scientific reports



OPEN

Cross-linking of fungal β -glucosidase on Al_2O_3 nanocrystals synthesized using *Cajanus cajan* L. Millsp. extracts for *in suit* genistein manufacture

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This present study deals with the cross-linking of fungal β -glucosidase on Al_2O_3 nanocrystals (NCs) synthesized in *C. cajan* for in-suit genistein production. The *Cajanus cajan* leaves were dried and used to prepare their extract at 65 °C by agitation for 30 min. For enzyme production under submerged culture, 50 mL of medium at pH 8.6 with an inoculum volume of 2 mL; was incubated for 72 h with optimized parameters at 30 °C. The Al_2O_3 NCs were synthesized by adding 30 mM Al_2NO_3 to 25 mL of leaf extract with NaOH at 65 °C for 50 min which enhanced the β -glucosidase specific activity when immobilized. Genistein by biotransformation was obtained using both free (0.67 ± 0.42 mg/ mL) and Al_2O_3 immobilized β -glucosidase (1.3 ± 0.66 mg/mL) for 48 h. The substrate level and enzyme concentration were 2.5 and 1 mL respectively. The UV visible spectra for leaf extract; free and cross-linked β -glucosidase and Al_2O_3 NCs were at 225, 235, 300, and 210 nm. The bands for Al_2O_3 NCs were achieved at 500–750 cm⁻¹ which showed the FTIR analysis to check the change in functional groups of free and Al_2O_3 cross-linked β -glucosidase. In XRD analysis, peaks depicted the crystalline structure of Al_2O_3 NCs ranging from 10–50°. The size of NCs was confirmed by using different magnifications (1.01, 2.00, 3.00, 5.00, 7.02, and 10 K X) of SEM images obtained. For zeta potential measurements, the peak was obtained at -21.0 mV.

Keywords Green synthesis, Aspergillus oryzae, Cost-effective, Enzyme, X-ray diffraction (XRD), Zeta Sizer

Enzymes play a crucial role in modern industries because of their efficient and specific properties¹. Among these, β -glucosidase enhances the estrogenic activity by converting glycoside to aglycone isoflavone. The enzyme immobilized on metallic nanocrystals (NCs) can improve its stability as required for effective biotransformation of sophoricoside to genistein². The Al_2O_3 NCs were being used for the first time to immobilize genistein. Naturally, the leaves of *Cajanus cajan* common name "Pigeon pea" L. Mill sp. contain a substantial amount of sophoricoside which can be extracted in liquid media. Further, the sophoricoside³ can be microbiologically converted to genistein using an extracellular β -glucosidase from *A. oryzae*^{4,5}. The *A. oryzae* is known for its subtle character due to its presence between intra plus interspecies and also its structure analogous to that of

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A. flavus. It has a sturdy competence to secrete huge amounts of hydrolytic enzymes plus post-translational modification 6,7 . β-Glucosidase is apprehensive about the succession of glycosidic bonds and their cleavage 8 . β-Glucosidase is a set of enzymes that are classically divided in GH1 as well as GH3 families, by less occurrence in GH families 9,10 . The β-glucosidase enzyme is responsible for β-glycosidic linkage hydrolysis in amino, aryl, and oligo groups 11 . B-Glucosidases used for synthetic special effects also play an important role in both biological as well as biotechnological processes 12 . β-D-Glucoside glucohydrolase is the systemic name of β-glucosidase and is found in miscellaneous Aspergillus species for its production 13 . C. cajan is also known as the 'orphan crop' and 'poor people's meat' because of its great protein content and amino acids such as methionine, lysine as well as tryptophan, and is a vital source of dietary vitamins plus minerals, particularly B vitamins. Pea plant is liable to a variety of fungal diseases such as foot, root as well as collar rots 14,15 . Extracts from two pea cultivars increased the activity of several enzymes, importantly β-glucosidase and xylanase. C. cajan was mainly used for many years to treat diabetes, sores, skin irritation, hepatitis, measles, jaundice, and dysentery as well as to expel bladder stones and stabilize the menstrual period 16 . Pigeon pea leaves in Chinese folk medicine are widely used to stop the blood, as a painkiller furthermore to kill parasites. It has many therapeutic applications as antioxidant, anticancer, antitumor, antimalarial, and antibacterial properties 17 .

Different analytical techniques such as SEM as well as XRD were used for the determination of NCs size and other properties ^{18,19}. UV-visible spectroscopy is used for the quantization of NCs with accuracy²⁰. Zeta potential measurements were used for the establishment of the size of NCs at extremely different levels. Scanning electron microscopy (SEM) was performed on cross-linked enzymes in toting up to carry the control by use of the XL-30 ESEM model². The XRD analysis is employed through copper anode radiation by using a diffractometer²¹.

The research depends on the cross-linking of β -glucosidase from Aspergillus oryzae and the green synthesis of Al_2O_3 NCs from C. cajan. Remarkably, the study manifests the foremost exploitation of C. cajan which is also known as "pigeon pea", to take it as a resource of sophoricoside for in-suit genistein production. Moreover, it was the earliest illustration to employ for Al_2O_3 NCs synthesis of C. cajan leaf extracts. An exclusive immobilization method for cross-linking of β -glucosidase by Al_2O_3 NCs was formed as Al_2O_3 NCs endow with chemical stability, mechanical strength, and high surface area plus used for the biotechnological and biomedical processes. The in-suit produced genistein is a precious and helpful amalgam that contains considerable due to its capability to have a chemical structure similar to that of estradiol also called phytoestrogen and health benefits for the sake of treatment of many diseases. The pioneering advancement has the prospective to improve the in-suit genistein production by the proficient employment of sophoricoside and the immobilization method for cross-linking of β -glucosidase on Al_2O_3 NCs.

Materials and methods

All chemicals and reagents used in the present study were of analytical grade and were obtained from the chemical store of IIB, GC University Lahore.

Collection of C. Cajan L. Mill sp. leaves

The cleaned and washed leaves of *C. cajan* were obtained from the Government College University Botanic Garden, Lahore (N 31° 33' 24.102", E 74° 19' 40.8432").

Preparation of leave extracts

The fully dried leaves of *C. cajan* were ground into semi-amorphous powder. Twenty-five mL (25mL) of distilled water with 0.5 g of leaf powder was placed in a conical flask of 100 mL and kept for agitation at 80–120 rpm for 30 min at 30 °C. After that, the extract was centrifuged at 3500 rpm for 20 min using a centrifuge model Sigma Laboratory Centrifuges 3K30.

Optimization of different extracting parameters of C. Cajan leaves

The leaf powder (0.5 g) was added in each flask containing 25 mL of distilled water, deionized water, potassium acetate buffer, phosphate buffer, and weakly alkaline distilled water plus saline peptone. Then the saline-peptone water was used as an optimized extracting agent. The peptone (0.85% w/v) and NaCl (0.05% w/v) were added to 100 mL of distilled water at natural pH for the preparation of saline-peptone. The saline peptone water with six different levels was (6.25 mL, 12.5 mL, 18.75 mL, 25 mL, 31.25 mL, 37.5 mL) set on agitation at different temperatures i.e. 40 °C, 50 °C, 60 °C, 70 °C for 75 min at 80–120 rpm.

Production of extracellular β-glucosidase

Organism and fermentation procedure

Wild-type A. oryzae Isl-9 was acquired from the Culture Bank of the Department of Microbiology, Dr. Ikramul-Haq Institute of Industrial Biotechnology, GCU Lahore. The strains were preserved for a short term at 4 °C on the potato dextrose agar slants for use. The 250 mL Erlenmeyer flasks for submerged media were obtained for β -glucosidase production with 50 mL of medium consisting of 0.3 g/l sucrose, 0.4 g/l NH $_4$ Cl, 0.5 g/l NaCl, 0.1 g/l K $_2$ HPO $_4$, as well as 0.05 g/l MgSO $_4$ at (11 pH). The medium was then autoclaved for sterilization at 15 psi (121°C) for 20 min. Then, 1 mL of conidial suspension was added to the cooled autoclaved media 22 and incubated at 30°C in a shaking incubator for 72 h 23 . After filtration, clear supernatant was stored at -4 °C for the enzyme assay as well as for biotransformation.

Optimizing parameters for β -glucosidase production

The cane sugar and beet sugar were employed as carbohydrate sources by using eight altered concentrations from 0.75, 1.5, 2.25, 3.75, 4.5, 5.25, to 6% w/v. The $\rm K_2HPO_4$ and $\rm KH_2PO_4$ were taken for the sake of potassium source with six different concentrations from 0.025, 0.05, 0.075, 0.1, 0.125, to 0.15% w/v. The submerged medium was

Fig. 1. Biotransformation of sophoricoside to genistein.

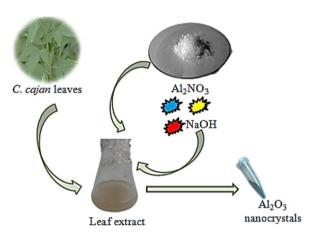


Fig. 2. Green synthesis of Al₂O₂ NCs by reaction of Al₂NO₂ with NaOH in C. cajan leaf extract.

incubated at four different temperatures 24 °C, 30 °C, 36 °C and 42 °C. At two different periods of 30 and 60 min, a β -glucosidase activity assay was obtained.

Biotransformation of natural sophoricoside to genistein

For biotransformation as shown in Figs. 1 and 25 mL of free and cross-linked enzyme supernatant was added with equivalent leaf extract having sophoricoside at basic pH in 250 mL of the conical flask⁴. After that, this aqueous solution was incubated for 73 h at 35°C.

Optimizing parameters for sophoricosidase biotransformation

The free and cross-linked β -glucosidase (2mL) separately was added in test tubes with six changed substrate levels (leaf extract) from 0.5, 1, 1.5, 2, 2.5 to 3mL and the volume made up to 5mL by adding distilled water along with the blank without leaf extract ran parallel to other test tubes for incubation.

Green synthesis of Al2O3 NCs

The green synthesis of ${\rm Al_2O_3}$ NCs using plant extracts was obtained because the plant source was free from toxicants as well as was set up to be at low cost reported by Manikandan et al. ²⁴. The NCs were prepared by having 50 milliliters of ${\rm Al_2NO_3}$ solution and stirred at 37 °C for 10 min while adding ten milliliters of leaf extract. The same method was also carried out to compare with distilled water. One molar NaOH was added drop-wise (till pH 11.0 was achieved) to the solution for its reaction with ${\rm Al_2NO_3}$ as shown in Fig. 2. The change in the color of the solution from light yellow to dark yellow was acquired at 60 °C for half an hour correspondence to the synthesis of ${\rm Al_2O_3}$ NCs. The NCs were then centrifuged at 6,000 rpm for 20 min to yield large pellets and the NCs were dried at room temperature ¹⁹.

For efficient green synthesis of Al₂O₃ NCs, three different parameters were optimized. Six concentrations of Al₂NO₃ (15, 30, 45, 60, 75, and 90 mM) were employed and agitated at 65 °C for 30 min from 80 to 120 rpm.

Immobilization by cross-linking of β-glucosidase on Al2O3 NCs

For the immobilization through cross-linking method, Al_2O_3 NCs (100 mg) were added to 0.14 Units of β -glucosidase for 20 h in phosphate buffer saline (PBS) of 10 mM (pH 7.0) at 4 °C. The PBS ran thoroughly to Al_2O_3 immobilized β -glucosidase while unbounded enzymes and impurities were washed out by centrifugation at 6000 rpm for 5 min².

Characterization analysis of Al2O3 NCs

Several characterization techniques were employed as UV-VIS spectroscopy, FTIR spectroscopy, XRD analysis, and SEM.

UV-VIS spectroscopy

UV visible spectroscopy was obtained using a UV visible spectrophotometer in the Nanotechnology Laboratory of the Department of Chemistry, GCU Lahore. The sample was all set by the above-mentioned optimized conditions. The diluted sample deliberated by light absorption by diverse rays with a variety of ranges of wavelength. The free and Al_2O_3 linked β -glucosidase was taken as 3 mL in a quartz cuvette and observed under a UV-VIS spectrophotometer from 200 to 800 nm range investigated by De Caro and Claudia²⁵.

Fourier transform infrared spectroscopy

The analysis was employed to reveal the functional groups with capping as well as reducing characters. The diluted samples of free β -glucosidase and Al $_2$ O $_3$ immobilized β -glucosidase were sent to the Centre for Advanced Studies in Physics (CASP), GCU Lahore. The spectrum was recognized on a machine model of Spectrum-100, Perkin-Elmer, and St. Louis, USA. The % transmittance was estimated to range from 4000 to 400 per centimeter at 25 °C 26 . A diminutive quantity of the samples was positioned on a holder. The signal from the machine was calculated for the both blank and the samples. The spectra were obtained for the presence of the peaks to show the change of functional groups, while on the other hand, detected data was transferred to the CD to save the information of the peaks.

XRD analysis

The liquid sample of nano-Al $_2$ O $_3$ - β -glucosidase was directed to the Central Research Laboratory, Lahore College for Women University (LCWU) for XRD analysis. The machine model (D8 discover, Bruker) was set at an angle from 10° to 80° with a scan speed of 0.05/s with 2 theta determining the crystalline nature of NCs that immobilize the enzyme. The detector produced at a wavelength of (0.160 nm) by having a Cu anode 27 . The sample in powder form was positioned on the holder and then x-ray crossed through the sample on the holder. The X-ray hit on the sample and diffracted all along causing interference and scattering.

SEM analysis

SEM was carried out for confirmation of size plus topography. The powder sample of Al_2O_3 immobilized β -glucosidase was delivered to Central Research Laboratory, LCWU. The SEM model was from ZEISS Company of EVOLS10. The size distribution of the Al_2O_3 immobilized enzyme was investigated. The sample was weighed down to the chamber in the machine. The beam of electrons passed through the chamber containing a sample of NCs and from different angles the signals were produced. The scanning electron micrographs were executed by different magnifications from 40,000X to 100,000X observed by Bejar et al. ²⁸.

Zeta potential measurement

For the zeta distributions, the sample was prepared in powdered form for analysis and was transferred to Lahore University of Management Sciences (LUMS), Lahore. The sonication process was done for the sample and then inspected by Zetasizer Nano ZS instruments model number (Malvern Instruments Ltd, Zetasizer Ver. 7.10). The examination was accomplished in a range from -150 mV to 150 mV apparent at $37 \, ^{\circ}\text{C}^{29}$. The measurement was attuned at $5.5 \, \text{millimeters}$ with the size of $-21.0 \, \text{mV}$.

β-Glucosidase activity assay and protein determination

The enzyme activity was evaluated by using the method presented by Watanabe et al. With minor adjustments. The one unit of enzyme was known as the mass of enzyme that yielded 1 μ mol of p-NP per minute in the conditions provided by the assay. The enzyme activity progress was considered when 0.1% p-NPG (0.2 mL) was added to the 0.2 mL of phosphate buffer with pH 6 plus 0.1 mL of enzyme supernatant incubated at 30 °C. After that the reaction stopped with the addition of 1 M Na₂CO₃ (2.5 mL) and the reaction ran parallel to the blank 1. The experiment was placed in duplicate. Moreover, the discharge was calculated at 400 nm. At the same time, an array of several concentrations of the reacted solution was employed to put up the standard curve by using Bradford's reagent 12.

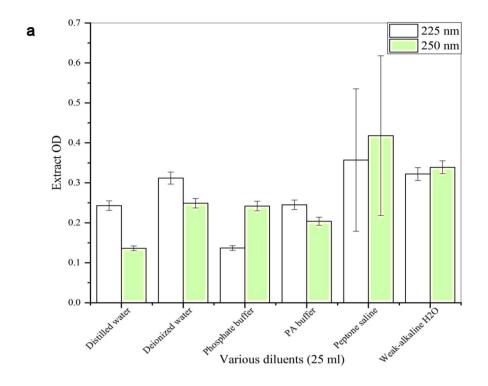
Statistical analysis

All the investigations were carried out in triplicate with mean ± standard deviation. All plots were drawn using the Origin95 software. XRD was plotted using X'pert HighScore plus, and SEM sizing was completed using ImageJ software. The p-value < 0.05 was considered significant.

Results and discussion Preparation of *C. Cajan* leaf extracts

Effect of various diluents for leaf extraction

The partially amorphous leaves of C. cajan were employed to check the effect of different diluents to get the optimized option for leaf extraction preparation. The results are given in Fig. 3a. The peptone essentially used for the diluent as an optimized extractant includes 2% ethanol plus 0.1% (w/v) dextrose. Whereas peptone saline was all set by 0.085% (w/v) peptone along with 0.005% (w/v) NaCl. The outcome of the optimized value was achieved at a pH of 7. The peptone saline was exploited at OD of 0.357 as well as 0.418 among the wavelengths at 225 nm plus 250 nm correspondingly. The biggest pellet by peptone saline was obtained as $(0.2\pm0.07 \text{ mg/mL})$ by 225 nm moreover $(0.2\pm0.10 \text{ mg/mL})$. It could be due to the favorable activity of peptone-saline as the best hydrosylate while the other ethanol or aqueous extracts also show high antioxidant activity with high phenolic concentration. It revealed that they have important constituents using polar solvents as reported by Mahitha et al.³³. Lastly, the peptone saline may have no extra worth than previous sources such as chloroform along with acetone. Other than this peptone reduced the unenthusiastic effects of oxidative tension in leaf tissues



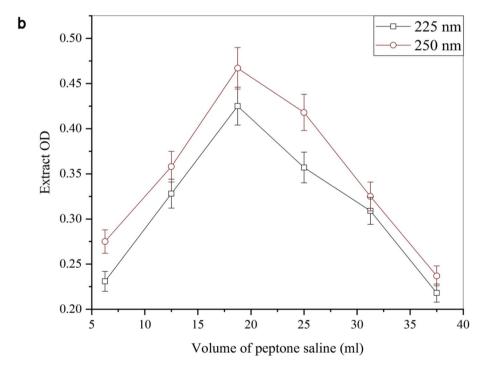


Fig. 3. Preparation of extracts from the pre-treated *C. cajan* leaf powder by using various diluents (a), different volume of peptone-saline (b), and different temperatures (c) at pH 7.0 with volume of diluent 25 mL at 30 °C and error bars at $\rho \le 0.05$ indicating standard deviation amongst the values.

of *C. cajan* furthermore is highly important for having nitrogenous substances as well as is great for amino acids concentration³⁴. In previous studies, petroleum ether, ethyl acetate, and n-butanol were also used for the preparation of leaf extracts as compared to the peptone-saline³⁵. The leaf extract was used for the synthesis of NCs and peptone-saline produced smaller sizes of NCs than other diluents. So it's important to consider the diluent which gives reliable results.

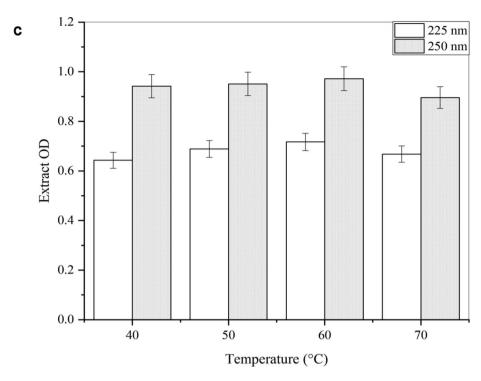


Figure 3. (continued)

Effect of volume of peptone saline for leaf extraction

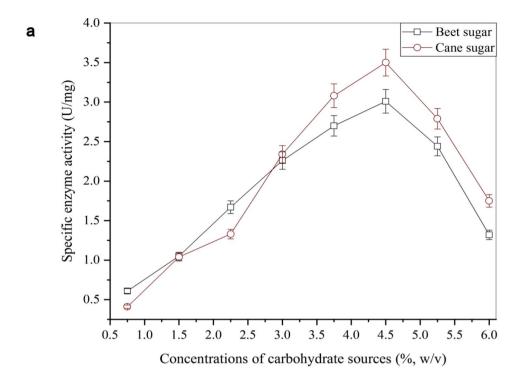
The outcomes of the volume of peptone saline for C. cajan leaves extraction were executed. The result is given in Fig. 3b. The variety of volume with several concentrations (6.25 mL, 12.5 mL, 18.75 mL, 25 mL, 31.25 mL, and 37.5 mL) was used for the extract preparation. The suitable outcome was as $(0.31 \pm 0.078 \text{ mg/mL})$ plus (0.34 ± 0.08 mg/mL) was achieved with 18.75 mL; also with the volume of peptone saline as 31.25 mL. However, 18.75 mL has a privileged fold of pellet while the other volumes with lower pellets and the ODs were 0.425 plus 0.467 at 225 as well as at 250 nm correspondingly. In Fig. 3b, 250 nm presents a red line that demonstrates the maximum peak with 18.75 mL whereas the minimum peak with 5.50 mL. The wavelength of 225 nm illustrated the black line which showed a similar result. The descriptive standard deviation was designed for 18.75 mL of volume attained at 225 nm for improved effect. Higher than the 31.25 mL, the extract commotion reduced to 0.218 at a wavelength of 225 nm. The extract by acetone confirmed the utmost yield of 17.72 mL which is more relevant to peptone saline slightly than the extract from chloroform and ethyl acetate³⁶. The leaf extract preparation of C. cajan was arranged with 150 mL of ethanol solutions using maceration³³. Depending upon the treatment dose groupings, different volumes of solvent and solute were acquired by Manzo and Vitor³⁷ for the preparation of extracts. 300 mL of methanol and water were used as solvents for the preparation of extracts with C. cajan leaf extract by Rinthong and Maneechai³⁸. Yield of extraction depends on the well-defined optimized volume of diluent which gives controlled and quantified results. With the increase or decrease in volume, the necessary components are discharged from the solvent.

Effect of different temperatures for leaf extraction

The result for different temperatures of C. C cajan leaf powder for the extraction preparation was executed as given in Fig. 3c. Different temperatures (40 °C, 50 °C, 60 °C, and 70 °C) were employed for leaf extraction for other optimized conditions for its preparation. The great appropriate pellets were achieved as $(0.64 \pm 0.07 \text{ mg/mL})$ plus $(0.89 \pm 0.08 \text{ mg/mL})$ for optimization and the ODs were 0.717 plus 0.972 at 225 nm as well as 250 nm at 60 °C. Fig shows the highest peak of 60 °C at 250 nm along with the lowest peak of 40 °C. The same results were obtained at 225 nm which has a maximum peak at 60 °C. The result is in association with the conclusions made by Nagati et al.³⁹ who attained agitation of extract at 60 °C, while 40 °C was reported by Raghuwanshi et al.⁴⁰. The solubility of the extracts increases with the increase in temperature however too high temperature can cause impurities and loss of solvent⁴¹. The same optimal temperature (60 °C) condition was also reported by Nehra et al.³⁶.

Production of β -glucosidase from *A. oryzae* Isl-9 strain under submerged culture Effect of different carbohydrate sources

The different sources of carbohydrate were used in submerged fermentation media intended for the β -glucosidase production and their effects are given in Fig. 4a. Different concentrations from 0.75, to 6% (w/v) with 0.75% (w/v) differentiation of sucrose such as cane sugar as well as beet sugar were evaluated by submerged culture. The concentrations utilized were for confirmation of both sources to gain the prime production of mycelia mass. The largest pellets with 4.5% (w/v) were acquired by both carbohydrate sources used and then compared. The utmost



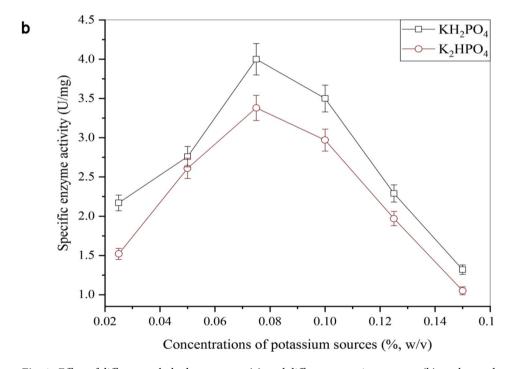


Fig. 4. Effect of different carbohydrate sources (a), and different potassium sources (b) on the production of β-glucosidase from A. oryzae Isl-9 under submerged culture with K_2HPO_4 0.1% (w/v) in 60 min at 30 °C and error bars at $\rho \leq 0.05$ indicating standard deviation amongst the values.

activity of β -glucosidase at 3.5 ± 0.10 U/mg was achieved using cane sugar as a source of sucrose. The highest production of mycelia mass was considered as 2.03 ± 1.06 mg/mL having the greatest peak with cane sugar whereas 1.88 ± 0.85 mg/mL of production was obtained by beet sugar with the lowest peak. But if the sucrose concentration is increased or decreased, the cell mass production also decreases. It may be due to the increase or decrease of other nutrient source concentrations in fermentation media which can cause inappropriate conduct

to *A. oryzae* Isl-9 during shaking investigated by Carlile et al.⁴². The same outcome was revealed in Yu et al.⁴³ studies for the production of β -glucosidase while Liao et al.⁴⁴ presented a production of β -glucosidase with 2% (w/v). Olajuyigbe et al.⁴⁵ reported the β -glucosidase production with 1% (w/v) under submerged culture.

Effect of different Potassium sources

The effects from different sources of potassium were evaluated for the enzyme production by submerged fermentation through *A. oryzae* Isl-9 strain and the comparison between potassium sources used is given in Fig. 4b. These potassium sources were KH_2PO_4 and K_2HPO_4 with different concentrations from 0.025, to 0.15% (w/v) with a variation of 0.025. The high β -glucosidase production was achieved by using KH_2PO_4 (0.075% w/v). The maximum β -glucosidase production was attained by large pellets construction with this potassium source than any other sources. The highest β -glucosidase activity at 4.01 ± 0.87 U/mg was acquired by optimized KH_2PO_4 concentration whereas β -glucosidase activity at 3.38 ± 0.98 U/mg was obtained with K_2HPO_4 concentration. The highest peak with KH_2PO_4 was obtained as 2.67 ± 0.96 mg/mL while the lowest peak was 2.25 ± 0.89 mg/mL with K_2HPO_4 . The potassium source KH_2PO_4 played an important role in the maintenance of pH in fermentation media and both sources provided phosphates to microbe in media attaining optimal conditions. The cellular transport enhanced in the media is also ascribed by KH_2PO_4 . The concentration of KH_2PO_4 is (0.5% w/v) utilized for the production of β -glucosidase via submerged fermentation media production. Later the 0.3% (w/v) concentration of KH_2PO_4 was used in Molina et al. investigation. Results are related to Olajuyigbe et al. as 0.08% (w/v) concentration of KH_2PO_4 was utilized for better production of the enzyme.

Effect of different time intervals

The variation between periods for the β -glucosidase activity assay and their effects are given in Fig. 5a. The β -glucosidase produced by submerged culture medium through *A. oryzae* strain Isl-9. The reaction solution was all set in test tubes having 0.1 mL of filtrate from submerged medium, phosphate buffer (0.2 mL), p-NPG (0.2 mL), and then 2.5 mL of Na $_2$ CO $_3$ was used as blockage for the reaction with periods of 30 and 60 min for incubation. Then comparison between these two time periods was evaluated to find out the highest β -glucosidase activity. With 60 min, the highest β -glucosidase activity was achieved as 2.58 ± 0.67 U/mg, and lower β -glucosidase activity was acquired by a 30 min time period. In this Fig, the time of incubation with 60 min showed the utmost peak of 2.31 ± 0.07 mg/mL and the smallest peak of 2.06 ± 0.04 mg/mL with 30 min. The results showed a resemblance to the work of Almeida et al.⁴⁷. The enzyme activity was attained by evaluating the quantity of p-NP freed from p-NPG.

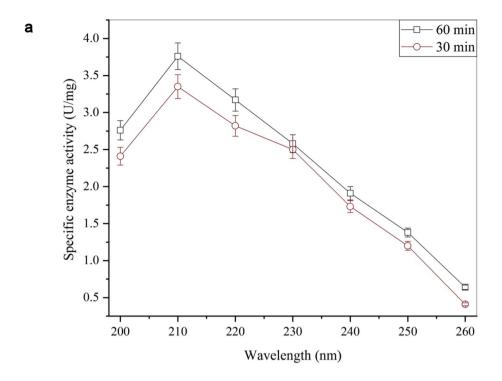
Effect of different incubation temperatures

The effects of the different temperatures of incubation to produce the enzyme from A. oryzae strain through submerged fermentation were attained as given in Fig. 5b. The temperatures for the incubation were from 24 °C, 30 °C, 36 °C, to 42 °C for the comparison of optimization. The high β -glucosidase production was attained at 30 °C. The maximum β -glucosidase production was achieved by presenting large pellets with this optimized temperature than that of the pellets with other temperatures. The uppermost β -glucosidase activity at 5.55 ± 0.081 U/mg was obtained through 30 °C whereas the β -glucosidase activity for other 3 other temperatures was 4.79 ± 0.003, 4.58 ± 0.06 as well as 3.97 ± 0.008 U/mg. The temperature of 30 °C showed the highest peak with 5.0 ± 0.481 mg/mL and the peak with 4.51 ± 0.25 mg/mL was at 24 °C whereas the lowest peaks at 4.12 ± 0.48 mg/mL and 3.37 ± 0.59 mg/mL at 36 °C and 42 °C. The estimated incubation temperature (30 °C) was achieved as the enzyme became inactivated if the temperature became low or higher than this optimized temperature. The stability of the enzyme was significantly affected by other temperatures than this as high temperature can cause denaturation. This parameter was also followed in Yu et al. 43 research studies. The effect of temperature was furthermore given by Yuksekdag et al. 48 which has great influence on the specific enzyme activity. The temperature 32 °C was utilized by Molina et al. 46 as an optimized temperature for the higher production of β -glucosidase with great specific activity.

Characterization of Al2O3 NCs

UV visible spectroscopy

The analysis showed absorbance that corresponds to different ranges of wavelength from 200 nm to 800 nm for several samples of various natures. The sample got enlightened through the rays passed by it; this showed the level of absorbance given by De Caro and Claudia²⁵. The sturdy bands calculated under a wavelength of 300 nm for the Al_2O_3 NCs as given in Fig. 6a. In these spectra, it was presented the line with distilled water as a solvent to Al_2NO_3 has the lowest peak having 0.1 of absorbance with wavelength at 210 nm. The other line presented Al_2NO_3 in *C. cajan* leaf extract as a solvent having a maximum peak at 210 nm of wavelength and absorbance of 0.7. The ODs for both samples were 0.487 and 0.213 for leaf extract and distilled water respectively which showed a higher value for leaf extract as a solvent. The various concentrations of aluminum nitrate were taken for the comparison of optimized results intended for the green synthesis of aluminum oxide NCs as given in Fig. 6b. The spectra showed the green synthesis of Al_2O_3 NCs by using different concentrations of Al_2NO_3 for the synthesis of Al_2O_3 NCs. One line that presented 60 mM showed the maximum peak with 3.75 absorbance at 200 nm as well as the minimum peak with 3.0 absorbance at 225 nm of wavelength. The smallest peak with 75 mM was with an absorbance of 2.5 at 210 nm of wavelength. The optimized concentration of 30 mM with an absorbance of 3.0 showed its peak at 230 nm and the other larger peak with an absorbance of 3.25 at 215 nm of wavelength.



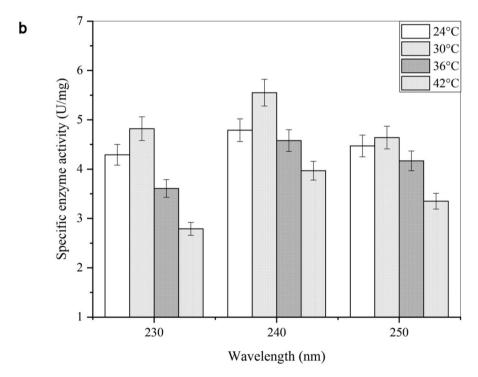
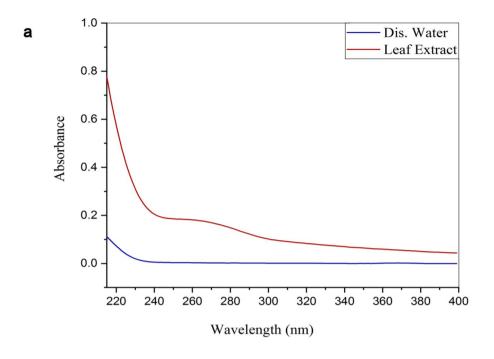


Fig. 5. Effect of different time intervals (a), and different temperatures (b) on the production of β -glucosidase from A. oryzae Isl-9 under submerged culture with cane sugar 4.5% (w/v), KH2PO4 0.075% (w/v) at 30 °C and error bars at $\rho \le 0.05$ indicating standard deviation amongst the values.

 $Fourier\ transform\ infrared\ spectroscopy\ (FTIR)$

By using FTIR spectra were gained with a decree (8 cm $^{-1}$). The transmission range was from 400 cm $^{-1}$ to 4000 cm $^{-1}$ used for FTIR 26 . These spectra presented the occurrence of formLess arrangement and also the chaotic structure between both free and cross-linked enzymes in the peaks. This showed the existence of an improved crystallized point after that presence for the Al_2O_3 NCs also distinguished in these spectra. Notably,



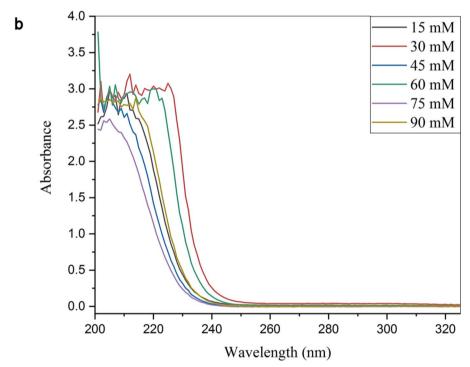


Fig. 6. UV-VIS absorption spectrum (a) effect of different solutions (b) effect of different concentrations of Al₂NO₃.

the incorporation of Al_2O_3 nanocomposites is evident in the immobilized enzyme spectra, as indicated by distinct peaks at 561, 599, and 665 cm $^{-1}$. These peaks correspond to the Fe–O stretching vibrations and the Al–O stretching mode of AlO_6 , respectively, confirming the participation of these functional groups in the synthesis and stabilization of Al_2O_3 NCs. The results were in correspondence to Romero et al. ⁴⁹ as shown in Fig. 7.

X-ray diffraction (XRD)

XRD method for Al_2O_3 NCs cross-linked β -glucosidase was employed as given in Fig. 8. This happened by the scan rate at 0.05/s from ranging in 10–80 degrees. The trial section started with the preparation of the sample in powdered form. The device produced a wavelength of 0.15406 nm by utilizing copper anode as presented by Tsybulya and Kryukovaa²⁷, . The crystalline arrangement of Al_2O_3 NCs depicted peaks accessible at 10.17, 13.32,

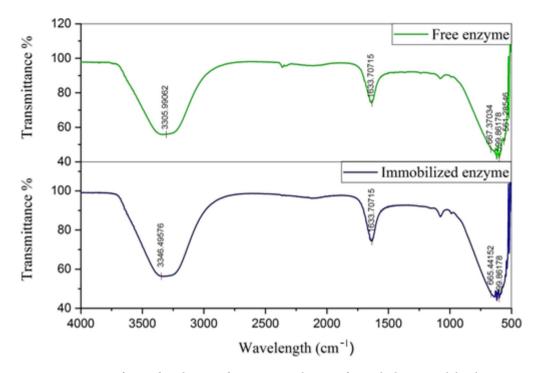


Fig. 7. Fourier transform infrared spectra for comparison between free and Al_2O_3 immobilized enzyme.

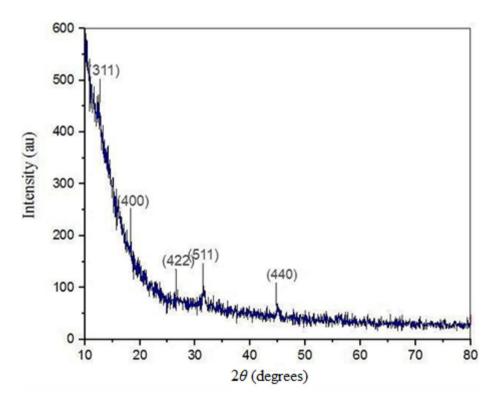


Fig. 8. XRD patterns of Al₂O₃ NCs.

17.83, 27.91, 32.25, 38.92, and 45.56 with real intensities of about 100, 85, 40, 22, 25, 35 and 20% respectively. The crystalline nature of $\mathrm{Al_2O_3}$ NCs plays a pivotal role in enhancing the immobilization and activity of β -glucosidase, as the ordered lattice structure facilitates stable interactions between the enzyme and the support material. Similar trends have been observed results investigated from Talaei et al.⁵⁰ who demonstrated the significance of crystallinity in nanocomposites for improving enzyme functionality. The size investigation by

XRD has shown that NCs have 51.5 nm size. The similar results from XRD spectra were also analogous as evaluated by Piriyawong et al.⁵¹ investigation.

SEM

The Al₂O₃ NCs size distribution was acquired by calculating the sizes of all particles in the sample. All organic and inorganic subjects can be measured on the nanoscale of SEM. The results are shown in Fig. 9. Different magnifications for (1 μm at 5.00 KX in Fig. 9d μm at 1.01, 2.00, and 3.00 KX in Fig. 9a, b,c were exploited to have the micrographs from the sample containing Al₂O₃ NCs. The size of the particles in the sample was obtained through the software of image J as 41.8 nm as shown in Fig. 10. The cumulative particles known as aggregate were performed by different angles at 1 µm to obtain micrographs. The micrographs presented normally the nature as well as the shape of particles in the sample as presented by Guo et al.².

Zeta potential measurements

The results are shown in Fig. 11. The highest peak for zeta potential was -21.0 mV which showed the NCs having a negative charge and dispersed in the medium. The negative value of zeta potential suggested that the NCs were stable. The particles in the sample with zeta potential was enough negative than -30 mV were usually measured secured and in 180 min, 0.8% v/v of concentration was sonicated providing similar results also investigated in Khoshnevisan and Barkhi⁵² studies. The stability observed can be attributed to the surface chemistry of the Al₂O₂ NCs, which provides sufficient repulsion between particles. The negatively charged surface arises from the

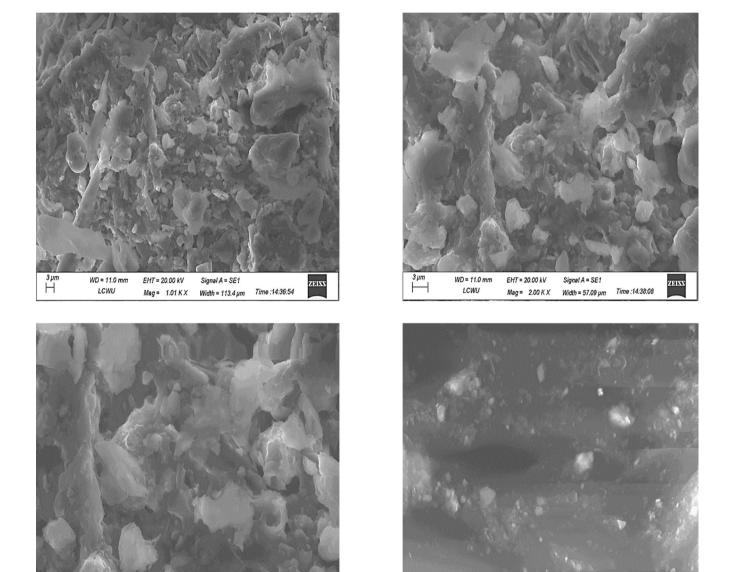


Fig. 9. SEM micrographs of Al₂O₃ NCs immobilized β-glucosidase resolved at different magnifications.

| https://doi.org/10.1038/s41598-025-89973-3

WD = 11.0 mm

EHT = 20.00 kV

Mag = 5.00 K X

Signal A = SE1

Width = 22.87 um

Signal A = SE1

Width = 38.06 µm

Time:14:37:47

EHT = 20.00 kV

Mag = 3.00 K X

WD = 11.0 mm

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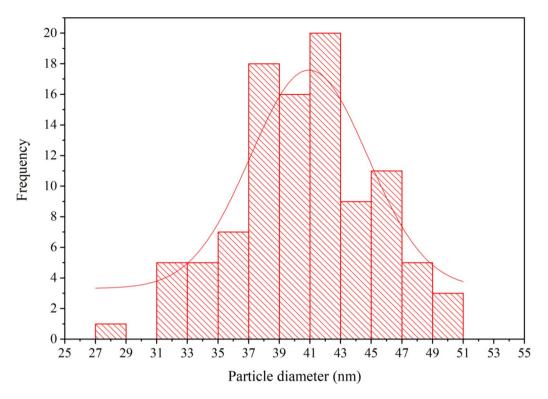


Fig. 10. Histogram of size analysis of Nanoparticles.

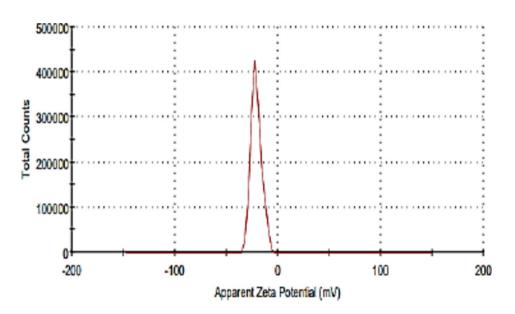


Fig. 11. Zeta potential distribution.

deprotonation of hydroxyl groups or the adsorption of negatively charged species in the medium. Such behavior has been widely reported in the literature, where the surface charge of metal oxides like ${\rm Al_2O_3}$ plays a crucial role in stabilizing NCs⁵³.

Optimization of biotransformation of sophoricoside to genistein Effect of substrate level

For the optimal substrate level intended to biotransform the sophoricosidase for in-suit genistein production by free β -glucosidase and Al₂O₃ NCs cross-linked β -glucosidase, its effects were employed. As an optimizing parameter, substrate level gained a significant role for further studies. The diverse phenolic compounds from the natural sources used in food technology and biological activities; act as anticancers⁵⁴, anti-allergic, anti-

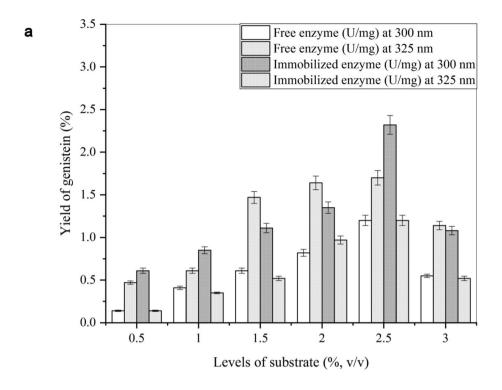
inflammatory, antioxidant, and antidepressant⁵⁵. The different volumes of substrate levels as *C. cajan* leaf extract from 0.5, 1, 1.5, 2, 2.5, to 3 mL) were utilized as 2 mL for both free β-glucosidase as well as the Al_2O_3 NCs cross-linked β-glucosidase. After that the volume in the test tubes containing both enzymes, the volume raised to 5 mL. The comparison between both free and Al_2O_3 NCs cross-linked β-glucosidase was obtained by spectrophotometer at 300 and 325 nm respectively as given in Fig. 12a. The process of biotransformation held for the 3 days at 35 °C through incubation at 160 rpm. The free and Al_2O_3 NCs cross-linked β-glucosidase with wavelengths of 300 and 325 nm created iso-flavones with 1.11 ± 0.38 , 0.62 ± 0.36 , and 1.2 ± 0.68 as well as 0.61 ± 0.39 mg/mL, correspondingly. The optimized level of substrate (*C. cajan* leaf extract) as 2.5% v/v was achieved at 300 nm. The Al_2O_3 NCs cross-linked β-glucosidase illustrated greater outputs than that of free β-glucosidase. The highest yield production of genistein was obtained with immobilized enzyme at 300 nm than free enzyme. It was the optimum level of substrate that produced a good yield of genistein; if the substrate level increased or decreased the yield decreased with it. There was a resemblance of the research results for this parameter to the Singhvi and Zinjarde⁵⁶ investigation on biotransformation for conversion of glycosides to isoflavones. The *C. cajan* leaf extract level utilized in Mei et al.⁴ studies.

Effect of different enzyme concentrations

The effects of using the several changed concentrations of β -glucosidase were employed for the demonstration of optimal to convert natural sophoricoside to suit genistein production. Both the free β -glucosidase and Al₂O₃ NCs cross-linked β-glucosidase enzymes were compared to which one has the best production of genistein. The results demonstrated that the immobilized enzyme achieved superior genistein yields compared to its free counterpart under identical reaction conditions, as shown in Fig. 12b. These findings are consistent with previous studies highlighting the advantages of enzyme immobilization in improving catalytic efficiency, stability, and reusability. For the optimization study, various concentrations of enzymes (0.25, 0.5, 0.75, 1.0, 1.25, 1.5, 1.75, 2.0% v/v) were tested, maintaining a substrate concentration of 2.5% v/v and a reaction time of 48 h. The optimal enzyme concentration for both free and immobilized β-glucosidase was found to be 1.0% v/v. At this concentration, the immobilized enzyme achieved a maximum genistein yield which was significantly higher than the free enzyme. The superior performance of the Al₂O₃ NCs cross-linked enzyme can be attributed to the enhanced stability and reduced diffusional limitations provided by the immobilization matrix⁵⁷. Moreover, the immobilized enzyme exhibited consistent activity over the extended incubation period of 72 h, maintaining its efficiency under continuous agitation at 35 °C. There was increased production of isoflavones with the immobilized enzyme achieving a concentration of 1.3 ± 0.66 mg/mL compared to 0.67 ± 0.42 mg/mL for the free enzyme and these results were similar to that of Pandjaitan et al.⁵⁸. Compared to free enzymes, immobilized enzymes demonstrated a two-fold increase in isoflavone production suggesting their potential for industrialscale applications.

Conclusion

Most patients want to seek natural compounds as an alternative medicine for treatment rather than anti-inflammatory drugs that are accompanied by undesirable effects. That's why this study is an effort to find an innovative and cost-effective approach of producing genistein by immobilizing fungal β -glucosidase on Al_2O_3 nanocrystals synthesized using *Cajanus cajan* (pigeon pea) leaf extracts. The β -glucosidase production by submerged culture was performed and then the highest specific enzyme activity was obtained by immobilization on Al_2O_3 NCs synthesized by leaf extracts. Both the free β -glucosidase and Al_2O_3 NCs cross-linked β -glucosidase were compared for the in-situ production of genistein by natural sophoricosidase. All the results from different analyses are remarkably promising and futuristic for the long-lasting effects of genistein in cancer cells, the pathogenesis of asthma, obesity, arthritis, and neurodegenerative diseases. This method of genistein production is cost-effective and has a high yield with no side products. Moreover, these findings emphasize the green synthesis of nanocrystals, optimized enzyme activity, and biotransformation efficiency, offering substantial potential for applications in biomedicine and biotechnology.



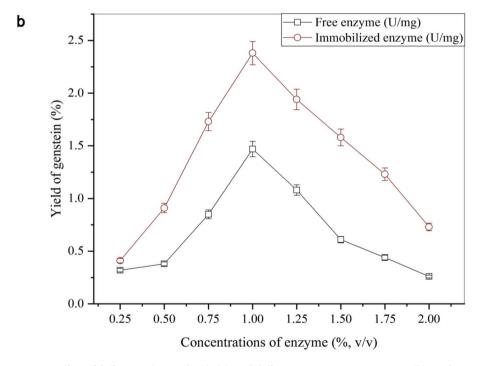


Fig. 12. Effect of different substrate levels (a), and different enzyme concentrations (b) on the biotransformation of natural sophoricoside (in *C. cajan* leaf extract) to genistein by free and immobilized β -glucosidase with rate of biotransformation 72 h, enzyme concentration 2 mL (v/v) and error bars at $\rho \le 0.05$ indicating standard deviation amongst the values.

Data availability

All the data is contained within the article.

Received: 13 August 2024; Accepted: 10 February 2025

Published online: 25 February 2025

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Acknowledgements

The authors extend their appreciation to the Researchers Supporting Project number (RSP2025R418), King Saud University, Riyadh, Saudi Arabia.

Author contributions

A.I: Conceptualization, methodology, software, data curation, writing – original draft; S.A., M.U.A: Methodology, validation, investigation, resources, visualization, supervision, draft revision for reviewers' comments; M.U.H., Z.S.: Software, data curation, project administration, formal analysis, visualization; S.E: Methodology, investigation, validation, project administration; B.A. R.M.A. M.S.E.: Software, data curation, writing – review and editing, visualization; M.A.J., T.M.: Data curation, validation, formal analysis, resources, funding acquisition.

Funding

This research received no external funding.

Declarations

Competing interests

The authors declare no competing interests.

Additional information

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