

## CFTR, a Rectifying, Non-Rectifying Anion Channel?

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CFTR is an anion channel with high conductance for  $\text{Cl}^-$  and other halides. Abnormalities in this channel which decrease plasma membrane  $\text{Cl}^-$  conductance result in the inherited disease, cystic fibrosis (CF). The fact that this disease is also characterized clinically by significant abnormalities in  $\text{HCO}_3^-$  secretion in the pancreas (1, 2), provides a priori evidence that this anion channel is critical to  $\text{HCO}_3^-$  transport in some, possibly all, exocrine epithelia where CFTR is expressed. If CFTR is significantly permeable to  $\text{HCO}_3^-$  and if  $\text{HCO}_3^-$  is physiologically transported through CFTR, the defect in  $\text{HCO}_3^-$  in CF might be self-evident. Thus, it is of fundamental importance to determine the  $\text{HCO}_3^-$  permeability characteristics of this channel in defining its role in epithelial  $\text{HCO}_3^-$  management and its impact on the disease process. Moreover, since there are no other anion channels characterized as " $\text{HCO}_3^-$  selective", the possibility that CFTR may function as such makes the question academically intriguing. Recently, a renewed interest in defining the role of CFTR in  $\text{HCO}_3^-$  transport has arisen (3-14). Most studies of the function of CFTR in this regard have been performed in cultured cells or tissues using whole cell or membrane patch techniques and have reported that in these ex vivo systems the permeability of  $\text{HCO}_3^-$  relative to  $\text{Cl}^-$  is between 0.1 to 0.2. In early studies using the intact, native, microperfused human sweat duct, which abundantly expresses CFTR in its apical membrane, we were unable to detect differences between  $\text{HCO}_3^-$  and gluconate (15) (we took gluconate as a standard for impermeant anions). In the present study, we undertook to determine the reason for this discrepancy, if possible, and to further define the relevance of CFTR to  $\text{HCO}_3^-$  movement.

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### Methods

We isolated single sweat ducts from fresh biopsies of human skin. Segments of ducts greater than 500  $\mu\text{m}$  were microperfused with a double barrel luminal micropipette that served to perfuse and record transepithelial voltage on one side and to pass constant current pulses on the other. This arrangement allowed estimation of the specific membrane conductance from the cable equation (16). After confirming the integrity of the perfused tubule, we applied 1,000-5,000 units/mL of  $\alpha$ -toxin from *Staphylococcus aureus* to the bath solution in order to selectively permeabilize the basolateral membrane (17). This procedure leaves the epithelium with an intact and functional apical membrane and a non-selective basal membrane permeable to molecules of up to about 5,000 mwu. Since activation of CFTR is exquisitely sensitive to ATP and cAMP, its activity can be readily controlled in this preparation by controlling the presence of either of these nucleotides in the cytosolic bathing solution. That is, addition of 10  $\mu\text{M}$  cAMP plus 5 mM ATP activates CFTR in seconds whereas removal of either cAMP or ATP from the cytosolic bath deactivates the channel. We then sought to determine the permeability of the apical membrane to  $\text{HCO}_3^-$  and  $\text{Cl}^-$  relative to the impermeant gluconate anion by measuring the trans apical membrane diffusion potential generated by chemical gradients for these anions and the simultaneous changes in membrane conductances.

In our previous investigations in which we had not seen a  $\text{HCO}_3^-$  conductance, we had thought it prudent not to exceed the maximum anticipated physiological concentration of 25 mM for  $\text{HCO}_3^-$  in the sweat duct. Subsequent studies from other laboratories have uniformly used essentially isosmotic concentrations of about 140 mM  $\text{HCO}_3^-$ . Thus, we felt compelled to examine the behavior of CFTR in the sweat duct at these higher  $\text{HCO}_3^-$  concentrations as well. Our general experimental protocol was to first inactivate CFTR and perfuse the cytosol with K-gluconate while changing the composition of the luminal perfusate from gluconate to  $\text{Cl}^-$  to  $\text{HCO}_3^-$ , not necessarily in that order (we used either the  $\text{K}^+$  salt or the  $\text{Na}^+$  salt plus  $10^{-5}$  M amiloride to block  $\text{Na}^+$  shunt conductance; the  $\text{K}^+$  permeability of the duct apical membrane is nil). We then activated CFTR and repeated the luminal perfusate changes. Next, we deactivated

CFTR by withdrawing ATP and/or cAMP and continued to perfuse the lumen of the tubule with 150 mM K-gluconate while we proceeded to change the composition of the cytosolic bath from gluconate to  $\text{Cl}^-$  to  $\text{HCO}_3^-$  as before. These maneuvers were repeated again after activating CFTR a second time and the transepithelial diffusion potential differences ( $\Delta V_{\text{ap}}$ ) and conductances ( $\Delta G_{\text{ap}}$ ) were recorded after each change on either side of the membrane. The fact that there were no significant changes in electrical parameters until cAMP and ATP were added provides strong evidence that CFTR is the only significant activatable ionic conductance in the membrane under these conditions. After establishing that there were no significant changes in specific conductance or diffusion potential through the apical membrane so long as CFTR remained inactivated, we omitted these steps in the protocol in order to preserve tissue viability.

## Results and Discussion

First, after activating CFTR with cAMP and ATP, we kept the cytosol perfused with K gluconate while changing the luminal perfusate. We found that the diffusion potential ( $V_{\text{ap}}$ ) for  $\text{HCO}_3^-$  vs. gluconate in the lumen did not change (mean  $\Delta V_{\text{ap}} = -1.6$  mV) while that for  $\text{Cl}^-$  vs. gluconate increased by nearly 40 mV (mean  $\Delta V_{\text{ap}} = -39.5$  mV). Similarly, the conductance ( $G_{\text{ap}}$ ) did not change when  $\text{HCO}_3^-$  replaced gluconate (mean

$\Delta G_{\text{ap}} = -0.7$  mS/cm<sup>2</sup>), but when  $\text{Cl}^-$  replaced gluconate,  $G_{\text{ap}}$  increased by more than 31 mS/cm<sup>2</sup> (Table 1a). On the surface, we took these data as confirmation of our initial impression that CFTR is impermeable to  $\text{HCO}_3^-$  even at these non-physiological concentrations of  $\text{HCO}_3^-$ . On the other hand, when we kept the lumen perfused with gluconate and changed the composition of the cytosolic perfusion fluid from gluconate to  $\text{HCO}_3^-$ , the diffusion potential became more negative by 15 mV and the corresponding  $G_{\text{ap}}$  increased by almost 12 mS/cm<sup>2</sup>. The changes to  $\text{Cl}^-$  in the lumen were even more dramatic. Mean  $V_{\text{ap}}$  changed by 53 mV with a simultaneous mean increase in  $G_{\text{ap}}$  of 124 mS/cm<sup>2</sup> (Table 1a).

These combined results suggest that CFTR effectively rectifies  $\text{HCO}_3^-$  permeation from cell to lumen. To our knowledge, channel behavior that completely rectifies one ion while remaining permeable to another has not been documented and is not easily reconciled from first principles. Further, inspection of the data reveals that both the magnitude of the diffusion p.d. for  $\text{Cl}^-$  and its associated  $G_{\text{Cl}}$  are reduced when gluconate is present in the cytosolic fluid. These considerations in addition to a previous report (18) that large organic anions produce a block of CFTR when present on the cytosolic surface suggested that the effect might be due to anomalous rectification induced by our "standard" impermeant anion, gluconate.

Thus, we proceeded to determine the  $\Delta V_{\text{ap}}$  and  $\Delta$

**Table 1.** Data show the differences in transapical membrane potentials ( $\Delta V_{\text{ap}}$ ) and conductances ( $\Delta G_{\text{ap}}$ ) of the microperfused sweat duct when the bath (...in Cell) or luminal fluids (...in Lumen) are kept at a constant composition of 150 mM Gluconate (Gluconate) or 150 mM Cl (Cl), indicated by headings in bold, while the fluid composition on the contralateral side of the membrane was changed from 150 mM gluconate to 150 mM  $\text{HCO}_3^-$  ( $\text{Glu}^- \rightarrow \text{HCO}_3^-$ ) and from 150 mM gluconate to 150 mM Cl ( $\text{Glu}^- \rightarrow \text{Cl}^-$ ), as indicated in the first column. Significant differences from values determined when 150 mM gluconate was present on both sides of the apical membrane were calculated using the Student's T-test for paired values.

Table 1a	Gluconate in Cell		Gluconate in Lumen	
	$\Delta V_{\text{ap}}$ (mV)	$\Delta G_{\text{ap}}$ (mS/cm <sup>2</sup> )	$\Delta V_{\text{ap}}$ (mV)	$\Delta G_{\text{ap}}$ (mS/cm <sup>2</sup> )
$\text{Glu}^- \rightarrow \text{HCO}_3^-$	-1.6	-0.7	15.0*	11.9*
sem =	2.1	2.5	2.5	31.2
$\text{Glu}^- \rightarrow \text{Cl}^-$	-39.5*	31.2*	53.2†	124.0†
sem =	6.0	9.8	6.7	38.6
	n=7		n=8	

  

Table 1b	Cl in Cell		Cl in Lumen	
	$\Delta V_{\text{ap}}$ (mV)	$\Delta G_{\text{ap}}$ (mS/cm <sup>2</sup> )	$\Delta V_{\text{ap}}$ (mV)	$\Delta G_{\text{ap}}$ (mS/cm <sup>2</sup> )
$\text{Glu}^- \rightarrow \text{HCO}_3^-$	-8.9	14.7*	18.7†	11.1
sem =	4.4	8.5	5.1	7.0
$\text{Glu}^- \rightarrow \text{Cl}^-$	-41.9*	56.0*	61.7†	75.6†
sem =	10.0	13.7	8.2	23.4
	n=4		n=10	

\*Significantly different:  $p < 0.05$

†Significantly different:  $p < 0.01$

Gap for  $\text{HCO}_3^-$  and  $\text{Cl}^-$  (compared to gluconate) when the contralateral anion was 150 mM  $\text{Cl}^-$ , instead of gluconate. With 150 mM  $\text{Cl}^-$  on the luminal side, we changed the cytosolic bath from gluconate to  $\text{HCO}_3^-$  to  $\text{Cl}^-$  as before and found that cytosolic  $\text{HCO}_3^-$  hyperpolarized Vap by about 19 mV and increased Gap by about 11 mS/cm<sup>2</sup> (Table 1b). Changes to cytosolic  $\text{Cl}^-$  hyperpolarized Vap by 62 mV and increased Gap by about 75 mS/cm<sup>2</sup>. When we held KCl on the cytosolic surface, changing the lumen from gluconate to  $\text{HCO}_3^-$  depolarized Vap by 9 mV and increased Gap by about 15 mS/cm<sup>2</sup> whereas changing to luminal  $\text{Cl}^-$  depolarized Vap by 42 mV and increased Gap by about 56 mS/cm<sup>2</sup>.

These results indicate that at these high concentrations of  $\text{HCO}_3^-$ , a CFTR dependent  $\text{HCO}_3^-$  conductance is present, but low, relative to  $\text{Cl}^-$  conductance. The statistically similar changes in Vap and Gap when the  $\text{Cl}^-$  and  $\text{HCO}_3^-$  gradients are reversed suggest that CFTR does not significantly rectify conductance with either anion in the absence of gluconate. However, the data indicate that in the cytosol, gluconate interacts with a cytosolic portion of the CFTR molecule to effectively block  $\text{HCO}_3^-$  permeation from the luminal surface. Consistent observations of asymmetrical cytosolic blocking have been reported for several organic anions, including gluconate, on single channel CFTR  $\text{Cl}^-$  conductance in membrane patch-clamp studies (18). These effects raise the question of whether native organic anions normally in the cell might exert similar effects physiologically. Such an effect might explain the lack of  $\text{HCO}_3^-$  permeability observed in the intact unpermeabilized native sweat duct (15). Whether or not this is the case, the question remains as to whether the apparent, relatively low  $\text{HCO}_3^-$  permeability of CFTR is adequate to play a physiological role in transporting  $\text{HCO}_3^-$ . It is tempting to speculate that the channel in its native environment may rectify  $\text{HCO}_3^-$  and allow movement out of the cell, but block its re-entry by virtue of cytoplasmic anions. Indeed, relative to normal cells, CF cells isolated from the rat<sup>19</sup> and human duodenum have been reported to recover poorly from alkaline load (20).

Lastly, the uncomfortable question remains as to whether explorations of channel function in the presence of such high  $\text{HCO}_3^-$  concentration are physiologically consonant. The lumen of the pancreatic duct is one of the few, perhaps only site, where concentrations of this magnitude are known to occur. Our data may suggest that at concentrations generally thought to occur in the cell and in the extracellular fluid, CFTR may not be a principal route for  $\text{HCO}_3^-$  permeation through the plasma membrane, especially if significant amounts of blocking anions are present in the cytoplasm. Nevertheless, if CFTR does not function as a physiological conductive

$\text{HCO}_3^-$  permeation pathway, we are left without explanation of how it exerts such a pronounced deleterious effect on  $\text{HCO}_3^-$  transport in CF.

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