



Article Lightweight Vapor-Permeable Plasters for Building Repair Detailed Experimental Analysis of the Functional Properties

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Abstract: Three types of lightweight plasters for building repair were prepared and tested. The composition of plasters was designed in respect to their compatibility with materials used in the past in historical masonry. For the hardened plasters, detailed testing of microstructural and macrostructural parameters was realized together with the broad experimental campaign focused on the assessment of mechanical, hygric, and thermal properties. As the researched plasters should find use in salt-laden masonry, specific attention was paid to the testing of their durability against salt crystallization. The mechanical resistance, porosity, water vapor transmission properties, and water transport parameters of all the researched plasters safely met criteria of WTA directive 2-9-04/D and standard EN 998-1 imposed on repair mortars. Moreover, the tested materials were ranked as lightweight plasters and due to their low thermal conductivity they can be used for the improvement of thermal performance of repaired masonry. The salt crystallization test caused little or no damage of the plasters, which was due to their high porosity that provided free space for salt crystallization. The developed plasters can be recommended for application in repair of damp and salt masonry and due to their compatible composition also in historical, culture heritage buildings. The added value of plasters is also their good thermal insulation performance.

Keywords: lightweight plasters; perlite; vapor permeability; salt crystallization resistance; water and salt transport properties

1. Introduction

Many different requirements are placed on the properties of both interior and exterior plasters. Nevertheless, their thermal insulation function has gained importance in recently. As the consumption of energy used for the temperature control in both residential and commercial buildings through heating and air conditioning is still increasing [1], the energy performance of buildings represents the driving force for the improvement and the design of the advanced thermal insulation materials. Worldwide, the buildings and construction sector were responsible for 36% of the total energy use and 39% of energy and process-related CO_2 emissions in 2018 [2]. In the European Union (EU), the heating and cooling of buildings represents around half of the EU's final energy consumption and is the biggest energy end-use sector, ahead of transport and electricity. Moreover, 85% of the energy used for heating and cooling is produced from natural gas, coal, and oil products and only 15% is generated from renewable energy sources [3]. The key targets for 2030 adopted by the European Commission under the 2030 Climate and Energy Framework included at least 40% cuts in greenhouse gas emissions (from 1990 levels) and at least 32.5% improvement in energy efficiency [4].



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The most common way how to reduce the energy demands for heating of older buildings is application of the External Thermal Insulation Composite Systems (ETICS). This solution enables to effectively insulate the whole envelope of the building and reduce thermal bridges that would negatively affect the overall building's hygrothermal performance [5–7]. However, in the case of older and historical buildings, application of ETICS is often forbidden especially due to the requirements of culture heritage authorities that insist on the preservation of the original and decorative appearance of the architectural style of facades and their elements, such as balusters, on brackets balconies, ornamental brackets, pillars and pilasters, gables, etc. Therefore, the compromise solution in the improvement of the hygrothermal performance of such buildings is to enhance their envelopes with appropriate thermal insulation plasters possessing other advanced functional parameters [8]. Many investigations were carried out focused on the integration of different kinds of insulating materials into the composition of plasters in order to achieve their low thermal conductivity and thus thermal insulation efficiency [9–21]. Generally, the authors reported that the addition of lightweight aggregates or fibers into plaster composition greatly improves thermal conductivity, increases porosity, and decreases the mechanical parameters of the hardened plasters. Improvements in sorption properties and water vapor permeability were also reported.

The main aim of this paper is the comprehensive analysis of three types of novel lightweight thermal insulation plasters with enhanced composition containing expanded perlite and designed for interiors, including areas with increased humidity, such as kitchens and bathrooms, or exteriors. The plasters should find use also in repair of historical masonry, where they should act as durable materials for the moderation of the indoor climate of cultural heritage buildings. During restoration works, the compatibility between the new repair mortars and the original components is essential for an adequate intervention on the monument. As pure Portland cement mortars are incompatible with most of the traditional materials inbuilt in historical structures, lime-, natural hydraulic lime-, lime-pozzolan-, gypsum–, and cement–lime-based mortars are considered as materials applicable for the repair of historical masonry and structures. In this sense, the two cement-lime plasters and one lime-gypsum plaster with expanded perlite were characterized in terms of their structural, mechanical, thermal, and hygric parameters. As the durability of plasters is a very important parameter especially in their use in restoration of salt-laden masonry, the tests of salt transport properties were conducted together with the analysis of salt crystallization resistance. To the best of the authors' knowledge, such comprehensive analysis of plasters intended to be used for complex solution of thermal insulation and repair problems was not presented yet and can be considered as a further step for the improvement of eco-efficiency of both contemporary and older building stock. Moreover, the increasing interest in the knowledge of the properties of mortars for restoration purposes justifies the research carried out.

2. Materials and Methods

2.1. Composition of Researched Plasters

Plaster mixtures were prepared from hydrated lime CL 90-S (Čertovy Schody Inc., Tmaň, Czech Republic, member of the Lhoist group), Portland cement CEM I 42.5 R (Českomoravský cement Inc., Mokrá-Horákov, Czech Republic, member of the HeidelbergCement Group), and gypsum binder (Gypstrend Ltd., Kobeřice, Czech Republic). As aggregates, washed quartz sand of 0/1 or 0/2 mm fraction delivered from Filtrační písky, Ltd., Chlum u Doks, Czech Republic (loose bulk density 1668 kg·m⁻³) and expanded perlite EP 150 PB (Perlit Praha Ltd., Praha, Czech Republic) of fraction 0/1 or 0/2 mm and having loose bulk density of 178 kg·m⁻³, were used. Due to the short setting times of gypsum, citric acid monohydrate (Inchema Ltd., Horní Počernice, Czech Republic) was added as a setting retarder.

The chemical composition of the employed materials was obtained from X-Ray Fluorescence (XRF) analysis which was conducted by an ARL QUANT'X EDXRF Spectrometer (Thermo Scientific, Madison, WI, USA), equipped with a Rh X-Ray tube and Si(Li) detector crystal. The data were collected and evaluated using the UniQuant ED 6.32 software (Thermo Scientific, West Palm Beach, FL, USA).

The CLM1 plaster was prepared with a binder-to-aggregate volume ratio of 1:1.15. This volume ratio corresponds with the 1:4 weight ratio commonly used in preparation of lime and cement–lime renders in research and practice [22,23]. This binder to aggregate ratio was also well documented in historical masonry [24–26]. Similar dosage of the blended binders and particular aggregates was then followed in CLM2 and LGM plasters. In the preparation of samples, the use of a constant binder/aggregate volume ratio guarantees an equal proportion of binder and aggregate in the individual mixtures for materials with different loose bulk densities. This is especially important when using lightweight aggregate, such as perlite, where the amount of aggregate would increase disproportionately at a constant weight ratio.

The volume of water was adjusted to maintain the similar and normal consistency of all plasters; flow 160 \pm 5 mm. The flow was verified in accordance with EN 1015-3 [27]. Based on the research on gypsum-based mortars reported in [28], the amount of citric acid was of 0.03 wt. % of the dry compounds. The composition of the investigated plasters is given in Table 1. The casted specimens were demolded after 48 h and stored at temperature 23 \pm 2 °C and relative humidity 50 \pm 5% until their testing. The casted samples were 40 mm \times 40 mm \times 160 mm prisms, 100 mm cubes, and circular plates having a radius of 120 mm and thickness of 30 mm.

Plaster	Lime (g)	Cement (g)	Gypsum (g)	Sand 0/1 (g)	Sand 0/2 (g)	Perlite (g)	Citric Acid (g)	Water (g)
CLM1	50	50	-	70	-	75	-	30
CLM2	45	55	-	-	70	75	-	28
LGM	30	-	70	70	-	75	0.07	45

Table 1. Composition of the investigated plasters.

For the dry plaster mixtures, the particle size was measured using a standard sieve analysis. The sieves with apertures of 0.063, 0.09, 0.125, 0.25, 0.5, 1.0, and 2.0 mm were used.

The 28-days hardened samples were tested. For each batch of plaster, a minimum of 5 samples were examined.

2.2. Assessment of Structural and Microstructural Parameters

Among the macrostructural parameters, bulk density, specific density, and total open porosity were measured. The dry bulk density ρ_b (kg·m⁻³) of dry renders was determined according to the European standard EN 1015-10 [29]. The samples were dried in the vacuum condition at 60 °C. The specific density ρ_s (kg·m⁻³) was assessed using a helium pycnometer Pycnomatic ATC (Porotec, Hofheim, Germany). Based on the knowledge of dry bulk density and specific density values, the total open porosity ψ (-) was calculated [30]. The expanded combined uncertainties of the bulk density, specific density, and porosity determination were 1.4%, 1.2%, and 2.0%. The microstructure of the investigated plasters was analyzed by mercury intrusion porosimetry (MIP), which was conducted by the use of porosimeters of Pascal series, Pascal 140 and Pascal 440 (Thermo Fisher Scientific, Waltham, MA, USA). The measured parameters were average pore diameter, total pore volume, and incremental and cumulative pore volume distributions. The typical dry sample mass in MIP test was approx. 1.0 g.

2.3. Analysis of Mechanical Parameters

Flexural strength, compressive strength, and dynamic modulus of elasticity were the researched mechanical parameters. The strength tests were realized according to the standard EN 1015-11 [31]. In the three-point bending test, the flexural strength f_f (Mpa) was measured on the standard 40 mm × 40 mm × 160 mm prisms. The load speed was 50

 $N \cdot s^{-1}$. The compressive strength f_c (Mpa) was measured on the fragments from the flexural strength test. The uniaxial compression force (100 $N \cdot s^{-1}$) was applied on the 40 mm × 40 mm cross section of the specimens. For the dynamic modulus of elasticity E_d (Gpa) measurement, a Vikasonic apparatus (Schleinbinger Geräte, Buchbach, Germany) was utilized. The mechanical parameters were measured on 5 samples of the particular tested plaster. The expanded combined uncertainty of the mechanical parameters assessment was 1.4%, 1.4%, and 2.3% for f_f , f_c , and E_d , respectively.

2.4. Determination of Thermal Parameters

Heat transport and storage parameters of dry hardened samples were determined by a thermal constants analyzer ISOMET 2114 (Applied Precision, Bratislava, Slovakia) operating on a transient impulse technique principle [32]. For the measurement, circular surface probe IPS 1105 was used. The measurement range of the applied probe was 0.04–0.3 W·m⁻¹·K⁻¹ for the thermal conductivity λ (W·m⁻¹·K⁻¹) and 4.0 × 10⁴–1.5 ×10⁶ J·m⁻³·K⁻¹ for the volume heat capacity c_v (J·m⁻³·K⁻¹).

2.5. Hygric Parameters

The water vapor transmission rate in the examined plasters was quantified using a cup method in both wet-cup and dry-cup arrangements [33]. The tests were conducted according to EN ISO 12572 [34]. For the cup measurements, the circular plate samples having diameter of 120 mm and thickness of 30 mm were conditioned at constant temperature T = (23 ± 0.5) °C and relative humidity RH = (50 ± 5) % until they reached constant mass. The samples were placed in steel cups and then sealed by technical plasticine to ensure 1-D water vapor transport through their exactly known cross-sectional area. In the wet-cup test, the cup contained saturated solution of KNO₃ that generated below the specimen placed in cup RH of (93 \pm 5)%, while in the dry-cup measurement the cup contained silica gel that maintained RH of approx. 2%. The upper side of the specimen was exposed to RH of $(50 \pm 5)\%$ which was maintained by an automatically controlled climate chamber. From the measured steady-state specimen mass gain (dry cup) or loss (wet cup), water vapor permeability δ (s) of a particular plaster was obtained. The water vapor permeability of air δ_a (s) was determined based on the temperature and atmospheric measurements using Schirmer's equation [35]. The water vapor resistance factor μ (-) was calculated as δ_a/δ ratio. The duration of the cup test was approximately 8 days depending on the material permeability for water vapor. The expanded combined uncertainty of the water vapor diffusion test was for the water vapor permeability and the water vapor resistance factor 2.8%.

The plasters' water vapor adsorption capacity was characterized by the measurement of sorption and desorption isotherms. The samples were placed in a set of desiccators that contained saturated solutions that enabled to maintain selected RH. To acquire required relative humidities that covered the whole hygroscopic moisture range, following salts were used: LiCl, K₂CO₃, NaCl, KCl, and K₂SO₄. At the maintained constant temperature *T* = (23 ± 0.5) °C, these salts provided RH of 11%, 43%, 75%, 85%, and 98%, respectively. In every desiccator, 3 dry specimens of each studied plaster were placed on the plastic grid above the saturated salt solution. The 40 mm × 40 mm × 10 mm samples were cut from the standard prisms. The mass of the specimens was monitored until they mass difference was <0.1%. Then, the moisture content by mass was calculated, statistically evaluated, related to the corresponding RH, and one point of sorption/desorption isotherm was plotted. In this way, the adsorption and desorption isotherms were constructed based on the static gravimetric method [36]. The hysteresis in desorption process [37] was also monitored.

The water absorption coefficient A_w (kg·m⁻²·s^{-1/2}) and 24-h water absorption W_a (kg·m⁻²) was assessed in accordance with the EN 1015-18 [38]. The 40 mm cubes were on lateral sides insulated by epoxy resin and their bottom side submerged 5 mm in water. The 24-h water absorption and water absorption coefficient were measured not only for

penetration of water, but also for 1% NaCl and 1% Na₂SO₄ water solutions. The expanded combined uncertainty of both water absorption tests was 1.2%.

2.6. Salt Crystallization Resistance

As to date no commonly accepted methodology for the accelerated salt aging test of porous building materials is available, the testing procedure for the assessment of salt crystallization resistance was originally designed based on the standard EN 12370 [39]. In order to reflect the real situation in practice, two salts were chosen for the test procedure, namely sodium sulfate (anhydrous) Na₂SO₄ and sodium chloride NaCl, and moreover, the concentration of both salts was lower than prescribed in the standard EN 12370 [39], which overestimates salt concentration in the crystallization experiment. According to the recommendations of Lubelli et al. [40] and Granneman et al. [41], the amount of each salt used was chosen to be 2% (weight salt/weight dry specimen). Oven dried specimens having dimensions of 40 mm \times 40 mm \times 40 mm were subjected to 10 crystallization cycles, each cycle consisted of samples immersion in salt solution for 2 h followed by drying in an oven at 70 °C for at least 16 h. After drying, specimens were removed from the oven and left to cool for 2 h. Each sample was placed in its own container, which was during the immersion and cooling phase covered with the cap to prevent evaporation. For the evaluation of the salt crystallization effect, visual observation, light microscopy (LM) analysis, compressive strength tests, and ultrasonic measurement were conducted. The Light Microscopy (LM) was performed by Navitar (Rochester, NY, USA) macro-optics with optical zoom up to 110X and recorded with digital camera Sony 2/3", with a resolution of 5 Mpix. The compressive strength and the dynamic modulus of elasticity tests were conducted in a similar way as described above. Moreover, loss or gain of specimen mass after crystallization cycles was recorded, similarly as prescribed in EN 12370 [39]. Based on the performed experiments, the compressive strength and the dynamic modulus of elasticity ratios of samples that underwent salt crystallization and reference samples were assessed.

Table 2 summarizes a nomenclature of symbols used.

Parameter	Symbol	Unit
Specific density	$ ho_{s}$	(kg⋅m ⁻³)
Bulk density	$ ho_{ m b}$	$(kg \cdot m^{-3})$
Total open porosity	ψ	(%)
Hg porosity	ψ_{Hg}	(%)
Flexural strength	f_{f}	(MPa)
Compressive strength	f_{c}	(MPa)
Dynamic modulus of elasticity	E_{d}	(GPa)
Thermal conductivity	λ	$(W \cdot m^{-1} \cdot K^{-1})$
Thermal diffusivity	а	$(m^2 \cdot s^{-1})$
Volumetric heat capacity	$\mathcal{C}_{\mathbf{V}}$	$(J \cdot m^{-3} \cdot K^{-1})$
Water vapor permeability	δ	(s)
Water vapor resistance factor	μ	(-)
Water absorption coefficient	$A_{ m w}$	$(kg \cdot m^{-2} \cdot s^{-1/2})$
24-h water absorption	Wa	$(kg \cdot m^{-2})$

 Table 2. The nomenclature of used symbols.

3. Results and Discussion

The chemical composition of the employed materials obtained by XRF is introduced in Table 3.

Material	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	CaO	MgO	K ₂ O	Na ₂ O	P_2O_5	TiO ₂	SO ₃
Lime	0.23	0.13	0.16	98.74	0.43	-	-	-	-	0.13
Cement	20.35	4.91	3.24	65.29	1.47	0.91	0.14	0.08	0.45	3.13
Gypsum	6.91	2.47	1.03	42.41	0.68	0.43	-	0.05	0.27	45.60
Quartz sand	98.71	0.45	0.18	0.02	0.03	0.08	0.01	0.03	0.10	0.02
EP	67.72	18.04	1.83	4.34	0.40	2.30	4.43	0.14	0.10	0.10

Table 3. Chemical composition of initial materials (wt.%) obtained by XRF.

The particle size distribution curves of the prepared dry plasters mixtures are plotted in Figure 1. The recorded particle size distribution corresponds to the particle size of raw materials contained in plasters' composition and their dosage.



Figure 1. Particle size distribution of the analyzed plasters.

The results of the conducted tests and analyses were evaluated using the specifications for masonry and repair mortars summarized in EN 998-1 [42] and the WTA directive 2-9-04/D [43].

The macrostructural parameters of the investigated plasters are summarized in Table 3. The agreement between the bulk density and total open porosity values is quite obvious. The WTA directive 2-9-04/D [43] prescribes the bulk density < 1400 (kg·m⁻³) for the repair plaster. Similarly, the porosity must be >40%. Both these conditions were safely encountered for all tested materials. In Table 4, the total open porosity measured by mercury intrusion ψ_{Hg} (-) is also presented. Taking into consideration the principles of the applied total porosity assessment methods, low weight of samples for MIP tests, and inhomogeneity of the tested plasters, the difference in the porosity values can be considered as insignificant.

Table 4. The macrostructural properties of the hardened plasters.

Material	$ ho_{ m s}$ (kg·m $^{-3}$)	$ ho_{ m b}$ (kg·m $^{-3}$)	ψ (%)	ψ_{Hg} (%)
CLM1 CLM2	2587 ± 31 2536 ± 30	1269 ± 18 1120 ± 16	50.9 ± 1.0 55.8 ± 1.1	48.7 54 1
LGM	$\frac{2330 \pm 30}{2389 \pm 29}$	1046 ± 15	56.2 ± 1.1	59.7

The pore size distribution curves obtained by mercury intrusion porosimetry are given in Figure 2 and Figure S1 (Supplementary Materials). Apparently, the microstructural data corresponds with the total porosity data presented in Table 4. The pore size distribution parameters acquired by MIP are presented in Table S1 (Supplementary Materials). They clearly characterize highly open porous structure of the examined materials, which is positive with respect to the requirements imposed on the repair plasters. In the whole studied pore diameter range, the volume of pores was the lowest for plaster CLM1. Contrary to that the relative volume of pores in the recorded pore radii was the highest for LGM. The relative volumes of pores of plaster CLM2 were in the middle.



Figure 2. Incremental pore volume distribution of the researched plasters.

The mechanical resistance of the investigated plasters is apparent from Table 5. These are the results of three combined effects, the nature of the used binder, use of lightweight admixture (perlite), and porosity. The highest flexural strength and compressive strength exhibited lime–gypsum plaster LGM. On the other hand, the dynamic modulus of elasticity of this material was in the middle. Similar performance of lime–gypsum plasters was reported, e.g., in [44,45]. From the practical point of view, all plasters can be classified in category CS II [42] which well satisfies the condition of the WTA directive 2-9-04/D [43]. According to this directive, the compressive strength of repair plasters must be in the 1.5–5.0 MPa. This criterion safely met all prepared plastering mortars.

Material	$f_{\rm f}$ (MPa)	SD *	f _c (MPa)	SD *	E _d (GPa)	SD *
CLM1	0.9	0.10	1.7	0.17	2.3	0.16
CLM2	1.4	0.10	2.4	0.21	3.5	0.19
LGM	1.9	0.16	3.6	0.24	2.7	0.14
*CD standard	d and a time.					

Table 5. The mechanical and thermal properties of the hardened plasters.

* SD—standard deviation.

Heat transport and storage in the investigated plasters were characterized by the thermal conductivity λ (W·m⁻¹·K⁻¹), thermal diffusivity *a* (m²·s⁻¹), and volumetric heat capacity c_v (J·m⁻³·K⁻¹). These parameters are summarized in Table 6. As there is not any strict requirement on the thermal characteristics of plaster intended for repair applications, the thermal insulation potential was evaluated in respect to EN 998-1 [42]. According to this standard, all tested plasters can be ranked as lightweight plasters for interior and in the case of CLM materials also for exterior use. Moreover, lime–gypsum plaster satisfied the criteria imposed on thermal insulation plaster of T2 type. The thermal conductivities of both cement–lime plasters were slightly higher than required limit for T2 ($\lambda < 0.2$), but they were still acceptable for the improvement of thermal performance of repaired masonry.

Material	λ (W·m ⁻¹ ·K ⁻¹)	$a \ (\mathrm{m}^2 \cdot \mathrm{s}^{-1})$	$c_v (J \cdot m^{-3} \cdot K^{-1})$
CLM1	0.227	$0.950 imes 10^{-6}$	$0.239 imes 10^6$
CLM2	0.211	$0.938 imes10^{-6}$	$0.225 imes 10^6$
LGM	0.191	$0.964 imes10^{-6}$	$0.198 imes10^6$

Table 6. The thermal properties of the hardened plasters.

The water vapor transmission properties are important parameters of plasters for repair applications. The results of the cup test that was conducted in both dry cup and wet cup arrangements are presented in Table 7. All materials had the water vapor resistance factor <12.0, which is strictly limited by the WTA directive 2-9-04/D [43]. Such plasters enable water vapor release from the interior and possible drying of the structures suffering from the excessive moisture presence. It must be noted, the water vapor resistance factor criterion of the WTA directive is higher than that prescribed for repair mortars in the EN 998-1 [42] which requires μ < 15.0. The water vapor permeability was higher for wet cup arrangement of the test than that of assessed in the dry cup analysis. Similar material performance in the different conditions of the cup experiment was observed, e.g., in [46–48]. The acceleration of water vapor transmission in the wet cup test can be attributed to the reduced surface binding forces between water vapor molecules and pores due to the filling by water molecules within samples conditioning for the test [49].

Table 7. The water vapor transport parameters of the hardened plasters.

Matorial	Dry C	up	Wet C	up
Iviateriai	$\delta imes 10^{-10}$ (s)	μ (-)	$\delta imes 10^{-10}$ (s)	μ (-)
CLM1	1.73	11.4	2.63	7.5
CLM2	2.03	9.7	2.90	6.8
LGM	1.94	10.2	3.13	6.3

Basically, the water vapor permeability of materials is considered to be dependent on its macrostructural and microstructural parameters and binder nature [50,51]. In our case, not only the total pore volume, but also pore size distribution, their shape, and tortuosity played a role in the water vapor transmission process [52]. Quantitatively, both the examined vapor transport parameters were similar to those published in [53–58].

The sorption and desorption isotherms are plotted in Figure 3. Both the sorption and desorption curves obtained for the tested materials are quite different. Based on IUPAC isotherm classification [59], which provides fundamental guidance how to interpret sorption isotherms for the purpose of structural characterization, the measured sorption/desorption data corresponds to the Type IVa isotherm, typical for mesoporous materials. In the case of a Type IVa isotherm, capillary condensation is accompanied by hysteresis. This occurs, when the pore width exceeds a certain critical width, which is dependent on the adsorption system and temperature.

As expected, the lowest water vapor absorption capacity exhibited plaster CLM1; up to 43% of relative humidity (RH), the gravimetric water content was in the range of detection error. For the higher relative humidity, the moisture content increases slightly versus RH to reach about 0.6%. Plasters CLM2 and LGM were more sensitive to the RH changes of the environment. Since RH > 40%, the capillary condensation arose [60], the micropores and mesopores were filled by water molecules [61], and the water content has strongly increased to reach about 5.1% and 3.6% at 98% RH for plasters CLM2 and LGM, respectively.



Figure 3. Sorption and desorption isotherms of the researched plasters.

The hysteresis was well visible for all researched plasters. In case of CLM1, it reached about 0.4% and it was almost constant in the RH range 11–80%. The hysteresis of CLM2 was in the range 4.3–3.3%, and the desorption curve exhibited linearly decreasing character in the RH range 11–75%. The hysteresis of LGM was about 2.5% and was almost unchanged in the RH range 11–75%. The observed hysteresis in desorption process is usually assigned to the capillary condensation hysteresis [62], the contact angle hysteresis [63], the inkbottle effect [64], or chemical interaction of material with water molecules [65]. A part of the residual moisture may also be due to the partial lime carbonation that could start at high relative humidity. Similar residual mass of lime-based plasters observed, e.g., Mazhoud et al. [58].

The 24 h water absorption W_w and water absorption coefficient A_w assessed in accordance with EN 1015-18 [38] are presented in Table 8. All studied plasters exhibited high water absorption capability and safely met criteria of water absorption rate prescribed in EN 998-1 [42] and WTA directive 2-9-04/D [43]. According to EN 998-1 [42], the repair mortar must have $W_a \ge 0.3 \text{ kg} \cdot \text{m}^{-3}$, which was well fulfilled. Quantitatively, the water ingress into the studied plasters corresponded with the pore size distribution and microstructural parameters, which were the determining factors affecting the overall water imbibition. Namely the volume of capillary pores (0.01–10 µm) in the particular plasters affected the water absorption rate and the total water ingress. Therefore, not the total open porosity, but the share of the volume of capillary pores on overall porosity was the dominant parameter for moisture transport. Similarly, the transport of the tested salt solutions was also governed by the porous structure parameters.

	Water		NaCl		Na_2SO_4	
Material	$A_{\rm w} (\rm kg \cdot m^{-2} \cdot s^{-1/2})$	W_{a} (kg·m ⁻²)	$A_{ m NaCl}$ (kg·m ⁻² ·s ^{-1/2})	W _{NaCl} (kg⋅m ⁻²)	$A_{ m Na2SO4}$ (kg·m ⁻² ·s ^{-1/2})	W_{Na2SO4} (kg·m ⁻²)
CLM1	0.187	12.7	0.178	12.8	0.190	12.2
CLM2	0.149	12.4	0.114	11.7	0.113	12.1
LGM	0.121	12.0	0.110	11.5	0.109	12.0

Table 8. The hygric parameters of the researched plasters.

The differences between the observed hygric parameters assessed for penetration of tap water and 1% NaCl and Na_2SO_4 solutions were small, but some deceleration in the transport of salt solutions in comparison with the transport of pure water can be distinguished.

In Table 9, the mass change of samples subjected to the salt crystallization tests is given. The ratios of the compressive strength and dynamic modulus of samples exposed to 10 wetting drying cycles (water, NaCl, Na₂SO₄) and that of the reference samples are also introduced. The examined plasters showed excellent resistance against crystallization of NaCl and Na₂SO₄ solutions. Both CLM plasters exhibited even improved compressive strength after they underwent crystallization tests and wetting/drying cycles. This was assigned to the continuous hydration of cement/lime binder whose positive contribution to the total mechanical strength prevailed against other effects. The resistance against salt crystallization was caused by the high open porosity of prepared plasters that enabled crystallization of salts from applied solutions in the free porous space without causing damage. Quantitatively bigger problem for the durability of the tested materials appeared for the action of NaCl solution that caused the biggest mass loss and drop in dynamic modulus of elasticity ratio for all the plasters. However, the damage parameters were in this case also small. In summary, considering the results of salt crystallization tests and measured plasters' residual parameters, the prepared plasters can be recommended as plastering materials for salt laden masonry.

Material	Water	NaCl	Na ₂ SO ₄
CLM1	-0.80	-1.85	-0.83
CLM2	-0.41	-0.24	-0.18
LGM	-0.23	-1.02	-0.25
	Compressive s	trength ratio (-)	
CLM1	1.01	1.08	1.19
CLM2	1.18	0.99	1.03
LGM	0.99	0.91	1.02
	Dynamic modulus	of elasticity ratio (-)	
CLM1	0.96	0.88	0.95
CLM2	0.98	0.89	0.99
LGM	0.85	0.90	0.93

Table 9. Mass loss after the salt crystallization test (wt.%), the compressive strength, and dynamic modulus of elasticity ratios of samples subjected to salt crystallization tests and reference samples.

The results of light microscopy imaging are introduced in Figure 4. No cracks or any surface damage was observed on the plasters' fracture surface, which proves the high salt crystallization resistance of the analyzed materials. Photographical observation has not detected any damage of samples subjected to the salt crystallization test as apparent from Figure S2 (Supplementary Materials).



Figure 4. Structure of the hardened plasters analyzed by light microscopy.

4. Conclusions

Lightweight vapor permeable plasters for building repair were designed and tested within the broad experimental campaign. The conducted complex research involved chemical analysis of the base materials, standard sieve analysis of plasters dry mixtures, measurement of macrostructural, microstructural and mechanical properties, and assessment of hygrothermal parameters of the hardened plasters. Specific attention was paid to the analysis of the salt crystallization resistance and determination of the residual strength and stiffness after the wetting/drying cycles in the condition of water, NaCl and Na₂SO₄ solutions. The following main results were obtained and highlighted:

- (i) The bulk density and porosity of the developed plasters met requirements prescribed for repair mortars;
- Based on the compressive strength values, all plasters were classified in category CS II, which is positive with respect to the encountered requirements imposed on the repair plastering materials;
- (iii) The tested plasters can be ranked as lightweight plasters for interior or exterior application (CLM materials). Lime–gypsum plaster was classified as the thermal insulation plaster of T2 type. Although the thermal conductivities of cement–lime plasters slightly exceed the limit for T2, they are still applicable for the improvement of the thermal insulation of repaired masonry;
- (iv) The plasters were found highly permeable for water vapor, which enables the drying of the treated substrates suffering from the excessive moisture action;
- (v) All studied plasters exhibited high water absorption capability and safely met criteria of the water absorption rate;
- (vi) Considering the results of salt crystallization tests and measured plasters' residual parameters, the prepared plasters can be recommended as plastering materials for masonry suffering from salt action.

With respect to the obtained experimental results, it was summarized that the developed materials can find use as lightweight repair plasters possessed of sufficient mechanical strength, high water vapor and water permeability, thermal insulation performance, and durability against salt crystallization and cyclic wetting/drying. They can be therefore applied on salt-laden masonry, and in case of the excessive moisture presence will ensure water evaporation and thus drying the repaired masonry.

Supplementary Materials: The following are available online at https://www.mdpi.com/article/10 .3390/ma14102613/s1, Figure S1: Cumulative pore volume distribution of the researched plasters, Figure S2: Photos of the researched plasters before and after salt crystallization test: (**a**) Plaster CLM1; (**b**) Plaster CLM2; (**c**) Plaster LGM, Table S1: The pore size distribution parameters.

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