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The phase transition of Pb₈F₁₄I₂

Matthias Weil¹

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Abstract The reversible phase transition of $Pb_8F_{14}I_2$ is of continuous type and takes place at about 107 °C as monitored by temperature-dependent single crystal and powder X-ray diffraction measurements, optical microscopy, and differential scanning calorimetry. The low-temperature ferroelastic phase crystallizes in the orthorhombic crystal system (23 °C, *Bmmb*, Z = 2, a = 6.0699(6) Å, b = 6.0165(6) Å, c = 25.077(2) Å, 1487 structure factors, 41 parameter, $R(F^2) = 0.0346$, $wR(F^2) = 0.0771$) and changes its symmetry to the tetragonal crystal system into the high-temperature paraelastic phase (130 °C, I4/mmm, Z = 1, a = 4.2667(12) Å, c = 25.388(7) Å, 430 structure $R(F^2) = 0.0575,$ 303 parameter, factors, $wR(F^2) = 0.1564$). Group–subgroup relationships between the two structures and a hypothetical intermediate structure are presented.

Graphical abstract

$$\begin{array}{ccc} \text{LT-Pb}_8\text{F}_{14}\text{I}_2 & 107 \text{ °C} & \text{HT-Pb}_8\text{F}_{14}\text{I}_2 \\ \text{Bmmb}, Z = 2 & \overleftrightarrow & I4/mmm, Z = 1 \end{array}$$

Keywords Lead fluoride halide ·

Crystal structure determination · Thermal analysis · Group–subgroup relationships

Matthias Weil matthias.weil@tuwien.ac.at

Introduction

In the pseudo-binary system PbF_2/PbI_2 , the phases $(PbF_2)_7(PbI_2)$, $(PbF_2)_5(PbI_2)$, and $(PbF_2)_4(PbI_2)$ have been synthesized in the form of single crystals and their crystal structures reported by Aurivillius [1]. Synthesis and unit cell parameter of polycrystalline matlockite-type [2, 3] PbFI have been described by Rulmont [4] and Aurivillius [5]; the crystal structure of the latter phase was refined afterwards from single crystal data [6].

The crystal structure of $(PbF_2)_7(PbI_2)$ has originally been determined by Aurivillius on the basis of integrated room temperature Weissenberg data in the orthorhombic space group Bmmb (standard setting Cmcm, No. 63) with lattice parameters a = 6.0711(6) Å, b = 6.0198(5) Å, and c = 25.084(2) Å [1]. For that purpose single crystals were originally grown by boiling and recrystallizing a sample with nominal composition of 10PbF2·PbI2 in water. The author reported a systematic twinning of the very tiny and thin crystals in the way that the a- and b-axes are interchanged. Due to the method of data collection, the platy crystal form and clearly visible twinning of the investigated single crystal, a straightforward structure refinement was hampered at that time, and the reliability index of 0.109 based on |F| was rather high. However, all atomic positions could be derived, and for Pb and I atoms anisotropic temperature factors were considered in the final model. The nearly equal lengths of the a- and b-axes and an I-centred pseudo-tetragonal subcell make it appear likely that a phase transition from the room-temperature modification (lowtemperature (LT) phase) into a tetragonal high-temperature modification (high-temperature (HT) phase) occurs. Although the author of the original study has theoretically derived an idealized structure model in space group I4/ mmm with the unit cell relations: $a_{\text{ideal.}} \approx (a_{\text{LT}} + b_{\text{LT}})/$



¹ Division of Structural Chemistry, Institute for Chemical Technologies and Analytics, TU Wien, Getreidemarkt 9/164-SC, 1060 Vienna, Austria

 $\sqrt{8}$; $c_{ideal.} \approx c_{LT}$, neither experimental details regarding a possible phase transition nor structure data of the assumed high-temperature phase were reported at that time or afterwards. Therefore, a more detailed examination of a possible phase transition and a redetermination of the crystal structure of $(PbF_2)_7(PbI_2)$ (= $Pb_8F_{14}I_2$) seemed worthwhile.

In this article preparation, ferroelastic behaviour, phase transition and crystal structures of $Pb_8F_{14}I_2$ at room temperature (LT modification) and at 130 °C (HT modification) as well as their relationships are reported.

Results and discussion

Phase transition

Examination of Pb₈F₁₄I₂ (LT) single crystals under a microscope in transmitted polarized light revealed clearly visible multiple domains for most of the crystals, in accordance with the observations by Aurivillius [1]. The domain crystals show ferroelastic behaviour [7] and can be partially or completely reorientated by application of stress with a pair of tweezers along the diagonal or parallel to the a- or b-axes (in the setting of space group Bmmb). By heating single crystals in inert silicon oil on a Kofler heating stage, a spontaneous change from a biaxial to an uniaxial crystal system in terms of the change of the birefringence of crystal faces was observed at 107(2) °C, indicating a phase transition from the orthorhombic to the tetragonal crystal system. This process was reversible and showed virtually no hysteresis. The observed transition temperatures are in good agreement with a complementary DSC measurement, with on-set temperatures of 104.6 °C (heating; endothermal effect) and 104.0 °C (cooling; exothermal effect) (Fig. 1).

The temperature-dependence of the unit cell parameters for the LT- and HT-phases, as evidenced by X-ray diffraction measurements, is given in Fig. 2. The derived transition temperature of the diffraction experiments is somewhat higher (118(8) °C) for this kind of measurement than those from the optical examinations or DSC measurements. This behaviour is ascribed to the more inaccurate temperature measurement during the diffraction studies where the temperature sensing device was displaced a couple of millimetres from the actual sample. Given the different temperature sensing for the three methods, the derived transition temperature from the DSC experiment appears to be that with the highest accuracy.

The evolution of the orthorhombic unit cell parameters a and b (Fig. 2a) reveals a more obvious change of a towards the transition point than that of b. The c unit cell parameter appears not to be directly involved in the



Fig. 1 DSC curves obtained during heating (*red*) and cooling (*blue*) of $Pb_8F_{14}I_2$ (colour figure online)

transition mechanism (Fig. 2b), with a non-linear behaviour for the temperature-dependence of the LT-phase and a linear behaviour for the HT-phase. The resulting change of the cell volume with temperature is depicted in Fig. 2c. The ratio (a - b)/(a + b), where a and b are the lengths of the orthorhombic unit cell at a given temperature, is a suitable order parameter η for classification of the type (continuous, discontinuous) of the phase transition. The development of the temperature-dependence of $\eta = (a - b)/(a + b)$ is given in Fig. 2d and indicates a continuous phase transition from orthorhombic to tetragonal symmetry upon heating. This classification is supported by the continuous change of the unit cell volume with temperature close to the transition point and the observation of the very slight hysteresis observed during optical examination of crystals at the Kofler stage in polarized light or in the DSC curve (Fig. 1) during heating and cooling above and below the transition point.

Above the transition temperature $Pb_8F_{14}I_2$ is tetragonal, representing the paraelastic phase in space group *I4/mmm*; below the transition temperature it is orthorhombic, representing the ferroelastic phase in space group *Bmmb*. There is no direct group–subgroup relation between these two space groups [8, 9], because *Bmmb* (or in its standard setting *Cmcm*) is not a maximal subgroup but a general subgroup of *I4/mmm* with *Fmmm* being the intermediate space group type. The group–subgroup relationships of the paraelastic (HT) and ferroelastic (LT) structures of Pb_8 - $F_{14}I_2$, together with the hypothetical intermediate structure model in *Fmmm*, are presented in the form of a Bärnighausen family tree [10] in Fig. 3.

The symmetry reduction from *I*4/*mmm* to *Fmmm* is of *translationengleiche* (t) kind with an index of 2, accompanied by a loss of the fourfold rotation axes present in the







25.5

25.4

25.3

25.2

25.1

25.0

20

40

60

Fig. 2 Evolution of unit cell parameters of $Pb_8F_{14}I_2$ with temperature. The crystal has tetragonal symmetry at $T > T_C$ and orthorhombic symmetry at $T < T_C$, with a > b; single crystal diffraction data from Aurivillius (*grey squares*) [1], current single crystal diffraction data (*black squares*), powder X-ray diffraction data diffraction (*black circles*); the size of the symbols is greater that the standard deviation

of individual values. **a** *a* and *b* axes (the identical unit cell parameters *a* and *b* in the tetragonal HT-phase can be extrapolated into the temperature interval of the orthorhombic LT-phase, *dashed line*); **b** *c* axis; **c** volume; **d** plot of the temperature-dependent course of the order parameter η of Pb₈F₁₄I₂

higher-symmetry structure. The relationship between the *Fmmm* structure and *Bmmb* is of *klassengleiche* (k) kind with an index of 2 under change of the Bravais centering from *F* to *B*. The first *translationengleiche* transition of index 2 from *I*4/*mmm* \rightarrow *Fmmm* is a ferroic transition [11] and explains the presence of two twin domains in the resulting crystals of the LT-Pb₈F₁₄I₂ polymorph.

Structure description: LT-phase

The current structure refinement confirms the original model [1]; however, with increased precision and accuracy. The crystal structure is made up of four-layer "PbF₂"-type blocks

consisting of four unique Pb²⁺ cations. The blocks are stacked along [001] with a mean Pb–F distance of 2.63 Å which is in good agreement with the sum of the ionic radii [12] for Pb²⁺ and F⁻ (2.60 Å; weighted according to the different coordination numbers as discussed below). The "PbF₂"-type blocks are separated by interstitial I⁻ ions that are located at $z \approx \frac{1}{4}$ and $\frac{3}{4}$ (Fig. 4, left) with a mean Pb–I distance of 3.70 Å. The "PbF₂"-type blocks contain additional ordered F⁻ ions (F1) with one close contact of 2.261(10) Å to Pb2. The latter defines one of the boundaries of the blocks and is additionally bonded to the interstitial I⁻ ions. Due to the additional F1 ions, the coordination numbers (CN) of the central Pb²⁺ cations in each block are increased in comparison with fluorite-type PbF₂ (CN = 8), to Fig. 3 Group–subgroup relations for the HT- and LTforms of $Pb_8F_{14}I_2$ and the hypothetical intermediate phase in *Fmmm*. Coordinates of HTand LT- $Pb_8F_{14}I_2$ refer to the refined coordinates (deposited as CIFs, see *Experimental*) but here due to truncation with somewhat lower precision



HT-phase





 Table 1
 Details of single

 crystal X-ray data collections
 and structure refinements

Phase	$Pb_8F_{14}I_2$ (LT)	$Pb_8F_{14}I_2$ (HT)				
Temperature/°C	23	130				
$M_{ m r}$	2177.32					
Crystal size/mm ³	$0.015 \times 0.100 \times 0.100$					
Crystal colour, form	Colourless, plate					
Crystal system	Orthorhombic	Tetragonal				
Space group, no.	<i>Bmmb</i> , 63	I4/mmm, 139				
a/Å	6.0699(6)	4.2667(12)				
b/Å	6.0165(6)					
c/Å	25.077(2)	25.388(7)				
V/Å ³	915.80(16)	462.2(3)				
Ζ	2	1				
$D_{\rm calcd}/{\rm g}~{\rm cm}^{-3}$	7.896	7.823				
$\mu(MoK_{\alpha})/mm^{-1}$	76.864	76.027				
Theta range/°	1.624–38.733	3.210-37.513				
hkl range						
h	-10 to 10	-7 to 7				
k	-10 to 10	-7 to 7				
l	-44 to 43	-42 to 43				
Reflections measured	34113	15499				
Reflections unique	1487	430				
Reflections $[I > 2\sigma(I)]$	956	303				
R _{int}	0.1396	0.1091				
Extinction coefficient	0.00023(2)	0.0008(3)				
Twin ratio	0.507:0.493					
Parameters refined	41	19				
$R(F^2)/wR(F^2) \ [I > 2\sigma(I)]$	0.0346/0.0689	0.0575/0.1420				
$R(F^2)/wR(F^2)$ [all reflections]	0.0695/0.0771	0.0864/0.1564				
GooF (F^2)	0.895	1.189				
$\Delta ho_{ m fin}$ (max/min)/e ⁻ Å ⁻³	5.106/-5.602	4.401/-9.657				
CSD number	431846	431847				
			-			

phase, the mean Pb–F distance in the "PbF₂"-type blocks slightly increases to 2.70 Å, whereas the mean Pb–I distance of the LT-phase and that of the HT-phase are more or less the same. Comparison of the individual Pb–F and Pb–I bond lengths reveals a shift of individual atoms up to 0.25 Å for the lighter F and 0.15 Å for heavy I atoms.

Experimental

Preparation

Single crystals of $Pb_8F_{14}I_2$ were grown under hydrothermal conditions from stoichiometric amounts of PbI_2 (Riedel-de Haën, pure) and PbF_2 (Aldrich, 99 + %). The starting materials were homogenized by grinding and placed in a 5-cm³ Teflon[®] container that was filled up to two-thirds of its volume with demineralized water. The Teflon[®] container was then closed and placed in a steel autoclave and subjected to the following heating protocol: $25 \rightarrow 250$ °C [2 h], 250 °C [10 days], $250 \rightarrow 25$ °C [10 h]. Nearly colourless laminated crystals of Pb₈F₁₄I₂ and few very thin crystals with a plate-like form and light-yellow colour of PbFI [6] were obtained. Microcrystalline samples of Pb₈F₁₄I₂ were prepared by reaction of stoichiometric amounts of PbF₂ and PbI₂ in sealed and evacuated silica ampoules at 380 °C for 1 week.

DSC measurements

DSC measurements were performed with ~20 mg samples on a NETZSCH DSC 200F3 system in the temperature range 30–150 °C (aluminium crucibles with pierced lid, flowing argon atmosphere (20 cm³/min), heating rate 5 °C min⁻¹).

Table 2 Selected bond lengths/Å

Pb ₈ F ₁₄ I ₂ (LT)			$Pb_8F_{14}I_2$ (HT)		
Pb1–F2	2.492(5)	$4 \times$	Pb1–F2	2.5404(9)	4×
Pb1–F3	2.693(6)	$4 \times$	Pb1–F3	2.762(11)	$4 \times$
Pb1–F1	3.0400(7)	$2\times$	Pb1–F1	3.00(6)	
Pb2–F1	2.261(10)		Pb1–F1	3.027(5)	$4 \times$
Pb2–F3	2.434(5)	$4 \times$	Pb2–F3	2.390(8)	$4 \times$
Pb2–I1	3.8415(9)	$2 \times$	Pb2–F1	2.59(6)	
Pb2–I1	3.8594(9)	$2\times$	Pb2–I1	3.6967(14)	$4 \times$
Pb3–F2	2.579(6)	$4 \times$			
Pb3–F3	2.789(6)	$4 \times$			
Pb3–F1	2.884(10)				
Pb3–F1	3.0298(13)	$2\times$			
Pb4–F3	2.381(5)	$4 \times$			
Pb4–I1	3.5504(8)	$2 \times$			
Pb4–I1	3.5756(8)	$2 \times$			

X-ray diffraction and single crystal structure analysis

Polycrystalline samples were characterized by X-ray powder diffraction with $CuK_{\alpha 1,2}$ radiation ($\lambda = 1.54060$, 1.54439 Å). At room temperature a Philips X'pert system was used. Temperature-dependent measurements were performed on a Philips PW1012/10 diffractometer equipped with a home-built heat controlling system. Unit cell parameters were refined with the program TOPAS [13].

Single crystal X-ray data were collected on a Bruker APEX II CCD diffractometer with MoK_{α} radiation ($\lambda = 0.71079$ Å). For single crystal X-ray measurements at room temperature, optically controlled single domain crystals of Pb₈F₁₄I₂ (RT, 23 °C) were fixed with cyanoacrylate adhesive on thin silica glass fibres. For hightemperature measurement of Pb₈F₁₄I₂ (HT, 130 °C), a more temperature-resistant two-component adhesive was used. Heating was provided with a gas-flow heater using a stream of nitrogen. Diffraction data for the two measurements were corrected for absorption effects with the program HABITUS [14]. Full-matrix least-squares refinements on F^2 for the two data sets and correction of extinction effects were carried out with the program SHELXL-2014 [15].

For the LT-phase the coordinates of the reported structure model [1] were taken as starting parameters for refinement in the space group *Bmmb*. The measured crystal was twinned by pseudo-merohedry with a rotation by 90° along [001] as twin element (refined twin ratio of the twin domains 1:1). For the HT-phase the proposed coordinates of the idealized structure model [1] in the space group *I*4/ *mmm* were used. For the two models anisotropic displacement parameters for all atoms were refined. Details of data collections and structure refinements are listed in Table 1. Selected bond lengths are collated in Table 2. Further details of the crystal structure investigations may be obtained from Fachinformationszentrum Karlsruhe, 76344 Eggenstein-Leopoldshafen, Germany (fax: (+49)7247-808-666; e-mail: crysdata@fiz-karlsruhe.de, https://www.fiz-karlsruhe.de/icsd.htm) on quoting the appropriate CSD number listed at the end of Table 1. Drawings of structural details were produced using the program ATOMS [16].

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References

- 1. Aurivillius B (1980) Chem Scripta 15:153
- 2. Nieuwenkamp W, Bijvoet JM (1932) Z Kristallogr 81:469
- 3. Pasero M, Perchiazzi N (1996) Mineral Mag 60:833
- 4. Rulmont A (1973) CR Acad Sci Ser C 276:775
- 5. Aurivillius B (1976) Chem Scripta 10:206
- 6. Weil M, Kubel F (2001) Acta Crystallogr E57:i80
- Salje EKH (1990) Phase transitions in ferroelastic and co-elastic crystals. Cambridge University Press, Cambridge
- Ivantchev S, Kroumova E, Madariaga G, Perez-Mato JM, Aroyo MI (2000) J Appl Crystallogr 33:1190
- 9. Müller U (2013) Symmetry relationships between crystal structures. Oxford University Press, Oxford

- 14. Herrendorf W (1997) HABITUS. University of Gießen, Germany
- 15. Sheldrick GM (2015) Acta Crystallogr C 71:3
- 16. Dowty E (2008) ATOMS for Windows. Shape Software, Kingsport
- Bärnighausen H (1980) MATCH 9:139
 Wadhawan VK (2006) Phase Transitions 64:165
- 12. Shannon RD (1976) Acta Crystallogr A 32:751
- 13. Bruker-AXS (2012) TOPAS 4.2. Bruker-AXS, Karlsruhe