organic compounds

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Methyl (2Z)-2-{(2Z)-3-[(cyclopentylidene)amino]-4-oxo-2-phenylimino-1,3-thiazolidin-5-ylidene}acetate

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Key indicators: single-crystal X-ray study; T = 100 K; mean σ (C–C) = 0.002 Å; R factor = 0.029; wR factor = 0.075; data-to-parameter ratio = 13.5.

In the title compound, $C_{17}H_{17}N_3O_3S$, the thiazole ring is nearly planar [maximum deviation = 0.015 (1) Å for the ring N atom] and the cyclopentane ring has a twist conformation. The molecular conformation is stabilized by a hypervalent interaction between the S atom and the ester group carbonyl O atom, with an S···O distance of 2.7931 (10) Å. In the crystal, $C-H \cdots O$ interactions generate chains of molecules propagating along [110] and π - π stacking interactions [centroidcentroid distance = 3.4677 (7) Å] between the thiazole rings organize these chains into (001) layers.

Related literature

For the synthesis and similar structures, see: Akkurt *et al.* (2009); Li *et al.* (2011); Mague *et al.* (2013); Mohamed *et al.* (2013*a,b*); Pomés Hernández *et al.* (1996); Sundar *et al.* (2003). For the general biological significance of thiazolidinone scaffold compounds, see: Pfützner *et al.* (2007); Schianca *et al.* (2012); Jain *et al.* (2012); Lant (1986); Rock *et al.* (1991).



Experimental

Crystal data $C_{17}H_{17}N_3O_3S$ $M_r = 343.41$ Monoclinic, $P2_1/c$ a = 9.9684 (2) Å b = 9.9657 (2) Å c = 16.9818 (3) Å $\beta = 105.9290$ (6)°

Data collection

Bruker D8 VENTURE PHOTON 100 CMOS diffractometer Absorption correction: multi-scan (*SADABS*; Bruker, 2013) $T_{\rm min} = 0.76, T_{\rm max} = 0.84$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.029$ $wR(F^2) = 0.075$ S = 1.082951 reflections 218 parameters
$$\begin{split} & Z = 4 \\ & \text{Cu } K \alpha \text{ radiation} \\ & \mu = 1.96 \text{ mm}^{-1} \\ & T = 100 \text{ K} \\ & 0.17 \times 0.16 \times 0.09 \text{ mm} \end{split}$$

V = 1622.23 (5) Å³

17510 measured reflections 2951 independent reflections 2769 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.023$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C10-H10B\cdotsO1^{i}$ $C11-H11A\cdotsO1^{i}$	0.99 0.99	2.57 2.58	3.2889 (17) 3.2547 (16)	130 125

Symmetry code: (i) -x + 1, -y + 2, -z + 1.

Data collection: *APEX2* (Bruker, 2013); cell refinement: *SAINT* (Bruker, 2013); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS2013* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2013* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *WinGX* (Farrugia, 2012) and *PLATON* (Spek, 2009).

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Supporting information for this paper is available from the IUCr electronic archives (Reference: GK2605).

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supplementary materials

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Methyl (2*Z*)-2-{(2*Z*)-3-[(cyclopentylidene)amino]-4-oxo-2-phenylimino-1,3-thiazolidin-5-ylidene}acetate

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1. Comment

Diversity in the biological response profile of thiazolidinone and analogous scaffolds has attracted much attention to the exploration of this skeleton for a variety of therapeutic applications. Thus, the successful pharmaceutical applications of pioglitazone as a hypoglycemic agent (Pfützner *et al.*, 2007; Schianca *et al.*, 2012), thiazolidomycin activity against streptomyces species (Jain *et al.*, 2012), etozoline as an antihypertensive (Lant, 1986), and ralitoline as a potent anti-convulsant (Rock *et al.*, 1991) have established the wide spectrum potential of the thiazolidinone moiety. As part of our ongoing program of drug design and discovery, we report the synthesis and crystal structure of the title compound.

As shown in Fig. 1, the thiazole ring (S1/N1/C1–C3) of the title compound (I) is nearly planar with a maximum deviation of 0.015 (1) Å for N1. The cyclopentane ring (C7–C11) is twisted around the C9–C10 bond. The dihedral angle between the thiazole and phenyl rings is 53.84 (6)°. The C1–C3–C4–C5, C3–C4–C5–O2, C3–C4–C5–O3, C4–C5–O3–C6 and O2–C5–O3–C6 torsion angles are 178.90 (12), 6.7 (2), -174.42 (12), -178.11 (11) and 2.99 (19)°, respectively. All the bond lengths in (I) are comparable to those observed in similar structures (Akkurt *et al.*, 2009; Li *et al.*, 2011; Mague *et al.*, 2013; Mohamed *et al.*, 2013*a,b*; Pomés Hernández *et al.*, 1996; Sundar *et al.*, 2003).

Molecular conformation of (I) is stabilized by a short intramolecular (hypervalent) contact between the S1 atom and the ester group carbonyl O2 atom of 2.7931 (10) Å. In the crystal packing, the C—H…O interactions (Table 1, Fig. 2) with H…O distance of 2.57 - 2.58 Å generate chains of molecules propagating along the [110] direction. Furthermore, π - π stacking interactions [*Cg*1…*Cg*1 (1 - *x*, 1 - *y*, 1 - *z*) = 3.4677 (7) Å; where *Cg*1 is a centroid of the S1/N1/C1–C3 ring] between the thiazole rings organize these chains into the (001) layers.

2. Experimental

A solution of 1 mmol (233 mg) 2-cyclopentylidene-*N*-phenylhydrazinecarbothioamide in 15 ml ethanol was added dropwise to a solution of 1 mmol (142 mg) dimethyl but-2-ynedioate in 10 ml ethanol. The reaction mixture was stirred and refluxed at 351 K. The reaction progress was monitored by TLC until completion. On cooling a solid yellow product was precipitated, filtered off under vacuum and recrystallized from ethanol to furnish block-shaped yellow crystals (m.p. 541–543 K).

3. Refinement

All H atoms were positioned geometrically and treated as riding atoms, with C—H = 0.95 Å (aromatic H), 0.98 Å (methyl H) and 0.99 Å (methylene H), with $U_{iso}(H) = 1.5 U_{iso}(C)$ for methyl H atoms and $U_{iso}(H) = 1.2 U_{iso}(C)$ for the others. The components of the displacement parameters in the direction of the bond between non-hydrogen atoms were restrained to be equal within an effective standard deviation of 0.01 (DELU instruction).

Computing details

Data collection: *APEX2* (Bruker, 2013); cell refinement: *SAINT* (Bruker, 2013); data reduction: *SAINT* (Bruker, 2013); program(s) used to solve structure: *SHELXS2013* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL2013* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *WinGX* (Farrugia, 2012) and *PLATON* (Spek, 2009).



Figure 1

Perspective view of the title compound with 50% probability displacement ellipsoids.



Figure 2

Crystal packing of the title compound viewed down the *a* axis. Only hydrogen atoms involved in C-H···O interactions (dashed lines) are shown.

Methyl (2Z)-2-{(2Z)-3-[(cyclopentylidene)amino]-4-oxo-2-phenylimino-1,3-thiazolidin-5-ylidene}acetate

F(000) = 720
$D_{\rm x} = 1.406 {\rm ~Mg} {\rm ~m}^{-3}$
Cu <i>K</i> α radiation, $\lambda = 1.54178$ Å
Cell parameters from 9943 reflections
$\theta = 4.4 - 68.3^{\circ}$
$\mu = 1.96 \text{ mm}^{-1}$
T = 100 K
Block, yellow
$0.17 \times 0.16 \times 0.09 \text{ mm}$

Data collection

Bruker D8 VENTURE PHOTON 100 CMOS diffractometer Radiation source: INCOATEC I μ S micro-focus source Mirror monochromator Detector resolution: 10.4167 pixels mm ⁻¹ ω scans Absorption correction: multi-scan	$T_{\min} = 0.76, T_{\max} = 0.84$ 17510 measured reflections 2951 independent reflections 2769 reflections with $I > 2\sigma(I)$ $R_{int} = 0.023$ $\theta_{\max} = 68.2^{\circ}, \theta_{\min} = 4.6^{\circ}$ $h = -12 \rightarrow 12$ $k = -11 \rightarrow 12$
(SADABS; Bruker, 2013)	$l = -20 \rightarrow 20$
Refinement	
Refinement on F^2 Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.029$	H-atom parameters constrained
$wR(F^2) = 0.075$	$w = 1/[\sigma^2(F_o^2) + (0.038P)^2 + 0.6771P]$
S = 1.08	where $P = (F_o^2 + 2F_c^2)/3$
2951 reflections	$(\Delta/\sigma)_{\rm max} < 0.001$
218 parameters	$\Delta ho_{ m max} = 0.30 \ { m e} \ { m \AA}^{-3}$
61 restraints	$\Delta \rho_{\rm min} = -0.21 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. Bond distances, angles *etc*. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted *R*-factors *wR* and all goodnesses of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating *-R*-factor-obs *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
S1	0.49073 (3)	0.41667 (3)	0.35998 (2)	0.0187 (1)	
01	0.66547 (10)	0.71951 (9)	0.49052 (6)	0.0244 (3)	
O2	0.72709 (10)	0.25889 (9)	0.37200 (6)	0.0258 (3)	
03	0.93931 (9)	0.30447 (9)	0.45742 (6)	0.0238 (3)	
N1	0.44614 (11)	0.65036 (10)	0.41830 (6)	0.0190 (3)	
N2	0.36915 (11)	0.75447 (11)	0.44449 (7)	0.0214 (3)	
N3	0.24379 (11)	0.55652 (11)	0.33548 (7)	0.0204 (3)	
C1	0.58785 (13)	0.63691 (13)	0.44889 (8)	0.0193 (3)	
C2	0.37387 (13)	0.54855 (12)	0.36801 (8)	0.0184 (3)	
C3	0.63174 (13)	0.50466 (13)	0.42232 (7)	0.0186 (3)	
C4	0.76427 (13)	0.46352 (13)	0.44650 (8)	0.0202 (4)	
C5	0.80432 (13)	0.33216 (13)	0.42077 (8)	0.0207 (4)	
C6	0.98665 (15)	0.17461 (14)	0.43752 (9)	0.0276 (4)	
C7	0.36894 (13)	0.86381 (13)	0.40585 (8)	0.0202 (4)	
C8	0.43371 (17)	0.89572 (14)	0.33790 (9)	0.0298 (4)	
C9	0.41616 (18)	1.04928 (15)	0.32757 (11)	0.0364 (5)	
C10	0.28241 (16)	1.07780 (14)	0.35110 (9)	0.0300 (4)	
C11	0.29241 (14)	0.98521 (13)	0.42434 (8)	0.0241 (4)	

C12	0.17495 (13)	0.45669 (13)	0.27910 (8)	0.0209 (3)
C13	0.05398 (14)	0.39885 (14)	0.28986 (9)	0.0249 (4)
C14	-0.01893 (15)	0.30450 (15)	0.23424 (9)	0.0283 (4)
C15	0.02709 (15)	0.26783 (14)	0.16740 (9)	0.0291 (4)
C16	0.14570 (15)	0.32725 (16)	0.15559 (9)	0.0300 (4)
C17	0.21879 (14)	0.42310 (14)	0.21052 (8)	0.0252 (4)
H4	0.83320	0.51970	0.48060	0.0240*
H6A	0.94260	0.10380	0.46180	0.0410*
H6B	1.08820	0.16880	0.45940	0.0410*
H6C	0.96120	0.16350	0.37790	0.0410*
H8A	0.38460	0.84850	0.28680	0.0360*
H8B	0.53350	0.87020	0.35310	0.0360*
H9A	0.49630	1.09710	0.36430	0.0440*
H9B	0.40730	1.07640	0.27030	0.0440*
H10A	0.19940	1.05620	0.30550	0.0360*
H10B	0.27760	1.17310	0.36660	0.0360*
H11A	0.34520	1.02850	0.47600	0.0290*
H11B	0.19850	0.96050	0.42870	0.0290*
H13	0.02130	0.42400	0.33530	0.0300*
H14	-0.10100	0.26480	0.24210	0.0340*
H15	-0.02240	0.20230	0.12990	0.0350*
H16	0.17740	0.30250	0.10970	0.0360*
H17	0.29840	0.46550	0.20120	0.0300*

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S 1	0.0170 (2)	0.0147 (2)	0.0233 (2)	0.0017(1)	0.0036 (1)	-0.0023 (1)
01	0.0223 (5)	0.0191 (5)	0.0305 (5)	-0.0023 (4)	0.0053 (4)	-0.0060 (4)
O2	0.0216 (5)	0.0233 (5)	0.0304 (5)	0.0028 (4)	0.0037 (4)	-0.0056 (4)
O3	0.0179 (5)	0.0226 (5)	0.0294 (5)	0.0051 (4)	0.0040 (4)	-0.0028 (4)
N1	0.0194 (5)	0.0141 (5)	0.0238 (5)	0.0017 (4)	0.0063 (4)	-0.0014 (4)
N2	0.0222 (6)	0.0160 (5)	0.0274 (6)	0.0036 (4)	0.0092 (4)	-0.0026 (4)
N3	0.0181 (5)	0.0181 (5)	0.0245 (6)	0.0016 (4)	0.0051 (4)	0.0009 (4)
C1	0.0200 (6)	0.0176 (6)	0.0211 (6)	0.0003 (5)	0.0072 (5)	0.0012 (5)
C2	0.0210 (6)	0.0146 (6)	0.0206 (6)	0.0016 (5)	0.0076 (5)	0.0014 (5)
C3	0.0210 (6)	0.0161 (6)	0.0191 (6)	-0.0013 (5)	0.0063 (5)	0.0000 (5)
C4	0.0188 (6)	0.0193 (7)	0.0218 (6)	-0.0009(5)	0.0043 (5)	-0.0006(5)
C5	0.0178 (6)	0.0219 (7)	0.0223 (6)	0.0016 (5)	0.0054 (5)	0.0016 (5)
C6	0.0247 (7)	0.0240 (7)	0.0329 (7)	0.0085 (6)	0.0060 (6)	-0.0022 (6)
C7	0.0170 (6)	0.0182 (7)	0.0239 (6)	-0.0004(5)	0.0032 (5)	-0.0041 (5)
C8	0.0371 (8)	0.0209 (7)	0.0371 (8)	0.0065 (6)	0.0197 (7)	0.0048 (6)
C9	0.0487 (10)	0.0221 (8)	0.0451 (9)	0.0068 (7)	0.0243 (8)	0.0076 (7)
C10	0.0359 (8)	0.0185 (7)	0.0347 (8)	0.0058 (6)	0.0084 (6)	0.0005 (6)
C11	0.0253 (7)	0.0172 (7)	0.0302 (7)	0.0034 (5)	0.0084 (5)	-0.0032 (5)
C12	0.0180 (6)	0.0174 (6)	0.0247 (6)	0.0037 (5)	0.0015 (5)	0.0024 (5)
C13	0.0220 (7)	0.0241 (7)	0.0281 (7)	0.0000 (5)	0.0062 (6)	0.0009 (6)
C14	0.0235 (7)	0.0249 (7)	0.0336 (8)	-0.0032 (6)	0.0029 (6)	0.0028 (6)
C15	0.0286 (7)	0.0230 (7)	0.0295 (7)	-0.0002 (6)	-0.0025 (6)	-0.0017 (6)
C16	0.0280 (7)	0.0351 (8)	0.0242 (7)	0.0060 (6)	0.0024 (6)	-0.0035 (6)

supplemen						ntary materials
C17	0.0200 (7)	0.0279 (7)	0.0262 (7)	0.0016 (5)	0.0038 (5)	0.0005 (6)
Geomet	ric parameters (Å,	, °)				
S1—C2	2	1.7861 (13)	1	C13—C14		1.388 (2)
S1—C3	5	1.7461 (13)	1	C14—C15		1.385 (2)
01—C	1	1.2148 (16)	1	C15—C16		1.385 (2)
O2—C5	5	1.2091 (16)	1	C16—C17		1.393 (2)
O3—C5	5	1.3471 (16)	1	C4—H4		0.9500
03—Се	5	1.4489 (17)	1	С6—Н6А		0.9800
N1N2	2	1.4320 (15)	1	C6—H6B		0.9800
N1-C	1	1.3713 (17)	1	С6—Н6С		0.9800
N1-C2	2	1.3929 (16)	1	C8—H8A		0.9900
N2-C2	7	1.2717 (17)	1	C8—H8B		0.9900
N3—C2	2	1.2648 (18)	1	С9—Н9А		0.9900
N3—C	12	1.4201 (17)	1	С9—Н9В		0.9900
C1C3	3	1.4968 (18)	1	C10—H10A		0.9900
C3—C4	1	1.3358 (19)		C10—H10B		0.9900
C4—C5	5	1.4696 (18)	1	C11—H11A		0.9900
С7—С8	3	1.503 (2)		C11—H11B		0.9900
C7—C1	11	1.5086 (19)	1	C13—H13		0.9500
C8—C9)	1.545 (2)		C14—H14		0.9500
С9—С1	10	1.520 (2)		C15—H15		0.9500
C10—C	211	1.530 (2)		C16—H16		0.9500
C12—C	C13	1.393 (2)		C17—H17		0.9500
C12—C	217	1.3924 (19)	1			
C2—S1	—С3	91.01 (6)		O3—C6—H6A		109.00
C5—O3	3—С6	115.04 (10)		O3—C6—H6B		109.00
N2—N	1—C1	122.50 (10)	1	O3—C6—H6C		109.00
N2—N	1—C2	119.15 (11)		H6A—C6—H6B		109.00
C1—N	1—C2	117.82 (11)		H6A—C6—H6C		110.00
N1N2	2—С7	112.69 (11)		H6B—C6—H6C		109.00
C2—N3	3—C12	119.89 (11)		С7—С8—Н8А		111.00
01—C	1—N1	125.42 (12)	1	С7—С8—Н8В		111.00
01—C	1—C3	125.44 (12))	С9—С8—Н8А		111.00
N1C1	1—C3	109.14 (11)		С9—С8—Н8В		111.00
S1—C2	2—N1	110.15 (9)		H8A—C8—H8B		109.00
S1—C2	2—N3	128.67 (10)	1	С8—С9—Н9А		111.00
N1-C2	2—N3	121.19 (12)	1	С8—С9—Н9В		111.00
S1—C3	G-C1	111.82 (9)		С10—С9—Н9А		111.00
S1—C3	6—C4	126.58 (10)	1	С10—С9—Н9В		111.00
C1-C3	3—C4	121.59 (12)	1	H9A—C9—H9B		109.00
C3—C4	4—C5	120.56 (12)	1	C9—C10—H10A		111.00
O2—C3	5—03	124.08 (12)	1	C9—C10—H10B		111.00
O2—C3	5—C4	124.60 (12)	1	С11—С10—Н10А		111.00
O3—C3	5—C4	111.32 (11)		C11—C10—H10B		111.00
N2-C	7—С8	129.59 (12)	1	H10A—C10—H10B	3	109.00
N2-C2	7—C11	120.60 (12)	1	C7—C11—H11A		111.00
C8—C7	7—C11	109.79 (11)		C7—C11—H11B		111.00

C7—C8—C9	103.75 (12)	C10-C11-H11A	111.00
C8—C9—C10	103.62 (13)	C10-C11-H11B	111.00
C9—C10—C11	103.58 (12)	H11A—C11—H11B	109.00
C7—C11—C10	103.79 (11)	С12—С13—Н13	120.00
N3—C12—C13	118.49 (12)	C14—C13—H13	120.00
N3—C12—C17	121.86 (12)	C13—C14—H14	120.00
C13—C12—C17	119.47 (12)	C15—C14—H14	120.00
C12—C13—C14	120.09 (13)	C14—C15—H15	120.00
C13—C14—C15	120.46 (14)	С16—С15—Н15	120.00
C14—C15—C16	119.58 (14)	C15—C16—H16	120.00
C15—C16—C17	120.43 (14)	С17—С16—Н16	120.00
C12—C17—C16	119.90 (13)	С12—С17—Н17	120.00
C3—C4—H4	120.00	С16—С17—Н17	120.00
C5—C4—H4	120.00		
C3—S1—C2—N1	-1.19 (9)	O1—C1—C3—C4	3.2 (2)
C3—S1—C2—N3	179.13 (13)	N1—C1—C3—S1	1.81 (13)
C2—S1—C3—C1	-0.35 (10)	N1—C1—C3—C4	-177.24 (12)
C2—S1—C3—C4	178.65 (12)	S1—C3—C4—C5	0.00 (19)
C6—O3—C5—O2	-2.99 (19)	C1—C3—C4—C5	178.90 (12)
C6—O3—C5—C4	178.11 (11)	C3—C4—C5—O2	6.7 (2)
C1—N1—N2—C7	84.66 (15)	C3—C4—C5—O3	-174.42 (12)
C2—N1—N2—C7	-103.83 (13)	N2	-171.37 (15)
N2—N1—C1—O1	-11.7 (2)	C11—C7—C8—C9	10.31 (16)
N2—N1—C1—C3	168.71 (10)	N2-C7-C11-C10	-164.51 (13)
C2-N1-C1-01	176.71 (13)	C8—C7—C11—C10	14.00 (15)
C2—N1—C1—C3	-2.90 (15)	C7—C8—C9—C10	-30.79 (16)
N2—N1—C2—S1	-169.23 (8)	C8—C9—C10—C11	39.85 (15)
N2—N1—C2—N3	10.49 (18)	C9—C10—C11—C7	-33.10 (14)
C1—N1—C2—S1	2.68 (14)	N3-C12-C13-C14	177.64 (13)
C1—N1—C2—N3	-177.61 (12)	C17—C12—C13—C14	2.6 (2)
N1—N2—C7—C8	1.4 (2)	N3—C12—C17—C16	-178.17 (13)
N1—N2—C7—C11	179.58 (11)	C13—C12—C17—C16	-3.2 (2)
C12—N3—C2—S1	-5.62 (19)	C12-C13-C14-C15	-0.5 (2)
C12—N3—C2—N1	174.72 (11)	C13—C14—C15—C16	-0.9 (2)
C2—N3—C12—C13	130.97 (14)	C14—C15—C16—C17	0.2 (2)
C2—N3—C12—C17	-54.06 (18)	C15—C16—C17—C12	1.9 (2)
O1—C1—C3—S1	-177.80 (11)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	D—H	H···A	D····A	<i>D</i> —H··· <i>A</i>
C10—H10 <i>B</i> ····O1 ⁱ	0.99	2.57	3.2889 (17)	130
C11—H11A···O1 ⁱ	0.99	2.58	3.2547 (16)	125

Symmetry code: (i) -x+1, -y+2, -z+1.