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## Structure Reports

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# (E)-1-[2-(4-Chloro-2-nitrostyryl)-1-phenylsulfonyl-1H-indol-3-yl]propan-1-one

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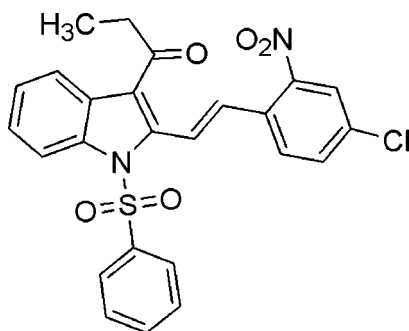
Received 8 November 2013; accepted 12 November 2013

Key indicators: single-crystal X-ray study;  $T = 295$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.050;  $wR$  factor = 0.150; data-to-parameter ratio = 20.6.

In the title compound,  $\text{C}_{25}\text{H}_{19}\text{ClN}_2\text{O}_5\text{S}$ , the phenyl ring forms dihedral angles of  $79.62$  (12) and  $80.02$  (13)° with the indole ring system and the benzene ring, respectively. The nitro group is twisted at an angle of  $22.39$  (11)° with respect to the attached benzene ring. In the crystal, molecules assemble into double layers in the  $ab$  plane via  $\text{C}-\text{H}\cdots\text{O}$  interactions.

## Related literature

For the biological activity of indole derivatives, see: Okabe & Adachi (1998); Srivastava *et al.* (2011). For related structures, see: Chakkaravarthi *et al.* (2008, 2010).



## Experimental

### Crystal data

$\text{C}_{25}\text{H}_{19}\text{ClN}_2\text{O}_5\text{S}$   
 $M_r = 494.93$   
Triclinic,  $P\bar{1}$   
 $a = 8.4658$  (3) Å  
 $b = 8.6643$  (3) Å  
 $c = 16.1126$  (6) Å  
 $\alpha = 84.196$  (2)°  
 $\beta = 87.768$  (3)°  
 $\gamma = 79.541$  (2)°  
 $V = 1156.01$  (7) Å<sup>3</sup>  
 $Z = 2$   
Mo  $K\alpha$  radiation  
 $\mu = 0.30$  mm<sup>-1</sup>  
 $T = 295$  K  
 $0.28 \times 0.24 \times 0.20$  mm

### Data collection

Bruker Kappa APEXII diffractometer  
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.922$ ,  $T_{\max} = 0.943$   
25531 measured reflections  
6333 independent reflections  
4536 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.029$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.050$   
 $wR(F^2) = 0.150$   
 $S = 1.03$   
6333 reflections  
308 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.49$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.39$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C10}-\text{H10}\cdots\text{O1}^i$	0.93	2.60	3.327 (3)	136
$\text{C16}-\text{H16B}\cdots\text{O5}^{\text{ii}}$	0.97	2.37	3.260 (3)	152

Symmetry codes: (i)  $x, y + 1, z$ ; (ii)  $x + 1, y, z$ .

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

The authors wish to acknowledge the SAIF, IIT Madras, for the data collection.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK5271).

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## supplementary materials

*Acta Cryst.* (2013). E69, o1784 [doi:10.1107/S1600536813031073]

**(E)-1-[2-(4-Chloro-2-nitrostyryl)-1-phenylsulfonyl-1*H*-indol-3-yl]propan-1-one**

M. Umadevi, V. Saravanan, R. Yamuna, A. K. Mohanakrishnan and G. Chakkaravarthi

**1. Introduction****2. Experimental****2.1. Synthesis and crystallization**

A solution of 1-(2-(bromomethyl)-1-(phenylsulfonyl)-1*H*-indol-3-yl)propan-1-one (5 g, 12.31 mmol) and triphenylphosphine (3.5 g, 13.54 mmol) in dry THF (100 ml) was refluxed for 6 h. After consumption of the starting material, the solvent was removed under vacuum and the solid washed with diethyl ether to give the phosphonium salt. Then, the mixture of phosphonium salt (8 g, 11.97 mmol), 4-chloro-2-nitrobenzaldehyde (2.45 g, 13.17 mmol) and K<sub>2</sub>CO<sub>3</sub> (3.30 g, 23.95 mmol) in DCM (70 ml) was stirred at room temperature for 24 h. After completion of the reaction (monitored by TLC), it was diluted using DCM (30 ml), washed with water (2 x 100 ml) and dried (Na<sub>2</sub>SO<sub>4</sub>). Removal of solvent *in vacuo* followed by trituration of the crude product with MeOH (20 ml) afforded the title compound.

**2.2. Refinement**

Crystal data, data collection and structure refinement details are summarized in Table 1. H atoms were positioned geometrically and refined using riding model, with C—H = (0.93–0.97) Å and  $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$ . Owing to poor agreement, the (001) reflection was omitted from the final cycles of refinement.

**3. Results and discussion**

In continuation of our studies on indole derivatives which are known to exhibit anti-microbial, anti-biotic, analgesic and anti-cancer activities (Okabe and Adachi, 1998; Srivastava *et al.*, 2011), we herewith report the crystal structure of the title compound (I). The geometric parameters of (I) (Fig. 1) are in close agreement with similar structures (Chakkaravarthi *et al.*, 2008; 2010).

The phenyl ring makes the dihedral angle of 79.62 (12)° with the indole ring system. The phenyl ring (C1—C6) and the benzene ring (C20—C25) are inclined at an angle of 80.02 (13)°. The nitro group is twisted at an angle of 22.39 (11)° with respect to the attached benzene ring (C20—C25). The N1 atom is *sp*<sup>2</sup> hybridised as the bond angles around N1 atom sum 352.4°. Details of the C—H...O interactions are given in Table 2 - these lead to layers in the *ab* plane.

**Computing details**

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT* (Bruker, 2004); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008).

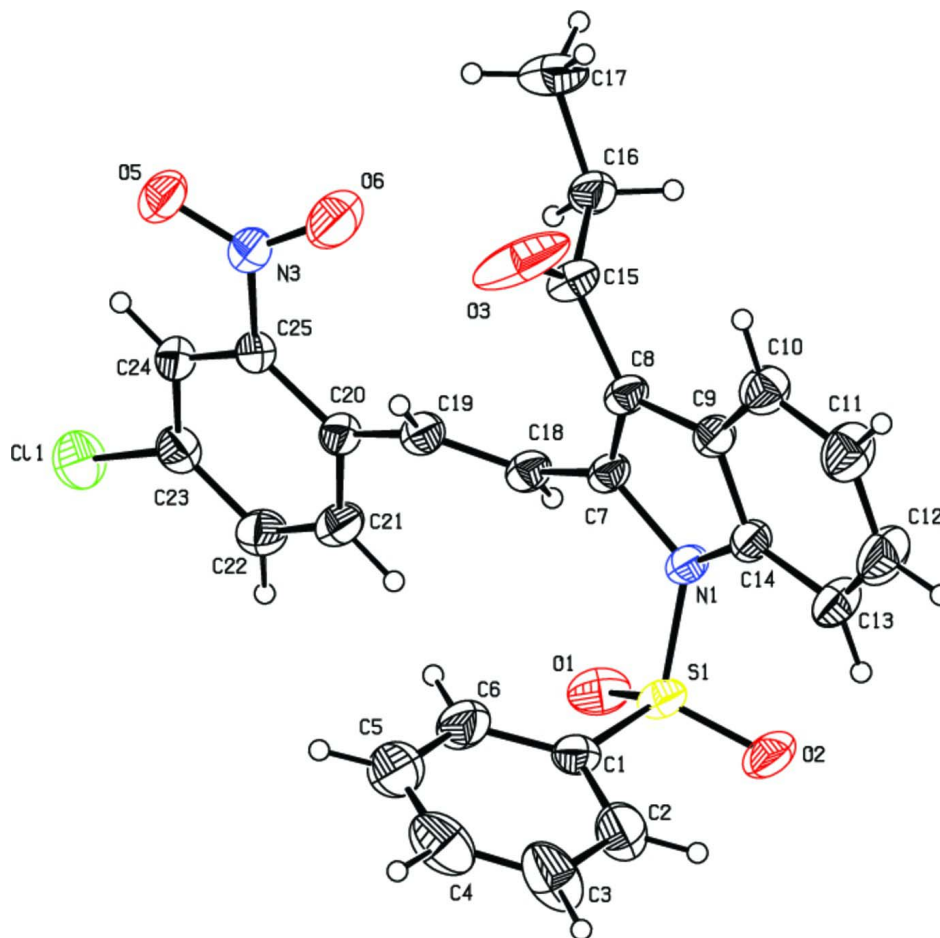


Figure 1

The molecular structure of (I), with atom labels and 30% probability displacement ellipsoids for non-H atoms.

**(E)-1-[2-(4-Chloro-2-nitrostyryl)-1-phenylsulfonyl-1*H*-indol-3-yl]propan-1-one**

*Crystal data*

$C_{25}H_{19}ClN_2O_5S$   
 $M_r = 494.93$   
 Triclinic,  $P\bar{1}$   
 Hall symbol: -P 1  
 $a = 8.4658 (3) \text{ \AA}$   
 $b = 8.6643 (3) \text{ \AA}$   
 $c = 16.1126 (6) \text{ \AA}$   
 $\alpha = 84.196 (2)^\circ$   
 $\beta = 87.768 (3)^\circ$   
 $\gamma = 79.541 (2)^\circ$   
 $V = 1156.01 (7) \text{ \AA}^3$

$Z = 2$   
 $F(000) = 512$   
 $D_x = 1.422 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$   
 Cell parameters from 8917 reflections  
 $\theta = 2.5\text{--}28.4^\circ$   
 $\mu = 0.30 \text{ mm}^{-1}$   
 $T = 295 \text{ K}$   
 Block, colourless  
 $0.28 \times 0.24 \times 0.20 \text{ mm}$

*Data collection*

Bruker Kappa APEXII  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 $\omega$  and  $\varphi$  scan

Absorption correction: multi-scan  
 (*SADABS*; Sheldrick, 1996)  
 $T_{\min} = 0.922$ ,  $T_{\max} = 0.943$   
 25531 measured reflections  
 6333 independent reflections

4536 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.029$   
 $\theta_{\text{max}} = 29.5^\circ$ ,  $\theta_{\text{min}} = 2.4^\circ$

$h = -11 \rightarrow 11$   
 $k = -11 \rightarrow 11$   
 $l = -20 \rightarrow 22$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.050$   
 $wR(F^2) = 0.150$   
 $S = 1.03$   
 6333 reflections  
 308 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0709P)^2 + 0.3517P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.49 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.39 \text{ e } \text{\AA}^{-3}$

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.6620 (2)	0.1560 (2)	0.44142 (11)	0.0497 (4)
C2	0.7468 (3)	0.1501 (4)	0.51342 (14)	0.0765 (7)
H2	0.8570	0.1131	0.5140	0.092*
C3	0.6662 (4)	0.1997 (5)	0.58429 (16)	0.0984 (11)
H3	0.7215	0.1945	0.6336	0.118*
C4	0.5062 (4)	0.2564 (4)	0.58242 (17)	0.0920 (9)
H4	0.4537	0.2952	0.6298	0.110*
C5	0.4199 (4)	0.2573 (4)	0.51124 (19)	0.0991 (10)
H5	0.3093	0.2923	0.5115	0.119*
C6	0.4985 (3)	0.2060 (3)	0.43989 (15)	0.0781 (7)
H6	0.4420	0.2053	0.3916	0.094*
C7	0.6425 (2)	0.3393 (2)	0.23941 (10)	0.0408 (4)
C8	0.6569 (2)	0.4934 (2)	0.22907 (10)	0.0424 (4)
C9	0.7938 (2)	0.5131 (2)	0.27424 (11)	0.0439 (4)
C10	0.8617 (3)	0.6446 (2)	0.28430 (14)	0.0584 (5)
H10	0.8179	0.7435	0.2589	0.070*
C11	0.9950 (3)	0.6244 (3)	0.33266 (18)	0.0738 (7)
H11	1.0405	0.7113	0.3412	0.089*
C12	1.0626 (3)	0.4773 (3)	0.36883 (18)	0.0771 (7)
H12	1.1536	0.4675	0.4008	0.093*
C13	0.9999 (3)	0.3449 (3)	0.35916 (15)	0.0625 (5)
H13	1.0466	0.2461	0.3836	0.075*
C14	0.8635 (2)	0.3653 (2)	0.31128 (11)	0.0440 (4)
C15	0.5596 (2)	0.6261 (2)	0.17657 (12)	0.0516 (5)
C16	0.5615 (3)	0.6198 (3)	0.08490 (12)	0.0593 (5)
H16A	0.5066	0.5359	0.0729	0.071*
H16B	0.6722	0.5927	0.0657	0.071*
C17	0.4848 (4)	0.7702 (3)	0.03593 (16)	0.0880 (8)
H17A	0.3710	0.7884	0.0469	0.132*
H17B	0.5053	0.7613	-0.0226	0.132*
H17C	0.5291	0.8566	0.0523	0.132*
C18	0.5229 (2)	0.2611 (2)	0.20724 (10)	0.0442 (4)

H18	0.5563	0.1621	0.1884	0.053*
C19	0.3680 (2)	0.3249 (2)	0.20359 (11)	0.0467 (4)
H19	0.3352	0.4250	0.2212	0.056*
C20	0.2462 (2)	0.2449 (2)	0.17303 (11)	0.0453 (4)
C21	0.2582 (3)	0.0828 (2)	0.19095 (14)	0.0566 (5)
H21	0.3466	0.0271	0.2201	0.068*
C22	0.1449 (3)	0.0018 (2)	0.16730 (14)	0.0589 (5)
H22	0.1580	-0.1070	0.1796	0.071*
C23	0.0107 (2)	0.0832 (2)	0.12499 (12)	0.0518 (4)
C24	-0.0071 (2)	0.2427 (2)	0.10536 (12)	0.0487 (4)
H23	-0.0963	0.2976	0.0764	0.058*
C25	0.1097 (2)	0.3207 (2)	0.12932 (11)	0.0452 (4)
N1	0.77198 (17)	0.25460 (17)	0.28902 (9)	0.0427 (3)
N3	0.0859 (2)	0.4903 (2)	0.10290 (11)	0.0554 (4)
O1	0.6646 (2)	0.00599 (16)	0.31000 (9)	0.0651 (4)
O2	0.92318 (19)	0.02047 (18)	0.37066 (10)	0.0682 (4)
O3	0.4897 (3)	0.7364 (2)	0.20882 (12)	0.1283 (11)
O5	-0.04694 (18)	0.55877 (18)	0.08461 (11)	0.0692 (4)
O6	0.2012 (2)	0.5539 (2)	0.09605 (17)	0.1115 (9)
S1	0.76175 (6)	0.08992 (5)	0.35118 (3)	0.04869 (14)
Cl1	-0.13374 (7)	-0.01698 (7)	0.09611 (4)	0.07319 (19)

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0570 (11)	0.0502 (10)	0.0409 (9)	-0.0105 (8)	-0.0042 (8)	0.0040 (7)
C2	0.0641 (14)	0.121 (2)	0.0511 (12)	-0.0321 (14)	-0.0066 (10)	-0.0076 (13)
C3	0.098 (2)	0.169 (3)	0.0489 (13)	-0.071 (2)	0.0058 (13)	-0.0239 (17)
C4	0.114 (3)	0.112 (2)	0.0570 (15)	-0.0362 (19)	0.0260 (16)	-0.0210 (15)
C5	0.0807 (18)	0.127 (3)	0.0700 (17)	0.0174 (18)	0.0207 (14)	0.0128 (17)
C6	0.0669 (14)	0.106 (2)	0.0486 (12)	0.0085 (14)	-0.0037 (10)	0.0089 (12)
C7	0.0405 (8)	0.0465 (9)	0.0314 (7)	0.0034 (7)	-0.0039 (6)	-0.0038 (6)
C8	0.0444 (9)	0.0458 (9)	0.0330 (8)	0.0039 (7)	-0.0042 (7)	-0.0044 (7)
C9	0.0421 (9)	0.0493 (9)	0.0373 (8)	-0.0004 (7)	-0.0008 (7)	-0.0048 (7)
C10	0.0591 (12)	0.0499 (11)	0.0654 (13)	-0.0074 (9)	-0.0074 (10)	-0.0043 (9)
C11	0.0639 (14)	0.0660 (14)	0.0959 (18)	-0.0182 (11)	-0.0190 (13)	-0.0110 (13)
C12	0.0568 (13)	0.0819 (17)	0.0943 (19)	-0.0116 (12)	-0.0320 (13)	-0.0059 (14)
C13	0.0503 (11)	0.0655 (13)	0.0677 (13)	0.0003 (9)	-0.0196 (10)	0.0001 (10)
C14	0.0398 (8)	0.0495 (9)	0.0400 (9)	-0.0013 (7)	-0.0029 (7)	-0.0035 (7)
C15	0.0597 (11)	0.0460 (10)	0.0443 (10)	0.0046 (8)	-0.0140 (8)	-0.0024 (8)
C16	0.0538 (11)	0.0768 (14)	0.0393 (9)	0.0031 (10)	-0.0010 (8)	0.0057 (9)
C17	0.117 (2)	0.0850 (18)	0.0532 (13)	-0.0059 (16)	-0.0168 (14)	0.0213 (12)
C18	0.0476 (9)	0.0459 (9)	0.0369 (8)	-0.0009 (7)	-0.0060 (7)	-0.0045 (7)
C19	0.0471 (9)	0.0481 (10)	0.0424 (9)	-0.0022 (8)	-0.0022 (7)	-0.0042 (7)
C20	0.0412 (9)	0.0489 (10)	0.0430 (9)	-0.0024 (7)	0.0006 (7)	-0.0014 (7)
C21	0.0533 (11)	0.0505 (11)	0.0612 (12)	-0.0003 (9)	-0.0098 (9)	0.0045 (9)
C22	0.0625 (12)	0.0478 (10)	0.0644 (13)	-0.0080 (9)	-0.0033 (10)	0.0017 (9)
C23	0.0503 (10)	0.0574 (11)	0.0486 (10)	-0.0131 (9)	0.0045 (8)	-0.0052 (8)
C24	0.0388 (9)	0.0563 (11)	0.0482 (10)	-0.0044 (8)	0.0006 (7)	-0.0001 (8)
C25	0.0395 (9)	0.0479 (9)	0.0447 (9)	-0.0028 (7)	0.0035 (7)	0.0015 (7)

N1	0.0420 (7)	0.0436 (8)	0.0389 (7)	0.0004 (6)	-0.0064 (6)	0.0001 (6)
N3	0.0441 (8)	0.0534 (9)	0.0655 (10)	-0.0064 (7)	-0.0053 (7)	0.0068 (8)
O1	0.0916 (11)	0.0455 (7)	0.0589 (9)	-0.0115 (7)	-0.0143 (8)	-0.0052 (6)
O2	0.0624 (9)	0.0575 (8)	0.0722 (10)	0.0174 (7)	-0.0096 (7)	0.0063 (7)
O3	0.205 (3)	0.0840 (13)	0.0636 (11)	0.0794 (16)	-0.0502 (14)	-0.0264 (10)
O5	0.0509 (8)	0.0593 (9)	0.0903 (12)	0.0050 (7)	-0.0109 (8)	0.0038 (8)
O6	0.0654 (11)	0.0756 (12)	0.189 (2)	-0.0286 (9)	-0.0420 (13)	0.0514 (14)
S1	0.0554 (3)	0.0407 (2)	0.0451 (2)	0.00326 (19)	-0.00822 (19)	0.00039 (18)
Cl1	0.0676 (4)	0.0741 (4)	0.0840 (4)	-0.0276 (3)	-0.0047 (3)	-0.0088 (3)

*Geometric parameters (Å, °)*

C1—C6	1.374 (3)	C15—O3	1.185 (2)
C1—C2	1.380 (3)	C15—C16	1.483 (3)
C1—S1	1.747 (2)	C16—C17	1.506 (3)
C2—C3	1.373 (4)	C16—H16A	0.9700
C2—H2	0.9300	C16—H16B	0.9700
C3—C4	1.355 (5)	C17—H17A	0.9600
C3—H3	0.9300	C17—H17B	0.9600
C4—C5	1.383 (5)	C17—H17C	0.9600
C4—H4	0.9300	C18—C19	1.329 (3)
C5—C6	1.378 (4)	C18—H18	0.9300
C5—H5	0.9300	C19—C20	1.468 (3)
C6—H6	0.9300	C19—H19	0.9300
C7—C8	1.355 (3)	C20—C21	1.391 (3)
C7—N1	1.427 (2)	C20—C25	1.399 (2)
C7—C18	1.454 (2)	C21—C22	1.373 (3)
C8—C9	1.437 (2)	C21—H21	0.9300
C8—C15	1.493 (2)	C22—C23	1.387 (3)
C9—C10	1.393 (3)	C22—H22	0.9300
C9—C14	1.394 (2)	C23—C24	1.369 (3)
C10—C11	1.371 (3)	C23—C11	1.725 (2)
C10—H10	0.9300	C24—C25	1.382 (3)
C11—C12	1.379 (4)	C24—H23	0.9300
C11—H11	0.9300	C25—N3	1.468 (2)
C12—C13	1.373 (3)	N1—S1	1.6720 (15)
C12—H12	0.9300	N3—O6	1.202 (2)
C13—C14	1.388 (3)	N3—O5	1.205 (2)
C13—H13	0.9300	O1—S1	1.4153 (16)
C14—N1	1.418 (2)	O2—S1	1.4230 (15)
C6—C1—C2	121.4 (2)	C17—C16—H16A	108.5
C6—C1—S1	118.79 (16)	C15—C16—H16B	108.5
C2—C1—S1	119.75 (18)	C17—C16—H16B	108.5
C3—C2—C1	119.1 (3)	H16A—C16—H16B	107.5
C3—C2—H2	120.5	C16—C17—H17A	109.5
C1—C2—H2	120.5	C16—C17—H17B	109.5
C4—C3—C2	120.0 (3)	H17A—C17—H17B	109.5
C4—C3—H3	120.0	C16—C17—H17C	109.5
C2—C3—H3	120.0	H17A—C17—H17C	109.5

C3—C4—C5	121.1 (3)	H17B—C17—H17C	109.5
C3—C4—H4	119.5	C19—C18—C7	122.80 (17)
C5—C4—H4	119.5	C19—C18—H18	118.6
C6—C5—C4	119.6 (3)	C7—C18—H18	118.6
C6—C5—H5	120.2	C18—C19—C20	123.04 (17)
C4—C5—H5	120.2	C18—C19—H19	118.5
C1—C6—C5	118.7 (2)	C20—C19—H19	118.5
C1—C6—H6	120.6	C21—C20—C25	115.37 (17)
C5—C6—H6	120.6	C21—C20—C19	119.73 (16)
C8—C7—N1	108.46 (15)	C25—C20—C19	124.82 (16)
C8—C7—C18	129.63 (15)	C22—C21—C20	122.68 (18)
N1—C7—C18	121.90 (15)	C22—C21—H21	118.7
C7—C8—C9	108.69 (15)	C20—C21—H21	118.7
C7—C8—C15	128.84 (16)	C21—C22—C23	119.55 (19)
C9—C8—C15	122.37 (17)	C21—C22—H22	120.2
C10—C9—C14	119.98 (17)	C23—C22—H22	120.2
C10—C9—C8	132.32 (17)	C24—C23—C22	120.32 (19)
C14—C9—C8	107.70 (16)	C24—C23—C11	119.70 (16)
C11—C10—C9	118.2 (2)	C22—C23—C11	119.98 (16)
C11—C10—H10	120.9	C23—C24—C25	118.76 (17)
C9—C10—H10	120.9	C23—C24—H23	120.6
C10—C11—C12	121.1 (2)	C25—C24—H23	120.6
C10—C11—H11	119.5	C24—C25—C20	123.31 (17)
C12—C11—H11	119.5	C24—C25—N3	115.63 (15)
C13—C12—C11	122.2 (2)	C20—C25—N3	121.01 (16)
C13—C12—H12	118.9	C14—N1—C7	107.57 (13)
C11—C12—H12	118.9	C14—N1—S1	120.91 (11)
C12—C13—C14	116.9 (2)	C7—N1—S1	123.89 (12)
C12—C13—H13	121.5	O6—N3—O5	122.39 (18)
C14—C13—H13	121.5	O6—N3—C25	118.61 (16)
C13—C14—C9	121.65 (19)	O5—N3—C25	118.88 (16)
C13—C14—N1	130.80 (18)	O1—S1—O2	120.22 (10)
C9—C14—N1	107.55 (14)	O1—S1—N1	106.22 (8)
O3—C15—C16	122.04 (18)	O2—S1—N1	106.01 (9)
O3—C15—C8	119.14 (17)	O1—S1—C1	109.63 (10)
C16—C15—C8	118.71 (16)	O2—S1—C1	109.04 (10)
C15—C16—C17	114.9 (2)	N1—S1—C1	104.52 (8)
C15—C16—H16A	108.5		
C6—C1—C2—C3	-1.9 (4)	C25—C20—C21—C22	0.3 (3)
S1—C1—C2—C3	-178.8 (2)	C19—C20—C21—C22	177.18 (19)
C1—C2—C3—C4	-1.3 (5)	C20—C21—C22—C23	-1.1 (3)
C2—C3—C4—C5	3.6 (5)	C21—C22—C23—C24	1.4 (3)
C3—C4—C5—C6	-2.7 (5)	C21—C22—C23—C11	-179.18 (17)
C2—C1—C6—C5	2.7 (4)	C22—C23—C24—C25	-0.7 (3)
S1—C1—C6—C5	179.7 (2)	C11—C23—C24—C25	179.79 (14)
C4—C5—C6—C1	-0.4 (5)	C23—C24—C25—C20	-0.1 (3)
N1—C7—C8—C9	-1.87 (19)	C23—C24—C25—N3	177.42 (17)
C18—C7—C8—C9	179.19 (17)	C21—C20—C25—C24	0.4 (3)

N1—C7—C8—C15	174.39 (17)	C19—C20—C25—C24	-176.37 (17)
C18—C7—C8—C15	-4.6 (3)	C21—C20—C25—N3	-177.05 (18)
C7—C8—C9—C10	179.8 (2)	C19—C20—C25—N3	6.2 (3)
C15—C8—C9—C10	3.2 (3)	C13—C14—N1—C7	179.4 (2)
C7—C8—C9—C14	0.9 (2)	C9—C14—N1—C7	-1.50 (18)
C15—C8—C9—C14	-175.62 (16)	C13—C14—N1—S1	28.8 (3)
C14—C9—C10—C11	-1.5 (3)	C9—C14—N1—S1	-152.11 (13)
C8—C9—C10—C11	179.8 (2)	C8—C7—N1—C14	2.11 (19)
C9—C10—C11—C12	1.6 (4)	C18—C7—N1—C14	-178.85 (15)
C10—C11—C12—C13	-0.7 (4)	C8—C7—N1—S1	151.63 (13)
C11—C12—C13—C14	-0.3 (4)	C18—C7—N1—S1	-29.3 (2)
C12—C13—C14—C9	0.4 (3)	C24—C25—N3—O6	-155.4 (2)
C12—C13—C14—N1	179.3 (2)	C20—C25—N3—O6	22.2 (3)
C10—C9—C14—C13	0.5 (3)	C24—C25—N3—O5	20.7 (3)
C8—C9—C14—C13	179.57 (18)	C20—C25—N3—O5	-161.72 (19)
C10—C9—C14—N1	-178.64 (17)	C14—N1—S1—O1	-178.89 (13)
C8—C9—C14—N1	0.39 (19)	C7—N1—S1—O1	35.43 (16)
C7—C8—C15—O3	121.1 (3)	C14—N1—S1—O2	-49.93 (15)
C9—C8—C15—O3	-63.1 (3)	C7—N1—S1—O2	164.38 (14)
C7—C8—C15—C16	-62.5 (3)	C14—N1—S1—C1	65.22 (15)
C9—C8—C15—C16	113.3 (2)	C7—N1—S1—C1	-80.47 (15)
O3—C15—C16—C17	6.7 (4)	C6—C1—S1—O1	-37.0 (2)
C8—C15—C16—C17	-169.6 (2)	C2—C1—S1—O1	139.98 (19)
C8—C7—C18—C19	-39.8 (3)	C6—C1—S1—O2	-170.50 (19)
N1—C7—C18—C19	141.39 (18)	C2—C1—S1—O2	6.5 (2)
C7—C18—C19—C20	-178.51 (16)	C6—C1—S1—N1	76.5 (2)
C18—C19—C20—C21	37.1 (3)	C2—C1—S1—N1	-106.5 (2)
C18—C19—C20—C25	-146.33 (19)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C10—H10...O1 <sup>i</sup>	0.93	2.60	3.327 (3)	136
C16—H16B...O5 <sup>ii</sup>	0.97	2.37	3.260 (3)	152

Symmetry codes: (i) *x*, *y*+1, *z*; (ii) *x*+1, *y*, *z*.