

Article

# Constituents of *Gastrodia elata* and Their Neuroprotective Effects in HT22 Hippocampal Neuronal, R28 Retinal Cells, and BV2 Microglial Cells

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**Abstract:** *Gastrodia elata* is widely used in traditional medicine and contains various types of metabolites with pharmacological activity. In the course of searching for neuroprotective molecules associated with the potential of *G. elata* in the treatment of neurodegenerative disorders, two new phenolic compounds (**1** and **2**) and a new tripeptide (**3**), together with 16 known compounds (**4–19**), were isolated from the rhizomes of *G. elata*. The structures of the compounds were determined by the interpretation of spectroscopic data, including nuclear magnetic resonance and mass spectrometry data. All obtained compounds were assessed for their ability to protect neuronal cells against neurotoxicity and neuroinflammation. Of these, **4** and **5** were found to possess moderate activities in HT22 hippocampal neuronal cells, whereas **2**, **6**, and **7** showed weak activities in R28 retinal cells. Additionally, compound **9** showed moderate inhibitory activity on lipopolysaccharide-induced nitric oxide production in BV2 microglial cells.

**Keywords:** *Gastrodia elata;* phenolic compound; tripeptide; neuroprotection; neuronal cell; microglial cell

#### 1. Introduction

Oxidative stress is implicated in several neurological and ophthalmic diseases, including Alzheimer's disease, Parkinson's disease, ischemic stroke, and glaucoma [1,2]. Glutamate or hydrogen peroxide ( $H_2O_2$ ) toxicity is a common model used to study oxidative stress-induced neuronal or optic neuronal cell death, which is associated with acute and chronic cellular insults [1]. In neuronal cells, a high concentration of extracellular glutamate can inhibit the production of glutathione (GSH) by disturbing the uptake of cystine into cells, and leading to the accumulation of reactive oxygen species (ROS). An increase in intracellular ROS induces neuronal cell death by multiple pathways, including apoptosis, necrosis, ferroptosis, or other forms of cell death [3].

Retina cells are a type of neuron that have originated from a similar lineage of neuronal cells. Retina tissue is high oxygen-consuming and converts visual information into electronic signals that can transfer from the eye to the brain. An increase in oxygen consumption can induce oxidative stress that, in turn, can induce damage to the retina and represents a risk factor for several retinal disorders, including glaucoma [4,5]. Therefore, protecting neuronal and retina cells from oxidative stress is a major therapeutic strategy in preventing the pathogenesis of oxidative stress-related neurological and ophthalmic diseases [6].



Besides, the modulation of nitric oxide (NO) secretion in microglia cells can be beneficial to neurodegenerative diseases. The microglia cell is a main glial cell type that plays pivotal roles in the innate immune response and phagocytosis that can remove cell debris and damaged neurons. In disease conditions, microglia activated by lipopolysaccharide (LPS) significantly accelerates neuroinflammation which releases cyclooxygenase-2, NO synthase as well as interleukin-1b and -6, tumor necrosis factor- $\alpha$ , and NO. In particular, a high level of NO can induce oxidative stress and neuron destruction contributing to a neurodegenerative disorder [7].

*Gastrodia elata* Blume (family Orchidaceae), commonly known as tianma, has been used for the treatment of neurological disorders such as headache, dizziness, limb numbness, infantile convulsion, vertigo, epilepsy, and tonic spasm [8]. Previous phytochemical investigations of *G. elata* have revealed the presence of various phenolic compounds, sterols, and organic acids, which exhibit a range of biological activities, including neuroprotective, antioxidant, anti-inflammatory, and anticonvulsant effects [9–12]. In particular, several phenolic compounds, including gastrodin, gastrodigenin, and vanillin, have been shown to protect neuronal cells and restore the function of the brain by suppressing oxidative stress and inflammatory responses [13].

These results have encouraged further studies in an attempt to discover natural neuroprotective molecules from the rhizomes of *G. elata*. Nineteen compounds, including two new phenolic compounds (1 and 2) and a new tripeptide (3), were isolated and structurally characterized from *n*-hexane-, ethyl acetate (EtOAc)-, and water (H<sub>2</sub>O)-soluble fractions of the rhizomes of *G. elata*. The neuroprotective and optic nerve-protective effects of all isolated compounds were examined against glutamate- or  $H_2O_2$ -induced toxicity in HT22 hippocampal neuronal and R28 retinal cells, respectively. In addition, all obtained compounds were also assessed for their anti-neuroinflammatory effects in BV2 microglial cells through inhibiting LPS-induced NO production.

Herein, this paper describes the structural identification of three new compounds (1–3) and biological assessment of the compounds associated with the potential of *G. elata* in the treatment of neurodegenerative disorders.

#### 2. Results and Discussion

## 2.1. Isolation and Identification of Compounds from the Rhizomes of G. elata

Compound 1 was isolated as a yellowish amorphous powder, and its molecular formula,  $C_{20}H_{18}O_3$ , was determined by high-resolution mass spectrometry (HRMS) data (Figure S8). The ultraviolet (UV) spectrum (Figure S6) exhibited maximum absorption at 227 and 281 nm, and the infrared (IR) spectrum (Figure S7) displayed absorption bands at 3334, 1228, 1105, and 1014 cm<sup>-1</sup>, indicating that 1 is a phenolic compound with hydroxy and ether groups. The <sup>1</sup>H nuclear magnetic resonance (NMR) data (Table 1 and Figure S1) showed three pairs of AA'BB' type signals at  $\delta_H$  7.28 (2H, d, *J* = 8.5 Hz, H-2, H-6), 6.84 (2H, d, *J* = 8.5 Hz, H-3, H-5), 7.10 (2H, d, *J* = 8.5 Hz, H-2', H-6'), 6.89 (2H, d, *J* = 8.5 Hz, H-3', H-5'), 7.02 (2H, d, *J* = 8.0 Hz, H-2", H-6"), and 6.74 (2H, d, *J* = 8.0 Hz, H-3", H-5"), as well as two methylene signals at  $\delta_H$  4.94 (2H, s, H-7) and 3.80 (2H, s, H-7'). The <sup>13</sup>C NMR data (Table 1 and Figure S2) displayed 20 carbon signals, suggesting the presence of three sets of 1,4-disubstituted phenol groups. These assignments were supported by a detailed analysis of the two-dimensional (2D) NMR spectra (Figures S3 and S5). The heteronuclear multiple bond correlation (HMBC) cross-peaks from H-7 to C-2 ( $\delta_C$  130.3) and C-4' ( $\delta_C$  158.2), as well as the cross-peaks from H-7' to C-2' ( $\delta_C$  130.4) and C-2" ( $\delta_C$  130.5), indicated the location of each phenol group. Consequently, the structure of compound 1 was determined to be 4-[[4-(4-hydroxybenzyl)phenoxy]methyl]phenol (Figure 1).

Compound **2** was isolated as a yellowish oil. The molecular formula,  $C_{16}H_{18}O_3$ , was determined by HRMS data. The <sup>1</sup>H and <sup>13</sup>C NMR data of **2** (Table 1) were superimposable to those of the already reported compound, 4-hydroxy-3-(4-hydroxybenzyl)benzyl methyl ether [14], except for the replacement of a methoxy group at C-7' by an ethoxy group. The <sup>1</sup>H NMR signals of the ethoxy group appeared at  $\delta_H$  3.57 (2H, q, *J* = 7.0 Hz, H-1") and 1.24 (3H, t, *J* = 7.0 Hz, H-2"). Its connectivity was demonstrated by the HMBC cross-peak from H-1" to C-7' ( $\delta_C$  72.8). Furthermore, an additional interpretation of the 2D NMR suggested that compound **2** is 4-(ethoxymethyl)-2-(4-hydroxybenzyl) phenol.

Position	1 (CD <sub>3</sub> COCD <sub>3</sub> )		2 (CDCl <sub>3</sub> )	
	δC	$\delta_{ m H}$	$\delta_{\rm C}$	$\delta_{ m H}$
1	129.2		131.6	
2	130.3	7.28, d (8.5)	129.9	6.99, d (8.5)
3	116.0	6.84, d (8.5)	115.6	6.62, d (8.5)
4	158.1		154.3	
5	116.0	6.84, d (8.5)	115.6	6.62, d (8.5)
6	130.3	7.28, d (8.5)	129.9	6.99, d (8.5)
7	70.4	4.94, s	35.6	4.41, s
1′	135.1		130.2	
2′	130.4	7.10, d (8.5)	131.1	7.08, d (1.5)
3'	115.6	6.89, d (8.5)	127.6	
4'	158.2		153.7	
5'	115.6	6.89, d (8.5)	115.8	6.67, d (8.0)
6'	130.4	7.10, d (8.5)	128.0	7.06, dd (8.0, 1.5)
7′	40.7	3.80, s	72.8	3.85,s
1″	133.5		65.9	3.57, q (7.0)
2″	130.5	7.02, d (8.0)	15.3	1.24, t (7.0)
3″	116.0	6.74, d (8.0)		
4″	156.5	,		
5″	116.0	6.74, d (8.0)		
6″	130.5	7.02, d (8.0)		

Table 1. Nuclear magnetic resonance (NMR) spectroscopic data for compounds 1 and 2.

Compound **3** was isolated as a pale yellowish amorphous powder. The UV and IR data of **3** were almost the same as those of **17**, indicating that the compounds possess a similar skeleton, *S*-(4-hydroxybenzyl)glutathione. The <sup>1</sup>H NMR data of **3** were superimposable to that of **17**, with the only difference being the presence of an additional ethoxy group at  $\delta_H$  4.16 (2H, q, *J* = 7.0 Hz) and 1.25 (3H, t, *J* = 7.0 Hz), as further supported by HRMS data. Furthermore, the basic skeleton was reconfirmed by <sup>1</sup>H–<sup>1</sup>H correlation spectroscopy (COSY) and HMBC data. In particular, the connectivity between glutathione and 4-hydroxybenzyl groups was established by the HMBC correlation from H-3' to C-7''' ( $\delta_C$  36.5), suggesting that the 4-hydroxybenzyl group was connected to the sulfur atom of the cysteinyl unit, and the location of the ethoxy group was determined by the HMBC correlation from the methylene proton signal at  $\delta_H$  4.16 to C-1'' ( $\delta_C$  171.1). Therefore, the structure of compound **3** was elucidated as ethyl *S*-(4-hydroxybenzyl) glutathione.

The previously reported compounds were identified to be 4-[[4-(ethoxymethyl)phenoxy] methyl]phenol (4) [15], gastrol A (5) [16], bis(4-hydroxyphenyl)methane (bisphenol F) (6) [10], 4-hydroxybenzyl vanillyl ether (7) [17], bis(4-hydroxybenzyl)ether (8) [18], 2,4-bis(4-hydroxybenzyl)phenol (9) [10], gastrodigenin (10) [18], 4-hydroxybenzyl ethyl ether (11) [19], gastrodin (12) [18], 4-hydroxybenzaldehyde (13) [18], 3,5-dimethoxybenzoic acid-4-*O*- $\beta$ -D-glucopyranoside (14) [20], parishin E (15) [19], adenosine (16) [21], *S*-(4-hydroxybenzyl)glutathione (17) [22], palmitic acid ethyl ester (18) [23], and linoleic acid ethyl ester (19) [23] by comparing the UV, IR, and NMR spectroscopic data with published values.

Although ethylated metabolites have been reported in *G. elata*, as in the example of compounds **4** and **11**, compound **2** may represent artifacts produced during the ethanol (EtOH) extraction process [8,22]. However, the purpose of this study is to evaluate the value of the EtOH extract of *G. elata* as a neuroprotective agent by identifying bioactive compounds extracted from EtOH. Therefore, if effective and reproducible, ethylated constituents of the *G. elata* EtOH extract may also be suitable for the purposes of this study.



Figure 1. Chemical structures of compounds 1–19 isolated from the rhizomes of G. elata.

# 2.2. Neuroprotective and NO Inhibitory Effects of Compounds from the Rhizomes of G. elata

To discover potent neuroprotective compounds, all obtained compounds **1–19**, isolated from *G. elata*, were assessed for their activities against glutamate-induced toxicity in HT22 cells (Table S1). The HT22-immortalized mouse hippocampal neuronal cell line has commonly been used to evaluate the ability of compounds to protect neuronal cells [24]. *N*-acetyl-L-cysteine (NAC) was used as a positive control. Among the 19 compounds, compounds **4** and **5** showed more than 90% cell viability and were further analyzed for dose-dependency using a six-point concentration series. Even though cell viability was decreased to 60% with 5 mM of glutamate, treatment with a concentration ranging from 5.6 to 0.6  $\mu$ M of **4**, or a concentration ranging from 16.7 to 0.6  $\mu$ M of **5**, was associated with a protective effect of ≥90%, measured as cell viability, against glutamate-induced toxicity (Figure 2). Both an effective concentration and cell viability are regarded as important factors in a cell protection assay.

R28 retinal cells were used to investigate protective compounds for optical neurons [25]. Among the isolated compounds, three compounds (**2**, **6**, and 7) were selected through a screening system (Table S2) to be further studied using a six-point concentration series. NAC was used as a positive control. After the addition of 300  $\mu$ M of H<sub>2</sub>O<sub>2</sub>, cell viability decreased to 50% due to oxidative stress-induced cell death but was restored to 60–70% after treatment with with 50 to 16.6  $\mu$ M of either **2**, **6**, or **7** (Figure 3).



**Figure 2.** Protective effect of **4** and **5** against glutamate-induced toxicity in HT22 cells (\*\* P < 0.05, and \*\*\* P < 0.0002).



**Figure 3.** Protective effects of **2**, **6**, and **7** against  $H_2O_2$ -induced toxicity in R28 cells (\*\* P < 0.005).

Besides, the inhibitory effects of compounds isolated from the EtOH extract of *G. elata* on NO production were evaluated in LPS-treated murine BV2 microglial cells at a concentration ranging from 0.02 to 50.0  $\mu$ M (Table S3). The cytotoxicity of these compounds was measured using the MTT assay. Inhibition of NO production in LPS-induced microglial cells is considered an effective approach to alleviating the inflammatory response in neurons [26]. Only compound **9** showed moderate NO inhibitory activities with IC<sub>50</sub> 38.98  $\mu$ M, while **9** showed no cytotoxicity at all concentrations treated in the cell lines (50 to 0.2  $\mu$ M) (Figure 4).



Figure 4. Inhibitory effect of 9 against LPS-induced nitric oxide production in BV2 cells (\*\*\* *P* < 0.001).

Although the compound **6** possesses potential bioactivity for neuroprotection, its neurotoxicity toward a zebrafish embryos was reported [27], so this potential toxicity should be considered for the use of *G. elata* extract or bisphenol F (**6**). However, the toxicity of bisphenol F (**6**) might be much weaker than that of bisphenol A. Bisphenol A could cause developmental abnormalities in the neuronal system through possibly inducing degradation of hypoxia-inducible factor  $1\alpha$  (HIF- $1\alpha$ ). A recent report has

documented that the HIF-1 $\alpha$ -related toxicity of bisphenol A disappeared by the removal of two central methyl groups in bisphenol A to obtain bisphenol F [28].

Although the neuroprotective effect of gastrodin (12) on glutamate-treated PC12 cells and  $H_2O_2$ -treated SH-SY5Y cells was reported, it was not effective in our toxicity-inducing model using glutamate in HT22 cells or  $H_2O_2$  in R28 cells [29,30]. Gastrodin, a major phenolic component of *G. elata*, can prevent hypoxia-induced neurotoxicity in cultured rat cortical neurons [31] and has shown antioxidative and neuroprotective effects in a transient focal brain ischemia model in the rodent [32]. In our bioassay model for LPS-induced nitric oxide inhibition, gastrodin showed no activity at 50  $\mu$ M. However, it is reported that the treatment of gastrodin (40  $\mu$ M and 60  $\mu$ M) can inhibit LPS-induced enhancement of expression of the iNOS in microglial cells [33]. Although 4-hydroxybenzaldehyde (13) was not effective in our neuroprotection assay either, it possesses an antioxidative effect on brain lipid peroxidation in pentylenetetrazole-induced convulsions in rats [34] and can inhibit glutamate-induced apoptosis in IMR-32 neuroblastoma cells [35].

The neuroprotective effects of compounds **2**, **4**, **5**, **6** and **7** on  $H_2O_2$  or glutamate-induced toxicity in hippocampal and retinal cells are first reported in this study. The inhibition of **9** on LPS-induced nitric oxide production was not reported in the previous literature either. This present study was limited to the discovery of additional neuroprotective compounds derived from *G. elata*. In order to classify the various neuroprotective constituents of *G. elata* according to their biological activity range and investigate the potentials for drug development, additional structure-activity relationship (SAR), mechanism of action (MOA) and in vivo efficacy studies of bioactive compounds from this study should be conducted.

#### 3. Materials and Methods

#### 3.1. General Procedures

UV spectra were obtained on a Spectramax M5 (Molecular Devices). IR spectra were obtained using a FT/IR-4200 spectrometer (Jasco). HRMS data were acquired using a Waters xevo G2 quadrupole time-of-flight (QTOF) mass spectrometer equipped with electrospray ionization. NMR experiments were conducted on a Varian 500 MHz and a Jeol 500 MHz using tetramethylsilane or solvent residues as reference signals, and the chemical shifts were expressed as  $\delta$  values. Thin-layer chromatography (TLC) analyses were performed on silica gel 60 F<sub>254</sub> plates and RP-18 F<sub>2545</sub> plates (0.25 mm, Merck), and compounds were visualized by UV light (254 and 365 nm) and 20% (v/v) H<sub>2</sub>SO<sub>4</sub> reagent (Sigma-Aldrich). Silica gel (70–230 or 230–400 mesh, Merck), Sephadex LH-20 (Amersham Pharmacia Biotech), and Diaion HP-20 (Mitsubishi Chemical Co.) were used for column chromatography. Prepacked cartridges, Redi Sep-C<sub>18</sub> (13 g, 26 g, 43 g, 130 g, Teledyne Isco), were used for flash chromatography. Semi-preparative high-performance liquid chromatography (HPLC) was performed using a Gilson 321 pump and a Gilson 155 UV/VIS detector, with a YMC Pack ODS-A column (5 µm, 250 × 20 mm i.d.). All solvents used for the chromatographic separations were distilled before use.

#### 3.2. Plant Material

The rhizomes of *G. elata* Blume (Orchidaceae) were obtained from a domestic Korean market (Muju Firefly Market, Muju-gun, Jeollabuk-do), in April 2015. The origin of the herbal material was identified by Prof. Dae Sik Jang and a voucher specimen (GAEL1-2015) has been deposited in the Lab. of Natural Product Medicine, College of Pharmacy, Kyung Hee University, Republic of Korea.

#### 3.3. Extraction and Isolation

The rhizomes of fresh *G. elata* (17.4 kg) were extracted with EtOH (35 L) twice at room temperature. The EtOH extract was partitioned with *n*-hexane (18 L  $\times$  2) and EtOAc (18 L  $\times$  2) to yield *n*-hexane (5.8 g), EtOAc- (36.3 g), and H<sub>2</sub>O-soluble extracts (500.0 g), respectively.

The *n*-hexane-soluble extract (5.8 g) was subjected to silica gel column chromatography (CC) (70–230 mesh,  $\emptyset$  4.5 × 45.0 cm) using *n*-hexane–EtOAc–methanol (MeOH) mixtures (19:1:0 to 0:0:1, v/v) as the mobile phase to produce 18 fractions (H1 to H14). Fraction H2 (1.1 g) was further fractionated using silica gel CC (230-400 mesh,  $\emptyset$  3.5 × 27.0 cm) eluted with *n*-hexane–EtOAc (49:1 to 0:1, v/v) to produce compounds **18** (135.5 mg, yield 0.0249% by weight of the ethanol extract) and **19** (15.3 mg, yield 0.0028%).

The EtOAc-soluble extract (36.3 g) was chromatographed over silica gel CC (70–230 mesh,  $\emptyset 4.7 \times 45.0$  cm) with dichloromethane (CH<sub>2</sub>Cl<sub>2</sub>)–MeOH mixtures (19:1 to 0:1,  $\nu/\nu$ ) to produce 15 fractions (E1 to E15). Fraction E2 (7.1 g) was subjected to silica gel CC (230–400 mesh,  $\phi$  4.9 × 34.9 cm) with *n*-hexane–EtOAc mixtures (7:3 to 3:2, v/v) to produce 14 subfractions (E2-1 to E2-14). Subfraction E2-3 (4.7 g) was further fractionated to 13 subfractions (E2-3-1 to E-2-3-13) using silica gel (230–400 mesh, ø 4.3 × 37.9 cm) with *n*-hexane–EtOAc mixtures (7:3 to 0:1, *v*/*v*). Subfractions E2-3-4 (2.1 g) and E2-3-5 (75.1 mg) were separated by a flash chromatography system using Redi Sep- $C_{18}$  cartridges (130 g, MeOH-H<sub>2</sub>O, 3:7 to 7:3, v/v; 26 g, MeOH-H<sub>2</sub>O, 3:7 to 4:1, v/v), respectively, to obtain compound 4 (121.4 mg, yield 0.0223%). Subfraction E2-6 (570.1 mg) was separated by a flash chromatography system using Redi Sep-C<sub>18</sub> (43 g, MeOH–H<sub>2</sub>O, 3:7 to 7:3, v/v), followed by preparative HPLC with a YMC Pack ODS-A column to obtain compound 1 (10.7 mg, yield 0.0020%). Fraction E3 (2.4 g) was fractionated using silica gel CC (230–400 mesh,  $\emptyset$  3.7 × 26.3 cm) with *n*-hexane–EtOAc (7:3 to 3:2, v/v) to produce 13 subfractions (E3-1 to E3-13). Subfractions E3-2 (729.3 mg) and E3-5 (725.4 mg) were separated using a flash chromatography system and Redi Sep-C<sub>18</sub> cartridges (43 g, MeOH–H2O, 1:3 to 9:11, v/v; 130 g, MeOH–H<sub>2</sub>O, 3:7 to 1:1, v/v), respectively, to obtain compounds **2** (178.7 mg, yield 0.0328%), **6** (248.9 mg, 0.0457%), 11 (247.5 mg, 0.0454%), and 13 (14.1 mg, 0.00259%). Subfraction E3-7 (216.9 mg, yield 0.0398%) was separated on a flash chromatography system with Redi Sep-C<sub>18</sub> (26 g, MeOH–H2O, 1:3 to 1:1, v/v), followed by preparative HPLC with a YMC Pack ODS-A column, to yield compounds 5 (2.7 mg, yield 0.0005%), 7 (2.6 mg, 0.0005%), 8 (6.4 mg, 0.0012%), and 10 (13.7 mg, 0.0025%). Fraction E5 (1.8 g) was separated using silica gel CC (230–400 mesh,  $\emptyset$  3.6  $\times$  27.5 cm, *n*-hexane–EtOAc, 3:2 to 0:1, v/v) to produce 13 subfractions (E5-1 to E5-13). Subfractions E5-4 (118.6 mg) and E5-5 (187.5 mg) were separated through the flash chromatography system using a Redi Sep-C<sub>18</sub> cartridge (26 g, MeOH-H<sub>2</sub>O, 1:1 to 3:1, v/v) to obtain compounds 9 (68.9 mg, yield 0.0126%) and 10 (35.8 mg, 0.0066%), respectively.

The H<sub>2</sub>O-soluble extract (500.0 g) was chromatographed over Diaion HP-20 CC ( $\emptyset$  8.0 × 57.0 cm) and eluted with MeOH–H<sub>2</sub>O mixtures (0:1 to 1:0, v/v) to produce 18 fractions (W1 to W18). Fraction W5 (8.1 g) was further fractionated using Sephadex LH-20 ( $\emptyset$  5.0 × 53.5 cm) with MeOH–H<sub>2</sub>O mixtures (3:7 to 1:0, v/v), followed by preparative HPLC with a YMC Pack ODS-A column, resulting in the isolation of compounds **12** (8.9 mg, yield 0.0016%) and **15** (14.5 mg, 0.0027%). Fraction W8 (5.2 g) was further subjected to Sephadex LH-20 ( $\emptyset$  4.6 × 51.0 cm) with MeOH–H<sub>2</sub>O mixtures (2:3 to 1:0, v/v) to produce eight subfractions (W8-1 to W8-8). Subfraction W8-5 (1.46 g) was fractionated using silica gel CC (230–400 mesh,  $\emptyset$  3.6 × 26.0 cm) with EtOAc–MeOH–H<sub>2</sub>O mixtures (40:9:1 to 0:1:0, v/v) to produce nine subfractions (W8-5-1 to W8-5-9). Compound **16** (143.2 mg, yield 0.0263%) was purified by recrystallization with MeOH from subfraction W8-5-4 (28.4 mg). Subfractions W8-5-5 (73.3 mg) and W8-5-6 (62.5 mg) were separated by a flash chromatography system with Redi Sep-C<sub>18</sub> (13 g, MeOH–H<sub>2</sub>O, 0:1 to 7:3, v/v), followed by preparative HPLC with a YMC Pack ODS-A column, to obtain compounds **3** (3.6 mg, yield 0.0007%) and **14** (4.8 mg, 0.0009%). Subfraction W8-5-9 (790.0 mg) was purified using a flash chromatography system and Redi Sep-C<sub>18</sub> (43 g, MeOH–H<sub>2</sub>O, 0:1 to 1:1, v/v), and recrystallized with MeOH, resulting in the isolation of compound **17** (233.3 mg, yield 0.0428%).

#### 3.3.1. 4-[[4-(4-Hydroxybenzyl)phenoxy]methyl]phenol (1)

Yellowish amorphous powder; UV (MeOH);  $\lambda_{max}$  (log  $\varepsilon$ ) 227 (4.60), 281 (4.00) nm; IR (ATR)  $\nu_{max}$  3334, 1611, 1509, 1228, 1105, 1014, 820 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (500 and 125 MHz, CD<sub>3</sub>COCD<sub>3</sub>), see Table 1; HRESIMS (negative mode) *m*/*z* 305.1174 [M – H]<sup>-</sup> (calcd. for C<sub>20</sub>H<sub>17</sub>O<sub>3</sub>, 305.1178).

#### 3.3.2. 4-(Ethoxymethyl)-2-(4-hydroxybenzyl)phenol (2)

Yellowish oil; UV (MeOH);  $\lambda_{max}$  (log  $\varepsilon$ ) 229 (4.53), 280 (3.90) nm; IR (ATR)  $\nu_{max}$  3324, 1612, 1511, 1236, 1069, 820 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data (500 and 125 MHz, CDCl<sub>3</sub>), see Table 1; HRESIMS (negative mode)  $m/z = 257.1177 [M - H]^-$  (calcd. for C<sub>16</sub>H<sub>17</sub>O<sub>3</sub>, 257.1178).

## 3.3.3. Ethyl S-(4-hydroxybenzyl) Glutathione (3)

Pale yellowish amorphous powder;  $\lambda_{max}$  (log  $\varepsilon$ ) 228 (5.21), 279 (4.43) nm; IR (ATR)  $\nu_{max}$  3273, 2981, 2938, 1746, 1647, 1514, 1451, 1415, 1375, 1339, 1216, 1032, 835 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CD<sub>3</sub>OD)  $\delta$  1.25 (3H, t, *J* = 7.0 Hz, CH<sub>2</sub>CH<sub>3</sub>), 2.07 (1H, m, H<sub>2</sub>-3), 2.15 (1H, m, H<sub>2</sub>-3), 2.51 (2H, m, H<sub>2</sub>-4), 2.67 (1H, dd, *J* = 13.5, 9.0 Hz, H<sub>2</sub>-3'), 2.91 (1H, dd, *J* = 13.5, 5.5 Hz, H<sub>2</sub>-3'), 3.63 (1H, dd, *J* = 6.5, 6.5 Hz, H-2), 3.68 (2H, s, H<sub>2</sub>-7'''), 3.89 (1H, d, *J* = 17.5 Hz, H<sub>2</sub>-2''), 3.95 (1H, d, *J* = 17.5 Hz, H<sub>2</sub>-2''), 4.16 (2H, q, *J* = 7.0 Hz, CH<sub>2</sub>CH<sub>3</sub>), 4.55 (1H, dd, *J* = 9.0, 5.5 Hz, H-2'), 6.72 (2H, d, *J* = 8.5 Hz, H-3''', H-5'''), 7.15 (2H, d, *J* = 8.5 Hz, H-2'''), 42.1 (C-2''), 54.1 (C-2'), 55.3 (C-2), 62.3 (CH<sub>2</sub>CH<sub>3</sub>), 116.2 (C-3''', H-5'''), 130.0 (C-1'''), 131.2 (C-2''', C-6'''), 157.6 (C-4'''), 171.1 (C-1''), 173.6 (C-1'), 173.9 (C-1), 175.0 (C-5); HRESIMS (negative mode) *m*/*z* = 440.1487 [M – H]<sup>-</sup> (calcd. for C<sub>19</sub>H<sub>26</sub>N<sub>3</sub>O<sub>7</sub>S, 440.1491).

## 3.4. Cell Culture

HT22 hippocampal neuronal cells were purchased from the Korean Cell Line Bank (Seoul, South Korea). R28 retinal cells were purchased from Kerafast (Kerafast, Inc., Boston, MA, USA), and BV2 microglial cells were obtained from Prof. Choon-Gon Jang, Sungkyunkwan University (Suwon, South Korea).

HT22 hippocampal neuronal cells and R28 retinal cells were cultured in Dulbecco's modified Eagle's minimum essential medium (DMEM) supplemented with penicillin (100 U/mL), streptomycin (100  $\mu$ g/mL), and 10% fetal bovine serum (FBS) (Gibco) in a humidified atmosphere of 5% CO<sub>2</sub> at 37 °C. Cells were cultured every two days using 0.05% trypsin.

BV2 microglial cells were cultured in DMEM mixed with penicillin (100 U/mL), streptomycin (100  $\mu$ g/mL), and 10% FBS in a humidified atmosphere of 5% CO<sub>2</sub> at 37 °C. Cells were cultured every two days using a pipette for detaching cells.

## 3.5. Sample Preparation

The 10 mg/mL DMSO stock solutions using the compounds were prepared and serially diluted 3-fold with DMSO. The amount and concentration of DMSO were always maintained in all conditions, regardless of the concentrations of the compounds of interest. Using a liquid handling station equipped with a pin tool system, we typically transferred 500 nL of the compound in DMSO solution into 100  $\mu$ L of the medium, which resulted in a consistent DMSO concentration of 0.5%.

#### 3.6. Cell Viability Assay

Cell viability in HT22 and R28 cells was evaluated using the MTT assay. The HT22 cells were seeded at a density of  $3 \times 10^3$  cells/well in a 96-well plate and R28 cells were plated at  $10 \times 10^3$  cells/well in a 96-well plate. After incubation for 24 h, compounds including NAC (1 mM) were added over 2 h, followed by the addition of 5 mM of glutamate for 22 h. A total of 50 µL of EZ-Cytox reagent (Daeil Lab) was added over 2 h. The plate was measured at 450 nm using a UV/VIS plate reader [36,37].

Cytotoxicity of compounds against the BV2 cells was determined using the MTT assay. The cells were added at the density of  $3 \times 10^4$  cells/well in a 96-well plate. After incubating for 24 h, the cells were treated with compounds and incubated for 24 h. After 24 h, 50 µL of media was removed and 50 µL of EZ-Cytox reagent was added over 1 h, and the plate was measured at 450 nm using a UV/VIS plate reader.

#### 3.7. NO Production Assay

BV2 microglial cells were seeded at the density of  $3 \times 10^4$  cells/well in a 96-well plate. The cells were treated with compounds for 1 h before exposure of 1 µg/mL of LPS. After 24 h incubation with compound and LPS, nitrite in culture media was measured to assess NO production in BV2 cells using Griess reagent. In a 96-well plate, 50 µL of aliquots were mixed with 50 µL of Griess reagent (1% sulfanilamide, 0.1% naphtylethylenediamine dihydrochloride in 2% phosphoric acid) and incubated at room temperature for 15 min.; the plate was measured at 544 nm using a UV/VIS plate reader.

# 4. Conclusions

Three new compounds (1–3), together with 16 known compounds (4–19), were isolated and structurally characterized using chromatographic and spectroscopic methods. Among the compounds, phenolic compounds 4 and 5 were found to possess moderate neuroprotective activities in HT22 cells, while phenolic compounds 2, 6, and 7 exhibited weak activities in R28 cells. Compound 9 showed a moderate inhibitory effect on nitric oxide production in LPS-treated BV2 microglial cells.

Previous reports have revealed that phenolic compounds isolated from *G. elata* can protect neuronal cells through antioxidative, anti-inflammatory, anticonvulsive, neuroprotective, and circulatory system modulating effects, etc [8]. However, the investigation of other neuroprotective compounds needs to be evaluated to provide clarification for potential antagonistic or synergistic effects of multi-components contained in *G. elata*. Therefore, our investigation could provide additional information of neuroprotective compounds on hippocampal neuronal, retinal precursor, and microglial cells, which can contribute to the development of therapeutic agents for neuronal disease using phenolic compound mixtures derived from *G. elata*.

**Supplementary Materials:** The following are available online at http://www.mdpi.com/2223-7747/9/8/1051/s1, S1: The <sup>1</sup>H NMR spectral data of compounds **4–19**, Table S1: The screening of all isolated compounds for protective effects against HT22 cell death, Table S2: The screening of all isolated compounds (50 μM) for protective effects on R28 cell death, Table S3: The screening of all isolated compounds for inhibitory effects of nitric oxide production on LPS-treated BV2 cell lines. Figures S1–S8: The <sup>1</sup>H, <sup>13</sup>C, COSY, HSQC, and HMBC NMR, UV, IR, and HRESIMS spectrum of compound **1**, Figures S1–S24: The <sup>1</sup>H, <sup>13</sup>C, COSY, HSQC, and HMBC NMR, UV, IR, and HRESIMS spectrum of compound **2**, Figures S17–S24: The <sup>1</sup>H, <sup>13</sup>C, COSY, HSQC, and HMBC NMR, UV, IR, and HRESIMS spectrum of compound **3**.

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