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Data on A parametric temperature dependent potential for β -PbF₂: A numerical investigation by molecular dynamics



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ABSTRACT

This article presents the data on a parametric temperature dependent potential for β -PbF₂ using molecular dynamics (MD) simulations in the rigid ion approach. The β -PbF₂ is an important ionic conductor that exhibit a super ionic behavior at 711 K. The understanding of the temperature effect in its properties is crucial for possible applications in electrode for solid state batteries, Cherenkov detectors, and rare earth host for scintillation screen. The simulations were done in the DL_POLY Classic 1.9 package employing the Buckingham pair-potential type. The data have not been reported nor discussed in the research paper to be submitting.

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Table 2Lattice parameter for different ρ_{ij} values at 930 K.

$\rho(T)$ (eV)	Lattice parameter a_0 (Å)
0.495	5.941
0.496	5.950
0.497	5.985
0.498	6.002
0.499	6.165
0.500	6.120

Table 3Adjustment constants of the potentials that describe the β -PbF₂ by MD [11].

Atomic pairs	A_{ij} (eV)	ρ_{ij} (Å)	C_{ij} (eV Å ⁶)
Pb – Pb	0.0	0.0	0.0
Pb – F	122.7	Tab. 1–2.	0.0
F – F	10255	0.225	107.3

2. Experimental design, materials, and methods

2.1. MD simulation detail

The data reported here was obtained using DL_POLY Classic 1.9 package develop by Smith et al. [8] at the Daresbury Laboratory. In this work, the calculations were performed in a cubic simulation box with 768 atoms and long size of 23.720 Å. VESTA [9] was used to prepare the unit cell, while the supercell was created with AtomsK package [10]. Periodic boundary has been used in order to reproduce the bulk properties. The system was previously equilibrated at environmental conditions: 300 K and 1 atm, respectively. In order to compute the crystal expansion (lattice parameters), the simulations were performed in a NVT ensemble and then relaxed into a NPT ensemble, where the number of the ions (N), temperature (T) and pressure (P) are kept constant. A 5 fs integration time is used to find the $\rho(T)$, then a 1 fs integration time is used to performance a new simulation at the $\rho(T)$ correct values, with finality

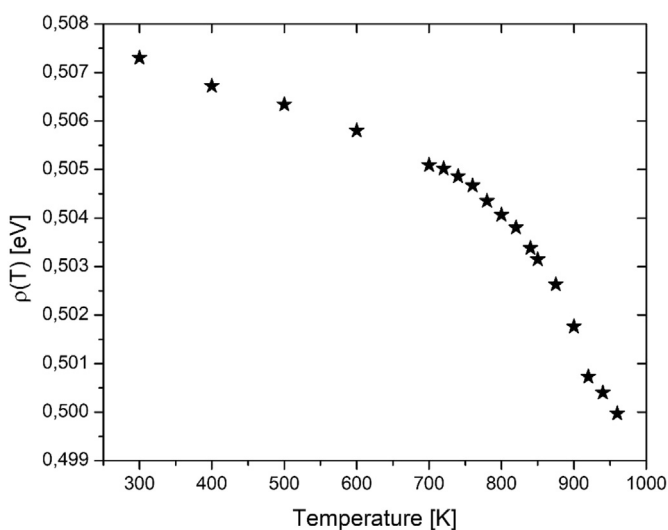
**Fig. 1.** Values for the adjustment parameter $\rho(T)$ as a function of temperature.

Table 4Values of the enthalpy as a function of temperature for β -PbF₂.

Temperature (K)	Enthalpy (kJ mol ⁻¹)
300	-2239.19618
350	-2236.83185
370	-2235.87824
400	-2234.44629
420	-2233.46642
450	-2231.99567
470	-2230.98597
500	-2229.42186
520	-2228.37485
550	-2226.68132
570	-2225.5548
600	-2223.77367
620	-2222.5076
650	-2220.55895
670	-2219.266
700	-2217.2881
720	-2215.84356
750	-2213.80256
770	-2212.31626
800	-2210.2459

to corroborate the accurate lattice parameter at each temperature. In both cases, a 10 Å cutoff is employed, and the Ewald sum is used to compute the Coulomb long range potential. The used potential parameters are summarized in Table 3. The data is obtained from Walker et al. [11] as well the modifications proposed in the Section 1.

2.2. Parametric temperature dependent potential for β -PbF₂

From the Tables 1 and 2 a linear fitting is done for each temperature. In order to find the better lattice parameter value, thermal expansion measurements for PbF₂ obtained by Goff et al. [7] by neutron diffraction at different temperatures were employed. The $\rho(T)$ values are shown in Fig. 1.

In order to make a first approximation on the validation of the potential data shown in Fig. 1, the enthalpy of the atomic system was recorded (refer to Table 4).

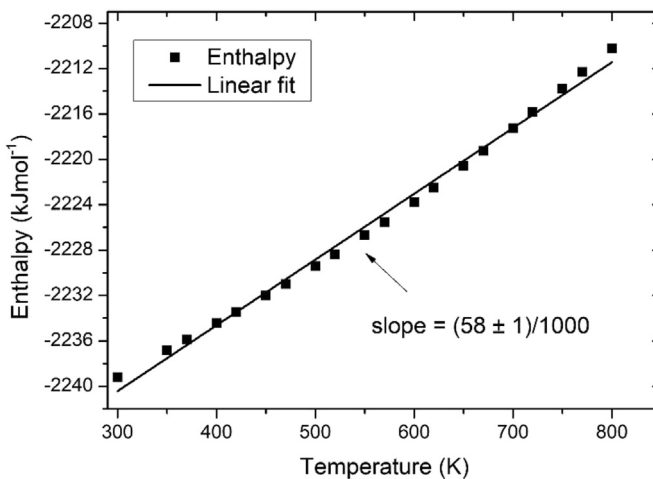


Fig. 2. The enthalpy for β -PbF₂. In filled square, data obtained by MD; in solid line, the linear fit.

The specific heat capacity at constant pressure, C_p , is calculated from Table 4 and the slope of the linear fit shown in Fig. 2. The C_p value obtained by MD is $58 \pm 1 \text{ J mol}^{-1} \text{ K}^{-1}$ which is in acceptable agreement with reported value of $69 \pm 7 \text{ J mol}^{-1} \text{ K}^{-1}$ between 400 and 640 K [12].

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Conflict of Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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