

# Marine Sponge/H<sub>3</sub>PO<sub>4</sub>: As a Naturally Occurring Chiral Catalyst for Solvent-free Fischer-Indole Synthesis

Mohammad Reza Shushizadeh<sup>\*,1,2</sup>, Azar Mostoufi<sup>1</sup>, Rashid Badri<sup>3</sup>, Somaye Azizyan<sup>3</sup>

<sup>1</sup> Department of Medicinal Chemistry, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, IR Iran

<sup>2</sup> Marine Pharmaceutical Science Research Center, Ahvaz Jundishapur University of Medical Sciences, IR Iran

<sup>3</sup> Islamic Azad University-Khoozestan Science and Research Center, Ahvaz, IR Iran

\*Corresponding author: Mohammad Reza Shushizadeh, Department of Medicinal Chemistry, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, IR Iran, P.O. Box: 61357-73135. Tel: +986114445655, E-mail: m.r.shushizadeh@gmail.com, m.r.shushizadeh@ajums.ac.ir.

Received: April 26, 2013; Revised: May 27, 2013; Accepted: June 15, 2013

**Background:** A new and efficient method have been developed for the synthesis of different indole derivatives from various ketones, having at least one hydrogen atom attached to each of their  $\alpha$ -carbon atoms, and hydrazines in solvent-free conditions, using marine sponge/H<sub>3</sub>PO<sub>4</sub> as a naturally occurring chiral catalyst.

**Objectives:** This study recommended the use of marine sponge/H<sub>3</sub>PO<sub>4</sub> as a naturally occurring chiral catalyst for preparation of phenylhydrazones from ketones having one  $\alpha$ -hydrogen and subsequent cyclisation of the products to indoles.

**Materials and Methods:** The reaction was carried out by mixing the phenylhydrazine, ketone, and marine sponge/H<sub>3</sub>PO<sub>4</sub> powder in mortar and pestle; the mixture was ground at room temperature in an appropriate time until TLC show the completion of the reaction. The product extracted by CH<sub>2</sub>Cl<sub>2</sub> and evaporation of solvent yields the products.

**Results:** In this research work, several indoles are synthesized using phenylhydrazine and aliphatic or aromatic ketone as starting materials, in the presence of marine sponge/H<sub>3</sub>PO<sub>4</sub> powder as a natural catalyst under solvent-free condition.

**Conclusions:** We found marine sponge/H<sub>3</sub>PO<sub>4</sub> to be an effective catalyst for indolisation of phenylhydrazones from ketones having  $\alpha$ -hydrogens in solvent-free conditions.

**Keywords:** Ketones; Indole; Phenylhydrazine

## 1. Background

Indole ring systems became important structural components in many natural pharmaceutical agents (1). Their synthesis and functionalization are a major area of focus for synthetic organic chemists. Numerous methods have been developed for the synthesis of indoles (2, 3). The Fischer-indole synthesis, which uses ketones and arylhydrazines is the most widely employed synthetic procedure, especially for large- scale productions of biologically active compounds. This procedure involves an acid- catalyzed 3,3-sigmatropic rearrangement of an N-arylhydrazone intermediate, followed by elimination of ammonia. Several Bronsted acids (H<sub>2</sub>SO<sub>4</sub>, HCl, PPA, AcOH, TsOH, and ionic liquids), Lewis acids (ZnCl<sub>2</sub>, TiCl<sub>4</sub>, BiNO<sub>3</sub>, and Sc(OTf)<sub>3</sub>), and solid acids (zeolite, and Montmorillonite clay) have been reported to catalyze the Fischer-indole synthesis (4-19). Recently, propyl phosphonic acid in combination with cyclic anhydride (T3P) has also been reported recently to catalyze this synthesis (20). Although these methods are satisfactory for the synthesis of many molecules and other purposes, they have certain disadvantages, such as long time reaction, harsh reaction, us-

ing high amount of acid catalyst, low products, and using toxic, corrosive, expensive, or non-reusable catalysts. Another major drawback is that these methods do not exhibit any enantioselectivities induced by chiral and natural catalysts. These disadvantages limit their practical utility in large- scale synthesis processes. Consequently, there is a need to develop alternative reagents for these types of reactions. Marine sponges are known as a prolific source of biologically active and structurally unique metabolites (21, 22). We described the catalytic, chirality and absorbent abilities of marine sponge powder of Iranian coast of Persian Gulf in organic reaction such as sulfonamides synthesis (23), and researches about its application in organic reactions such as various oxidation, and reduction reactions (24); therefore we decided to study shallow sponges (*Desmospongia* sp.) of Qeshm and Bush-ehr Islands in offshore zone as an efficient chiral catalyst.

Other advantages of the marine sponge are that it can act as an absorbent to activate the C-N bond for nucleophilic preparation and cyclisation of phenylhydrazone, with high and predictable asymmetric induction, and can be removed easily from the product. For better catalyst, with respect to operational simplicity, greater yields,

### Implication for health policy makers/practice/research/medical education:

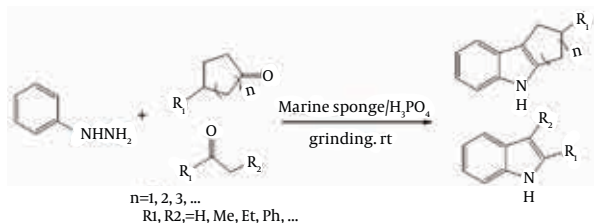
Marine natural catalyst catalyzed efficiently the Fischer-indole synthesis in solvent-free condition.

Copyright © 2013, School of Pharmacy, Ahvaz Jundishapur University of Medical Sciences; Published by DOCS. This is an open-access article distributed under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/3.0>), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

indolisation reactions of cyclic and acyclic ketones more research in this field is demanded.

## 2. Objectives

In continuation of our studies on the solvent-free and marine natural catalytic reactions, it was decided to report the use of marine sponge/ $H_3PO_4$  as a naturally occurring chiral catalyst for preparation of phenylhydrazones from ketones having one  $\alpha$ -hydrogen and subsequent cyclisation of the products to indoles, as shown in Figure 1.



**Figure 1.** Preparation of Indoles by Cyclization of Hydrazones in Presence of Marine Sponge/ $H_3PO_4$

## 3. Materials and Methods

### 3.1. Reagents and Materials

All starting materials were purchased from Merck and Aldrich Companies. The IR spectra were recorded on a Perkin-Elmer RXI infrared spectrometer.  $^1H$  NMR spectra were recorded on a 400 MHz Bruker FT-NMR spectrometer. The SEM image was recorded on 1455 VP LEO-Germany. TLC accomplished the purity of substrates and reactions monitored on silica gel (Merck, Germany) polygram SIGL/UV254 plates. The melting points are uncorrected.

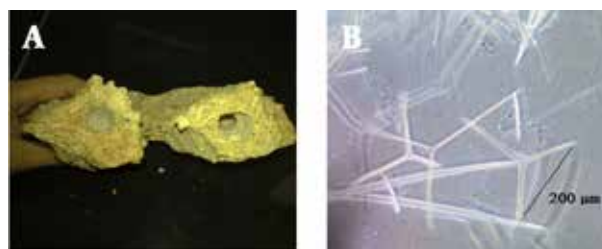
### 3.2. Preparation of Marine Sponge Powder

In this study, marine sponges (*Demospongiae* sp.) collected from Nakhiloo Island, Bushehr, Iran (North coast of Persian Gulf), in May 2010 at a depth between 5 and 10 m, and were washed several times using methanol and then deionized water to remove some organic compounds, extraneous and salts. Then, they were dried in an oven at  $60^\circ C$  for 48 hours. The dried marine sponge was chopped and the particles were separated according to their sieves. Identification of sponges was carried out kindly by Khoramshahr marine science and Technology University. It was indicated as *Demospongiae* sp., that has siliceous ( $SiO_2$ ) spicules, as shown in Figure 2.

### 3.3. Preparation of Marine Sponge/ $H_3PO_4$ Mixture

Marine sponge/ $H_3PO_4$  mixture was prepared simply by

grinding 0.9 g marine sponge powder, and 0.1 mL concentrated phosphoric acid in a mortar until a fine and homogeneous mixture was obtained.



**Figure 2.** A. Marine Sponge (*Demospongiae* sp.), B. SEM of Siliceous Spicules

### 3.4. General Procedure

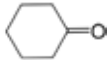
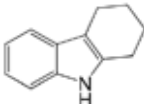

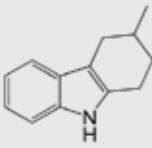
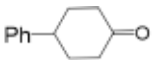
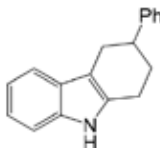
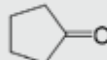
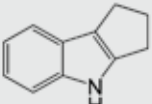
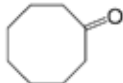
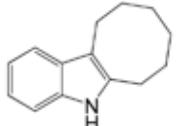

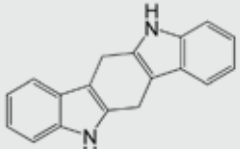

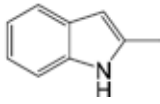
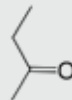
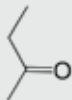

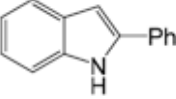
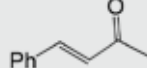
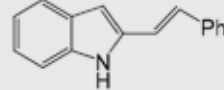
The reaction was carried out by mixing the phenylhydrazine (50 mmol), ketone (50 mmol), and 0.1 g of marine sponge/ $H_3PO_4$  mixture in mortar and pestle; the excessive amount of ketones such as acetone and ethyl methyl ketone with low boiling point, was used. The mixture was ground at room temperature in appropriate time reported in Table 1. The progress of reaction was monitored by TLC. After completion of the reaction, the mixture was added to 25 mL of ice/water and stirred. The product was extracted with  $CH_2Cl_2$  ( $2 \times 25$  mL), and washed with 10 mL of 5% sodium hydrogen carbonate and 10 mL of water. Then the solution was dried over anhydrous  $CaCl_2$  and filtered. The solvent was evaporated under reduced pressure, and the crude product was purified by recrystallization in 50% ethanol-water to make the pure product. Note: phenylhydrazine is a suspected carcinogen; therefore, gloves should be worn whenever working with compound. Some products are readily oxidized, so they must not be dried in an oven.

## 4. Results

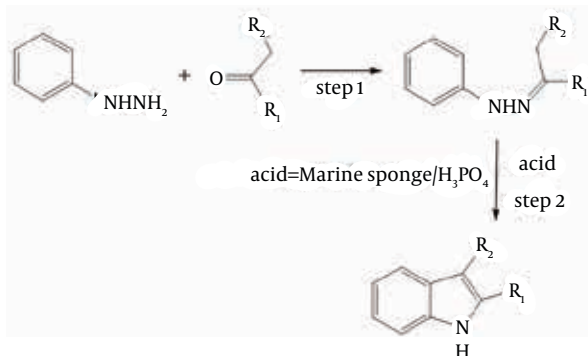
Indoles bearing various substituents at positions 2 and 3 can be synthesized via the Fisher-indole synthesis, which involves two steps and uses phenylhydrazine and an aliphatic or aromatic ketone as starting materials. During the synthesis in our experiments, phenylhydrazine reacted with ketone to produce a phenylhydrazone, as shown in step 1 of Figure 3. An acid was added to catalyze, the cyclization reaction and the subsequent loss of one of the nitrogen atoms as ammonia, as shown in step 2 of the Figure 3 (25, 26). Various acid have been used for catalyzing the second reaction.

In order to find out the most effective catalyst for indolisation, we performed the Fischer- indole reaction of 1:1 equimolar phenylhydrazine with ketones at room temperature using various catalysts, as shown in Table 2.

**Table 1.** Solvent-Free Fischer-Indole Synthesis of Phenyl-Hydrazine and Ketones Using Marine Sponge/H<sub>3</sub>PO<sub>4</sub><sup>a</sup>

Entry	Ketone	Product	Time, min	Yield, %	Melting Point, °C	Enantiomeric Execc, %
1			5	90	116-118	-
2			12	85	109-111	92
3			7	89	123-125	90
4			15	55	101-103	-
5			13	81	150-152	-
6			25	69	201-203	-
7			17	70	58-60	-
8			15	75	106-108	-
9			10	78	187-189	-
10			30	62	179-181	-

<sup>a</sup> All products were confirmed by comparison with authentic samples (IR, <sup>1</sup>H NMR and TLC). The reaction was carried out by mixing the phenylhydrazine (50 mmol), ketone (50 mmol), and 0.1 g of marine sponge/H<sub>3</sub>PO<sub>4</sub> mixture; the excessive amount of ketones such as acetone and ethyl methyl ketone with low boiling point, was used.



**Figure 3.** Mechanism of Indoles Synthesis in Presence of Marine Sponge/ $H_3PO_4$

**Table 2.** Solvent-free Fischer-indole Reaction of Phenylhydrazine With Cyclohexanone as Ketone, in the Presence of Different Catalysts

Catalysts	Time, min	Yield, %	Condition
Marine sponge/ $H_3PO_4$	5	90	r.t.
[cmmim][BF <sub>4</sub> ] (12)	120	92	140 °C
Propylphosphonic Anhydride (20)	5	90	MW, 100 °C
PPA (16)	15	89	95 °C

## 5. Discussion

As the results indicate, marine sponge/ $H_3PO_4$  are more advantageous over previously used catalysts, which are as follows: requirement of a very small amount of the catalyst, mild reaction, room temperature, good solid absorbent, easy to handle, and products in good-to-high yields. More importantly, this catalyst produced enantioselective products with high enantiomeric excess, as shown in Table 1 (entries 2 and 3). Structures of the products were characterized by their spectral ( $^1H$  NMR, IR, and MS) data (27, 28).

### 5.1. Characterization of Products

Selected spectral data for the products in Table 1 are given:

6,7,8,9-tetrahydro-5H-carbazole (14-15)-(entry 1): IR( $cm^{-1}$ , KBr): 3396 (-NH), 1614, 1583 (C=C), 738, 635 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.67 (1H, bs, NH), 7.55 (1H, d, J=7.52), 7.41 (1H, d, J=7.91), 7.22 (1H, t, J=7.75), 7.05 (1H, t, J=7.1), 2.9 (2H, t,  $CH_2$ , J=3.1), 2.7 (2H, t,  $CH_2$ , J=5.2), 1.8 (4H, m, 2 $CH_2$ );  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 136.1, 134.1, 127.7, 121.1, 119.0, 117.7, 110.3, 109.0, 24.7, 22.9, 22.0, 21.2.

3-Methyl-2,3,4,9-Tetrahydro-1H-carbazole-(entry 2): IR( $cm^{-1}$ , KBr): 3396 (-NH), 1619, 1583 (C=C), 740, 635 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.6 (1H, bs,

NH), 7.5 (1H, d, J=7.5 Hz), 7.38 (1H, d, J=7.95 Hz), 7.25 (1H, t, J=7.7 Hz), 7.03 (1H, t, J=7.1 Hz), 2.9 (2H, t,  $CH_2$ , J=9 Hz), 2.7 (2H, d,  $CH_2$ , J=6.5 Hz), 2.1 (2H, m,  $CH_2$ ), 1.6 (1H, m), 1.2 (3H, d,  $CH_3$ , J=6.45 Hz);  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 135.7, 134.1, 128.8, 124.1, 119.0, 117.7, 111.4, 110.0, 30.9, 29.0, 28.3, 26.2, 21.7.

3-Phenyl-2,3,4,9-Tetrahydro-1H-carbazole-(entry 3): IR( $cm^{-1}$ , KBr): 3396 (-NH), 1644, 1604 (C=C), 749, 682 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.65 (1H, bs, NH), 7.54 (1H, d, J=7.55 Hz), 7.45 (1H, d, J=8 Hz), 7.36 (1H, t, J=7.7 Hz), 7.30 (1H, t, J=7.0 Hz), 7.2 (2H, t, J=8.1 Hz), 7.15 (3H, m), 3.0 (2H, d,  $CH_2$ , J=11 Hz), 2.8 (2H, d,  $CH_2$ , J=6.9), 2.2 (2H, m,  $CH_2$ ), 1.9 (1H, m);  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 145.8, 136.0, 135.7, 132.90, 132.92, 128.9, 126.7, 126.8, 124.9, 124.1, 119.0, 117.7, 111.4, 110.0, 38.0, 32.3, 30.3, 26.8.

1,2,3,4-Tetrahydrocyclopenta[b]indole (14-15)-(entry 4): IR( $cm^{-1}$ , KBr): 3402 (-NH), 1600, 1502 (C=C), 749, 693 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.59 (1H, bs, NH), 7.45 (1H, d, J=7.15 Hz), 7.33 (1H, d, J=7.05 Hz), 7.2 (1H, t, J=7.4 Hz), 7.05 (1H, t, J=7.25 Hz), 3.0 (2H, t,  $CH_2$ , J=7 Hz), 2.6 (2H, t,  $CH_2$ , J=7 Hz), 1.59 (2H, p,  $CH_2$ , J=7 Hz);  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 137.0, 135.6, 125.8, 122.0, 120.8, 120.0, 112.1, 111.0, 26.9, 25.8, 24.9.

6,7,8,9,10,11-Hexahydrocycloocta[b]indole (14-15)-(entry 5): IR( $cm^{-1}$ , KBr): 3391 (-NH), 1619, 1599 (C=C), 743, 698 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.5 (1H, bs, NH), 7.49 (1H, d, J=7.3 Hz), 7.35 (1H, d, J=7.55 Hz), 7.2 (1H, t, J=7.1 Hz), 7.04 (1H, t, J=7.45 Hz), 2.75-2.85 (4H, d, 2 $CH_2$ ), 2.5-2.65 (4H, m, 2 $CH_2$ ), 2.1 (4H, m, 2 $CH_2$ );  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 136.6, 135.1, 129.7, 124.7, 119.0, 117.7, 112.6, 110.4, 28.6, 26.5, 26, 25.9, 25.7, 22.2.

5,6,11,12-Tetrahydro-6,12-diaza-indeno[1,2-b]fluorene-(entry 6): IR( $cm^{-1}$ , KBr): 3391 (-NH), 1614, 1583 (C=C), 741, 697 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.7 (2H, bs, 2NH), 7.6 (2H, d, J=8.3 Hz), 7.43 (2H, d, J=7.8 Hz), 7.03 (2H, t, J=8.15 Hz), 6.7 (2H, t, J=7.3 Hz), 4.5 (4H, m, 2 $CH_2$ );  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 136.1 (2C), 132.4 (2C), 124.4 (2C), 118.9 (2C), 118 (2C), 111.7 (2C), 107.7 (4C), 30.9 (2C).

2-Methylindole (14-15)-(entry 7): IR( $cm^{-1}$ , KBr): 3399 (-NH), 1558 (C=C), 738, 691 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.6 (1H, bs, NH), 7.51 (1H, d, J=7.2 Hz), 7.4 (1H, d, J=7.6 Hz), 7.23 (1H, t, J=7.44 Hz), 7.1 (1H, t, J=7.1 Hz), 6.7 (1H, m), 1.9-2.3 (3H, s,  $CH_3$ );  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 137.1, 136.0, 129.6, 120.9, 120.0, 119.6, 110.2, 100.1, 15.9.

2,3-Dimethylindole (14-15)-(entry 8): IR( $cm^{-1}$ , KBr): 3402 (-NH), 1618, 1558 (C=C), 738, 636 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.45 (1H, bs, NH), 7.35 (1H, d, J=7.5 Hz), 7.35 (1H, d, J=8 Hz), 7.15 (1H, t, J=7.78 Hz), 7.05 (1H, t, J=7 Hz), 2.7 (3H, s,  $CH_3$ ), 2.5 (3H, s,  $CH_3$ );  $^{13}C$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 137.1, 129.0, 121.4, 120.1, 118.0, 113.6, 110.2, 106.1, 15.9, 10.1.

2-Phenylindole-(entry 9): IR( $cm^{-1}$ , KBr): 3350 (-NH), 1599 (C=C), 744, 682 (aromatic),  $^1H$  NMR (400 MHz,  $CDCl_3$ , TMS,  $\delta$  ppm): 7.9 (1H, bs, NH), 7.65 (1H, d, J=7.5 Hz), 7.5 (1H, d,

J=8.2 Hz), 7.41 (1H, t, J=7.4 Hz), 7.3 (1H, t, J=7.05 Hz), 7.25 (3H, m), 7.1 (2H, t, J= 8 Hz), 7 (1H, m); <sup>13</sup>C NMR (400 MHz, CDCl<sub>3</sub>, TMS, δ ppm): 137.7, 135.2, 132.3, 128.3, 128.2, 128, 127, 124.2, 124.1, 121.5, 119.9, 119.1, 110.1, 100.1.

2-Styrylindole-(entry 10): IR(cm<sup>-1</sup>, KBr): 3345 (-NH ), 1601, 1504 (C=C), 748, 693 (aromatic), <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>, TMS, δ ppm): 8.7 (1H, bs, NH), 7.8 (1H, d, J=7.6 Hz), 7.5 (1H, d, J=7.33 Hz), 7.42 (1H, t, J=7.53 Hz), 7.32 (1H, t, J=7 Hz), 7.2 (3H, m), 7.1 (2H, t, J= 8.3 Hz), 7 (1H, m), 6.9 (1H, d, J=15 Hz), 6.71 (1H, d, J=18 Hz); <sup>13</sup>C NMR (400 MHz, CDCl<sub>3</sub>, TMS, δ ppm): 145.3, 141.2, 139.2, 129.3 (2C), 128.4, 128.2 (2C), 128.0, 125.6, 120.2, 113.3, 107.4 (2C).

In conclusion, we found marine sponge/H<sub>3</sub>PO<sub>4</sub> to be an effective catalyst for indolisation of phenylhydrazones from ketones having α-hydrogens in solvent-free conditions. The quality of indolic products is good. This method offered marked improvement compared to previously reported ones. Its advantages included operational simplicity, low reaction time, and high yields of pure products.

## Acknowledgements

The authors wish to thank Marine Pharmaceutical Research Center of Ahvaz Jundishapur University of Medical Science; and professor Sayed Mohammad Bagher Nabavi for financial support of this work.

## Authors' Contribution

None declared.

## Financial Disclosure

There is no financial disclosure.

## Funding/Support

This article has been supported by Ahvaz Jundishapur University of Medical Sciences.

## References

- Humphrey GR, Kueth JT. Practical methodologies for the synthesis of indoles. *Chem Rev*. 2006;**106**(7):2875-911.
- Gribble GW. Recent developments in indole ring synthesis—Methodology and applications. *J Chem Soc Perkin Trans*. 2000;**7**:1045-75.
- Sundberg RJ. *Indoles (Best Synthetic Methods)*. In: Katritzky AR, Meth-Cohn O, Rees CS, editors. London: Academic press; 1996.
- Inman M, Moody CJ. A two step route to indoles from haloarenes—a versatile variation on the Fischer indole synthesis. *Chem Commun (Camb)*. 2011;**47**(2):788-90.
- Ashcroft CP, Hellier P, Pettman A, Watkinson S. Second-Generation Process Research Towards Eletriptan: A Fischer Indole Approach. *Org Proc Res Dev*. 2010;**15**(1):98-103.
- Wahab B, Ellames G, Passey S, Watts P. Synthesis of substituted indoles using continuous flow micro reactors. *Tetrahedron*. 2010;**66**(21):3861-5.
- Jiang H, Wang Y, Wan W, Hao J. p-TsOH promoted Fischer indole synthesis of multi-substituted 2-trifluoromethyl indole derivatives. *Tetrahedron*. 2010;**66**(14):2746-51.
- Xu DQ, Wu J, Luo SP, Zhang JX, Wu JY, Du XH, et al. Fischer indole synthesis catalyzed by novel SO<sub>3</sub>H-functionalized ionic liquids in water. *Green Chem*. 2009;**11**(8):1239-46.
- Park IK, Suh SE, Lim BY, Cho CG. Aryl hydrazide beyond as surrogate of aryl hydrazine in the Fischer indolization: the synthesis of N-Cbz-indoles, N-Cbz-carbazoles, and N,N'-bis-Cbz-pyrrolo[2,3-f]indoles. *Org Lett*. 2009;**11**(23):5454-6.
- Jeanty M, Blu J, Suzenet F, Guillaumet G. Synthesis of 4- and 6-azaindoles via the Fischer reaction. *Org Lett*. 2009;**11**(22):5142-5.
- Sudhakar A, Jayadevappa H, Mahadevan KM, Hulikal V. Efficient synthesis of 2-ethoxycarbonyl indoles. *Synth Commun*. 2009;**39**(14):2506-15.
- Varma PP, Sherigara BS, Mahadevan KM, Hulikal V. Efficient and straightforward synthesis of tetrahydrocarbazoles and 2, 3-dimethyl indoles catalyzed by CAN. *Synth Commun*. 2008;**39**(1):158-65.
- Chen J, Chen W, Hu Y. Microwave-Enhanced Fischer Reaction: An Efficient One-Pot Synthesis of γ-Carbolines. *Synlett*. 2008;**2008**(1):77-82.
- Xu DQ, Yang WL, Luo SP, Wang BT, Wu Jian, Xu ZY. Fischer indole synthesis in brønsted acidic ionic liquids: a green, mild, and regio-specific reaction system. *Eur J Org Chem*. 2007;**2007**(6):1007-12.
- Robinson B, Robinson B. *The Fischer indole synthesis*. New York: Wiley; 1982.
- Kissman HMF, Farnsworth DW, Witkop B. Fischer indole syntheses with polyphosphoric acid. *J Am Chem Soc*. 1952;**74**(15):3948-9.
- Bratulescu G. A new and efficient one-pot synthesis of indoles. *Tetrahedron Lett*. 2008;**49**(6):984-6.
- Yi FP, Sun HY, Pan XH, Xu Y, Li JZ. Synthesis of Fischer indole derivatives using carboxyl-functionalized ionic liquid as an efficient and recyclable catalyst. *Chin Chem Lett*. 2009;**20**(3):275-8.
- Chaskar A, Deokar H, Padalkar V, Phatangare K, Patil SK. Highly Efficient and Facile Green Approach for One-Pot Fischer Indole Synthesis. *J Korean Chem Soc*. 2010;**54**(4):411-3.
- Desroses M, Wieckowski K, Stevens M, Odell LR. A microwave-assisted, propylphosphonic anhydride (T3P®) mediated one-pot Fischer indole synthesis. *Tetrahedron Lett*. 2011;**52**(34):4417-20.
- Rifai S, Fassouane A, El-Abbouyi A, Wardani A, Kijjoa A, Van Soest R. Screening of antimicrobial activity of marine sponge extracts. *J Med Mycol*. 2005;**15**(1):33-8.
- McCaffrey EJ, Edean R. Antimicrobial activity of tropical and subtropical sponges. *Marine Biol*. 1985;**89**(1):1-8.
- Shushizadeh MR, Mostoufi A, Fakhrian M. Marine Sponge/CuO Nanocrystal: A Natural and Efficient Catalyst for Sulfonamides Synthesis. *Jundishapur J Nat Pharm Prod*. 2012;**7**(4):134-9.
- Sarma NS, Krishna M, Rao SR. Sterol ring system oxidation pattern in marine sponges. *Marine Drugs*. 2005;**3**(3):84-111.
- Hughes DL. Progress in the Fischer indole reaction. A review. *Org Prep Proced Int*. 1993;**25**(6):607-32.
- Przheval'skii NM, Kostromina LY, Grandberg II. Mechanism of Fischer reaction. Dependence of thermal indolization of cyclohexanone arylhydrazones on nature of substituent in the benzene ring. *Chem Heterocycl Comp*. 1988;**24**(2):154-8.
- Dhakshinamoorthy A, Pitchumani K. Facile clay-induced Fischer indole synthesis: A new approach to synthesis of 1, 2, 3, 4-tetrahydrocarbazole and indoles. *Appl Catal A Gen*. 2005;**292**:305-11.
- Chen J, Hu Y. Microwave Assisted One-Pot Synthesis of 1, 2, 3, 4-Tetrahydrocarbazoles. *Synth Commun*. 2006;**36**(11):1485-94.