



Article Four New Pterosins from *Pteris cretica* and Their Cytotoxic Activities

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Abstract: Phytochemical investigation of the aerial parts of *Pteris cretica* led to the isolation and elucidation of nine pterosins, including four new pterosins, creticolacton A (1), 13-hydroxy-2(R),3(R)-pterosin L (2), creticoside A (3), and spelosin 3-O- β -D-glucopyranoside (4), together with five known pterosins 5–9. Their structures were identified mainly on the basis of 1D and 2D NMR spectral data, ESI-MS and literature comparisons. Compounds 1 and 3 were new type of petrosins with a six membered ring between C-14 and C-15. The new compounds were tested in vitro for their cytotoxic activities against four human tumor cell lines (SH-SY5Y, SGC-7901, HCT-116, Lovo). Results showed that compounds 1 and 2 exhibited cytotoxic activity against HCT-116 cells with IC₅₀ value of 22.4 μ M and 15.8 μ M, respectively.

Keywords: Pteris cretica Linn.; pterosins; cytotoxic activity

1. Introduction

The genus Pteris (Pteridaceae) comprises about 300 species, which are mainly distributed over the tropical and temperate zones throughout the world. There are about 66 species of the genus in China, especially in South and Southwest of China [1]. Pteris is commonly known as "Jue-Cai" in Chinese. Most species of this genus such as Pteris multifida, Pteris cretica, and Pteris nervosa are used as traditional Chinese medicines to clear heat, remove dampness and cool blood in medical practice [2]. In recent years, the phytochemical investigation of this genus has resulted in the isolation and identification of sesquiterpenoids (known as pterosins), diterpenes, flavonoids and various quinic acids [3–5]. Among of above secondary metabolites, pterosins, illudane-type sesquiterpenoids are most significant characteristic constituents of Pteris. Moreover, pterosins are used as the chemical markers for Pteridaceae [6]. Pterosins can be classified into three categories according to the number of skeletal carbon, such as 13 carbon, 14 carbon and 15 carbon pterosin derivatives [6,7]. Pharmacological investigations showed that pterosins had many bioactivities, such as antitumor, anti-inflammatory, anti-diabetes and anti-tuberculosis properties [8–11]. However, most studies on pterosins from this genus focused on Pteris multifida [7]. Other species of the Pteris genera probably possess pharmacological activities and bioactive secondary metabolites that are similar to *Pteris multifida*. As a part of our ongoing research aimed at finding potentially bioactive components from medicinal plants of the genus Pteris [12–14], we have carried out a detailed phytochemical investigation of the aerial parts of Pteris cretica Linn. (P. cretica). Herein described are the isolation and structural elucidation of nine pterosins, including four pterosins (compounds 1, 2, 5 and 6), along with five pterosides (compounds 3, 4 and 7-9).

2. Results and Discussion

The 70% ethanol extract of *P. cretica* was extracted with petroleum ether, dichloromethane, EtOAc and *n*-BuOH, separately. The *n*-BuOH extraction was repeatedly subjected to thin-layer, normal-phase column and reversed phase semi-preparative column chromatography to afford four new pterosins **1–4**, together with five known pterosins, namely pterosin D (5) [15], (3*R*)-pterosin W (6) [8], (3*R*)-pterosin D 3-*O*- β -D-glucopyranoside (7) [16], 3(*R*)-pteroside W (8) [17], 2*R*,3*R*-pterosin L 3-*O*- β -D-glucopyranoside (9) [18]. The structures of compounds **1–9** are illustrated in Figure 1.



Figure 1. Structures of compounds 1-9.

Compound 1 was isolated as colorless amorphous powder (MeOH) and showed a molecular formula of $C_{15}H_{16}O_4$ with eight degrees of unsaturation, as determined by a HRESI-MS peak at m/z261.1127 $[M + H]^+$ (calcd for $C_{15}H_{17}O_4$, 261.1127) (See Supplementary Materials). The UV spectrum exhibited the characteristic absorptions of pterosin-type sesquiterpenes at 215 (4.03), 258 (3.71), 310 (2.97) nm [15]. In the ¹H-NMR spectrum (Table 1), three singlet methyl signals, including two geminal dimethyl groups located at C-2 ($\delta_{\rm H}$ 1.10 and 1.25, each 3H) and one aromatic methyl group ($\delta_{\rm H}$ 2.49, 3H), a carbinyl proton at C-3 ($\delta_{\rm H}$ 4.80, s), an aromatic proton ($\delta_{\rm H}$ 7.80, s), and two methine groups $[\delta_{\rm H} 3.06 (2H, m, H-13) \text{ and } 4.55 (1H, m, H-14a), 4.49 (1H, m, H-14b)]$ were observed. The ¹³C-NMR displayed 15 carbon resonances (Table 1), including three methyls at $\delta_{\rm C}$ 20.0 (C-10), $\delta_{\rm C}$ 20.3 (C-12) and δc 22.6 (C-11), two methylenes at δc 27.0 (C-13) and δc 67.9 (C-14), two methines at δc 77.1 (C-3) and δc 132.4 (C-4), and eight quaternary carbons. Comparison of the NMR data of 1 with those of pterosin D (5) [15], which was isolated at the same time, revealed that they shared a similar structure. The differences between them were that one ester carbonyl group (δc 163.5) replacing the methyl group (C-15) of pterosin D was observed in 1. The proton and carbon chemical shifts of C-14 appeared at the low field region due to the deshielding influence of the oxygenatom suggested 1 possessed a lactone link between positions C-14 and C-15. This was further confirmed by the HMBC spectrum (Figure 2) that showed the correlations from H-14 ($\delta_{\rm H}$ 4.49, 4.55) to C-15 ($\delta_{\rm C}$ 163.5). The absolute configuration of **1** was also determined as 3*R* by the CD spectrum which exhibited a positive cotton effect at 326 nm [19]. Based on these results, compound 1 was identified as creticolactone A, which represents the new type of petrosin possessing a six membered lactone ring between positions C-14 and C-15.

Compound **2** was isolated as colorless amorphous powder (MeOH) and showed a molecular formula of $C_{15}H_{20}O_5$ with six degrees of unsaturation as determined by the HRESI-MS peak at m/z 315.1001 [M + Cl]⁻ (calcd for $C_{15}H_{20}O_5$ Cl, 315.0999) (See Supplementary Materials). The UV spectrum exhibited the characteristic absorptions of pterosin-type sesquiterpenes at 222 (4.19), 265 (3.82), 305 (2.78) [15]. In the ¹H-NMR spectrum (Table 1), three singlet methyl signals, including one geminal

dimethyl group located at C-2 ($\delta_{\rm H}$ 1.15) and two aromatic methyl groups ($\delta_{\rm H}$ 2.58 and 2.76), a carbinyl proton at C-3 ($\delta_{\rm H}$ 4.84) and an aromatic proton ($\delta_{\rm H}$ 7.35) were observed.

Position	1		2	
	$\delta_{ m H}$ (J in Hz)	$\delta_{\rm C}$	$\delta_{ m H}$ (J in Hz)	δ _C
1	-	206.1	-	209.7
2	-	53.4	-	57.0
3	4.80 (s)	77.1	4.84 (s)	77.4
4	7.80 (s)	132.4	7.35 (s)	126.9
5	-	144.4	-	146.6
6	-	144.1	-	140.4
7	-	123.9	-	138.2
8	-	134.0	-	133.1
9	-	155.1	-	155.6
10	1.10 (s)	20.0	3.73 (d, <i>J</i> = 11.0) 3.68 (d, <i>J</i> = 11.0)	67.1
11	1.25 (s)	22.6	1.15 (s)	19.2
12	2.49 (s)	20.3	2.58 (s)	22.4
13	3.06 (m)	27.0	5.32 (dd, <i>J</i> = 4.9, 8.3)	72.7
14	4.49 (m) 4.55 (m)	67.9	3.93 (dd, <i>J</i> = 8.3, 11.4) 3.64 (dd, <i>J</i> = 4.9, 11.4)	65.6
15	-	163.8	2.76 (s)	15.2

Table 1. The NMR spectroscopic data of compounds **1** and **2** (MeOH-*d*₄, ¹H-NMR 600 MHz, ¹³C-NMR 150 MHz).

The ¹³C-NMR displayed 15 carbon resonances, including three methyls at δc 15.2 (C-15), δc 19.2 (C-11) and δc 22.4 (C-12), two oxygenated methylenes at δc 65.6 (C-14) and δc 67.1 (C-10), three methines at δc 72.7 (C-13), 77.4 (C-3) and δc 126.9 (C-4), and eight quaternary carbons. The ¹H-NMR and 13 C-NMR spectra of compound **2** showed similar features to those of 2(R),3(R)-pterosin L [15]. The difference between the ¹H-NMR of **2** and that of 2(R),3(R)-pterosin L was that 2(R),3(R)-pterosin L exhibited an A₂X₂ coupled system (2H × 2, t, I = 8.0 Hz), but 2 showed characteristic ABX signals [$\delta_{\rm H}$ 5.32 (1H, dd, J = 4.9, 8.3 Hz, H-13), 3.93 (1H, dd, J = 8.3, 11.4 Hz, H-14a), 3.64 (1H, dd, J = 4.9, 11.4 Hz, H-14b)]. This suggested a 1,2-glycol was located at C-6 of 2. This was also confirmed by the carbon chemical shifts of C-13 (δ_C 72.7) appearing in the low field region due to the deshielding influence of the oxygen atom. Furthermore, in HMBC spectrum (Figure 2), the methine proton at $\delta_{\rm H}$ 5.32 (H-13) showed cross-peaks with C-5, C-6 and C-7, which further indicated that the attached hydroxyl group could be assigned to C-13. The *cis*-configuration of the methyl at C-2 and the H at C-3 in **2** was confirmed by NOESY correlations between H-3 and H-11. The absolute configuration of 2 was determined by its CD spectrum, which exhibited a positive Cotton effect at 329 nm in MeOH, indicating that the oxygenated group at C-3 existed in a pseudoaxial conformation irrespective of the configuration at C-2 [14,19]. Based on these results, compound **2** was assigned as 13-hydroxyl-2(R), 3(R)-pterosin L.

Compound **3** was isolated as a colorless amorphous powder (MeOH) and showed a molecular formula of $C_{21}H_{28}O_8$ with eight degrees of unsaturation as determined by a HRESI-MS peak at m/z 431.1672 [M + Na]⁺ (calcd. for $C_{21}H_{28}O_8Na$, 431.1682) (See Supplementary Materials). The UV spectrum exhibited the characteristic absorptions of pterosin-type sesquiterpenes at 228 (4.28), 270 (3.99), 310 (3.27) nm [15]. The ¹H-NMR spectrum (Table 2) showed signals of one singlet methyl (δ_H 1.14, H-11), one aromatic methyl (δ_H 2.35, H-12), a methylene group [2.80 (d, *J* = 17.0 Hz, H-3a), 3.50 (d, *J* = 17.0 Hz, H-3b)], an aromatic proton [δ_H 7.22 (s, H-4)], an ethanol grouping located at C-6 [δ_H 2.74 (2H, t, *J* = 5.5 Hz, H-13), 3.98 (2H, m, H-14)], and two oxygenated methylene groups [δ_H 4.10 (1H, d, *J* = 9.4 Hz, H-10a), 3.48 (1H, d, *J* = 9.4 Hz, H-10b), 5.08 (2H, d, *J* = 4.8 Hz, H-15)]. Furthermore, the presence of a β -configuration glucose moiety was evident from the signals assignable to one anomeric proton (δ_H 4.20, d, *J* = 7.9 Hz), two oxygenated methine protons [δ_H 3.78 (dd, *J* = 1.6, 11.8 Hz)

and 3.61 (dd, J = 5.7, 11.8 Hz)], and four overlapping protons ($\delta_{\rm H}$ 3.03–3.28) [20]. This was further confirmed by GC-MS analysis after acid hydrolysis [21]. The connectivity of the glucose part was identified by HMBC correlations between the H-glc-1 ($\delta_{\rm H}$ 4.20) and the C-10 ($\delta_{\rm C}$ 74.86).



Figure 2. The key HMBC (\rightarrow) and NOESY (\leftrightarrow) of compounds **1–4**.

Position	3		4	
	$\delta_{ m H}$ (J in Hz)	$\delta_{\rm C}$	$\delta_{ m H}$ (J in Hz)	$\delta_{\rm C}$
1	-	211.9	-	211.2
2	-	51.2	-	52.8
3	2.80 (d, <i>J</i> = 17.0) 3.50 (d, <i>J</i> = 17.0)	38.5	4.84 (s)	86.0
4	7.22 (s)	126.9	7.52 (s)	127.8
5	-	146.4	-	146.2
6	-	132.5	-	140.7
7	-	136.2	-	139.1
8	-	130.1	-	131.4
9	-	153.9	-	152.7
10	4.10 (d, <i>J</i> = 9.4) 3.48 (d, <i>J</i> = 9.4)	74.86	1.08 (s)	22.3
11	1.14 (s)	21.8	1.28 (s)	22.8
12	2.35 (s)	19.9	2.57 (s)	22.0
13	2.74 (t, J = 5.5)	27.2	5.32 (dd, <i>J</i> = 4.9, 8.4)	72.7
14	3.98 (m)	65.7	3.92 (dd, <i>J</i> = 8.4, 11.5) 3.63 (dd, <i>J</i> = 4.9,11.5)	65.5
15	5.08 (d, J = 4.8)	67.5	2.74 (s)	15.2
glc-1	4.20 (d, <i>J</i> = 7.9)	104.7	4.57 (d, <i>J</i> = 7.8)	105.8
glc-2	3.21 (m)	71.5	3.36 (m)	71.7
glc-3	3.20 (m)	77.9	3.42 (m)	78.0
glc-4	3.03 (m)	74.9	3.28 (m)	75.3
glc-5	3.28 (m)	78.1	3.27 (m)	78.2
glc-6	3.78 (dd, <i>J</i> = 1.6, 11.8) 3.61 (dd, <i>J</i> = 5.7, 11.8)	62.8	3.85 (dd, <i>J</i> = 2.0, 12.0) 3.75 (dd, <i>J</i> = 5.4, 12.0)	62.8

Table 2. The NMR spectroscopic data of compounds **3** and **4** (MeOH- d_4 , ¹H-NMR 600 MHz, ¹³C-NMR 150 MHz).

Compound **3** exhibited ¹H- and ¹³C-NMR data closely resembling to those of rhedynoside A [22]. The difference between NMR spectroscopic data of **3** (Table 2) and those of rhedynoside A was **3** possesses an ether link. The presence of an ether link was assigned to be between positions C-15 and C-14 as their proton and carbon chemical shifts appeared at the low field region due to the deshielding influence of the oxygen atom. This was further confirmed by the HMBC spectrum that showed the correlations from H-15 ($\delta_{\rm H}$ 5.08) to C-14 ($\delta_{\rm C}$ 65.6) and from H-14 ($\delta_{\rm H}$ 3.98) to C-15 ($\delta_{\rm C}$ 67.5). The absolute configuration of **3** was also determined as 2*S* by the CD spectrum which exhibited vibronic *n*- π^* transition which concurs with C-2 having 2*S* configuration [22]. Based on these results, compound **3** was assigned as creticoside A.

Compound 4 was isolated as colorless amorphous powder (MeOH) and showed a molecular formula of $C_{21}H_{30}O_9$ with seven degrees of unsaturation as determined by a HRESI-MS peak at m/z $461.1581 \text{ [M + Cl]}^-$ (calcd for C₂₁H₃₀O₉Cl, 461.1578) and 471.1874 [M + COOH]^- (calcd for C₂₂H₃₁O₁₂, 471.1866) (See Supplementary Materials). The UV spectrum exhibited the characteristic absorptions of pterosin-type sesquiterpenes at 219 (4.16), 265 (3.80), 303 (2.88) nm [15]. In the ¹H-NMR spectrum (Table 2), four singlet methyl signals, including two geminal dimethyl groups located at C-2 ($\delta_{\rm H}$ 1.08 and 1.28, each 3H) and two aromatic methyl groups ($\delta_{\rm H}$ 2.57 and 2.74, each 3H), a carbinyl proton at C-3 ($\delta_{\rm H}$ 4.84, s), an aromatic proton ($\delta_{\rm H}$ 7.52, s), and a set of ABX signals [$\delta_{\rm H}$ 5.32 (dd, *J* = 4.9, 8.4 Hz), 3.92 (dd, J = 8.4, 11.5 Hz), 3.63 (dd, J = 4.9, 11.5 Hz)] assignable to a 1,2-glycol located at C-6 were observed. Compound 4 exhibited ¹H and ¹³C-NMR data closely resembling to those of spelosin [15]. Comparison of the UV and NMR spectroscopic data of 4 (Table 2) with those of spelosin showed several differences, the main one being the presence of an additional hexose sugar moiety having a signal for an anomeric-H at $\delta_{\rm H}$ 4.57 (d, J = 7.8 Hz) and remaining sugar proton signals at $\delta_{\rm H}$ 3.27–3.85. The ¹³C-NMR spectrum of 4 contained an anomeric carbon signal of a hexose moiety at $\delta_{\rm H}$ 105.8 and signals for remaining five sugar carbons at $\delta_{\rm H}$ 62.8–78.2, which were in good agreement with those reported for glucoside compounds [22]. Furthermore, the coupling constant (J = 7.8 Hz) is consistent with a *trans* ${}^{3}J_{\text{H-H}}$, showing that the glucose moiety is in a β -configuration [20]. This was further confirmed by GC-MS analysis after acid hydrolysis [21]. The connectivity of glucose part was identified by HMBC correlations between the H-1" ($\delta_{\rm H}$ 4.57) and the C-3 ($\delta_{\rm C}$ 86.0) (Figure 2). The absolute configuration of 4 was also determined as 3*R* by the CD spectrum which exhibited a positive Cotton effect at 329 nm [19]. Based on these results, compound 4 was assigned as spelosin 3-O- β -D-glucopyranoside.

Compounds 1–4 were evaluated for cytotoxic activities against four human cell lines (SH-SY5Y (neuroblastoma cell line), SGC-7901 (gastric cancer cell line), HCT-116 (colon cancer cell line) and Lovo (colorectal cancer cell line). Compounds 1–4 were inactive ($IC_{50} > 100 \mu M$) to SH-SY5Y, SGC-7901, and HCT-116 cell lines. Compounds 1 and 2 exhibited cytotoxic activity against HCT-116 cells with IC_{50} value of 22.4 μM and 15.8 μM , respectively.

3. Experimental Section

3.1. General Information

1D and 2D-NMR spectra were obtained on an AV-600 spectrometer (Bruker, Rheinstetten, Germany) with TMS as internal reference, and methanol- d_4 as solvent. Electrospray ionisation (ESI) mass spectra were acquired in the positive ion mode on a LCQ DECAXP instrument (Thermo Finnigan, San Jose, CA, USA) equipped with an ion trap mass analyzer. HRESI-MS were obtained in the positive ion mode on a Waters UPLC Premier Q-TOF system (Waters Inc., Milford, MA, USA). CD spectra were obtained on an Olis DSM 1000 spectrometer (Olis Inc., Augusta, GA, USA). Optical rotations were acquired on Shenguang SGW-1 digital polarimeter (Jingke, Shanghai, China). GC-MS were obtained on a Thermo Finnigan Trace DSQ (TR-5MS column: 60 m × 0.25 mm × 2.5 μ m) (ThermoFinnigan, Bremen, Germany). TLC plates were HSGF254 SiO₂ from Yantai Jiangyou Silica Gel Development Co., Ltd. (Yantai, China). Column chromatography (CC) silica gel (SiO₂; 200–300 mesh; Qingdao Haiyang Chemical Co., Ltd., Qingdao, China), Sephadex LH-20 (GE-Healthcare Bio-Sciences AB, Uppsala, Sweden), ODS (Grace

C18, Grace Davison Discovery Sciences, Columbia, MD, USA) were employed as packing materials, semi-preparative HPLC (Grace Prevail C18 column, 5 μ m, 10.0 mm I.D \times 250 mm, Grace Davison Discovery Sciences, Columbia, MD, USA). All other chemicals were of analytical reagent grade.

3.2. Plant Materials

The aerial parts of *P. cretica* were collected in Jiangxi China, and identified by Prof. Xiaomei Fu of Jiangxi University of Traditional Chinese Medicine. A voucher specimen (NO. 20161007) was deposited at the Key Laboratory of Modern Preparation of TCM, Jiangxi University of Traditional Chinese Medicine, China.

3.3. Extraction and Isolation

The aerial parts of *P. cretica* (5 kg) were extracted with 70% (v/v) aqueous ethanol (about 60 L). The 70% EtOH extract was concentrated under reduced pressure to give a residue (803 g), which was then extracted with petroleum ether, dichloromethane, EtOAc and *n*-BuOH, respectively. The above extracts were concentrated to yield 81 g of petroleum ether fraction, 108 g of dichloromethane fraction, 154 g of EtOAc fraction and 132 g of *n*-BuOH fraction. The UV analysis of the above fractions monitoring the characteristic absorption bands in the 304 to 205 nm region of pterosin derivatives [15,16], showed that the *n*-BuOH extract contained more pterosins in comparison with other extracts. Therefore, the *n*-BuOH extract was subjected to further purification. Part of the dried *n*-BuOH extract (about 130 g) was subjected to silica gel column chromatography with gradient mixtures of $CH_2Cl_2/MeOH$ (from 10:1 to 1:10). Each 200 mL fraction was collected and evaporated, and some fractions was combined after analytical TLC inspection (CHCl₃/MeOH 10:1 to 3:1). Finally, 33 fractions (Fr.1–33) were obtained. Fr.8 (1.7 g) was further subjected to silica gel CC (CH₂Cl₂/MeOH 10:1, 8:1, 4:1). Each 50 mL of eluent was collected and combined after analytical TLC inspection (CHCl₃/MeOH 8:1 to 5:1). 12 fractions (Fr.8.1–12) were obtained. Fr.8.4 (256 mg) was further subjected to Sephadex LH-20 column chromatography with MeOH. 15 fractions (Fr.8.4.1–15) were obtained. Fr.8.4.6 (57 mg) was further purified by semi-preparative HPLC (65% MeOH/H₂O, 3.0 mL/min) to yield 1 (7.2 mg, t_R18.2 min), 5 (9.7 mg, t_R14.5 min), 6 (11.1 mg, t_R10.5 min). Fr.12 (2.2 g) was further subjected to silica gel CC (CH₂Cl₂/MeOH 9:1, 7:1, 3:1). Twenty-three fractions (Fr.12.1–23) were obtained. Fr.12.9 (195 mg) was further subjected to Sephadex LH-20 column chromatography with MeOH. Eighteen fractions (Fr.12.9.1–18) were obtained. Fr.12.9.5 (24 mg) was further purified by semi-preparative HPLC (59% MeOH/H₂O, 3.0 mL/min) to yield 2 (6.8 mg, t_R 18.3 min). Fr.12.9.16 (21 mg) was further purified by semi-preparative HPLC (55% MeOH/H₂O, 3.0 mL/min) to yield **3** (8.1 mg, t_R15.6 min). Fr.18 (1.0 g) was further subjected to silica gel CC (CH₂Cl₂/MeOH 9:1, 7:1, 3:1). Twenty fractions (Fr.18.1–20) were obtained. Fr.18.6 (195 mg) was further subjected to Sephadex LH-20 column chromatography with MeOH. Eleven fractions (Fr.18.6.1–11) were obtained. Fr.18.6.5 (19 mg) was further purified by semi-preparative HPLC (50% MeOH/H₂O, 3.0 mL/min) to yield 4 (7.3 mg, t_R17.6 min). Fr.18.6.9 (25 mg) was further purified by semi-preparative HPLC (50% MeOH/H₂O, 3.0 mL/min) to yield 7 (10.0 mg, t_R15.5 min). Fr.22 (897 mg) was further subjected to Sephadex LH-20 column chromatography with MeOH. Sixteen fractions (Fr.22.1–16) were obtained. Fr.22.7 (68 mg) was further purified by semi-preparative HPLC (45% MeOH/H₂O, 3.0 mL/min) to yield 8 (14.3 mg, t_R15.3 min). Fr.22.11 (35 mg) was further purified by semi-preparative HPLC (45% MeOH/H₂O, 3.0 mL/min) to yield 9 (15.0 mg, t_R19.5 min).

3.4. Spectral Data

Creticolactone A (1): Colorless amorphous powder (MeOH). $[\alpha]_D^{25} = +51.4^\circ$ (c =0.032, MeOH). UV (MeOH) λ_{max} (log ϵ) 215 (4.03), 258 (3.71), 310 (2.97) nm; HRESI-MS at *m*/*z* 261.1127 [M + H]⁺ (calcd for C₁₅H₁₇O₄, 261.1127); ¹H-NMR and ¹³C-NMR see Table 1.

13-Hydroxy-2(R),3(R)-pterosin L (2): Colorless amorphous powder (MeOH). $[\alpha]_D^{25} = +30.0^{\circ}$ (c =0.032, MeOH). UV (MeOH) λ_{max} (log ε) 222 (4.19), 265 (3.82), 305 (2.78) nm; HRESI-MS at *m*/z 315.1001 [M + Cl]⁻ (calcd for C₁₅H₂₀O₅Cl, 315.0999); ¹H-NMR and ¹³C-NMR see Table 1.

Creticoside A (3): Colorless amorphous powder (MeOH). $[\alpha]_D^{25} = -32.0^\circ$ (c =0.032, MeOH). UV (MeOH) λ_{max} (log ϵ) 228 (4.28), 270 (3.99), 310 (3.27) nm; HRESI-MS at *m/z* 431.1672 [M + Na]⁺ (calcd for C₂₁H₂₈O₈Na, 431.1682); ¹H-NMR and ¹³C-NMR see Table 2.

Spelosin 3-*O*-β-*D*-*glucopyranoside* (4): Colorless amorphous powder (MeOH). $[\alpha]_D^{25} = -17.8^\circ$ (c =0.032, MeOH). UV (MeOH) λ_{max} (log ε) 219 (4.16), 265 (3.80), 303 (2.88) nm; HRESI-MS at *m*/*z* 461.1581 [M + Cl]⁻ (calcd for C₂₁H₃₀O₉Cl, 461.1578); ¹H-NMR and ¹³C-NMR see Table 2.

3.5. Cytotoxic Activity

The cytotoxicity of the isolated compounds (the purities of compounds **1–4** were 95.5%, 93.0%, 92.5% and 95.6%, respectively) against four human tumor cell lines (SH-SY5Y, SGC-7901, HCT-116, Lovo) was evaluated using the MTT assay performed according to the method described by Liu [23].

3.6. Acid Hydrolysis of 3-4

The procedure of the absolute configuration determination of glucose (compounds **3** and **4**) was as previously reported [21].

4. Conclusions

This study describes the successful isolation and identification of nine pterosin-type sesquiterpenoids from the aerial part of *P. cretica*, including four new pterosins (compounds 1–4). Their chemical structures were elucidated by 1D- and 2D-NMR spectroscopic analysis. creticolactone A (1) and creticoside A (3) belong to a new type of pterosin with a six membered ring between positions C-14 and C-15. To our best knowledge, this is the first report on the occurrence of this new class of pterosins from the genus *Pteris*, which implies that they might be chemotaxonomic markers for *P. cretica*. Furthermore, all of the new isolated compounds 1–4 were evaluated for cytotoxic activity. Compounds 1 and 2 exhibited cytotoxic activity against HCT-116 cells with IC_{50} values of 22.4 μ M and 15.8 μ M.

Supplementary Materials: The Supplementary Materials are available online.

Author Contributions: The contributions of the respective authors were as follows: J.S. designed the research. J.L. (Jian Lu) and S.C. were responsible for the isolation of the compounds. C.P. and Q.M. were responsible for the bioactivities investigation of new compounds. J.L. (Jianqun Liu) and J.S. were responsible for identification of the isolated compounds and wrote the whole article.

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Conflicts of Interest: The authors declare no conflict of interest.

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Sample Availability: Samples of the compounds are not available from the authors.



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