



Flavanols from Nature: A Phytochemistry and Biological Activity Review

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Abstract: Flavanols, a common class of secondary plant metabolites, exhibit several beneficial health properties by acting as antioxidant, anticarcinogen, cardioprotective, anti-microbial, anti-viral, and neuroprotective agents. Furthermore, some flavanols are considered functional ingredients in dairy products. Based on their structural features and health-promoting functions, flavanols have gained the attention of pharmacologists and botanists worldwide. This review collects and summarizes 121 flavanols comprising four categories: flavan-3-ols, flavan-4-ols, isoflavan-4-ols, and flavan-3,4-ols. The research of the various structural features and pharmacological activities of flavanols and their derivatives aims to lay the groundwork for subsequent research and expect to provide mentality and inspiration for the research. The current study provides a starting point for further research and development.

Keywords: flavanols; flavonoids; natural products; biological activities

1. Introduction

Various flavanols have been isolated from Nature over the past few decades. As their potential beneficial properties have been recognized, they have been increasingly studied by scientists worldwide [1]. Up to now, flavanols have been found in common foods, including cereals, legumes, fruits, vegetables, forages, hops, beers, red wine, tea, cocoa, grapes, and apples [2]. In addition, flavanols were used as quality markers for nuts and cereals as well. Therefore, there is a growing awareness of the benefits of natural flavanols. As a distinct sub-group of flavonoids, flavanols are broadly characterized by the absence of a no double bond between C-2 and C-3 and the absence of a carbonyl group on the C ring (C-4), while featuring a hydroxyl group(s) on C-3 or C-4. Given the above structural characteristics, four main types of flavanols have been found in Nature: (i) flavan-3-ols, (ii) flavan-4-ols, (iii) isoflavan-4-ols, and (iv) flavan-3,4-ols, whose basic skeletons are shown in Figure 1. Flavan-3-ols are the most commonly reported among the four types, followed by flavan-4-ols and flavan-3,4-ols, and lastly, isoflavan-4-ols. The classification and quantity of each category of flavanols are displayed in Figure 2. Due to aggregation, oligomerized flavan-3-ols and oligomerized flavan-3,4-ols have also been discovered in succession. The four subtypes of flavan-4-ols are based on the position of the linked saccharide or the seven-membered oxygen heterocycle between the sugar and aglycone. In general, 121 flavanols and their derivatives were found from the natural plant kingdom, distributed in 52 species and 29 families (Figure 2). They range from simple monomers to oligomers, and from aglycones to glycosides. Through the analysis and data statistics, plants rich-in flavanols are mainly include 10 species (Pronephrium penangianum, Camellia sinensis, Theobroma cacao, Astragalus membranaceus, Rhoicissus tridentate, Juniperus



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Copyright: © 2022 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). communis var. depressa, Camellia sinensis var. assamica, Livistona chinensis, Thespesia populnea and Wisteria floribunda), are marked in blue color (Figure 2) and should be considered in the future development of natural flavanols. Several of these plants were used in the treatment of various diseases. For example, Pronephrium penangiana (previously named Abacopteris penangiana) is widely distributed in the south of China and used as a folk medicine to treat blood circulation stasis, barriers to the study of blood flow, edema and inflammation for patients suffering from metabolic syndrome [3]. Astragalus mongholicus (also named Huangqi in China), a well-known herb amongst traditional Chinese medicine practitioners, has been used to improve resistance to infections, in immunological disorders and viral infections, and used as a hepatoprotective heart tonic, and in the treatment of nephritis, and diabetes [4]. Hovenia dulcis is commonly used in Korean and Chinese traditional medicine to treat liver diseases and as a detoxifying agent for alcohol poisoning. Garcinia dulcis is an Asian medicinal plant used in folk medicines and its stem bark has been used as an anti-inflammatory agent in Thailand [5].







Flavan-4-ols

Isoflavan-4-ols

Flavan-3,4-ols

Figure 1. Basic skeletons of flavanols.



Figure 2. Plant sources of 121 flavanols (Families and species and the corresponding quantity of compounds).

Although flavanols feature good bioactivity, there is no systematic and comprehensive review for phytochemistry and biological activities [6,7]. This review is focused on the phytochemical investigations and biological activities of naturally occurring flavanols, covering the literature from 1974–2021.

2. Flavan-3-ols

Flavan-3-ols can be classified into simple flavan-3-ols, alkaloid flavan-3-ols and oligomeric flavan-3-ols by analyzing the structural characteristics of published compounds (Table 1, Figure 3).

No.	Name	Plant Source	Plant Organ	Ref.
1	(+)-afzelechin	Wisteria floribunda Nelumbo nucifera; Hovenia dulcis; Zizyphus jujuba Mill. var. spinosa	fruit peels seed epicarp stem barks roots	[8] [9] [10] [11]
		Livistona chinensis Thespesia populnea	fruits barks	[12] [13]
2	(+)-afzelechin-7-Ο-β-D-glucopyranoside	Juniperus communis var. depressa Dioscorea villosa	stems and leaves rhizomes	[14] [15]
3	(+)-afzelechin-5-Ο-β-D-glucopyranoside	Wisteria floribunda Dioscorea villosa	fruit peels rhizomes	[8] [15]
4	(+)-catechin	Wisteria floribunda Astilbe rivularis Uncaria gambier Rhoicissus tridentata Zizyphus jujuba Mill. var. spinosa Euonymus laxiflorus Livistona chinensis Thespesia populnea Chaenomeles speciosa Picea abies Lawsonia inermis	fruit peels rhizomes leaves roots roots trunk barks fruits barks fruits root barks aerial parts	[8] [16] [9] [17] [11] [18] [12] [13] [19] [20] [21]
5	catechin-5- <i>O</i> -β-D-glucoside	Chaenomeles speciosa	fruits	[19]
6	(+)-3'-O-methylcatechin-5-O-β-D- glucopyranoside	Juniperus communis var. depressa	stems and leaves	[14]
7	3'- deoxycatechin-3-O-α-L- rhamnopyranoside	Saraca asoca	barks	[22]
8	(+)-4'-O-methylcatechin 5-O-β-D-glucopyranoside	Juniperus communis var. depressa	stems and leaves	[14]
9	(+)-gallocatechin	Rhoicissus tridentata Thespesia populnea	roots barks	[17] [13]
10	(–)-fisetinidol	Bauhinia pentandra Rhoicissus tridentata Acacia molissima	stems roots woods	[23] [17] [24,25]
11	(–)-robinetinidol	Lysidice rhodostegia	barks	[26]
12	(–)-epigallocatechin	Thespesia populnea Zizyphus jujuba Mill. var. spinosa	barks roots	[13] [11]

Table 1. The numbers, plant sources, corresponding organs of flavan-3-ols.

No.	Name	Plant Source	Plant Organ	Ref.
		Astilbe rivularis;	rhizomes	[16]
		Hovenia dulcis	stem barks	[10]
		Iuniperus	stems and	[14]
		communis var. depressa	leaves	[]
13	(-)-catechin	Plicosenalus curviflorus	leaves	[27]
15	() catecrist	Fuonumus laviflorus	trunk bark	[18]
		Artocarnus vanthocarnus	leaves	[28]
		Saraca asoca	icuves	[20]
		<i>Surucu usocu</i>	roots	[21]
			10013	[41]
14	(2S,3R)-3,3',4',5,7-pentahydroxyflavane-5-	Plicosepalus curviflorus	leaves	[26]
	<i>O-</i> gallate	, ,		
15	(–)-afzelechin	Astilbe rivularis	rhizomes	[16]
		Euonymus laxiflorus	trunk barks	[18]
16	(–)-gallocatechin	Champ		
		Livistona sinensis	leaves	[12]
	(2 <i>S</i> ,3 <i>R</i>)-3,3',4',5,5',7-hexahydroxyflavane-		1	
17	4',5-di-O-gallate	Plicosepalus curviflorus	leaves	[26]
		Wisteria floribunda		[8]
		Cassia grandis	fruit peels	[0]
		Cussiu grunuis	leaves	[29]
		Reaction the transfer	leaves	[7]
		Knoicissus triuentata	roots	[17]
10	() and a table	Lysiaice moaostegia	roots	[25]
18	(–)-epicatechin	Juniperus	stems and leaves	[14]
		communis var. depressa	fruits	[10]
		Livistona chinensis	roots	[12]
		Cassia sieberiana	barks	[30]
		Thespesia populnea	fruits	[13]
		Chaenomeles speciosa		[19]
10	3'-O-methylepicatechin-7-O-β-D-	Juniperus communis	stoma and laavaa	[14]
19	glucopyranoside	var.depressa	sterns and leaves	[14]
	3'-O-deoxycatechin-3-O-α-L-			[04]
20	rhamnopyranoside	Turpinia formosana	roots	[31]
	1 5	Camallia cinancia	laguas	[0]
01	$()$ anisotophin 2 \cap collete	Cumentu sinensis	leaves	[9]
21	(-)-epicatecinii-5-0-ganate	Knowski structuru	roots	[17]
			roots	[23]
22	(–)-epicatechin-3- <i>O</i> -β-D-allopyranoside	Turpinia formosana	roots	[31]
23	5-O-methylepicatechin-7-O-β-D-	Juniperus communis	stems and leaves	[14]
20	glucopyranoside	var.depressa	stenis and reaves	
0.4	3,7,4'-trihydroxy-5,3'-dimethoxyflavan-7-	Champer and and filter	11	[20]
24	Ö-β-D-glucopyranoside	Styrax suberifolius	barks	[32]
		Wisteria floribunda	fruit peels	[8]
		Astilbe rivularis	rhizomes	[16]
		Cassia grandis	leaves	[29]
25	(–)-epiafzelechin	Zizyphus jujuba Mill.	roots	[11]
		var. spinosa		[···-]
		Thespesia nonulnea	barks	[13]
		Saraca asoca	barks	[21]
				[+]
26	(+)-epiafzelechin	Wisteria floribunda	fruit peels	[8]
	7- <i>O</i> -β-D-gluco-pyranoside	Saraca asoca	barks	[21]
	3'-O-methylepicatechin-7-O-B-D-	Juniperus		[4,4]
27	glucopyranoside	communis var.depressa	stems	[14]
	U 17	1		

Table 1. Cont.

Table 1. Cont.

No.	Name	Plant Source	Plant Organ	Ref.
28	(2R,3R)-3,5,7,3',5'-penthahydroxyflavane	Camellia sinensis	leaves	[13]
29	(2R,3R)-3,5,6,7,8,3′,5′- heptahydroxyflavane	Livistona chinensis	fruits	[12]
30	(2 <i>R</i> ,3 <i>R</i>)-3,5,6,7,8,4'-hexahydroxyflavane	Livistona chinensis	fruits	[12]
31	(–)-epigallocatechin	Rhoicissus tridentata Camellia sinensis Zizyphus jujuba Mill. var. spinosa	roots leaves roots	[17] [9] [11]
		Thespesia populnea	barks	[13]
32	(–)-epigallocatechin gallate	Camellia sinensis.	leaves	[33]
33	(2 <i>S</i> ,3 <i>S</i>)-3,5,7,3′,5′-pentahydroxyflavane	Livistona. chinensis	fruits	[12]
34	dulcisflavan	Garcinia dulcis	fruits	[34]
35	(—)-6-(5′′S)-2-pyrrolidinone- epiafzelechin	Actinidia arguta	roots	[35]
36	(–)-6-(5''R)-N-ethyl-2-pyrrolidinone- epicatechin-3-O-gallate	Camellia sinensis var. assamica	leaves	[36]
37	(–)-6-(5′′S)-N-ethyl-2-pyrrolidinone- epicatechin-3-O-gallate	Camellia sinensis var. assamica	leaves	[36]
38	(–)-8-(5′′R)-(2-pyrrolidinone-5-yl)-(–)- epicatechin	Camellia sinensis var. assamica Actinidia arguta	leaves roots	[36] [36]
39	(−)-8-(5′′S)-(2-pyrrolidinone-5-yl)-(−)- epicatechin	Camellia sinensis var. assamica Actinidia arguta	leaves roots	[36] [35]
40	(–)-8-(5′′R)-N-ethyl-2-pyrrolidinone- epicatechin	Camellia sinensis var. assamica	leaves	[36]
41	(–)-8-(5′′R)-N-ethyl-2-pyrrolidinone- epicatechin	Camellia sinensis var. assamica	leaves	[36]
42	(–)-8-(5′′R)-N-ethyl-2-pyrrolidinone- epicatechin-3-O-gallate	Camellia sinensis var. assamica	leaves	[36]
43	(–)-8-(5′′S)-N-ethyl-2-pyrrolidinone- epicatechin-3-O-gallate	Camellia sinensis var. assamica	leaves	[36]
44	kopsirachin	Kopsia dasyrachis	leaves	[37]
45	(–)-uncariagambiriine	Uncaria gambir	leaves	[38]
46	lotthanongine	Trigonostemon reidioides; Baliospermum reidioides	leaves leaves	[39] [39]
47	8'-ethylpyrrolidinonyltheasinensin A	Camellia sinensis	leaves	[40]
48	(–)-epicatechin-($2\beta \rightarrow O-7;$ $4\beta \rightarrow 8$)-(+)-catechin	Camellia sinensis	leaves	[41]
49	(-)-epicatechin-($2\beta \rightarrow O-7;$ $4\beta \rightarrow 8$)-(-)-catechin	Theobroma cacao	fruits	[42]
50	3T- <i>O</i> -α-L-arabinopyranosyl-ent- epicatechin- $(2\alpha \rightarrow 7, 4\alpha \rightarrow 8)$ -epicatechin	Theobroma cacao	fruits	[42]
51	3T- <i>O</i> -β-D-galactopyranosyl-ent- epicatechin- $(2\alpha \rightarrow 7, 4\alpha \rightarrow 8)$ -epicatechin	Theobroma cacao	fruits	[42]
52	$3T-O-\alpha-L$ -arabinopyranosyl-ent- epicatechin- $(2\alpha \rightarrow 7, 4\alpha \rightarrow 8)$ -epicatechin	Theobroma cacao	fruits	[42]

No.	Name	Plant Source	Plant Organ	Ref.
53	procyanidin B4	Rhoicissus tridentata Picea abies Camellia sinensis	roots root barks leaves	[17] [20] [41]
54	procyanidin B3	Rhoicissus tridentata Picea abies	roots root barks	[17] [21]
55	fisetinidol-($4\beta \rightarrow 8$)-catechin	Rhoicissus tridentata	roots	[17]
56	fisetinidol-($4\alpha \rightarrow 8$)	Rhoicissus tridentata	roots	[17]
57	procyanidin B2	Theobroma cacao	fruits	[42]
58	procyanidin B5	Theobroma cacao	fruits	[42]
59	methylene 8,8-bis-(+)-catechin	Potentilla fruticosa	leaves	[43]
60	methylene-6,8-bis-(+)-catechin	Potentilla fruticosa	leaves	[43]
61	methylene 6,8-bis-(7-O-glucosyl)-(+)-catechin	Potentilla fruticosa	leaves	[43]
62	theaflavin	Camellia sinensis	leaves	[41]
63	theaflavin-3-galloyl	Camellia sinensis	leaves	[41]
64	theaflavin-3,3'-galloyl	Camellia sinensis	leaves	[41]
65	theaflavin-3'-galloyl	Camellia sinensis	leaves	[41]
66	procyanidin C1	Theobroma cacao	fruits	[41]
67	procyanidin-C1-tri-gallate	Theobroma cacao	fruits	[42]
68		Theobroma cacao	fruits	[42]
69		Theobroma cacao	fruits	[42]
70	3T-O-α-L-arabinopyranosylcinnamtannin B1	Theobroma cacao	fruits	[42]
71	3T-O-β-D-galactopyranosylcinnamtannin B1	Theobroma cacao	fruits	[42]
72	cinnamtannin B1	Theobroma cacao	fruits	[42]
73	cinnamtannin A2	Theobroma cacao	fruits	[42]
74	angular tannin	Acacia mangium	barks	[44]
75	linear tannin	Acacia mangium	barks	[44]
76	twice-angular tannin	Acacia mangium	barks	[44]

Table 1. Cont.

"T" and "ent" mean the top unit of the trimeric flavan-3-ols and enantiomer, respectively.

2.1. Simple Flavan-3-ols

As the most common group of flavanols in the diet, simple flavan-3-ols are found in a wide range of raw-food materials [45]. Due to their significant health benefits, these ingredients are candidates in developing dietary recommendations [46]. Representing the most common flavonoid consumed in the American diet, the flavan-3-ols and their condensation products are regarded as functional ingredients in various beverages, herbal remedies and supplements [47]. They often occur in food and contribute to astringency, bitterness, sourness, sweetness, salivary viscosity, aroma, and color formation as food quality parameters. Furthermore, they are also used as food quality parameters. Moreover, flavan-3-ols also improve food's microbial, oxidative, and heat stability [47].

Structurally, flavan-3-ols rarely occur as glycosides, they are characterized by an absence of a double bond between C-2 and C-3, the presence of a hydroxyl group on position C-3, and the absence of the C-4 carbonyl in ring-C. Thus, flavan-3-ols have two chiral centers at C-2 and C-3.

Simple flavan-3-ols



ЮΗ

OН

HO.

R₃O



.OH

OR₄

ЮН

.OH

(2R,3R)

ÓR

(2S,3R)

HO.



13 R = H

14 R = galloyl

10

ΌΗ



27 $R_1 = Glu, R_2 = OCH_3$





11

15

ÓН

′ОН

'OR₁









 $\begin{array}{l} \textbf{19} \ R_1 = R_2 = H, \ R_3 = Glu, \ R_4 = CH_3 \\ \textbf{20} \ R_1 = Rha, \ R_2 = R_3 = H, \ R_4 = H \\ \textbf{21} \ R_1 = galloyl, \ R_2 = R_3 = H, \ R_4 = H \\ \textbf{22} \ R_1 = All, \ R_2 = R_3 = H, \ R_4 = H \\ \textbf{23} \ R_1 = H, \ R_2 = CH_3, \ R_3 = Glu, \ R_4 = H, \\ \textbf{24} \ R_1 = H, \ R_2 = CH_3, \ R_3 = H, \ R_4 = OCH_3 \end{array}$





Figure 3. Cont.



Figure 3. Cont.



Figure 3. Structures of flavan-3-ols.

At present, a total 34 simple flavan-3-ols have been reported. Generally speaking, based on the two chiral centers C-2 and C-3 of ring-C, flavan-3-ols can exist as four possible stereoisomers: (2*R*,3*S*), (2*S*,3*R*), (2*R*,3*R*), (2*S*,3*S*), as shown in Figure 3. Most of the natural flavan-3-ols are 2*R* isomers (compounds **1–11**, **17–32**) while 2*S* isomers (**12–16**, **33**, **34**) only occur in a few plants such as *Euonymus laxiflorus*, *Ulmus davidiana* and *Plicosepalus curvifloru*. Hydroxyl is the most common group on ring-A and ring-B of flavanols. The substitution positions of hydroxyl groups on the ring-A of most simple flavan-3-ols are mainly at C-5 and C-7, while a few compounds like **9**, **10** and **17** only have one hydroxyl group at the C-7 position. The main substitution positions of hydroxyls on ring-B are C-3', C-4' and C-5'. Furthermore, hydroxyl groups at different positions could also be methylated, esterified with gallic acid, and generate glycosidic bonds with different monosaccharides such as glucose, rhamnose and allose.

2.2. Alkaloid Flavan-3-ols

In the review, a group of exceptional flavan-3-ols attached to pyrrolidone, indole moiety and skytanthine moieties (compounds **35–47**) is uniformly classified as alkaloid flavan-3-ols [48]. They are usually linked with five-membered lactam rings at the C-6 or C-8 positions of ring-A. Interestingly, a new chiral center is formed naturally based on the γ -position carbon of the lactam ring attached to the aromatic ring. These alkaloid flavan-3-ols often appear in the form of a couple of epimers with extremely similar NMR data. In addition, an ethyl group was usually connected to the nitrogen atom and galloyl group as a common moiety link to the C-3 position. Compound **44**, also named kopsirachin, featured two skytanthine moieties linked to the C-6 and C-8 positions, respectively. Compound **45** is characterized by a monoterpene indole unit connected at the C-8 position, and compound **46** features an acylamino indole unit connected at the C-4 position. They share a common biosynthetic precursor, tryptophan. Compound **47** is a unique dimer in alkaloid flavan-3-ols, having an *N*-ethyl-2-pyrrolidinone moiety linked to the C-8 position, and it also has a dimeric structure.

2.3. Oligomeric Flavan-3-ols

Some basic flavan-3-ols can be condensed to form oligomeric flavan-3-ols, usually found in cereals, legumes, fruits, vegetables, forages, and beverages. They play a crucial role in the generation of phenyl- γ -valerolactones and phenylvaleric acids, which feature appealing health benefits.

So far, several oligomeric flavan-3-ols, including dimers, trimers, tetramers, pentamers and hexamers, have been reported. Depending on the inter-flavanol linkages, six types of dimers can be distinguished. Compounds 48–56 are formed by condensation between the C-4 of the upper monomer and the C-8 or C-6 of the adjacent lower units. Additionally, in compounds 48–51, two pairs of isomers are observed, possessing an additional C-2– O-C-7 linkage between monomeric ones. Compounds 59 and 61 were determined to be dimeric flavan-3-ols through a methylene linkage between C-8 and C-8 or C-6 and C-6 of two basic units. Compound 60 is determined to be a dimeric flavan-3-ol glycoside. The dimeric flavan-3-ols 62–65 with the characteristic 5/7 or 7/5 fused ring benzotropolone skeleton system, namely theaflavins, were generated by a special pattern between ring-C of two basic flavan-3-ols. It has been reported that theaflavins heavily influence the astringency of black teas and therefore affect tea tasting evaluations. In theaflavins, the most common group linked at 3-OH is galloyl. Two types of trimers can be classified by inter flavanol linkage inter flavan-3-ols. Compounds 66-69 consist of three flavan-3-ol units linked by a C-4-C-8 bond between basic flavan-3-ols. In contrast, compounds 70-72 possess an additional C-2–O–C-7 linkage between two simple units. Compound 73 is the only tetrameric flavan-3-ol isolated from *Theobroma cacao*, in which four (–)-epicatechin units are linked by C-4–C-8 bonds. Pentameric and hexameric flavan-3-ols are mainly linear oligomers formed via C-4-C-8 bonds and occasionally contain a branched unit via a C-4–C-6 bond between flavanols.

3. Flavan-4-ols

As the name implies, flavan-4-ols have a hydroxyl group at the C-4 position of ring C (in Table 2, Figure 4). Only a few of these compounds, such as 77 and 78, exist in the form of aglycones. Other flavan-4-ols **79–100** with the same 6,8-dimethyl features usually possess glucose or glucuronic acid moieties attached to C-5 or C-7 of ring-A to form phenolic *O*-glycosides. Notably, some 5-*O*-glucosides or 7-*O*-glucosides of flavan-4-ols can generate seven-membered oxygen heterocycles by dehydration between the 2-OH of glucose and an aglycone hydroxyl.

No.	Name	Plant source	Plant Organ	Ref.
77	(2R,4S)-7-hydroxy-flavan-4-ol	Morus alba	leaves	[49]
78	(2R,4S)-2',4'-dihydroxy-2H-furan-(3'',4'':8,7)- flavan-4-ol	Morus alba	leaves	[49]
79	(2 <i>S,</i> 4 <i>R</i>)-7,4'-dihydroxy-6,8-dimethyl-4,2''- oxidoflavan-5- <i>O</i> -β-D-glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
80	eruberin A	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
81	(2 <i>S</i> ,4 <i>R</i>)-6,8-dimethyl-4'-methoxy-4,2''- oxidoflavan-5,7-di- <i>O</i> -β-D-glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
82	abacopterin G	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
83	(2 <i>S,4R</i>)-7,4'-dihydroxy-6-methoxymethyl-8- methyl-4,2''-oxidoflavan-5-О-β-D- glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
84	(2S,4R)-5,7-dihydroxy-4,4'-dimethoxy-6,8- dimethylflavan-5-Ο-β-D-6- acetylglucopyranoside-7-Ο-β-D- glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
85	(2 <i>S</i> ,4 <i>R</i>)-4,5,7-trihydroxy-4′-methoxy-6,8- dimethylflavan-5- <i>O</i> -β-D-6-glucopyranoside-7- <i>O</i> - β-D-glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
86	6''-O-acetyltriphyllin A	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[51]
87	eruberin B	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50,52]
88	4'-hydroxypneumatopterin B	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[51]
89	(2 <i>S,</i> 4 <i>R</i>)-4,5,7-trihydroxy-4′-methoxy-6,8- dimethylflavan-5-Ο-β-D-6- acetylglucopyranoside-7-Ο-β-D-glucopyranosid	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[50]
90	triphyllin A	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[53]
91	eruberin C	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[52]
92	abacopterin D	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[53]
93	abacopterin K	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[54]
94	(2 <i>R,4S</i>)-6,8-dimethyl-7-hydroxy-4'-methoxy- 4,2''-oxidoflavan-5-Ο-β-D-glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[55]

 Table 2. The numbers, plant sources and corresponding organs of flavan-4-ols.

No.	Name	Plant source	Plant Organ	Ref.
95	(2R,4S)-6,8-dimethyl-7-hydroxy-4'-methoxy- 4,2''-oxidoflavan-5-Ο-β-D-6''-O-acetyl- glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[56]
96	(2 <i>S</i> ,4 <i>R</i>)-7-hydroxy-6-methoxymethyl-8-methyl- 4,2''-oxidoflavan-5- <i>O</i> -β-D-glucopyranoside	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[56]
97	abacopterin E	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[57]
98	abacopterin C	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[56]
99	abacopterin A	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[56]
100	abacopterin B	Abacopteris penangiana or Pronephrium penangiana	rhizomes	[56]

Table 2. Cont.



 $\begin{array}{l} \textbf{79} \ R_1 = H, \ R_2 = H, \ R_3 = H \\ \textbf{80} \ R_1 = H, \ R_2 = H, \ R_3 = CH_3 \\ \textbf{81} \ R_1 = H, \ R_2 = Glu, \ R_3 = CH_3 \\ \textbf{82} \ R_1 = OCH_3, \ R_2 = H, \ R_3 = H \\ \textbf{83} \ R_1 = OCH_3, \ R_2 = H, \ R_3 = CH_3 \end{array}$

QН

OHOH





 $\begin{array}{l} \textbf{84} \ R_1 = GluA, \ R_2 = H, \ R_3 = H, \ R_4 = CH_3 \\ \textbf{85} \ R_1 = GluA, \ R_2 = H, \ R_3 = R_4 = CH_3 \\ \textbf{86} \ R_1 = GluA, \ R_2 = OH, \ R_3 = H, \ R_4 = CH_3 \\ \textbf{87} \ R_1 = Glu, \ R_2 = H, \ R_3 = H, \ R_4 = CH_3 \\ \textbf{88} \ R_1 = Glu, \ R_2 = H, \ R_3 = R_4 = H \\ \textbf{89} \ R_1 = Glu, \ R_2 = H, \ R_3 = R_4 = CH_3 \\ \textbf{90} \ R_1 = Glu, \ R_2 = OH, \ R_3 = R_4 = H \\ \textbf{91} \ R_1 = Glu, \ R_2 = OH, \ R_3 = H, \ R_4 = CH_3 \\ \end{array}$











98 $R_1 = H, R_2 = H, R_3 = CH_3$ 99 $R_1 = H, R_2 = Ac, R_3 = CH_3$ 100 $R_1 = Ac, R_2 = H, R_3 = CH_3$

Figure 4. Structures of flavan-4-ols.

Until now, a total of 24 flavan-4-ols have been described in the literature, most of them were isolated from *Pronephrium penangianum*, which is a traditional national herb medicine of the Tujia nationality in Hunan Province of China and has a good pharmacological activity on relaxing muscles and tend on stop romoting blood circulation, stanching bleeding and relieving pain in the Tujia system of medicine. Because of its red chicken blood color, the Tujia people name it "jixueqi". With the further investigation of "jixueqi", botanists found that its Chinese name is "pizhenxinyuejue".

4. Isoflavan-4-ols

Isoflavan-4-ols (compounds **101–109**) are a small group of flavanols isolated from the *Astragalus* genus (in Table 3, Figure 5), which is the largest genus in the Leguminosae family. Inconsistent with flavan-4-ols, the ring-B of isoflavan-4-ols is directly linked to C-3, and the 4-OH does not exist in the form of free hydroxyl. Consequently, a furan ring is easily formed between C-3, C-4 of ring-C and C-1', C-2' of ring B.

Table 3. The numbers, plant sources, and corresponding organs of isoflavan-4-ols.

No.	Name	Plant Source	Plant Organ	Ref.
101	maackiain	Astragalus cicer; Astragalus membranaceus; Astragalus mongholicus; Astragalus trojanus	roots roots	[58,59]
102	(6aR,11aR)-10-hydroxy-3,9,10-dimeth oxypterocarpan	Astragalus membranaceus;	roots	[60,61]
103	(6aR,11aR)-3,9,10-trimethoxypterocarpan	Astragalus membranaceus; Astragalus mongholicus	roots roots	[61]
104	vesticarpan	Astragalus membranaceus	roots	[62]
105	(–)-methylinissolin	Astragalus membranaceus	roots	[62]
106	(−)-methylinissolin-3- <i>O</i> -β-D-{6'-[2''-(<i>E</i>)- ene-penyl]}-glucoside	Astragalus membranaceus	roots	[62]
107	(—)-methylinissolin-3-O-β-D-(6′-acetyl)- glucoside	Astragalus membranaceus	roots	[62]
108	(−)-methylinissolin-3-O-β-D-glucoside	Astragalus membranaceus	roots	[62]
109	licoagroside D	Astragalus membranaceus	roots	[62]



Figure 5. Structures of isoflavan-4-ols.

5. Flavan-3,4-ols

Flavan-3,4-ols are classified into simple flavan-3,4-ols and dimeric flavan-3,4-ols based on aggregation (in Table 4, Figure 6).

No.	Name	Plant Source	Plant Organ	Ref.
110	guibourtacacidin	Acacia farnesiana	heartwoods	[63]
111	mollisacacidin	Acacia farnesiana	heartwoods	[63]
112	leucopelargonidin	Saraca asoca	barks	[25]
113	leucocyanidin	Saraca asoca	barks	[25]
114	teracacidin	Acacia galpinii	heartwoods	[63]
115	melacacidin	<i>Acacia nilotica</i> subsp. indica	heartwoods	[63]
116	(+)-2,3-trans-3,4-cis-peltogynol	Acacia melanoxylon	barks	[64] [65]
117	(–)-2,3-cis-3,4-cis-peltogynol	Acacia peuce; Acacia carneorum; Acacia crombiei; Acacia fasciculifera	barks barks barks barks	[64] [65]
118	[4-O-4]-bis-(2,3- <i>cis</i> -3,4- <i>trans</i> -3,3',4',7,8)- pentahydroxyflavan	Acacia peuce; Acacia carneorum; Acacia crombiei; Acacia fasciculifer	barks barks barks barks barks	[66]
119	2,3-cis-3,4-trans-3,3',4',7,8- pentahydroxflavan-[4-O-4]-2,3-cis-3,4-cis- 3,3',4',7,8-pentahydroxyflavan	Acacia peuce; Acacia carneorum; Acacia crombiei; Acacia fasciculifera	barks barks barks barks barks	[66]
120	[3,4,3',4]-O,O-linked-bis-(2,3- <i>trans</i> -3,4- <i>cis</i> - 3',4',7-trihydroxyflavan)	Acacia peuce; Acacia carneorum; Acacia crombiei; Acacia fasciculifera	barks barks barks barks barks	[67]
121	2,3-trans-3,4-trans-2',3'-trans-3',4'-cis- diastereoisomer	Acacia peuce; Acacia carneorum; Acacia crombiei; Acacia fasciculifera	barks barks barks barks barks	[67]

Table 4. The numbers, 1	plant sources and	corresponding of	organs of flavan-3,4-ols.
			0



Figure 6. Structures of flavan-3,4-ols.

5.1. Simple Flavan-3,4-ols

Only a small percentage of flavan-3,4-ols have been found in Nature. They present a hydroxyl group or a glycoside on position C-3 and C-4 and generate three chiral centers at C-2, C-3 and C-4. Based on the chiral centers, four compounds (110–113) are distinguished by their absolute configuration. However, compounds **114** and **115** have a six-membered oxygen heterocyclic ring linking C-2, C-3, C-1' and C-2'.

5.2. Oligomeric Flavan-3,4-ols

Oligomeric flavan-3,4-ols consists of several simple units and have additional chiral centers at C-3 and C-4 of all terminal monomer. Compounds **118–121** differ from other flavonoids by the presence of a fourth ring formed by a one-carbon bridge between an oxy functional group at C-3 of a flavonoid and a carbon of the flavonoid B ring. All the oligomeric flavan-3,4-ols have been isolated from *Acacia peuce*, *A. carneorum* and *A. crombiei*.

6. Pharmacological Activities

The pharmacological and biological properties of flavanols have been extensively studied to reveal their anti-oxidation, anti-inflammatory, anti-cancer, anti-viral effects, and protective cardiovascular properties, among others.

6.1. Anti-Oxidation

Flavanols are unanimously considered a natural source of antioxidant compounds. Flavan-3-ols exert their antioxidant effects via three main mechanisms. Flavan-3-ols have been shown three main mechanisms of antioxidants: (i) free radical scavenging, (ii) chelation of transition metals, (iii) inhibition of enzymes [68]. Free radicals are highly reactive species that contain an unpaired electron in the valence shell. The activity is a redox transition involving the donation of a single electron (or H-atom, equivalent to donation of an electron and an H^+) to a free radical species(\mathbb{R}^{\bullet}). The scavenging of the hydroxyl radical (OH $^{\bullet}$), superoxide (O₂ $^{\bullet-}$), and H₂O₂ is a common mechanism for anti-oxidation. DPPH (1,1-diphenyl-2-picrylhydrazyl) and ABTS [2,2'-amino-di(2-ethyl-benzothiazoline sulphonic acid-6)ammonium salt] are the most frequently encountered stable free radicals. Their antioxidative power and activity are measured by their ability to deactivate free radicals [69]. Free state iron and copper are usually combined with free radicals through Fenton and Haber-Weiss reactions. Flavan-3-ols bind to such divalent transition metals and effectively reduce the concentration of these cations, attenuating the extent of their oxidative activity [6]. Enzyme inhibition is also a vital mechanism to flavan-3-ols, especially for oligomeric flavan-3-ols [70–72]. They exhibit antioxidant activity through inhibition of prooxidative enzymes and lipoxygenase. (-)-epicatechin played a significant role in cellular redox regulation. It has been found to modulate the expression of enzymes producing superoxide anion, hydrogen peroxide, and nitric oxide [73], together with NADPH oxidases [74,75], xanthine oxidase [75], cyclooxygenase [76], and nitric oxide synthases (NOS). Furthermore, (-)-epicatechin interacts with cell membranes [77–79], reducing the lipid oxidation rate and regulating membrane-bound enzyme/receptor activity [80]. It also improves mitochondrial function and structure, decreasing the formation of superoxide anion and hydrogen peroxide. Moreover, it inhibits endoplasmic reticulum (ER) stress [81], leading to increased oxidant production. In addition, (–)-epicatechin upregulates antioxidant defense mechanisms via Nrf2 (NF-E2-related factor 2) activation [82–84]. Epicatechin, epigallocatechin, epicatechin gallate, and epigallocatechin gallate demonstrate excellent free radical scavenging activity. (+)-Catechin, (-)-catechin, (-)-afzelechin and (-)-epicatechin showed potent antioxidant activity, measured by the concentration of the extract required for 50% reduction in DPPH radical absorbance (13.5, 13.6, 21.8 and 20.9 μ M, respectively). In contrast, the positive control Trolox required 48.8 µM [20]. Compounds 18, 50, 51, 57, 58, 66, 70, 71, 73 exhibited anti-oxidant activities attributed to their inhibitory effects on nicotinamide adenine dinucleotide phosphate dependent lipid peroxidation in microsomes and on the autoxidation of linoleic acid. These effects are related to the

radical-scavenging activity in the peroxidation chain reactions [46]. Compounds **18**, **50**, **51**, **57**, **58**, **66**, **70**, **71**, **73** have inhibitory effects of on the autoxidation of linoleic acid initiated by the addition of V-70 with an IC₅₀ value of $0.62-5.3 \ \mu\text{g} / \text{mL}$. These compounds have radical-scavenging effects on the DPPH radical with an EC₅₀ value of $1.4-3.9 \ \mu\text{M}$. Compound **18** demonstrated dose-dependent scavenging potential and have particularly advantageous capacity against peroxynitrite (3.37–13.26 mmol AA/g), hydroxyl radical (5.03–8.91 mmol AA/g) and superoxide radical (3.50–5.50 mmol AA/g) (ascorbic acid (AA) was used as positive control.) [85]. Many flavan-4-ol glycosides also showed remarkable antioxidant and free radical scavenging properties. D-Gal is a normal, reducing sugar in vivo which can be oxidized into free radicals when present in high levels, causing oxidative stress and cellular damage. Abacopterin E, one of the flavan-4-ol isolated from *Pronephrium penangiana*, can scavenge free radicals, modulate the activities of antioxidant enzymes and reduce the lipid peroxidation in the hippocampus of D-Gal-treated mice [57].

6.2. Anti-Inflammation

To date, several experiments have demonstrated the anti-inflammatory activity of flavanols. NO levels, nuclear factor kappa-B (NF- κ B) and tumor necrosis factor-alpha (TGF- α) are well-established markers for the inflammatory process [86]. After infection, NO and several interleukins are generated by macrophages to combat the infectious agent. NF- κ B regulates the expression of the cytokines, including interleukin-1 beta (IL-1 β) and TNF- α , which are essential mediators of chronic inflammation and are implicated in leukocytosis, hyperplasia, and tissue break down [12]. The nuclear transcription factor NF-KB signaling pathway is the main feature of the inflammatory responses in oxidative-dependent diabetic aortic pathology. It is a major proinflammatory switch that can regulate the efflux of cytokines, such as monocyte chemoattractant protein (MCP)-1 and vascular endothelial growth factor (VEGF). (+)-Catechin inhibits the release of NO by two pathways: direct elimination of NO activity and inhibition of nitric oxide synthase (NO-iNOS) protein expression [87]. In the studies performed by Bui HuuTaia and Trinh Nam Trung, (+)-catechins' anti-inflammation activities were also reported [12]. (+)-Catechin and (–)-epicatechin have inhibitory effects on TNF- α -induced NF- κ B activation in HepG2 cells with IC₅₀ values of 14.1, 16.5 μ M, respectively. The study also showed that (+)-catechin, (+)-afzelechin, (-)-epiafzelechin and dulcisflavan inhibit the pro-inflammatory inducible nitric oxide synthase and cyclooxygenase-2 proteins in TNF- α -stimulated HepG2 cells at concentrations as low as 0.1 μ M. Catechin-5-O- β -D-glucoside showed anti-inflammatory activity with an inhibition rate of over $33.0\% \pm 4.0$ [23]. Theaflavins exhibit anti-inflammation effects by down-regulating the activation of NF-KB in macrophages. Theaflavin-3,3'-digallate, which contains two gallic acid moieties, exhibited the strongest anti-inflammatory activity as judged by its suppression on inducible NO synthase induction. Theaflavin, which featuring no gallic acid moiety, exhibited the least inhibitory effects [88]. From these results, we conclude that the gallic acid moiety is crucial for the anti-inflammatory activity of theaflavins. The same mechanism is observed in epigallocatechin gallate. It has been reported that (+)-catechin, (-)-epicatechin and its dimeric fraction inhibit NF- κ B by reducing the production of IL-2 in T cells during pro-inflammatory stimulation [89,90]. (–)-Gallocatechin-3-gallate, (–)gallocatechin and gallic acid inhibit lipopolysaccharide (LPS)-activated induction of nitric oxide synthase (NOS) in mouse peritoneal macrophages [91], because their gallocatechin and hydroxyl groups have anti-inflammatory effects by inhibiting NF-KB activation through NOS transcription [88]. Furthermore, 6-(2-pyrrolidinone-5-yl)-(–)-epicatechin and 8-(2pyrrolidinone-5-yl)-(-)-epicatechin have inhibitory activities against AGEs formation, with IC_{50} values of 13.5, 17.9 μ g / mL. A few flavan-4-ols also demonstrate significant antiinflammation activities. Abacopterin A exerts its anti-inflammatory properties by inhibiting NF- κ B expression and reducing the inflammatory response. These findings represent a potential agent of APA (Abacopterin A) for the treatment of inflammation [92].

6.3. Anti-Cancer

Cancer is one of the most serious diseases in the world and is closely related to improper regulation of apoptosis. Apoptosis is an active genetically determined process that automatically ends cell life, and is often called programmed cell death. A damaged or blocked apoptotic pathway results in uncontrolled cell division eventually leading to tumor formation [6]. Epigallocatechin-3-gallate exhibited strong effects on the induction of apoptosis and regulation of cell cycle in human and mouse carcinoma cells. Epigallocatechin-3-gallate exhibits strong apoptotic induction effects and regulates the cell cycle in human and mouse carcinoma cells. Epigallocatechin-3-gallate, epigallocatechin-3-gallate, epigallocatechin, and epicatechin-3-gallate, also led to the formation of inter nucleosomal DNA fragments and resulted in apoptosis in A431 cells and SMMC-7721 cells [93,94]. (2S,3S)-3,5,7,3',5'-pentahydroxyflavane has significant antiproliferative effects against HL-60 and CNE-1 with IC₅₀ of 0.2 \pm 0.01 and 1.0 \pm 0.1 μ M [16]. 12-O-tetradenoylphorbol 13-acetate (TPA) is a potent tumor promoter. Oligomeric flavanols can inhibit TPA-induced ornithine decarboxylase (ODC) activity to against cancer. Furthermore, this inhibition increases with the degree of oligomeric flavan-3-ols (trimer > dimer > monomer) [95]. Abacopterin C, abacopterin D, eruberin C and triphyllin A have exhibited a strong anticancer activity in vitro with a more than 60% inhibition rate of HepG2 at a dose of 50 μ M. Based on the strong anti-cancer properties of abacopterin C, our team showed a new drug-delivery system using a compound derived from Pronephrium penangianum (abacopterin C) for the treatment of cervical cancer. The system provides a potential synergistic platform for cancer therapy. An enhanced silver nanocluster system for cytochrome c detection and natural drug screening was also established. The release of Cyt c from apoptotic tumor cells was monitored by the novel fluorescence-enhanced nanoprobe, DNA-AgNCs@tween 80, confirming its effectiveness. Abacopterin C is a potential drug for treating anti-colon tumors [96].

6.4. Anti-Viral

Epigallocatechin-3-gallate and theaflavins (theaflavin, theaflavin-3-gallate, theaflavin 3'-gallate, and theaflavin-3,3'-digallate) exhibit antiviral properties against various viruses. Epigallocatechin-3-gallate effectively inhibits viral infections such as PRRSV (porcine reproductive and respiratory syndrome virus) [97], hepatitis C virus (HCV) [98], (west nile virus) WNV, (human immunodeficiency virus) HIV [99] and so on. Theaflavins showed inhibitory effects against HCV [100], TMV (Tobacco Mosaic Virus) [101] and other viral [102] infections.

COVID-19 is a viral disease that affects the epithelial cells of the respiratory system and causes inflammation of the mucosal membrane. Its outbreak poses a threat to the people's lives all over the world. COVID-19 is an infectious disease caused by a novel positive-sense single-27 stranded RNA corona virus called SARS-CoV-2. 3CL pro, a vital enzyme found in SARS-CoV-2, is a druggable target owing to its important function in viral cell replication. (–)-Epicatechin-3-O-gallate and theaflavins inhibit 3CL pro at lesser concentrations and also exhibit good binding to this target [103]. In molecular docking, Epigallocatechin-3-gallate and theaflavin-3-gallate showed that the tea polyphenols had good docking scores with the possible targets of COVID-19 [104]. (–)-Epicatechin-3-O-gallate and theaflavins for the treatment of COVID-19.

6.5. Cardiovascular Protection

Cardiovascular and cerebrovascular diseases are major life-threatening conditions, and lipid-lowering therapy has been widely used to reduce the incidence of these diseases. ACE activity is tightly linked to cardiovascular and cerebrovascular diseases. The inhibitory activity of angiotensin I-converting enzyme (ACE) was examined with metabolites of compound **21** produced by intestinal bacteria, together with tea catechins [105]. Oligomeric flavan-3-ols (dimer and hexamer) and epigallocatechin significantly inhibited enzyme activity and ACE activity [106]. Another molecular cardioprotective mechanism of

oligomeric flavan-3-ols is LDL oxidation inhibition (low-density lipoprotein). LDL in the blood may accumulate within the arterial wall and is prone to oxidation [107]. Flavan-3-ols can reduce cholesterol levels, restore lipid balance, and prevent the conversion of LDL to harmful oxidized chemical states [108].

6.6. Vascular Protection

Flavan-3-ols have therapeutic benefits in neurodegenerative conditions such as Alzheimer's and Parkinson's diseases. Alzheimers disease (AD) is a type of dementia, is one of the most common neurodegenerative diseases [109]. It is mostly characterized by extensive protein aggregation with amyloid-beta deposition in senile plaques and tau aggregation in neurofibrillary tangles. The ubiquitin-proteasome system (UPS) and theautophagy-lysosomal pathway are the two major catabolic pathways in eukaryotic cells. The UPS controls the degradation of cytosolic and nuclear proteins, including short-lived proteins, whereas autophagy is responsible for the clearance of protein aggregates and damaged organelles. Due to an age-dependent decline in their activity, aberrant proteins accumulate, contributing to the onset and development of age-related disorders, such as AD. Phenyl-γ-valerolactonesare microbiota-derived metabolites of flavan-3-ols; some of them activate the autophagic pathway in the SH-SY5Y control cell line. According to the cell line, we can find the change of LC3II (the lipidated form of LC3 that localizes in autophagosomal membranes) and p62 (a substrate of autophagy that accumulates in cells when autophagy is inhibited) [110]. SH-SY5Y control cells showed increased levels of LC3II but almost no change in the expression of the p62 protein, suggesting the activation of the early steps of the autophagic pathway in this cell line [111].

6.7. Miscellaneous Activities

Flavanols have also been associated with other beneficial health effects by acting as antimicrobials, anti-virals, anti-parasitics, anti-AChE and anti-alkaline phosphatase. Chunget al. summarized the anti-microbial effects of several tannin extracts on yeast, filamentous fungi, bacteria, and viruses [112]. (–)-8-*N*-ethyl-2-pyrrolidinone-epiafzelechin-3-*O*-gallate showed inhibitory activity against AChE with IC₅₀ values of 14.23 μ M. (–)-epicatechin-3-*O*- β -D-allopyranoside has alkaline phosphatase (ALP) activity by increasing mineralization up 120.8%, and 134.6% at 80 and 100 μ M.

6.8. Metabolism and Pharmacokinetics

Phenyl-γ-valerolactones (PVLs) and their related phenylvaleric acids (PVAs) are the main metabolites of flavan-3-ols. In 1958, PVLs were first reported to be flavan-3-ol metabolites by Oshima and Watanabe [113]. As more PVLs and PVAs were discovered, their activities in humans have attracted attention. 5-(3', 5'-dihydroxyphenyl)-γ-valerolactone, the main ring-fission metabolite of compound **21**, showed slightly higher blood-brain barrier (BBB) permeability than **21** and had neuritogenic activity in SH-SY5Y cells at 0.05 μ M [114]. In addition, metabolites of **21** degraded by rat intestinal flora have been confirmed to have antioxidative activity [115]. 5-(3',4'-dihydroxyphenyl)-γ-valerolactone is the major microbial metabolite of proanthocyanidins. By down regulating TNF-α-stimulated expressions of the two biomarkers of atherosclerosis vascular cell adhesion molecule-1 and monocyte chemotactic protein-1, the metabolite prevents THP-1 monocyte-endothelial cell adhesion, activation of nuclear factor kappa B transcription and phosphorylation of I kappa-B kinase and IκBα [116]. The radical scavenging abilities of the metabolites were found to be stronger than or nearly equal to those of Trolox (positive medicine) [114].

7. Conclusions

The review summarized 121 flavanols among flavan-3-ols, flavan-4-ols, isoflavan-4-ols and flavan-3,4-ols. Few systematic studies have reported flavanols among four types, which highly impact biological activities. The Theaceae plants *Camellia sinensis* and the Thelypteridaceae plant *Pronephrium penangianum* possess the most abundant flavanols.

Pronephrium penangianum harbors flavan-4-ols as its major constituents. *Camellia sinensis* var. *assamica*, another Theaceae plant, is the main source of alkaloid flavan-3-ols. Most of them showed strong inhibitory activity on α -glucosidase and acetyl-cholinesterase. *Astragalus* root is a very old and well-known drug in traditional Chinese medicine. The *Astragalus* genus has been reported to be rich in flavanols as one of its characteristic constituents. Another notable species is *Acacia sensustricto*. It is a dominant species in most Australian landscapes, with reported uses in the traditional medicine of Australian Indigenous people, and in traditional medicine systems of the Pacific Islands and Asia. Five plants consisting of *Acacia melanoxylon, Acacia peuce, Acacia carneorum, Acacia crombiei, Acacia fasciculifera* contain flavanols. These findings may be of important chemotaxonomic significance and provide a foundation to find more flavanols.

Most flavanols exhibit anti-oxidative, anti-inflammatory, anti-cancer, anti-viral activities, and have vascular protective effects on the cardiovascular system. Catechins and their derivatives generally exhibit strong antioxidant activity based on their structures, bioactivities, and pharmacokinetics. The most promising flavanols for medicinal development include the anti-cancer agents abacopterin A and abacopterin C. With the rapid development of nanomaterials chemistry, there are abundant types of nanoformulations, such as nanoflabers, nanoliposomes, and micelles, that can be used to improve flavanols targeting and druggability. (–)-epicatechin-3-*O*-gallate and theaflavins were previously reviewed for their antiviral activities, they are potential antiviral agents which should be explored as treatment and prophylactic alternatives for COVID-19. The potential of the flavanols for the treatment of cardiovascular, cerebrovascular diseases, Alzheimer's and Parkinson's diseases are also noticeable. Furthermore, the data of the genome of all kinds of flavanols, in parallel with the use of transcriptomic, proteomic, and metabolomic technologies, will aid in the elucidation of the genes involved in pharmacology activities.

However, the study on flavanols are far from enough. On the one hand, the content and purity of the active compounds isolated by traditional methods are usually low and subject to varying environmental factors, making it difficult to meet the huge demand of the international market. This highlights the need to develop new strategies for the preparation of these characteristic compounds. On the other hand, the biggest obstacle to large-scale production of flavanols is the lack of knowledge regarding the biosynthetic pathways. Meanwhile, a deeper work of isolation and biosynthetic pathways should be explored, and more pharmacological activities studies are necessary to define the active constituents of flavanols. The discovery of new compounds and transcriptomic, proteomic, and metabolomic technologies of flavanols are important and promising, such researches can provide clues for the development of new drugs. In this paper, the current research status of flavanols was reviewed in order to provide some scientific theoretical basis and reference for the follow-up research of flavanols. So far, few results have been published in this regard. It is expected that in the next few years more research will be carried out on this topic.

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