

## 4-Hydroxypyridinium hydrogen sulfate

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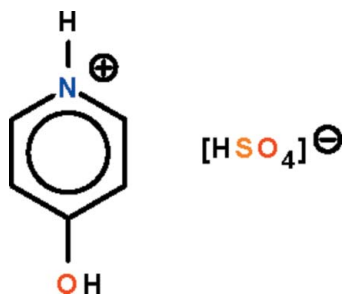
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Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.036;  $wR$  factor = 0.098; data-to-parameter ratio = 12.6.

The crystal structure of the title salt,  $\text{C}_5\text{H}_6\text{NO}^+\cdot\text{HSO}_4^-$ , consists of planar (r.m.s. deviation = 0.001 Å) 4-hydroxypyridinium cations and hydrogen sulfate anions which are hydrogen bonded into a layer motif. In the anion, the S—O bond [1.551 (2) Å] involving the O atom bearing the acid H atom is longer than the other three S—O bonds, which range from 1.437 (1) to 1.454 (1) Å.

### Related literature

For the crystal structures of bis(4-hydroxypyridinium) sulfate monohydrate and tris(4-hydroxypyridinium) hydrogen disulfate monohydrate, see: Xu *et al.* (2009*a,b*).



### Experimental

#### Crystal data

$\text{C}_5\text{H}_6\text{NO}^+\cdot\text{HSO}_4^-$

$M_r = 193.18$

Monoclinic,  $P2_1/c$

$a = 10.4541$  (7) Å

$b = 10.7017$  (6) Å

$c = 6.8397$  (4) Å

$\beta = 96.503$  (2)°

$V = 760.28$  (8) Å<sup>3</sup>

$Z = 4$

Mo  $K\alpha$  radiation

$\mu = 0.41$  mm<sup>-1</sup>

$T = 293$  K

0.27 × 0.21 × 0.15 mm

#### Data collection

Rigaku R-Axis RAPID IP

diffractometer

Absorption correction: multi-scan

(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.898$ ,  $T_{\max} = 0.941$

7273 measured reflections

1729 independent reflections

1612 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.017$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$

$wR(F^2) = 0.098$

$S = 1.03$

1729 reflections

137 parameters

7 restraints

All H-atom parameters refined

$\Delta\rho_{\text{max}} = 0.52$  e Å<sup>-3</sup>

$\Delta\rho_{\text{min}} = -0.39$  e Å<sup>-3</sup>

Table 1

Selected bond lengths (Å).

S1—O1	1.445 (1)	S1—O3	1.437 (1)
S1—O2	1.551 (2)	S1—O4	1.454 (1)

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O2—H2 $\cdots$ O4 <sup>i</sup>	0.85 (1)	1.77 (1)	2.603 (2)	168 (3)
O5—H5 $\cdots$ O1	0.85 (1)	1.77 (1)	2.6166 (19)	175 (3)
N1—H1 $\cdots$ O3 <sup>ii</sup>	0.84 (1)	2.04 (1)	2.8529 (19)	163 (2)

Symmetry codes: (i)  $x, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (ii)  $x + 1, y, z$ .

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalClear* (Rigaku/MSK, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2677).

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**supplementary materials**

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## 4-Hydroxypyridinium hydrogen sulfate

L.-H. Huo, Y.-M. Xu, S. Gao and S. W. Ng

### Experimental

The compound is a side product that was obtained when commercially available 4-hydroxypyridine-3-sulfonic acid was recrystallized from water. Its crystals were obtained from a water solution.

### Refinement

Carbon-bound H-atoms refined with a C–H distance restraint of  $0.95 \pm 0.01$  Å; their temperature factors were refined. The nitrogen- and oxygen-bound H-atoms were refined with a distance restraint of  $N-H = O-H = 0.85 \pm 0.01$  Å; their temperature factors were refined.

### Figures

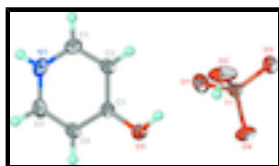


Fig. 1. Thermal ellipsoid plot (Barbour, 2001) of  $[C_5H_6NO]^+ [HSO_4]^-$  at the 50% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

## 4-Hydroxypyridinium hydrogen sulfate

### Crystal data



$$M_r = 193.18$$

Monoclinic,  $P2_1/c$

Hall symbol:  $-P 2ybc$

$$a = 10.4541 (7) \text{ \AA}$$

$$b = 10.7017 (6) \text{ \AA}$$

$$c = 6.8397 (4) \text{ \AA}$$

$$\beta = 96.503 (2)^\circ$$

$$V = 760.28 (8) \text{ \AA}^3$$

$$Z = 4$$

$$F_{000} = 400$$

$$D_x = 1.688 \text{ Mg m}^{-3}$$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 6627 reflections

$$\theta = 3.6\text{--}27.4^\circ$$

$$\mu = 0.41 \text{ mm}^{-1}$$

$$T = 293 \text{ K}$$

Prism, colorless

$$0.27 \times 0.21 \times 0.15 \text{ mm}$$

### Data collection

Rigaku R-Axis RAPID IP  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

1729 independent reflections

1612 reflections with  $I > 2\sigma(I)$

$$R_{\text{int}} = 0.017$$

# supplementary materials

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$T = 293$  K  $\theta_{\max} = 27.4^\circ$   
 $\omega$  scan  $\theta_{\min} = 3.6^\circ$   
Absorption correction: Multi-scan  
(ABSCOR; Higashi, 1995)  $h = -13 \rightarrow 13$   
 $T_{\min} = 0.898$ ,  $T_{\max} = 0.941$   $k = -13 \rightarrow 13$   
7273 measured reflections  $l = -8 \rightarrow 8$

## Refinement

Refinement on  $F^2$  Secondary atom site location: difference Fourier map  
Least-squares matrix: full Hydrogen site location: inferred from neighbouring sites  
 $R[F^2 > 2\sigma(F^2)] = 0.036$  All H-atom parameters refined  
 $wR(F^2) = 0.098$   $w = 1/[\sigma^2(F_o^2) + (0.0587P)^2 + 0.3712P]$   
 $S = 1.03$  where  $P = (F_o^2 + 2F_c^2)/3$   
1729 reflections  $(\Delta/\sigma)_{\max} = 0.001$   
137 parameters  $\Delta\rho_{\max} = 0.52 \text{ e } \text{\AA}^{-3}$   
7 restraints  $\Delta\rho_{\min} = -0.39 \text{ e } \text{\AA}^{-3}$   
Primary atom site location: structure-invariant direct methods Extinction correction: none

## Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.15352 (4)	0.36664 (3)	0.22053 (6)	0.02984 (15)
O1	0.28767 (13)	0.37320 (14)	0.2986 (3)	0.0580 (4)
O2	0.1438 (2)	0.41518 (14)	0.0056 (2)	0.0657 (5)
O3	0.07735 (11)	0.45271 (12)	0.3196 (2)	0.0423 (3)
O4	0.10415 (14)	0.23950 (12)	0.21136 (18)	0.0430 (3)
O5	0.46894 (12)	0.20467 (13)	0.2977 (2)	0.0486 (4)
N1	0.82340 (14)	0.36324 (15)	0.3509 (2)	0.0416 (4)
C1	0.72047 (18)	0.43863 (17)	0.3364 (3)	0.0388 (4)
C2	0.59869 (16)	0.39037 (16)	0.3180 (3)	0.0351 (4)
C3	0.58308 (15)	0.26042 (15)	0.3148 (2)	0.0320 (3)
C4	0.69235 (16)	0.18433 (16)	0.3312 (3)	0.0359 (4)
C5	0.81115 (16)	0.23848 (19)	0.3484 (3)	0.0410 (4)
H1	0.8975 (14)	0.395 (2)	0.366 (4)	0.063 (7)*
H2	0.134 (3)	0.357 (2)	-0.078 (4)	0.079 (9)*
H5	0.412 (2)	0.2615 (19)	0.293 (4)	0.063 (7)*
H1A	0.735 (2)	0.5257 (10)	0.338 (3)	0.044 (6)*
H2A	0.5283 (16)	0.4452 (18)	0.302 (3)	0.049 (6)*
H4	0.6824 (19)	0.0967 (9)	0.330 (3)	0.040 (5)*
H5A	0.8881 (16)	0.191 (2)	0.356 (4)	0.061 (7)*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0273 (2)	0.0272 (2)	0.0352 (2)	-0.00275 (13)	0.00428 (15)	-0.00138 (13)
O1	0.0264 (7)	0.0495 (9)	0.0963 (12)	0.0041 (5)	-0.0012 (7)	-0.0159 (8)
O2	0.1253 (16)	0.0347 (7)	0.0384 (7)	-0.0135 (9)	0.0154 (8)	0.0017 (6)
O3	0.0312 (6)	0.0428 (7)	0.0538 (8)	0.0005 (5)	0.0090 (5)	-0.0101 (6)
O4	0.0581 (8)	0.0317 (6)	0.0391 (6)	-0.0130 (5)	0.0045 (5)	0.0004 (5)
O5	0.0254 (6)	0.0346 (7)	0.0852 (11)	-0.0021 (5)	0.0041 (6)	-0.0049 (6)
N1	0.0289 (7)	0.0507 (9)	0.0457 (8)	-0.0105 (6)	0.0064 (6)	-0.0084 (7)
C1	0.0439 (9)	0.0337 (8)	0.0397 (9)	-0.0068 (7)	0.0086 (7)	-0.0045 (7)
C2	0.0329 (8)	0.0303 (8)	0.0424 (9)	0.0037 (6)	0.0052 (7)	-0.0022 (6)
C3	0.0268 (7)	0.0316 (8)	0.0376 (8)	-0.0003 (6)	0.0037 (6)	-0.0026 (6)
C4	0.0315 (8)	0.0308 (8)	0.0454 (9)	0.0034 (6)	0.0037 (7)	-0.0031 (7)
C5	0.0279 (8)	0.0485 (10)	0.0466 (9)	0.0043 (7)	0.0040 (7)	-0.0058 (8)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

S1—O1	1.445 (1)	N1—H1	0.841 (10)
S1—O2	1.551 (2)	C1—C2	1.366 (2)
S1—O3	1.437 (1)	C1—H1A	0.945 (10)
S1—O4	1.454 (1)	C2—C3	1.400 (2)
S1—O2	1.5514 (15)	C2—H2A	0.938 (10)
O2—H2	0.848 (10)	C3—C4	1.397 (2)
O5—C3	1.3273 (19)	C4—C5	1.363 (2)
O5—H5	0.849 (10)	C4—H4	0.943 (9)
N1—C1	1.340 (2)	C5—H5A	0.947 (10)
N1—C5	1.341 (3)		
O3—S1—O1	111.16 (8)	C2—C1—H1A	121.6 (13)
O3—S1—O4	114.04 (8)	C1—C2—C3	118.86 (16)
O1—S1—O4	112.73 (9)	C1—C2—H2A	119.0 (14)
O3—S1—O2	104.60 (9)	C3—C2—H2A	122.1 (14)
O1—S1—O2	106.91 (11)	O5—C3—C4	117.63 (15)
O4—S1—O2	106.71 (8)	O5—C3—C2	123.36 (15)
S1—O2—H2	113 (2)	C4—C3—C2	119.01 (15)
C3—O5—H5	107.5 (19)	C5—C4—C3	119.20 (16)
C1—N1—C5	121.59 (15)	C5—C4—H4	121.5 (12)
C1—N1—H1	119.2 (19)	C3—C4—H4	119.3 (12)
C5—N1—H1	119 (2)	N1—C5—C4	120.59 (16)
N1—C1—C2	120.75 (16)	N1—C5—H5A	116.9 (16)
N1—C1—H1A	117.6 (13)	C4—C5—H5A	122.5 (16)
C5—N1—C1—C2	-0.1 (3)	O5—C3—C4—C5	179.79 (17)
N1—C1—C2—C3	0.1 (3)	C2—C3—C4—C5	-0.4 (3)
C1—C2—C3—O5	179.94 (17)	C1—N1—C5—C4	-0.1 (3)
C1—C2—C3—C4	0.2 (2)	C3—C4—C5—N1	0.4 (3)

## supplementary materials

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### Hydrogen-bond geometry (Å, °)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O2—H2 $\cdots$ O4 <sup>i</sup>	0.85 (1)	1.77 (1)	2.603 (2)	168 (3)
O5—H5 $\cdots$ O1	0.85 (1)	1.77 (1)	2.6166 (19)	175 (3)
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Symmetry codes: (i)  $x, -y+1/2, z-1/2$ ; (ii)  $x+1, y, z$ .

Fig. 1

