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Bis[5-methoxy-2-(methoxycarbonyl)-phenyl] methylphosphonate

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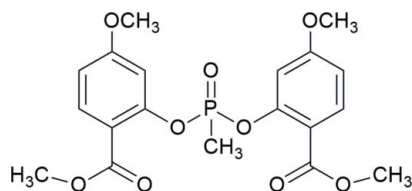
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Key indicators: single-crystal X-ray study; $T = 153$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.040; wR factor = 0.102; data-to-parameter ratio = 21.9.

In the title phosphonate, $\text{C}_{19}\text{H}_{21}\text{O}_9\text{P}$, the dihedral angle between the benzene rings is $63.33(3)^\circ$, and the P atom has a distorted tetrahedral geometry, with angles in the range $101.30(6)$ – $120.38(6)^\circ$. No significant intermolecular interactions are observed in the crystal structure, and π – π interactions between symmetry-related benzene rings are beyond 4 Å.

Related literature

For amidation and esterification of phosphoric acid, see: Kasemsuknimit *et al.* (2011). For the biological activity of phosphonic acids and their ester derivatives, see: Hilderbrand & Henderson (1983); Das *et al.* (2009); Wang *et al.* (2012).



Experimental

Crystal data

 $\text{C}_{19}\text{H}_{21}\text{O}_9\text{P}$
 $M_r = 424.33$ Triclinic, $P\bar{1}$
 $a = 7.852(2)$ Å $b = 10.616(3)$ Å
 $c = 12.200(3)$ Å
 $\alpha = 70.774(9)^\circ$
 $\beta = 77.800(9)^\circ$
 $\gamma = 84.172(10)^\circ$
 $V = 938.0(4)$ Å³ $Z = 2$
Mo $K\alpha$ radiation
 $\mu = 0.20$ mm⁻¹
 $T = 153$ K
 $0.60 \times 0.32 \times 0.23$ mm

Data collection

Rigaku AFC10/Saturn724+
diffractometer
Absorption correction: multi-scan
(*CrystalClear*; Rigaku, 2008)
 $T_{\min} = 0.890$, $T_{\max} = 0.955$ 13574 measured reflections
5875 independent reflections
5141 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.028$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.102$
 $S = 1.00$
5875 reflections268 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.38$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.50$ e Å⁻³

Data collection: *CrystalClear* (Rigaku, 2008); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2008) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *CrystalStructure* (Rigaku, 2008).

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Supporting information for this paper is available from the IUCr electronic archives (Reference: BH2491).

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supplementary materials

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Bis[5-methoxy-2-(methoxycarbonyl)phenyl] methylphosphonate

Saghir Hussain, Yang Deli, Shagufta Parveen, Xin Hao and Changjin Zhu

1. Comment

Phosphonic acids are very important in understanding and modulating some biological processes (Hilderbrand & Henderson, 1983). For instance, aminophosphonate *N*-derivatives are potent and selective inhibitors of protein tyrosine phosphatases (Wang *et al.*, 2012). Within this field, phosphonic esters and their derivatives have attracted interest due to their biological activities (Das *et al.*, 2009; Kasemsuknimit *et al.*, 2011). As an extension of these studies, we report herein on the structure of the title compound, C₁₉H₂₁O₉P.

The molecule (Fig. 1) is lying in general position in a triclinic cell. The molecular conformation is centered on a distorted tetrahedral P atom, and benzene rings of oxo substituents make a dihedral angle of 63.33 (3)°. In the crystal structure (Fig. 2), very weak C—H···O intermolecular contacts exist, which involve P=O and C=O groups as acceptors.

2. Experimental

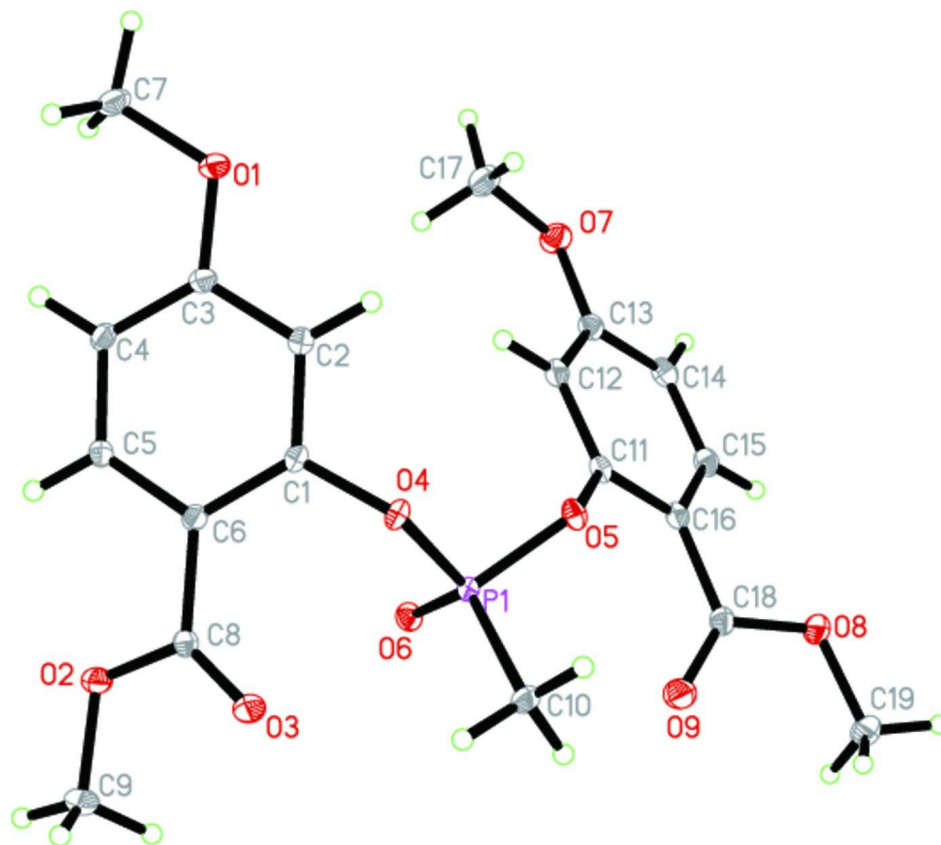
A mixture of diethyl methylphosphite (10 mmol) and phosphoryl chloride (10.5 mmol) was refluxed at 55 °C for 48 h. Then, 4-methoxy-methylsalicylate (10 mmol) in CH₂Cl₂ (10 ml) was added dropwise in the presence of triethylamine (4 mmol) to the viscous yellowish reaction mixture and stirred for 30 minutes at 0 °C, and then at room temperature for 48 h. The reaction mixture was extracted with CH₂Cl₂ (3×10 ml) and H₂O (10 ml). The organic layer was washed with 1M HCl (3×10 ml), H₂O (10 ml), dried over MgSO₄, concentrated, and the residue was purified by column chromatography with petroleum ether and ethyl acetate (3:1) as eluent, affording a colorless solid.

3. Refinement

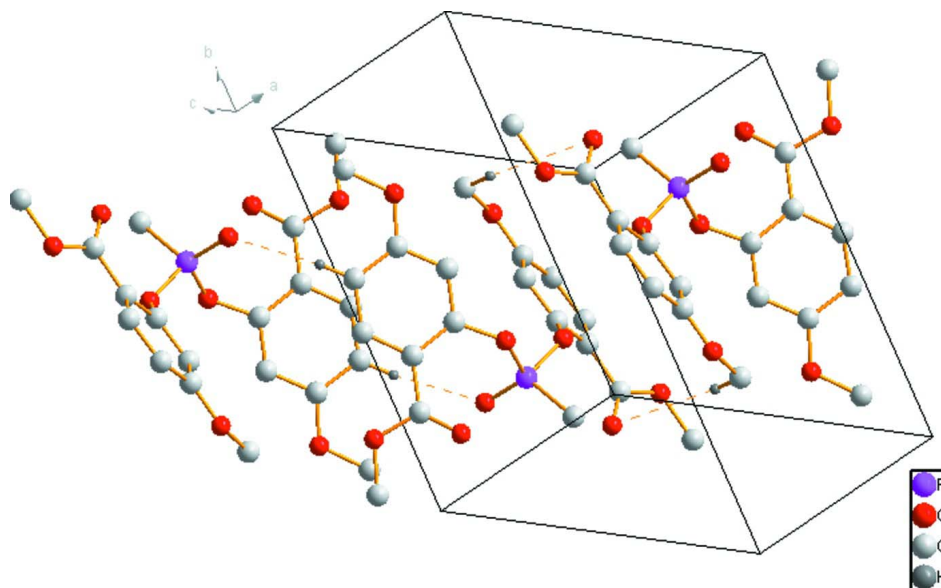
All H atoms were included in the riding model approximation with C—H distances of 0.95 (aromatic CH) or 0.98 Å (methyl CH₃), and with $U_{\text{iso}}(\text{H})=1.2U_{\text{eq}}(\text{C aromatic})$ or $1.5U_{\text{eq}}(\text{C methyl})$.

Computing details

Data collection: *CrystalClear* (Rigaku, 2008); cell refinement: *CrystalClear* (Rigaku, 2008); data reduction: *CrystalClear* (Rigaku, 2008); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2008) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *CrystalStructure* (Rigaku, 2008).

**Figure 1**

The molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level.

**Figure 2**

A view of intermolecular interactions (dotted lines) in the crystal structure of the title compound. H atoms not involved in contacts were omitted for clarity.

Bis[5-methoxy-2-(methoxycarbonyl)phenyl] methylphosphonate

Crystal data

$C_{19}H_{21}O_9P$	$Z = 2$
$M_r = 424.33$	$F(000) = 444$
Triclinic, $P\bar{1}$	$D_x = 1.502 \text{ Mg m}^{-3}$
Hall symbol: -P 1	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 7.852 (2) \text{ \AA}$	Cell parameters from 3814 reflections
$b = 10.616 (3) \text{ \AA}$	$\theta = 2.3\text{--}31.5^\circ$
$c = 12.200 (3) \text{ \AA}$	$\mu = 0.20 \text{ mm}^{-1}$
$\alpha = 70.774 (9)^\circ$	$T = 153 \text{ K}$
$\beta = 77.800 (9)^\circ$	Prism, colorless
$\gamma = 84.172 (10)^\circ$	$0.60 \times 0.32 \times 0.23 \text{ mm}$
$V = 938.0 (4) \text{ \AA}^3$	

Data collection

Rigaku AFC10/Saturn724+ diffractometer	13574 measured reflections
Radiation source: Rotating Anode	5875 independent reflections
Graphite monochromator	5141 reflections with $I > 2\sigma(I)$
Detector resolution: 28.5714 pixels mm^{-1}	$R_{\text{int}} = 0.028$
φ and ω scans	$\theta_{\text{max}} = 31.0^\circ$, $\theta_{\text{min}} = 2.3^\circ$
Absorption correction: multi-scan (<i>CrystalClear</i> ; Rigaku, 2008)	$h = -11 \rightarrow 11$
$T_{\text{min}} = 0.890$, $T_{\text{max}} = 0.955$	$k = -15 \rightarrow 14$
	$l = -17 \rightarrow 16$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.040$	H-atom parameters constrained
$wR(F^2) = 0.102$	$w = 1/[\sigma^2(F_o^2) + (0.0463P)^2 + 0.36P]$
$S = 1.00$	where $P = (F_o^2 + 2F_c^2)/3$
5875 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
268 parameters	$\Delta\rho_{\text{max}} = 0.38 \text{ e \AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.50 \text{ e \AA}^{-3}$
0 constraints	Extinction correction: <i>SHELXL97</i> (Sheldrick, 2008), $F_c^* = kF_c[1 + 0.001 \times F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.018 (2)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
P1	0.33809 (4)	0.27109 (3)	0.77167 (3)	0.01429 (8)
O1	0.32051 (13)	0.73969 (9)	0.92979 (9)	0.0254 (2)
O2	0.05122 (13)	0.15477 (9)	1.15552 (9)	0.0245 (2)
O3	0.29370 (14)	0.10828 (9)	1.03853 (9)	0.0278 (2)
O4	0.42225 (11)	0.33306 (9)	0.85008 (8)	0.01764 (17)
O5	0.40476 (11)	0.36743 (9)	0.64092 (8)	0.01864 (18)
O6	0.14807 (11)	0.26036 (9)	0.80413 (8)	0.01992 (18)
O7	0.12590 (14)	0.79574 (10)	0.48656 (9)	0.0291 (2)
O8	0.33993 (14)	0.27659 (10)	0.35051 (9)	0.0279 (2)
O9	0.32347 (16)	0.17592 (10)	0.54559 (9)	0.0318 (2)

C1	0.33063 (15)	0.40112 (12)	0.92554 (10)	0.0156 (2)
C2	0.36082 (15)	0.53541 (12)	0.89291 (11)	0.0180 (2)
H2	0.4363	0.5776	0.8207	0.022*
C3	0.27920 (15)	0.60907 (12)	0.96717 (11)	0.0177 (2)
C4	0.16437 (16)	0.54847 (12)	1.07129 (11)	0.0183 (2)
H4	0.1058	0.5987	1.1206	0.022*
C5	0.13718 (16)	0.41317 (12)	1.10167 (11)	0.0184 (2)
H5	0.0588	0.3718	1.1728	0.022*
C6	0.22035 (15)	0.33527 (12)	1.03192 (10)	0.0165 (2)
C7	0.24992 (19)	0.81449 (14)	1.00919 (13)	0.0268 (3)
H7A	0.1223	0.8166	1.0222	0.032*
H7B	0.2911	0.9059	0.9747	0.032*
H7C	0.2882	0.7720	1.0848	0.032*
C8	0.19625 (16)	0.18952 (12)	1.07215 (11)	0.0182 (2)
C9	0.0165 (2)	0.01386 (14)	1.19956 (14)	0.0306 (3)
H9A	0.0128	-0.0178	1.1332	0.037*
H9B	-0.0959	-0.0006	1.2544	0.037*
H9C	0.1090	-0.0355	1.2409	0.037*
C10	0.47462 (17)	0.12702 (12)	0.77410 (12)	0.0216 (2)
H10A	0.4780	0.0735	0.8562	0.026*
H10B	0.5927	0.1540	0.7321	0.026*
H10C	0.4286	0.0739	0.7354	0.026*
C11	0.30478 (14)	0.45119 (12)	0.56325 (10)	0.0167 (2)
C12	0.25839 (15)	0.57749 (13)	0.57246 (11)	0.0192 (2)
H12	0.2823	0.6012	0.6361	0.023*
C13	0.17609 (16)	0.66892 (13)	0.48675 (12)	0.0210 (2)
C14	0.14194 (16)	0.63410 (14)	0.39278 (12)	0.0225 (2)
H14	0.0855	0.6965	0.3346	0.027*
C15	0.19083 (16)	0.50850 (13)	0.38535 (11)	0.0215 (2)
H15	0.1691	0.4857	0.3206	0.026*
C16	0.27195 (15)	0.41295 (13)	0.47087 (11)	0.0186 (2)
C17	0.1721 (2)	0.83997 (16)	0.57463 (14)	0.0350 (3)
H17A	0.1159	0.7847	0.6532	0.042*
H17B	0.1331	0.9334	0.5626	0.042*
H17C	0.2989	0.8321	0.5686	0.042*
C18	0.31427 (16)	0.27628 (13)	0.46351 (11)	0.0210 (2)
C19	0.3690 (2)	0.14795 (15)	0.33229 (14)	0.0308 (3)
H19A	0.4530	0.0945	0.3797	0.037*
H19B	0.4152	0.1601	0.2484	0.037*
H19C	0.2586	0.1019	0.3564	0.037*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
P1	0.01521 (13)	0.01433 (14)	0.01354 (15)	-0.00011 (10)	-0.00176 (10)	-0.00540 (11)
O1	0.0306 (5)	0.0164 (4)	0.0281 (5)	-0.0045 (4)	0.0032 (4)	-0.0099 (4)
O2	0.0277 (5)	0.0176 (4)	0.0259 (5)	-0.0049 (3)	0.0029 (4)	-0.0075 (4)
O3	0.0359 (5)	0.0169 (4)	0.0263 (5)	0.0033 (4)	0.0007 (4)	-0.0067 (4)
O4	0.0172 (4)	0.0215 (4)	0.0173 (4)	0.0015 (3)	-0.0030 (3)	-0.0111 (3)
O5	0.0155 (4)	0.0235 (4)	0.0140 (4)	0.0004 (3)	-0.0028 (3)	-0.0024 (3)

O6	0.0168 (4)	0.0226 (4)	0.0200 (4)	-0.0034 (3)	-0.0009 (3)	-0.0069 (4)
O7	0.0344 (5)	0.0224 (5)	0.0298 (5)	0.0054 (4)	-0.0085 (4)	-0.0078 (4)
O8	0.0411 (6)	0.0256 (5)	0.0178 (5)	-0.0018 (4)	-0.0029 (4)	-0.0096 (4)
O9	0.0496 (6)	0.0235 (5)	0.0222 (5)	-0.0008 (4)	-0.0092 (5)	-0.0058 (4)
C1	0.0159 (5)	0.0181 (5)	0.0150 (5)	0.0022 (4)	-0.0041 (4)	-0.0081 (4)
C2	0.0190 (5)	0.0185 (5)	0.0167 (5)	-0.0012 (4)	-0.0027 (4)	-0.0061 (4)
C3	0.0197 (5)	0.0149 (5)	0.0195 (6)	-0.0002 (4)	-0.0048 (4)	-0.0063 (4)
C4	0.0212 (5)	0.0177 (5)	0.0175 (6)	0.0014 (4)	-0.0033 (4)	-0.0083 (4)
C5	0.0213 (5)	0.0182 (5)	0.0150 (5)	-0.0001 (4)	-0.0019 (4)	-0.0055 (4)
C6	0.0188 (5)	0.0157 (5)	0.0155 (5)	0.0009 (4)	-0.0042 (4)	-0.0055 (4)
C7	0.0300 (6)	0.0203 (6)	0.0336 (7)	-0.0014 (5)	-0.0027 (6)	-0.0153 (6)
C8	0.0231 (5)	0.0178 (5)	0.0144 (5)	-0.0004 (4)	-0.0051 (4)	-0.0049 (4)
C9	0.0386 (8)	0.0192 (6)	0.0310 (7)	-0.0095 (5)	0.0006 (6)	-0.0059 (5)
C10	0.0265 (6)	0.0170 (5)	0.0211 (6)	0.0040 (4)	-0.0037 (5)	-0.0076 (5)
C11	0.0140 (4)	0.0206 (5)	0.0130 (5)	-0.0015 (4)	-0.0016 (4)	-0.0022 (4)
C12	0.0175 (5)	0.0229 (6)	0.0168 (6)	-0.0021 (4)	-0.0017 (4)	-0.0064 (5)
C13	0.0179 (5)	0.0208 (6)	0.0213 (6)	-0.0007 (4)	-0.0010 (4)	-0.0045 (5)
C14	0.0201 (5)	0.0256 (6)	0.0187 (6)	-0.0006 (5)	-0.0057 (4)	-0.0016 (5)
C15	0.0208 (5)	0.0266 (6)	0.0165 (6)	-0.0038 (5)	-0.0045 (4)	-0.0046 (5)
C16	0.0173 (5)	0.0220 (6)	0.0157 (5)	-0.0030 (4)	-0.0013 (4)	-0.0053 (5)
C17	0.0497 (9)	0.0257 (7)	0.0301 (8)	0.0022 (6)	-0.0056 (7)	-0.0118 (6)
C18	0.0215 (5)	0.0248 (6)	0.0174 (6)	-0.0031 (4)	-0.0023 (4)	-0.0077 (5)
C19	0.0414 (8)	0.0285 (7)	0.0275 (7)	-0.0033 (6)	-0.0056 (6)	-0.0153 (6)

Geometric parameters (Å, °)

P1—O6	1.4667 (10)	C7—H7A	0.9800
P1—O5	1.5956 (10)	C7—H7B	0.9800
P1—O4	1.5984 (9)	C7—H7C	0.9800
P1—C10	1.7720 (13)	C9—H9A	0.9800
O1—C3	1.3596 (15)	C9—H9B	0.9800
O1—C7	1.4375 (16)	C9—H9C	0.9800
O2—C8	1.3476 (15)	C10—H10A	0.9800
O2—C9	1.4465 (16)	C10—H10B	0.9800
O3—C8	1.2099 (15)	C10—H10C	0.9800
O4—C1	1.3927 (14)	C11—C12	1.3878 (17)
O5—C11	1.3865 (14)	C11—C16	1.3953 (17)
O7—C13	1.3633 (16)	C12—C13	1.3939 (18)
O7—C17	1.4275 (19)	C12—H12	0.9500
O8—C18	1.3494 (16)	C13—C14	1.3971 (19)
O8—C19	1.4428 (17)	C14—C15	1.3764 (19)
O9—C18	1.2063 (17)	C14—H14	0.9500
C1—C2	1.3791 (17)	C15—C16	1.4057 (18)
C1—C6	1.4048 (17)	C15—H15	0.9500
C2—C3	1.4009 (17)	C16—C18	1.4829 (18)
C2—H2	0.9500	C17—H17A	0.9800
C3—C4	1.3914 (17)	C17—H17B	0.9800
C4—C5	1.3872 (17)	C17—H17C	0.9800
C4—H4	0.9500	C19—H19A	0.9800
C5—C6	1.3977 (16)	C19—H19B	0.9800

C5—H5	0.9500	C19—H19C	0.9800
C6—C8	1.4801 (17)		
O6—P1—O5	113.96 (5)	O2—C9—H9C	109.5
O6—P1—O4	114.87 (5)	H9A—C9—H9C	109.5
O5—P1—O4	102.31 (5)	H9B—C9—H9C	109.5
O6—P1—C10	120.38 (6)	P1—C10—H10A	109.5
O5—P1—C10	101.53 (6)	P1—C10—H10B	109.5
O4—P1—C10	101.30 (6)	H10A—C10—H10B	109.5
C3—O1—C7	116.45 (10)	P1—C10—H10C	109.5
C8—O2—C9	115.46 (10)	H10A—C10—H10C	109.5
C1—O4—P1	125.45 (8)	H10B—C10—H10C	109.5
C11—O5—P1	127.58 (8)	O5—C11—C12	117.91 (11)
C13—O7—C17	117.67 (11)	O5—C11—C16	119.54 (11)
C18—O8—C19	116.33 (11)	C12—C11—C16	122.11 (11)
C2—C1—O4	116.28 (10)	C11—C12—C13	118.90 (12)
C2—C1—C6	121.84 (11)	C11—C12—H12	120.6
O4—C1—C6	121.81 (11)	C13—C12—H12	120.6
C1—C2—C3	119.46 (11)	O7—C13—C12	123.89 (12)
C1—C2—H2	120.3	O7—C13—C14	115.62 (12)
C3—C2—H2	120.3	C12—C13—C14	120.48 (12)
O1—C3—C4	124.13 (11)	C15—C14—C13	119.34 (12)
O1—C3—C2	115.46 (11)	C15—C14—H14	120.3
C4—C3—C2	120.41 (11)	C13—C14—H14	120.3
C5—C4—C3	118.65 (11)	C14—C15—C16	121.86 (12)
C5—C4—H4	120.7	C14—C15—H15	119.1
C3—C4—H4	120.7	C16—C15—H15	119.1
C4—C5—C6	122.68 (11)	C11—C16—C15	117.30 (12)
C4—C5—H5	118.7	C11—C16—C18	122.28 (11)
C6—C5—H5	118.7	C15—C16—C18	120.39 (11)
C5—C6—C1	116.89 (11)	O7—C17—H17A	109.5
C5—C6—C8	120.60 (11)	O7—C17—H17B	109.5
C1—C6—C8	122.47 (10)	H17A—C17—H17B	109.5
O1—C7—H7A	109.5	O7—C17—H17C	109.5
O1—C7—H7B	109.5	H17A—C17—H17C	109.5
H7A—C7—H7B	109.5	H17B—C17—H17C	109.5
O1—C7—H7C	109.5	O9—C18—O8	122.93 (12)
H7A—C7—H7C	109.5	O9—C18—C16	126.15 (12)
H7B—C7—H7C	109.5	O8—C18—C16	110.91 (11)
O3—C8—O2	122.46 (12)	O8—C19—H19A	109.5
O3—C8—C6	125.90 (12)	O8—C19—H19B	109.5
O2—C8—C6	111.62 (10)	H19A—C19—H19B	109.5
O2—C9—H9A	109.5	O8—C19—H19C	109.5
O2—C9—H9B	109.5	H19A—C19—H19C	109.5
H9A—C9—H9B	109.5	H19B—C19—H19C	109.5
O6—P1—O4—C1	-14.43 (11)	C1—C6—C8—O3	19.6 (2)
O5—P1—O4—C1	109.57 (10)	C5—C6—C8—O2	20.16 (16)
C10—P1—O4—C1	-145.81 (10)	C1—C6—C8—O2	-162.06 (11)

O6—P1—O5—C11	8.28 (12)	P1—O5—C11—C12	85.25 (13)
O4—P1—O5—C11	-116.33 (10)	P1—O5—C11—C16	-102.25 (12)
C10—P1—O5—C11	139.24 (10)	O5—C11—C12—C13	172.42 (10)
P1—O4—C1—C2	-112.42 (11)	C16—C11—C12—C13	0.12 (18)
P1—O4—C1—C6	70.66 (14)	C17—O7—C13—C12	4.73 (19)
O4—C1—C2—C3	-177.17 (10)	C17—O7—C13—C14	-174.46 (12)
C6—C1—C2—C3	-0.26 (18)	C11—C12—C13—O7	-179.55 (11)
C7—O1—C3—C4	5.11 (18)	C11—C12—C13—C14	-0.39 (18)
C7—O1—C3—C2	-174.95 (11)	O7—C13—C14—C15	179.04 (11)
C1—C2—C3—O1	178.09 (11)	C12—C13—C14—C15	-0.18 (19)
C1—C2—C3—C4	-1.97 (18)	C13—C14—C15—C16	1.06 (19)
O1—C3—C4—C5	-177.96 (12)	O5—C11—C16—C15	-171.47 (10)
C2—C3—C4—C5	2.11 (18)	C12—C11—C16—C15	0.71 (17)
C3—C4—C5—C6	-0.03 (19)	O5—C11—C16—C18	10.56 (17)
C4—C5—C6—C1	-2.08 (18)	C12—C11—C16—C18	-177.27 (11)
C4—C5—C6—C8	175.82 (11)	C14—C15—C16—C11	-1.30 (18)
C2—C1—C6—C5	2.22 (17)	C14—C15—C16—C18	176.71 (12)
O4—C1—C6—C5	178.97 (10)	C19—O8—C18—O9	3.7 (2)
C2—C1—C6—C8	-175.63 (11)	C19—O8—C18—C16	-175.42 (11)
O4—C1—C6—C8	1.11 (17)	C11—C16—C18—O9	26.3 (2)
C9—O2—C8—O3	-1.95 (18)	C15—C16—C18—O9	-151.63 (14)
C9—O2—C8—C6	179.59 (11)	C11—C16—C18—O8	-154.61 (11)
C5—C6—C8—O3	-158.23 (13)	C15—C16—C18—O8	27.48 (16)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C7—H7B \cdots O3 ⁱ	0.98	2.52	3.3134 (18)	138
C9—H9A \cdots O6 ⁱⁱ	0.98	2.79	3.3198 (18)	115

Symmetry codes: (i) $x, y+1, z$; (ii) $-x, -y, -z+2$.