



Article

Glucose Tolerance-Improving Activity of Helichryoside in Mice and Its Structural Requirements for Promoting Glucose and Lipid Metabolism

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Abstract: An acylated flavonol glycoside, helichryoside, at a dose of 10 mg/kg/day per os for 14 days, improved the glucose tolerance in mice without affecting the food intake, visceral fat weight, liver weight, and other plasma parameters. In this study, using hepatoblastoma-derived HepG2 cells, helichryoside, *trans*-tiliroside, and kaempferol 3-*O*- β -D-glucopyranoside enhanced glucose consumption from the medium, but their aglycones and *p*-coumaric acid did not show this activity. In addition, several acylated flavonol glycosides were synthesized to clarify the structural requirements for lipid metabolism using HepG2 cells. The results showed that helichryoside and related analogs significantly inhibited triglyceride (TG) accumulation in these cells. The inhibition by helichryoside was more potent than that by other acylated flavonol glycosides, related flavonol glycosides, and organic acids. As for the TG metabolism-promoting activity in high glucose-pretreated HepG2 cells, helichryoside, related analogs, and their aglycones were found to significantly reduce the TG contents in HepG2 cells. However, the desacyl flavonol glycosides and organic acids derived from the acyl groups did not exhibit an inhibitory impact on the TG contents in HepG2 cells. These results suggest that the existence of the acyl moiety at the 6'' position in the D-glucopyranosyl part is essential for glucose and lipid metabolism-promoting activities.

Keywords: helichryoside; acylated flavonol glycoside; glucose tolerance-improving activity; lipid metabolism-promoting activity

1. Introduction

Flavonoids are one of the most abundant classes of secondary plant metabolites. Flavonoids are biosynthesized by the shikimate and acetate-malonate pathways and are comprised of compounds that possess a common C₆-C₃-C₆ skeleton, where two aromatic rings (named ring A and B) are linked via a heterocyclic 4H-pyran ring (ring C). Modification of the 15-carbon skeleton through different oxidation levels and substituents to ring C gives rise to different classes of flavonoids, such as flavones, flavonols, flavanones, chalcones, dihydroflavonols (flavanonols), isoflavones, aurones, anthocyanidins,

leucoanthocyanidines (flavan-3,4-diols), and flavan-3-ols. They naturally occur in not only aglycone forms, but also as glycosylated and/or acylated derivatives and oligomeric and polymeric structures, such as the flavan-3-ol-derived condensed tannins and proanthocyanidins [1–5]. Flavonoid health benefits are well-recognized, such as their antioxidant properties, properties for weight management, cardiovascular disease protection, anti-allergic activity, vascular fragility, prevention of viral and bacterial infections, anti-inflammatory activity, age-related neurodegenerative disease prevention, anti-platelet aggregation effects, and cancer protection, etc. [2–8]. Our studies on bioactive constituents from medicinal and/or food resources have reported several bio-functional properties of flavonoids. These included aldose reductase inhibitory [9–12], anti-platelet aggregation [9], anti-allergic [12–14], anti-inflammatory [12,15–18], aminopeptidase N inhibition [11,17,19], hepatoprotective [20], gastroprotective [21], melanogenesis inhibition [22], and dipeptidyl peptidase-IV inhibitory [23] activities. This paper deals with the practical synthesis and glucose tolerance-improving activity of helichryoside (**1** = quercetin 3-*O*-(6''-*O*-*trans*-*p*-coumaroyl)- β -D-glucopyranoside), isolated from *Helichrysum kraussii* and *H. stoechas* [24,25] by other research groups. Furthermore, synthetic studies of the related analogs of **1** (**2**–**15**, Figure 1) were also carried out, as well as characterization of its structural requirements for glucose and lipid metabolism in HepG2 cells. In our previous report, an acylated flavonol glycoside, *trans*-tiliroside (**16** = kaempferol 3-*O*-(6''-*O*-*trans*-*p*-coumaroyl)- β -D-glucopyranoside), isolated from the fruit of *Rosa canina*, was found to suppress visceral fat weight gain and improve glucose tolerance in mice [26]. The structures of **1** and **16** are similar: the former has a *p*-coumaroyl ester at the 6-position in the β -D-glucopyranosyl moiety of quercetin 3-*O*- β -D-glucopyranoside (isoquercitrin, **17**), while the latter has the common acyl group at the same position of kaempferol 3-*O*- β -D-glucopyranoside (**18**).

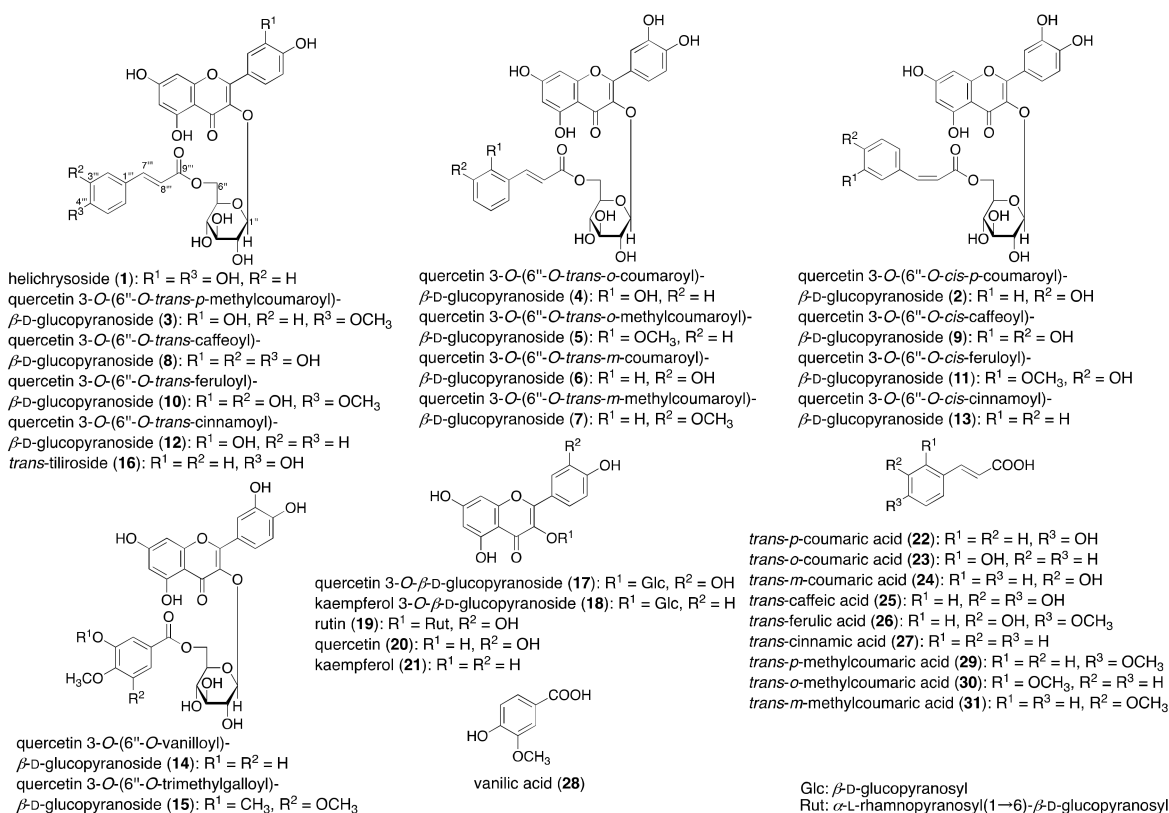


Figure 1. Structures of acylated flavonol glycosides (**1**–**16**) and related compounds (**17**–**31**).

2. Results and Discussion

2.1. Synthesis of Acylated Flavonol Glycosides (1–15)

Rutin (**19** = quercetin 3-*O*- α -L-rhamnopyranosyl(1 \rightarrow 6)- β -D-glucopyranoside), constructed with quercetin (**20**) as an aglycone, is one of the most widely distributed naturally occurring flavonoids and has been reported to have several pharmacological activities, such as anti-oxidant, anti-inflammatory, anti-diabetic, anti-adipogenic, and neuroprotective effects, and has been used in hormone therapy [27–30]. In order to achieve the practical synthesis of **1** from **19**, the most inexpensive and commercially available flavonoid, the optimal conditions for enzymatic hydrolysis of the terminal rhamnosyl part were investigated. Therefore, the practical derivation from **19** to **17** was carried out using naringinase (from *Penicillium decumbens*) under an optimal pH and temperature (pH 7 and 50 °C), and the time course of the reaction mixture was monitored by high performance liquid chromatography (HPLC) analysis (Figure S1 and Table S1). As shown in Table 1, the highest content of **17** in the reaction mixture of 2 h was observed. By applying these conditions, a large-scale derivation of **17** (6.50 g and 14.0 mmol, 56.9%) from **19** (15.0 g and 24.6 mmol) was achieved.

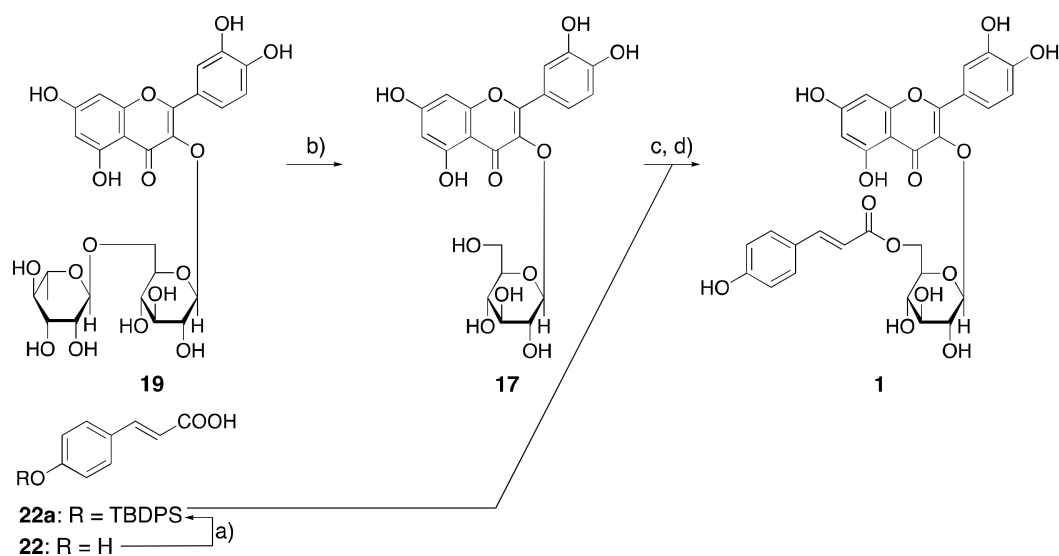
Table 1. Peak area ratio of rutin (**19**), quercetin 3-*O*- β -D-glucopyranoside (**17**), and quercetin (**20**) in the reaction mixture.

Reaction Time	Peak Area (%)		
	Rutin (19)	Quercetin 3- <i>O</i> -Glc (17)	Quercetin (20)
0 min	99.1	0.0	0.0
5 min	96.6	1.6	0.7
30 min	60.8	30.7	7.5
1 h	29.7	46.3	22.9
1.5 h	16.4	50.1	32.6
2 h	6.3	49.6	43.2
3 h	0.0	40.8	57.7
4.5 h	0.0	25.9	72.5
8 h	0.0	9.0	89.4
24 h	0.0	1.5	95.5

Linearities for **19**, **17**, and **20** in the HPLC analytical condition were calculated as shown in Table S1.

Synthesis of the acylated flavonol glycosides, including helichryoside (**1–15**) from **17**, using the corresponding acylation reactions, was carried out. Therefore, protection of a phenol group in *p*-coumaric acid (**22**) with *tert*-butyldiphenylsilyl chloride (TBDPSCI) yielded the corresponding silyl ether, **22a**. Acylation of **17** with **22a** in the presence of 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (EDC·HCl) and 4-dimethylaminopyridine (4-DMAP) in pyridine, followed by deprotection with tetrabutylammonium fluoride (TBAF), provided **1** with a 36.9% yield (Scheme 1).

In a similar procedure to that of **1** [31,32], compounds **3**, **4**, **5**, **6**, **7** [33], **8** [34], **10** [33], **12** [33,35,36], **14**, and **15** were synthesized with **17** and the corresponding organic acids. The *cis*-isomer of **1**, quercetin 3-*O*-(6''-*O*-*cis*-*p*-coumaroyl)- β -D-glucopyranoside (**2**), was derived under a UV lamp in a methanol solution of **1**. Using a similar procedure, quercetin 3-*O*-(6''-*O*-*cis*-caffeoyl)- β -D-glucopyranoside (**9**), quercetin 3-*O*-(6''-*O*-*cis*-feruloyl)- β -D-glucopyranoside (**11**), and quercetin 3-*O*-(6''-*O*-*cis*-cinnamoyl)- β -D-glucopyranoside (**13**) were also isomerized from **8**, **10**, and **12**, respectively. Among these synthetic products, known compounds (**1**, **7**, **8**, **10**, and **12**) were identified by a comparison of their physicochemical data with those of authentic samples or with reported values. The structural determination of new compounds (**2–6**, **9**, **11**, and **13–15**) was elucidated on their spectroscopic properties, including the ¹³C-NMR data, as shown in Table S2.



Scheme 1. Reagents and conditions: (a) *tert*-butyldiphenylsilyl chloride (TBDPSCI), imidazole/*N,N*-dimethylformamide (DMF), 40 °C, 16 h, 76.5%; (b) naringinase/H₂O, 50 °C, 2 h, 56.9%; (c) **22a** (1.2 eq), 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide hydrochloride (EDC·HCl), 4-dimethylaminopyridine (4-DMAP)/pyridine, 50 °C, 12 h; (d) tetrabutylammonium fluoride (TBAF)/tetrahydrofuran (THF), r.t., 1 h, 36.9% (two steps from **17**).

2.2. Effect of Helichrysin (1) on the Liver Triglyceride (TG) Content and Glucose Tolerance Test after 14 Days of Administration in Mice

Diabetes is characterized by a high incidence of cardiovascular disease and poor control of hyperglycemia caused by insulin resistance (IR). IR can be defined as the inability of insulin to stimulate glucose uptake into the liver, skeletal muscle, or adipose tissue. Hyperglycemia is an important factor contributing to the development of atherosclerosis, and is relevant to the pathophysiology of late diabetic complications. Therefore, improving IR may form part of the strategy for the prevention and management of cardiovascular disease in diabetes [37,38]. We have reported that several anti-diabetogenic therapeutic candidates obtained from natural resources, such as acylated flavonol glycosides from *Sinocrassula indica* [39]; saponins from *Borassus flabellifer* [40]; and thiosugars from *Salacia reticulata*, *S. oblonga*, and *S. chinensis* [41–46], showed the inhibition of postprandial hyperglycemia and/or improvement of glucose tolerance in sugar-loaded animal models. As mentioned above, the structure of **16** isolated from *R. canina* [26] is quite similar to **1**, so we presumed that **1** also exhibits similar anti-diabetogenic activity to **16** in an in vivo study. To continue our search for new candidates of the anti-diabetogenic and/or anti-diabetic principles and to evaluate the anti-diabetogenic effect of **1**, the effect of 14 days of the continuous administration of **1** on glucose tolerance was performed in mice. Following this continuous administration, **1** was found to significantly suppress the increase in blood glucose levels at doses of 1 and 10 mg/kg/day per os (p.o.), at 60 min post glucose loading (Figure 2 and Table S3). The area under the curve (AUC) of blood glucose levels was significantly reduced at the dose of 10 mg/kg/day (p.o.). As indicated in Table S4, the continuous administration of **1** tended to reduce the weights of visceral fat and the liver and the liver TG content, without affecting the food intake and other plasma parameters, including plasma TG, total cholesterol, and free fatty acids (FFA).

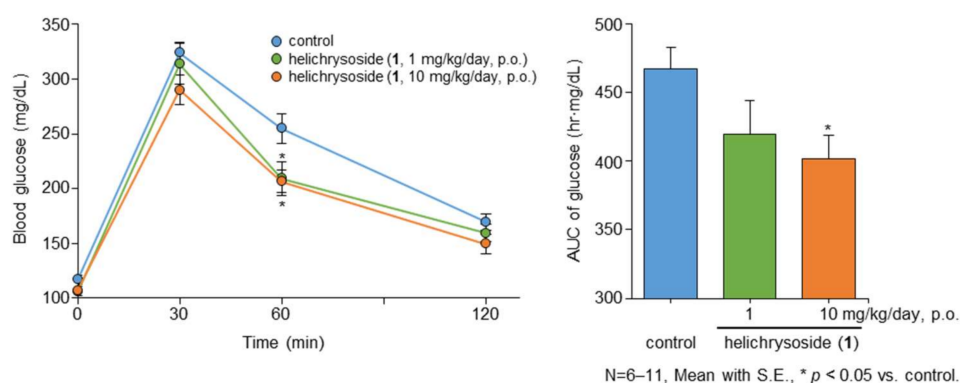


Figure 2. Effect of helichryoside (1) on the glucose tolerance test after 14 days of administration in mice.

2.3. Effects on Glucose Consumption in HepG2 Cells

The liver is one of the tissues important for maintaining blood glucose homeostasis and greatly affects the formation of abnormal glucose tolerance [47]. Since the improving activity of helichryoside (1) in terms of glucose tolerance in mice was observed in the previous section, we investigated the effects of 1 and its related compounds (16–18 and 20–22), to clarify the structural requirement of glucose consumption using human hepatoblastoma-derived HepG2 cells. As shown in Table 2, the glucose concentration in the medium was found to be significantly reduced at 6 days pretreatment with 1, *trans*-tiliroside (16), kaempferol 3-*O*- β -D-glucopyranoside (18), and metformin. On the other hand, the desacyl derivative of 1, quercetin 3-*O*- β -D-glucopyranoside (17); the aglycones of 1 and 16, quercetin (20) and kaempferol (21); and *trans-p*-coumaric acid (22) did not result in changes in the glucose concentration in the medium. These results suggested that the *p*-coumaroyl moiety at the 6'' position in the D-glucopyranosyl part was essential for promoting glucose consumption. Recent related studies have reported that compounds 17, 20, and 21 promoted glucose uptake into muscle and hepatocytes [48–50]. Due to the long-term treatment of test samples of cells in our study, the treatment with compounds 17, 20, and 21 showed cytotoxicity at the concentration of 30–100 μ M.

Table 2. Effects of helichryoside (1) and related compounds (16–18 and 20–22) on glucose consumption in HepG2 cells.

Treatment	Glucose in the Medium (% of Control)(Protein (% of Control))				
	0 μ M	3 μ M	10 μ M	30 μ M	100 μ M
Helichryoside (1)	100.0 \pm 2.6 (100.0 \pm 5.4)	94.7 \pm 1.1 ** (96.6 \pm 2.0)	86.7 \pm 1.2 ** (98.3 \pm 2.0)	89.3 \pm 1.1 ** (102.9 \pm 1.5)	82.9 \pm 1.5 ** (105.2 \pm 1.4)
<i>Trans</i> -tiliroside (16)	100.0 \pm 6.4 (100.0 \pm 4.7)	84.6 \pm 0.9 ** (105.4 \pm 1.5)	90.1 \pm 0.6 ** (102.3 \pm 2.7)	96.8 \pm 1.7 (103.4 \pm 1.9)	79.8 \pm 1.0 ** (110.0 \pm 1.0)
Quercetin 3- <i>O</i> -Glc (17)	100.0 \pm 1.1 (100.0 \pm 1.3)	100.9 \pm 1.4 (104.0 \pm 1.0)	100.6 \pm 1.3 (104.1 \pm 0.9)	241.7 \pm 3.8 ** (47.2 \pm 0.6 **)	269.4 \pm 5.2 ** (23.1 \pm 1.4 **)
Kaempferol 3- <i>O</i> -Glc (18)	100.0 \pm 3.0 (100.0 \pm 3.5)	95.9 \pm 0.7 (96.4 \pm 2.0)	98.9 \pm 2.7 (94.2 \pm 1.4 *)	93.4 \pm 0.4 ** (100.1 \pm 0.4)	92.8 \pm 1.1 ** (98.7 \pm 1.4)
Quercetin (20)	100.0 \pm 2.0 (100.0 \pm 0.5)	98.6 \pm 0.4 (102.0 \pm 0.9)	97.2 \pm 1.7 (103.4 \pm 0.8 **)	105.6 \pm 2.0 (98.9 \pm 0.6)	249.2 \pm 3.9 ** (41.7 \pm 0.4 **)
Kaempferol (21)	100.0 \pm 3.1 (100.0 \pm 0.7)	98.4 \pm 2.2 (101.0 \pm 0.8)	95.2 \pm 1.5 (100.7 \pm 0.5)	98.1 \pm 0.9 (100.3 \pm 1.2)	224.4 \pm 3.1 ** (55.4 \pm 0.5 **)
<i>Trans-p</i> -coumaric acid (22)	100.0 \pm 3.6 (100.0 \pm 3.7)	96.9 \pm 1.5 (99.7 \pm 0.6)	96.5 \pm 0.5 (98.2 \pm 1.7)	96.8 \pm 1.7 (98.3 \pm 0.4)	97.7 \pm 1.8 (105.8 \pm 1.8 *)
	0 μ M	62.5 μ M	125 μ M	250 μ M	500 μ M
Metformin	100.0 \pm 1.5 (100.0 \pm 1.2)	98.9 \pm 1.9 (102.0 \pm 1.8)	95.1 \pm 0.7 (100.9 \pm 1.7)	97.7 \pm 1.8 (100.5 \pm 2.1)	87.5 \pm 4.8 * (100.6 \pm 2.3)

Each value represents the mean \pm S.E. ($n = 4$ or 8); asterisks denote significant differences from the control group, where * $p < 0.05$ and ** $p < 0.01$.

2.4. Effects on Lipid Metabolism in HepG2 Cells

A fatty liver is recognized as a significant risk factor for serious liver diseases [51,52]. A strong causal link has been identified between fatty liver diseases and hyperinsulinemia, caused by insulin resistance [53,54]. Therefore, a fatty liver is considered to be closely associated with obesity and type 2 diabetes [54]. In previous studies on the identification of anti-fatty liver principles from natural medicines, several flavonoids [55–58] were revealed to inhibit lipid accumulation in HepG2 cells. Similarly, we also reported that several megastigmanes [59], diterpenes [60], and limonoids [61] inhibited lipid metabolism in high glucose-pretreated HepG2 cells.

Intracellular TG accumulated in HepG2 cells via increasing the expression of lipogenesis-related proteins, such as sterol regulatory element-binding protein 1c (SREBP-1c) and fatty acid synthase (FAS), when cultured in high glucose-containing medium [55,62]. To characterize this phenomenon, we examined the inhibitory effects of the acylated flavonol glycosides (1–16) and related compounds (17–22, 24, 26, and 28–31) on (i) high glucose-induced TG accumulation in HepG2 cells and (ii) TG contents in high glucose-pretreated HepG2 cells.

As shown in Table 3, several acylated flavonol glycosides, such as helichryoside (1), quercetin 3-*O*-(6''-*O*-*cis*-*p*-coumaroyl)- β -D-glucopyranoside (2), quercetin 3-*O*-(6''-*O*-*trans*-*p*-methylcoumaroyl)- β -D-glucopyranoside (3), quercetin 3-*O*-(6''-*O*-trimethylgalloyl)- β -D-glucopyranoside (15), and *trans*-tiliroside (16), significantly inhibited high glucose-induced TG accumulation in HepG2 cells (% of control at 100 μ M: 1 (76.9 \pm 2.3%), 2 (80.4 \pm 1.2%), 3 (58.2 \pm 7.5%), 15 (85.8 \pm 4.1%), and 16 (82.3 \pm 3.0%)). In contrast, the other acylated flavonol glycosides (4–14), related flavonol glycosides (17–19), and organic acids, which related to the corresponding acyl groups (22–31), did not show significant inhibitory activity up to a concentration of 100 μ M. As for the inhibitory effects of the corresponding aglycones, quercetin (20, 45.7 \pm 0.4% at 100 μ M) and kaempferol (21, 25.5 \pm 1.4% at 100 μ M) showed stronger activity than that of the acylated flavonol glycosides, with cytotoxicity under the effective concentrations (data not shown). The structural requirements of the acylated flavonol glycosides were assessed and showed that (1) the acylated flavonol glycosides with a *p*-coumaroyl, *p*-methylcoumaroyl, or trimethylgalloyl moiety as the acyl group in the 6'' position of the D-glucopyranosyl part are essential for the activity, and (2) the glycoside structure contributes to reducing the cytotoxicity.

On the other hand, all the tested acylated flavonol glycosides (1–16) and their aglycones (20 and 21) were found to significantly inhibit the TG content in high glucose-pretreated HepG2 cells at a concentration of 100 μ M, as shown in Table 4. Specifically, the 6''-*O*-acylated quercetin 3-*O*- β -D-glucopyranosides structure, having a *trans*- and *cis*-*p*-coumaroyl (% of control at 10 μ M: 1 (82.9 \pm 1.3) and 2 (86.0 \pm 3.2%)), *trans*-*p*-methylcoumaroyl (3, 87.4 \pm 0.9%), *trans*-*m*-coumaroyl (6, 87.5 \pm 4.2%), *trans*-*m*-methylcoumaroyl (7, 88.4 \pm 2.0%), *trans*- and *cis*-cinnamoyl (12 (87.2 \pm 2.1%) and 13 (88.9 \pm 1.8%)), and vanilloyl moiety (14, 75.0 \pm 7.3%), showed potent activities. However, the corresponding flavonol glycosides (17–19) and organic acids (22, 24, 26, and 28–31) lacked this potency. Based on these results, the following structural requirements can be concluded: the acylated flavonol glycosides with a *p*- or *m*-coumaroyl, *p*- or *m*-methylcoumaroyl, cinnamoyl, or vanilloyl moiety with the acyl group in the 6'' position of the D-glucopyranosyl part are essential for the potent inhibition of TG content in high glucose-pretreated HepG2 cells.

Table 3. Effects of acylated flavonol glycosides (1–16) and related compounds (17–22, 24, 26, and 28–31) on high glucose-induced triglyceride accumulation in HepG2 cells.

Treatment	TG/Protein (% of Control)				
	0 μ M	3 μ M	10 μ M	30 μ M	100 μ M
Helichryoside (1)	100.0 \pm 1.6	100.9 \pm 7.1	96.0 \pm 1.5	90.2 \pm 2.2	76.9 \pm 2.3 **
Quercetin 3-O-(6''-O-cis-p-coumaroyl)-Glc (2)	100.0 \pm 0.7	102.0 \pm 2.1	97.3 \pm 2.1	93.2 \pm 1.2 *	80.4 \pm 1.2 **
Quercetin 3-O-(6''-O-trans-p-methylcoumaroyl)-Glc (3)	100.0 \pm 4.8	99.8 \pm 10.2	107.4 \pm 2.3	112.9 \pm 2.6	58.2 \pm 7.5 **
Quercetin 3-O-(6''-O-trans-o-coumaroyl)-Glc (4)	100.0 \pm 2.6	104.6 \pm 4.7	98.2 \pm 3.0	98.3 \pm 2.4	93.4 \pm 1.2
Quercetin 3-O-(6''-O-trans-o-methylcoumaroyl)-Glc (5)	100.0 \pm 6.7	84.4 \pm 11.7	73.5 \pm 12.2	107.6 \pm 13.8	125.9 \pm 3.6
Quercetin 3-O-(6''-O-trans-m-coumaroyl)-Glc (6)	100.0 \pm 3.7	97.4 \pm 10.9	99.0 \pm 2.7	96.4 \pm 5.6	106.0 \pm 4.8
Quercetin 3-O-(6''-O-trans-m-methylcoumaroyl)-Glc (7)	100.0 \pm 3.4	108.0 \pm 4.5	80.5 \pm 10.6	114.2 \pm 8.5	92.2 \pm 8.5
Quercetin 3-O-(6''-O-trans-caffeoyl)-Glc (8)	100.0 \pm 0.7	108.4 \pm 1.6 *	107.0 \pm 1.7 *	103.6 \pm 2.3	98.1 \pm 1.7
Quercetin 3-O-(6''-O-cis-caffeoyl)-Glc (9)	100.0 \pm 2.2	102.4 \pm 1.8	102.2 \pm 1.6	100.7 \pm 1.7	93.8 \pm 2.2
Quercetin 3-O-(6''-O-trans-feruloyl)-Glc (10)	100.0 \pm 3.2	104.7 \pm 5.3	110.6 \pm 2.4	108.4 \pm 4.2	114.9 \pm 3.6
Quercetin 3-O-(6''-O-cis-feruloyl)-Glc (11)	100.0 \pm 7.0	104.9 \pm 3.3	101.6 \pm 2.7	100.3 \pm 2.5	92.9 \pm 5.2
Quercetin 3-O-(6''-O-trans-cinnamoyl)-Glc (12)	100.0 \pm 1.3	106.8 \pm 3.5	100.6 \pm 2.0	102.8 \pm 0.9	115.6 \pm 2.9 **
Quercetin 3-O-(6''-O-cis-cinnamoyl)-Glc (13)	100.0 \pm 4.0	106.3 \pm 4.1	100.2 \pm 3.9	107.8 \pm 3.2	121.0 \pm 3.1 **
Quercetin 3-O-(6''-O-vanilloyl)-Glc (14)	100.0 \pm 3.0	96.3 \pm 4.5	101.7 \pm 2.7	105.9 \pm 3.2	104.5 \pm 1.3
Quercetin 3-O-(6''-O-trimethylgalloyl)-Glc (15)	100.0 \pm 2.9	99.7 \pm 0.7	96.4 \pm 1.1	92.3 \pm 3.4	85.8 \pm 4.1 **
Trans-tiliroside (16)	100.0 \pm 2.0	99.1 \pm 1.2	95.5 \pm 2.0	109.6 \pm 3.8	82.3 \pm 3.0 **
Quercetin 3-O-Glc (17)	100.0 \pm 1.1	97.1 \pm 1.9	124.5 \pm 1.3 **	96.0 \pm 3.1	94.3 \pm 5.2
Kaempferol 3-O-Glc (18)	100.0 \pm 3.5	96.8 \pm 3.2	95.5 \pm 2.0	99.8 \pm 1.3	91.9 \pm 2.1
Rutin (19)	100.0 \pm 4.1	102.1 \pm 7.6	108.4 \pm 3.4	103.9 \pm 2.5	103.1 \pm 2.6
Quercetin (20)	100.0 \pm 0.9	99.8 \pm 1.1	93.8 \pm 0.4 **	82.6 \pm 2.4 **	45.7 \pm 0.4 **
Kaempferol (21)	100.0 \pm 1.2	94.2 \pm 1.2	100.0 \pm 2.4	90.2 \pm 2.4 **	25.5 \pm 1.4 **
Trans-p-coumaric acid (22)	100.0 \pm 0.8	98.4 \pm 2.9	97.6 \pm 3.4	99.6 \pm 1.7	102.7 \pm 2.1
Trans-o-coumaric acid (24)	100.0 \pm 1.1	99.1 \pm 0.7	97.4 \pm 1.7	97.1 \pm 1.5	96.2 \pm 2.2

Table 3. Cont.

Treatment	TG/Protein (% of Control)				
	0 μ M	3 μ M	10 μ M	30 μ M	100 μ M
<i>Trans-m-coumaric acid (26)</i>	100.0 \pm 1.1	99.2 \pm 0.6	97.7 \pm 2.1	97.6 \pm 1.9	99.6 \pm 1.9
<i>Trans-caffeic acid (28)</i>	100.0 \pm 3.8	97.2 \pm 1.9	105.1 \pm 4.3	99.9 \pm 3.3	89.5 \pm 4.7
<i>Trans-ferulic acid (29)</i>	100.0 \pm 3.4	95.0 \pm 1.8	105.3 \pm 4.5	101.9 \pm 3.4	102.0 \pm 2.6
<i>Trans-cinnamic acid (30)</i>	100.0 \pm 4.7	103.4 \pm 2.7	107.1 \pm 1.3	100.0 \pm 2.5	103.2 \pm 5.6
Vanillic acid (31)	100.0 \pm 2.4	97.2 \pm 1.5	100.6 \pm 0.7	99.3 \pm 2.3	101.3 \pm 1.4
	0 mM	0.125 mM	0.25 mM	0.5 mM	1 mM
Metformin	100.0 \pm 0.4	86.8 \pm 1.5 **	75.6 \pm 1.5 **	64.8 \pm 1.0 **	61.1 \pm 2.8 **

Each value represents the mean \pm S.E. ($n = 4$); asterisks denote significant differences from the control group, where * $p < 0.05$ and ** $p < 0.01$.

Table 4. Effects of acylated flavonol glycosides (1–16) and related compounds (17–22, 24, 26, and 28–31) on the triglyceride content in high glucose-pretreated HepG2 cells.

Treatment	TG/Protein (% of Control)				
	0 μ M	3 μ M	10 μ M	30 μ M	100 μ M
Helichryoside (1)	100.0 \pm 3.5	90.5 \pm 0.6	82.9 \pm 1.3*	83.3 \pm 4.6*	62.1 \pm 3.6**
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>cis-p</i> -coumaroyl)-Glc (2)	100.0 \pm 3.7	86.0 \pm 0.8 *	86.0 \pm 3.2 *	81.6 \pm 1.4 **	74.6 \pm 2.3 **
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>trans-p</i> -methylcoumaroyl)-Glc (3)	100.0 \pm 2.2	90.3 \pm 1.0 **	87.4 \pm 0.9 **	83.3 \pm 1.5 **	66.4 \pm 1.4 **
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>trans-o</i> -coumaroyl)-Glc (4)	100.0 \pm 2.3	93.3 \pm 1.5	89.5 \pm 2.6**	79.6 \pm 1.6**	64.5 \pm 1.8 **
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>trans-o</i> -methylcoumaroyl)-Glc (5)	100.0 \pm 3.0	94.5 \pm 2.7	93.4 \pm 1.6	86.4 \pm 1.2 **	64.6 \pm 2.0 **
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>trans-m</i> -coumaroyl)-Glc (6)	100.0 \pm 1.3	91.3 \pm 3.6	87.5 \pm 4.2 *	83.7 \pm 3.1 **	69.8 \pm 3.2 **
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>trans-m</i> -methylcoumaroyl)-Glc (7)	100.0 \pm 2.0	91.2 \pm 1.9 *	88.4 \pm 2.0 **	83.8 \pm 1.5 **	76.2 \pm 1.5 **
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>trans-caffeoyl</i>)-Glc (8)	100.0 \pm 13.8	91.1 \pm 6.0	87.0 \pm 5.2	84.2 \pm 6.3	65.8 \pm 5.4 **
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>cis-caffeoyl</i>)-Glc (9)	100.0 \pm 2.9	95.6 \pm 1.4	91.8 \pm 2.0	87.4 \pm 4.2 *	87.4 \pm 4.6 *
Quercetin 3- <i>O</i> -(6''- <i>O</i> - <i>trans-feruloyl</i>)-Glc (10)	100.0 \pm 7.9	89.2 \pm 3.9	94.0 \pm 4.3	74.3 \pm 5.8 **	75.8 \pm 3.7 **

Table 4. Cont.

Treatment	TG/Protein (% of Control)				
	0 μ M	3 μ M	10 μ M	30 μ M	100 μ M
Quercetin 3-O-(6''-O-cis-feruloyl)-Glc (11)	100.0 \pm 10.6	86.8 \pm 4.7	90.7 \pm 2.2	82.4 \pm 4.8 *	77.3 \pm 3.5 **
Quercetin 3-O-(6''-O-trans-cinnamoyl)-Glc (12)	100.0 \pm 3.4	91.4 \pm 2.6	87.2 \pm 2.1 *	80.9 \pm 2.1 **	68.7 \pm 1.3 **
Quercetin 3-O-(6''-O-cis-cinnamoyl)-Glc (13)	100.0 \pm 2.1	93.2 \pm 0.8	88.9 \pm 1.8 **	89.9 \pm 0.5 **	82.5 \pm 1.0 **
Quercetin 3-O-(6''-O-vanilloyl)-Glc (14)	100.0 \pm 9.7	102.5 \pm 7.4	75.0 \pm 7.3 **	70.2 \pm 5.2 **	60.8 \pm 6.1 **
Quercetin 3-O-(6''-O-trimethylgalloyl)-Glc (15)	100.0 \pm 5.7	92.2 \pm 2.0	93.1 \pm 1.0	83.7 \pm 1.3 **	81.7 \pm 3.9 **
Trans-tiliroside (16)	100.0 \pm 1.7	98.9 \pm 2.9	96.2 \pm 1.2	86.1 \pm 3.1 **	72.8 \pm 3.1 **
Quercetin 3-O-Glc (17)	100.0 \pm 2.1	100.9 \pm 2.6	100.5 \pm 2.5	108.3 \pm 1.2	108.0 \pm 1.4
Kaempferol 3-O-Glc (18)	100.0 \pm 3.2	98.6 \pm 3.3	96.0 \pm 2.3	88.8 \pm 3.3	92.4 \pm 5.1
Rutin (19)	100.0 \pm 9.2	95.3 \pm 4.0	91.8 \pm 5.8	99.3 \pm 2.7	94.9 \pm 7.6
Quercetin (20)	100.0 \pm 2.4	92.5 \pm 1.1	92.1 \pm 3.5	71.5 \pm 3.3 **	63.2 \pm 0.7 **
Kaempferol (21)	100.0 \pm 2.4	103.4 \pm 2.8	95.0 \pm 4.6	82.8 \pm 1.5 **	56.1 \pm 3.6 **
Trans-p-coumaric acid (22)	100.0 \pm 3.3	95.1 \pm 4.5	99.8 \pm 4.9	103.7 \pm 3.3	118.6 \pm 6.2 *
Trans-o-coumaric acid (24)	100.0 \pm 2.2	97.2 \pm 0.9	93.8 \pm 3.3	97.5 \pm 4.1	102.4 \pm 4.2
Trans-m-coumaric acid (26)	100.0 \pm 2.3	93.7 \pm 1.0	97.1 \pm 3.4	92.1 \pm 0.8	94.4 \pm 1.7
Trans-caffeic acid (28)	100.0 \pm 13.2	89.9 \pm 6.6	89.6 \pm 0.6	91.1 \pm 2.6	97.6 \pm 1.4
Trans-ferulic acid (29)	100.0 \pm 11.2	88.6 \pm 6.3	97.4 \pm 3.1	100.5 \pm 4.7	104.3 \pm 6.6
Trans-cinnamic acid (30)	100.0 \pm 3.0	102.8 \pm 2.7	99.2 \pm 7.0	98.5 \pm 2.9	102.0 \pm 5.3
Vanillic acid (31)	100.0 \pm 9.2	92.9 \pm 4.7	98.4 \pm 2.8	95.1 \pm 8.4	106.7 \pm 2.5
	0 μM	0.125 mM	0.25 mM	0.5 mM	1 mM
Metformin	100.0 \pm 1.2	92.1 \pm 1.4 *	88.6 \pm 1.5 **	87.8 \pm 2.2 **	81.4 \pm 1.4 **

Each value represents the mean \pm S.E. ($n = 4$ or 8); asterisks denote significant differences from the control group, where * $p < 0.05$ and ** $p < 0.01$.

Recently, it has been reported that derivatives of **16** activate adenosine 5'-monophosphate-activated protein kinase (AMPK) in 3T3-L1 cells [63] and stimulate glucose transporter (GLUT) 4 translocation in skeletal muscle cells [64]. AMPK is known as a key molecule involved in regulating glucose and lipid metabolism in the liver. From the similarity of the structure, the activities exhibited by **1** and its analogs in this study may have been caused via the same mechanism, but further investigations are needed to clarify the details.

3. Materials and Methods

3.1. Chemicals and Reagents

Rutin and naringinase were purchased from Sigma-Aldrich Co. LLC., St. Louis, MO, USA. DMF, *tert*-butyldiphenylsilyl chloride (TBDPSCI), EDC·HCl, 4-DMAP, TBAF, and tetrahydrofuran (THF) were purchased from Tokyo Chemical Industry Co., Ltd., Tokyo, Japan. Flavonols (**17**, **18**, and **20–21**), organic acids (**22–31**), and other chemicals, unless otherwise indicated, were purchased from Nakalai Tesque Inc., Kyoto, Japan.

3.2. General Experimental Procedures

The following instruments were used to obtain spectroscopic data: specific rotations, Horiba SEPA-300 digital polarimeter ($l = 5$ cm); UV spectra, Shimadzu UV-1600 spectrometer; IR spectra, Shimadzu FTIR-8100 spectrometer; FAB-MS and high-resolution MS, JEOL JMS-SX 102A mass spectrometer; ESIMS and HRESIMS, Exactive Plus mass spectrometer (Thermo Fisher Scientific Inc., MA, USA); $^1\text{H-NMR}$ spectra, JEOL JNM-ECA600 (600 MHz) and JNM-ESC400 (400 MHz) spectrometers; $^{13}\text{C-NMR}$ spectra, JEOL JNM-ECA600 (150 MHz) and JNM-ESC400 (100 MHz) spectrometers, with tetramethylsilane as an internal standard; HPLC detector, Shimadzu SPD-10A vp UV-VIS detectors; and HPLC column, Cosmosil 5C $_{18}$ -MS-II (Nacalai Tesque Inc.). The following experimental conditions were used for column chromatography (CC): ordinary-phase silica gel CC, silica gel 60N (Kanto Chemical Co., Tokyo, Japan; 63–210 mesh, spherical, neutral), normal-phase thin-layer chromatography (TLC), pre-coated TLC plates with silica gel 60F $_{254}$ (Merck, Darmstadt, Germany; 0.25 mm), with detection achieved by spraying with 1% Ce(SO $_4$) $_2$ –10% aqueous H $_2$ SO $_4$, and followed by heating.

3.3. Enzymatic Hydrolysis of Rutin (**19**) Monitored by HPLC

A suspension of rutin (**19**, 100.0 mg) in H $_2$ O (50 mL) was mixed and stirred at 50 °C in a water bath for a few minutes. Then, naringinase (5.0 mg) was added to the suspension to start the reaction. Aliquots (1 mL) of the reaction mixture after 0, 5, and 30 min and 1, 1.5, 2, 3, 4.5, 8, and 24 h were transferred into a 10 mL volumetric flask and methanol was added to make up the volume, respectively. Each solution was filtered through a syringe filter (0.45 μm), and an aliquot of 1 μL was subjected to the following HPLC analytical conditions.

A series LC-20A Prominence HPLC system (version 3.40, Shimadzu Co., Kyoto, Japan) was equipped with a UV-VIS detector, a binary pump, a degasser, an autosampler, a thermostatic column compartment, and a control module. The chromatographic separation was performed on a Cosmosil 5C $_{18}$ -MS-II (3 μm particle size, 150 \times 2.0 mm i.d., Nacalai Tesque Inc., Kyoto, Japan) operated at 40 °C with mobile phase A (acetonitrile) and B (H $_2$ O containing 0.1% acetic acid). The gradient program was as follows: 0–3 min (A:B = 20:80, v/v) \rightarrow 10–15 min (90:10, v/v) \rightarrow 15–25 min (20:80, v/v, hold). The flow rate was 0.2 mL/min with UV detection at 254 nm and the injection volume was 1 μL . The standard curves were prepared with five concentration levels in the range of 25–400 $\mu\text{g/mL}$ (25, 50, 100, 200, and 400 $\mu\text{g/mL}$, respectively). Linearity for each compound, such as rutin (**19**), quercetin 3-*O*- β -D-glucopyranoside (**17**), and quercetin (**20**), was plotted using linear regression of the peak area versus concentration. The coefficient of correlation (R^2) was used to judge the linearity (Figure S1 and Table S1).

3.4. Practical Derivation from Rutin (19) to Quercetin 3-O- β -D-glucopyranoside (17) by Naringinase

Naringinase (750.0 mg) was added to a suspension of rutin (**19**, 15.0 g, 24.6 mmol) in H₂O (7.5 L), and the mixture was stirred at 50 °C for 2 h. Removal of the solvent from the reaction mixture was carried out under reduced pressure using EtOH as an azeotropic solvent to give a crude product, which, on silica gel CC (500 g, CHCl₃/MeOH/H₂O (10:3:0.4, v/v/v)), gave a title compound (**17**, 6.50 g, 56.9%).

3.5. Synthesis of Helichryoside (1)

Under an argon atmosphere, imidazole (1.50 g, 22.0 mmol, 3.2 eq) and TBDPSCI (4.54 g, 16.5 mmol, 2.4 eq) were added to a solution of *trans-p*-coumaric acid (**22**, 1.30 g, 6.88 mmol) in dry-DMF (12.0 mL), and the mixture was stirred at 40 °C for 16 h. The reaction mixture was poured into ice-water and extracted with EtOAc, before being washed with brine. The extract was condensed under a reduced pressure to give a white solid, which was dissolved in CHCl₃/MeOH (10:7, v/v, 17 mL) and acidified by 1 M HCl until pH 3.0. After stirring at room temperature for 1.5 h, the reaction mixture was condensed under a reduced pressure to give a pale yellow oil, which was crystallized in *n*-hexane/EtOAc (9:1, v/v, 20 mL) to give **22a** (2.12 g, 76.5%).

4-*O*-*tert*-Butyldiphenylsilyl ether of *trans-p*-coumaric acid (**22a**): ¹H NMR (400 MHz, CDCl₃): δ 1.10 (9H, s, *tert*-Bu), 6.23, 7.65 (1H each, both d, J = 16.0 Hz, H-8 and 7), 6.76, 7.30 (2H each, both d, J = 8.7 Hz, H-3,5 and 2,6), [7.37 (4H, m), 7.43 (2H, m), 7.70 (4H, dd, J = 1.9, 8.2 Hz), arom.]. ¹³C NMR (100 MHz, CDCl₃): δ 127.2 (C-1), 129.8 (C-2,6), 120.3 (C-3,5), 158.1 (C-4), 146.7 (C-7), 114.8 (C-8), 172.3 (C-9), 132.3, 135.4, 127.9, 130.1 (arom. C-1, 2,6, 3,5, 4), 19.4 (CH₃C-), 26.4 (CH₃C-).

Compound **22a** (0.40 g, 0.96 mmol, 1.2 eq), EDC·HCl (0.31 g, 1.60 mmol, 2.0 eq), and 4- DMAP (0.15 g, 1.20 mmol, 1.5 eq) were added to a solution of **17** (0.37 g, 0.80 mmol) in pyridine (4.0 mL), and the mixture was stirred at 50 °C for 12 h. The reaction mixture was condensed under a reduced pressure to give a crude product. Then, a solution of the crude product in THF (5.0 mL) was added to TBAF (*ca.* 1.0 mol/L in THF, 800 μ L, 0.80 mmol, 1.0 eq) at room temperature for 1 h. The reaction mixture was quenched in H₂O, and removal of the solvent under a reduced pressure then furnished a residue, which, on silica gel CC (10 g, CHCl₃/MeOH/H₂O (10:3:0.4, v/v/v)), gave a title compound (**1**, 0.18 g, 36.9%).

3.6. Synthesis of 3–8, 10, 12, 14, and 15

In a manner similar to that used for **22a** (*vide supra*), *trans-o*-coumaric acid (**24**), *trans-m*-coumaric acid (**26**), *trans*-caffeic acid (**28**), *trans*-ferulic acid (**29**), and vanillic acid (**31**) were derived to yield the corresponding silyl ether derivatives, **24a** (95.1%), **26a** (94.6%), **28a** (393.0%), **29a** (87.3%), and **31a** (98.5%), respectively.

2-*O*-*tert*-Butyldiphenylsilyl ether of *trans-o*-coumaric acid (**24a**): ¹H NMR (400 MHz, CDCl₃): δ 1.15 (9H, s, *tert*-Bu), 6.49, 8.53 (1H each, both d, J = 16.0 Hz, H-8 and 7), 6.48 (1H, dd, J = 1.8, 8.7 Hz, H-3), 6.88 (1H, br dd, J = *ca.* 8, 8 Hz, H-5), 6.96 (1H, ddd, J = 1.8, 8.3, 8.7 Hz, H-4), 7.59 (1H, dd, J = 1.8, 7.8 Hz, H-6), (7.39 (4H, m), 7.44 (2H, m), 7.72 (4H, dd, J = 1.8, 8.3 Hz), arom.). ¹³C NMR (100 MHz, CDCl₃): δ 125.0 (C-1), 154.6 (C-2), 119.9 (C-3), 131.5 (C-4), 121.4 (C-5), 127.5 (C-6), 142.3 (C-7), 117.1 (C-8), 172.6 (C-9), 132.2, 135.4, 127.9, 130.1 (arom. C-1, 2,6, 3,5, 4), 19.6 (CH₃C-), 26.5 (CH₃C-).

3-*O*-*tert*-Butyldiphenylsilyl ether of *trans-m*-coumaric acid (**26a**): ¹H NMR (400 MHz, CDCl₃): δ 1.12 (9H, s, *tert*-Bu), 6.17, 7.59 (1H each, both d, J = 16.0 Hz, H-8 and 7), 6.78 (1H, br d, J = *ca.* 8 Hz, H-4), 6.94 (1H, br s, H-2), 7.04 (br d, J = *ca.* 8 Hz, H-6), 7.10 (dd, J = 7.8, 7.8 Hz, H-5), (7.38 (4H, m), 7.44 (2H, m), 7.71 (4H, m), arom.). ¹³C NMR (100 MHz, CDCl₃): δ 135.2 (C-1), 119.1 (C-2), 156.0 (C-3), 122.2 (C-4), 129.7 (C-5), 121.5 (C-6), 146.8 (C-7), 117.3 (C-8), 172.1 (C-9), 132.2, 135.5, 127.9, 130.1 (arom. C-1, 2,6, 3,5, 4), 19.5 (CH₃C-), 26.5 (CH₃C-).

3,4-Di-*O*-*tert*-butyldiphenylsilyl ether of *trans*-caffeic acid (**28a**): ¹H NMR (400 MHz, CDCl₃): δ 1.14, 1.18 (9H each, both s, *tert*-Bu), 5.48, 7.17 (1H each, both d, J = 16.0 Hz, H-8 and 7), 6.39 (1H, d,

$J = 8.2$ Hz, H-5), 6.53 (1H, dd, $J = 1.8, 8.2$ Hz, H-6), 6.59 (1H, d, $J = 1.8$ Hz, H-2), (7.40 (8H, m), 7.45 (4H, m), 7.79 (8H, m), arom.). ^{13}C NMR (100 MHz, CDCl_3): δ_{C} 126.8 (C-1), 119.4 (C-2), 146.5 (C-3), 148.9 (C-4), 120.5 (C-5), 122.3 (C-6), 146.6 (C-7), 114.4 (C-8), 172.0 (C-9), 132.5/132.9, 135.4/135.6, 127.9/128.0, 130.0/130.2 (arom. C-1, 2,6, 3,5, 4), 19.5, 19.5 ($\text{CH}_3\text{C}-$), 26.6, 26.8 ($\text{CH}_3\text{C}-$).

4-*O*-*tert*-Butyldiphenylsilyl ether of *trans*-ferulic acid (**29a**): ^1H NMR (400 MHz, CDCl_3): δ 1.11 (9H, s, *tert*-Bu), 3.60 (3H, s, $-\text{OCH}_3$), 6.23, 7.64 (1H each, both d, $J = 16.0$ Hz, H-8 and 7), 6.69 (1H, d, $J = 8.2$ Hz, H-5), 6.86 (1H, dd, $J = 1.8, 8.2$ Hz, H-6), 6.95 (1H, d, $J = 1.8$ Hz, H-2), (7.35 (4H, m), 7.41 (2H, m), 7.70 (4H, dd, $J = 1.8, 7.8$ Hz), arom.). ^{13}C NMR (100 MHz, CDCl_3): δ_{C} 127.7 (C-1), 111.2 (C-2), 150.8 (C-3), 147.9 (C-4), 120.4 (C-5), 122.5 (C-6), 147.1 (C-7), 114.7 (C-8), 172.3 (C-9), 133.1, 135.3, 127.6, 129.8 (arom. C-1, 2,6, 3,5, 4), 19.8 ($\text{CH}_3\text{C}-$), 26.6 ($\text{CH}_3\text{C}-$).

3-*O*-*tert*-Butyldiphenylsilyl ether of vanillic acid (**31a**): ^1H NMR (400 MHz, CDCl_3): δ 1.12 (9H, s, *tert*-Bu), 3.61 (3H, s, $-\text{OCH}_3$), 6.37 (1H, d, $J = 9.2$ Hz, H-5), 7.47 (1H, dd, $J = 1.8, 9.2$ Hz, H-6), 7.48 (1H, d, $J = 1.8$ Hz, H-2), (7.35 (4H, m), 7.41 (2H, m), 7.69 (4H, m), arom.). ^{13}C NMR (100 MHz, CDCl_3): δ_{C} 122.4 (C-1), 113.5 (C-2), 150.4 (C-3), 150.3 (C-4), 119.8 (C-5), 124.0 (C-6), 171.7 (C-9), 132.9, 135.2, 127.6, 129.8 (arom. C-1, 2,6, 3,5, 4), 19.8 ($\text{CH}_3\text{C}-$), 26.5 ($\text{CH}_3\text{C}-$).

In a manner similar to that used for the preparation of **1** (*vide supra*), dehydration condensation of **17** (500.0 mg, 1.08 mmol) and **24a**, **26a**, **28a**, **30a**, and **31a** (521.1, 521.1, 521.1, 849.6, 558.6, and 526.3 mg, respectively, 1.29 mmol, 1.2 eq) were conducted to yield the corresponding 6''-*O*-acylated quercetin 3- β -*D*-glucopyranosides, **4** (134.6 mg, 31.0%), **6** (147.7 mg, 34.0%), **8** (248.5 mg, 35.1%), **10** (176.9 mg, 38.0%), and **14** (130.7 mg, 29.8%), respectively. Syntheses of the 6''-*O*-acylated quercetin 3- β -*D*-glucopyranosides, **3** (39.0 mg, 20.3%), **5** (63.2 mg, 32.9%), **7** (57.2 mg, 29.8%), **12** (53.2 mg, 33.3%), and **15** (75.2 mg, 32.9%) were also carried out by the dehydration condensation of **17** with *trans*-*p*-methylcoumaric acid (**23**), *trans*-*o*-methylcoumaric acid (**25**), *trans*-*m*-methylcoumaric acid (**27**), *trans*-cinnamic acid (**30**), and trimethylgallic acid, respectively, without the deprotection procedure of the *tert*-butyldiphenylsilyl (TBDPS) ether group.

Quercetin 3-*O*-(6''-*O*-*trans*-*p*-methylcoumaroyl)- β -*D*-glucopyranoside (**3**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{31}\text{H}_{28}\text{O}_{14}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 647.1377. Found: 647.1373. ^1H NMR (600 MHz, $\text{DMSO}-d_6$): δ 3.21 (1H, dd, $J = 8.9, 9.0$ Hz, H-4''), 3.29 (1H, dd, $J = 8.8, 9.0$ Hz, H-3''), 3.31 (1H, dd, $J = 7.4, 8.8$ Hz, H-2''), 3.40 (1H, ddd, $J = 2.0, 6.9, 8.9$ Hz, H-5''), 3.82 (3H, s, $-\text{OCH}_3$), (4.30 (1H, dd, $J = 2.0, 11.8$ Hz), 4.66 (1H, dd, $J = 6.9, 11.8$ Hz), H₂-6''), 5.49 (1H, d, $J = 7.4$ Hz, H-1''), 6.13, 6.33 (1H each, both d, $J = 2.0$ Hz, H-6 and 8), 6.17, 7.36 (1H each, both d, $J = 16.0$ Hz, H-8''' and 7'''), 6.83 (1H, d, $J = 8.4$ Hz, H-5'), 6.95, 7.44 (2H each, both d, $J = 8.8$ Hz, H-3''' and 5'''), 7.53 (1H, dd, $J = 2.2, 8.4$ Hz, H-6'), 7.55 (1H, d, $J = 2.2$ Hz, H-2'), 12.60 (1H, br s, 5-OH). ^{13}C NMR (150 MHz, $\text{DMSO}-d_6$): δ_{C} given in Table S2.

Quercetin 3-*O*-(6''-*O*-*trans*-*o*-coumaroyl)- β -*D*-glucopyranoside (**4**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{30}\text{H}_{26}\text{O}_{14}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 633.1220. Found: 633.1226. ^1H NMR (600 MHz, $\text{DMSO}-d_6$): δ 3.22 (1H, dd, $J = 8.8, 9.5$ Hz, H-4''), 3.28 (1H, dd, $J = 8.8, 8.8$ Hz, H-3''), 3.30 (1H, dd, $J = 7.3, 8.8$ Hz, H-2''), 3.40 (1H, ddd, $J = 2.2, 6.4, 9.5$ Hz, H-5''), (4.07 (1H, dd, $J = 6.4, 11.9$ Hz), 4.29 (1H, dd, $J = 2.2, 11.9$ Hz), H₂-6''), 5.47 (1H, d, $J = 7.3$ Hz, H-1''), 6.15, 6.38 (1H each, both d, $J = 2.0$ Hz, H-6 and 8), 6.34, 7.73 (1H each, both d, $J = 16.1$ Hz, H-8''' and 7'''), 6.83 (1H, ddd, $J = 1.7, 7.8, 8.7$ Hz, H-5'''), 6.84 (1H, d, $J = 8.2$ Hz, H-5''), 6.93 (1H, dd, $J = 1.7, 8.3$ Hz, H-3'''), 7.22 (1H, ddd, $J = 1.6, 8.3, 8.7$ Hz, H-4'''), 7.38 (1H, dd, $J = 1.6, 7.8$ Hz, H-6'''), 7.52 (1H, dd, $J = 2.0, 8.2$ Hz, H-6''), 7.55 (1H, d, $J = 2.0$ Hz, H-2''), 12.58 (1H, br s, 5-OH). ^{13}C NMR (150 MHz, $\text{DMSO}-d_6$): δ_{C} given in Table S2.

Quercetin 3-*O*-(6''-*O*-*trans*-*o*-methylcoumaroyl)- β -*D*-glucopyranoside (**5**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{31}\text{H}_{28}\text{O}_{14}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 647.1377. Found: 647.1381. ^1H NMR (600 MHz, $\text{DMSO}-d_6$): δ 3.21 (1H, m, H-4''), 3.33 (2H, m, H-2'', 3''), 3.41 (1H, ddd, $J = 2.3, 6.9, 9.5$ Hz, H-5''), 3.82 (3H, s, $-\text{OCH}_3$), (4.11 (1H, dd, $J = 6.9, 12.1$ Hz), 4.30 (1H, dd, $J = 2.3, 12.1$ Hz), H₂-6''), 5.48 (1H, d, $J = 7.5$ Hz, H-1''), 6.07, 6.30 (1H each, both d, $J = 2.0$ Hz, H-6 and 8), 6.32, 7.70 (1H each, both d, $J = 16.1$ Hz, H-8''' and 7'''), 6.83 (1H, d, $J = 8.9$ Hz, H-5''), 6.98 (1H, br dd, $J = \text{ca. } 8, 8$ Hz, H-5'''), 7.05 (1H, br d, $J = \text{ca. } 8$ Hz, H-3'''), 7.41 (1H, ddd, $J = 1.7, 8.2, 8.3$ Hz, H-4'''), 7.47 (1H, dd,

$J = 1.7, 8.2$ Hz, H-6'''), 7.52 (1H, dd, $J = 2.3, 8.9$ Hz, H-6''), 7.53 (1H, d, $J = 2.3$ Hz, H-2''), 12.57 (1H, br s, 5-OH). ^{13}C NMR (150 MHz, DMSO- d_6): δ_{C} given in Table S2.

Quercetin 3-*O*-(6''-*O*-*trans*-*m*-coumaroyl)- β -*D*-glucopyranoside (**6**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{30}\text{H}_{26}\text{O}_{14}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 633.1220. Found: 633.1226. ^1H NMR (600 MHz, DMSO- d_6): δ 3.22 (1H, dd, $J = 8.8, 9.0$ Hz, H-4''), 3.28 (1H, dd, $J = 7.6, 8.9$ Hz, H-2''), 3.31 (1H, dd, $J = 8.8, 8.9$ Hz, H-3''), 3.40 (1H, ddd, $J = 2.0, 6.6, 9.0$ Hz, H-5''), (4.08 (1H, dd, $J = 6.6, 11.9$ Hz), 4.30 (1H, dd, $J = 2.0, 11.9$ Hz), H₂-6''), 5.46 (1H, d, $J = 7.6$ Hz, H-1''), 6.12, 6.33 (1H each, both br s, H-6, 8), 6.24, 7.35 (1H each, both d, $J = 16.0$ Hz, H-8''' and 7'''), 6.82 (1H, d, $J = 8.4$ Hz, H-5'), 6.83 (1H, br d, $J = \text{ca. } 8$ Hz, H-4'''), 6.91 (1H, br d, $J = \text{ca. } 8$ Hz, H-6'''), 6.94 (1H, br s, H-2''), 7.20 (1H, dd, $J = 8.0, 8.0$ Hz, H-5'''), 7.52 (1H, dd, $J = 2.2, 8.4$ Hz, H-6'), 7.54 (1H, d, $J = 2.2$ Hz, H-2'), 12.57 (1H, br s, 5-OH). ^{13}C NMR (150 MHz, DMSO- d_6): δ_{C} given in Table S2.

Quercetin 3-*O*-(6''-*O*-vanilloyl)- β -*D*-glucopyranoside (**14**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{29}\text{H}_{26}\text{O}_{15}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 637.1169. Found: 637.1174. ^1H NMR (600 MHz, DMSO- d_6): δ 3.24 (1H, dd, $J = 8.9, 9.1$ Hz, H-4''), 3.30 (1H, dd, $J = 8.8, 8.9$ Hz, H-3''), 3.33 (1H, dd, $J = 7.4, 8.8$ Hz, H-2''), 3.45 (1H, ddd, $J = 2.1, 6.9, 9.1$ Hz, H-5''), 3.70 (3H, s, -OCH₃), (4.13 (1H, dd, $J = 6.9, 11.9$ Hz), 4.40 (1H, dd, $J = 2.1, 11.9$ Hz), H₂-6''), 5.54 (1H, d, $J = 7.4$ Hz, H-1''), 6.18, 6.35 (1H each, both d, $J = 1.9$ Hz, H-6 and 8), 6.70 (1H, d, $J = 8.2$ Hz, H-5'''), 6.78 (1H, d, $J = 8.4$ Hz, H-5'), 7.19 (1H, dd, $J = 2.0, 8.2$ Hz, H-6'''), 7.27 (1H, d, $J = 2.0$ Hz, H-2'''), 7.51 (1H, d, $J = 2.2$ Hz, H-2'), 7.53 (1H, dd, $J = 2.2, 8.4$ Hz, H-6'), 12.58 (1H, br s, 5-OH). ^{13}C NMR (150 MHz, DMSO- d_6): δ_{C} given in Table S2.

Quercetin 3-*O*-(6''-*O*-trimethylgalloyl)- β -*D*-glucopyranoside (**15**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{31}\text{H}_{30}\text{O}_{16}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 681.1432. Found: 681.1436. UV (MeOH, nm (log ϵ)): 259 (3.99), 295 (3.62), 359 (3.77). IR (KBr): 3590, 1701, 1655, 1597, 1503, 1341, 1202, 1075 cm^{-1} . ^1H NMR (600 MHz, DMSO- d_6): δ 3.23 (1H, dd, $J = 9.1, 9.7$ Hz, H-4''), 3.30 (2H, m, H-2'', 3''), 3.50 (1H, ddd, $J = 2.4, 7.3, 9.7$ Hz, H-5''), (3.67 (6H, s), 3.73 (3H, s), -OCH₃), (4.26 (1H, dd, $J = 7.3, 11.9$ Hz), 4.43 (1H, dd, $J = 2.4, 11.9$ Hz), H₂-6''), 5.49 (1H, d, $J = 7.6$ Hz, H-1''), 6.10, 6.30 (1H each, both d, $J = 2.1$ Hz, H-6 and 8), 6.74 (1H, d, $J = 8.6$ Hz, H-5'), 7.48 (1H, br s, H-2'), 7.49 (1H, br d, $J = \text{ca. } 9$ Hz, H-6'), 12.48 (1H br s, 5-OH). ^{13}C NMR (150 MHz, DMSO- d_6): δ_{C} given in Table S2.

3.7. Isomerization of **1**, **8**, **10**, and **12**

A methanol solution (20 mL) of **1** (100.0 mg, 0.17 mmol) in a Pyrex tube was left standing for 8 h under a UV lamp (short wave) at room temperature. The reaction mixture was condensed under a reduced pressure to give a crude product, which, on HPLC (Cosmosil 5C₁₈-MS-II, MeOH-1% aqueous AcOH (55:45, v/v)), gave the *cis*-isomer **2** (25.1 mg) and a recovering compound (**1**, 69.8 mg). Using the similar procedure, a methanol solution (10 mL) of **8**, **10**, or **12** (each 50.0 mg) was isomerized to the corresponding *cis*-isomer **9** (12.2 mg (recovered **8**, 25.8 mg)), **11** (10.8 mg (recovered **10**, 26.9 mg)), or **13** (14.2 mg (recovered **12**, 30.1 mg)).

Quercetin 3-*O*-(6''-*O*-*cis*-*p*-coumaroyl)- β -*D*-glucopyranoside (**2**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{30}\text{H}_{26}\text{O}_{14}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 633.1220. Found: 633.1226. ^1H NMR (600 MHz, CD₃OD): δ 3.29 (1H, dd, $J = 8.9, 9.0$ Hz, H-4''), 3.42 (2H, m, H-3'', 5''), 3.49 (1H, dd, $J = 7.3, 8.8$ Hz, H-2''), 4.14-4.22 (2H, m, H₂-6''), 5.19 (1H, d, $J = 7.3$ Hz, H-1''), 5.51, 6.69 (1H each, both d, $J = 12.8$ Hz, H-8''' and 7'''), 6.18, 6.30 (1H each, both d, $J = 1.8$ Hz, H-6 and 8), 6.67, 7.49 (2H each, both d, $J = 8.6$ Hz, H-3''', 5''' and 2''', 6'''), 6.80 (1H, d, $J = 8.7$ Hz, H-5'), 7.56 (2H, m, H-2', 6'). ^{13}C NMR (150 MHz, CD₃OD): δ_{C} given in Table S2.

Quercetin 3-*O*-(6''-*O*-*cis*-caffeoyle)- β -*D*-glucopyranoside (**9**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $\text{C}_{30}\text{H}_{26}\text{O}_{15}\text{Na}$ ($\text{M}+\text{Na}$) $^+$: 649.1169. Found: 649.1170. ^1H NMR (600 MHz, DMSO- d_6): δ 3.18-3.36 (4H, m, H-2''-5''), 3.71(3H, s, -OCH₃), (4.04 (1H, dd, $J = 6.4, 11.9$ Hz), 4.19 (1H, dd, $J = 2.2, 11.9$ Hz), H₂-6''), 5.42 (1H, d, $J = 7.5$ Hz, H-1''), 5.42, 6.58 (1H each, both d, $J = 12.8$ Hz, H-8''' and 7'''), 6.20, 6.36 (1H each, both d, $J = 1.7$ Hz, H-6 and 8), 6.28 (1H, d, $J = 2.0$ Hz, H-2'''), 6.67 (1H, d, $J = 8.5$ Hz, H-5'''), 6.82 (1H, d, $J = 8.6$ Hz, H-5'), 6.98 (1H, dd, $J = 2.0, 8.5$ Hz, H-6'''), 7.52 (2H, m, H-2', 6'), 12.59 (1H, br s, 5-OH). ^{13}C NMR (150 MHz, DMSO- d_6): δ_{C} given in Table S2.

Quercetin 3-O-(6''-O-cis-feruloyl)- β -D-glucopyranoside (**11**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $C_{31}H_{28}O_{15}Na$ (M+Na)⁺: 663.1326. Found: 663.1330. ¹H NMR (600 MHz, DMSO-*d*₆): δ 3.19 (1H, m, H-4''), 3.27 (2H, m, H-2'', 3''), 3.36 (1H, ddd, *J* = 2.3, 6.0, 9.6 Hz, H-5''), 3.71 (3H, s, -OCH₃), 4.18 (1H, dd, *J* = 6.4, 11.9 Hz), 4.19 (1H, dd, *J* = 1.8, 11.9 Hz, H₂-6''), 5.46 (1H, d, *J* = 7.4 Hz, H-1''), 5.48, 6.65 (1H each, both d, *J* = 12.8 Hz, H-8''' and 7'''), 6.19, 6.31 (1H each, both d, *J* = 1.8 Hz, H-6 and 8), 6.72 (1H, d, *J* = 8.2 Hz, H-5'''), 6.81 (1H, d, *J* = 8.7 Hz, H-5'), 7.08 (1H, dd, *J* = 1.8, 8.2 Hz, H-6'''), 7.45 (1H, br d, *J* = ca. 9 Hz, H-6'), 7.52 (1H, br s, H-2'), 7.59 (1H, d, *J* = 1.8 Hz, H-2'''), 12.57 (1H, br s, 5-OH). ¹³C NMR (150 MHz, DMSO-*d*₆): δ_C given in Table S2.

Quercetin 3-O-(6''-O-cis-cinnamoyl)- β -D-glucopyranoside (**13**): A yellow powder, high-resolution positive-ion FABMS: Calcd for $C_{30}H_{26}O_{13}Na$ (M+Na)⁺: 617.1271. Found: 617.1277. ¹H NMR (600 MHz, DMSO-*d*₆): δ 3.14 (1H, dd, *J* = 8.6, 9.5 Hz, H-4''), 3.25 (1H, dd, *J* = 7.5, 8.9 Hz, H-2''), 3.28 (1H, dd, *J* = 8.6, 8.9 Hz, H-3''), 3.33 (1H, ddd, *J* = 2.3, 6.6, 9.5 Hz, H-5''), (4.07 (1H, dd, *J* = 6.6, 11.8 Hz), 4.18 (1H, dd, *J* = 2.3, 11.8 Hz), H₂-6''), 5.69, 6.81 (1H each, both d, *J* = 12.9 Hz, H-8''' and 7'''), 6.19, 6.33 (1H each, both d, *J* = 2.0 Hz, H-6 and 8), 6.82 (1H, d, *J* = 8.9 Hz, H-5'), 7.27 (4H, m, H-2''', 6''', 3''', 5'''), 7.43 (1H, m, H-4'''), 7.51 (1H, dd, *J* = 2.0, 8.9 Hz, H-6'), 7.51 (1H, d, *J* = 2.0 Hz, H-2'), 12.56 (1H, br s, 5-OH). ¹³C NMR (150 MHz, DMSO-*d*₆): δ_C given in Table S2.

3.8. Animals

Male ddY mice were purchased from Kiwa Laboratory Animal Co., Ltd., (Wakayama, Japan). The animals were housed at a constant temperature of 23 ± 2 °C and fed a standard laboratory chow (MF, Oriental Yeast Co., Ltd., Tokyo, Japan). All experiments were performed with conscious mice, unless otherwise noted. The experimental protocol was approved by Kindai University's Committee for the Care and Use of Laboratory Animals (KAPR-26-004, 1 April 2014).

3.9. Effects on the Glucose Tolerance Test in Mice

Effects on the glucose tolerance test after 14 days of administration of **1** in mice were determined according to the previously described protocol [26]. A test sample was administrated orally to male ddY mice (11 weeks old and fed a standard laboratory chow) once a day (10:00–12:00) for 14 days. Body weight and food intake were measured every day before administration of the test sample. Fasting for 20 h was carried out after the final administration, and 10% (w/v) glucose solution was intraperitoneally (i.p.) administrated to mice at 10 mL/kg. Blood samples (ca. 0.2 mL) were collected in tubes containing 10 units of heparin sodium from the infraorbital venous plexus before (0 h) and 0.5, 1, and 2 h after the loading of glucose. Mice were then killed by cervical dislocation, and the epididymal, mesenteric, and paranephric fat pads were removed and weighed. Plasma glucose, TG, total cholesterol, and FFA levels were determined using commercial kits (Glucose CII-test Wako, Triglyceride E-test Wako, Cholesterol E-test Wako, and NEFA C-test Wako, respectively, FUJIFILM Wako Pure Chemical Corporation, Tokyo, Japan). After removing the liver, ca. 300 mg of liver tissue was cut and homogenized with 9 mL of distilled water. An aliquot of the homogenate (500 μ L) was diluted with distilled water (1 mL) and the TG concentration in the suspension was determined using Triglyceride E-test Wako.

3.10. Cell Culture

HepG2 cells (RCB1648, Riken Cell Bank, Tsukuba, Japan) were maintained in Minimum Essential Medium Eagle (MEM, Sigma-Aldrich Co. LLC., St. Louis, MO, USA) containing 10% fetal bovine serum, 1% MEM non-essential amino acids (FUJIFILM Wako Pure Chemical Corporation, Tokyo, Japan), penicillin G (100 units/mL), and streptomycin (100 μ g/mL) at 37 °C under 5% CO₂ atmosphere.

3.11. Effects on Glucose Consumption in HepG2 Cells

HepG2 cells were inoculated in a 48-well tissue culture plate (10⁵ cells/well in 150 μ L/well in MEM). After 20 h, the medium was replaced with 150 μ L/well of Dulbecco's Modified Eagle's Medium (DMEM) containing low-glucose (1000 mg/L) and a test sample. Cells were cultured for 6 days, and

the medium was replaced every 2 days. The medium was then transferred to 200 μL /well of DMEM containing high-glucose (4500 mg/L) and the cells were cultured. After 20 h, the glucose content in the medium was determined using commercial kits (Glucose CII-test Wako, FUJIFILM Wako Pure Chemical Corporation, Tokyo, Japan). Medium was removed, and the cells were homogenized in distilled water (105 μL /well) by sonication. The protein content in the homogenate was determined using the BCA protein Assay Kit (FUJIFILM Wako Pure Chemical Corporation, Tokyo, Japan). Each test compound was dissolved in DMSO and added to the medium (final DMSO concentration was 0.5%). An anti-diabetic agent, metformin, was used as a reference compound.

3.12. Effects on High Glucose-Induced TG Accumulation in HepG2 Cells

HepG2 cells were inoculated in a 48-well tissue culture plate (10^5 cells/well in 150 μL /well in MEM). After 20 h, the medium was replaced with 150 μL /well of DMEM containing high-glucose and a test sample, which was cultured for 4 days, with medium containing a test sample being replaced every 2 days. Medium was then removed, and the cells were homogenized in distilled water (105 μL /well) by sonication. The TG and protein content in the homogenate were determined using commercial kits (Triglyceride E-test Wako and BCA protein Assay Kit, respectively, FUJIFILM Wako Pure Chemical Corporation, Tokyo, Japan). Data were expressed as the % of control of TG/protein ($\mu\text{g}/\text{mg}$). Each test compound was dissolved in DMSO and was added to the medium (final DMSO concentration was 0.5%). An anti-diabetic agent, metformin, was used as a reference compound.

3.13. Effects on TG contents in High Glucose-Pretreated HepG2 Cells

Effects on TG metabolism-promoting activity in high glucose-pretreated HepG2 cells were evaluated according to the method described previously [61], with slight modifications. HepG2 cells were inoculated in a 48-well tissue culture plate (10^5 cells/well in 150 μL /well in MEM). After 20 h, the medium was replaced with 150 μL /well of DMEM containing high-glucose and cultured for 6 days, with the medium being replaced every 2 days. After accumulation of the lipid, the medium was transferred to 150 μL /well of DMEM containing low-glucose and a test sample, and the cells were cultured. After 20 h, the TG and protein content in the cells were determined by the same manner as described above. Data were expressed as the % of control of TG/protein ($\mu\text{g}/\text{mg}$). Each test compound was dissolved in DMSO and added to the medium (final DMSO concentration was 0.5%). An anti-diabetic agent, metformin, was used as a reference compound.

3.14. Statistics

Values are expressed as means \pm S.E. One-way analysis of variance (ANOVA) followed by Dunnett's test was used for statistical analysis. Probability (p) values of less than 0.05 were considered significant.

4. Conclusions

The present study demonstrated that helichryoside (**1**), an acylated flavonol glycoside, improved glucose tolerance in ddY mice. In the study, using HepG2 cells, helichryoside (**1**) was shown to significantly enhance glucose consumption in the medium, inhibit high glucose-induced TG accumulation in cells, and promote the effect of TG metabolism in high glucose-pretreated cells. The results from various acylated flavonol glycosides, flavonol glycosides, flavonols, and organic acids indicated that the acyl group at the 6'' position in the D-glucopyranosyl part was essential for the improved glucose tolerance activities. Previous evidence, along with this study, suggests that helichryoside (**1**) might be considered as a possible candidate for the prevention of glucose and lipid metabolism-related disorders.

Supplementary Materials: Supplementary materials can be found at <http://www.mdpi.com/1422-0067/20/24/6322/s1>.

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