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## Short Communication

# Highly efficient preparation of 1-lysophosphatidylcholine via high proportion of Novozym<sup>®</sup> 435 (lipase B from *Candida antarctica*)catalyzed ethanolysis

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(100 wt% of PC), 97 % ethanol, in 72 h at 40 °C. During the reaction, acyl migration from 1-LPC to 2-LPC was rare. This novel synthetic method is expected to expand the practical applications of 1-LPC. © 2020 Published by Elsevier B.V. This is an open access article under the CC BY-NC-ND license (http://

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### 1. Introduction

Lysophosphatidylcholine (LPC) is a class of biomolecules derived from the cleavage of phosphatidylcholine (PC) via phospholipase  $A_1$  and  $A_2$  (PLA<sub>2</sub>) [1,2] or by the transfer of fatty acids to free cholesterol via lecithin-cholesterol acyltransferase [3]. LPC has two isomers, characterized by a hydroxyl group at the sn-1 position (1-LPC) or the sn-2 position (2-LPC). It exerts a variety of physiological functions; for example, it induces MCP-1 expression, increases inflammatory cytokines, disrupts mitochondrial integrity, activates macrophages, and induces oligodendrocyte demyelination [4,5]. However, to facilitate investigation regarding the physiological roles of these compounds and accelerate related research, a simple method for LPC synthesis is urgently needed.

To prepare 2-LPC, PLA<sub>2</sub>-catalyzed hydrolysis of PC is a simple and well-established method [6,7]. The sn-2 positional regiospecificity of PLA<sub>2</sub> enables 2-LPC production in slightly alkali reaction conditions with calcium cations as co-activators of PLA<sub>2</sub>. This technique is often used to analyze the fatty acid composition at the sn-1, -2 positions of phospholipids [8]. The lipase-catalyzed acylation of glycerophosphocholine constitutes an alternative approach [9-11]. Carmen and Adlercreutz obtained 2-LPC with high conversion (>95 %) by acylation using several vinyl fatty acids as acyl donors and Novozym® 435 (lipase B from Candida

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antarctica) as a catalyst in a mixture of 50 % t-butanol and 0.5 mmol water at 25 °C for 100 h [9]. Mnasri et al. obtained 2-LPC with a yield of 75 % by acylation using oleic acid as an acyl donor and Lipozyme<sup>®</sup> RM-IM (*Rhizomucor miehei*) as a catalyst in a solvent-free system at 50 °C for 24 h (or 40 °C for 72 h) [11].

To prepare 1-LPC, phospholipase A<sub>1</sub> (PLA<sub>1</sub>)-catalyzed hydrolysis has been applied by Lim et al. [12]. They obtained 83.7 mol% 1-LPC via PLA1 (Thermomyces lanuginosus)-catalyzed hydrolysis under a bi-phasic system of water and hexane [12]. However, the industrial application of PLA<sub>1</sub> is limited by the high cost. In addition, quantitative yields have not been obtained using sn-1, 3 regiospecific lipase. 1-LPC production via lipase-catalyzed alcoholysis, first developed in 1994 by Sarney et al., has been demonstrated using Lipozyme<sup>®</sup> IM-60 (Mucor miehei) as a catalyst [13,14]. Quantitative yields (>98 %) of 1-LPC in several alcohols, such as ethanol, 2-propanol, and 1-butanol, have been obtained at 22 °C for 24 h [13]. In 2013, alcoholysis reactions catalyzed by PLA1 (Thermomyces lanuginosus), Novozym<sup>®</sup> 435, and Lipase PS (Burkholderia cepacia), were reported by Baeza-Jiménez et al. [15] with LPC yields of 50 %, 58.5 %, and 80 %, respectively. Although the LPC isomer obtained by Novozym<sup>®</sup> 435 was not reported in this previous study, it is expected to be 1-LPC because the enzyme shows *sn*-1 positional specificity to glycerophospholipids but no regiospecificity to triacylglycerols [9].

In this study, a simple preparation method of 1-LPC via Novozym<sup>®</sup> 435-catalyzed ethanolysis without any toxic solvents was demonstrated (scheme 1). The effects of the water content in ethanol, reaction temperature, and reaction time on 1-LPC yield were investigated.

ABSTRACT Efficient preparation methods for 1-lysophosphatidylcholine (1-LPC), a physiologically important compound, are lacking. Here, we established a method for 1-LPC preparation via Novozym<sup>®</sup> 435 (a lipase B from Candida antarctica)-catalyzed ethanolysis. Novozym<sup>®</sup> 435 showed sn-1 regiospecificity to phosphatidylcholine, although it does not exhibit regiospecificity to triacylglycerol. In particular, quantitative 1-LPC yields (96.5  $\pm$  0.2 mol%) were reliably obtained in the presence of Novozym<sup>®</sup> 435







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Scheme 1. Preparation of 1-LPC via Novozym<sup>®</sup> 435 (lipase B from Candida antarctica)-catalyzed ethanolysis of PC (R; alkyl).

#### 2. Materials and methods

#### 2.1. Materials

PC from soybean (Phospholipon 90 G) was purchased from H. Holstein Co., Ltd. (Tokyo, Japan). Novozym<sup>®</sup> 435 (lipase B from *Candida antarctica*) was purchased from MIK Pharm Co., Ltd. (Tokyo, Japan). LPC ( $\iota$ - $\alpha$ -lysophosphatidylcholine) from egg yolk, used as the 2-LPC standard, was obtained from FUJIFILM Wako Pure Chemical Corporation (Osaka, Japan).

All solvents and other chemicals used in this study were of analytical grade.

#### 2.2. Enzymatic reaction

The typical enzymatic reaction was performed as follows. PC (50  $\mu$ mol) was dissolved with 1 mL of 97 % ethanol, followed by the addition of Novozym<sup>®</sup> 435 (100 wt% of PC). The reaction was allowed to proceed at 40 °C and 900 rpm in the dark. After the reaction, the enzyme was removed by filtration. The reaction mixture was adjusted to 5 mL with ethanol containing 0.1 % formic acid to avoid acyl migration and subjected to high-performance liquid chromatography (HPLC) for calculation of the 1-LPC yield.

### 2.3. Calculation of the 1-LPC yield

The reaction mixture (Section 2.2) was subjected to HPLC to quantify 1-LPC, as described by Adlercreutz and Wehtje with slight modifications [16]. The HPLC system consisted of a Waters 2695 Separations Module (Milford, MA, USA) and a Refractive Index Detector Model 133 (GILSON, Middleton, WI, USA). An InertSustain<sup>®</sup> NH<sub>2</sub> ( $4.6 \times 250 \text{ mm}$ ,  $5 \mu \text{m}$ ; GL Sciences, Tokyo, Japan) column was used. Samples were then eluted by isocratic elution of the mobile phase at 95 % ethanol- and 20 mM oxalic acid with a ratio of 94:4 (v/v). The flow rate was maintained at 1.0 mL/ min and the column temperature was maintained at 25 °C. Calibration curves were prepared with substrate PC and 2-LPC (see Section 2.1.) and levels of residual PC (mol%), 2-LPC (mol%), and 1-LPC yield (mol%, 2-LPC equivalent) were calculated. 1-LPC (mol%) was calculated by the Eq. (1). In these HPLC conditions, retention times (min) of PC, 1-LPC, and 2-LPC were 9.3, 11.5, and 12.6, respectively.

1-LPC (mol%) = 1-LPC (mol) / [PC (mol) + 1-LPC (mol) + 2-LPC (mol)] x 100 (1)

#### 2.4. Statistical analysis

All values are expressed as the means  $\pm$  SD (n = 3). Statistical differences were determined using Scheffe's tests, with a significance threshold of P < 0.05.

### 3. Results and discussion

3.1. Effect of ethanol concentration of the reaction solvent on 1-LPC yield

We first investigated the effect of the ethanol concentration on the ethanolysis reaction mediated by Novozym<sup>®</sup> 435. The reaction mixture included PC (50  $\mu$ mol), solvents with different concentrations of ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (100 wt% of PC). The reaction proceeded for 24 h at 60 °C and 900 rpm in the dark.

The highest average yield  $(75.1 \pm 1.3 \text{ mol}\%)$  was obtained using 97 % ethanol as a reaction solvent, although no significance differences in yield were observed for ethanol concentrations in the range of 92%–98% (Fig. 1). The yield decreased with the water content of the reaction solvent. In reaction solvents with high water contents, PC was not perfectly dissolved, resulting in low yields. For 50 % ethanol, only 16.0  $\pm$  1.1 mol% 1-LPC was obtained and residual PC was  $83.9 \pm 0.9 \text{ mol\%}$ , suggesting that acyl migration and hydrolysis at the sn-2 position did not occur in this reaction system. At high concentrations of ethanol, the 1-LPC yield was also low;  $11.2 \pm 5.5 \text{ mol}\%$  1-LPC was obtained in 99.5 % ethanol. This may be explained by enzyme denaturation caused by ethanol, which can absorb essential water [17], as small amounts of water are essential for the ethanolysis reaction to proceed using Novozym<sup>®</sup> 435 as a catalyst. These results are consistent with those of Sarney et al., who found that 5% water in ethanol was optimal although a high 1-LPC yield could be obtained up to 14 % [13].

Based on these results, 97 % ethanol was selected as the reaction solvent in the reaction system.

#### 3.2. Effect of reaction temperature on 1-LPC yield

An ethanolysis reaction was performed under different temperatures. The reaction mixture consisted of PC (50  $\mu$ mol), 97 % ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (100 wt% of PC). The reaction proceeded for 24 h at 4–60 °C at 900 rpm in the dark.

The highest 1-LPC yields were obtained at 40 °C ( $80.9 \pm 8.7 \text{ mol}$ %), although the yields at this temperature did not differ significantly from those at 60 °C (Fig. 2). At 20 °C, the 1-LPC yield was 62.7 ± 8.5 mol% which was moderate, and at 4 °C, the yield was lower. These results showed that the reaction system is suitable for a substrate containing unsaturated fatty acids, such as marine phospholipids with eicosapentaenoic acid and docosahexaenoic acid, which are sensitive to oxidative deterioration.

Baeza-Jiménez et al. also investigated the effect of temperature on the Novozym<sup>®</sup> 435-catalyzed ethanolysis of PC, showing that the degree of ethanolysis increased clearly as the reaction temperature increased from 40 °C to 60 °C [15]. They used Novozym<sup>®</sup> 435 with 15 wt% of PC, compared with 100 wt% of PC in our study. This difference in enzyme dose might influence reaction temperature because the reaction rate is high in reaction



Fig. 1. Effect of the ethanol concentration on 1-LPC yield.

Reaction mixture; PC (50  $\mu$ mol), solvent with different concentrations of ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (100 wt% of PC). The reaction proceeded for 24 h at 60 °C, 900 rpm in the dark. Different letters indicate significant differences (P < 0.05).

systems with high enzyme doses. Nevertheless, it is challenging to precisely clarify the effect of temperature.

3.3. Effect of enzyme dose on 1-LPC yield and residual PC

The 1-LPC yield did not differ significantly between reaction temperatures of 40 °C and 60 °C; accordingly, 40 °C was selected as the optimal reaction temperature in this system to avoid acyl migration at high temperatures.

Next, the effect of enzyme dose on 1-LPC yield and residual PC was investigated. The reaction mixture consisted of PC (50  $\mu$ mol), 97 % ethanol (1.0 mL), and different doses of Novozym<sup>®</sup> 435 (12.5–100 wt % of PC). The reaction proceeded for 24 h at 40 °C, 900 rpm in the dark.



Fig. 2. Effect of the reaction temperature on 1-LPC yield.

Reaction mixture; PC (50 μmol), 97 % ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (100 wt% of PC). The reaction proceeded for 24 h at various temperatures and 900 rpm in the dark. Different letters indicate significant differences (P < 0.05).



Fig. 3. Effect of enzyme dose on 1-LPC yield and residual PC.

Reaction mixture; PC (50 μmol), 97 % ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (12.5–100 wt% of PC). The reaction proceeded for 24 h at various temperatures and 900 rpm in the dark. Different letters indicate significant differences (P < 0.05).

1-LPC yield decreased with decreasing enzyme dose (Fig. 3). In accordance with decreasing 1-LPC yield, residual PC increased whereas no 2-LPC yield was observed in any trial. The highest 1-LPC yield was obtained when 100 wt% of Novozym<sup>®</sup> 435 was used (this value was obtained from the trial shown in Fig. 2). Notably, even the use of 50 wt% of Novozym<sup>®</sup> 435 afforded significantly lower 1-LPC yield than that obtained with 100 or 75 wt%. When 12.5 wt% of Novozym<sup>®</sup> 435 was used, the standard amount applied

in current Novozym<sup>®</sup> 435-mediated processes, < 20 mol% of 1-LPC yield was obtained. Yang et al. reported 97.7 % yield of 1-LPC via Novozym<sup>®</sup> 435 mediated ethanolysis of PC using only 10 wt% of biocatalyst loading [18]. However, we have not been able reproduce this result.

This result indicated that this reaction system requires a substantial enzyme dose to reliably obtain high 1-LPC yield, likely because short chain alcohols, such as methanol and ethanol, tend



Fig. 4. Time course of Novozym<sup>®</sup> 435-catalyzed ethanolysis of PC.

Reaction mixture; PC (50 µmol), 97 % ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (100 wt% of PC). The reaction proceeded for 1–72 h at 40 °C, 900 rpm in the dark. Symbol: 1-LPC (mol%), open circle; PC (mol%), closed circle; 1-LPC + PC (mol%), square.

to dehydrate enzyme protein and thereby deactivate the enzyme [19]. Although immobilized enzymes such as Novozym<sup>®</sup> 435, which is immobilized on acrylic resin, can overcome this problem by maintaining their active structure [20], a large enzyme dose was required in a reaction media containing high concentration of ethanol which is essential to shift reaction equilibrium to produce 1-LPC.

#### 3.4. Effect of reaction time on 1-LPC yield

Finally, the effect of reaction time was investigated using a mixture consisting of PC (50  $\mu$ mol), 97 % ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (100 wt% of PC). The reaction was continued for 72 h at 40 °C, 900 rpm in the dark.

1-LPC yield increased as the reaction time increased and reached a plateau at 60 h (Fig. 4). The maximum 1-LPC yield of  $96.5 \pm 0.2 \text{ mol}\%$  was obtained at 72 h. Residual PC substrate decreased as the reaction time increased. The total amount of 1-LPC + residual PC was nearly 100 mol% during the reaction, suggesting that acyl migration from 1-LPC to 2-LPC was rare in this reaction system.

Alternatively, Baeza-Jiménez et al. reported that the degree of ethanolysis increases until 48 h and apparently decreases thereafter at several reaction temperatures (30–60 °C) [15]. We note that acyl migration from 1-LPC to 2-LPC might occur in this previously established reaction system. In addition, the discrepancy between the results of the previous [15] and current study may arise from the differences in the concentration of the reaction system. Specifically, Baeza-Jiménez et al. used an enzyme concentration of 15 wt% (to PC) in 5 mL of 95 % ethanol (150 mg PC /mL), whereas we applied 100 wt% enzyme in 1.0 mL of 97 % ethanol (approximately 40 mg PC /mL). However, additional research is required to clarify this issue.

#### 4. Conclusion

1-LPC was successfully prepared under the following optimized reaction conditions: PC (50  $\mu$ mol), 97 % ethanol (1.0 mL), and Novozym<sup>®</sup> 435 (100 wt% of PC), with a 72-h reaction time at 40 °C and 900 rpm in the dark. This resulted in a 1-LPC yield of 96.5  $\pm$  0.2 mol%. In this reaction system, Novozym<sup>®</sup> 435 showed regiospecificity at the *sn*-1 position of PC and acyl migration from 1-LPC to 2-LPC was rare.

#### **CRediT** authorship contribution statement

**Sayumi Yasuda:** Investigation. **Yukihiro Yamamoto:** Conceptualization, Methodology, Writing – original draft.

#### **Declaration of Competing Interest**

None.

#### Appendix A. Supplementary data

Supplementary material related to this article can be found, in the online version, at doi:https://doi.org/10.1016/j.btre.2020. e00505.

#### References

- N. Ridgway, R. McLeod, Biochemistry of Lipids, Lipoproteins and Membranes, 6th ed., Elsevier, Amsterdam, The Netherlands, 2016.
- [2] M. Gauster, G. Rechberger, A. Sovic, G. Horl, E. Steyrer, W. Sattler, S. Frank, Endothelial lipase releases saturated and unsaturated fatty acids of high density lipoprotein phosphatidylcholine, J. Lipid Res. 46 (2005) 1517–1525.
- [3] R.P. Dullaart, R.T. Gansevoort, B.D. Dikkeschei, D. de Zeeuw, P.E. de Jong, A. van Tol, Role of elevated lecithin: cholesterol acyltransferase and cholesteryl ester transfer protein activities in abnormal lipoproteins from proteinuric patients, Kidney Int. 44 (1993) 91–97.
- [4] R.A. Rabini, R. Galassi, P. Fumelli, N. Dousset, M.L. Solera, P. Valdiguie, G. Curatola, G. Ferretti, M. Taus, L. Mazzanti, Reduced Na(+)-K(+)-ATPase activity and plasma lysophosphatidylcholine concentrations in diabetic patients, Diabetes 43 (1994) 915–919.
- [5] S.H. Law, M.L. Chan, G.K. Marathe, F. Parveen, C.H. Chen, L.Y. Ke, An updated review of lysophosphatidylcholine metabolism in human diseases, Int. J. Mol. Sci. 20 (2019) 1149.
- [6] Z. Guo, A.F. Vikbjerg, X. Xu, Enzymatic modification of phospholipids for functional applications and human nutrition, Biotechnol. Adv. 23 (2005) 203–259.
- [7] J. Kim, C.S. Lee, J. Oh, B.G. Kim, Production of egg yolk lysolecithin with immobilized phospholipase A2, Enzyme Microb. Technol. 29 (2001) 587–592.
- [8] L. Amate, M. Ramírez, A. Gil, Positional analysis of triglycerides and phospholipids rich in long-chain polyunsaturated fatty acids, Lipids 34 (1999) 865–871.
- [9] V. Carmen, P. Adlercreutz, Lysophosphatidylcholine synthesis with Candida antarctica lipase B (Novozym 435), Enzyme Microb. Technol. 26 (2000) 630– 635.
- [10] S.I. Hong, Y. Kim, C.T. Kim, I.H. Kim, Enzymatic synthesis of lysophosphatidylcholine containing CLA from sn-glycero-3phosphatidylcholine (GPC) under vacuum, Food Chem. 129 (2011) 1–6.
- [11] T. Mnasri, F. Ergan, J. Hérault, G. Pencreac'h, Lipase-catalyzed synthesis of oleyl-lysophosphatidylcholine by direct esterification in solvent-free medium without water removal, J. Oleo Sci. 66 (2017) 1009–1016.
- [12] C.W. Lim, B.H. Kim, I.H. Kim, M.W. Lee, Modeling and optimization of phospholipase A1-catalyzed hydrolysis of phosphatidylcholine using response surface methodology for lysophosphatidylcholine production, Biotechnol. Prog. 31 (2014) 35–41.
- [13] D.B. Sarney, G. Fregapane, E.N. Vulfson, Lipase-catalyzed synthesis of lysophospholipids in a continuous bioreactor, J. Am. Oil Chem. Soc. 71 (1994) 93–96.
- [14] T. Mnasri, J. Hérault, L. Gauvry, C. Loiseau, L. Poisson, F. Ergan, G. Pencreac'h, Lipase-catalyzed production of lysophospholipids, Oilseeds and fats, Crops Lipids 24 (2017) D405.
- [15] R. Baeza-Jiménez, L.X. López-Martínez, C. Otero, I.H. Kim, H.S. Garía, Enzymecatalysed hydrolysis of phosphatidylcholine for the production of lysophosphatidylcholine, J. Chem. Technol. Biot. 88 (2013) 1859–1863.
- [16] D. Adlercreutz, E. Wehtje, A simple HPLC method for the simultaneous analysis of phosphatidylcholine and its partial hydrolysis products 1- and 2-acyl lysophosphatidylcholine, J. Am. Oil Chem. Soc. 78 (2001) 1007–1011.
- [17] A. Zaks, A.M. Klibanov, Enzymatic catalysis in nonaqueous solvents, J. Biol. Chem. 263 (1988) 3194–3201.
- [18] G. Yang, R. Yang, J. Hu, Lysophosphatidylcholine synthesis by lipase-catalyzed ethanolysis, J. Oleo Sci. 64 (2015) 443–447.
- [19] J.S. Dordick, Designing enzymes for use in organic solvents, Biotechnol. Prog. 8 (1992) 259–267.
- [20] R.C. Rodriques, C. Ortiz, Á. Berenguer-Murcia, R. Torres, R. Fernández-Lafuente, Modifying enzyme activity and selectivity by immobilization, Chem. Soc. Rev. 42 (2013) 6290–6307.