

Original Article

Yanangdaengin, a dihydrochalcone glucoside galloyl ester as active antioxidative agent from leaves of *Lysiphyllum strychnifolium* (syn. *Bauhinia strychnifolia*)

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ABSTRACT

Objective: To isolate and identify the major bioactive components from the leaves of *Lysiphyllum strychnifolium*, an indigenous herb used in traditional Thai medicine for detoxification, longevity, and some other health related issues.

Methods: Comparative HPLC analyses of the crude extracts from three provenances were carried out for an overview of characteristic compound profiles. Isolation of the major compounds was undertaken with chromatographic methods. Chemical structures were elucidated by NMR spectroscopic techniques and mass spectrometry. DPPH scavenging assay was carried out to determine the free radical scavenging activity of isolated compounds.

Results: Yanangdaengin (**3**), a dihydrochalcone glucoside galloyl ester, has been isolated together with its corresponding dihydrochalcone glucoside trilobatin (**2**) as major compounds from the leaves of *L. strychnifolium*. Additionally, gallic acid (**1**) was co-chromatographically identified. Free radical scavenging activity of isolated compounds were determined. Compound **3** exhibited higher free radical scavenging activities in comparison to Trolox and quercetin.

Conclusion: The isolated compounds could be used as chemical markers for quality assessment. The present work could promote the quality control and herbal medicinal product development of this plant.

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1. Introduction

Lysiphyllum strychnifolium (Craib) A. Schmitz, is an endemic herb commonly known as “Kha-Yan” or “Ya-Nang-Daeng” in Thailand (Larsen & Larsen, 1984; Pooma & Suddee, 2014). This plant was previously listed as *Bauhinia strychnifolia* Craib and reclassified based on results of a phylogenetic study of nuclear ribosomal DNA (Hao et al., 2003). This species occurs mainly in dry deciduous dipterocarp forests in northern, central and eastern parts of Thailand (Tangnak et al., 2018). In traditional Thai medicine (TTM), aqueous extracts from leaves and/or stems of *L. strychnifolium* are used for detoxification, longevity and some other health-related issues (Luengthong et al., 2016; Maitree et al., 2018; Sutiyaporn et al., 2018; Wutthithammawet, 1997). From

extracts of this plant, various activities such as cytotoxic effects against human cancer cell lines (Kaewpiboon et al., 2012; Yuenyongsawad et al., 2013), anti-HIV-1 and anti-allergic (Bunluepuech et al., 2013), and antihyperuricemic effects (Sutiyaporn et al., 2018) have been reported. Although this plant is commonly used by Thai traditional practitioners, there are no reports on the economic importance of this plant. Mostly, Thai traditional practitioners cultivate this plant in their gardens for medicinal purposes. Scientific research in depth of its phytochemical composition and pharmacological mode of actions may promote this plant as a potential economic crop for developing herbal medicinal products.

To date, several flavonoids such as quercetin, 3,5,7,3',5'-penta-hydroxy-flavanonol-3-O- α -L-rhamnopyranoside, 3,5,7-trihydroxy-chromone-3-O- α -L-rhamnopyranoside and the triterpenoids β -sitosterol, and stigmasterol are known from stems (Bunluepuech et al., 2013; Yuenyongsawad et al., 2013). Gallic acid was detected

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in leaves of *L. strychnifolium* (syn. *B. strychnifolia*) (Maitree et al., 2018; Sutiyaoporn et al., 2018). However, despite the long traditional use of this plant, the quality control and pharmacological research have been hindered because the major bioactive compounds were unknown. Herein, we are able to report the isolation and structure elucidation of yanangdaengin (3), a dihydrochalcone glucoside galloyl ester of trilobatin. Furthermore, the dihydrochalcone trilobatin (2) and gallic acid (1) could be identified. Compounds 2 and 3 represent the major active constituents of this plant. From all three compounds we assessed the free-radical scavenging activity, especially compounds 1 and 3 showed impressive activities in comparison to Trolox.

2. Materials and methods

2.1. Reagents

HPLC grade methanol and glacial acid were purchased from Labscan (Thailand). Deionized water was purified by Ultra Clear system (Siemen Water Technologies Corp., USA). Gallic acid was purchased from Tokyo Chemical Industry Co., Ltd. (Japan). 1,1-diphenyl-2-picrylhydrazyl radical (DPPH) was purchased from Sigma (St. Louis, MO).

2.2. Plant materials

The samples were collected in the provinces Ratchaburi, Udonthani and Uttaradit (Thailand) and also purchased from Charoen-suk Pharma Supply Co., Ltd., Nakhon Pathom, Thailand. The samples were identified by Ms. Pajaree Inthachub. Voucher specimens were deposited at Drug Discovery and Development Center, Office of Advanced Science and Technology, Thammasat University, Thailand. The samples were dried at 50 °C for 72 h.

2.3. Extraction and isolation

The dried and powdered leaves (overall 200 g) were macerated with methanol for 3 × 72 h with occasional shaking. After HPLC profiling, the extracts were pooled, filtered, and the solvent was evaporated using a rotary evaporator. This yielded 16 g of crude methanolic extract. This extract was roughly separated by column chromatography (CC) (Merck silica gel 60, 70–230 mesh) eluted with ethyl acetate and methanol (95:5, volume percent). Fractions were analyzed by TLC (silica gel 60 F₂₅₄) and HPLC. Further purification was made using CC (Merck LiChroprep RP-18, 40–63 μm) with methanol and water (50:50, volume percent). Purification by CC over Sephadex LH-20 eluted isocratic with methanol yielded 34 mg of trilobatin (2) and 8 mg of yanangdaengin (3).

2.4. NMR and mass spectrometry

All NMR spectra were recorded at room temperature either on a Bruker Avance II 400 (resonance frequencies 400.13 MHz for ¹H and 100.63 MHz for ¹³C) or a Bruker Avance III 600 (resonance frequencies 600.25 MHz for ¹H and 150.95 MHz for ¹³C) with standard Bruker pulse programs. The samples were dissolved in 0.6 mL of MeO-d₄ (99.9% D). Chemical shifts are given in ppm, referenced to residual solvent signals (δ_{H} 3.31, δ_{C} 49.0).

Mass spectra were recorded on a high-resolution time-of-flight (HR-TOF) mass spectrometer (maXis, Bruker Daltonics) by direct infusion electrospray ionization (ESI) in positive ionization mode (mass accuracy $\pm 5 \times 10^{-6}$) as well as in negative mode (mass accuracy $\pm 10 \times 10^{-6}$). HR-TOF MS measurements have been performed within the selected mass range of *m/z* 100–2500. ESI was made by the capillary voltage of 4 kV to maintain a (capillary) cur-

rent between 30 and 50 nA. Nitrogen temperature was maintained at 180 °C using a flow rate of 4.0 L/min and the N₂ nebulizer gas pressure at 0.3 bar.

2.5. HPLC analysis

HPLC analyses were performed on an Agilent 1260 Series (Agilent Technologies) equipped with a 1260 Quat pump VL quaternary pump, 1260 ALS autosampler, 1260 TCC column thermostat, 1260 DAD VL diode array detector and a Hypersil BDS C₁₈ column (4.6 mm × 100 mm; 3.5 μm particle size); injection volume was 10 μL and the wavelength of detection was set at 254 nm. The mobile phases were (A) 0.5% acetic acid in water and (B) methanol. Gradient elution was used from 0% to 100% B for 40 min, 100% B for 10 min. The column was equilibrated with 100% A for 10 min prior to each analysis. The flow rate was set at 1.0 mL/min at 25 °C.

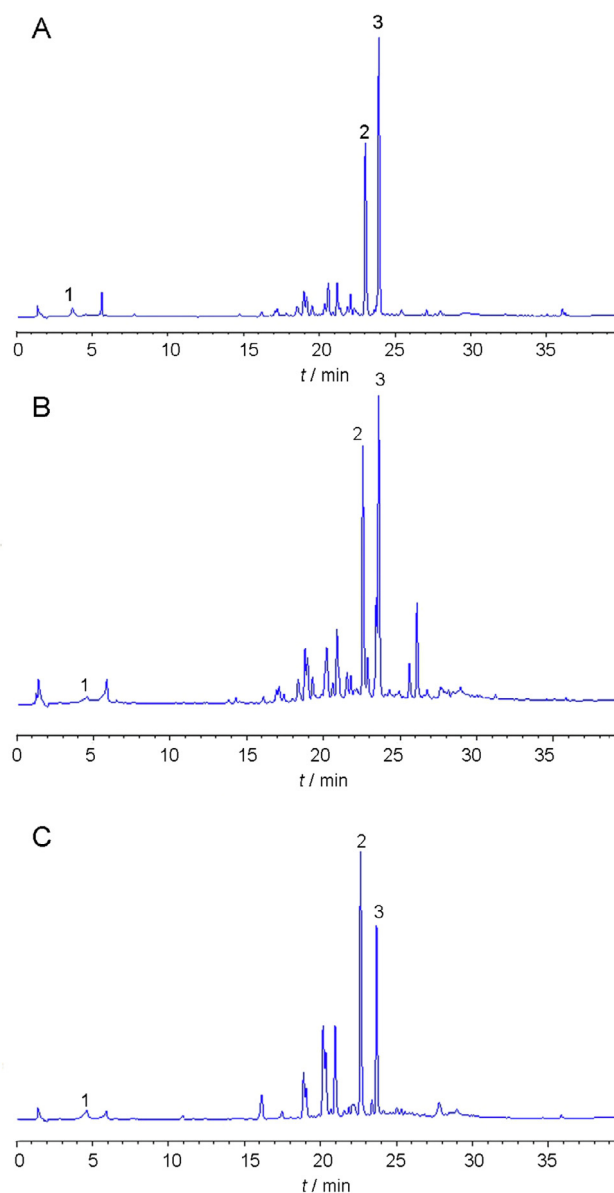


Fig. 1. HPLC at 254 nm overview of crude extracts of *L. strychnifolium* leaves demonstrating same pattern of major compounds (gallic acid (1), trilobatin (2) and yanangdaengin (3)) from Uttaradit (A), Udonthani (B) and Ratchaburi (C).

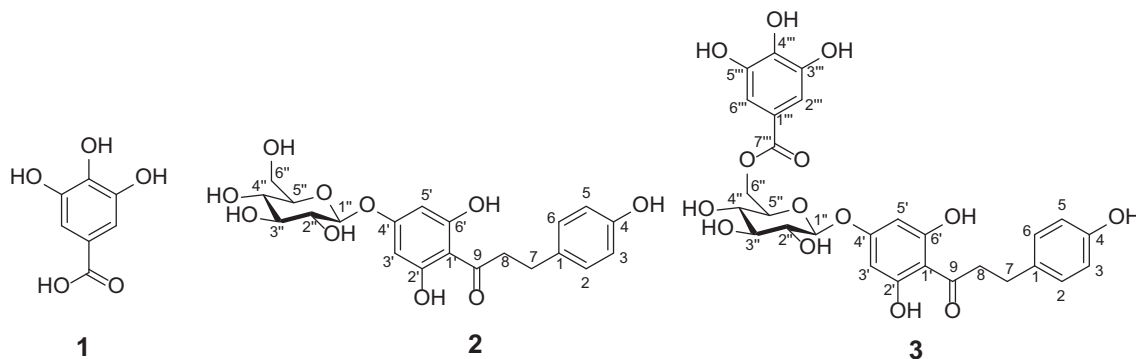


Fig. 2. Chemical structures of gallic acid (1), trilobatin (2) and yanangdaengin (3).

2.6. Determination of free radical scavenging activity

The free radical scavenging activity of isolated compounds was determined using the DPPH radical scavenging assay (Sithisarn et al., 2015; Vongsak et al., 2015). The stock solution of the plant extract (1 mg/mL) was serially diluted to concentrations of 1.56 to 100 µg/mL in a microplate (100 µL sample in each well) and 100 µL DPPH[•] solution (152 µmol/L in methanol) were added into each well. After incubation at 25 °C for 30 min, the absorbance at 517 nm was measured using a microplate reader. The corresponding blank was also determined and percent inhibition was calculated as follows: Scavenging activity (%) = $[1 - (A_1 - A_2) / A_0] \times 100\%$, where A_0 was the absorbance of control (DPPH[•] solution without sample), A_1 = absorbance of DPPH[•] solution in the presence of the sample; A_2 = was the absorbance without DPPH[•] solution.

3. Results and discussion

Dried and ground leaves of *L. strychnifolium* samples were extracted with 50% methanol in water. In order to get an overview of characteristic compound profiles, comparative HPLC analyses of

the crude extracts from different provenances were carried out (Fig. 1). With regard to previous studies of this plant (Bunluepuech et al., 2013; Yuenyongsawad et al., 2013), none of the reported constituents could be detected in major peaks. Therefore isolation of the major compounds was undertaken. This led to yanangdaengin (3), a galloyl ester of dihydrochalcone glucoside, together with its known derivative trilobatin (2). Their structures were elucidated by NMR and MS analyses and the spectroscopic data were compared with published data. Although, the structure of yanangdaengin (3) has been mentioned in Tao et al. (2012) as 4,2',6'-trihydroxy-dihydrochalcone-4'-O-(6''-galloyl)-β-D-glucopyranose without supporting data and was referenced to Tanaka et al. (2005), the structure of yanangdaengin (3) could not be found in this reference. Therefore, in the present work we now provide the NMR as well as HR-ESI-MS data of compound 3. Identification of gallic acid (1) in *L. strychnifolium* leaves was done by HPLC coupled with UV diode array detection and TLC comparison with an authentic standard. These results are well in line with previous studies (Maitree et al., 2018; Sutiyaiporn et al., 2018).

Yanangdaengin (phloretin 4'-O-(6''-O-galloyl)-β-D-glucoside, 3) was isolated as a white amorphous powder (m.p.165–166 °C).

Table 1
¹H and ¹³C NMR spectroscopic data for compounds 2 and 3.

| Carbons | δ_H | | δ_C | |
|-------------|-----------------------------------|--|-------------------------|-------------------------|
| | 2 ^a | 3 ^b | 2 ^c | 3 ^d |
| 1 | – | – | 133.8 (C) | 133.9 (C) |
| 2, 6 | 7.04 (2H, d, $J = 8.5$ Hz) | 7.04 (2H, d, $J = 8.6$ Hz) | 130.3 (CH) | 130.3 (CH) |
| 3, 5 | 6.69 (2H, d, $J = 8.5$ Hz) | 6.69 (2H, d, $J = 8.6$ Hz) | 116.1 (CH) | 116.1 (CH) |
| 4 | – | – | 156.5 (C-OH) | 156.5 (C-OH) |
| 7 | 2.86 (2H, m) | 2.85 (2H, m) | 31.2 (CH ₂) | 31.2 (CH ₂) |
| 8 | 3.30 (2H, m) | 3.30 (2H, m) | 47.6 (CH ₂) | 47.5 (CH ₂) |
| 9 | – | – | 207.0 (C = O) | 207.0 (C = O) |
| 1' | – | – | 106.8 (C) | 107.0 (C) |
| 2', 6' | – | – | 165.4 (C-OH) | 165.4 (C-OH) |
| 3', 5' | 6.09 (2H, s) | 6.08 (2H, s) | 96.4 (CH) | 96.4 (CH) |
| 4' | – | – | 165.0 (C) | 164.9 (C) |
| 1'' | 4.93 (1H, d, $J = 7.4$ Hz) | 4.98 (1H, d, $J = 7.5$ Hz) | 101.1 (CH) | 101.1 (CH) |
| 2'' | 3.43 (1H, m) | 3.47 (1H, m) | 74.6 (CH) | 74.6 (CH) |
| 3'' | 3.39 (1H, m) | 3.50 (1H, m) | 77.9 (CH) | 77.7 (CH) |
| 4'' | 3.45 (1H, m) | 3.53 (1H, m) | 71.1 (CH) | 71.1 (CH) |
| 5'' | 3.45 (1H, m) | 3.74 (1H, ddd, $J = 9.3, 4.8, 2.2$ Hz) | 78.3 (CH) | 75.7 (CH) |
| 6'' | 3.90 (1H, dd, $J = 12.1, 2.1$ Hz) | 4.55 (1H, dd, $J = 12.1, 2.2$ Hz) | 62.3 (CH ₂) | 64.3 (CH ₂) |
| 1''' | 3.71 (1H, dd, $J = 12.1, 5.5$ Hz) | 4.46 (1H, dd, $J = 12.1, 4.8$ Hz) | – | 121.2 (C) |
| 2''' , 6''' | – | – | – | 110.2 (CH) |
| 3''' , 5''' | – | – | – | 146.5 (C-OH) |
| 4''' | – | – | – | 139.8 (C-OH) |
| 7''' | – | – | – | 168.3 (C = O) |

^a ¹H NMR spectrum (600 MHz, methanol d_4).

^b ¹H NMR spectrum (400 MHz, methanol d_4).

^c ¹³C NMR spectrum (150 MHz, methanol d_4).

^d ¹³C NMR spectrum (100 MHz, methanol d_4).

Table 2
Content of compounds **1–3** in leaves of three examined samples and their free radical scavenging activity.

| Compounds | Content/% | | | IC ₅₀ /(mmol·L ⁻¹) ^a |
|----------------------------|--------------|--------------|--------------|--|
| | Uttaradit | Udonthani | Ratchaburi | |
| Gallic acid (1) | 0.15 (±0.01) | 0.10 (±0.01) | 0.17 (±0.01) | 5.99 ± 0.29 |
| Trilobatin (2) | 3.42 (±0.06) | 5.40 (±0.19) | 3.94 (±0.23) | 51.59 ± 1.67 |
| Yanangdaengin (3) | 1.34 (±0.24) | 0.97 (±0.13) | 0.67 (±0.23) | 5.03 ± 0.37 |
| Quercetin | | | | 8.52 ± 0.25 |
| Trolox | | | | 12.25 ± 0.39 |

^aData expressed as mean ± SD for triplicate analysis.

The ¹H NMR spectrum was closely related to that of trilobatin (**2**) (Qin et al., 2015), with the signals of the phloroglucinol ring at δ 6.08 (s, 2H, H-3' and H-5'), the *para*-hydroxybenzene moiety at δ 7.04 (d, 2H, H-3, H-5), 6.69 (d, 2H, H-2, H-6), two mutually coupled methylene groups at δ 3.30 (m, H-8) and 2.85 (m, H-7), and the resonances of a glucose group. The ¹³C NMR data also confirmed the presence of the phloretin-4'-*O*-β-glucoside subunit. Additional ¹³C resonances at δ 168.3, 146.5 (2C), 139.8, 121.2, and 110.2 (2C) and a two-proton singlet in the ¹H NMR at δ 7.08 were indicative for a galloyl group. The connecting position of the galloyl group was determined as CH₂-O of the glucoside (C-6) moiety on the basis of characteristic low field shifts of H-6''[δ4.54 (dd, *J* = 12.1, 2.2 Hz) and 4.45 (dd, *J* = 12.1, 4.8 Hz)] compared to those of trilobatin (**2**) and furthermore by long-range cross-peaks of these protons to the galloyl carboxyl carbon C-7'''. Accordingly, the structure of compound **3** was elucidated as phloretin 4'-*O*-(6''-*O*-galloyl)-β-*D*-glucoside (Fig. 2), which was also confirmed by high-resolution mass spectral data revealing a [M + Na]⁺ peak at *m/z* 611.1360 (calcd for C₂₈H₂₈O₁₄Na: 611.1371) in positive mode and in the negative mode a [M-H]⁻ peak at *m/z* 587.1414 (calcd for C₂₈H₂₇O₁₄: 587.1406), respectively. The assignment of all the ¹H and ¹³C NMR data was accomplished by ¹H-¹H COSY, ¹H-¹³C HSQC, and ¹H-¹³C HMBIC NMR data are listed in Table 1.

The free-radical scavenging activities of the isolated compounds were determined by the DPPH• free radical scavenging assay and the IC₅₀ values are reported in Table 2. The obtained results clearly showed the high potentials of compounds **1** and **3**, whereas **2** only showed moderate anti-oxidative properties. We assume that the gallic acid moiety of **3** and free gallic acid (**1**) content in the leaves are responsible for this positive effect. The higher amount of **3** in the leaves justify the usage as an anti-oxidative drug. This proposed active principle of the herbal drug now may promote the quality control and herbal medicinal product development.

4. Conclusion

Yanangdaengin (**3**) has been isolated together with its corresponding dihydrochalcone glucoside trilobatin (**2**) as major compounds from the leaves of *L. strychnifolium*. Additionally, gallic acid (**1**) was also identified. Free radical scavenging activities were determined, revealing high potentials of compounds **1** and **3**, whereas **2** only showed moderate anti-oxidative properties. These compounds could also be used as active chemical markers for quality assessment. The present study provided a useful basis for quality control and further development of this plant.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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