

Citation: Wang Y, Jiang S, Wang H, Bie H (2017) A mucoadhesive, thermoreversible *in situ* nasal gel of geniposide for neurodegenerative diseases. PLoS ONE 12(12): e0189478. <u>https://doi.org/10.1371/journal.pone.0189478</u>

Editor: Bing Xu, Brandeis University, UNITED STATES

Received: May 14, 2017

Accepted: November 27, 2017

Published: December 14, 2017

Copyright: © 2017 Wang et al. This is an open access article distributed under the terms of the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original author and source are credited.

Data Availability Statement: All relevant data are within the paper and its Supporting Information files.

Funding: This work was supported by The Natural Science Foundation of Shandong Province, ZR2014HL095 (http://www.sdnsf.gov.cn/sdnsfms/ jsp/loginUp2.jsp). The funders had no role in study design, data collection and analysis, decision to publish, or preparation of the manuscript.

Competing interests: The authors have declared that no competing interests exist.

RESEARCH ARTICLE

A mucoadhesive, thermoreversible *in situ* nasal gel of geniposide for neurodegenerative diseases

Yingting Wang, Shulong Jiang*, Hongli Wang, Haiyan Bie

Jining No.1 People's Hospital, Shandong Province, China

* 3416048269@qq.com

Abstract

Neurodegenerative diseases are becoming prevalent as the population ages. Geniposide could inhibit oxidative stress, reduce apoptosis, protect neuron, and has been used for therapy of the neurodegenerative diseases. The bioavailability of geniposide by nasal route is greater than that by oral administration. However, mucociliary clearance is a rate-limiting factor for nasal route administration. The objective of this study was to develop and evaluate a mucoadhesive, thermoreversible in situ nasal gel of geniposide. The poloxamers (P407, P188) and the hydroxypropyl methylcellulose were used as thermoreversible and mucoadhesive polymers, respectively. Borneol was used as a permeation enhancer. The hydrogel was prepared with the cold method and optimized by the response surface methodologycentral composite design. Gelation temperature, pH, clarity, gel strength, mucoadhesive strength, in vitro and ex vivo release kinetics of formulations were evaluated. The optimized amounts of poloxamer407 (P407), poloxamer188 (P188) and hydroxypropyl methylcellulose were determined to be 19.4-20.5%, 1.1-4.0% and 0.3-0.6% respectively. The secondorder polynomial equation in terms of actual factors indicated a satisfactory correlation between the independent variables and the response ($R^2 = 0.9760$). An ANOVA of the empirical second-order polynomial model indicated the model was significant (P<0.01). P407, P188, P407×P188, P407² and P188² were significant model terms. The effects of P407 on gelation temperature were greater than those of other independent variables. The pH values of all the formulations were found to be within 6.3-6.5 which was in the nasal physiological pH range 4.5–6.5. The drug content, gel strength, mucoadhesive strength of the optimized formulations were 97-101%, 25-50 sec and 4000-6000 dyn/cm² respectively. The in vitro release kinetics of cumulative release of geniposide was fitted to the zeroorder model. The ex vivo cumulative release kinetics of geniposide was fitted to the Weibull model. This study concludes that the release of geniposide is controlled by gel corrosion, and that the permeation of geniposide is time-dependent. The more residence time, mucoadhesive, thermoreversible in situ nasal gel of geniposide for neurodegenerative diseases is of compliance and potential application.

Introduction

Neurodegenerative diseases are becoming prevalent as the population ages. The neurodegenerative diseases include Alzheimer's disease (AD), Parkinson's disease (PD), amyotrophic lateral sclerosis (ALS), frontotemporal dementia (FTD) and so on [1-4]. Geniposide is a pharmacologically active compound in Gardenia jasminoides Ellis (Rubiaceae) used for the treatment of hepatic disease, inflammation disorders, contusions and brain disorders [5-7]. Accumulated research data showed geniposide could inhibit oxidative stress and mitochondrial dysfunction [8], improve cognition [9], inhibit the interaction between amyloid-beta peptide and RAGE, attenuate A β -induced neuronal injury [10,11]. The order of absolute bioavailability of geniposide was oral (F = 9.74%) < nasal drops (F = 49.54%) < intramuscular (F = 72.69%). The nasal route delivered geniposide to brain directly through the olfactory region [12]. The pharmacokinetics parameters of intranasal (i.n.) and intragastric (i.g.) administration were compared with those of intravenous (i.v.) administration. The bioavailabilities of geniposide were 85.38% (i.n.) and 28.76% (i.g.) [13]. Nasal delivery of drugs could improve better patient compliance than intravenous (i.v) administration [14]. The nasal route has been considered as a viable and efficacious alternative for drugs which have extensive first pass metabolism [15]. One of the major disadvantages to deliver drugs through nasal route is the mucociliary clearance [16]. To address this issue, mucoadhesive in situ gel formulation was devised to increase the residence time in the nasal cavity [17]. The bioavailability of geniposide might be promoted if the drug remains longer time inside the nasal cavity. The borneol could increase transportation of geniposide across the human nasal epithelial cell [18, 19]. Poloxamer was temperature-triggered, nontoxic, nonirritating and non-sensitizing polymer [20-22]. Poloxamer and hydroxypropyl methylcellulose have been used for the in situ gel [23-25]. Until now, mucoadhesive, thermoreversible in situ nasal gel of geniposide has not been reported.

Taking the above factors into consideration, the present study developed and evaluated a more residence time, mucoadhesive, thermoreversible *in situ* nasal gel of geniposide for the neurodegenerative diseases.

Materials and methods

Materials

Geniposide (95.0% purity) and borneol (86.3% purity) were provided by the First People's Hospital in Jining. Geniposide standard (97.5% purity) was purchased from the National Institute for Food and Drug Control (Beijing, China). Poloxamers (P407, P188) were purchased from Sinopharm Chemical Reagent Co., Ltd (Shanghai, China). Hydroxypropyl methylcellulose (HPMC K4M) was purchased from Anhui Sunhere Pharmaceutical Excipients Co., Ltd (Anhui, China). Acetonitrile (HPLC grade) was purchased from Avantor Performance Materials Trading Co., Ltd (Shanghai, China). Benzalkonium chloride (BC) and sodium chloride were purchased from Sinopharm Chemical Reagent Co., Ltd (Shanghai, China). All other reagents were of analytical grade.

Determination of geniposide

Geniposide was quantified by high performance liquid chromatography equipped with Waters e2695 separation module and 2998 photodiode array detector (Waters, the United States). The chromatographic separation was achieved using a Waters X-Bridge C18 column (5 μ m, 4.6 mm × 250 mm). The mobile phase was an acetonitrile: purified water (15: 85) mixture. The detector was set 238 nm. The column temperature was controlled 25°C. The volume of each injection was 20 μ l.

Preparation of formulation

The cold method was adopted for preparing formulation [26–28]. First of all, purified water was stored overnight at 4°C in refrigerator. Secondly, the required amount of poloxamer (P407 and P188) was slowly added into the required volume of cold purified water with continuous stirring, and then the dispersion was kept overnight at 4°C until a transparent hydrogel was obtained. Thirdly, the required amount of hydroxypropyl methylcellulose, benzalkonium chloride (0.001%, w/v) and sodium chloride (0.9%, w/v) were dispersed into the hydrogel with continuous stirring. Finally, geniposide and borneol were added into the upper solution. The volume was adjusted, and then stored in refrigerator again until transparent formulations were prepared.

Clarity of formulation

Formulation was observed visually under black and white background. The clarity of formulation was graded as follows: turbid: +, clear: ++, very clear (glassy): +++ [29].

pH of formulation

The pH value of formulation was determined by using the pH meter (HANNA, P211). The pH meter was first calibrated using solutions of pH 7.01 and pH 4.01.

Gelation temperature of formulation

The gelation temperature of formulation was determined by the tube inversion method as reported previously [30, 31]. Briefly, the hydrogel (0.5 ml) was transferred to small vial (2 ml, $12 \text{ mm} \times 32 \text{ mm}$), which was sealed and immersed into a thermostat controlled-electric water bath at an initial temperature of 20°C. The temperature of the water bath was increased in increment of 0.5°C/min. The mercury bulb of a thermometer with a minimum readable scale of 0.2°C was placed at the same level with the hydrogel. The meniscus of the hydrogel didn't move when the vial was tilted 90 degree angle. The temperature on the thermometer was identified as the gelation temperature. The gelation temperature was recorded and measured in triplicate.

Gel strength of formulation

The gel strength of formulation was determined as reported previously [32-34]. A sample of 50 g of hydrogel was put into a 100 ml graduated cylinder and gelled in a thermostatically controlled water bath at $37\pm0.5^{\circ}$ C. A weight of 35 g was placed onto the gel. The gel strength, which was an indication of viscosity, was determined by the time in second required by the weight to penetrate 5 cm into the gel.

Mucoadhesive strength of formulation

The mucoadhesive strength of formulation was determined as reported previously [35, 36]. A section of goat nasal mucosa was obtained from local slaughter house immediately after its sacrificesacrifice. Two cylindrical glass vials with 2 cm diameter and modified balance instrument were taken. The goat nasal mucosa was tied to one side of the both vials. Fifty milligrams of hydrogel was placed on one nasal mucosa of one vial. The two vials' nasal mucosa were attached together for 2 min. Water was poured drop by drop into the container of the balance instrument until the two vials got detached from each other. The water was weighed. The mucoadhesive strength of formulation was expressed as the detachment stress in dyne/cm²,

and was determined by minimal.

Mucoadhesive strength (dyne/cm²) = $m \times g/A$ (m = Weight required for detachment of two vials in grams, g = Acceleration due to gravity (980 cm/s²), A = The area of nasal mucosa exposed).

In vitro release kinetics of formulation

In vitro corrosion of *in situ* gel and the release of geniposide from the *in situ* gel were studied simultaneously through a membraneless method as reported previously [37-39]. The membraneless model allowed the release medium to directly contact the gels surface. The cold hydrogel (5 g) was transferred into a graduated test tube (1 cm diameter), which was placed in water bath ($34\pm0.5^{\circ}$ C) and maintained 10 min. A saline phosphate buffer (pH = 6.4, 2.5 ml) used as release medium, pre-equilibrated at $37\pm0.5^{\circ}$ C, was layered over the surface of the gel. After removing the medium at predetermined one-hour interval, the test tube was cleared, weighted and layered with fresh saline phosphate buffer (2.5 ml). The repeated test procedure was finished until less than 10% of the gel was remained. The amount of geniposide in the samples was determined by the high-performance liquid chromatography method. The experiment was performed in triplicate.

Cumulative gel dissolved rate = cumulative gel dissolved/initial gel \times 100%

Cumulative geniposide released rate = cumulative geniposide released/initial geniposide \times 100%

Ex vivo drug permeation

Ex vivo drug permeation was studied as previously reported [40-42]. Nasal cavity of goat was obtained from local slaughter house. It was safely transported to laboratory by keeping it in the saline phosphate buffer (pH6.4). The intact nasal mucosa was separated, cleaned and stored in the saline phosphate buffer. The study was conducted using a Franz diffusion system (RYJ-12B, Shanghai China). The nasal mucosa was fixed on the Franz diffusion cell having effective permeation area of 2.8 cm². After 30 min of incubation time, the optimized formulation 0.5 g was placed in the donor compartment. The temperature of the chamber was maintained at $34\pm0.5^{\circ}$ C. The saline phosphate buffer (pH6.4, 6.5 ml) was used as receptor medium. Receptor medium 1 ml was withdrawn from the receptor chamber at the predetermined 30 min interval, and immediately replaced by the fresh saline phosphate buffer maintained at $34\pm0.5^{\circ}$ C. The geniposide of sample was determined by the high- performance liquid chromatography method. The experiment was performed in triplicate. The *ex vivo* geniposide permeation data were evaluated in different mathematical models.

http://dx.doi.org/10.17504/protocols.io.ks2cwge.

Results

Pre-formulation on gelation temperature

As shown in <u>Table 1</u>, the poloxamer exhibited the phenomenon of reverse thermal gelling under a certain concentration and temperature. The formulation with P407(16.0–24.0%) and P188 (1.0–8.0%) formed a semisolid transparent gel at a certain temperature. The gelation temperature decreased gradually as the P407 concentration increased. However, the temperature increased as the P188 concentration increased. Therefore, P188 was used in combination with P407 to regulate the gelation temperature for more suitable gel formulations.

P188(%, w/v)	P407 (%, w/v)				
	16.0	18.0	20.0	22.0	24.0
0	40.1±0.2	34.5±0.2	28.6±0.1	23.4±0.1	18.0±0.3
1.0	42.1±0.1	35.3±0.1	30.1±0.2	26.2±0.1	21.9±0.2
2.0	46.3±0.3	37.1±0.1	34.6±0.2	28.9±0.1	23.7±0.1
4.0	48.3±0.3	39.6±0.1	36.6±0.3	32.0±0.1	26.5±0.1
6.0	49.6±0.2	42.7±0.2	40.8±0.2	33.3±0.1	29.8±0.1
8.0	51.3±0.2	44.8±0.2	44.1±0.3	36.6±0.2	32.3±0.3

Table 1. Effects of P188 and P407 concentration on the gelation temperature (n = 3, °C).

Effects of additives and geniposide on gelation temperature

Drug and additives changed the gelation temperature (Table 2). The gelation tempera-ture decreased from 33.2 °C to 30.3 °C as the hydroxypropyl methylcellulose (HPMC) concentration increased from 0.1% to 1.0% (F3-F5). Benzalkonium chloride (BC) and sodium chloride (NaCl) decreased gelation temperature from 36.6 °C to 33.8 °C (F1 vs F2). However, the addition of the geniposide and borneol increased gelation temperature from 33.8 °C to 35.2 °C (F2 vs F6). The amounts of P407, P188 and HPMC should be optimized to achieve a more suitable gelation temperature for the thermoreversible *in situ* nasal gel of geniposide. All the formulations were transparent.

Optimization of formulation

P407 (18.0–24.0%), P188 (1.0–8.0%) and HPMC (0.1–1.0%) were the independent variables. Gelation temperature was the response variable. A three-factor and five-level (-Alpha, -1, 0, 1, +Alpha) full factorial design was employed for the optimization of *in situ* gel of formulations geniposide (Table 3).

Central composite design (CCD) was employed to evaluate the influence of the three independent variables in runs of 20 experiments. As shown in <u>Table 4</u>, based on the experimental results, a second-order polynomial equation using actual values demonstrated the empirical relationships between the independent variables and the response as follows:

Tgel = 192.7753–13.4382×P407+3.2916×P188-8.7255×HPMC-0.1687×P407×P188+ 0.1389×P407×HPMC+0.1667×P188×HPMC+0.2702×P407²+0.1905×P188²+2.9784 ×HPMC² (R^2 = 0.9760). Regression analysis with a R^2 value 0.9760 indicated a satisfactory correlation between the independent variables and the response. Analysis of variance (ANOVA) was

	Concentration (w/v)	Clarify	рН	Tgel (°C)
F1	P407 20%+P188 4%	+++	6.3	36.6±0.3
F2	P407 20%+P188 4%+ BC 0.001% + NaCl 0.9%	+++	6.4	33.8±0.2
F3	P407 20%+P188 4%+ BC 0.001% + NaCl 0.9% +HPMC 0.1%	++	6.2	33.2±0.1
F4	P407 20%+P188 4%+ BC 0.001% + NaCl 0.9% +HPMC 0.5%	++	6.1	32.7±0.1
F5	P407 20%+P188 4%+ BC 0.001% + NaCl 0.9% +HPMC 1.0%	++	6.2	30.3±0.2
F6	P407 20%+P188 4%+ BC 0.001% + NaCl 0.9% +geniposide 1.0%+ borneol 1.1%	++	6.5	35.2±0.2

Table 2. Effects of BC, NaCl, HPMC,	geniposide and borneol on g	elation temperature (n = 3).
-------------------------------------	-----------------------------	------------------------------

Note: turbid: +, clear: ++, very clear (glassy): +++



Independent variables		Levels and ranges					
	-α	low	medium	high	+α		
	-1.6818	-1.0	0	+1.0	+1.6818		
P407 (%)	18.0	19.2	21.0	22.8	24.0		
P188 (%)	1.0	2.4	4.5	6.6	8.0		
HPMC (%)	0.1	0.3	0.6	0.8	1.0		

Table 3. Independent variables and natural levels.

https://doi.org/10.1371/journal.pone.0189478.t003

applied to evaluate the adequacy of the empirical second- order polynomial model (Table 5), the Model *F*-value of 45.1780 (P<0.0001) implied the model was significant. In the present study, P407, P188, P407×P188, P407² and P188² were significant model terms. Three-dimensional surfaces and contours (Fig 1) as graphical representations of the regression equation showed the considerable influences of P407, P188 and HPMC on the gelation temperature. Gelation temperature increased smoothly with the increase of P188 from 1.0% to 8.0%.Gelation temperature obviously decreased with the increase of P407 from 18.0% to 24.0%. Effect of HPMC on gelation temperature was slight.

Gel strength and mucoadhesive strength of the optimized formulations

Eight groups of formulations which met the criteria (Table 6) were listed (Table 7). Clarity, pH, gelation temperature, gel strength and mucoadhesive strength of the optimized formulations were evaluated gradually (Table 7). All the optimized formulations were transparent. The pH values of all the formulations were found to be within 6.3–6.5 which was in the nasal physiological pH range 4.5–6.5. Gelation temperature values of the optimized formulations were 29.6–31.3 °C, which were in the nasal physiological temperature range 29–34 °C. The gel strength values of the optimized formulations were between 29 sec and 67 sec. Mucoadhesive

Runs	Independent variables in coded form			Ind	Independent variables in their natural form			Tgel value (°C)	
	407 (%)	188 (%)	HPMC (%)	407 (%)	188 (%)	HPMC (%)	Predicted	Actual	
1	1.0	1.0	-1.0	22.8	6.6	0.3	30.4	29.4±0.1	
2	-1.0	1.0	-1.0	19.2	6.6	0.3	41.8	41.9±0.3	
3	0.0	0.0	-1.6818	21.0	4.5	0.1	31.9	31.8±0.2	
4	1.0	-1.0	1.0	22.8	2.4	0.8	24.4	24.3±0.3	
5	0.0	0.0	1.6818	21.0	4.5	1.0	30.4	30.5±0.2	
6	1.0	-1.0	-1.0	22.8	2.4	0.3	25.4	25.2±0.1	
7	1.0	1.0	1.0	22.8	6.6	0.8	29.9	28.6±0.3	
8	-1.6818	0.0	0.0	18.0	4.5	0.6	41.5	40.0±0.2	
9	-1.0	1.0	1.0	19.2	6.6	0.8	41.0	41.1±0.3	
10	-1.0	-1.0	1.0	19.2	2.4	0.8	33.0	34.0±0.2	
11	1.6818	0.0	0.0	24.0	4.5	0.6	24.7	26.2±0.1	
12	0.0	1.6818	0.0	21.0	8.0	0.6	38.5	39.7±0.3	
13	-1.0	-1.0	-1.0	19.2	2.4	0.3	34.2	35.4±0.2	
14	0.0	-1.6818	0.0	21.0	1.0	0.6	27.5	26.3±0.1	
15~20	0.0	0.0	0.0	21.0	4.5	0.6	30.6	30.6±0.1	

Table 4. Central composite design experiments and experimental results.

Source	Sum of squares	df	Mean Square	<i>F</i> -value	P-value
Model	510.9341	9	56.7705	45.1780	< 0.0001
A-P407 (%)	339.6683	1	339.6683	270.3087	< 0.0001
B-P188 (%)	145.8883	1	145.8883	116.0981	< 0.0001
C-HPMC (%)	2.7124	1	2.7124	2.1586	0.1725
AB	3.2512	1	3.2512	2.5874	0.0188
AC	0.0313	1	0.0313	0.0249	0.8778
BC	0.0613	1	0.0613	0.0487	0.8297
A ²	11.0482	1	11.0482	8.7922	0.0142
B ²	10.1740	1	10.1740	8.0965	0.0174
C ²	0.4994	1	0.4994	0.3974	0.5426

Table 5. ANOVA for response surface quadratic model.

https://doi.org/10.1371/journal.pone.0189478.t005

strength values of the optimized formulations were between 3885 dyn/cm² and 1 0935 dyn/ cm^2 .

In vitro release kinetics of geniposide

As shown in Fig 2 and Table 8, the cumulative release rate of the geniposide from the optimized formulations (S3, S4, S5) versus time increased slowly without burst release condition, and the average rate was 99.4% after 6 h. The cumulative corrosion average rate of gel was 95.2% after 6 h. To understand the release mechanism of geniposide from *in situ* gel, data were analyzed with DD Solver (1.0) software. The order of suitable model was Zero-order model > Weibull model >First-order model > Higuchi model (Fig 3 and Table 9). A representative HPLC chromatogram of geniposide and a chromatogram of geniposide of the optimized formulations were presented in Fig 4.

Ex vivo permeation kinetics of geniposide

As shown in Table 10, the cumulative geniposide permeation rates of the optimized formulations were 83.1–86.8% after 6 h. To understand the release and permeation mechanism of geniposide, data were analyzed with DD Solver (1.0) software. The order of suitable model was Weibull model > Zero-order model > First-order model > Higuchi model (Table 11).

Discussion

Nasal drug delivery has received a significant attention as a convenient and reliable route for local administration of drugs [43, 44]. The nasal cavity offered a distinctive advantage for potential direct drug delivery to the brain along the olfactory nerves [34, 45, 46]. The rapid mucociliary clearance was important rate-limiting factor for nasal drug absorption [15, 47]. For the reason, researches have oriented toward the application of the bioadhesive polymers to extend formulations' residence time in the nasal cavity for better drug bioavailability [48, 49]. *In situ* forming polymeric formulation was the ideal drug delivery for nasal drops [50]. In this study, the poloxamer407 and the poloxamer188 were used as the basic excipients. The ideal formulation and gelation temperature were got by optimizing proportion of poloxamer407 and poloxamer188.

Determination of gelation temperature is a major step in the preparation of the thermoreversible gel [51, 52]. The thermoreversible hydrogel for nasal drug delivery should be gelled in 25–34°C [23, 53]. As shown in <u>Table 1</u>, the gelation temperature increased when the concentration of poloxamer407 was decreased. The same phenomenon was found when the



Fig 1. Three-dimensional surfaces and contours showing the influences of P407, P188 and HPMC. (A) 3D surface plot of P407 and P188 on Tgel (HPMC = 0.4%). (B) Contour of P407 and P188 on Tgel (HPMC = 0.4%). (C) 3D surface plot of P407 and HPMC on Tgel (P188 = 3.1%). (D) Contour of P407 and HPMC on Tgel (P188 = 3.1%). (E) 3D surface plot of P188 and HPMC on Tgel (P407 = 21.0%). (F) Contour of P188 and HPMC on Tgel (P407 = 21.0%).

PLOS ONE

concentration of poloxamer188 was increased. When the concentration of poloxamer407 was \geq 24% and the concentration of poloxamer188 was \leq 2.0%, the gelation temperature of the formulation was <25°C. The formulation easily gelled during manufacturing and was unsuitable for administration. If the concentration of poloxamer407 was \leq 18%, the gelation temperature of formulation was >34°C. The formulations couldn't be used because they remained in liquid

Name	Goal values	Lower limit	Upper limit
P407 (%)	in range	18.0	22.8
P188 (%)	in range	1.0	8.0
HPMC (%)	in range	0.1	1.0
Tgel (°C)	31.0	30.0	32.0

Table 6.	Criteria for	the optimized	formulations.

state and were easily washed away in nasal cavity. When the concentration of poloxamer407 was >18% and <24%, some gelation temperature was >34°C, too. The addition of poloxamer188 provided more alternative in the optimization of the formulation in the present study. The differential effects of P188 and P407 on the gelation temperature resulted from the different proportions of PPO and PEO subunits. As shown in Table 2, the addition of drug and additives had some effects on the gelation temperature, too. The gelation temperature decreased with the addition of the hydroxypropyl methylcellulose, because the hydroxypropyl methylcellulose could bind to polyethylene oxide chains in poloxamer molecules. This binding hindered the interactions between water and poloxamer molecules with more extensive intermolecular hydrogen bonding [54, 55]. The gelation temperature decreased when sodium chloride was added. The decrease in the gelation temperature was possible due to the salting-out effect of NaCl on PEO segments. It was known that the cloud point of PEO surfactants decreased due to the salting out effects caused by Na⁺ and Cl⁻[56, 57]. However, the addition of geniposide increased the gelation temperature by its water solubility characteristics.

The response surface methodology is a kind of mathematical and statistical technique for designing experiments, building models, evaluating the relative significance of several independent variables, and determining the optimum conditions for desirable response [58–60]. In this study, the central composite design was employed for determining the optimum condition of the gelation temperature. A 3-factor, 5-level full factorial design was employed for the optimization of the mucoadhesive, thermoreversible *in situ* nasal. To evaluate the adequacy of the model, analysis of variance (ANOVA) was applied. The ANOVA of the empirical second-order polynomial indicated that the model was highly significant (Table 5). The response surface methodology played an important auxiliary role in optimizing the thermoreversible *in situ* nasal gel.

Mucoadhesive strength of the formulation is another important index [61]. Nasal mucociliary clearance decreased contact time and drug absorption by transporting the drug to the

Table 7. Parameters of the optimized formulations (n = 3).

F	P407 (%)	P188 (%)	HPMC (%)	рН	Clarity	Tgel (°C)	Gel strength (sec)	Mucoadhesive strength (dyn/cm ²)
S1	21.1	4.0	0.1	6.3	++	30.6±0.2	29±1.7	4560±106
S2	20.2	1.5	0.2	6.4	++	30.8±0.2	35±2.4	3885±68
S3	20.5	3.3	0.3	6.5	++	30.8±0.1	37±2.5	4450±46
S4	19.7	1.3	0.5	6.4	++	29.7±0.2	38±0.9	4555±82
S5	19.9	2.4	0.6	6.4	++	30.2±0.1	46±1.2	5330±53
S6	19.6	1.5	0.8	6.3	++	31.0±0.3	58±1.8	6780±72
S7	19.6	1.6	0.9	6.5	++	31.3±0.2	59±1.2	8945±115
S8	19.4	1.1	1.0	6.5	++	29.6±0.3	67±2.2	10935±62

Note: turbid: +, clear: ++, very clear (glassy): +++



Fig 2. The release features of formulation in vitro. (A) Cumulative release of geniposide from the optimized formulations. (B) Cumulative corrosion of gel of the optimized formulations.

nasopharynx and then to the gastric intestinal tract. The mucoadhesive strength between 4000-6000 dyn/cm² was considered adequate [17, 62]. As shown in Table 7, the concentration of HPMC influenced the mucoadhesive strength greatly. As concentration of HPMC increased from 0.1% to 1.0%, there was a significant increase in mucoadhesive strength from 4560 dyn/cm² to 1 0935 dyn/cm².

The gel strength is another important criterion. The gel strength values between 25–50 sec were essential [15, 63]. As shown in Table 7, as concentration of HPMC increased from 0.1% to 1.0%, there was a significant increase in gel strength from 29 sec to 67 sec. The gel strength values of optimized formulations (S6, S7) were more than 50 sec, which meant the formulations were too stiff and would cause discomfort to the mucosal surface. Gel strength of the optimized formulations (S3, S4 and S5) was suitable.

The pH values of all the formulations were 6.3–6.5 which were in the nasal physiological pH range 4.5–6.5 (Table 7) [64]. The drug and agents didn't effects the pH of the formulations. The drug contents of all optimized formulations were checked and found in the range of 97–101% (Table 8).

The membraneless model was applied to study geniposide release from thermosensitive gel *in vitro*. In general, this model allowed the release medium to directly contact the gel surface and was closer to the *in vivo* condition [37–39, 65]. The in vitro release profiles of geniposide from the optimized formulations and the gel corrosion profiles were obtained simultaneously.

Time	Cum	ulative corrosion of g	el (%)	Cu	Cumulative release of geniposide (%)		
(h)	S3	S4	S5	S3	S4	S5	
1	16.2±4.6	21.1±5.6	18.5±4.6	21.3±4.6	23.2±4.1	19.5±3.6	
2	32.1±2.5	39.3±3.1	31.7±3.9	37.1±2.6	36.1±2.2	36.7±1.8	
3	43.3±2.1	47.8±2.8	52.7±2.6	49.3±1.9	51.8±2.6	50.5±2.8	
4	62.1±2.8	60.3±1.2	63.6±1.9	67.1±2.7	66.3±1.7	67.6±2.8	
5	78.6±1.9	78.3±3.6	76.0±3.9	80.3±3.3	81.0±3.6	81.9±3.0	
6	94.6±5.8	97.7±4.3	93.3±5.1	97.6±6.5	99.3±8.0	101.3±7.0	

Table 8. Corrosion of gel and release of geniposide from the optimized formulations.



Fig 3. Release kinetics of geniposide from the optimized formulations.

The in vitro release data of geniposide were evaluated kinetically using mathematical models like the Zero-order, First-order, Higuchi and Weibull (Table 9). The best-fit model was the Zero-order model ($R^2 = 0.9935-0.9959$). Release of geniposide was controlled by gel corrosion.

Ex vivo drug permeation could give more valuable informations about drug behavior in vivo [41–42, 66]. In this study, the cumulative release of geniposide from the optimized thermoreversible *in situ* nasal gel was 83.1–89.8% (Table 10). As shown in Table 11, the geniposide permeation kinetics model of the optimized formulations was fitted to the Weibull model ($R^2 = 0.9943-0.9979$), which indicated the permeation of geniposide across nasal mucosa was possibly time-dependent. The more residence time of *in situ* nasal gel of geniposide was essential.

Conclusions

The release of geniposide is controlled by gel corrosion. The permeation of geniposi- de is time dependent. The more residence time, mucoadhesive, thermoreversible *in situ* nasal gel of geniposide for neurodegenerative diseases is of compliance and potential application. The attractive *in situ* nasal gel of geniposide will be evaluated in further animal study.

Table 9. In vitro release kinetics models of geniposide from the optimized formulations.

Model	Zero-order	First-order	Higuchi	Weibull
S3	Y = 3.512+1.003*X	Y = 100*[1-Exp(-0.018*X)]	Y = 8.597*X^0.5	Y = 100*{1-Exp[-(X^1.449)/324.419]}
	Rsqr = 0.9959	Rsqr = 0.9493	Rsqr = 0.9171	Rsqr = 0.9835
S4	Y = 0.480+1.029*X	Y = 100*[1-Exp(-0.017*X)]	Y = 8.430*X^0.5	Y = 100*{1-Exp[-(X^1.765)/1250.221]}
	Rsqr = 0.9935	Rsqr = 0.9132	Rsqr = 0.8782	Rsqr = 0.9827
S5	Y = 1.072*X-0.339	Y = 100*[1-Exp(-0.018*X)]	Y = 8.602*X^0.5	Y = 100*{1-Exp[-(X^1.661)/797.704]}
	Rsqr = 0.9941	Rsqr = 0.9084	Rsqr = 0.8721	Rsqr = 0.9660

Note: X: Cumulative corrosion of gel (%). Y: Cumulative release of geniposide (%).





Fig 4. Determination by HPLC method. (A) Representative HPLC chromatogram of geniposide (5.967 min). (B) Chromatogram of geniposide of the optimized formulations (5.990 min).

Time (h)	S3	S4	S5
0.5	8.1±1.2	7.9±2.3	11.2±1.0
1.0	16.8±1.9	18.6±2.3	16.3±2.1
1.5	25.3±3.2	26.9±1.6	23.9±3.2
2.0	33.7±2.8	38.1±3.7	35.2±3.3
2.5	46.1±3.1	45.6±5.5	43.3±2.4
3.0	53.5±1.6	55.5±3.1	50.5±4.6
3.5	61.3±5.6	63.6±3.4	61.1±3.8
4.0	74.1±2.1	73.0±3.0	67.4±4.2
4.5	79.5±3.6	77.4±1.7	72.8±3.6
5.0	82.6±2.1	83.1±1.6	79.1±1.0
5.5	84.6±6.3	86.3±5.6	80.6±5.6
6.0	86.8±5.3	89.8±4.6	83.1±7.0

Table 10. *Ex vivo* cumulative permeation rate of geniposide from optimized formulations.

Model	Zero-order	First-order	Higuchi	Weibull
S3	Y = 4.442+15.369*t	Y = 100*[1-Exp(-0.282*t)]	Y = 50.553*t^0.5–37.762	Y = 100*{1-Exp[-t^1.457)/6.159]}
	Rsqr = 0.9678	Rsqr = 0.9423	Rsqr = 0.9824	Rsqr = 0.9943
S4	Y = 5.702+15.317*t	Y = 100*[1-Exp(-0.290*t)]	Y = 50.410*t^0.5–31.3999	Y = 100*{1-Exp[-(t^1.408)/5.600]}
	Rsqr = 0.9771	Rsqr = 0.9532	Rsqr = 0.9928	Rsqr = 0.9979
S5	Y = 5.976+14.174*t	Y = 100*[1-Exp(-0.260*t)]	Y = 46.469*t^0.5–28.049	Y = 100*{1-Exp[-(t^1.327)/5.763]}
	Rsqr = 0.9757	Rsqr = 0.9621	Rsqr = 0.9839	Rsqr = 0.9947
S5	Y = 5.976+14.174*t	Y = 100*[1-Exp(-0.260*t)]	Y = 46.469*t^0.5-28.049	Y = 100*{1-Exp[-(t/1.327)
	Rsqr = 0.9757	Rsqr = 0.9621	Rsqr = 0.9839	Rsqr = 0.9947

Table 11. Ex vivo permeation kinetics models of geniposide from the optimized formulations.

Note: Y: Cumulative release of geniposide (%). t: Time (h)

https://doi.org/10.1371/journal.pone.0189478.t011

Supporting information

S1 Fig. Cube plot of the effects of P408, P188 and HPMC on Tgel at a time. The predicted values from the coded model were P407 (19.2–22.8%), P188 (2.4–6.6%) and HPMC (0.3–0.8%). (TIF)

S2 Fig. The normal plot of residuals. Distribution of Tgel points indicates that the transformation of the response may provide a better analysis.

(TIF)

S1 Table. Diagnostic statistics. The report of the residual, leverage, fitted value DFFITS and other statistics indicate that not all actual values are ideal and some are suitable. So the optimized formulations must be testified and achieve the suitable formulation. (DOC)

Acknowledgments

The authors are thankful to professors Jianzhong Zhu and Shijin Bu for providing advices.

Author Contributions

Conceptualization: Shulong Jiang.

Data curation: Yingting Wang, Shulong Jiang, Hongli Wang, Haiyan Bie.

Formal analysis: Yingting Wang, Shulong Jiang, Hongli Wang, Haiyan Bie.

Funding acquisition: Shulong Jiang.

Investigation: Shulong Jiang.

Methodology: Shulong Jiang.

Project administration: Yingting Wang, Shulong Jiang.

Resources: Shulong Jiang.

Software: Yingting Wang, Shulong Jiang, Hongli Wang, Haiyan Bie.

Supervision: Shulong Jiang.

Validation: Yingting Wang, Shulong Jiang.

Visualization: Yingting Wang, Shulong Jiang, Hongli Wang, Haiyan Bie.

Writing - original draft: Yingting Wang, Hongli Wang, Haiyan Bie.

Writing - review & editing: Yingting Wang, Shulong Jiang.

References

- 1. Junn E, Mouradian MM. MicroRNAs in Neurodegenerative Diseases and Their Therapeutic Potential. Pharmacol Ther. 2012; 133(2):142–150. https://doi.org/10.1016/j.pharmthera. PMID: 22008259
- Chen P, Wang LX, Wang Y, Li S, Shen L, Liu Q, et al. Phosphoproteomic Profiling of Selenatetre-ated Alzheimer's Disease Model Cells. PLoS One. 2014; 9(12):e113307. https://doi.org/10.1371/journal. pone. PMID: 25485856
- Bryan L, Kaye W, Antao V, Mehta P, Muravov O, Horton DK. Preliminary Results of National Amyotrophic Lateral Sclerosis (ALS) Registry Risk Factor Survey Data. PLoS One. 2016; 11(4): e0153683. https://doi.org/10.1371/journal.pone.0153683 PMID: 27124833
- Callegari L, Mattei C, Benassi F, Krueqer F, Grafman J, Yaldizli Ö, et al. Agomelatine Improves Apathy in Frontotemporal Dementia. Neurodegener Dis. 2016; 16(5–6):352–6. <u>https://doi.org/10.1159/</u> 000445873 PMID: 27229348
- Zhou TT, Fan GR, Hong ZY, Chai YF, Wu YT. Large-scale isolation and purification of geniposide from the fruit of Gardenia jasminoides Ellis by high-speed counter-current chromatography. JChromatogr A. 2005; 1100(1):76–80. https://doi.org/10.1016/j.chroma.2005.09.026 PMID: 16203006
- Koo HJ, Lim KH, Jung HJ, Park EH. Anti-inflammatory evaluation of gardenia extract, geniposide and genipin. J Ethnopharmacol. 2006; 103(3):496–500. https://doi.org/10.1016/j.jep.2005.08.011 PMID: 16169698
- Li L, Zhao J, Liu K, Li GL, Han YQ, Liu Z. Geniposide prevents rotenone-induced apoptosis in primary cultured neurons. Neural Regen Res. 2015; 10(10):1617–21. <u>https://doi.org/10.4103/1673-5374</u>. 167760 PMID: 26692859
- Liu JH, Yin F, Zheng XX, Jing JJ, Hu YH. Geniposide, a novel agonist for GLP-1 receptor, prevents PC12 cells from oxidative damage via MAP kinase pathway. Neurochem Int. 2007; 51(6–7):361–9. https://doi.org/10.1016/j.neuint.2007.04.021 PMID: 17629357
- Wang L, Yan SJ, Zhang WS. Effects of an early and late-stage treatment with Geniposide on cognitive dysfunction in a transgenic mouse model relevant to Alzheimer's disease. Molecular Neurodegeneration. 2012; 7(7):1–1. https://doi.org/10.1186/1750-1326-7-S1-S4
- Lv C, Wang L, Liu X, Cong X, Yan SS, Wang Y, et al. Geniposide attenuates oligomeric Aβ (1–42)induced (1–42)-induced inflammatory response by targeting RAGE-dependent signaling in BV2 cells. Curr Alzheimer Res. 2014; 11(5):430–40 PMID: 24801214
- Zhao CH, Lv C, Li H, Du SS, Liu XL, Li Z, et al. Geniposide Protects Primary Cortical Neurons against Oligomeric Aβ1-42-Induced Neurotoxicity through a Mitochondrial Pathway. PLoS One. 2016, 11(4): e0152551. https://doi.org/:0.1371/journal.pone.0152551 PMID: 27046221
- Yang M, Chen XY, Zhang HY, Wang JM, Lu Q, Song W. Pharmacokinetics of geniposide through 4 routes of administration. Chinese Journal of New Drugs. 2010; 19(9):769–754.
- Lu Y, Du SY, Bai J, Li PY, Wen R, Zhao XJ. Bioavailability and Brain-Targeting of Geniposide in Gardenia-Borneol Co-Compound by Different Administration Routes in Mice. Int. J. Mol. Sci.2012; 13 (11):14127–35. https://doi.org/10.3390/ijms131114127 PMID: 23203054
- 14. Pires A, Fortuna A, Alves G, Falcao A. Intranasal drug delivery: how, why and what for? JPharm Pharm Sci. 2009; 12(3):288–311. https://doi.org/10.18433/J3NC79 PMID: 20067706
- Karavasili C, Fatouros DG. Smart materials: *in situ* gel-forming systems for nasal delivery. Drug Discov Today. 2016; 21(1):157–166. https://doi.org/10.1016/j.drudis.2015.10.016 PMID: 26563428
- Illum L. Nasal drug delivery: new developments and strategies. Drug Discov Today. 2002; 7(23): 1184– 9. PMID: 12547019
- Singh RM. Kumar A, Pathak k. Thermally Triggered Mucoadhesive In Situ Gel of Loratadine: β-Cyclodextrin Complex for Nasal Delivery. AAPS PharmSciTech. 2013; 14(1): 412–424. <u>https://doi.org/10.</u> 1208/s12249-013-9921-9 PMID: 23358934
- Du SY, Lu R, Yao ZL, Chen XL, Zhai YS, Wu Q, et al. Effect of Components in Xingnaojing on Rat Nasal Absorption of Geniposide from Gardenia Extract. Chinese Journal of Experimental Tra-ditional Medical Formulae. 2012; 18(4): 146–150.
- Chen ZZ, Gong X, Lu Y, Du SY, Yang ZH, Bai J, et al. Enhancing effect of borneol and muscone on geniposide transport across the human nasal epithelial cell monolayer. PloS One. https://doi.org/10.1371/ journal.pone.0101414 PMID: 24992195
- Yapar EA, İnal Ö. Poly (ethylene oxide)–Poly (propylene oxide)-Based Copolymers for Transd-ermal Drug Delivery: An Overview. Tropical J Pharm Res 2012; 11 (5): 855–866. <u>https://doi.org/10.4314/tjpr.v11i5.20</u>

- Zhang K, Shi XL, Lin X, Yao CX, Shen L, Feng Y. Poloxamer-based *in situ* hydrogels for controlled delivery of hydrophilic macromolecules after intramuscular injection in rats. Drug Delive. 2015; 22(3):375–382. https://doi.org/10.3109/10717544.2014.891272 PMID: 24601854
- Wang HL, Bie HY, Si CP, Wang YT. Optimization of the Formulation of Azelastine Hydroch-loride Thermosensitive In-situ Gel Nasal Drops by Central Composite Design-Response Surface Methodology. China Pharmacy. 2016; 27(1):98–100.
- Klouda L, Mikos AG. Thermoresponsive hydrogels in biomedical applications. Eur J Pharm Bio-pharm. 2008; 68(1):34–45. https://doi.org/10.1016/j.ejpb.2007.02.025 PMID: 17881200
- Asasutjarit R, Thanasanchokpibull S, Fuongfuchat A, Veeranondha S. Optimization and evalua- tion of thermoresponsive diclofenac sodium ophthalmic *in situ* gels. Int J Pharmaceut. 2011; 411(1–2):128– 35. https://doi.org/10.1016/j.ijpharm.2011.03.054 PMID: 21459137
- 25. Yu Z, Guo F, Guo Y, Zhang Z, Wu F, Luo X. Optimization and evaluation of astragalus poly-saccharide injectable thermoresponsive in-situ gels. PLoS One. 2017; 12(3): e0173949. <u>https://doi.org/10.1371/journal.pone.0173949</u> PMID: 28350799
- Rdivojša M, Grabnar I, Ahlin Grabnar P. Thermosensitive *in situ* gelling poloxamer-based syste-ms with chitosan nanocomplexes for prolonged subcutaneous delivery of heparin: Design and in vitro evaluation. Euro J Pharm Sci. 2013; 50 (1): 93–101. <u>https://doi.org/10.1016/j.ejps.2013.03.002</u> PMID: 23524253
- Shi XL, Lin X, Yao CX, Shen L, Feng Y. Injectable long-acting *in situ* forming systems for Radix Ophiopogonis polysaccharide. Int J Biol Macromol 2015; 72(2): 553–559. https://doi.org/10.1016/j.ijbiomac. 2014.09.009 PMID: 25236608
- Bie HY, Wang HL, Wang YT. Optimization of Azelastine Hydrochloride Thermosensive in situ gel eye drops by Central Composite Design-response Surface Methodology. Chinese Journal New Drug. 2016; 25(8):1–7
- Bhandwalkar MJ, Avachat AM. Thermoreversible Nasal *In Situ* Gel of Venlafaxine Hydroc-hloride: Formulation, Characterization, and Pharmacodynamic Evaluation. AAPS Pharm Sci Tech.2013; Tech.2013; 14(1):101–10. https://doi.org/10.1208/s12249-012-9893-1 2322938123229381
- Ur-Rehman T, Tavelin S, Gröbner G. Chitosan *in situ* gelation for improved drug loading and ret-ention in poloxamer 407 gels Int J Pharm. 2011; 409(1–2):19–29. <u>https://doi.org/10.1016/jijpharm.2011.02</u>. 017 PMID: 21335076
- Marwah H, Garg T, Rath G, Goyal AK. Development of transferosomal gel for trans-dermal deli-very of insulin using iodine complex. Drug Deliv. 2016; 23(5):1636–1644 https://doi.org/10.3109/10717544. 2016.1155243 PMID: 27187718
- Yong CS, Choi JS, Quan QZ, Rhee JD, Kim CK, Lim SJ et al. Effect of sodium chloride on the gelation temperature, gel strength and bioadhesive force of poloxamer gels containing diclofenac sodium. Int J Pharm. 2001; 226(1–2):195–205. PMID: 11532582
- Belgamwar VS, Chauk DS, Mahajan HS, Jain SA, Gattani SG, Surana SJ. Formulation and evalu-ation of *in situ* gelling system of dimenhydrinate for nasal administration. Pharm Dev Technol. 2009; 14 (3):240–8. https://doi.org/10.1080/10837450802498910 PMID: 19235555
- Kempwade A, Taranalli A. Formulation and evaluation of thermoreversible, mucoadhesive in situ intranasal gel of rizatriptan benzoate. J Sol-Gel Sci Technol. 2014; 72(1): 43–48. <u>https://doi.org/10.1007/</u> s10971-014-3422-5
- Shelke S, Shahi S, Jadhav K, Dhamecha D, Tiwari R, Patil H. Thermoreversible nanoethosomal gel for the intranasal delivery of Eletriptan hydrobromide. J Mater Sci Mater Med. 2016; 27(6):103. https://doi. org/10.1007/s10856-016-5713-6 PMID: 27091045
- 36. Galgatte UC, Kumbhar AB, Chaudhari PD. Development of *in situ* gel for nasal delivery: design, optimization, in vitro and in vivo evaluation. Drug Deliv. 2014; 21(1):62–73. <u>https://doi.org/10.3109/</u> 10717544.2013.849778 PMID: 24191774
- Yeole PG, Iyer D. Thermosensitive *in situ* gel of Timolol Maleate for the treatment of open angle glaucoma. RJPBCS. 2011; 2(3):1048–1064.
- Jabarian LE, Rouini MR, Atyabi F, Foroumadi A, Nassiri SM, Dinarvand R, et al. In vitro and in vivo evaluation of an *in situ* gel forming system for the delivery of PEGylated octreotide. Eur J, Pharm Sci 2013; 48(87): 87–96. https://doi.org/10.1016/j.ejps.2012.09.017 PMID: 23131800
- Phaechamud T, Mahadlek J, Chuenbarn T. In situ forming gel comprising bleached shellac loaded with antimicrobialdrugs for periodontitis treatment. Materials & Design. 2016; 89(5): 294–303. https://doi. org/10.1016/j.matdes.2015.09.138
- Chu KD, Chen LD, Xu W, Li H, Zhang YQ, Xie WR, et al. Preparation of a Paeonol Containing Temperature-Sensitive *In Situ* Gel and Its Preliminary Efficacy on Allergic Rhinitis. Int J Mo Sci. 2013; 14(3): 6499–6515. https://doi.org/10.3390/ijms14036499

- Patel N, Baldaniya M, Raval M, Sheth N. Formulation and Development of *In Situ* Nasal Gelling Systems for Quetiapine Fumarate-Loaded Mucoadhesive Microemulsion. J Pharm Innov. 2015; 10 (4):357–373. https://doi.org/10.1007/s12247-015-9232-7
- Pathak R, Dash RP, Misra M, Nivsarkar M. Role of mucoadhesive polymers in enhancing deliv-ery of nimodipine microemulsion to brain via intranasal route. Acta Pharm Sin B. 2014; 4(2):151–160 https://doi.org/10.1016/j.apsb.2014.02.002 PMID: 26579378
- Rokade M, Tambe B, Ruparel M. IN SITU GEL -SUSTAINED NASAL DRUG DELIVERY. IJPSR, 2015; Vol. 6(12): 4958–4966. https://doi.org/10.13040/IJPSR.0975-8232
- Illum L. Nasal drug delivery-possibilities, problems and solutions. J Control Release. 2003; 87(1– 3):187–98. http://doi.org/10.1016/S0168-3659(02)00363-2 PMID: 12618035
- De A, Chakraborty S, Mukherjee A, Chattopadhyay J, Ghatak S. DESIGN AND OPTIMIZAT-ION OF NASAL IN SITU GEL OF ONDANSETRON USING FACTORIAL DESIGN. IJRPC. 2013, 3(3):659– 673.
- 46. Rao M, Agrawal DK, Shirsath C. Thermoreversible mucoadhesive *in situ* nasal gel for treatment of Parkinson's disease. Drug Dev Ind Pharm.2017; 43(1):142–150. <u>https://doi.org/10.1080/03639045.2016</u>. 1225754 PMID: 27533244
- Jadhav KR, Gambhire MN, Shaikh IM, Kadam VJ, Pisal SS. Nasal Drug Delivery System-Factors Affecting and Applications. Current Drug Therapy. 2007; 2(1):27–38.<u>https://doi.org/10.2174/</u> 157488507779422374
- Dondeti P, Zia H, Needham TE. Bioadhesive and formulation parameters affecting nasal absor- ption. Inter J Pharma. 1996; 127(2):115–133. https://doi.org/10.1016/0378-/10.1016/0378-5173(95)04115-X
- Ugwoke MI, Agu RU, Verbeke N, Kinget R. Nasal mucoadhesive drug delivery: Background, applications, trends and future perspectives. Adv Drug Deliv Rev. 2005; 57(11):1640–1665. <u>https://doi.org/10. 1016/j.addr.2005.07.009</u> PMID: 16182408
- Madan M1, Bajaj A, Lewis S, Udupa N, Baig JA. *In situ* forming polymeric drug delivery syst-ems. Indian J Pharm Sci. 2009; 71(3):242–51. https://doi.org/10.4103/0250-474X.56015 PMID: 20490289
- Fernandez EJQ, Lopez D, Mijangos C, Duskovasmrckova M, Ilavsky M, Dusek K. Rheological and thermal properties of agarose aqueous solutions and hydrogels. Journal of Polymer Science Part B.2008; 46(3):322–328. https://doi.org/10.1002/polb.21370
- Kempwade A, Taranalli A. Formulation and evaluation of thermoreversible, mucoadhesive in situ intranasal gel of rizatriptan benzoate. J Sol-Gel Sci Technol. 2014; 72(1):43–48. https://doi.org/10.1007/ s10971-014-3422-5
- Yu L, Zhang Z, Zhang H, Ding JD.Biodegradability and biocompatibility of thermoreversible hydrogels formed from mixing a sol and a precipitate of block copolymers in water Biomacromo- lecules. 2010; 11 (8):2169–2178. https://doi.org/10.1021/bm100549g PMID: 20690723
- Lai M, Chang KC, Hsu SC, Chou MC, Hung W, Hsiao YR, et al. *In situ* gelation of PEG- PLGA-PEG hydrogels containinro and in vivo characterstics. Bi-omed Mater. Materials. 2014; 9(1): 015011. <u>https://</u> doi.org/10.1088/1748-6041/9/1/015011 PMID: 24457223
- Lu CC, Liu MJ, Fu HL, Zhang W, Peng GG, Zhang YL, et al. Novel thermosensitive in situ gel based on poloxamer for uterus delivery. Eur J Pharm Sci. 2015; 77:24–8. <u>https://doi.org/10.1016/j.ejps.2015.05.</u> 014 PMID: 25981887
- Jiang J, Li CH, Lombardi J, Colby RH, Rigas B, Rafailovich MH, et al. The effect ofphysiologic-ally relevant additives on the rheological properties of concentrated Pluronic copolymer gels. Pol-ymer. 2008; 49(16):3561–3567. https://doi.org/10.1016/j.Polymer.2008.05.038
- Seo HJ, Kim JC. Effects of additives on phase transitions of Poloxamer 407/Poloxamer 188 mixtureand release property of monoolein cubic phase containing the poloxamers. J Ind Eng Chem.2012; 18 (1):88–91.https://doi.org/10.1016/j.jiec.2011.11.077
- Nour V, Trandafir L, Cosmulescu S. Central Composite Design Applied to Optimize the Hydroal-coholic Extraction of Bilberry (Vaccinium Myrtillus L.) Fruits. J Food Biochem 2015; 39 (2): 179–188.https://doi. org/10.1111/jfbc.12119
- Kuo CH, Huang WH, Lee CK, Liu YC, Chang CMJ, Yang H, et al. Biofuel Cells Composed by Using Glucose Oxidase on Chitosan Coated Carbon Fiber Cloth. Int. J. Electrochem. Sci., 2013, 8 9242–9255.
- Bie HY, Wang HL, Wang YT. Optimization of Azelastine Hydrochloride Thermosensive in situ gel eye drops by Central Composite Design-response Surface Methodology. Chinese Journal New Drug. 2016; 25(8):1–7
- Sharma N,Kulkarni GT, Sharma A. Development of Abelmoschus esculentus (Okra)-Based Mucoadhesive Gel for Nasal Delivery of Rizatriptan Benzoate. Tropical Journal of Pharmaceutical Research.2013; 12(2):149–153. https://doi.org/10.4314/tjpr.v12i2.3

- 62. Pathak R, Prasad Dash R, Misra M, Nivsarkar M. Role of mucoadhesive polymers in enhancing delivery of nimodipine microemulsion to brain viaintranasal route. Acta Pharm Sin B. 2014; 4(2):151–60. <u>https://doi.org/10.1016/j.apsb.2014.02.002</u> PMID: 26579378
- Gratieri T, Gelfuso GM, Rocha EM, Sarmento VH, de Freitas O, Lopez RF, et al. A poloxamer/chitosan in situ forming gel with prolonged retention time for ocular delivery Eur J Pharm Biopharm. 2010; 75 (2):186–93. https://doi.org/10.1016/j.ejpb.2010.02.011 PMID: 20188828
- Jagdale S, Shewale N, Kuchekar BS. Optimization of Thermoreversible In Situ Nasal Gel of Timolol Maleate. Scientifica (Cairo). 2016; 2016:6401267. https://doi.org/10.1155/2016/6401267 PMID: 27293975
- 65. Dos Santos AC, Akkari AC, Ferreira IR, Maruyama CR, Pascoli M, Guilherme VA, et al. Polox-amerbased binary hydrogels for delivering tramadol hydrochloride: sol-gel transitionstudies, dissolutionrelease kinetics, in vitro toxicity, and pharmacological evaluationInt JNanomedicine. 2015; 10:2391– 401. https://doi.org/10.2147/IJN.S72337
- Dash S, Murthy PN, Nath L, Chowdhury P. Kin Drug Release from Controlled Drug Delivery Systems. Acta Poloniae Pharmaceutica. 2010; 67(3):217–223. PMID: 20524422