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Enhancement of Mechanical and Bond Properties of Epoxy Adhesives Modified by SiO₂ Nanoparticles with Active Groups

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Abstract: In order to improve the mechanical and bond properties of epoxy adhesives for their wide scope of applications, modified epoxy adhesives were produced in this study with SiO₂ nanoparticles of 20 nm in size, including inactive groups, NH_2 active groups, and C_4H_8 active groups. The mechanical properties of specimens were examined, and an investigation was conducted into the effects of epoxy adhesive modified by three kinds of SiO₂ nanoparticles on the bond properties of carbon fiber reinforced polymer and steel (CFRP/steel) double lap joints. According to scanning electron microscopy (SEM), the distribution effect in epoxy adhesive of SiO_2 nanoparticles modified by active groups was better than that of inactive groups. When the mass fraction of $SiO_2-C_4H_8$ nanoparticles was 0.05%, the tensile strength, tensile modulus, elongation at break, bending strength, flexural modulus, and impact strength of the epoxy adhesives reached their maximum, which were 47.63%, 44.81%, 57.31%, 62.17%, 33.72%, 78.89%, and 68.86% higher than that of the EP, respectively, and 8.45%, 9.52%, 9.24%, 20.22%, 17.76%, 20.18%, and 12.65% higher than that of the inactive groups of SiO₂ nanoparticles, respectively. The SiO₂ nanoparticles modified with NH₂ or C₄H₈ active groups were effective in improving the ultimate load-bearing capacity and bond properties of epoxy adhesives glued to CFRP/steel double lap joints, thus increasing the strain and interface shear stress peak value of the CFRP surface.

Keywords: epoxy adhesives; mechanical properties; SiO₂ nanoparticles; active groups; CFRP/steel double lap joints

1. Introduction

Due to its excellent mechanical properties, bond properties, insulation properties, thermal stability, alkali resistance, stable chemical properties, and low cost, epoxy adhesive is now widely used in bonding metal or non-metal with thermosetting materials [1–6], such as the bonding layer between ultrahigh performance concrete (UHPC) and a thin asphalt layer, or the bond of carbon fiber reinforced polymer (CFRP) with fatigue damaged steel structures. However, such examples require epoxy adhesives to have excellent mechanical properties, surface roughness, and toughness, yet unmodified epoxy adhesive composites are too brittle and vulnerable to cracking [7–9]. The aforementioned properties constrain the applications of epoxy adhesives in civil engineering, military industry, national defense, chemical industry, aerospace, and other high-technology industries [10,11]. Therefore, it is necessary to explore how to improve the mechanical properties of epoxy adhesives effectively [12–18].

A common method of reinforcing and toughening epoxy resin is to introduce a suitable filler. The surface atoms of nanomaterials are highly unsaturated, which makes them capable of reaction with the active groups in the epoxy adhesives to form a force. Therefore, many researchers have shown considerable interest in determining the optimal nanofiller. In [19], the authors studied the effects of adding a mineral filler on the mechanical properties



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of epoxy composites, and it was proven that the addition of ocher to the epoxy composite provides an increase in the thermal stability of the composite. Fouly et al. [20] found that the fracture toughness and ductility were enhanced by 47.3% and 12%, respectively, for the epoxy composite modified by 0.4 wt% alumina nanoparticles. Interestingly, using a fly ash filler classified as industrial waste, the results with fly ash/epoxy composite showed that the tensile strength increased with fly ash content up to a critical threshold, and then decreased with higher ash content [21]. As a result, the composite material leads to the development of more micro-cracks in different directions, acting as a toughening agent [22–33].

However, nanomaterials can cause small size effects and tend to agglomerate when mixed with epoxy resin, thus resulting in a high heterogeneity of composite materials and causing stress concentration in the process of bearing loads. In [34], the field emission scanning electron microscope images confirmed that nanocomposites with organo-modified MMT loading beyond 1 wt% exhibit large tactoid particles and agglomeration. In addition to diminishing the improvement of mechanical properties for composite materials, this also leads to an increased viscosity and difficulties in production [35,36]. The modification of the surface of the nanomaterials by a large number of active groups is capable of improving the compatibility between nanomaterials and epoxy adhesives. The nanomaterial can be effectively dispersed in epoxy adhesives, which is conducive to improving the mechanical properties of the composite [37-41]. In most of the existing research on epoxy adhesive toughening, the focus is placed on adjusting the mix ratio of nanomaterials and functionalizing coupling agents to improve the mechanical properties of epoxy adhesives [42–45]. Selimov et al. [46] examined the effects of silane coupling agents on the dispersion of particles in hybrid carbon fiber composites, and they found that the surface treatments improved the dispersion of the particles and reduced sedimentation. In [47], the authors reported the effects of silane coupling agent (3-aminopropyl triethylsilane) on alumina particles, which were subsequently used as fillers for an epoxy resin matrix. Results from mechanical testing indicated improvements in terms of mechanical properties as a result of the coupling treatment. Aksimentyeva et al. [48] investigated the surface modification of magnetite-polymer nanoparticles by luminescent BaZrO₃ nanocrystals and polyaniline as conducting substances, and the proposed method of surface modification may be used for developing sensors and functional materials for diagnostic methods in medicine. Tsebriienko et al. [49] proposed the poly (titanium oxide) of interpenetrating polymer networks, which showed that the presence of poly (titanium oxide) increases the compatibility of the components of the interpenetrating polymer. Another interesting finding reported in [50] is about the comparison of the mechanical properties of untreated and surface-modified alumina/epoxy nanocomposites, which corroborate the sensitivity of the strength-to-particle surface modification.

SiO₂ nanoparticles are hydrophilic and readily agglomerate in epoxy adhesives due to the fact that the surface of SiO_2 nanoparticles contains a large amount of hydroxyl (–OH). Therefore, it is necessary to modify the surface of SiO_2 nanoparticles to enhance their compatibility with epoxy adhesives and improve the effect of SiO_2 nanoparticles on the mechanical properties of epoxy adhesives. The modification methods of SiO_2 nanoparticles include the alcohol esterification method, the silane coupling agent method, the polymer graft modification method, the surfactant method, the in situ modification method, and the high-energy modification method, among which the most commonly used and traditional method is the silane coupling agent method. The silane coupling agent is a chemical substance with a bidirectional reaction function that can make the bonding interface of epoxy adhesive and SiO_2 nanoparticles into a chemical bond, which can significantly improve the reinforcement performance of SiO_2 nanoparticles. In [51], the study investigated the epoxy-based composites containing SiO₂ modified by various silane coupling agents, which showed that the mechanical properties of the modified composite were improved by incorporation of the silane coupling agents at lower silica content, especially the failure strain. In [52,53], the studies found that the coupling agents form

hydrogen bonds with cellulose chains, which can enhance the interaction energy between nano-SiO₂ and cellulose chains, increasing the adsorption between nano-SiO₂ and cellulose chains. In another interesting study, ref. [54] showed that the dichlorodimethylsilane-modified SiO₂ has a higher content of organic groups and a higher grafting density than the phenyltrimethoxysilane-modified SiO₂. As suggested by Li et al. [30], the modified SiO₂ nanoparticles contain a large number of active groups on the surface, which can improve the impact strength and shear yield of the composite by reacting with the epoxy groups in the epoxy adhesive matrix. To sum up, the introduction of active groups in modified SiO₂ nanoparticles is capable of improving the mechanical properties of epoxy adhesive matrix composites, which provides an effective solution to their reinforcement and toughening.

Allowing for the weakness of the existing research, this paper aims to reveal the impact of several nanomaterials modified by active groups on the mechanical characteristics of adhesives. An investigation was conducted into the mechanical properties of epoxy adhesives as modified by adding three nanoparticles, SiO₂ nanoparticles, SiO₂-NH₂ nanoparticles, and SiO₂-C₄H₈ nanoparticles: the properties investigated include the stress–strain constitutive relationship, tensile strength, elastic modulus, elongation at break, shear strength, impact strength, and so on. In addition to comparing the composition ratio in the epoxy adhesive, ultrasonic cell fragmentation and dispersion technology were applied to further improve the mechanical properties of the nano-modified epoxy adhesive, and scanning electron microscopy (SEM) was used to indicate the wear mechanism. Then, the modified adhesive was applied to a CFRP-reinforced steel plate to reveal the effects of SiO₂ nanoparticles as modified using the inactive groups, and of SiO₂ nanoparticles modified by NH₂ groups and C₄H₈ groups, on the interface bond properties of lap joints, such as ultimate load-bearing capacity, interfacial shear stress, and maximum stress transfer range. Finally, the optimum nanoparticles and extent for steel plate reinforcement were determined.

2. Materials and Methods

2.1. Preparation of Test Materials

Epoxy adhesive (E-51 bisphenol A; specific equivalent weight: 192–196 g/mol; viscosity at 25 °C: 11,000~14,000 mPa.s; density at 25 °C: 1.6–2.3 g/cm³; obtained from Changsha Nan Xing Chemical Technology Co., Ltd., of SINOPEC, Changsha, China); 3glycidoxypropyltrimethoxysilane (KH-560, Changsha Chemical Research Institute, Changsha, China); curing agent (viscosity at 25 °C: 500~1000 mPa.s obtained from Changsha Chemical Research Institute, China); SiO₂ nanoparticles (Bongrui New Material Technology Co., Ltd., Fuyang, China). Before the experiments in this paper, relevant studies have been carried out on SiO₂ nanoparticles with different mass fractions to modify epoxy adhesive, and the mass fractions of 0.02%, 0.05%, 0.10%, 0.20%, and 0.50% were selected as the values to be presented. First, 120 g of epoxy adhesive was added to specified amounts of SiO_2 nanoparticles (0.024 g, 0.06 g, 0.12 g, 0.24 g, 0.60 g) and 3 g of coupling agent. After 10 min of manual dispersion, it was put into an ultrasonic cell crusher (Xinzhi ultrasonic cell crusher was procured from Ningbo Xinzhi Biotechnology Co., Ltd., Ningbo, China) for 2 h. Then, after the mixture was fully dispersed, 35 g of curing agent was added and thoroughly stirred. Next, the mixture was poured into the specimen mold at a uniform speed, and the specimen was placed horizontally indoors and cured at room temperature for 7 days. The production process of the epoxy adhesive specimen is shown in Figure 1, and the detailed parameters of SiO_2 nanoparticles are displayed in Table 1.



Figure 1. Production process of epoxy adhesive specimen.

Fable 1. Detailed par	ameters of SiO ₂ n	anoparticles.
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Types	Particle Size	Specific Surface Area	Volume Density	Density	Active Groups
SiO ₂ -0 SiO ₂ -NH ₂ SiO ₂ -C ₄ H ₈	20 nm	230 m ² /g	0.06 g/cm ³	2.2–2.6 g/cm ³	/ -NH ₂ -C ₄ H ₈

2.2. Measurement of Tensile Properties

The tensile properties of epoxy adhesives were tested in accordance with ASTM D638-10 [55], and five specimens were prepared for each ratio. The dimensions of the specimen are shown in Figure 2.



Figure 2. Dimensions of epoxy adhesive tensile specimen (unit: mm).

The quasi-static tensile properties of epoxy adhesive were tested using a 50 kN electronic universal testing machine (WDW-300D; Ji'nan Ke Sheng Testing Machine Equipment Co., Ltd., Ji'nan, China). The specimen was clamped between the upper and lower fixtures of the testing machine shown in Figure 3. Slippage between specimen and fixtures was prevented. Meanwhile, the central axis of the specimen was consistent with the central line of the upper and lower fixtures to ensure axis loading. Thereafter, the extensometer was installed in the middle section of the specimen. The test was uniformly and continuously loaded at a rate of 2 mm/min until failure. At that time, the failure load value was recorded, and the thickness at the middle and at both ends of the tensile section was measured and averaged.



Figure 3. Tensile test loading diagram.

2.3. Measurement of Flexural Properties

The flexural properties of epoxy adhesives were tested in accordance with ASTM D790-10 [56], and five flexural specimens with dimensions of $10 \text{ mm} \times 5 \text{ mm} \times 120 \text{ mm}$ were made for each ratio. The diagram of the specimen is shown in Figure 4.



Figure 4. Dimensions of epoxy adhesive flexural specimen (unit: mm).

The flexural properties of epoxy adhesives were tested using a 50 kN electronic universal testing machine. A three-point bending test was used for the loading shown in Figure 5. The three loading points of the specimen were 10 mm from the left edge, the center point, and 10 mm from the right edge, respectively. The loading rate and data processing were consistent with the tensile test.



Figure 5. Flexural test loading diagram.

2.4. Measurement of Impact Performance

The impact properties of epoxy adhesives were tested in accordance with GB/T 1043.1-2008 [57], and five impact specimens with type A notch were made for each ratio. A diagram of one such specimen is shown in Figure 6.



Figure 6. Dimensions of epoxy adhesive impact specimen (unit: mm).

The impact performance of epoxy adhesives was tested using a liquid crystal plastic pendulum impact testing machine (LZ21.400-B; Shenzhen Lambo Sansa Material Testing Co., Ltd., Shenzhen, China). A simple beam bearing with a span of 60 mm was selected in the test. The pendulum impact blade was exactly in the middle of the span. When loaded, the specimen extended beyond the left and right supports by 10 mm. The pendulum was turned counterclockwise by hand to the end and then released quickly, and the specimen was struck clockwise by the pendulum.

2.5. Measurement of Shear Performance

The shear properties of epoxy adhesives were examined according to ISO 4587-2003 international standards [58], and five specimens were prepared for each ratio. The dimensions of the specimen are shown in Figure 7.





Figure 7. Dimensions of epoxy adhesive shear specimen (unit: mm).

The shear properties of epoxy adhesives were tested using a 50 kN electronic universal testing machine. To verify that no torque was generated, a piece of equal-thickness steel plate was clamped at both ends. The test was consistently and continuously loaded at a rate of 1 mm/min until failure, and the failure load value was recorded at that time. The loading device is shown in Figure 8.



Figure 8. Shear test loading diagram.

2.6. Measurement of Bond Property

In terms of form and dimension, the design of the specimens for the tensile test of CFRP/steel double lap adhesive joints is referred to as type A in ASTM D3528-96 (R2008) [59]. The form of specimens and the installation position of sensors are shown in Figure 9. The total length of the specimens was 500 mm, the length of the steel plate used as bonding was 350 mm, and the length of the small steel block used as loading clamping was 70 mm. Three specimens were prepared for each ratio.



Figure 9. CFRP/steel double lap joints form and strain gauge arrangement (unit: mm).

The tensile test of CFRP/steel double lap joints was carried out on a 300 kN quasi-static tensile test machine with displacement control for loading at a rate of 0.3 mm/min. The loading device is shown in Figure 10. The stress and strain data were collected by a static strain testing system.



Figure 10. Tensile test of CFRP/steel double lap test piece.

3. Results and Discussion

3.1. Analysis of Mechanical Properties of Epoxy Adhesives Modified by SiO₂ Nanoparticles with *Active Groups*

3.1.1. Tensile Property

Figure 10 shows the relationship between the amounts of SiO_2 nanoparticles with different active groups and the tensile mechanical properties of epoxy adhesives. Figure 11a shows that the tensile strength of the epoxy adhesives were modified by SiO_2 -0 nanoparticles, SiO₂-NH₂ nanoparticles and SiO₂-C₄H₈ nanoparticles. The improvement in tensile strength of epoxy adhesives was greatest at the same mass fraction (0.05%) of SiO₂-NH₂ nanoparticles, SiO₂-C₄H₈ nanoparticles, and SiO₂-0 nanoparticles, which was 59.04 MPa, 59.05 MPa, and 54.45 MPa, respectively. The former two improved by 47.60% and 47.63% when compared to the EP, and improved by 8.43% and 8.45% when compared to SiO₂-0 nanoparticles. The tensile modulus of the epoxy adhesives modified with SiO₂-NH₂ nanoparticles or SiO_2 - C_4H_8 nanoparticles was greater than that of SiO_2 -0 nanoparticles, as shown in Figure 11b. When 0.05% SiO₂-NH₂ nanoparticles or SiO₂-C₄H₈ nanoparticles were added into the epoxy adhesives, the tensile modulus of the epoxy adhesives reached a peak value of 6438 MPa or 6517 MPa, respectively, which was improved by 43.06% and 44.81% when compared to the EP, and improved by 8.20% and 9.52%, respectively, when compared to SiO_2 -0 nanoparticles. The elongation break curves (Figure 11c) were similar to those showing the specimens' tensile strength, with an increased trend in the early stages and a progressive drop in the late stages. The elongation at break of epoxy adhesives modified by SiO_2 -NH₂ nanoparticles, SiO_2 -C₄H₈ nanoparticles or SiO_2 -0 nanoparticles reached its peak value at the same mass fraction (0.05%), which was 2.66%, 2.72% or 2.49%, respectively. The former two were 53.52% and 57.31% higher compared to the EP, and 6.83% and 9.24% higher compared to SiO₂-0 nanoparticles, respectively. With the increase in nanoparticles, the elongation at break of SiO_2 -C₄H₈ decreased very slowly and remained above 2.20%. The addition of small amounts of nanoparticles could greatly improve the tensile properties of epoxy adhesive. With the increase in nanoparticles, the tensile properties of epoxy adhesive tended to increase at first and then decrease. It is worth noting that the tensile strength, the tensile modulus, and the elongation at break of composites were lower than those of EP when the mass fraction of SiO_2 -0 increased to 0.5%. The excessive addition of SiO₂ nanoparticles was difficult to disperse effectively in the epoxy adhesive and prone to agglomeration, which resulted in stress concentration and a sharp decline in the tensile properties of the composites. The tensile properties of epoxy adhesives modified with SiO₂-C₄H₈ nanoparticles or SiO₂-NH₂ nanoparticles were greater than those of SiO₂-0

nanoparticles, a result which indicates that the nanoparticles modified by active groups have better compatibility with epoxy adhesives than those modified by inactive groups.

Figure 11d depicts the stress–strain curves of tensile tests of the EP and three types of SiO_2 nanoparticles with a mass of 0.05%. At the initial stage of loading, the nanoparticles were closely bonded to the resin matrix, no damage occurred in the epoxy adhesives, and the strain–stress showed a linear relationship. As the stress continued to increase, the integrity of the adhesives was gradually destroyed, the strain increment became larger and larger, and the slope decreased gradually. From the graph, it could be seen that the failure stress and slope of the colloidal nanoparticle epoxy adhesives were clearly higher than those of the EP. Meanwhile, SiO₂ nanoparticles containing active groups had a better effect on the tensile properties of the epoxy adhesives than those of SiO₂ nanoparticles containing inactive groups. Therefore, SiO₂ nanoparticles modified by NH₂ and C₄H₈ active groups had little difference in improving the tensile properties of epoxy adhesives, and SiO₂-C₄H₈ nanoparticles were slightly better than SiO₂-NH₂ nanoparticles in improving the tensile properties of the epoxy adhesives.



Figure 11. Relationship between the dosages of SiO_2 nanoparticles with different active groups and tensile mechanical properties of epoxy adhesives: (a) tensile strength; (b) tensile modulus; (c) elongation at break; (d) relationship between stress and strain. (The margin of error is within 5%).

The SEM diagram of the tensile specimen of pure epoxy adhesive was smooth and the crack development was single, as shown in Figure 12a. Figure 12b–d shows that the cross-section of epoxy adhesive with a mass fraction of 0.05% SiO₂ nanoparticles was much coarser than that of EP, and there were obvious folds. The cross-section in Figure 12b contained some aggregation, which indicates that the compatibility of nanoparticles modified by active groups with epoxy resin was better than that of nanoparticles with inactive groups. The SiO₂ nanoparticles in the colloidal section in Figure 12c,d were evenly dispersed, with

irregular and disordered fracture sections and a large number of microcracks, which is typical of ductile fracture morphology. The cross-section of Figure 12e was smoother than that of Figure 12b–d, with a large convex-shaped area on the surface, and the crack development tended to be single. The reason for this is that SiO₂ nanoparticles were not evenly dispersed in the epoxy adhesive and agglomeration occurred, which reduced the colloidal mechanical properties.





Figure 12. Cont.



Figure 12. SEM diagram of the section of the adhesive colloid tensile specimens: (**a**) EP; (**b**) 0.05% SiO₂-0; (**c**) 0.05% SiO₂-NH₂; (**d**) 0.05% SiO₂-C₄H₈; (**e**) 0.50% SiO₂-0.

3.1.2. Flexural Properties

Figure 13 shows the relationship between the amounts of SiO₂ nanoparticles with different active groups and the flexural properties of epoxy adhesives. Figure 13a demonstrates that the bending strength of epoxy adhesives modified with SiO₂ nanoparticles was greater than that of the EP. The bending strength of the epoxy adhesives reached a peak value of 79.40 MPa or 86.71 MPa when 0.05% SiO₂-NH₂ nanoparticles or SiO₂-C₄H₈ nanoparticles were added into the epoxy adhesives, respectively. They were 48.50% and 62.17% higher compared to the EP, and 10.08% and 20.22% higher compared to SiO₂-0 nanoparticles, respectively. The flexural modulus of the epoxy adhesives modified by SiO₂-NH₂ nanoparticles or SiO₂-C₄H₈ nanoparticles was superior to that of SiO₂-0 nanoparticles as shown in Figure 13b. The improvement of epoxy adhesives modified by SiO₂-NH₂ nanoparticles, SiO₂-C₄H₈ nanoparticles, and SiO₂-0 nanoparticles reached its peak value at the same mass fraction (0.05%), which was 4620.97 MPa, 5019.86 MPa, and 4262.88 MPa respectively. Compared with the EP, these improvements were by 23.09%, 33.72%, and 13.55%, respectively.

In general, the addition of a small amount of SiO₂ nanoparticles can greatly improve the bending properties of epoxy adhesives. The bending properties of epoxy adhesives modified by SiO₂ nanoparticles with active groups were better than those with inactive groups. When the mass fraction of SiO₂-C₄H₈ nanoparticles was 0.05%, the bending property of epoxy adhesive reached its best effect. When the mass fraction of SiO₂ nanoparticles exceeded 0.05%, the bending properties of the epoxy adhesives modified bySiO₂-NH₂ nanoparticles decreased faster than those of SiO₂-C₄H₈ nanoparticles. The flexural properties of the composite were lower than those of EP when the mass fraction of SiO₂-NH₂ nanoparticles was 0.50%, which also indicates that the distribution effect of the C₄H₈ active group in epoxy adhesive was better than that of the NH₂ active group.



Figure 13. Relationship between the dosages of SiO_2 nanoparticles with different active groups and bending mechanical properties of epoxy adhesives: (**a**) bending strength; (**b**) flexural modulus. (The margin of error is within 5%).

3.1.3. Impact Performance

At the moment of contact between the impact specimen and the pendulum, the pendulum immediately cut the specimen into two sections along the V-notch, accompanied by a few fragments of sputtering, and the crack extended perpendicularly to the direction of load. As shown in Table 2, the impact strength of the epoxy adhesives modified by SiO₂ nanoparticles under different mass fractions was analyzed. When 0.02% SiO₂ nanoparticles were added to epoxy adhesives, the impact strength of the epoxy adhesives was significantly improved. With the increase in nanoparticles, the impact strength of epoxy adhesives steadily increased, and reached its peak value when the mass fraction of SiO₂ nanoparticles was 0.05%. The epoxy adhesive modified by 0.05% SiO₂-C₄H₈ nanoparticles had the best effect on improving the impact strength (32.76 kJ/m²), which was 78.89% higher compared to the EP (18.31 kJ/m²) and 20.18% higher compared to the SiO₂-0 nanoparticles (27.26 kJ/m²). When the mass fraction of SiO₂ nanoparticles was 0.10%, the impact properties of the three kinds of epoxy adhesives modified by SiO₂ nanoparticles all decreased, which might have been due to the large amounts of SiO₂ nanoparticles that were difficult to disperse effectively in the epoxy adhesive and prone to agglomeration.

Table 2. Impact test data of epoxy adhesives modified by SiO₂ nanoparticles with different active groups.

Mass Fractions (%)	Impact Strength (kJ/m ²)			
	SiO ₂ -NH ₂	SiO_2 - C_4H_8	SiO ₂ -0	
0	18.31	18.31	18.31	
0.02	22.21	23.90	22.59	
0.05	28.74	32.76	27.26	
0.10	25.81	28.27	24.12	

3.1.4. Shear Performance

Figure 14 shows the load-displacement curves of shear specimens modified by 0.05% SiO₂-C₄H₈ nanoparticles, 0.05% SiO₂-NH₂ nanoparticles, 0.05% SiO₂-0 nanoparticles, and the EP. As can be seen in the figure, the load of the specimens grew quickly as displacement increased at the beginning of the test. When the load value reached about 0.3 kN, the load increment tended to zero with increase in displacement, and the curve became flatter. Thereafter, the load increment was linearly correlated with the displacement increment, and the slope was slightly less than at the beginning of the test. The shear strength of the adhesives was greatly improved after being modified by nanoparticles, and the shear strength of epoxy adhesives modified by SiO₂-C₄H₈ nanoparticles or SiO₂-NH₂ nanoparticles was 68.86% (3.52 kN) or 67.93% (3.50 kN) higher compared to the EP (2.09 kN), and 12.65% or 2.03% higher compared to the SiO_2 -0 nanoparticles (3.13 kN), respectively. The shear properties of epoxy adhesive modified by SiO_2 nanoparticles with the two kinds of active groups were basically the same, and both better than those with no active groups. However, the flatter part of the curve for the C_4H_8 active group was significantly longer than that of the NH_2 active group, which might have been caused by the relative slip between the sheared specimen and the clamp.



Figure 14. Load-displacement curves of epoxy adhesives modified by nanoparticles with different active groups during shear test.

3.2.1. Load-Displacement Relationship

Figure 15 shows the load-displacement curves of CFRP/steel double lap specimens in the tensile process. With the increase in displacement, the load increment of the three specimens increased. The shear strength and ductility of the epoxy adhesives were improved to a certain extent after the addition of SiO₂ nanoparticles. It can be seen that there is an abrupt drop in the curves for SiO₂-0 nanoparticles and EP, which was due to relative slip between the loading device and the specimen during the loading process. As the displacement continued to increase, the specimen was re-loaded. The load peak value of the specimen modified by SiO₂-C₄H₈ nanoparticles, SiO₂-NH₂ nanoparticles, and SiO₂-0 nanoparticles was 57.57 kN, 55.55 kN, and 52.09 kN, which was 41.24%, 36.29% and 27.80% higher compared to the EP (40.76 kN), respectively. The bond properties of the epoxy adhesives modified by SiO₂ nanoparticles with the two kinds of active groups were better than those with no active groups, and SiO₂ nanoparticles modified by the C₄H₈ active group showed the best improvement.



Figure 15. Load-displacement curves of CFRP/steel double lap specimens.

3.2.2. Surface Strain Distribution of CFRP Plate

The strain distribution on the CFRP surface is shown in Figure 16. Under ultimate load, a considerable stress gradient of the double lap joints bonded by the EP arose near the free end of the steel plate. The strain of EP at 5 mm from the free end of the steel plate was 1339 $\mu\epsilon$; the strain at 15 mm from the free end of the steel plate dropped to 740 $\mu\epsilon$, decreasing by 44.73%; and the strain at 25 mm from the free end of the steel plate was 572 $\mu\epsilon$, decreasing by 57.28%. When SiO₂-0 nanoparticles with a mass fraction of 0.05% were added into the epoxy adhesives, the strain at 5 mm from the free end of the steel plate was 2750 $\mu\epsilon$; the strain at 15 mm from the free end of the steel plate was 2001 $\mu\epsilon$, decreasing by 27.24%; and the strain at 25 mm from the free end of the steel plate was 1863 $\mu\epsilon$, decreasing by 32.25%. When SiO₂-NH₂ nanoparticles with a mass fraction of 0.05% were added into the epoxy adhesives, the strain at 5 mm from the free end of the steel plate was 3113 $\mu\epsilon$; the strain at 15 mm from the free end of the steel plate was 2617 $\mu\epsilon$, decreasing by 15.93%; and the strain at 25 mm from the free end of the steel plate was 2246 $\mu\epsilon$, decreasing by 27.85%. Similarly, the strain at 5 mm from the free end of the steel plate of the epoxy adhesives modified by 0.05% SiO₂-C₄H₈ nanoparticles was 3230 $\mu\epsilon$; the strain at 15 mm from the free end of the steel plate was 2519 $\mu\epsilon$, decreasing by 22.01%. It is worth noting that the maximum strain at 25 mm from the steel plate's free end was $3002 \ \mu\epsilon$, which was considerably higher than the maximum strain at 15 mm, and a decrease of 7.06% compared to that at 5 mm from the free end of the steel plate. The results indicated that SiO₂ nanoparticles could improve the bond properties of epoxy adhesives, increase the strain and strain gradient range of the CFRP surface, and improve the final bearing capacity

of double lap joints. The effect of $SiO_2-C_4H_8$ nanoparticles and SiO_2-NH_2 nanoparticles resulted in a more effective improvement than that of SiO_2-0 nanoparticles. A tearing sound was noted, which was emitted during the loading procedure of the specimen by the peeling of a small section of the interface. The stripping of the CFRP plate and steel plate occurred as the load increased. A warning signal appeared before the failure of the specimen, which was characteristic of ductile failure.



Figure 16. Strain distribution of CFRP surface: (a) EP; (b) SiO_2 -0 nanoparticles; (c) SiO_2 -NH₂; (d) SiO_2 -C₄H₈ nanoparticles.

4. Conclusions

This paper proposes epoxy adhesives modified by different active groups of SiO₂ nanoparticles. With the introduction of SiO₂-0, SiO₂-NH₂, and SiO₂-C₄H₈ nanoparticles into epoxy adhesives at a mass fraction of 0.05%, the improvement in mechanical properties reached its maximum. When added to modify the epoxy adhesives, the SiO₂-C₄H₈ nanoparticles showed the most significant improvement in mechanical properties, while the SiO₂-0 nanoparticles showed the least improvement. When the mass fraction of SiO₂-C₄H₈ nanoparticles was 0.05%, the tensile strength, tensile modulus, elongation at break, bending strength, flexural modulus, and impact strength were 47.63% (59.05 MPa), 44.81% (6517 MPa), 57.31% (2.72%), 62.17% (86.71 MPa), 33.72% (5019.86 MPa), 78.89% (32.76 kJ/m²), and 68.86% (3.52 kN) higher than that of the EP, respectively. SiO₂ nanoparticles were capable of improving not only the mechanical properties of the epoxy adhesive, but also the strain on CFRP surfaces, the interface shear stress peak, and the transfer range. In general, the mechanical properties and bond properties of epoxy adhesives modified by adding active groups were superior to those modified by adding inactive groups, and the modification effect of C₄H₈ active groups was better compared to NH₂ active groups.

The experiments show that active groups can significantly improve the modification effect of nanoparticles, and that active groups could become a meaningful direction of nanoparticle research for practical engineering applications. The characterization of SiO₂-NH₂ and SiO₂-C₄H₈ (FTIR, XRF, etc.) were further studied in the follow-up study.

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