



Reactions of nitroxides XIII: Synthesis of the Morita–Baylis–Hillman adducts bearing a nitroxyl moiety using 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl as a starting compound, and DABCO and quinuclidine as catalysts

Jerzy Zakrzewski

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Address:

Institute of Industrial Organic Chemistry, Annopol 6, 03-236 Warsaw,
Poland

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Email:

Jerzy Zakrzewski - zakrzewski@ipo.waw.pl

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Abstract

The Morita–Baylis–Hillman adducts bearing a nitroxyl moiety were synthesized from 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl and aliphatic, aryl and heterocyclic aldehydes.

Introduction

In the Morita–Baylis–Hillman (MBH) reaction, aldehydes react with a double bond activated by an electron-withdrawing group (EWG). The vinylic carbon bearing an EWG undergoes substitution. The reaction is carried out in the presence of either a tertiary amine (e.g., DABCO [2–6], quinuclidine and its derivatives [7–12], DBU [13,14], DBN [13], DMAP and its derivatives [4,15,16], urotropine [17], brucine *N*-oxide [18]) or a phosphine [19] as a catalyst. The MBH reaction is a carbon–carbon bond forming process. This is the reason why a huge amount of research devoted to the reaction is reported

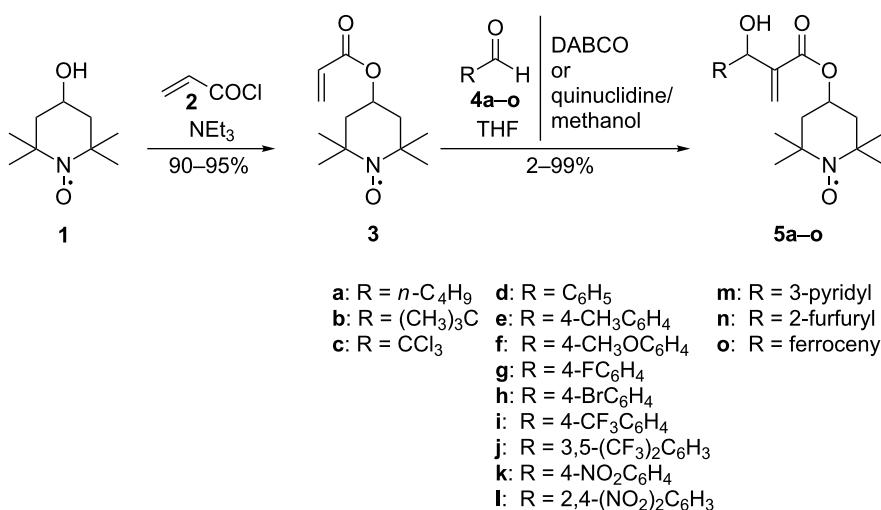
every year. The application and scope of the reaction has been summarized in many review articles, e.g., [20], as well as the latest ones [21–24]. Some recent results concerning MBH reaction have been presented [6,16,18,25–33]. MBH adducts themselves are reported to be antiproliferative agents [34]; however, they are often applied as a tool for building more complex target structures, usually of biological importance [35–43]. The MBH reaction is a rather slow process (complete reaction can take hundreds of hours), especially when acrylates are used [44]. As has been very well known since the 1960s,

stable nitroxides can react without affecting the unpaired electron. However, there are also reactions that do involve the free electron (e.g., many types of reductions, or a disproportionation in an acidic environment). To the best of our knowledge, nitroxyl radicals have not yet been applied in the MBH reaction (to date), and the potential influence of the nitroxide moiety on the MBH reaction is unknown. Herein we present the MBH reaction with a nitroxyl radical, 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl, used as a starting material as an olefin activated with EWG. Acrylates are considered as rather unreactive in the MBH reaction [44]. It was shown that the effects on the reaction of aryl, benzyl, alkyl, and functionalized alkyl acrylic esters with benzaldehyde and furfuraldehyde in the presence of DABCO, strongly depend upon the electronic and steric effects of the ester part. The “unreactivity” of acrylates increases with steric hindrance and with increasing chain length of the alcohol moiety in an acrylate [20,45]. The alcohol moiety in 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl is undoubtedly sterically hindered, so the expected reaction times will be long; however, this nitroxide has been chosen based on the availability of such nitroxyl esters. As an electrophilic partner in the MBH reaction, commonly used aldehydes were chosen, whose activity is broadly discussed in the literature. Aromatic aldehydes, especially those containing EWGs, are considered as reactive in the MBH reaction, in contrast to the aliphatic aldehydes (both *n*-butanal and pivalaldehyde), which are considered to be unreactive, although EWGs on the α -carbon atom (e.g., chloral) enhance their reactivity [20]. 2-Furaldehyde and nicotinic aldehyde were used because they were considered to be especially reactive in the MBH reaction [43].

Results and Discussion

4-Acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl (**3**) was obtained by esterification of acryloyl chloride (**2**) with 2,2,6,6-tetramethyl-4-piperidinol-1-oxyl (**1**) [46,47] in 90–95% yield. To synthesize MBH adducts (**5a–o**), **3** was reacted with aliphatic (**4a–c**), aryl (**4d–l**) and heterocyclic (**4m,n**) aldehydes, and (to obtain a compound bearing both nitroxyl and ferrocenyl moiety) ferrocenyl aldehyde (**4o**). Two catalytic systems were tested: DABCO, and quinuclidine with methanol as a cocatalyst. The latter system was chosen because it has been described as an excellent rate enhancer for the MBH reaction [9,11]. Methanol was chosen as an additive bearing a polar O–H bond, which activates both the aldehyde and “Michael” intermediate formed in the first step of the reaction, when an amine catalyst attacks an EWG activated olefin [11]. The reaction was carried out in THF as a solvent or the reagents were stirred neat. The synthesized MBH adducts (**5a–o**) are summarized in Scheme 1 and Table 1.

MBH adducts (**5a–o**) were obtained with very diverse results. Reaction times varied from one day to tens of days (**5f,g,o**). The reaction was significantly faster in the presence of quinuclidine/methanol system as a catalyst than in the presence of DABCO. The yields of the MBH adducts successfully obtained varied from negligible (**5o**, see below) to almost quantitative (**5c,m**). The use of 4-hydroxybenzaldehyde, 2-bromonicotinic aldehyde and 3-ferrocenylpropenal [48] did not result in any detectable MBH adducts in either catalytic system. The reaction with 2,4-dinitrobenzaldehyde (**4l**) provides some atypical intensely blue-colored side-products. The appropriate blue zones were isolated during column chromatography (Supporting Information



Scheme 1: MBH adducts **5a–o** from 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl.

Table 1: Yields and melting points (mp) of MBH adducts **5a–o**.

5	R	DABCO <i>t</i> [h]	DABCO yield [%]	quinuclidine/methanol <i>t</i> [h]	quinuclidine/methanol yield [%]	mp [°C]
a	<i>n</i> -C ₄ H ₉	333	36	96	81	83–85
b	(CH ₃) ₃ C	330	—	480	33	red oil
c	CCl ₃	91	48	26	99	154–156
d	C ₆ H ₅	115	78	48	92	116–118
e	4-CH ₃ C ₆ H ₄	672	—	336	42	130–133
f	4-CH ₃ OC ₆ H ₄	984	—	1112	27	121–124
g	4-FC ₆ H ₄	1008	27	336	23	105–109
h	4-BrC ₆ H ₄	672	24	168	56	127–130
i	4-CF ₃ C ₆ H ₄	768	76	72	77	132–135
j	3,5-(CF ₃) ₂ C ₆ H ₃	528	59	432	47	112–115
k	4-NO ₂ C ₆ H ₄	376	22	96	79	102–104
l	2,4-(NO ₂) ₂ C ₆ H ₃	90	69	30	56	50–57 (glass)
m	3-pyridyl	592	99	72	65	92–97
n	2-furyl	664	24	144	48	114–118
o	ferrocenyl	1128	—	1680	2	118–124

File 1); however, attempts to obtain their physicochemical and spectral data were unsuccessful. The blue color of the side-products suggests that a nitroso group may be formed as a result of the reduction of the nitro group. Such transformations of 2,4-dinitrobenzaldehyde are well known. Under light, 2-nitroso-4-nitrobenzoic acid (characterized as its methyl ester) is formed [49]. In dilute aqueous sodium hydroxide solution, 4-nitroso-2-nitrophenol and formic acid are formed [50–52]. Ferrocenyl aldehyde (**4o**) as the starting compound [53] did not afford any product when DABCO was used as a catalyst. Use of the quinuclidine/methanol system and thorough searching of a potential product by TLC, resulted in the separation of the expected adduct **5o** in 2% yield. However, its identity was confirmed unambiguously by (HR-)MS (both EI and ESI) and IR spectroscopy. This poor result is probably caused by the cumulative steric hindrance of both ferrocenyl and nitroxyl moieties (the sensitivity of the MBH reaction to the steric hindrance of the alcohol moiety in an acrylate was commented on in the introduction [20,45]). Due to the radical nature of the adducts **5a–o**, their structures were confirmed by (HR-)MS (both EI and ESI), and IR spectroscopy (see Supporting Information File 1 for EIMS and IR spectra). Most of the EIMS spectra show the *m/z* 124 peak as the base one (except **5k,m,o**), and the abundant intensity of *m/z* 109 peak (except **5c,f,l,o**). The both fragments: *m/z* 124 and 109 are originated from the nitroxyl moiety of the investigated compounds **5**. HRMS–EI for *m/z* 124: calcd for C₉H₁₆: 124.12520; found: 124.12515; for *m/z* 109: calcd for C₈H₁₃: 109.10173; found: 109.10079. However, direct characterization of the adducts **5a–o** by NMR is impossible, one of the adducts (**5d**, R = C₆H₅) was subjected to the exemplary experiment developed for nitroxides by Keana and coworkers in 1975

[54]. **5d** was reduced *in situ* with phenylhydrazine directly in an NMR tube to a nonradical, corresponding *N*-hydroxylamine. The ¹H NMR and ¹³C NMR spectra were recorded, but only their aliphatic part was found to be valuable due to the signals belonging to phenylhydrazine itself. The recorded spectral data are presented in the experimental part and the Supporting Information File 1 (together with spectra of phenylhydrazine itself, as a background) as well. The observed singlet in ¹H NMR at δ 5.54, and symmetric narrow multiplets at 5.76–5.90 and 6.28–6.45 assigned to the C₆H₅–CH(OH)–C(=CH₂)- fragment are well consistent with the spectra of the typical series of the MBH adducts of common aldehydes and methyl acrylate, presented in [32]. Synthesized compounds **5a–o** showed a weak antifungal activity. No insecticidal, acaricidal and herbicidal activity were shown.

Conclusion

In conclusion, it has been demonstrated that the use of 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl (**3**) as a starting compound allows us to obtain new MBH adducts **5a–o** bearing a nitroxyl moiety. The use of quinuclidine with methanol as a catalyst instead of DABCO decreases the time of the reaction. No influence of the radical nature of 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl (**3**) on the reaction course was observed.

Experimental

General: 2,2,6,6-Tetramethyl-4-hydroxypiperidin-1-oxyl (**1**) was synthesized by the oxidation of 2,2,6,6-tetramethyl-4-piperidinol with 30% hydrogen peroxide (76.5% yield, mp 71–73 °C), according to [55–57]. Liquid aldehydes were puri-

fied by using vacuum distillation. THF was distilled over sodium under argon in the presence of benzophenone as an indicator. The experiments were performed in a 5 mL round-bottom flask, equipped with a magnetic stirrer. Most of the MBH adducts were obtained as red solids. TLC was carried out on silica gel Merck Alurolle 5562, Alufolien 5554. Column chromatography was performed by using Merck 1.09385.1000 or Zeochem 60 hyd 40–63 µm (0.040–0.063 mm, 230–400 mesh). TLC visualisation: UV 254 nm light and/or iodine vapours. EIMS data were recorded by using AMD 604 and Agilent Technologies 5975 B mass spectrometers. HRMS–EI data were recorded by using an AMD 604 mass spectrometer. ESIMS and HRMS–ESI (positive ions, CH₃OH as solvent) were recorded by using a Micromass LCT apparatus. IR (ν , cm⁻¹) data were recorded by using an FT/IR Jasco 420 spectrophotometer. ¹H and ¹³C NMR data were collected by using a Varian UNITY-plus 200 spectrophotometer.

4-Acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl (3): 2,2,6,6-Tetramethyl-4-piperidinol-1-oxyl (**1**, 0.344 g, 2 mmol) in benzene (\approx 5 mL) was placed in a round bottomed flask of 50 mL capacity equipped with a magnetic stirrer. A solution of triethylamine (1.54 g, 15.3 mmol, 2.1–2.2 mL) in benzene (6 mL) was added dropwise from a pipette. A solution of acryloyl chloride (**2**, 0.187 g, 2.08 mmol, 170 µL) in benzene (3.5 mL) was added dropwise from a syringe at room temperature. The progress of the reaction was monitored by TLC (benzene/ethyl acetate 9:1, benzene/methanol 9:1). The reaction mixture was stirred at room temperature for 20 h, then the second portion of **2** (0.094 g, 1.04 mmol, 85 µL) in benzene (1.7 mL) was added dropwise from a syringe. The reaction mixture was stirred at room temperature for 1 h. Depending on the results of the TLC control of the progress of the reaction the third portion of **2** may be added (\approx 50 µL). After the reaction had been terminated, the precipitate of triethylamine hydrochloride was filtered off and the filtrate was concentrated under reduced pressure. The red residue was subjected to column chromatography (hexane/ethyl acetate 9:1 as a mobile phase). The red eluent was collected to give red crystals of 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl (**3**), yield: 0.41–0.43 g (90–95%); mp 102–104 °C (lit. [46]: 102.5–103 °C, [58]: 102 °C, [59]: 99 °C); EIMS m/z (% relative intensity): M⁺ 226 (16), 194 (10), 154 (15), 141 (7), 140 (17), 139 (21), 124 (84), 109 (95), 98 (6), 95 (8), 82 (21), 81 (20), 69 (11), 68 (20), 67 (19), 55 (100), 41 (34); HRMS–EI (m/z): calcd for C₁₂H₂₀NO₃, 226.14432; found, 226.14401; IR (KBr): 1720 (C=O), 1635 (C=C) cm⁻¹.

MBH adducts 5; general procedure without solvent with an excess of aldehyde (5a,b,d,n): 4-Acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl (**3**, 0.4 mmol), and a catalyst (about

10–20 mol % DABCO or quinuclidine), methanol (only if quinuclidine is used as catalyst) (10 µL), and an excess of freshly distilled aldehyde (**4a**, **4b**, **4d**, **4n**, \approx 1 mL) were stirred under argon at room temperature for the time mentioned in Table 1. The progress of the reaction was monitored by TLC (hexane/ethyl acetate 9:1, benzene/ethyl acetate 9:1, benzene/methanol 9:1). The MBH adduct **5** was isolated by direct column chromatography of the reaction mixture by using an appropriate mobile phase (hexane/ethyl acetate 9:1, benzene/ethyl acetate 95:5, benzene/methanol 95:5) to afford the desired adduct **5**.

MBH adducts 5; general procedure in THF (5c, e–m, o): An appropriate aldehyde, freshly distilled and added with a syringe if liquid (0.8 mmol (**4e**, **4f**, **4g**, **4h**, **4i**, **4j**, **4m**), 0.6 mmol (**4c**), or 0.4 mmol (**4k**, **4l**, **4o**)), 4-acryloyloxy-2,2,6,6-tetramethylpiperidine-1-oxyl (**3**, 0.4 mmol), a catalyst (about 10–20 mol % DABCO or quinuclidine), methanol (only if quinuclidine is used as catalyst, 10 µL), and anhydrous THF (1.0–1.5 mL), were stirred under argon at room temperature for the time mentioned in Table 1. The progress of the reaction was monitored by TLC (hexane/ethyl acetate 9:1, benzene/ethyl acetate 9:1, benzene/methanol 9:1). THF was evaporated under reduced pressure. The residue was subjected to column chromatography by using an appropriate mobile phase (hexane/ethyl acetate 9:1, benzene/ethyl acetate 95:5, benzene/methanol 95:5) to afford the desired adduct **5**.

4-(2-((n-Butyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5a): Yield (DABCO): 46 mg (36%); Yield (quinuclidine): 101 mg (81%); orange solid; mp 83–85 °C; EIMS m/z (% relative intensity): M⁺ 312 (10), 255 (7), 240 (10), 155 (45), 154 (23), 141 (11), 140 (35), 139 (17), 124 (100), 109 (69), 100 (18), 98 (9), 95 (30), 85 (10), 83 (44), 82 (19), 81 (16), 74 (14), 69 (31), 67 (17), 56 (18), 55 (29), 41 (32); HRMS–EI (m/z): calcd for C₁₇H₃₀NO₄: 312.21748; found: 312.21674; mass spectrum (ESI, m/z , %) 335 (100, [M + Na]⁺) 140 (46); HRMS–ESI (m/z): calcd for C₁₉H₂₆NO₄Na, 335.2073; found, 335.2067; IR (KBr): 1701 (C=O), 1633 (C=C) cm⁻¹.

4-(2-((tert-Butyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5b): Yield (quinuclidine): 41 mg (33%); red oil; EIMS m/z (% relative intensity): M⁺ 312 (11), 241 (6), 172 (7), 156 (44), 155 (64), 140 (73), 124 (100), 109 (65), 100 (49), 98 (17), 95 (22), 85 (13), 83 (22), 82 (27), 81 (24), 74 (55), 69 (46), 67 (23), 58 (17), 57 (98), 56 (38), 55 (42), 41 (65); HRMS–ESI (m/z): calcd for C₁₇H₃₀NO₄, 312.21748; found, 312.21818; ESIMS m/z (% relative intensity): [M + Na]⁺ 335 (100); HRMS–ESI (m/z): [M + Na]⁺ calcd for C₁₇H₃₀NO₄Na, 335.2073; found, 335.2042; IR (KBr): 1706 (C=O) 1627 (C=C) cm⁻¹.

4-(2-((Trichloromethyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5c): Yield (DABCO): 72 mg (48%); yield (quinuclidine): 148 mg (99%); yellow powder; mp 154–156 °C; EIMS m/z (% relative intensity): 374 (5), M^+ 372 (5), 240 (9), 203 (4), 201 (5), 167 (9), 165 (15), 155 (24), 154 (20), 140 (17), 139 (19), 124 (100), 110 (6), 100 (10), 98 (6), 95 (5), 85 (11), 83 (7), 82 (17), 81 (14), 69 (22), 68 (11), 67 (13), 56 (14), 55 (20); HRMS–EI (m/z): calcd for $C_{14}H_{21}NO_4Cl_3$, 372.05362; found, 372.05425; ESIMS m/z (% relative intensity): 397 (86), $[M + Na]^+$ 395 (100); HRMS–ESI (m/z): $[M + Na]^+$ calcd for $C_{14}H_{21}Cl_3NO_4Na$, 395.0434; found, 395.0429; IR (KBr): 1710 (C=O), 1633 (C=C) cm^{-1} .

4-(2-((Phenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5d): Yield (DABCO): 104 mg (78%); yield (quinuclidine): 122 mg (92%); orange crystals; mp 116–118 °C; EIMS m/z (% relative intensity): M^+ 332 (9), 302 (3), 284 (2), 154 (16), 140 (11), 133 (8), 124 (100), 117 (28), 116 (14), 115 (27), 109 (82), 79 (21); HRMS–EI calcd for $C_{19}H_{26}NO_4$, 332.1862; found, 332.1856; ESIMS m/z (% relative intensity): $[M + Na]^+$ 355 (100); HRMS–ESI (m/z): $[M + Na]^+$ calcd for $C_{19}H_{26}NO_4Na$, 355.1760; found, 355.1742; IR (KBr): 1707 (C=O), 1628 (C=C) cm^{-1} ; ^1H NMR (200 MHz, CDCl_3 , after reduction with phenylhydrazine in situ, in an NMR tube, in CDCl_3 , aliphatic part of the spectrum, in fact the spectrum of the corresponding hydroxylamine [54]) δ 1.16, 1.18, 1.19, 1.21 (4s, 12H, 4CH_3), 1.40–1.64 (m, 2H, CH_2), 1.75–1.95 (m, 2H, CH_2), 4.96–5.18 (m, 1H, $\text{CH}-\text{OC}(=\text{O})$), 5.54 (s, 1H, $\text{CH}-\text{OH}$), 5.76–5.90 (m, 1H, =CHH), 6.28–6.45 (m, 1H, =CHH) ppm; ^{13}C NMR (50 MHz, CDCl_3 , after reduction with phenylhydrazine in situ, in an NMR tube, in CDCl_3 , aliphatic part of the spectrum, in fact the spectrum of the corresponding hydroxylamine [54]) δ 20.65 (2 CH_3), 32.00 (2 CH_3), 43.75 (CH_2 , confirmed by DEPT 135, piperidine ring), 43.81 (CH_2 , confirmed by DEPT 135, piperidine ring), 43.97 (CH_2 , confirmed by DEPT 135, piperidine ring), 59.48 (2C, absent in DEPT 135, piperidine ring), 67.68 ($\text{CHOC}(=\text{O})$, 73.32 (CHOH), 126.12 ($\text{C}=\text{CH}_2$, confirmed by DEPT 135), 126.84 (2 CH_{ar}), 128.03 (CH_{ar}), 128.64 (2 CH_{ar}), 141.63 ($\text{C}_{\text{ar}}-\text{CHOH}$, absent in DEPT 135), 142.50 ($\text{C}=\text{CH}_2$, absent in DEPT 135), 165.98 ($\text{C}=\text{O}$, absent in DEPT 135) ppm.

4-(2-((4-Methylphenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5e): Yield (quinuclidine): 58 mg (42%); orange solid; mp 130–133 °C; EIMS m/z (% relative intensity): M^+ 346 (22), 191 (13), 175 (15), 173 (12), 156 (11), 155 (10), 154 (22), 147 (9), 140 (18), 139 (16), 131 (25), 130 (9), 129 (15), 124 (100), 109 (67), 93 (11), 91 (19), 82 (13), 81 (14), 77 (9), 74 (8), 69 (13), 68 (7), 69 (13), 56 (8), 55 (13), 41 (14); HRMS–EI (m/z): calcd for $C_{20}H_{28}NO_4$, 346.20183; found, 346.20093; ESIMS m/z (% relative intensity): $[2M + Na]^+$

715 (3), $[M + Na]^+$ 369 (100); HRMS–ESI (m/z): $[M + Na]^+$ calcd for $C_{20}H_{28}NO_4Na$, 369.1905; found, 369.1916; IR (KBr): 1709 (C=O), 1630 (C=C) cm^{-1} .

4-(2-((4-Methoxyphenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5f): Yield (quinuclidine): 39 mg (27%); orange solid; mp 121–124 °C; EIMS m/z (% relative intensity): M^+ 362 (23), 208 (13), 207 (17), 191 (23), 190 (14), 189 (19), 173 (12), 163 (9), 162 (8), 154 (42), 147 (10), 146 (14), 145 (20), 140 (31), 139 (19), 137 (23), 135 (36), 124 (100), 110 (7), 69 (23), 57 (6), 56 (11), 41 (21); HRMS–EI (m/z): calcd for $C_{20}H_{28}NO_5$, 362.19675; found, 362.19501; EIMS m/z (% relative intensity): $[M + Na]^+$ 385 (100), 119 (15); HRMS–ESI (m/z): $[M + Na]^+$ calcd for $C_{20}H_{28}NO_5Na$, 385.1865; found, 385.1841; IR (KBr): 1707 (C=O), 1611 (C=C) cm^{-1} .

4-(2-((4-Fluorophenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5g): Yield (DABCO): 38 mg (27%); Yield (quinuclidine): 33 mg (23%); yellow powder; mp 105–109 °C; EIMS m/z (% relative intensity): M^+ 350 (13), 195 (5), 180 (5), 179 (7), 177 (4), 154 (33), 140 (22), 139 (16), 135 (25), 134 (12), 133 (20), 124 (100), 109 (83), 97 (15), 83 (6), 82 (16), 69 (25), 67 (13), 56 (11), 55 (18), 41 (22); HRMS–EI (m/z): calcd for $C_{19}H_{25}NO_4F$, 350.17676; found, 350.17570; ESIMS m/z (% relative intensity): $[M + Na]^+$ 373 (20), 288 (100); HRMS–ESI (m/z): $[M + Na]^+$ calcd for $C_{19}H_{25}FNO_4Na$, 373.1665; found: 373.1652; IR (KBr): 1712 (C=O), 1631 (C=C) cm^{-1} .

4-(2-((4-Bromophenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5h): Yield (DABCO): 39 mg (24%); yield (quinuclidine): 92 mg (56%); orange powder; mp 127–130 °C; EIMS m/z (% relative intensity): 412 (12), M^+ 410 (11), 255 (5), 239 (7), 185 (6), 160 (23), 154 (43), 140 (15), 139 (17), 124 (100), 116 (21), 109 (66), 69 (18), 55 (14), 41 (16); HRMS–EI (m/z): calcd for $C_{19}H_{25}NO_4Br$, 410.09669; found, 410.09745; ESIMS m/z (% relative intensity): 435 (80), $[M + Na]^+$ 433 (80), 414 (95), $[M + 2H]^+$ 412 (100); HRMS–ESI (m/z): $[M + 2H]^+$ calcd for $C_{19}H_{27}NO_4Br$, 412.1123; found, 412.1109; HRMS–ESI (m/z): $[M + Na]^+$ calcd for $C_{19}H_{25}NO_4BrNa$, 433.0865; found, 433.0868; IR (KBr): 1708 (C=O), 1630 (C=C) cm^{-1} .

4-(2-((4-Trifluoromethylphenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5i): Yield (DABCO): 122 mg (76%); yield (quinuclidine): 123 mg (77%); orange-yellow powder; mp 132–135 °C; EIMS m/z (% relative intensity): M^+ 400 (13), 229 (6), 201 (10), 185 (15), 184 (8), 183 (12), 154 (39), 141 (7), 139 (16), 127 (16), 125 (15), 124 (100), 109 (82), 85 (10), 83 (6), 82 (14), 81 (16), 74 (9), 69

(22), 68 (10), 67 (12), 57 (7), 56 (11), 55 (17), 41 (22); HRMS–EI (m/z): calcd for $C_{20}H_{25}NO_4F_3$, 400.17357; found, 400.17225; ESIMS m/z (% relative intensity): [M + Na]⁺ 423 (100); HRMS–ESI (m/z): [M + Na]⁺ calcd for $C_{20}H_{25}F_3NO_4Na$, 423.1633; found, 423.1629; IR (KBr): 1712 (C=O), 1630 (C=C) cm⁻¹.

4-(2-((3,5-Bis(trifluoromethyl)phenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5j): Yield (DABCO): 111 mg (59%); yield (quinuclidine): 88 mg (47%); light-orange crystals; mp 112–115 °C; EIMS m/z (% relative intensity): M⁺ 468 (16), 454 (4), 434 (10), 253 (10), 252 (5), 251 (4), 249 (6), 243 (6), 241 (9), 233 (10), 229 (8), 213 (7), 195 (13), 154 (36), 140 (22), 139 (17), 124 (100), 109 (74), 85 (12), 82 (14), 81 (14), 69 (15), 68 (9), 67 (11), 57 (6), 56 (10), 55 (14), 41 (16); HRMS–EI (m/z): calcd for $C_{21}H_{24}NO_4F_6$, 468.16095; found, 468.16152; ESIMS m/z (% relative intensity): [M + Na]⁺ 491 (100); HRMS–ESI (m/z): [M + Na]⁺ calcd for $C_{21}H_{24}F_6NO_4Na$, 491.1507; found, 491.1459; IR (KBr): 1714 (C=O), 1638 (C=C) cm⁻¹.

4-(2-((4-Nitrophenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5k): Yield (DABCO): 33 mg (22%); yield (quinuclidine): 120 mg (79%); yellow powder; mp 102–104 °C; EIMS m/z (% relative intensity): M⁺ 377 (15), 363 (12), 189 (8), 160 (25), 154 (51), 140 (67), 139 (19), 124 (76), 109 (100), 85 (13), 82 (22), 81 (21), 69 (29), 68 (13), 67 (16), 57 (9), 56 (16), 55 (23), 41 (28); HRMS–EI (m/z): calcd for $C_{19}H_{25}N_2O_6$, 377.17126; found, 377.17090; ESIMS m/z (% relative intensity): [M + Na]⁺ 400 (100); HRMS–ESI (m/z): [M + Na]⁺ calcd for $C_{19}H_{25}N_2O_6Na$, 400.1610; found, 400.1603; IR (KBr): 1713 (C=O), 1636 (C=C), 1521 (NO₂), 1351 (NO₂) cm⁻¹.

4-(2-((2,4-Dinitrophenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5l): Yield (DABCO): 117 mg (69%); yield (quinuclidine): 95 mg (56%); red glass; mp 50–67 °C; EIMS m/z (% relative intensity): M⁺ 422 (8), 408 (5), 179 (8), 154 (34), 140 (34), 139 (15), 124 (100), 85 (10), 82 (15), 81 (15), 69 (21), 68 (9), 67 (13), 57 (10), 56 (13), 55 (21), 41 (22); HRMS–EI (m/z): calcd for $C_{19}H_{24}N_3O_8$, 422.15634; found, 422.15561; ESIMS m/z (% relative intensity): [M + 2H]⁺ 424 (100); HRMS–ESI (m/z): [M + 2H]⁺ calcd for $C_{19}H_{26}N_3O_8$, 424.1720; found, 424.1729; IR (KBr): 1717 (C=O), 1628, (C=C), 1537 (NO₂), 1347 (NO₂) cm⁻¹.

4-(2-((3-Pyridyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5m): Yield (DABCO): 132 mg (99%); yield (quinuclidine): 87 mg (65%); orange-yellow powder; mp 92–97 °C; EIMS m/z (% relative intensity): 334 (12), M⁺ 333 (17), 303 (6), 301 (7), 247 (9), 180 (100), 163 (7),

162 (13), 154 (15), 144 (8), 140 (24), 135 (17), 124 (59), 118 (23), 117 (18), 109 (73), 82 (11), 81 (13), 80 (13), 79 (6), 78 (7), 69 (19), 68 (8), 67 (14), 56 (11), 55 (20), 53 (9), 41 (26); HRMS–EI (m/z): calcd for $C_{18}H_{25}N_2O_4$, 333.18143; found, 333.18050; ESIMS m/z (% relative intensity): [M + Na]⁺ 356 (100); HRMS–ESI (m/z): [M + Na]⁺ calcd for $C_{18}H_{25}N_2O_4Na$, 356.1712; found, 356.1708; IR (film) 1714 (C=O), 1632 (C=C) cm⁻¹.

4-(2-((2-Furyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5n): Yield (DABCO): 31 mg (24%); yield (quinuclidine): 62 mg (48%); dark orange-yellow powder; mp 114–118 °C; EIMS m/z (% relative intensity): M⁺ 322 (20), 305 (4), 154 (25), 151 (29), 140 (30), 124 (100), 109 (84), 97 (24), 83 (9), 82 (17), 81 (17), 69 (43), 67 (19), 56 (13), 55 (19), 41 (36); HRMS–EI (m/z): calcd for $C_{17}H_{24}NO_5$, 322.16545; found, 322.16457; ESIMS m/z (% relative intensity): 346 (50), [M + Na]⁺ 345 (100); HRMS–ESI (m/z): [M + Na]⁺ calcd for $C_{17}H_{24}NO_5Na$, 345.1552; found, 345.1554; IR (KBr): 1706 (C=O), 1633 (C=C) cm⁻¹.

4-(2-((Ferrocenyl)hydroxymethyl)acryloyloxy)-2,2,6,6-tetramethylpiperidine-1-oxyl (5o): Yield (quinuclidine): 4 mg (2%); orange solid; mp 118–124 °C; EIMS m/z (% relative intensity): 441 (8), M⁺ 440 (16), 439 (1), 438 (2), 426 (6), 425 (14), 424 (2), 423 (7), 410 (2), 409 (4), 392 (3), 286 (64), 284 (9), 270 (14), 269 (6), 268 (7), 243 (8), 241 (5), 224 (25), 186 (7), 185 (9), 149 (18), 98 (35), 97 (22), 80 (78), 71 (34), 70 (15), 69 (31), 58 (39), 57 (64), 56 (18), 55 (52), 45 (13), 44 (67), 43 (36), 42 (22), 41 (48), 40 (100); HRMS–EI (m/z): calcd for $C_{23}H_{30}NO_4Fe$, 440.15242; found, 440.15327; ESIMS m/z (% relative intensity): [M + Na]⁺ 463 (95), M⁺ 440 (100); HRMS–ESI (m/z): calcd for $C_{23}H_{30}NO_4Fe$, 440.1524; found, 440.1495; IR (KBr): 1701 (C=O), 1633 (C=C) cm⁻¹.

Supporting Information

Supporting Information features EIMS and IR spectra of the synthesized compounds **5a–o**, ¹H and ¹³C NMR of **5d** with phenylhydrazine, and the chromatographic separation of **5l**.

Supporting Information File 1

Detailed spectrographic data.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-8-171-S1.pdf>]

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