as magnetic excitation^{22,23} or photothermal excitation^{24,25}, or a stiffer sensor is required.

The success of FM-AFM with stiff sensors and piezoelectric detection in vacuum^{26–28} demonstrates the advantages of using small amplitudes (< 100 pm) because they yield higher sensitivity to short range forces²⁹. These small amplitudes on the order of the decay length of the short-range interaction forces cannot be realized with a soft cantilever because of the well-known “jump-to-contact” problem²⁰.

Scanning in ambient conditions where the sample is covered with a hydration layer³⁰ was a first step towards imaging in a biologically-relevant environment. Optimizing imaging with the qPlus sensor in ambient conditions led to reproducible images with atomic resolution on various sample systems^{31–33} and sensor oscillation modes³⁴. In order to image with a qPlus sensor in liquid environments, specifically with conducting liquids, electric detection of the AFM signal demands that either only the tip is submerged or the sensor is covered by an insulation layer. As the second method could lower the Q -value, we pursued the first option. Previous measurements were performed with qPlus sensors in aqueous solutions done with adding just a drop of liquid on the sample¹⁶. Ichii *et al.*³⁵ showed that imaging of muscovite mica at the atomic scale was possible with qPlus sensors in such a drop. In Fig. 1c we show large lipid bilayer patches that are non-destructively imaged in a drop of buffer solution (~ 20 μl) with qPlus sensors using FM-AFM. As the small liquid volume evaporated within about 15 min., high resolution imaging was impractical.

Our goal is to achieve high spatial resolution of soft biological samples. We therefore moved from this setup which allowed us to demonstrate proof of principle to a more controlled environment that would allow us to achieve high spatial resolution. A sample holder with an integrated liquid bath ($A_{\text{cell}} = 85$ mm²) was created (Fig. 2). The new sample holder allows us to use up to 420 μl of solution. As a consequence, very long tips (~ 500 – 1000 μm) had to be glued to the sensor’s oscillating prong. The qPlus sensor with the tip submerged in liquid is shown in Fig. 2b and the AFM setup including the liquid cell is depicted in Fig. 2c. This setup allows us to use any kind of tip material. For this study, we used sapphire which is very hard and chemically inert. With this new setup,

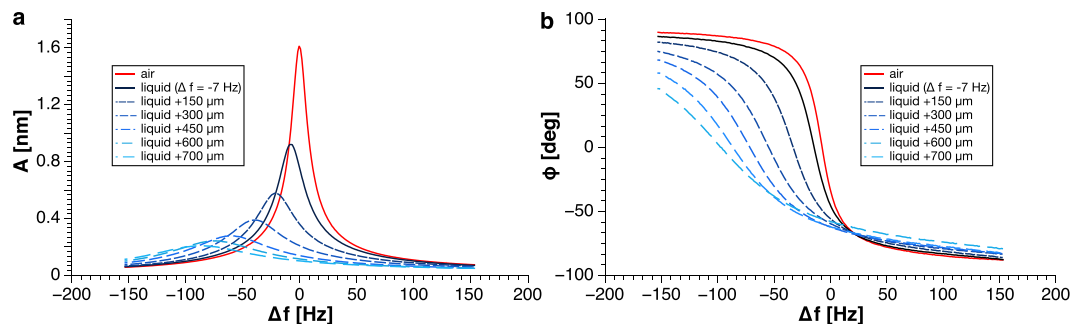


Figure 3. Resonance behaviour of qPlus sensor as a function of penetration depth. (a) Resonance curves in different penetration depths shown in a blue color-code and the reference one in air shown in red. The associated phases are depicted in (b) with the same color scheme and obviously it stays monotonic for all the time. The representative sensor had a resonance frequency of $f_0 = 23.41$ kHz and stiffness $k = 1800$ N/m in air.

we are able to image biological samples non-destructively. We show images of a lipid layer extracted directly from a biological sample in which the individual lipid head groups can be observed.

Results and Discussion

We first evaluated the influence of the penetration depth of the tip in liquid on the qPlus resonance curve. To do so, we recorded a resonance curve of the mechanically excited sensor in air and approached to the water layer. The distance where the tip starts penetrating the liquid can be seen by a shift in resonance frequency. To ensure that the tip is vibrating in liquid we defined $\Delta f = -7$ Hz as our first reference point in liquid and recorded resonance spectra at various penetration depths by approaching the qPlus sensor to the support using the coarse motor. The drive amplitude was kept constant for all data. The data shown in Fig. 3a display a continuous decrease of the oscillation amplitude at resonance starting from about 1.6 nm in air until it levels off at around 12.5% of its original value. The largest achievable penetration depth is limited by the length of the tip (representative sensor had a 700 μm long sapphire tip). It is important that the phase stays monotonic for all penetration depths as verified in Fig. 3b.

In FM-AFM, noise is characterized by the minimal detectable force gradient $\delta\langle k_{ts} \rangle_{\min}$, composed by detector $\delta k_{ts_{\text{det}}}$, thermal $\delta k_{ts_{\text{th}}}$ and oscillator noise $\delta k_{ts_{\text{osc}}}$ ³⁶. As these noise sources are statistically independent, $\delta\langle k_{ts} \rangle_{\min}$ can be written as¹⁶

$$\delta\langle k_{ts} \rangle_{\min} = \sqrt{\delta k_{ts_{\text{th}}}^2 + \delta k_{ts_{\text{det}}}^2 + \delta k_{ts_{\text{osc}}}^2} \quad (1)$$

The individual noise terms are given by¹⁶

$$\delta k_{ts_{\text{th}}} = \sqrt{\frac{4kk_B T B}{\pi f_0 A^2 Q}} \propto \sqrt{\frac{1}{Q}} \quad (2)$$

$$\delta k_{ts_{\text{det}}} = \sqrt{\frac{8}{3}} \frac{k n_q B^{\frac{3}{2}}}{f_0 A} \quad (3)$$

$$\delta k_{ts_{\text{osc}}} = \frac{k n_q \sqrt{2B}}{A Q} \propto \frac{1}{Q} \quad (4)$$

The above formulae include quality factor Q , the stiffness k , the Boltzmann constant k_B , the temperature T , the bandwidth B , the resonance frequency f_0 , the amplitude A and the deflection noise density n_q . Thermal and oscillator noise depend on the quality factor Q , which is related to the inverse loss of energy per cycle ΔE_{cycle} relative to the stored energy in the resonator E by $Q = E / (\Delta E_{\text{cycle}} \cdot 2\pi)$.

We calculated the Q factor from the slopes of the penetration depth dependent phases at f_0 . The depth dependence of the Q factor is plotted in Fig. 4a with the value in air marked in red as reference. There was a continuous decrease from 1600 in air until it leveled off at approximately 300. Q values up to 1000 in liquid are possible if the penetration depth is small enough. The data were reproducible for a variety of stiffnesses (ranging from 1800 to 8300 N/m), different tip geometries and even for thermal excitation of the sensor.

We also characterized the evaporation rate of water from the cell. The rate was 0.39 $\mu\text{l}/\text{min}$ for three different cells at a humidity of 40%. Assuming a measurement time of 30 min for a large overview scan, this results in a change of 138 μl . For a penetration depth of 600 μm , the Q factor will change from 300 to 350.

Figure 4b shows a calculation of the individual noise terms $\delta k_{ts_{\text{th}}}$, $\delta k_{ts_{\text{det}}}$ and $\delta k_{ts_{\text{osc}}}$ from Eqs (2–4) for a qPlus sensor as a function of Q . The parameters of a new custom qPlus sensor with higher f_0 used for the calculation can be found in Table 1. The curves show that for the given setup the thermal noise $\delta k_{ts_{\text{th}}}$ dominates. Oscillator noise

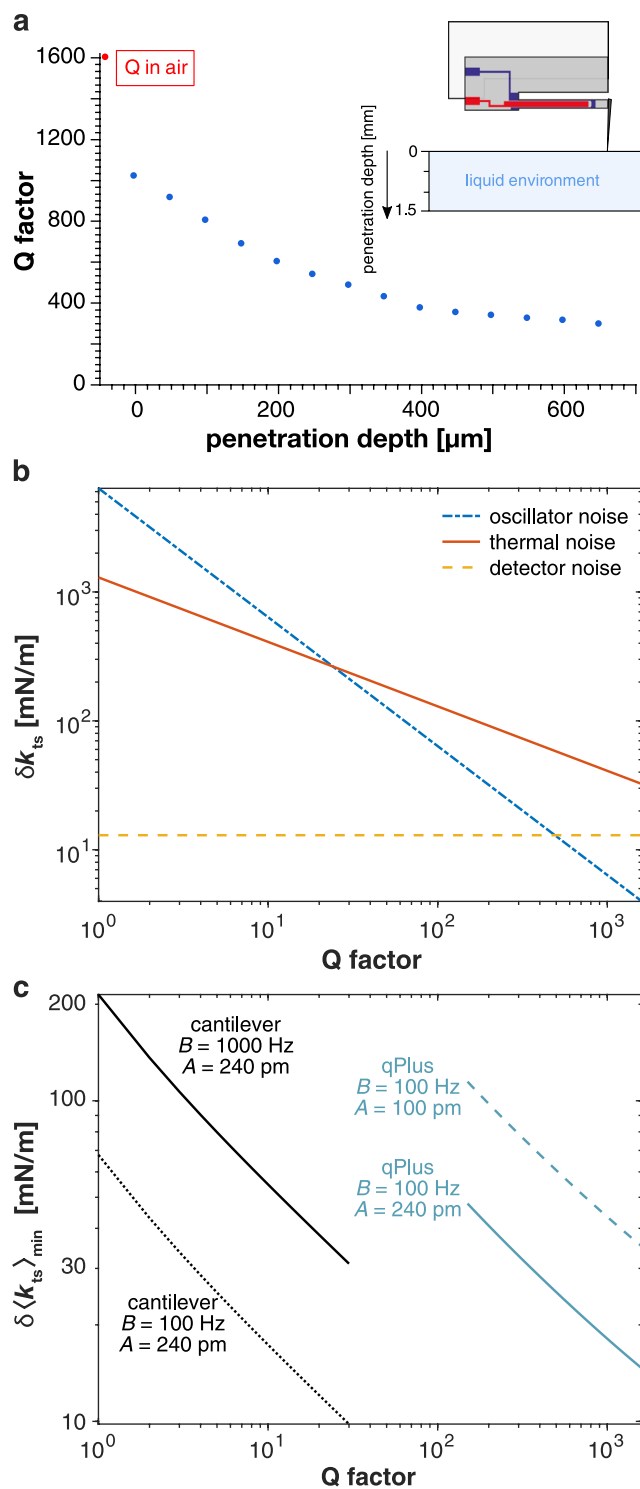


Figure 4. Analysing Q dependence of FM-AFM noise. **(a)** Dependence of the Q factor on the tip's penetration depth together with a schematic drawing. The reference Q value in air (1597) is marked with a red dot. **(b)** Thermal noise $\delta k_{ts,th}$ (solid, red), detector noise $\delta k_{ts,det}$ (dashed, yellow) and oscillator noise $\delta k_{ts,osc}$ (dashed-dotted, blue) of qPlus sensor plotted over Q . **(c)** Minimal detectable force gradient $\delta \langle k_{ts} \rangle_{\min}$ plotted vs. Q factor of silicon cantilever ($A = 240$ pm, colored black) with both a common $B = 1$ kHz and a small $B = 100$ Hz bandwidth compared to qPlus (colored blue, $B = 100$ Hz, $A = 100$ pm, dashed, and $A = 240$ pm, solid).

$\delta k_{ts,osc}$ becomes significant for Q values below 200 but is not as important as for soft silicon cantilevers where Q is in the range of 1 up to 30. The detector noise $\delta k_{ts,det}$ is independent of Q and therefore is dominated by the other noise terms for Q values below 200. The noise terms in Eqs (2–4) show that sensor properties k and f_0 are also

Parameter	qPlus	Cantilever
f_0 [kHz]	56.59	136
k [$\frac{\text{N}}{\text{m}}$]	1800	42
A [pm]	100	240
n_q [fm/ $\sqrt{\text{Hz}}$]	25	17
B [Hz]	100	1000
Q	150–1600	1–30

Table 1. Comparison of sensor and scan parameters which are influencing the total minimal detectable force gradient. The values for the cantilever are taken from Fukuma *et al.*¹⁴ and the ones for qPlus are typical specifications of our latest custom qPlus sensors which have a higher resonance frequency (as described in⁵⁶).

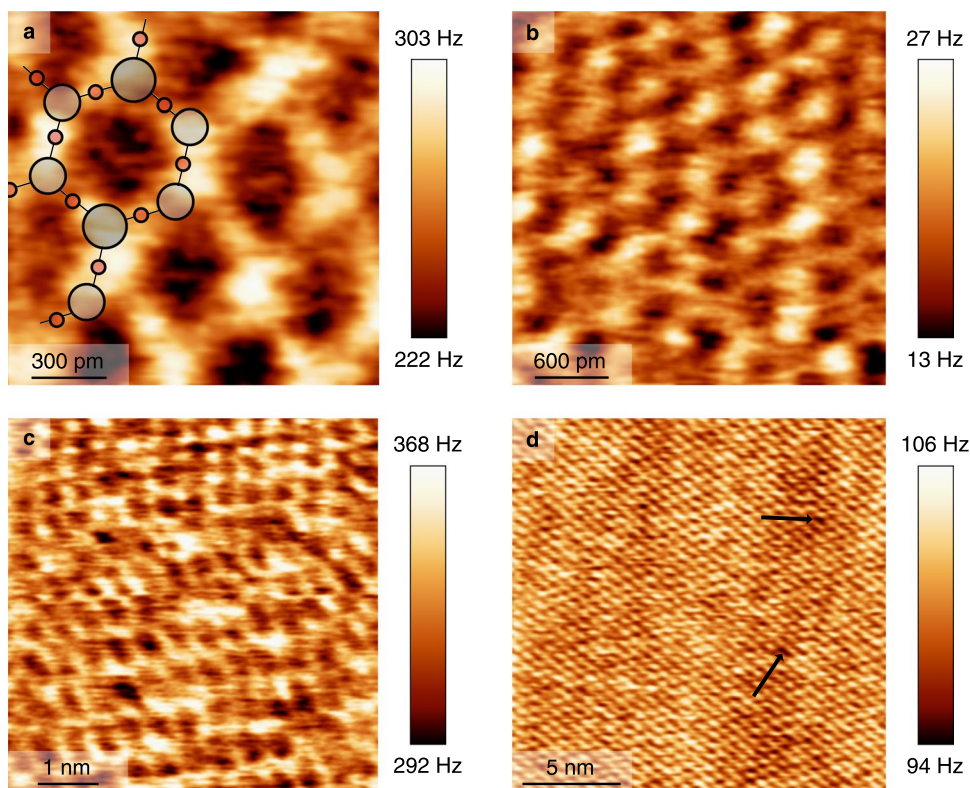


Figure 5. FM-AFM images in quasi-constant height mode (defined in the text) of muscovite mica in different environments. (a) Image taken in ambient conditions with $\Delta f = +257$ Hz and $A = 78$ pm. A possible lattice configuration of muscovite mica is overlaid. (b) Image taken in H_2O with $\Delta f = +20$ Hz and $A = 210$ pm. (c) Image taken in HAM culture medium with $\Delta f = +330$ Hz and $A = 112$ pm. (d) Image taken in Tris-HCl buffer with $\Delta f = +100$ Hz and $A = 100$ pm.

important in addition to Q . Silicon cantilevers have a higher f_0 and lower k than qPlus sensors which leads to an advantage in k/f_0 ratio. In liquid environments, the total force gradient noise is heavily influenced by the low Q . We compared the total noise in liquid to that from a standard cantilever by plotting the qPlus total noise over the Q range of 150 to 1600 and the cantilever noise over Q from 1 to 30 in Fig. 4c. The parameters for a cantilever taken from Fukuma *et al.*¹⁴ and a qPlus sensor used for the calculations can be found in Table 1. As the detector noise of qPlus is larger, the scanning speed has to be significantly slower to obtain a similar signal-to-noise ratio since the detector noise increases proportional to $B^{3/2}$ in FM-AFM as shown in Eq. (3). Therefore tapping mode with soft silicon cantilevers is typically still faster than FM-AFM imaging using the qPlus sensor.

Imaging was performed in various environments to evaluate the possibility of imaging at the atomic scale in biologically relevant environments. Our measurements were performed on muscovite mica $\text{KAl}_2(\text{OH,F})_2(\text{AlSi}_3\text{O}_{10})$ which is a well known substrate for immobilizing biological samples like lipids^{37–39} or membrane proteins^{40–42}. As biological environments we used ultrapure water, Tris buffer and a non-transparent cell culture medium. As a reference, Fig. 5a shows atomic resolution with overlaid lattice in air. The periodicity was found to be 0.52 nm which fits to the literature value⁴³.

For the measurements in ultra-pure (18 M Ω /cm) water we added 60 μl onto the freshly cleaved muscovite mica. The sensor's resonance frequency was determined to be 23.3 kHz in air and the Q value was 1777 in air

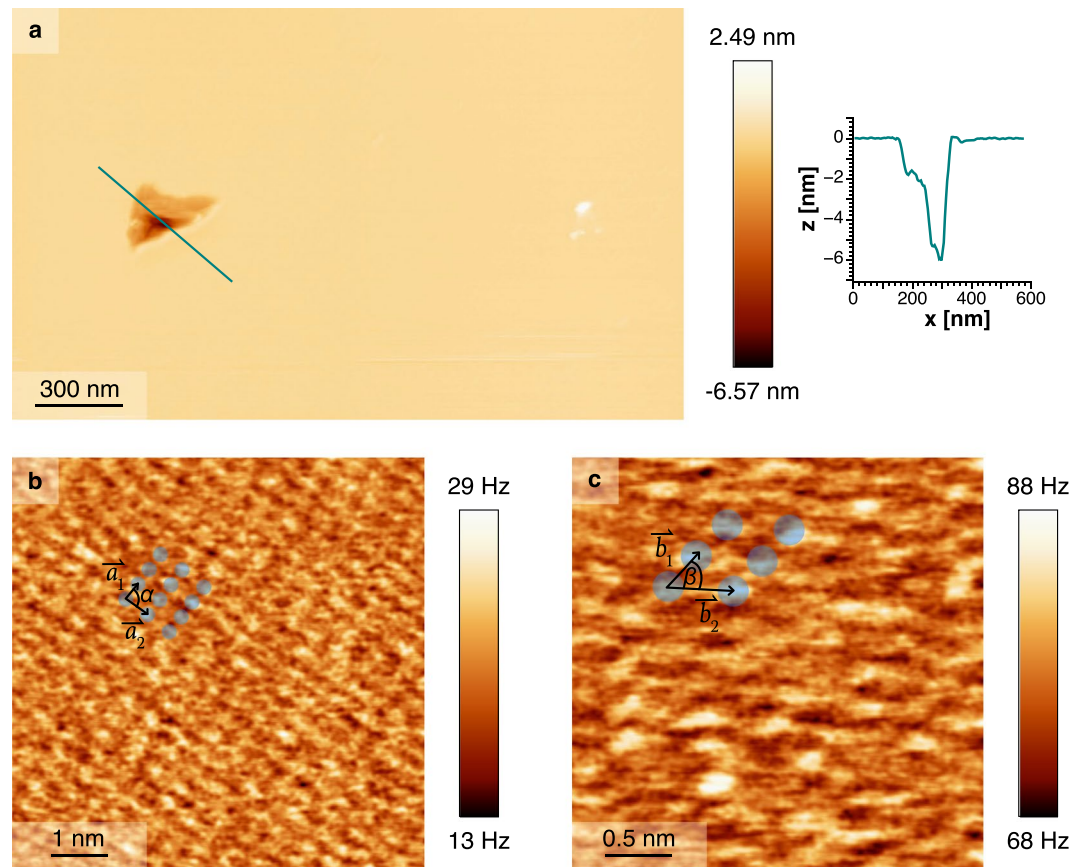


Figure 6. FM-AFM images of a lipid bilayer adsorbed on mica and measured in buffer solution. (a) Topography overview scan of the lipid bilayer with a large artificially created defect and an additional line profile along the cyan colored line is included. The imaging parameters were $A = 200$ pm and $\Delta f = +25$ Hz. (b) High resolution image of the lipid heads in Δf feedback taken in quasi-constant height mode with $A = 100$ pm and $\Delta f = +20$ Hz. The sensor's resonance frequency f_0 was 15.57 kHz. The rectangular lattice ($\alpha = 90^\circ$) is indicated by the headgroup positions marked by blue dots with lattice vectors \vec{a}_1 and \vec{a}_2 . (c) Image of the lipid heads in Δf feedback taken in quasi-constant height mode with $A = 106$ pm and $\Delta f = +76$ Hz. The sensor's resonance frequency f_0 was 15.72 kHz. The headgroup positions marked by blue dots indicate an oblique lattice. The angle β between the lattice vectors \vec{b}_1 and \vec{b}_2 was 64° .

($k = 1800$ N/m). The Q value in liquid was 714 near sample contact. The frequency shift image with atomic resolution depicted in Fig. 5b was recorded with an amplitude of 210 pm (best signal-to-noise ratio in this case) in quasi-constant height mode. This means that the feedback loop of the height control was set very slow to merely compensate the drift.

We then acquired images in a cell culture medium called HAM (see materials and methods) as shown in Fig. 5c. For imaging in cell culture medium we used a qPlus sensor with $k = 1800$ N/m, $f_0 = 56.59$ kHz and a Q of 1954 in air. We scanned in 100 μ l cell culture medium and Q decreased to 147 at close distance to the sample in liquid. The frequency shift image Fig. 5c, where again the honeycomb lattice could be imaged, was recorded with an amplitude of 112 pm in quasi-constant height mode.

Finally we added 200 μ l 10 mM Tris-HCl buffer solution (pH 8.0) onto freshly cleaved muscovite mica and used the same qPlus as for the H_2O measurement. The Q value was 275 during the measurement. The 20 nm by 20 nm frame pictured in Fig. 5d was measured in quasi-constant height with 100 pm amplitude and shows atomic resolution with naturally occurring defects marked by black arrows.

The new setup allowed us to study phospholipid bilayers with longer measurement time (~ 3 – 4 h at 50% humidity). Therefore, we adsorbed liposomes (made from egg-PC) onto a freshly cleaved mica sheet which formed lipid bilayers^{37,38,44,45}. Imaging was performed in approximately 250 μ l buffer solution. Figure 6a shows an overview scan where we artificially created a hole into the lipid bilayer by approaching 10 nm from a $\Delta f = +25$ Hz setpoint towards the sample and scratching along a 100×100 nm large frame. By analyzing the line profile over the scratched hole (Fig. 6a), one can see that the hole is ~ 5.3 nm deep which agrees to the height of a lipid bilayer⁴⁶. Furthermore, there is a ~ 1.5 nm step in the line profile visible that might refer to lipid headgroups having been bent over by the scratching. It is important to note that imaging with the stiff qPlus sensor was non-destructive since the image shows no further manipulated lipid layer than the artificially created defect.

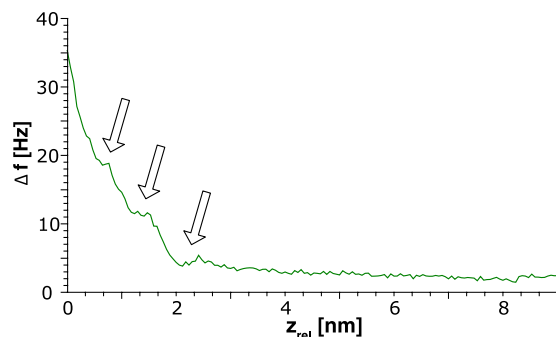


Figure 7. Frequency shift-distance spectrum on Egg-PC lipid bilayer. Arrows indicate peaks in the spectrum, discussed in detail in the text.

In Fig. 6b we scanned the lipids in a quasi-constant height mode to image the lipid headgroups with high spatial resolution. The image presents a rectangular lattice with lattice vectors \vec{a}_1 and \vec{a}_2 having magnitudes 0.34 nm and 0.45 nm shown in Fig. 6b. When scanning a second sample (liposomes extracted from the same sample) we observed an oblique lattice with lattice vectors \vec{b}_1 and \vec{b}_2 having magnitudes 0.26 nm and 0.55 nm and angle β between them of 64° , shown in Fig. 6c. The two structures can be explained by the various types of lipids (including different fatty acid distributions) which are present in natural lipid samples of Egg-PC. The area per lipid head can be calculated with the measured lattice vectors and was found to be 0.15 nm^2 for the rectangular lattice and 0.13 nm^2 for the oblique lattice.

These areal densities agree with previously quoted values (see, e.g. Chapter 10 in ref.⁴⁷), however they are significantly smaller than those values quoted by X-Ray diffraction studies^{46,48}. This might be because AFM can identify the individual domains of the lipid reconstruction, whereas X-Ray diffraction yields only an ensemble average.

As well as images of the Egg-PC lipid bilayer, we collected $\Delta f(z)$ spectroscopy. One such spectrum is shown in Fig. 7, in which we can observe plateaus in the $\Delta f(z)$ curve. Asakawa and co-workers acquired FM-AFM data of a model bilayer system (dipalmitoylphosphatidylcholine) in which they were able to observe similar features⁴⁵. They associated these features with ordered hydration layers. Previous work with the qPlus sensor has demonstrated that hydration layers can be observed above the calcite surface³³.

Conclusion

We have demonstrated that it is possible to image soft biological samples in liquid using stiff qPlus sensors. The qPlus sensors can be equipped with various kinds of tips such as metal tips that are used in tip enhanced Raman spectroscopy (TERS)^{49,50} and scanning near field optical microscopy (SNOM)^{51,52}. Furthermore FM-AFM with a PLL works without any further system adjustments due to the monotonic phase around f_0 in air and liquid. We showed that stiffer sensors like the qPlus can maintain high Q values greater than 150 in liquid. The Q factor plays a role for the minimal detectable force gradient $\delta\langle k_{ts} \rangle_{\min}$ and we note that silicon cantilevers suffer more from a liquid environment than qPlus sensors do. The comparison of $\delta\langle k_{ts} \rangle_{\min}$ for the two sensor types showed comparable values with the qPlus sensors. Challenges remain in improving imaging speed. With our liquid qPlus setup, atomic resolution on mica was achieved with amplitudes around one Angstrom in a variety of liquids including a non-transparent (pink) cell culture medium. These small amplitudes, with their high sensitivity to short-range forces, have been used to acquire images in vacuum environments with unprecedented spatial resolution. We hope that our future work will exploit this advantage and allow us to finally observe complex biological samples with atomic resolution in liquid environments. The development of a liquid bath allowed us to measure in a more controlled environment with longer measurement time. With it, we were able to image the lipid heads of a lipid bilayer adsorbed onto mica.

Materials and Methods

AFM setup. Our setup is a custom-design AFM head based on the Pan design⁵³ with a vertical approach and small mechanical loop for high stability and low thermal drift. Moreover the electrical wiring of the microscope was optimized including the use of a better operational amplifier design⁵⁴ to lower the deflection noise density n_f of the system from $35 \text{ fm}/\sqrt{\text{Hz}}$ (ref.¹⁶) down to $25 \text{ fm}/\sqrt{\text{Hz}}$. Imaging was performed with qPlus sensors equipped with sharp sapphire tips which were created by smashing a sapphire crystal with a hammer and gluing an appropriate splinter to the end of the prong.

Samples. For imaging we glued a thin muscovite mica (V1 grade purchased from Plano GmbH) disk to the bottom of the liquid cell, which was made out of Teflon, and cleaved it with mechanical exfoliation. Our results show that this leads to homogenous flat surfaces. For measurements in liquids we filled the liquid bath with the appropriate solution right after cleaving.

The cell culture medium we used was HAM medium, which is non-transparent (pink color). It contains salts (NaHCO_3 , CaSO_4 , Na_2HPO_4 , CaCl_2 , KCl , NaCl), a mineral solution, Resazurin, Na_2S and a small fraction of other solutes.

For Fig. 1b, lipid membranes were extracted together with purple membranes from *Halobacterium salinarum*⁵⁵. For adsorption, 2 μ l of the extracted lipid mixture was diluted in 120 μ l adsorption buffer (300 mM NaCl, 20 mM Tris, pH 7.5) and finally a drop (~20 μ l) of this solution was incubated on muscovite mica for 30 minutes. For imaging, the adsorption buffer was exchanged with imaging buffer (150 mM NaCl, 20 mM Tris, pH 7.5) and the sample was washed five times to remove unattached lipids.

For Fig. 6, liposomes were prepared in adsorption buffer (300 mM NaCl, 25 mM HEPES, pH 7.4) by extruding 25 mg/ml lipids (L- α -phosphatidylcholine from egg yolk, Egg-PC #840051, Avanti Polar Lipids) through a 0.1 μ m polycarbonate membrane using a Mini-Extruder (Avanti Polar Lipids, Inc.). For adsorption, liposomes were diluted to 25 μ g/ml in adsorption buffer and incubated on freshly cleaved mica for 30 min. After this, the adsorption buffer was exchanged with imaging buffer (150 mM NaCl, 25 mM HEPES, pH 7.4) and washed five times to remove unattached lipids.

Data availability. The datasets generated during and analysed during the current study are available from the corresponding author on reasonable request.

References

- Engel, A. & Müller, D. J. Observing single biomolecules at work with the atomic force microscope. *Nature Structural & Molecular Biology* **7**, 715–718 (2000).
- Fotiadis, D., Scheuring, S., Müller, S. A., Engel, A. & Müller, D. J. Imaging and manipulation of biological structures with the AFM. *Micron* **33**, 385–397 (2002).
- Hansma, P. *et al.* Tapping mode atomic force microscopy in liquids. *Applied Physics Letters* **64**, 1738–1740 (1994).
- Leung, C. *et al.* Atomic force microscopy with nanoscale cantilevers resolves different structural conformations of the DNA double helix. *Nano Letters* **12**, 3846–3850 (2012).
- Zhong, Q., Inniss, D., Kjoller, K. & Elings, V. Fractured polymer/silica fiber surface studied by tapping mode atomic force microscopy. *Surface Science* **290**, L688–L692 (1993).
- Baró, A. M. & Reifenberger, R. G. *Atomic force microscopy in liquid: biological applications* (John Wiley & Sons, 2012).
- Santos, N. C. & Castanho, M. A. An overview of the biophysical applications of atomic force microscopy. *Biophysical Chemistry* **107**, 133–149 (2004).
- Dufréne, Y. F. *et al.* Imaging modes of atomic force microscopy for application in molecular and cell biology. *Nature Nanotechnology* **12**, 295–307 (2017).
- Hörber, J. & Miles, M. Scanning probe evolution in biology. *Science* **302**, 1002–1005 (2003).
- Gross, L., Mohn, F., Moll, N., Liljeroth, P. & Meyer, G. The chemical structure of a molecule resolved by atomic force microscopy. *Science* **325**, 1110–1114 (2009).
- Huber, F. *et al.* Intramolecular force contrast and dynamic current-distance measurements at room temperature. *Physical Review Letters* **115**, 066101 (2015).
- Albrecht, T., Grütter, P., Horne, D. & Rugar, D. Frequency modulation detection using high-Q cantilevers for enhanced force microscope sensitivity. *Journal of Applied Physics* **69**, 668–673 (1991).
- Giessibl, F. J. & Bielefeldt, H. Physical interpretation of frequency-modulation atomic force microscopy. *Physical Review B* **61**, 9968 (2000).
- Fukuma, T., Kobayashi, K., Matsushige, K. & Yamada, H. True atomic resolution in liquid by frequency-modulation atomic force microscopy. *Applied Physics Letters* **87**, 034101 (2005).
- Ido, S. *et al.* Beyond the helix pitch: direct visualization of native DNA in aqueous solution. *ACS Nano* **7**, 1817–1822 (2013).
- Wutscher, E. & Giessibl, F. J. Atomic force microscopy at ambient and liquid conditions with stiff sensors and small amplitudes. *Review of Scientific Instruments* **82**, 093703 (2011).
- Humphris, A., Tamayo, J. & Miles, M. Active quality factor control in liquids for force spectroscopy. *Langmuir* **16**, 7891–7894 (2000).
- Tamayo, J., Humphris, A. D. & Miles, M. J. Piconewton regime dynamic force microscopy in liquid. *Applied Physics Letters* **77**, 582–584 (2000).
- Giessibl, F. J. High-speed force sensor for force microscopy and profilometry utilizing a quartz tuning fork. *Applied Physics Letters* **73**, 3956–3958 (1998).
- Giessibl, F. J. Advances in atomic force microscopy. *Reviews of Modern Physics* **75**, 949 (2003).
- Schäffer, T., Cleveland, J., Ohnesorge, F., Walters, D. & Hansma, P. Studies of vibrating atomic force microscope cantilevers in liquid. *Journal of Applied Physics* **80**, 3622–3627 (1996).
- Han, W., Lindsay, S. & Jing, T. A magnetically driven oscillating probe microscope for operation in liquids. *Applied Physics Letters* **69**, 4111–4113 (1996).
- Revenko, I. & Proksch, R. Magnetic and acoustic tapping mode microscopy of liquid phase phospholipid bilayers and DNA molecules. *Journal of Applied Physics* **87**, 526–533 (2000).
- Ratcliff, G. C. & Erie, D. A. & Superfine, R. Photothermal modulation for oscillating mode atomic force microscopy in solution. *Applied Physics Letters* **72**, 1911–1913 (1998).
- Labuda, A. *et al.* Comparison of photothermal and piezoacoustic excitation methods for frequency and phase modulation atomic force microscopy in liquid environments. *AIP Advances* **1**, 022136 (2011).
- Giessibl, F. J. Atomic resolution on si (111)-(7 \times 7) by noncontact atomic force microscopy with a force sensor based on a quartz tuning fork. *Applied Physics Letters* **76**, 1470–1472 (2000).
- Giessibl, F. J., Hembacher, S., Bielefeldt, H. & Mannhart, J. Subatomic features on the silicon (111)-(7 \times 7) surface observed by atomic force microscopy. *Science* **289**, 422–425 (2000).
- Gross, L. *et al.* Measuring the charge state of an adatom with noncontact atomic force microscopy. *Science* **324**, 1428–1431 (2009).
- Giessibl, F. J., Bielefeldt, H., Hembacher, S. & Mannhart, J. Calculation of the optimal imaging parameters for frequency modulation atomic force microscopy. *Applied Surface Science* **140**, 352–357 (1999).
- Santos, S. & Verdaguer, A. Imaging water thin films in ambient conditions using atomic force microscopy. *Materials* **9**, 182 (2016).
- Wastl, D. S., Weymouth, A. J. & Giessibl, F. J. Optimizing atomic resolution of force microscopy in ambient conditions. *Physical Review B* **87**, 245415 (2013).
- Wastl, D. S., Weymouth, A. J. & Giessibl, F. J. Atomically resolved graphitic surfaces in air by atomic force microscopy. *ACS Nano* **8**, 5233–5239 (2014).
- Wastl, D. S., Judmann, M., Weymouth, A. J. & Giessibl, F. J. Atomic resolution of calcium and oxygen sublattices of calcite in ambient conditions by atomic force microscopy using qPlus sensors with sapphire tips. *ACS Nano* **9**, 3858–3865 (2015).
- Ooe, H. *et al.* Amplitude dependence of image quality in atomically-resolved bimodal atomic force microscopy. *Applied Physics Letters* **109**, 141603 (2016).
- Ichii, T., Fujimura, M., Negami, M., Murase, K. & Sugimura, H. Frequency modulation atomic force microscopy in ionic liquid using quartz tuning fork sensors. *Japanese Journal of Applied Physics* **51**, 08KB08 (2012).

36. Kobayashi, K., Yamada, H. & Matsushige, K. Frequency noise in frequency modulation atomic force microscopy. *Review of Scientific Instruments* **80**, 043708 (2009).
37. Fukuma, T., Higgins, M. J. & Jarvis, S. P. Direct imaging of lipid-ion network formation under physiological conditions by frequency modulation atomic force microscopy. *Physical Review Letters* **98**, 106101 (2007).
38. Mingeot-Leclercq, M.-P., Deleu, M., Brasseur, R. & Dufrène, Y. F. Atomic force microscopy of supported lipid bilayers. *Nature Protocols* **3**, 1654 (2008).
39. Richter, R. P. & Brisson, A. R. Following the formation of supported lipid bilayers on mica: a study combining AFM, QCM-D, and ellipsometry. *Biophysical Journal* **88**, 3422–3433 (2005).
40. Müller, D. J. & Engel, A. Atomic force microscopy and spectroscopy of native membrane proteins. *Nature Protocols* **2**, 2191 (2007).
41. Pfreundschuh, M., Martinez-Martin, D., Mulvihill, E., Wegmann, S. & Muller, D. J. Multiparametric high-resolution imaging of native proteins by force-distance curve-based AFM. *Nature Protocols* **9**, 1113 (2014).
42. Liang, Y. *et al.* Organization of the G protein-coupled receptors rhodopsin and opsin in native membranes. *Journal of Biological Chemistry* **278**, 21655–21662 (2003).
43. Radoslovich, E. W. The structure of muscovite, $KAl_2(Si_3Al)O_{10}(OH)_2$. *Acta Crystallographica* **13**, 919–932 (1960).
44. Asakawa, H. & Fukuma, T. The molecular-scale arrangement and mechanical strength of phospholipid/cholesterol mixed bilayers investigated by frequency modulation atomic force microscopy in liquid. *Nanotechnology* **20**, 264008 (2009).
45. Asakawa, H. & Yoshioka, S. Nishimura, K.-i. & Fukuma, T. Spatial distribution of lipid headgroups and water molecules at membrane/water interfaces visualized by three-dimensional scanning force microscopy. *ACS Nano* **6**, 9013–9020 (2012).
46. Nagle, J. F. & Tristram-Nagle, S. Lipid bilayer structure. *Current Opinion in Structural Biology* **10**, 474–480 (2000).
47. Alberts, B. *et al.* *Molecular Biology of the Cell* (Garland Science, 2014).
48. Kučerka, N. *et al.* Lipid bilayer structure determined by the simultaneous analysis of neutron and X-ray scattering data. *Biophysical Journal* **95**, 2356–2367 (2008).
49. Stöckle, R. M., Suh, Y. D., Deckert, V. & Zenobi, R. Nanoscale chemical analysis by tip-enhanced raman spectroscopy. *Chemical Physics Letters* **318**, 131–136 (2000).
50. Chen, C., Hayazawa, N. & Kawata, S. A 1.7 nm resolution chemical analysis of carbon nanotubes by tip-enhanced Raman imaging in the ambient. *Nature Communications* **5**, 3312 (2014).
51. Pohl, D., Fischer, U. C. & Dürig, U. Scanning near-field optical microscopy (SNOM). *Journal of Microscopy* **152**, 853–861 (1988).
52. Betzig, E. & Chichester, R. J. Single molecules observed by near-field scanning optical microscopy. *Science* **262**, 1422–1425 (1993).
53. Pan, S., Hudson, E. & Davis, J. 3 He refrigerator based very low temperature scanning tunneling microscope. *Review of Scientific Instruments* **70**, 1459–1463 (1999).
54. Huber, F. & Giessibl, F. J. Low noise current preamplifier for qPlus sensor deflection signal detection in atomic force microscopy at room and low temperatures. *Review of Scientific Instruments* **88**, 073702 (2017).
55. Oesterhelt, D. & Stoekenius, W. Isolation of the cell membrane of Halobacterium halobium and its fractionation into red and purple membrane. *Methods in Enzymology* **31**, 667–678 (1974).
56. Pielmeier, F., Meuer, D., Schmid, D., Strunk, C. & Giessibl, F. J. Impact of thermal frequency drift on highest precision force microscopy using quartz-based force sensors at low temperatures. *Beilstein Journal of Nanotechnology* **5**, 407 (2014).

Acknowledgements

We thank the German Research Foundation (SFB 689, GRK 1570) and the European Commission (FP7-PEOPLE-2012-ITN-317348) for their support as well as Dominik Kirpal and Sonia Matencio for fruitful discussions and proofreading.

Author Contributions

A.J.W., F.J.G. and D.M. conceived of the initial measurements. K.Pf., E.M., A.J.W. and D.M. were involved in sample preparation and measurements of the purple membrane. K.P. built the liquid cell, performed the noise analysis, and acquired data of the mica surface and the egg-PC membrane. L.K. and M.P.K. performed the sample preparation of the egg-PC membrane. K.P., A.J.W. and F.J.G. wrote the final manuscript. All authors were included in proofreading the manuscript.

Additional Information

Competing Interests: Franz J. Giessibl holds patents in Germany, USA and China to the state-of-the-art qPlus sensor since 2011 and the initial version of the qPlus sensor in Germany and USA since 1996 and 1999 respectively. The authors declare no further competing interests.

Publisher's note: Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2018