

Catalytic Stereoconvergent Synthesis of Homochiral β -CF₃, β -SCF₃, and β -OCF₃ Benzylic Alcohols

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and in particular, needle-shaped crystals of representative stereopure products that exhibit either elastic or plastic flexibility, which opens the door to functional materials based on mechanically responsive chiral molecular crystals.

KEYWORDS: adaptive crystals, asymmetric catalysis, density functional calculations, drug design, fluorine, hydrogenation, kinetic resolution, ruthenium

INTRODUCTION

Fluorine atoms profoundly influence the properties of bioactive molecules on multiple levels, which results in half of blockbuster drugs and one-third of newly approved drugs being fluoro-pharmaceuticals.¹⁻⁶ Other fast-growing market segments are those of fluorinated materials for use in the electronics industry and in energy storage^{7,8} and of fluorinecontaining agrochemicals.⁹ Organofluorine chemistry is essentially man-made, as only a dozen fluorinated natural products have been identified on Earth.¹⁰ The consideration of new fluorinated chemotypes for advanced applications therefore inevitably follows the availability of the synthetic methods to access the relevant moieties. Outstanding progress was achieved in the preparation of a plethora of synthetic fluorine compounds.¹¹ A less developed area that is highly challenging but very rewarding is the asymmetric synthesis of stereogenic fluorinated molecules.¹²⁻¹⁵ In this context, we embarked on the asymmetric construction of chiral carbon atoms featuring a fluorinated motif with emphasis on the trifluoromethyl group C*-CF₃ and its heteroatomic homologues C*-SCF₃ and C^*-OCF_3 .

In particular, β -CF₃-substituted alcohols and amines are emerging structural motifs in medicinal chemistry (Figure 1A). For example, compound I, prepared as a mixture of stereoisomers, exhibits antibacterial activity,¹⁶ and racemic

compound **II** is an inhibitor of WD repeat-containing protein 5, which is overexpressed in some types of cancer.¹⁷ Stereochemically defined trifluoromethylated omarigliptin exhibits better pharmacokinetic and pharmacodynamic profiles compared to the parent drug molecule and is clinically evaluated as a super-long-acting antidiabetic.¹⁸

Surprisingly, however, there are no preceding literature reports on the asymmetric synthesis of the model 2-CF₃-1indanol **2a**, its amino analogue, or their higher homologues. Nonasymmetric approaches toward such cyclic benzo-fused β trifluoromethyl alcohols or amines have received significant attention in the past decade and are based on photoredox, electrochemical, or transition-metal-catalyzed oxy-trifluoromethylation^{19–26} or amino-trifluoromethylation^{27–32} of the corresponding olefins. There are only a handful of literature reports on the synthesis of the homochiral β -trifluoromethyl secondary alcohol motif (Figure 1B). The synthetic strategies are based on a two-step arrangement of the contiguous

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A) Bioactive β -CF₃ alcohols and amines:



B) Synthetic strategies towards homochiral β -CF₃ secondary alcohols:

Diastereoselective addition of vinyllithium, then asymmetric hydrogenation; ref. 33.



Asymmetric cycloaddition, then diastereoselective reduction; ref. 34.

$$Ar \xrightarrow{\text{NCbz}} + \begin{pmatrix} CF_3 \\ r.t. \\ OBoc \end{pmatrix} \xrightarrow{\text{Pd-cat.}} \\ \begin{array}{c} r.t. \\ dr > 98:2 \\ 99\% ee \end{pmatrix} \xrightarrow{\text{CF}_3} \begin{pmatrix} 1) O_3, -78 \ ^\circ\text{C} \\ 2) \ \text{NaBH}_4, -78 \ ^\circ\text{C} \\ dr = 90:10 \end{pmatrix} \xrightarrow{\text{CF}_3} \\ \begin{array}{c} \text{CF}_3 \\ \text{CbzN} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \begin{array}{c} \text{CbzN} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \begin{array}{c} \text{CbzN} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \begin{array}{c} \text{CbzN} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \begin{array}{c} \text{CbzN} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \begin{array}{c} \text{CbzN} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \begin{array}{c} \text{CbzN} \\ \end{array} \xrightarrow{\text{CF}_3} \\ \end{array}$$

One-step diastereoselective aldol reaction using a chiral auxiliary; ref. 35.



One-step catalytic enantioselective asymmetric transfer hydrogenation; this work:

0 0	Ru-cat. HCO ₂ H/Et ₂ N	он	 stereoconvergent mild conditions
	40 °C		 side reactions minimized
R L _ Phn =	dr >99:1, >99% ee	R	 extended to -SCF₃ and -OCF med-chem application meterials asiance application

Figure 1. (A) Bioactive compounds with β -CF₃ alcohol or amine motifs. (B) Synthetic strategies toward homochiral β -CF₃ secondary alcohols.

stereocenters employing diastereoselective addition of nucleophilic vinyllithium followed by substrate-controlled olefin hydrogenation,³³ or asymmetric cycloaddition followed by NaBH₄ reduction of the ketone intermediate.³⁴ A single-step approach via diastereoselective aldol or Reformatsky reactions using a chiral auxiliary has been described,^{35,36} but to the best of our knowledge no single-step catalytic enantioselective access to this class of molecules has ever been reported.

Dynamic kinetic resolution based on Noyori–Ikariya transfer hydrogenation (DKR-ATH) seemed like a fitting synthetic strategy for addressing the challenging simultaneous control of both chiral centers of the target compound class.^{37–41} DKR-ATH is a robust method for stereoconvergent access to enantiomerically pure secondary alcohols with multiple contiguous chiral centers starting from the readily available racemic α -substituted ketones,^{42–50} including fluorinated examples.^{51–58} This approach to β -CF₃ alcohols would involve in situ epimerization of α -CF₃ ketones via an enol or enolate-anion intermediate. Specifically, α -CF₃ enolates have been associated with decomposition due to fluoride elimination to furnish the corresponding unstable difluoroenone.^{59–61} This was foreseen as the major obstacle toward an efficient DKR-ATH-based catalytic asymmetric synthesis of β -CF₃ alcohols.

RESULTS AND DISCUSSION

A model racemic ketone 2-CF₃-1-indanone 1a was prepared in one step by triflic acid mediated annulation of benzene with 2-

 CF_3 -acrylic acid.⁶² It was subjected to DKR-ATH using a commonly used formic acid/triethylamine 3:2 mixture as a source of hydrogen and chlorobenzene as a cosolvent, and five representative Noyori–Ikariya type Ru(II) catalysts were tested (Table 1, runs 1–5). C1 is the archetypical Noyori

Table 1. Catalyst and Solvent Screening for Ru(II)-Catalyzed DKR-ATH of $1a^a$

(0 CF ₃ 1a) cat. : 100) //Et ₃ N . 40 °C	OH <i>cis-2a</i> >99:1 <i>cis/trans</i> >99% ee	O Ja	+ 0H 4a
	Ru(II) cat.	F/A	Cosolvent	Time	1a:2a:3a:4a
1	(<i>R</i> , <i>R</i>)-C1	3:2	PhCl	3 h	0:73:19:8
2	(<i>S</i> , <i>S</i>)-C2	3:2	PhCl	3 h	0:75:6:19
3	(<i>S</i> , <i>S</i>)-C3	3:2	PhCl	3 h	0:75:19:6
4	(<i>S</i> , <i>S</i>)-C4	3:2	PhCl	3 h	4:55:35:6
5	(3R,1'S)-C5	3:2	PhCl	3 h	0:75:0:25
6 (S	(s s) C c	3.2	-	3 h	5:24:71:0
	(3,3)-02	3.2		18 h	0:25:60:15
7 (\$\$)-C2	(S.S)-C2	5.2	-	3 h	21:74:5:0
1	(0,0) 02	0.2		18 h	5:75:20:0
8	(<i>S</i> , <i>S</i>)-C2	5:2	PhCl	3 h	0:99:0:1
9	(<i>S</i> , <i>S</i>)-C2	5:2	DMF	3 h	0:97:0:3
10	(<i>S</i> , <i>S</i>)-C2	5:2	dioxane	3 h	0:98:0:2
11	(<i>S</i> , <i>S</i>)-C2	5:2	1,2-DCE	3 h	0:98:0:2
Ts Ph -	Ru-Ru NH2 Ph (<i>R</i> , <i>R</i>)- C 1 (<i>S</i> , <i>S</i>)- C 1	Ts Ph 2 (S	Ph Ph S,S)-C3 (S,S)-C3	N N SO ₂ Ph S)-C4	O2 NH Ph (3 <i>R</i> ,1'S)- C5

^{*a*}DKR-ATH of **1a** (50 mg, 0.25 mmol) was carried out using a Ru(II) catalyst (1 mol %), HCO₂H/Et₃N (F/A) (0.25 mL) and cosolvent (0.5 mL) at 40 °C. The product ratio was determined by NMR analysis of reaction mixture aliquots, and the ratio of **2a** stereoisomers (*cis/trans* > 99:1; > 99% ee in all cases) was determined after isolation by ¹⁹F NMR and HPLC analysis using the chiral stationary phase. PhCl = chlorobenzene; DMF = *N*,*N*-dimethylformamide; dioxane = 1,4-dioxane; 1,2-DCE = 1,2-dichloroethane.

catalyst,^{63,64} and the rest are the so-called tethered catalysts, which proved to be superior for the reduction of structurally complex ketones.⁶⁵ Chronologically, **C2** was developed by Wills et al.,⁶⁶ followed by oxy-tethered catalyst **C3** by Ikariya et al.,⁶⁷ sulfamoyl-DPEN-cored **C4**,⁶⁸ and benzosultam-cored **C5** by Mohar and co-workers.^{69–71} The reactions using 1 mol % of catalysts **C1–C5** all reached >95% conversion within 3 h (Table 1, entries 1–5). Delightfully, all the catalysts yielded the product **2a** with excellent stereoselectivity⁷² (*cis/trans* > 99:1 and > 99% ee) as determined by ¹⁹F NMR and chiral HPLC, respectively. The absolute configuration of **2a** as (*S*,*S*) was determined by single-crystal X-ray diffraction (SCXRD) analysis of a product from the run with (*S*,*S*)-**C2** (Table 1, entry 2).

Disappointedly, significant amounts of side products, indanone 3a and/or indanol 4a (up to 41% total), were also detected in the reaction mixtures, indicating that detrifluoromethylation indeed took place during DKR-ATH. The catalysts performed differently regarding side product formation, and C2 was chosen for further studies because of its wide availability and favorable reaction kinetics (Table S1). Control experiments indicated that the trifluoromethyl moiety is eliminated from the ketone 1a rather than the product *cis*-2a via a non-ruthenium-catalyzed process involving the formation of the Et₃N/HF adduct (see the Supporting Information (SI)). To mitigate fluoride elimination, the use of HCO₂H/Et₃N in a 5:2 molar ratio with the most efficient (S,S)-C2 was attempted. Performing the DKR-ATH in neat HCO₂H/Et₃N 3:2 or 5:2 (Table 1, entries 6 and 7) revealed that, by increasing the relative amount of formic acid, the extent of detrifluoromethylation dramatically decreases while excellent stereoselectivities are still obtained. Further solvent screening revealed that the use of any cosolvent together with HCO₂H/Et₃N 5:2 was beneficial for the reaction yield as less than 3% of the side products were observed in chlorobenzene, DMF, 1,4-dioxane, or 1,2-dichloroethane (Table 1, entries 8-11). The first one was deemed optimal with only 1 mol % 1-indanol accompanying the target product 2a.

Computational modeling was further performed to corroborate the high level of stereoselectivities and realize the possible mechanism of **1a** racemization being the core process of DKR. The reaction between **1a** and the active form of precatalyst (*S*,*S*)-**C2** was studied using the M06-2X-D3/ SMD(chlorobenzene)/def2-qzvp//def2-svp method. Four diastereomeric transition states are possible (Figure 2).



Figure 2. Optimized transition state geometries en route to the four stereomeric products **2a** taking place with $R_{Ru\nu}\lambda$ structural arrangement of the (*S*,*S*)-**C2** catalyst active form (see text). The relative free energies are given in kcal·mol⁻¹. Some attractive and repulsive interactions are highlighted by green and red symbols, respectively. Noncritical H atoms are omitted for clarity.

For the $R_{Ru}\lambda$ -catalyst structural arrangement,^{73,74} observed in the solid-state of (S,S)-C2,⁶⁶ computations predict the ratio of the reaction rates leading to each stereoisomer as ~10⁹ (S,S):1800 (R,R):400 (S,R):1 (R,S).⁷⁵ This transforms into a *cis/trans* ratio of 2.5 × 10⁶ and enantioselectivity of 99.9996% for the *cis* product.⁷⁶ The discrepancy between experimentally and theoretically predicted % ee is likely due to the additional mechanisms of the generation of chirality.⁷⁷ However, the calculation reproduces and points to a high level of stereodiscrimination. Two spatial regions of the catalyst simultaneously control the final stereoselectivity: the region of the tethered η^6 -arene ligand and the region of the SO₂ moiety.^{71,77} Dynamic equilibrium and interplay of attraction and repulsion in each region through various noncovalent interactions lead to stabilization/destabilization of the corresponding stereoselectivity-determining transition states. The presence of the α -CF₃ functionality is crucial for exceptionally high stereoselectivity. As a comparison, DKR-ATH of 2-methyl-1-indanone using C3 yielded the corresponding alcohol with a lower *cis*-selectivity (*cis/trans* = 98:2, 98% ee),⁴⁵ whereas DKR-ATH of 2-acetamido-1-indanone (hydrogen bond donor α -substituent) using C5 was *trans*-selective (*cis/trans* = 9:91).⁷¹

A 3:2 mixture of HCO₂H/Et₃N is a typical choice for DKR with Noyori-Ikariya catalysts,⁷⁸ whereas a 5:2 mixture is usually used for ATH of simple ketones and imines.⁷⁹⁻⁸² Although generally not explained, an Et₃N or Et₃N/HCO₂H mixture might serve as a catalyst for the DKR-enabling rapid in situ racemization of the α -substituted ketones, consistent with the 3:2 choice.⁸³ Indeed, computations point that direct noncatalyzed epimerization of 1a is energetically prohibitive (Figure S1, top). On the contrary, 1a racemization catalyzed by Et₃N ("enolate-anion" pathway) and the concerted Et₃N/ HCO₂H process ("enol" pathway) are energetically plausible with the preference to the former by 4.3 kcal·mol⁻¹ (Figure S1, middle and bottom). Increasing the relative concentration of formic acid, associated with decreased fluoride elimination, pushes the major racemization pathway toward the concerted Et_3N/HCO_2H process.

With optimal conditions in hand, we turned our attention to DKR-ATH of various α -trifluoromethyl substituted benzofused cyclic ketones 1b-1m (Table 2). These were prepared as described for 1a,⁶² via radical desulfur-fragmentation followed by reconstruction of enol triflates⁸⁴ and radical trifluoromethylation of the corresponding olefins⁸⁵ and enol acetates,⁸⁶ respectively (see the SI). The ketones 1a-1m were all converted to the corresponding stereopure alcohols 2a-2m using the optimized reaction conditions (1 mol % of C2 in HCO₂H/Et₃N 5:2 and chlorobenzene at 40 °C) with reaction times to reach full conversion being between 1 and 6 h. Their (S,S)-absolute configuration was assigned based on SCXRD analysis of indan-cored 2a, 2d, and 2f and tetralin-cored 2k. The values of cis/trans ratio and ee in Table 2 are given as ">99", but the other three possible stereoisomers were in fact present below the limit of detection for most cases,⁸⁷ and the ee of the benzosuberol 2m was determined to be 99.2%. The tetramethyl substituted indanone 1c required a higher catalyst loading (5 mol %) to reach full conversion. The reaction yield was affected by competing detrifluoromethylaton which was generally more expressed during DKR-ATH of indan-cored ketones compared to their six-membered analogs. Nevertheless, the decomposition products 3a-3h and 4a-4h were readily removable by flash chromatography. 7-Acetamido analog 2h was formed in only 37% NMR yield with fast decomposition coupled to fast reduction in HCO₂H/Et₃N 3:2, which still outperformed the 5:2 ratio with 25% NMR yield and full conversion achieved only after 18 h. The tetralin derivatives 1i-1k were devoid of detrifluoromethylation, and the corresponding stereopure products 2i-2k were isolated directly after the extraction.

The method was then extended to the synthesis of stereopure 2-SCF₃ and 2-OCF₃ carbinols 2n-2p, where no side reactions were observed in either HCO₂H/Et₃N ratio tested. Stereoselectivities for both trifluoromethylthioethers 2n and 2o were determined to be *cis/trans* = 99.9:0.1 and 99.8% ee by ¹⁹F NMR and chiral GC, respectively, which gives an

Table 2. Scope of the DKR-ATH a



^{*a*}Unless otherwise specified, the reactions were carried out using (S_5S) -C2 (1 mol %) in HCO₂H/Et₃N 5:2 and chlorobenzene at 40 °C. ^{*b*}NMR yield based on integration of 2 relative to 1,3 and 4. Isolated yields after extraction and optional column chromatography were 1–15% lower. ^{*c*}5 mol % of (S_5S) -C2 used. ^{*d*}HCO₂H/Et₃N 3:2 used.

estimate of the detection limit. The starting 2-SCF₃ ketones 1n and 10 were prepared by means of Billard's reagent under acidic conditions from the corresponding bare ketones.⁸⁸ 2-Trifluoromethoxy-1-indanol 2p was obtained with somewhat lower stereopurity (*cis/trans* = 99:1, 96% ee) with the same sense of enantioselectivity (SCXRD analysis); its ketone precursor 1p was accessed via silver mediated oxidative trifluoromethylation of 2-hydroxy-1-indanone.⁸⁹ Pushing it further, the linear analogue 1q was successfully reduced within 7 h using the same standard conditions, delivering the product 2q as a 3:1 mixture of *anti* and *syn* diastereomers with 97.4% and 90.4% ee, respectively. The reduction of 1-SCF₃-2-indanone 1r to the corresponding alcohol 2r was unfortunately not highly enantioselective (45% ee), although a 95:5 *cis/trans* ratio was achieved.

We were pleased to find out that some of the novel enantiopure compounds prepared by our method crystallize as



Figure 3. (a) Three-point bending experiment with elastically flexible needle-shaped crystal of **2p**. (b) Crystal packing of **2p**, view along *c*-axis. (c) Bent plastically flexible crystal of **2o**. (d) Crystal packing of **2o**, view along *b*-axis.

needle-shaped crystals which are elastically (2a, 2d, 2p, and 4d) or plastically (2o) flexible (Figure 3 and SI). Mechanically responsive molecular crystals are being recognized as an unexplored platform for applications ranging from adaptive systems and actuators to biocompatible devices and all-organic soft robots.^{90–94} The crystal structures of 2a, 2d, 2o, 2p, and 4d exhibit some of the same features that were identified in other crystals with $elastic^{95-98}$ or plastic deformation behavior: $^{99-101}$ in particular, a short crystal axis (~5 Å), anisotropic packing, corrugated crystal packing, and a prominent intermolecular interaction that is highly directional (i.e., hydrogen-bonded chains parallel to the short *a*-crystallographic axis in structures with P2₁2₁2₁ symmetry and parallel to the short *b*-crystallographic axis in compounds crystallizing in $P2_1$ space group) with much weaker interactions in the perpendicular directions. The slippage of molecular layers lined with trifluoromethyl groups has previously been established to be the mechanism of the observed plastic deformation.¹⁰¹ In our case, chiral OH and the semisaturated benzo-fused scaffold clearly also contribute to mechanic responsiveness as detrifluoromethylated bromoindanol 4d was also to some degree elastically flexible.¹⁰² Moreover, for plastically flexible 20, two polymorphs (RT $P2_1$, and 100 K $P2_12_12_1$) were identified. On the other hand, the single crystals of 2f $(P2_12_12_1)$, 2k (P1), and 2n $(P2_12_12_1)$ exhibit a typical brittle behavior, suggesting that subtle differences in molecular structure and crystal packing determine the sweet spot of homochiral single-component flexible crystals.

From the medicinal chemistry point of view, the stereopure products **2** represent hitherto synthetically inaccessible building blocks featuring intrinsic nonplanarity, potential for specific interactions with the protein binding sites, and several growth vectors.^{103–105} Selected stereopure products **2** were thus prepared on the 1 mmol scale, and relevant further synthetic transformations were demonstrated (Scheme 1). **2g** was transformed to *trans*-configured **5** via iron-catalyzed diastereoselective Friedel–Crafts benzylation of 2-chloroanisole.^{106,107} This hydroxy-substituted 1-arylindan motif is characteristic of resveratrol dimer natural products.^{108,109} **20** was converted to azide **6** (*trans/cis* = 92:8) via nucleophilic substitution (S_N2) of the corresponding mesylate ester. It was further reduced to the corresponding amine 7 which was

Scheme 1. Further Synthetic Transformations of Stereopure DKR-ATH Products 2



isolated as a single stereomer after chromatography. 2i was Oalkylated to obtain stereopure clickable building block 8. 2d was converted to biaryl 9 via Suzuki coupling reaction, illustrating that unprotected 2-CF₃-carbinols are compatible with palladium catalysis. And finally, stereopure 2a and 2c were reoxidized using pyridinium chlorochromate to obtain enantioenriched 1a and 1c with 57% and 92% ee, respectively. To showcase the direct applicability of the developed synthetic methods in a medicinal chemistry setting, alkyne 8 was incorporated into 10 which represents a novel structural class of heat shock protein 90 (Hsp90) inhibitors. Compound 10 was designed using a molecular-dynamics-derived pharmacophore model (Figure S2).^{110,111} It was shown to inhibit Hsp90 in the luciferase refolding assay and display antiproliferative activity in the SkBr3 breast cancer cell line (IC_{50} = 51 \pm 2 μM).

CONCLUSION

In conclusion, we have successfully developed a highly efficient dynamic kinetic resolution strategy for the Noyori–Ikariya asymmetric transfer hydrogenation of racemic α -CF₃, α -SCF₃, and α -OCF₃ aryl ketones with excellent stereoselectivities (up to 99.9% ee, up to 99.9:0.1 dr) and suppressed detrifluoromethylation. The origin of DKR (in situ epimerization of the ketone substrate and stereoselectivity) were investigated by DFT calculations. Applicability in the field of medicinal chemistry was demonstrated by several further transformations of the stereopure products including incorporation into a promising in vitro anticancer compound. Moreover, an unprecedented class of homochiral small organic molecules, which crystallize as mechanically responsive single-component crystals, was identified. Overall, the presented synthetic methodology opens the door to new chiral fluorinated bioactive lead compounds and to materials science applications based on adaptive chiral molecular crystals.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsorginorgau.2c00019.

Movie 2d: movie showing the elastic flexibility of 2d (AVI)

Movie 4d: movie showing the elastic flexibility of 4d (AVI)

Experimental procedures, chiral HPLC and GC chromatograms, NMR spectra of the prepared compounds, cell-based assays, computational and SCXRD details, photos of mechanically responsive behavior (PDF)

Accession Codes

CCDC 2151748–2151755 and 2155509 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/ cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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