

Received: 24 August 2017 Accepted: 9 July 2018 Published online: 27 July 2018

# **OPEN** Ocean acidification drives community shifts towards simplified non-calcified habitats in a subtropical—temperate transition zone

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Rising atmospheric concentrations of carbon dioxide are causing surface seawater pH and carbonate ion concentrations to fall in a process known as ocean acidification. To assess the likely ecological effects of ocean acidification we compared intertidal and subtidal marine communities at increasing levels of pCO<sub>2</sub> at recently discovered volcanic seeps off the Pacific coast of Japan (34° N). This study region is of particular interest for ocean acidification research as it has naturally low levels of surface seawater pCO<sub>2</sub> (280-320 µatm) and is located at a transition zone between temperate and sub-tropical communities. We provide the first assessment of ocean acidification effects at a biogeographic boundary. Marine communities exposed to mean levels of pCO<sub>2</sub> predicted by 2050 experienced periods of low aragonite saturation and high dissolved inorganic carbon. These two factors combined to cause marked community shifts and a major decline in biodiversity, including the loss of key habitat-forming species, with even more extreme community changes expected by 2100. Our results provide empirical evidence that near-future levels of pCO2 shift sub-tropical ecosystems from carbonate to fleshy algal dominated systems, accompanied by biodiversity loss and major simplification of the ecosystem.

Rising atmospheric concentrations of carbon dioxide are causing surface seawater pH and carbonate ion concentrations to fall in a process known as ocean acidification. Marine organisms are expected to differ widely in their responses to this increase in CO<sub>2</sub> levels; partially due to the fact that altered carbonate chemistry stresses many organisms (e.g. negative effects on calcification, reproduction, feeding rate and early life-stage survival2), but is a resource for others (e.g. enhanced primary production and carbon fixation rates<sup>3</sup>). Most studies have focussed on the effects of ocean acidification on isolated organisms, yet these experiments seldom include community and ecosystem-level interactions and so it is difficult to assess how the results apply to natural ecosystems<sup>4-6</sup>.

Volcanic seeps can provide natural analogues for the effects of ocean acidification on the structure of marine ecosystems because they expose entire communities to a lifetime of elevated CO<sub>2</sub> levels. Use of CO<sub>2</sub> seeps in ocean acidification research has steadily increased over the past decade, with suitable sites now located across temperate<sup>7-10</sup>, sub-tropical<sup>11</sup>, and tropical ecosystems<sup>12,13</sup>. Most marine organisms have a planktonic stage in their life history, so recruitment into high CO2 seep sites will occur from populations that are not genetically adapted to future ocean acidification conditions. Therefore, the communities in the CO2 seep sites that become established will be comprised of organisms that are able to tolerate higher CO2 and lower carbonate saturation states for their entire lives (including those that live for many years). For clonal organisms, multiple generations of asexually reproducing individuals are exposed over periods of years, such as bryozoans and corals<sup>14,15</sup>, and seagrasses<sup>8,16</sup>. A few marine organisms have very limited larval dispersal, and there is evidence that populations of polychaetes

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at  $CO_2$  seeps are genetically distinct<sup>17</sup>, and that populations of molluscs that hatch benthic larvae have adapted to chronic ocean acidification over multiple generations through dwarfism<sup>18</sup>.

Since the beginning of the industrial era, atmospheric  $CO_2$  has increased from ~280  $\mu$ atm to present day levels of 400  $\mu$ atm<sup>1</sup>, and yet our understanding of the effects of ocean acidification that may have already occurred is limited; with most research focussed on potential impacts over the coming century. A recent study restored the carbonate chemistry saturation state of a coral reef flat to near pre-industrial levels and demonstrated that present-day net community calcification of coral reefs is already impaired by ocean acidification<sup>19</sup>. Due to the influence of the northward flowing Kuroshio Current<sup>20,21</sup> our study region in Japan has naturally low levels of surface seawater  $pCO_2$  (280–320  $\mu$ atm), which are near pre-industrial levels based on the global average (280  $\mu$ atm; ref.<sup>1</sup>). This particular chemical setting could therefore provide information on how the increase of  $CO_2$  since the pre-industrial period has already affected ecosystems in other parts of the world.

The presence of ecosystem engineers can modify habitats, promote spatial complexity and facilitate the presence of other species 22,23. Ocean acidification-driven changes to these habitat-forming organisms may therefore interact with the direct effects on those species residing in the habitat, and lead to lower species diversity in coral reefs, mussel beds and macroalgal habitats<sup>5</sup>. Previous studies in CO<sub>2</sub> seeps have demonstrated consistent patterns of ocean acidification impacts on the structure of marine ecosystems, with the observed ecological shifts in the acidified conditions showing a reorganisation of the community including reduced biodiversity, and habitat loss (as well as structural complexity)<sup>5,11,12</sup>. It is currently unclear, however, how those communities located at the boundaries of biogeographic regions are likely to respond to ocean acidification; in these regions many species overlap at their range margins and may demonstrate reduced fitness and performance relative to their range centre. Here, we investigate the effects of ocean acidification at a biogeographic transition zone on the Pacific coast of Japan. This region is a global biodiversity hotspot<sup>26,27</sup> where temperate and subtropical communities overlap, with the co-existence of both canopy-forming fleshy macroalgae and zooxanthellate scleractinian corals. These two groups are key habitat-forming species that provide a complex three-dimensional structure that sustains a diverse ecosystem. Such a location will therefore provide information on the effects of ocean acidification on range limits at subtropical—temperate transition zones globally.

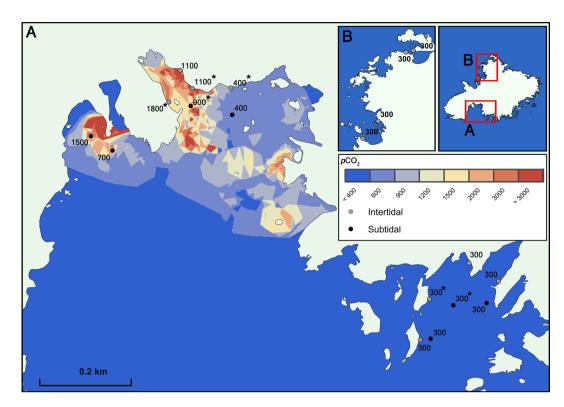
In the present study, we assessed the effects of chronic exposure to ocean acidification on intertidal and subtidal communities around a set of volcanic  $CO_2$  seeps off Shikine Island, Japan. We examined the community composition of benthic marine life at sites with reference levels of 300  $\mu$ atm  $pCO_2$  (near pre-industrial levels) and compared them with areas exposed to increasing levels of carbon dioxide gradually up to end-of-the-century  $pCO_2$  conditions to provide the first chemical and ecological assessment of the impact of ocean acidification at a subtropical—temperate transition zone.

### Methods

**Study Site and Carbonate Chemistry.** Shikine is a volcanic island east of the Izu peninsula in Japan  $(34^{\circ}19'9'' \text{ N}, 139^{\circ} 12'18'' \text{ E})$  with many CO<sub>2</sub> seeps in shallow waters that we surveyed using RV Tsukuba II. Different stations in the intertidal and subtidal zones (3-6 m) below Chart datum) were surveyed and given a classification based on their mean  $p\text{CO}_2$  levels. Those stations with a similar  $p\text{CO}_2$  level were then grouped together for subsequent analysis. The groupings used were 300, 400, 1100 and 1800 μatm  $p\text{CO}_2$  for the intertidal, and 300, 400, 700, 900, and 1500 μatm  $p\text{CO}_2$  for the subtidal. The intertidal stations included eight '300 μatm' stations, one '400 μatm' station, two '1100 μatm' stations and one '1800 μatm' station. The subtidal included three '300 μatm' stations, one '400 μatm' station, one '700 μatm' station, one '900 μatm' station and one '1500 μatm' station (see Fig. 1 for station locations). The abundance and distribution of rocky shore communities varies greatly in both space and time, even along the same stretch of coast<sup>28</sup>. Following the suggestion of refs<sup>29,30</sup> we used multiple reference stations (eight '300 μatm'  $p\text{CO}_2$  in the intertidal, and three '300 μatm'  $p\text{CO}_2$  in the subtidal) to assess the variability of 'normal' rocky shore communities to compare with our high CO<sub>2</sub> stations.

To describe the carbonate chemistry of the survey stations, pH, temperature, salinity and total alkalinity (TA) were measured through *in situ* measurements and/or discrete sampling at the respective stations using both a YSI sensor (YSI Pro Plus, USA) and a TOA-DKK multisensor (WQ-22C, TOA-DKK, Japan) in June 2015. Intertidal stations were surveyed by fixing the sensors to the shore (50 cm below the low water mark), with discrete samples collecting surface water. Subtidal stations were surveyed by fixing the sensors to the seafloor at 5–6 m depth. Discrete samples in the subtidal surveys were taken by SCUBA divers close to the bottom (5–6 m depth). Long term monitoring of pH, temperature, conductivity and dissolved oxygen of the bottom water at the '300 μatm' and '900 μatm' *p*CO<sub>2</sub> stations was carried out in June 2016 using durafet pH sensors (Seafet, Sea-Bird Scientific, Canada) calibrated on the total scale, Hobo conductivity loggers (U24-002-C) and Hobo dissolved oxygen data loggers (U26-001) (Bourne, Onset, USA) with the sensors fixed 30 cm above the seafloor at a depth of 5–6 m. A Horiba multi-parameter meter (U-5000G, Horiba Ltd, Kyoto Japan) coupled with a GPS (eTrex30x, Garmin) was used to document the spatial variation in carbonate chemistry, where we mapped the spatial distribution of *p*CO<sub>2</sub> using the nearest neighbour interpolation algorithm in ArcGIS (ESRI, New York, USA), see Fig. 1.

All pH meters, with the exception of the factory calibrated SeaFET sensors, were regularly calibrated to the NBS pH scale using three buffers 4.01, 7.00 and 10.01 (Thermo Scientific, USA). Total alkalinity of the intertidal stations was measured at the '300  $\mu$ atm' (n=6), '400  $\mu$ atm' (n=4), '1100  $\mu$ atm' (n=4) and '1800  $\mu$ atm' (n=3) stations. For the subtidal zone, bottom water samples were collected for TA measurements at the '300  $\mu$ atm' (n=10), '400  $\mu$ atm' (n=3), '700  $\mu$ atm' (n=11), '900  $\mu$ atm' (n=14), and '1500  $\mu$ atm' (n=14) stations. Water samples were immediately filtered at 0.45  $\mu$ m using disposable cellulose acetate filters (Dismic, Advantech, Japan) and stored at room temperature in the dark until measurement. TA was measured by titration (785 DMP Titrino, Metrohm) with HCl at 0.1 moll<sup>-1</sup>, and then calculated from the Gran function between pH 4.2 and 3.0. The titrations were cross-validated using a working standard (SD:  $\pm$ 9  $\mu$ mol kg<sup>-1</sup>) and against certified reference material purchased from the A.G. Dickson laboratory. Carbonate chemistry parameters were calculated using the CO<sub>2</sub>SYS software<sup>31</sup>. Measured pH<sub>NBS</sub>, TA, temperature and



**Figure 1.** Study area (Shikine-Jima, Japan) showing intertidal and subtidal stations, and the spatial variability in  $pCO_2$ . The spatial distribution of  $pCO_2$  was computed using the nearest neighbour algorithm in ArcGIS 10.2 software (http://www.esri.com/software/arcgis/). '\* Indicates sites where 24-hour measurements of carbonate chemistry were taken.

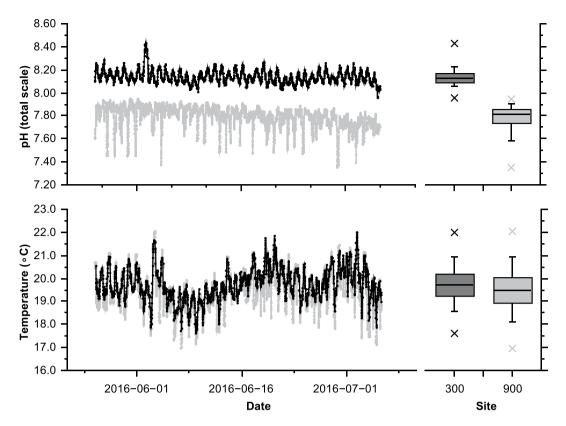
salinity were used as the input variables, alongside the disassociation constants from Mehrbach *et al.*<sup>32</sup>, as adjusted by Dickson *et al.*<sup>33</sup>, KSO<sub>4</sub> using Dickson<sup>34</sup>, and total borate concentrations from Uppström<sup>35</sup>.

**Biological Survey.** Percent cover of intertidal macroflora and encrusting fauna were visually assessed using  $25 \times 25$  cm quadrats. Other sessile macroinvertebrates were counted in  $50 \times 50$  cm quadrats. In both cases, 10-15 replicate quadrats were haphazardly deployed at least a metre apart on steeply sloping rock faces at each station. Macroalgae were grouped as: 'crustose coralline algae', 'non-calcareous encrusting algae' (such as Peyssonneliaceae spp.) and 'fleshy algae' (typically < 5 cm in height such as corticated *Ishige okamurae*, filamentous *Chaetomorpha spiralis*, and foliose *Ulva* spp.). Some fauna were surveyed as % cover, *viz.* 'sponges', 'hard corals', 'barnacles (>1 cm)', 'barnacles (<1 cm)', 'anemones' and 'colonial ascidians'. Groups of invertebrates that were counted as numbers of individuals were 'serpulids', 'spirobids', 'mussels', 'oysters', 'chitons', 'carnivorous gastropods', 'herbivorous gastropods' and 'decapods'. The community structure of shallow subtidal rock was assessed by SCUBA diving using haphazardly placed  $50 \times 50$  cm digital photoquadrats (n = 8-11) at the seven stations of '300 μatm', '400 μatm', '700 μatm', '900 μatm', and '1500 μatm'  $pCO_2$  levels. Photographs were analysed using Image] by overlaying 64 points on a grid, and recording the features and organisms at each point. Coverage in the photoquadrats was grouped as follows: 'canopy-forming algae' (5-50 cm in height), 'low-profile algae' (5 cm in height), 'turf algae' (filamentous algae and microalgae), 'non-calcareous encrusting algae', 'branched coralline algae', 'crustose coralline algae', 'hard corals', 'soft corals' and 'rock'.

In order to assess differences in species richness between the  $CO_2$  levels (since taxonomic groups were used to evaluate community changes across stations), species diversity was assessed during 30-minute searches in the intertidal zone at a '300  $\mu$ atm' station and a '1100  $\mu$ atm' station. Species richness in the subtidal zone at '300  $\mu$ atm', '400  $\mu$ atm' and '900  $\mu$ atm' stations was assessed by identifying the different species observed in the photoquadrats. The taxonomic groups that were assigned to the species observed in the intertidal and subtidal zones are shown in Tables S1 and S2.

For the statistical analyses, the stations were grouped based on their mean  $pCO_2$  classification (see 'Methods: Study Site and Carbonate Chemistry'). These groups were: '300  $\mu$ atm', '400  $\mu$ atm', '1100  $\mu$ atm' and '1800  $\mu$ atm' in the intertidal zone, and '300  $\mu$ atm', '400  $\mu$ atm', '700  $\mu$ atm', '900  $\mu$ atm' and '1500  $\mu$ atm' in the subtidal zone. Taxonomic groups were individually compared across  $CO_2$  levels using the Kruskal-Wallis test, with Fisher's least significant difference (Bonferroni-adjusted) as a *post-hoc* test (statistical significance tested at p < 0.05; with results for individual taxonomic groups presented in full in Table S3). All statistical analyses were performed using R software (Version 3.2.4) with the Kruskal function (agricolae package).

The variation in habitat complexity along  $pCO_2$  gradients was assessed based on the abundance of sessile taxa, along with a rank (between 0 and 5) for that taxon representing the biogenic habitat complexity provided. These ranks were: Minimum habitat complexity = 0, e.g. all encrusting algae; Very low complexity = 1, e.g. small



**Figure 2.** Variation of temperature and pH (total scale) over the month of June 2016 at a subtidal control site ('300  $\mu$ atm') and a subtidal elevated CO<sub>2</sub> site ('900  $\mu$ atm'). Measurements were carried out with the SeaFETs sensors deployed just above the seafloor.

spirorbids and small barnacles; Low complexity = 2, e.g., turf and low-profile fleshy algae, sponges and non calcifying anthozoans; Moderate complexity = 3, e.g. branched coralline algae and sparse oysters; High complexity = 4, e.g. canopy-forming algae, clumps of mussels; Exceptionally high habitat complexity = 5, e.g. hard corals (see Supporting Information, Table S3). The habitat complexity was calculated as follows: the abundance of each taxonomic group was normalised (using the decostand function, vegan package) and then had the rank (0–5 score) applied, providing a habitat complexity score for each quadrat. In order to provide a relative measure across the  $pCO_2$  sites, the habitat complexity score was normalised to between 0 and 1, where the quadrat showing the maximum complexity had a score of 1. These scores were then used to calculate the mean habitat complexity and its variability (standard error) for the different  $pCO_2$  stations.

# Results

**Seawater carbonate chemistry.** There were two areas with permanently high *p*CO<sub>2</sub> located in the northern part of the bay where CO<sub>2</sub> was bubbling up through the seabed (Fig. 2). Our stations closest to these seeps had mean  $pCO_2$  1773  $\pm$  1487  $\mu$ atm ('1800  $\mu$ atm', intertidal) and 1552  $\pm$  540  $\mu$ atm ('1500  $\mu$ atm', subtidal) (Figs 1, S1 and S2; Table 1). Variations in  $pCO_2$  over time were greatest at the highest  $pCO_2$  stations, the  $10^{th}$  and  $90^{th}$  percentiles are shown in Table 1 and boxplots showing the variation ranges of each parameters are shown in Fig. 2 and Supplementary Figs S1 and S2. These high levels of pCO<sub>2</sub> corresponded to mean  $\Omega_A$  values of 1.33  $\pm$  0.67 at station '1800  $\mu$ atm' and  $0.94 \pm 0.33$  at station '1500  $\mu$ atm', with minimum levels of  $\Omega_A$  observed being 0.30 and 0.60, respectively. Farther from the seep sites, intertidal stations had mean  $pCO_2$  levels of  $1182 \pm 672 \,\mu atm$  (station '1100  $\mu$ atm', intertidal) and subtidal stations with mean  $pCO_2$  levels of  $888 \pm 471 \,\mu$ atm (station '900  $\mu$ atm', subtidal) and  $714 \pm 20 \,\mu atm$  (stations '700  $\,\mu atm$ ', subtidal) (Figs 1, S1 and S2; Table 1). At stations with mean  $pCO_2$  levels of  $419 \pm 82 \,\mu$ atm (station '400  $\mu$ atm', intertidal) in the intertidal and  $460 \pm 40 \,\mu$ atm in the subtidal (stations '400  $\mu$ atm') variability in ocean acidification conditions was lower, which corresponded to a mean  $\Omega_A$  of  $2.49 \pm 0.35$  intertidally and  $2.23 \pm 0.31$  subtidally. Stations far from the seeps, had stable levels of pCO<sub>2</sub> with  $299 \pm 51$  µatm station '300 µatm', intertidal) and  $342\pm26\,\mu atm$  station '300  $\mu atm$ ', subtidal), where the mean  $\Omega_A$  recorded were  $3.27\pm0.37$  and  $2.75 \pm 0.14$ , respectively (Fig. 1; Table 1). Monitoring of the subtidal pH and temperature for over a month-period showed an average mean pH<sub>T</sub> of  $8.14\pm0.06$  at a subtidal '300  $\mu$ atm' station and an average mean pH<sub>T</sub> of  $7.79\pm0.10$ at the subtidal '900 µatm' station (Figs 2 and S3). The '300 µatm', '400 µatm' and '900 µatm' ('1100 µatm') stations represent near pre-industrial, present-day, and end of century conditions, respectively. Abiotic parameters such as dissolved oxygen, total alkalinity and depth did not differ across sites. The detailed carbonate chemistry of the surface and subtidal waters is presented in Table 1. Continuous measurements of seawater chemistry for long periods of time at stations marked with "\*" in Fig. 1 are presented in Figs 2, S1, S2 and S3.

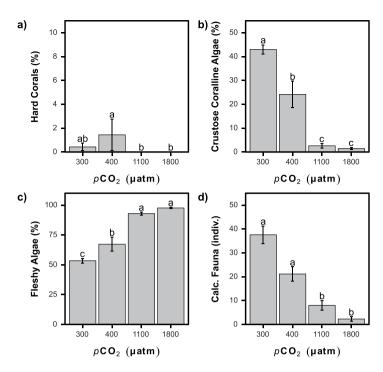
Station	$pH_{NBS}$	Temp. (°C)	Salinity (psu)	TA (μmol kg <sup>-1</sup>	pCO <sub>2</sub> (μatm)	DIC (μmol kg <sup>-1</sup> )	HCO <sub>3</sub> <sup>-</sup> (μmol kg <sup>-1</sup> )	CO <sub>3</sub> <sup>2-</sup> (μmol kg <sup>-1</sup> )	Ωcalcite	Ωaragonite
Intertidal			•	•	,	•			•	•
300	8.27 ± 0.06	$19.4 \pm 0.5$	$34.03 \pm 0.02$	2237 ± 1	298.6 ± 51.0	1938.9 ± 36.3	1719.6 ± 58.1	$209.4 \pm 23.5$	$5.04 \pm 0.57$	$3.27 \pm 0.37$
	(8.15-8.35)	(18.8-20.1)	(34.00-34.05)	(2236-2238)	(237.1-382.6)	(1888.4–1991.6)	(1638.4-1803.2)	(175.7-242.3)	(4.23-5.84)	(2.74-3.79)
400	8.15 ± 0.07	$17.5 \pm 1.1$	33.97 ± 0.09	2250±2	419.2 ± 82.3	$2028.1 \pm 35.9$	1853.0 ± 55.1	$160.5 \pm 22.4$	$3.86 \pm 0.54$	$2.49 \pm 0.35$
	(8.05-8.22)	(16.4-19.0)	(33.86-34.04)	(2248-2252)	(343.7-536.9)	(1985.6-2081.3)	(1786.5-1934.7)	(127.4-187.6)	(3.06-4.51)	(1.97-2.92)
1100	$7.81 \pm 0.22$	$18.5 \pm 0.7$	$33.97 \pm 0.03$	2282±31	1181.5 ± 672.4	$2186.8 \pm 82.4$	2054.4±103.6	$92.3 \pm 42.1$	$2.22 \pm 1.01$	$1.44 \pm 0.66$
	(7.50-8.07)	(17.6-19.5)	(33.94-34.01)	(2244-2326)	(518.1-2171.2)	(2089.9-2292.9)	(1929.3-2175.8)	(43.0-143.1)	(1.03-3.44)	(0.67-2.22)
1800	$7.70 \pm 0.30$	$21.3 \pm 0.4$	$34.30 \pm 0.21$	2261±3	1773.4 ± 1487.0	$2192.6 \pm 105.4$	$2052.5 \pm 105.7$	$84.5 \pm 42.9$	$2.04 \pm 1.03$	$1.33 \pm 0.67$
	(7.22-8.02)	(20.5-21.7)	(34.00-34.46)	(2259-2263)	(601.6-4308.8)	(2074.6-2360.2)	(1915.9-2199.0)	(25.1-140.0)	(0.60-3.37)	(0.39-2.20)
Subtidal	•								•	
300	8.22 ± 0.03	$16.5\pm0.0$	$34.51 \pm 0.03$	2260±3	341.5 ± 26.1	2007.6 ± 14.9	1816.6 ± 22.9	$178.8 \pm 9.0$	$4.28 \pm 0.22$	$2.75 \pm 0.14$
	(8.19-8.25)	(16.5-16.5)	(34.50-34.51)	(2256-2263)	(311.1-372.4)	(1989.0-2026.2)	(1790.1-1844.4)	(167.8-189.6)	(4.01-4.53)	(2.58-2.92)
400	8.11 ± 0.02	$18.1\pm0.2$	$34.75 \pm 0.06$	2252±3	459.5 ± 39.9	$2054.9 \pm 28.7$	1893.8 ± 45.4	$144.3\pm19.1$	$3.45 \pm 0.46$	$2.23 \pm 0.31$
	(8.08-8.13)	(17.8-18.4)	(34.69-34.80)	(2249-2256)	(431.4-503.1)	(2036.5-2066.4)	(1865.3-1911.3)	(137.3-156.1)	(3.28-3.73)	(2.12-2.41)
700	$7.94 \pm 0.01$	$16.3\pm0.1$	34.54 ± 0.05	2272±4	714.3 ± 20.2	2145.3 ± 4.3	2015.7 ± 5.7	$103.8 \pm 2.5$	$2.48 \pm 0.06$	$1.60\pm0.04$
	(7.92-7.96)	(16.2-16.3)	(34.50-34.60)	(2269-2276)	(686.2-743.0)	(2139.5-2150.1)	(2007.6-2021.8)	(100.5-107.3)	(2.40-2.57)	(1.54-1.65)
900	$7.90 \pm 0.17$	$19.0\pm0.7$	$34.47 \pm 0.06$	2270 ± 2	888.4 ± 471.3	2140.5 ± 61.2	2002.9 ± 78.7	$108.0 \pm 31.9$	$2.59 \pm 0.76$	$1.68 \pm 0.49$
	(7.69-8.09)	(18.3-19.8)	(34.40-34.50)	(2267-2271)	(490.5-1382.8)	(2067.5-2219.7)	(1902.2-2108.5)	(65.3-148.8)	(1.57-3.57)	(1.02-2.31)
1500	$7.66 \pm 0.15$	$17.5\pm0.2$	34.54±0.05	$2248 \pm 10$	1552.1 ± 539.5	2212.7 ± 49.5	2097.7 ± 52.4	$60.9 \pm 21.2$	$1.46 \pm 0.51$	$0.94 \pm 0.33$
	(7.47-7.87)	(17.2-17.6)	(34.50-34.60)	(2238-2263)	(855.8-2302.2)	(2142.6-2271.8)	(2020.8-2153.1)	(38.5-92.0)	(0.92-2.21)	(0.60-1.42)

Table 1. Carbonate chemistry in subtidal and intertidal waters off Shikine Island, Japan. pH<sub>NBS</sub>, temperature, salinity, and total alkalinity (TA) are measured values. Seawater  $pCO_2$ , dissolved inorganic carbon (DIC), bicarbonate (HCO<sub>3</sub><sup>-</sup>), carbonate (CO<sub>3</sub><sup>2-</sup>), saturation states for calcite (Ωcalcite) and aragonite (Ωaragonite) are values calculated using the carbonate chemistry system analysis program CO2SYS. Values are presented as mean  $\pm$  S.D. with 10<sup>th</sup> and 90<sup>th</sup> percentiles.



**Figure 3.** Representative ecological communities at increasing  $pCO_2$  levels. The top panels represent intertidal communities associated with mean levels of 300, 400 and 1100  $\mu$ atm  $pCO_2$ . The bottom panels represent subtidal communities associated with mean levels of 300, 400 and 900  $\mu$ atm  $pCO_2$ .

**Intertidal communities.** The rocky shores of Shikine Island were characterised by thick biogenic carbonate crusts formed by coralline algae, serpulids, barnacles and molluscs with an overgrowth of fleshy algae on the low shore in the intertidal zone. There was a significant difference in the abundance of macroflora and encrusting macrofauna between the different CO<sub>2</sub> levels (Figs 3 and 4). One of the most notable shifts in community



**Figure 4.** Changes in abundance (mean  $\pm$  SE) of taxa (mean  $\pm$  SE) with increasing  $pCO_2$  for the intertidal habitat. (a) Hard corals. (b) Coralline algae. (c) Fleshy algae. (d) Calcified fauna. A significant difference between  $pCO_2$  groups is indicated with a different letter (Kruskal-Wallis with Bonferroni-adjusted Fisher's least significant difference).

composition was a very clear decrease in the coralline algae as  $CO_2$  levels increased; this cover fell from  $43\pm21\%$  at '300  $\mu$ atm' to  $24\pm21\%$  at 400  $\mu$ atm,  $3\pm5\%$  '1100  $\mu$ atm' and  $1\pm2\%$  at '1800  $\mu$ atm', respectively (K-W: H=99.21, p<0.001; Figs 3 and 4a). There was also a significant decrease in sponge cover as  $CO_2$  levels increased (K-W: H=19.27, p<0.001; Fig. S4). Despite reductions in coralline algae and sponges, the amount of rock covered solely by biofilm remained low throughout the  $CO_2$  gradient, with  $2\pm6\%$  at '300  $\mu$ atm',  $4\pm6\%$  at '1100  $\mu$ atm' and  $1\pm2\%$  at '1800  $\mu$ atm'  $CO_2$  (K-W: H=11.46, p<0.01; Fig. S4). Fleshy algae increased in abundance from  $53\pm21\%$  at '300  $\mu$ atm' to  $98\pm3\%$  at '1800  $\mu$ atm' (K-W: H=47.57, p<0.001; Fig. 3c), and non-calcareous encrusting algae increased from  $4\pm7\%$  at '300  $\mu$ atm' to  $18\pm27\%$  and  $11\pm27\%$ , at '1100  $\mu$ atm' and '1800  $\mu$ atm', respectively (K-W: H=16.77, p<0.001; Fig. S4).

Most of the macrofauna investigated showed significant changes in abundance along the gradients in CO<sub>2</sub>. The abundance of calcifying organisms decreased as pCO<sub>2</sub> levels rose with a significant difference between the '300  $\mu$ atm' and '400  $\mu$ atm' versus the '1100  $\mu$ atm' and '1800  $\mu$ atm', with the number of individuals falling from  $38\pm38$  individuals at '300  $\mu$ atm' to  $2\pm3$  individuals at '1800  $\mu$ atm' (K-W: H=63.73, p<0.001; Figs 3 and 4d). The abundance of the larger barnacles (>1 cm in diameter including Megabalanus volcano and Megabalanus rosa) fell from  $18\pm24$  at '300  $\mu$ atm',  $15\pm8$  at '400  $\mu$ atm' and  $5\pm10$  individuals per 0.25 m² at '1100  $\mu$ atm', and they were almost absent at '1800  $\mu$ atm' (K-W: H=33.42, p<0.001; Fig. S4). Mussels were also reduced in abundance from  $7\pm13$  individuals at '300  $\mu$ atm' to  $2\pm3$  individuals in the '1800  $\mu$ atm' (K-W: H=20.80, p<0.001; Fig. S4). Oysters and decapods were less common at the elevated CO<sub>2</sub> stations but our surveys did not reveal statistically significant differences (K-W, p>0.05) as they were not very abundant in quadrats at '300  $\mu$ atm' (Fig. S4). The azooxanthellate coral *Tubastrea coccinea* was observed at '300  $\mu$ atm' and '400  $\mu$ atm' but absent at '1100  $\mu$ atm' and '1800  $\mu$ atm' pCO<sub>2</sub> (K-W: H=14.32, p<0.05; Fig. S4). Of the non-calcifying fauna, only sea anemones significantly increased in abundance at '1100  $\mu$ atm' (0.1  $\pm$  0.4 individuals per 0.25 m² at 300  $\mu$ atm vs. 1.0  $\pm$  1.4 at '1100  $\mu$ atm'), they were absent from the '1500  $\mu$ atm' station (K-W: H=53.89, p<0.001; Fig. S4).

Intertidal habitat complexity was provided by barnacles, mussels and oysters, and coralline algae (which formed a thick crust) at '300  $\mu$ atm'. These calcifying groups drastically decreased in abundance with an overall shift in the communities as  $pCO_2$  levels rose (Figs 3 and 5b). These shifts lead to a decrease in the complexity of the habitat (K-W: H=48.50, p<0.001), reducing two-fold from '300  $\mu$ atm' to '1100  $\mu$ atm', and more than 6-fold to '1800  $\mu$ atm' (Fig. 5a). At the high  $CO_2$  levels, the main habitat was low-profile fleshy algae (Figs 3 and 5a).

Despite the increasing abundance of fleshy algae with rising  $pCO_2$ , there was a 56% reduction in algal species richness from '300  $\mu$ atm' (18 spp.) to '1100  $\mu$ atm' (8 spp.) with little overlap in species composition between these two  $CO_2$  levels (Fig. 5c and Table S1). The '300  $\mu$ atm' stations had a diverse community of Rhodophyta with 16 species compared to only six species at '1100  $\mu$ atm'. There were 33 and 32 macrofaunal taxa at '300  $\mu$ atm' and '1100  $\mu$ atm' respectively, but the community composition was very different, with only seven species common to both sets of stations (Fig. 5c and Table S1).

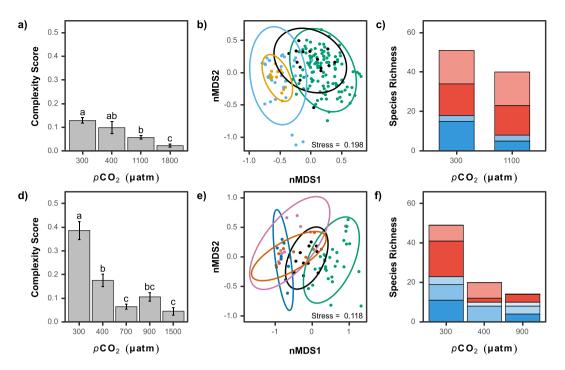


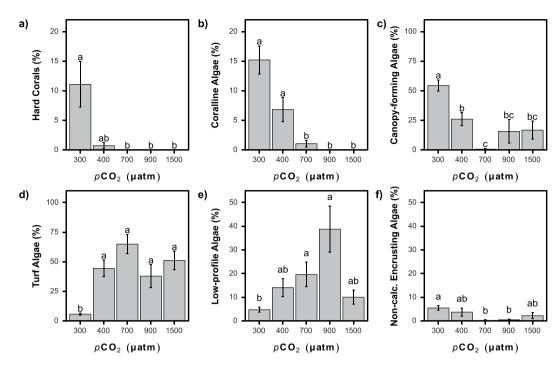
Figure 5. Changes in habitat complexity (mean  $\pm$  SE), communities, and species richness with increasing  $pCO_2$  for intertidal (**a–c**) and subtidal (**d–f**) habitats. (**a,d**) A significant difference between  $pCO_2$  groups is indicated with a different letter (Kruskal-Wallis with Bonferroni-adjusted Fisher's least significant difference). (**b,e**) The change in communities are illustrated by an nMDS plot based on Bray Curtis distance. The colour of each point represents the  $pCO_2$ : green: '300 μatm', black: '400 μatm', light blue: '1100 μatm' and orange: '1800 μatm' for the intertidal and green: '300 μatm', black: '400 μatm', blue: '700 μatm', red: '900 μatm' and pink: '1500 μatm' for the subtidal. Ellipses represent the 95% interval confidence. (**c,f**) Algal (blue) and faunal (red) species richness are shown with darker colours used for species only found in that site, and lighter shades for species that overlap across two sites. For the subtidal, the species overlap are graduated (from darkest to lightest) in the following order: 300–400 μatm, 300–900 μatm and 400–900 μatm (no species were common to all three sites).

**Subtidal zone.** Changes in the subtidal benthic community were remarkably similar to those observed intertidally, with reduced abundances of calcifying organisms as  $CO_2$  levels increased from '300 μatm' to '400 μatm' and again to '700 μatm' and beyond (Figs 3 and 6). The cover of coralline algae (K-W: H=42.46, p<0.001; Fig. 6b) and hard corals (K-W: H=19.67, p<0.001; Fig. 6a) was significantly reduced as  $CO_2$  levels rose. Hard corals were common at '300 μatm', where they had  $11\pm22\%$  cover, however, they were only sporadically found at '400 μatm' with just two colonies observed accounting for  $0.7\pm1.6\%$  cover, and absent from more highly elevated  $CO_2$  stations (Figs 3 and 6a). Soft corals and anemones were not recorded in the elevated  $CO_2$  stations corresponding to the end-of-the-century projections ('700 μatm', '900 μatm' and '1500 μatm') and were rare at '300 μatm' and '400 μatm'. Due to their low abundance at '300 μatm' and '400 μatm', the soft corals and the anemones did not significantly differ with  $CO_2$  level (K-W: H=6.32, p=0.18 and H=5.09, p=0.27 respectively) (Fig. S4).

The cover of non-calcifying macroalgae was high at all subtidal stations yet there were major shifts in community composition (Figs 3 and 6c-f). Large canopy forming macroalgae had significantly reduced abundance at '400  $\mu$ atm' and higher  $pCO_2$  end-of-the century projections (K-W: H=18.53, p<0.001, Fig. 6c) whereas low-profile algae and turf algae increased in cover as  $CO_2$  levels rose, with their cover significantly higher at '400  $\mu$ atm' and higher  $CO_2$  level stations compared to '300  $\mu$ atm' stations (K-W: H=23.41, p<0.001 and H=44.81, p<0.001 Fig. 6d,e). Due to the overall reduction in the percentage of calcified and non-calcifying macroalgae, the proportion of biofilm encrusted substrata significantly increased from  $2\pm3\%$  to  $20\pm14\%$  at '300  $\mu$ atm' and '1500  $\mu$ atm' respectively (K-W: H=32.88, p<0.001; Fig. S5).

Both hard corals and canopy forming macroalgae formed a biogenically complex habitat in the subtidal zone at '300  $\mu$ atm' CO<sub>2</sub> (Fig. 3). The sharp decrease of these two groups lead to significantly reduced habitat complexity (K-W: H=50.48, p<0.001) at CO<sub>2</sub> levels corresponding to the mid- ('400  $\mu$ atm') and end-of-the-century projections ('700  $\mu$ atm', '900  $\mu$ atm' and '1500  $\mu$ atm') (Figs 3 and 5c). The communities radically changed as pCO<sub>2</sub> rose with distinct communities observed at each pCO<sub>2</sub> site (Fig. 5d) with less complex low-profile and turf algae dominating at the highest pCO<sub>2</sub> (Fig. 3).

The species richness of the benthic flora and fauna was reduced by 71% as CO<sub>2</sub> levels rose, with a total of 49, 20 and 14 species at '300 µatm', '400 µatm' and '900 µatm', respectively (Fig. 5f and Table S2). This change in faunal species richness included seven hard coral species, a sea anemone, a soft coral species and a sponge, which were only observed at '300 µatm'. In addition, small gastropods were abundant at '300 µatm', but not at '400 µatm' or '900 µatm'. Other mobile benthic fauna that were only found in the '300 µatm' stations were sea cucumbers and coral boring serpulids and barnacles. Algal diversity was greatly reduced shifting from a diverse



**Figure 6.** Changes in abundance (mean  $\pm$  SE) of taxa with increasing  $pCO_2$  for the subtidal habitat. (a) Hard corals. (b) Coralline algae. (c) Canopy-forming fleshy algae. (d) Turf algae. (e) Low-profile fleshy algae. (f) Non-calcified encrusting algae. A significant difference between  $pCO_2$  groups is indicated with a different letter (Kruskal-Wallis with Bonferroni-adjusted Fisher's least significant difference).

community with 22 species at '300 µatm' to only 10 species at both the '400 µatm' and '900 µatm' stations, with only four species overlapping in species composition between '400 µatm' and '900 µatm' (Fig. 5f and Table S2). This is an underestimate of algal diversity as crustose coralline algae were counted as one taxon and most turf or very low-profile algae were not included, exceptions were larger algae such as *Lobophora variegata* and *Caulerpa chemnitzia* var. *peltata*. At '900 µatm', the only abundant canopy forming macroalgae was the red algae *Grateloupia elata* (Table S2).

# Discussion

Our comparisons of intertidal and subtidal rocky reef communities along natural gradients in  $\mathrm{CO_2}$  have revealed that ocean acidification is a threat to many marine organisms, as it can drive fundamental shifts in coastal marine ecosystems towards simplified, low diversity communities. Abrupt changes in subtidal and intertidal communities were revealed from present-day to near-future levels of  $\mathrm{CO_2}$  (300  $\mu$ atm to 400  $\mu$ atm), and then again to future levels (400  $\mu$ atm to 700  $\mu$ atm) and beyond. In natural coastal ecosystems, mean  $p\mathrm{CO_2}$  levels predicted for as soon as the mid-century will have periods of such low aragonite saturation and high availability of inorganic carbon that this will cause biodiversity loss driven by a decline in habitat-forming species (e.g. coralline algae, canopy-forming macroalgae, scleractinian corals, and barnacles) and an increase in low-profile and turf algae. Our observations suggest that ocean acidification will shift ecosystems at subtropical—temperate transition zones from complex calcified biogenic habitats towards less complex non-calcified habitats.

Increases in dissolved  $CO_2$  provide a resource for algae that cannot use bicarbonate ions for their photosynthesis<sup>37</sup> and is expected to increase the prevalence of macroalgae<sup>8,9,12,38</sup>. The significantly increased occurrence of low-profile fleshy algae with increasing  $pCO_2$  aligns with results from other shallow marine carbon dioxide seeps. However, not all macroalgae species respond in the same manner to the effects of elevated  $CO_2$ , with some species gaining a relative advantage over their counterparts<sup>3</sup>. The resulting pattern is that ocean acidification alters successional development due to competition for space by a few highly tolerant species<sup>39,40</sup>. The prevalence of low-profile fleshy algae in our elevated  $pCO_2$  sites may contribute towards the observed decline in canopy-forming macroalgae and corals<sup>41</sup>. In this context, natural analogues offer opportunities to assess competitive interactions and the effects of ocean acidification on ecological functions<sup>40</sup>.

The presence of highly calcified communities at all of our reference sites reflects the high carbonate saturation levels that typify this region due to naturally low background  $pCO_2$  levels<sup>20</sup>. At the highest  $CO_2$  stations the exposed shells and skeletons of calcifying organisms had visible signs of dissolution, as seen in other field studies worldwide<sup>42–45</sup>. The decrease in the abundance of calcifying macrofauna from our reference sites to '400  $\mu$ atm' sites, where even the lowest  $\Omega_{aragonite}$  remains higher than values typically observed nowadays in many parts of the ocean <sup>46,47</sup>, suggests that ocean acidification is already impairing the growth and survival of calcifiers<sup>19</sup>. This is a concern and provides an insight into the effects of ocean acidification in other parts of the world that have already experienced increases in  $pCO_2$  from 300  $\mu$ atm to 400  $\mu$ atm during the last century since the Industrial Revolution.

Communities of zooxanthellate scleractinian corals currently thrive at high latitude (here  $34^{\circ}$  N) in East Asia due to warm, northward flowing currents which bring low  $pCO_2$ , high carbonate saturated waters into the region<sup>48</sup>. These communities are an important reservoir of diversity for hermatypic corals, and a number of species found at our study site are endemic to the region<sup>49</sup>. We observed an abrupt decline in their abundance and diversity as  $CO_2$  levels rose and  $CaCO_3$  saturation state fell. Despite differences in biogeography, the major ecosystem changes we recorded along the  $CO_2$  gradients are broadly consistent with findings from other naturally acidified tropical coral reef settings<sup>12,13,50</sup>. Moreover, these patterns are comparable to those seen on tropical reefs in Florida, where present-day seasonal reductions in saturation state are contributing to reef dissolution, the die-back of scleractinians and an increase in low-profile fleshy algal growth<sup>44,51</sup>.

The Japanese subtropical-temperate transition zone is highly diverse due to a mix of subtropical and temperate species, which allows for the coexistence of diverse macroalgae with scleractinian zooxanthellate corals. This zone is at the leading-edge for subtropical species and the trailing-edge for temperate species, and this biogeographic boundary is likely to undergo fundamental shifts with future climate change<sup>52,53</sup>. With increased temperature threatening corals in the tropics<sup>54</sup>, it could be expected that higher latitudes will act as refugia, but this would require the loss of other ecologically important species that typically dominate these latitudes<sup>55</sup>. Our results support the notion that ocean acidification may constrain the shift of coral to higher latitudes<sup>56–58</sup>.

Biogenic complexity promotes the provisioning of habitats, allowing high levels of biodiversity to be sustained within an ecosystem<sup>5</sup>. Reductions in habitat complexity cause a reduction in biodiversity<sup>5,59</sup>. We found that as  $CO_2$  levels rose, there was a shift from structurally complex canopy-forming fleshy algae and corals to less complex low-profile fleshy algae and an absence of corals. This reduction in habitat complexity may have contributed towards the reduced species richness in our elevated  $pCO_2$  sites, as well as the minimal overlap in observed species among the different sites as many marine organisms rely on a particular habitat (e.g. ref.<sup>60</sup>). As the effect of ocean acidification could cause a simplification of the ecosystems, we can expect ocean acidification to also alter the delivery and the quality of the ecosystem services associated with these marine communities<sup>61</sup> and this should be a focus of future work.

Carbon dioxide seeps are open systems that allow recruitment from outside and this hinders genetic adaptation  $^{62}$ . Whilst organisms that survive at such seeps may upregulate genes to acclimate to high  $pCO_2$  levels  $^{63}$ , only species with very limited genetic dispersal can be expected to evolve to cope with the local conditions  $^{64}$ . The  $CO_2$  seep systems described in this report can nevertheless provide insights into how marine ecosystems have been changing under increased anthropogenic  $CO_2$  and into the near future. Thus, reference sites showed pre-industrial levels of  $CO_2$  and sites on the fringe of the  $CO_2$  gradient showed present-day and mid-century  $CO_2$  levels with minimum variations of these levels on short periods of time. Extreme variations of  $pCO_2$  concentrations at natural analogues are commonly observed  $^{11,13}$  and may bias the observed response of organisms  $^{65}$ . Such extreme variations in  $pCO_2$  levels were also observed at the highest  $pCO_2$  sites of the Shikine  $CO_2$  systems, yet we also located stations with small increases and variations in  $pCO_2$  that are well suited to projected levels of ocean acidification.

In conclusion, we found that an increase in  $\mathrm{CO}_2$  resulted in profound community-level changes in a biodiverse subtropical-temperate transition zone. Both intertidal and subtidal communities became highly simplified, with reduced biogenic habitat complexity and biodiversity. We highlight that ocean acidification may constrain tropical coral range expansion. Our findings suggest that a threshold for macroalgal and coral habitats at the subtropical-temperate transition zone is likely to be exceeded by 2050, with even more extreme changes expected by the end-of-the-century. Overall, ocean acidification is expected to simplify coastal marine communities throughout East Asia.

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# **Acknowledgements**

The authors would like to thank the technical staff of Shimoda Marine Research Center, University of Tsukuba for field assistance and Prof. Hiroyuki Fujimura, University of the Ryukyus, for total alkalinity measurements. This research was part of the 'Japanese Association of Marine Biology' project. This study was partially supported by the 'International Educational and Research Program', University of Tsukuba. Travel costs were funded the Daiwa Foundation (Grant Number: 10777/11517) for J.M.H.-S., and by a Japan Society for the Promotion of Science Short Term Invitation Fellowship (Grant Number: S16073) for M.M.

# **Author Contributions**

S.A., B.P.H. and J.M.H.-S. wrote the manuscript. S.A., S.W., K.K., M.M. and J.M.H.-S. designed and conducted the surveys. S.A., B.H., S.W., K.K. and K.I. analyzed the data. All authors contributed substantially to revisions.

# **Additional Information**

**Supplementary information** accompanies this paper at https://doi.org/10.1038/s41598-018-29251-7.

Competing Interests: The authors declare no competing interests.

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