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# Confinement-Controlled, Either syn- or anti-Selective Catalytic Asymmetric Mukaiyama Aldolizations of Propionaldehyde Enolsilanes

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**ABSTRACT:** Protected aldols (i.e., true aldols derived from aldehydes) with either *syn-* or *anti-* stereochemistry are versatile intermediates in many oligopropionate syntheses. Traditional stereoselective approaches to such aldols typically require several nonstrategic operations. Here we report two highly enantioselective and diastereoselective catalytic Mukaiyama aldol reactions of the TBS- or TES- enolsilanes of propionaldehyde with aromatic aldehydes. Our reactions directly deliver valuable silyl protected propionaldehyde aldols in a catalyst controlled manner, either as *syn-* or *anti-* isomer. We have identified a privileged IDPi catalyst motif that is tailored for controlling these aldolizations with exceptional selectivities. We demonstrate how a single atom modification in the inner core of the IDPi catalyst, replacing a CF<sub>3</sub>-group with a CF<sub>2</sub>H-group, leads to a dramatic switch in enantiofacial differentiation of the aldehyde. The origin of this remarkable effect was attributed to tightening of the catalytic cavity via unconventional C–H hydrogen bonding of the CF<sub>2</sub>H group.

Polyketides are pharmaceutically important secondary metabolites <sup>1</sup> Empharmaceutically important secondary metabolites. Erythromycin is a prototypical example, which as a synthetic target was declared by Woodward as "hopelessly complex ...in view of its plethora of asymmetric centers". This statement encouraged the beginning of several decades of intense and highly innovative method development in acyclic stereocontrol. Generations of chemists have contributed approaches to overcoming the synthetic challenges posed by oligopropionates, typically bearing linear stereopolyads with alternating methyl and hydroxyl groups.3 Nonetheless, only few truly reliable methods have found general utility in numerous syntheses of complex oligopropionates.4 Widely used approaches, based on chiral auxiliaries, rely on diastereoselective asymmetric propionate aldolizations or crotylation reactions (Figure 1).5,6 Very often, both approaches converge after several steps: following the critical diastereoselective C-C bond-formation, protecting group installation and redox manipulations lead to stable protected aldol intermediates of a general structure I, which are ideal for downstream functionalization to construct various polyketide motifs. Catalytic asymmetric crotylation methods developed more recently by Krische et al. provide an attractive alternative but feature a moderate step-economy when protected aldols of type I are needed. Direct stereoselective cross-aldol reactions of aldehydes have also been described, but an additional protection step is often unavoidable.8 A truly practical approach, from a total synthesis chemist's perspective, would directly deliver the protected aldols in a predictable catalytic manner, with full control over diastereoselectivity and enantioselectivity. In this regard, arguably the most useful variant of the Mukaiyama aldol reaction, the aldolization of propionaldehyde-derived enolsilanes with aldehydes, has

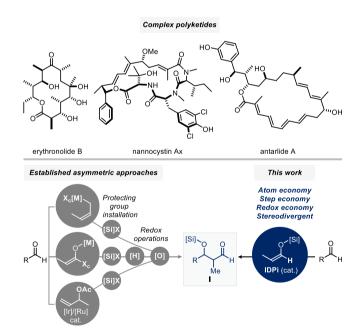


Figure 1. Complex bioactive polyketides. Established approaches to protected aldols I vs our single-step, stereodivergent method.

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remained elusive. In addition to directly delivering the protected aldols, such a reaction would favorably meet all of the metrics used to evaluate synthetic methods, such as atom-, 10 redox-, 11 and especially step-economy. 12

Yamamoto reported nonenantioselective examples using the bulky tris(trimethylsilyl)silyl (TTMSS, supersilyl) group to control selectivity toward single addition of aldehyde-derived enolsilanes. 13 The first diastereo- and enantioselective examples by Denmark utilized trichlorosilyl enolsilanes under Lewis base catalysis. 14 This approach is poorly atom-economic, as the silyl group is not retained in the final product and several steps are required to obtain aldols suitable for chain elongations. Kanai and Matsunaga reported an asymmetric Cu-catalyzed aldolization using in situ generated boron enolates of propionaldehyde, which gives mainly syn-aldols. 15 Although up to quadruple aldolizations were achieved, the existence of unprotected oligoaldols in various cyclic hemiacetal forms limits their selective elaboration to useful oligopropionate motifs. Recently we reported the first, highly enantioselective Mukaiyama cross-aldol reaction with simple triethylsilyl (TES) and tert-butyldimethylsilyl (TBS) enolsilanes of acetaldehyde and aliphatic and aromatic aldehydes using confined and strongly acidic imidodiphosphorimidate (IDPi) catalysts developed here. 16 Enzyme-like discrimination of the small substrate aldehyde over the larger product aldehyde is believed to be the origin of single aldolization without oligomerization. 17 From such a vantage point, exploitation of the unique reactivity of IDPi catalysts in the Mukaiyama cross-aldol reaction with propionaldehyde-derived enolsilanes would be even more valuable as it generates two (vicinal) stereogenic centers simultaneously, which are present in many bioactive polyketides (Figure 1).18 We envisioned developing a fully stereodivergent method giving access to all four stereoisomers. 19 Herein, we report that the IDPi family of catalysts provides a powerful solution to this long-standing goal.2

Our investigations commenced with an exploration of different IDPi catalysts in the aldolization of benzaldehyde 1 with (E)-enolsilanes 2a-b or (Z)-enolsilanes 4a-c (Table 1). At the onset we found that the diastereoselectivity was dependent on the nucleophile geometry, with (E)-enolsilanes providing syn-aldols and (Z)-enolsilanes giving anti-aldols, with varying degrees of diastereoselectivity, depending on the silyl group and the catalyst. IDPi 6, which was a preferred catalyst in our acetaldehyde-derived enolsilane additions, was tested in the reaction of (*E*)-enolsilane **2a** with benzaldehyde at -20 °C. Single aldolization product 3 was indeed formed in high yield, excellent diastereoselectivity (d.r. 97.5:2.5) in favor of the synaldol 3a ([Si] = TES), and with a promising enantiomeric ratio (e.r.) of 10.5:89.5 (Table 1A, entry 1). Given that fluorenesubstituted IDPi catalysts were especially privileged in our recent silylium-ion asymmetric counteranion-directed catalysis (Si-ACDC)<sup>21</sup> methodologies, <sup>22a-d</sup> we turned our attention to catalysts 7a-d (and S8a-c in the Supporting Information). Indeed, these IDPis emerged as preferred catalysts for our reactions. Spirocyclobutane-substituted IDPi 7a markedly stood out in terms of enantioselectivity, providing aldol 3a with a 95.5:4.5 e.r. (Table 1A, entry 2). Modification of the inner core of the IDPi from the Tf-group to a Nf-group further increased diastereoselectivity and enantioselectivity. Lowering the temperature to -40 °C led to a d.r. of 99:1 and a 98:2 e.r. (Table 1A, entry 5). Variation of the silyl group was welltolerated when the TBS-enolsilane 2b was used instead of the

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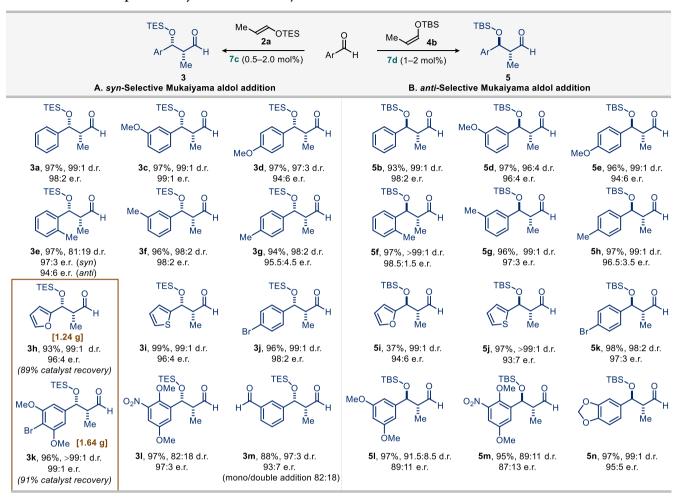
Γable 1. Reaction Development						
A. syn-Selective Mukaiyama aldol addition						
Ph H	н —	IDPi (2	O[Si] <b>n-b</b> mol%), Cl <sub>3</sub> (0.5 N	[Si] \ → Ph \	_	ı: [Si] = TES ɔ: [Si] = TBS
entry <sup>a</sup>	IDPi	[Si]	T (°C)	yield (%) <sup>b</sup>	syn/anti <sup>c</sup>	e.r. <sup>d</sup>
1	6	TES	-20	98	97.5:2.5	10.5:89.5
2	7a	TES	-20	99	95.5:4.5	95.5:4.5
3	7b	TES	-20	99	96:4	96:4
4	7c	TES	-20	99	97:3	97:3
5	7с	TES	-40	99	99:1	98:2
6	7с	TBS	-50	99	99:1	97:3
B. anti-Selective Mukaiyama aldol addition						
0			[Si]	[Si]	О    а	: [Si] = TES
Ph 1	H T (	IDPi (2 °C), CH	mol%), Cl <sub>3</sub> (0.5 N	vi <i>)</i>	- ''	: [Si] = TBS : [Si] = TIPS
entry <sup>a</sup>	IDPi	[Si]	T (°C)	yield (%) <sup>b</sup>	anti/syn	e.r. <sup>d</sup>
1	7a	TES	-60	99	86:14	75:25
2	7a	TBS	-60	99	94:6	89:11
3	7a	TIPS	-60	99	90:10	83:17
4	7b	TES	-60	96	63:37	39:61
5	7с	TES	-60	84	62:38	38:62
6	7d	TES	-60	99	96:4	87:13
7	7d	TBS	-60	99	99:1	97.5:2.5
8	7d <sup>e</sup>	TBS	-78	99	>99:1	98:2
IDPi						
= ,		I	=		= SC	$D_2CF_3$ ( <b>a</b> ) $D_2C_2F_5$ ( <b>b</b> ) $D_2-n-C_4F_9$ ( <b>c</b> )

<sup>a</sup>Reactions were conducted with benzaldehyde 1 (0.1 mmol), enolsilanes 2a-b or 4a-c (1.2 equiv), and IDPi (2 mol %) for 16-24 h at the indicated temperature. <sup>b</sup>Determined by <sup>1</sup>H NMR spectroscopy. <sup>c</sup>Determined by crude <sup>1</sup>H NMR analysis. <sup>d</sup>The e.r. was determined by HPLC. eAt -78 °C using a 5:4 CHCl<sub>3</sub>/n-hexane mixture. TES, triethylsilyl; TIPS, triisopropylsilyl; TBS, tertbutyldimethylsilyl. See the Supporting Information for determination of the absolute configuration.

TES-enolsilane 2a (Table 1A, entry 6). An opposite trend was observed in the anti-selective Mukaiyama aldol addition of (Z)-enolsilanes (Table 1B). Both the size of the perfluorinated sulfonamide in the inner core of IDPi and the silyl group had tremendous effects on the selectivity. Longer perfluorinated groups gave poorer d.r. and e.r. At -60 °C, the addition of

 $SO_2CF_2H(\mathbf{d})$ 

Table 2. Substrate Scope for the syn- and anti-Mukaiyama Aldol Additions<sup>a</sup>



<sup>a</sup>Reaction scale: 0.2-5 mmol. See the Supporting Information for full reaction conditions and the determination of e.r.

(Z)-enolsilane 4a ([Si] = TES) to benzaldehyde proceeded with modest d.r. and e.r. using catalysts 7b and 7c, which performed exceptionally well previously in the syn-aldolization (Table 1B, entries 4-5). In contrast, IDPi 7a with the shortest trifluoromethylsulfonamide core performed far better, but to our surprise, with inverted benzaldehyde enantiofacial preference. Among (Z)-enolsilanes 4a-4c with different silyl-groups (Table 1B, entries 1-3), enolsilane 4b with the TBS-group performed best, albeit only in 89:11 e.r. Hypothesizing that replacing one of the F atoms in the CF3-group to an H atom may help in positively influencing the selectivity without introducing significant steric changes, 23 IDPi 7d containing a CF<sub>2</sub>H-group was designed and synthesized. Specifically, we envisioned that CF2H···heteroatom interactions could lead to a modulation of the active site of the IDPi catalyst.<sup>24</sup> Strikingly, the single-atom modification of the inner core of the catalyst, replacing the CF<sub>3</sub>-groups with CF<sub>2</sub>H-groups, indeed resulted in a spectacular enhancement of both the d.r. (99:1) and e.r. (98:2) in the addition of enolsilane 4b (Table 1B, entries 7 and

With these results in hand, the scope of our *syn*-selective Mukaiyama aldol reaction with various aromatic aldehydes was explored, using catalyst 7c (Table 2A). Aromatic aldehydes with *o-*, *m-*, and *p*-substituents and heteroaromatic aldehydes (3a-3j) gave the corresponding products in excellent yields and stereoselectivity. Multisubstituted aromatic aldehydes

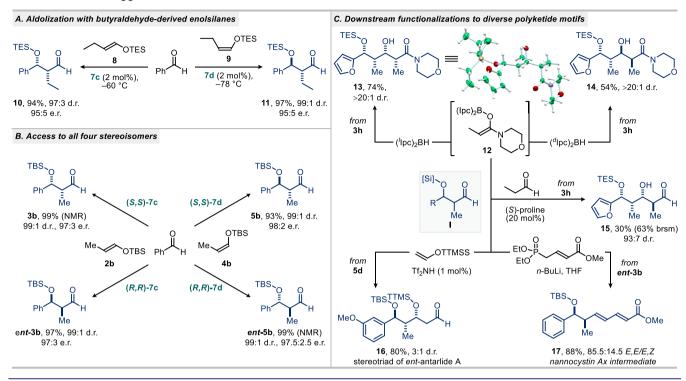
provided products 3k-3m, which contain substructures of complex polyketides. The catalyst loading could be reduced to 0.5 mol % without compromising the reaction time and stereoselectivity as shown with the gram scale synthesis of aldols 3h and 3k. A diastereoselective and enantioselective single aldolization of a dialdehyde substrate gave product 3m.

Essentially the same set of aromatic aldehydes performed equally well in our *anti*-aldolization process furnishing products 5b-k and 5m (Table 2B). Electron-rich aromatic aldehydes also delivered *anti*-aldols 5l and 5n with good diastereoselectivity and enantioselectivity. Only furfural gave somewhat lower yield of product 5i.

Furthermore, butyraldehyde-derived enolsilanes 8 and 9 also readily reacted to either *syn-* or *anti-*products 10 and 11 (Scheme 1A). Moreover, both enantiomers of our IDPi catalysts enabled access to all four possible stereoisomers of aldols as shown in Scheme 1B.

Our stereodivergent aldolization method is especially valuable in the context of the rapid generation of complex polyketide motifs (Scheme 1C). For example, when syn-aldol 3h was subjected to a follow-up propionate aldolization using in situ generated chiral boron enolates 12, 27 either the all-syn-stereotetrad 13 or its syn, anti, syn stereoisomer 14 were obtained as single diastereomers. The structure of the polyketide-like molecule 13 was unambiguously confirmed by

## Scheme 1. Further Applications of the Diastereoselective and Enantioselective Aldol Reactions



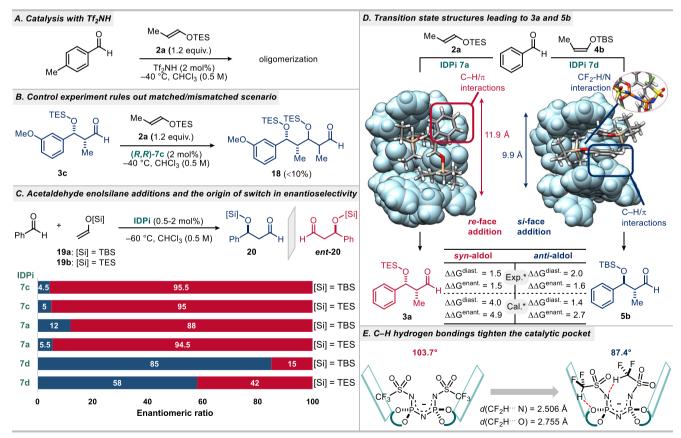


Figure 2. Mechanistic studies. (A–C) Control experiments. (D) Computed transition-state structures of major enantiomer of 3a (syn-selective addition) with 7a (left) and 5b (anti-selective addition) with 7d (right) at B3LYP-D3(BJ)/def2TZVP+CPCM(Chloroform)//ONIOM(PBE-D3/6-31G(d):PBE-D3/3-21G) level of theory. Distances between centroids of the two inner spirocyclobutyl-2-fluorenyl groups are shown. \*Energies in kcal/mol (see the Supporting Information for details). (E) Effect of C–H hydrogen bondings on the cavity size. Angles represent centroids of the two inner spirocyclobutyl-2-fluorenyl groups and the central nitrogen.

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X-ray crystallography. Alternatively, iterative aldolization using the proline-catalyzed cross-aldol addition with propionaldehyde delivered the syn, syn, anti-stereotetrad 15 with excellent diastereoselectivity. Further, anti-aldol 5d was converted to a fully protected double aldol adduct 16, containing the key anti, syn-stereotriad of antarlide A, 28 when our method was coupled with Yamamoto's supersilyl enolate technology. Stereotriad 16 is fully suited for further aldol addition toward antarlides. 1,3-Dienyl-6-oxy polyketide motif 17, an intermediate from a reported total synthesis of nannocystin Ax, was also obtained in only two steps starting from benzaldehyde (via ent-3b). The previous synthesis of 17 involved five steps, including an Evans-aldolization.<sup>29</sup> It is noteworthy that all of the polyketide motives 13-17 were obtained in just two steps from commercially available aldehydes such as furfural, manisaldehyde, and benzaldehyde.

To gain insight into the mechanism of our stereoselective aldol additions, experimental and computational studies were performed. Experiments were directed at probing aspects of our aldol reactions such as the origin of single aldolization and the unexpected change of facial selectivity during antialdolization using IDPi 7d. Initially, we confirmed that our IDPi catalysts were indeed unique in promoting the single aldolization of propionaldehyde enolsilanes: the well-established Mukaiyama aldol addition catalyst triflimide did not give even a trace of the desired single aldolization products because of complete enolsilane oligomerization (Figure 2A, see the Supporting Information, Figure S2 for details). syn-Aldol 3c, which was obtained using IDPi (S,S)-7c at −40 °C, underwent less than 10% conversion under the same conditions using the opposite enantiomer of the catalyst (R,R)-7c, excluding a potential matched-mismatched scenario (Figure 2B). The change in facial selectivity of aldehyde attack upon switching from the n-C<sub>4</sub>F<sub>9</sub>- and CF<sub>3</sub>-groups to the CF<sub>2</sub>H-group was also manifested when simple acetaldehyde-derived enolsilanes 19ab were used (Figure 2C). With catalysts 7a and 7c, acetaldehyde-derived enolsilane additions to benzaldehyde proceeded with re-selectivity, giving ent-20 irrespective of the silyl groups. In contrast, IDPi 7d, having a difluoromethanesulfonyl group in the core, reacted with si-selectivity.

In order to probe the origin of stereoselectivity and the switch of enantiofacial selectivity, an extensive DFT study was conducted for both syn- and anti-selective additions with IDPis 7a and 7d. Computed e.r.s and d.r.s were in good agreement with experimental observations in both cases (Figure 2D; see the Supporting Information, Figures S6-S10 for more details). When the optimized major transition-state structures of synand anti-selective aldolizations are compared, one of the most prominent differences appears to be the catalyst pocket size.<sup>30</sup> While catalyst 7a with CF<sub>3</sub>-cores has a relatively open cavity, the CF<sub>2</sub>H-groups in catalyst 7d engage in intramolecular hydrogen bonding interactions resulting in a more compact catalytic pocket (Figure 2D and 2E). Accordingly, for the synselective addition with 7a, the bulky (E)-enolsilane 2a bearing a smaller TES group would approach from the less hindered reface (Figure 2B, left). In contrast, for the case of anti-selective addition with 7d, the sterically less hindered (Z)-enolsilane 4b, with a slightly bulkier TBS group, provides a perfect fit into the narrower cavity, resulting in the complete switch of the facial selectivity (Figure 2D, right). The outcome of the acetaldehyde-derived enolsilane additions using catalysts 7a and 7d is also in good agreement with this model (Figures 2C, S12, and S13).

Additionally, our study has identified the involvement of  $CH/\pi$  interactions, indicated in Figure 2D,<sup>31</sup> between the spirocyclic methylene groups of the catalyst counteranion and the aromatic ring of the aldehyde substrate, which contribute to the high enantioselectivities in both transition states.<sup>32</sup> This is in agreement with our experimental observations, showing a strong effect of the spirocycle on the enantioselectivity (Figure S1).

Our highly stereoselective Mukaiyama aldol additions of propional dehyde enolsilanes give access to all stereoisomers of the stable and versatile protected ald ols in a predictable manner and can be used in rapid syntheses of complex polyketide motifs. Ultimately, our approach could aid in streamlining the synthesis of complex oligopropionates. We also uncovered an unusual enantior eversal effect by modifying a CF<sub>3</sub>-group to a CF<sub>2</sub>H-group. The origin of stereoselectivities and enantiofacial switch was rationalized through computational studies, revealing the cooperation of C–H hydrogen bonds and CH/ $\pi$  interactions to govern catalyst structure and transition states.

# ASSOCIATED CONTENT

# Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/jacs.1c07447.

Experimental details and analytical data for all new compounds (PDF)

## **Accession Codes**

CCDC 2097850–2097853 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data\_request/cif, or by emailing data\_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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#### Notes

The authors declare the following competing financial interest(s): A patent on the general catalyst class and its asymmetric catalysis has been filed.

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optimizing our catalysts for aliphatic aldehydes, and results will be published separately.