

ORIGINAL RESEARCH

Non-invasive and transdermal measurement of blood uric acid level in human by electroporation and reverse iontophoresis

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Correspondence: Congo Tak-Shing Ching Graduate Institute of Biomedicine and Biomedical Technology, National Chi Nan University, No. I University Rd, Puli, Nantou County 54561, Taiwan, Republic of China Tel +886 4929 10960 Ext 4774 Fax +886 4929 12434 Email tsching@ncnu.edu.tw **Abstract:** The aim of this study was to find out the optimum combination of electroporation (EP) and reverse iontophoresis (RI) on noninvasive and transdermal determination of blood uric acid level in humans. EP is the use of high-voltage electric pulse to create nano-channels on the stratum corneum, temporarily and reversibly. RI is the use of small current to facilitate both charged and uncharged molecule transportation across the skin. It is believed that the combination of these two techniques has additional benefits on the molecules' extraction across the human skin. In vitro studies using porcine skin and diffusion cell have indicated that the optimum mode for transdermal uric acid extraction is the combination of RI with symmetrical biphasic direct current (current density = 0.3 mA/cm^2 ; phase duration = 180 s) and EP with 10 pulses per second (voltage = 100 V/cm^2 ; pulse width = 1 ms). This optimum mode was applied to six human subjects. Uric acid was successfully extracted through the subjects' skin into the collection solution. A good correlation ($r^2 = 0.88$) between the subject's blood uric acid level and uric acid concentrations in collection solutions was observed. The results suggest that it may be possible to noninvasively and transdermally determine blood uric acid levels.

Keywords: Reverse iontophoresis, electroporation, uric acid, monitoring, noninvasive, transdermal

Introduction

Gout is one of the most common forms of arthritis,¹ affecting more than 700,000 adults in the UK² and nearly three million adults in the USA,³ and accounting for almost four million outpatient visits every year,⁴ with a substantial economic burden.⁵ Excessive amounts of uric acid, known as hyperuricemia, is one of most frequent metabolism disturbances. Hyperuricemia and high mean serum uric acid concentrations as risk factors for gout have been well documented in the past decades.⁶ Gouty arthritis, a rapid onset of joint inflammation, is a complication of hyperuricemia and is precipitated by deposits of uric acid crystals in the synovial fluid and synovial lining. Recent research has highlighted that patients with hyperuricemia and gout are at increased risk of morbidity and mortality related to cardiovascular disease.⁶ Therefore, blood uric acid monitoring is very important.

The routine method to determine blood uric acid level is achieved by blood sampling. However, this method is invasive, painful, and inconvenient. To the best of our knowledge, there is no noninvasive method to determine the blood uric acid level. Therefore, reverse iontophoresis (RI) and electroporation (EP), noninvasive techniques, were used in this study to transdermally and noninvasively extract uric acid in humans.

RI refers to the passage of a low level of current through the skin to promote the transport of both charged and neutral molecules.8 The main mechanisms that contribute to RI are the electro migration of charged species to the electrode of opposite polarity, electroosmosis of neutral molecules to the cathode or anode, or a combination of both. RI across the skin has been investigated as a noninvasive method for clinical monitoring. Its first clinical application was the transdermal extraction of glucose.10 EP is the use of an intense electric pulse to make the cell membrane transiently porous and then permeable to exogenous molecules present in the surrounding media. 11,12 EP is widely used for the loading of cells with drugs^{13,14} and the delivery of drugs into the cells of living tissues. 15,16 Recently, EP has been used for transcutaneous sampling of molecules, like glucose.¹⁷ Several papers have reported the use of combined RI and EP to enhance the effect of extracting metabolites, ^{18,19} but no papers have been found for the use of combined RI and EP for the extraction of uric acid and this was the novelty of this study. The aim of this study was to find out the optimum combination of EP and RI on noninvasive and transdermal determination of blood uric acid level in humans.

Materials and methods Reagents and solutions

All reagents used in this study were commercially available and used without further purification: phosphate buffered saline and uric acid were purchased from Sigma Chemical Company (St. Louis, MO) and uric acid assay kit (Catalog number: K608–100) from BioVision (Mountain View, CA).

De-ionized water, purified by a Millipore Milli-Q UFplus System (Bedford, MA) was used to prepare all solutions.

Equipment

The diffusion cell for in vitro studies was the same as previously described (Figure 1). The diffusion cell for human studies was newly-designed in this study, with a 5 mm diameter for each electrode chamber and the two electrode chambers were 11 mm apart (Figure 2). RI and EP devices were also newly-designed in this study. The RI device has the accuracy of \pm 0.01 μA on delivering current and \pm 0.01 ms on timing. The EP device has the accuracy of \pm 0.1 V on delivering voltage and \pm 0.01 ms on timing. A Bio-Rad 680 microplate reader (Bio-Rad, Hemel Hempstead, UK) was used for all colorimetric analysis.

Preparation of silver—silver chloride electrodes

Silver–silver chloride (Ag/AgCl) electrodes were used in both in vitro and human studies. The Ag/AgCl electrodes were prepared by chloridizing 99.99% pure silver wire, 1 mm in diameter and 25 mm in length (Aldrich Chemical Company Inc, Milwaukee, WI), immersed in 0.1 M HCl solution (Pt-cathode) for 90 minutes at an applied current of 314 μ A.

In vitro studies

All experiments were conducted using diffusion cells, in which both electrode chambers were located on the same surface side of a porcine skin (250 μ m) obtained by

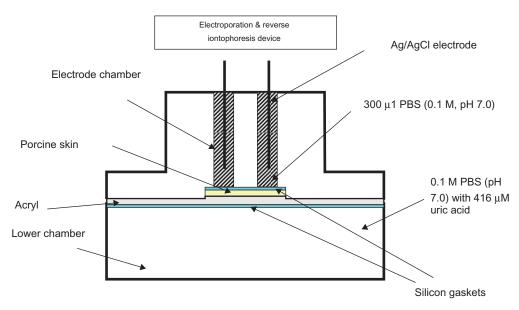


Figure I Diffusion cell for in vitro studies.

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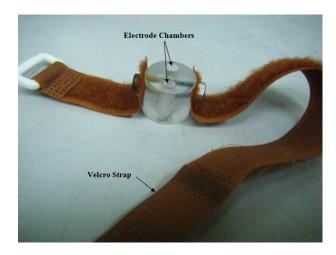


Figure 2 Diffusion cell (diameter = 27 mm) for human studies. Each electrode chamber had a diameter of 5 mm and they were I I mm apart.

dermatome. Both the electrode chambers and the lower chamber of the diffusion cell were filled with 0.1 M phosphate buffer (pH 7.0) while the lower chamber additionally contained 416 µM uric acid. Each electrode chamber contained a Ag/AgCl electrode. The surface area of the porcine skin exposed to the electrode in each chamber was 0.2 cm² and the electrode chambers were 11 mm apart. Different modes of EP and RI (Table 1 and Figure 3) were applied via the Ag/AgCl electrodes. The entire content of the electrode chambers were removed at the end of the experiment to determine the amount of uric acid extracted.

Human studies

Six subjects (four men and two women; mean age 45 ± 13 years) were recruited in this study. The study was approved by The Institutional Review Board of National Chi Nan University. Informed consent was obtained from each subject before the experiment.

The areas of skin of the subject's outer forearm where diffusion cell was to be located were prepared by briskly rubbing the areas for 6–8 s with alcohol preparation pads to remove dry skin, oils, and other contaminants. The areas were then allowed to dry thoroughly. Once the diffusion cell and Ag/AgCl electrodes were fixed in position onto the human forearm, 300 µl phosphate buffer (0.1 M; pH 7.0) was added into each electrode chamber of the diffusion cell. Then, the optimum combination of RI (symmetrical biphasic direct current (dc) with the current density of 0.3 mA/cm² and the phase duration of 180 s) and EP (100 V/cm² electric pulse with the pulse width of 1 ms and 10 pulses per second) was applied via the Ag/AgCl electrodes to the human skin. At the end of the experiment, phosphate buffer was pipetted from

the electrode chambers and stored in microcentrifuge tubes at 4°C for later quantification of uric acid. The subject's blood uric acid level was also measured before experiments at the central laboratory of the Puli Christian Hospital.

Determination of the extracted uric acid

Colorimetric assay was used to determine the amount of uric acid extracted through the porcine or human skin during the in vitro and human studies. An excellent linear relationship $(r^2 > 0.9)$ between uric acid concentrations and its relative absorbance was found, allowing the uric acid concentration to be calculated simply by linear regression.

Statistical analysis

One-way ANOVA was used to determine whether there were significant differences between EP parameters within a RI parameter, and between RI parameters within an EP parameter for the uric acid extraction. Post hoc comparisons were made with independent samples t-tests. A linear regression test was used to determine the relationship between the subject's blood uric acid level and uric acid concentrations in collection solutions. All statistical analyses were carried out using SPSS software (SPSS Inc., Chicago, IL), with the level of statistical significance set at P < 0.05.

Results and discussion

In vitro studies

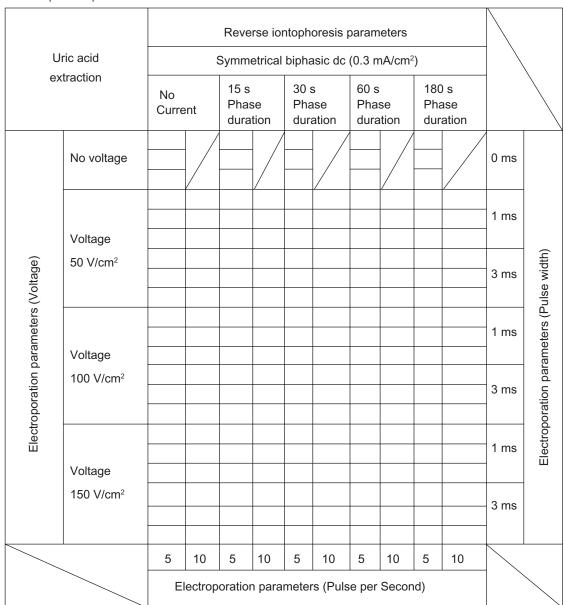
In this study, 195 combinations of RI and EP were chosen for the transdermal uric acid extraction in order to find out the optimum combination of EP and RI on noninvasive and transdermal determination of blood uric acid level in humans.

Uric acid is a negatively-charged molecule and its molecular weight is about 168 Da. Because of its small size and negative charge, uric acid was expected to be transdermally extractable. Electromigration was expected to be the dominant physical factor for the uric acid extraction.

As shown in Figure 4, the application of RI, EP, or the combination of both could facilitate a higher uric acid extraction as compared to diffusion alone. It was found that the uric acid extraction increased as the phase duration of the symmetrical biphasic dc (PDSBdc) increased. At 50 V/cm^2 , 100 V/cm^2 , and 150 V/cm^2 EP, uric acid extraction using the 180 s PDSBdc was found to be significantly higher (P < 0.001 in all cases) than that of protocols utilizing 0 s, 15 s, and 30 s PDSBdc conditions. Although an independent samples t-test revealed no significant difference (P = 0.617) between uric acid extraction using the 60 s and 180 s PDSBdc,

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Table I Experimental protocol for the in vitro studies



Notes: One hundred ninety five different combinations of reverse iontophoresis and electroporation parameters were selected for the in vitro transdermal uric acid extraction, in order to determine the optimum combination for uric acid extraction.

it was observed that the longer the PDSBdc, the better the uric acid extraction was. Therefore, 180 s PDSBdc was recommended as the optimum setting. On the other hand, the combination of RI and EP could facilitate a higher uric acid extraction as compared to RI or EP alone. This was because EP could create reversible nanochannels on the skin^{10,11} and this could further facilitate the uric acid extraction as compared with RI alone. It was also found that the higher the voltage of the EP, the better the uric acid extraction was. This meant that 150 V/cm² EP was better than 50 V/cm² EP (P < 0.001 in all cases of 15 s, 30 s, 60 s, and 180 s PDSBdc), as 150 V/cm² EP had relatively more energy to effectively create more nanochannels. On the other hand, there was no

significant difference (P = 0.413) between the 100 V/cm^2 EP and 150 V/cm^2 EP on the uric acid extraction. 100 V/cm^2 EP was therefore recommended as the optimum setting. Also, no significant difference (P = 0.552) was found between 1 ms and 3 ms pulse width of the EP on the uric acid extraction. As such, a 1 ms pulse width of the EP was recommended as the optimum setting because 1 ms pulse width was commonly used in clinic. Moreover, EP with 10 pulses per second facilitated significantly more uric acid extraction (P < 0.05) as compared with EP with 5 pulses per second. Therefore, EP with 10 pulses per second was recommended as the optimum setting. In a nutshell, the optimum combination of RI and EP on uric acid extraction was as follows:

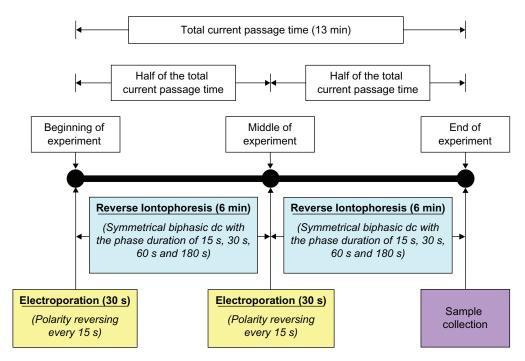


Figure 3 Experimental protocol showing the time sequence for the application of electroporation, reverse iontophoresis, and sample collection.

- RI: symmetrical biphasic dc with a current density of 0.3 mA/cm² and a phase duration of 180 s
- EP: 100 V/cm² electric pulse with a pulse width of 1 ms and 10 pulses per second.

Human studies

No erythema was observed on any subject's skin after the application of RI and EP. In addition, no subject reported pain or even unpleasant feeling during or after the experiment. Therefore, this human study demonstrated that RI and EP

could provide a safe, painless, and non invasive extraction of uric acid in humans.

In this human study, the combination of RI and EP significantly promoted transdermal uric acid extraction, but there are no other human studies for data comparison. As shown in Figure 5, a good correlation ($r^2 = 0.88$) between the subject's blood uric acid level and uric acid concentrations in collection solutions was observed. This meant that the subject's blood uric acid level could be predicted by the uric acid concentrations in collection solutions using the linear regres-

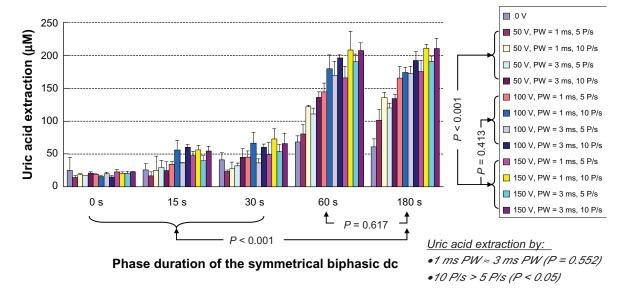


Figure 4 In vitro studies of the transdermal and noninvasive extraction of uric acid by different combinations of reverse iontophoresis (RI) and electroporation (EP). For the RI setting, it was a symmetrical biphasic dc with the current density of 0.3 mA/cm². The legend shows the electroporation setting where PW and P/s are the pulse width and pulse per second of the electroporation, respectively.

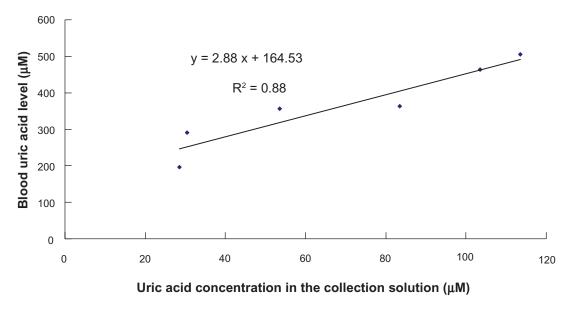


Figure 5 Comparison of real blood uric acid levels of subjects and uric acid concentrations in the collection solution after the application of the optimum combination of reverse iontophoresis and electroporation.

sion equation. Therefore, a transdermal and noninvasive method for blood uric acid monitoring was established.

Despite promising early results, the correlation between the subject's blood uric acid level and uric acid concentration in the collection solution was still not too high. Hence, it is necessary to test this method using larger subject groups in order to achieve a better correlation between the subject's blood uric acid level and uric acid concentration in the collection solution. On the other hand, subjects with a large difference of blood uric acid level are expected to improve the correlation as well.

Conclusion

Uric acid could be transdermally and noninvasively extracted by the combination of RI and EP. The optimum mode was the combination of the RI with symmetrical biphasic dc (current density = 0.3 mA/cm²; phase duration = 180 s) and EP with 10 pulses per second (voltage = 100V/cm²; pulse width = 1 ms). A good correlation between the subject's blood uric acid level and uric acid concentrations in collection solutions was observed. Therefore, a transdermal and noninvasive method for blood uric acid monitoring was established.

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Disclosure

The authors report no conflicts of interest in this work.

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