

Bis(isonicotinamide- κN)silver(I) trifluoromethanesulfonate acetonitrile disolvate

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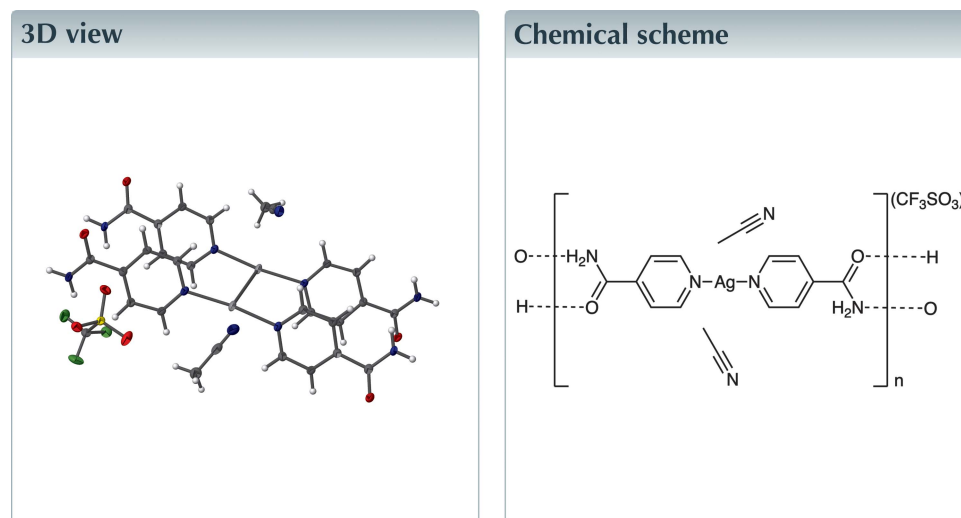
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Keywords: crystal structure; silver atom; isonicotinamide; trifluoromethanesulfonate ions; acetonitrile; polymeric structure.

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Structural data: full structural data are available from iucrdata.iucr.org

The central Ag^I atom of the title salt, [Ag(INAM)₂](CF₃SO₃)·2CH₃CN, where INAM is isonicotinamide (C₆H₆N₂O), is twofold coordinated by the pyridine N atoms of two isonicotinamide ligands creating a slightly distorted linear molecular geometry. The formation of polymeric chains {[Ag(INAM)₂]⁺}_n, held together by discrete hydrogen bonds through the amide group of the INAM ligand leaves voids for non-coordinating acetonitrile molecules that interact the silver metal center *via* regium bonds.



Structure description

Silver(I) isonicotinamide complexes have been investigated for the ability to form coordination complexes with a variety of molecular geometries due to amide hydrogen-bond synthons in their structure (Aakeröy & Beatty, 1998; Aakeröy *et al.*, 1998; Lian *et al.* 2007), luminescent properties (Yeşilel *et al.*, 2012), and antibacterial activity (Abu-Youssef *et al.*, 2007; Yu *et al.*, 2020). Our research group interest currently lies in the synthesis of novel metal complexes with biological activity; as part of our research in this area, herein, we describe the synthesis and structure of the title silver(I) complex.

As depicted in Fig. 1, the asymmetric unit of the title compound shows the Ag^I ion in a distorted linear coordination environment defined by two N-bonded isonicotinamide ligands. Two acetonitrile molecules and a trifluoromethanesulfonate ion complete the asymmetric unit; the acetonitrile molecules sit at opposite sides of the plane defined by N1–Ag1–N3 with the nitrile group facing the silver(I) metal center. All relevant bond lengths and angles involving the Ag atom are presented in Table 1. The angle N1–Ag1–N3 of 172.78 (7) is within the reported values (174.9, 180, and 171.1) in the comparable silver(I) isonicotinamide structures currently available in the CSD (version 5.42 with

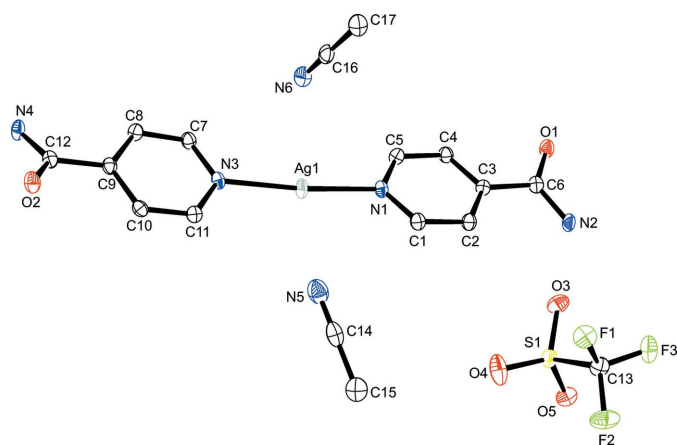


Figure 1
The molecular structure of the title compound with displacement ellipsoids drawn at the 50% probability level; H atoms are omitted for clarity.

update May 2021; Aakeröy & Beatty, 1998; refcode NISNEI; Bhogala *et al.*, 2004; refcode NABYOF; Abu-Youssef *et al.*, 2007; refcode XECZUB01).

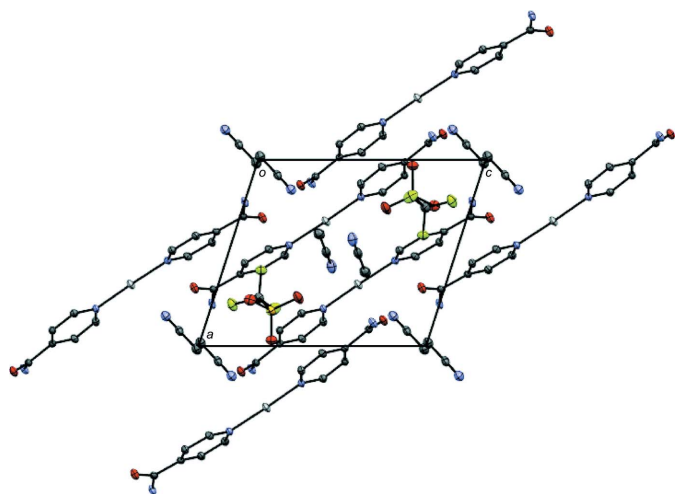


Figure 2
Perspective view of the packing structure of the title complex along the crystallographic *b* axis; H atoms are omitted for clarity.

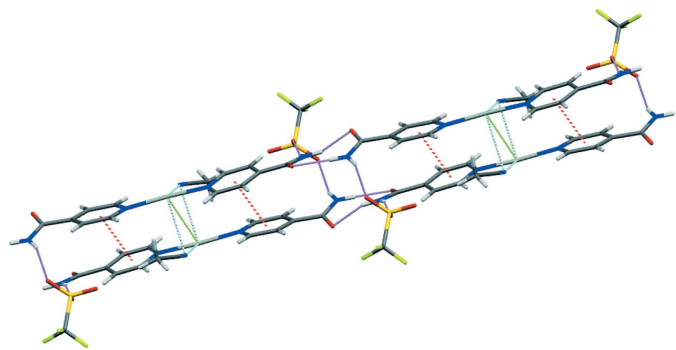


Figure 3
Capped sticks representation of the title molecule showing hydrogen bonds interactions (violet), regium bonds (light blue), Ag...Ag interactions (light green), and π - π stacking interactions (red).

Table 1
Selected geometric parameters (\AA , $^\circ$).

Ag1—N3	2.162 (2)	Ag1—N1	2.162 (2)
N1—Ag1—N3	172.78 (7)	N2—C6—C3	117.3 (2)
N4—C12—C9	118.3 (2)		

Table 2
Hydrogen-bond geometry (\AA , $^\circ$).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2 <i>A</i> ...O2 ⁱ	0.88	2.06	2.898 (3)	160
N2—H2 <i>B</i> ...O3	0.88	2.09	2.939 (3)	162
N4—H4 <i>A</i> ...O1 ⁱⁱ	0.88	2.05	2.927 (3)	171
N4—H4 <i>B</i> ...O5 ⁱⁱⁱ	0.88	2.22	3.033 (3)	154

Symmetry codes: (i) $x + 1, y, z - 1$; (ii) $x - 1, y, z + 1$; (iii) $-x + 1, -y + 1, -z + 1$.

Table 3
Experimental details.

Crystal data	
Chemical formula	[Ag(C ₆ H ₆ N ₂ O) ₂](CF ₃ O ₃ S)·2C ₂ H ₃ N
<i>M_r</i>	583.30
Crystal system, space group	Triclinic, <i>P</i> $\bar{1}$
Temperature (K)	100
<i>a</i> , <i>b</i> , <i>c</i> (\AA)	9.4566 (2), 11.0330 (3), 11.9848 (3)
α , β , γ ($^\circ$)	114.000 (2), 103.9287 (19), 95.129 (2)
<i>V</i> (\AA^3)	1083.72 (5)
<i>Z</i>	2
Radiation type	Cu <i>K</i> α
μ (mm^{-1})	9.00
Crystal size (mm)	0.27 \times 0.10 \times 0.07
Data collection	
Diffractometer	XtaLAB Synergy, Dualflex, HyPix
Absorption correction	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2019)
<i>T_{min}</i> , <i>T_{max}</i>	0.544, 1.000
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	24470, 4387, 4211
<i>R_{int}</i>	0.044
($\sin \theta/\lambda$) _{max} (\AA^{-1})	0.631
Refinement	
$R[F^2 > 2\sigma(F^2)]$, $wR(F^2)$, <i>S</i>	0.027, 0.072, 1.08
No. of reflections	4387
No. of parameters	300
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}$, $\Delta\rho_{\text{min}}$ (e \AA^{-3})	0.77, -0.78

Computer programs: *CrysAlis PRO* (Rigaku OD, 2019), *SHELXT* (Sheldrick, 2015a), *SHELXL2018/3* (Sheldrick, 2015b), and *OLEX2* (Dolomanov *et al.*, 2009).

Two types of hydrogen-bonding motifs are present in the crystal lattice, with numerical values collated in Table 2. In the crystal packing, molecules self-assemble into layers aligned along the *a*-axis direction (Fig. 2) *via* N—H...O interactions. The trifluoromethanesulfonate anions fill the void between the layers and interact with the isonicotinamide ligands through additional N—H...O interactions. The pyridyl rings of the isonicotinamide ligand show π - π stacking interactions with centroid-to-centroid (*C_g*...*C_g*) distances ranging from 3.7005 (13) to 3.8503 (14) \AA , and offset distances ranging from 1.940 to 2.056 \AA , respectively.

Two different supramolecular interactions involving the silver atom are also responsible for the observed crystal packing: an Ag...Ag interaction with a distance between silver atoms of 3.4258 (3) Å, comparable to other silver complexes found in the CSD database (Titov *et al.*, 2018; refcode FINWOR; Titov *et al.*, 2019; refcode PIRCUR); and regium bonds, between the nitrogen of the acetonitrile solvent molecules and the silver atom (Alkorta *et al.*, 2020; Zierkiewicz *et al.*, 2018), with lengths of 2.916 Å for Ag1–N5 and 2.955 Å for Ag1–N6. (Fig. 3)

Synthesis and crystallization

Silver trifluoromethanesulfonate (0.200 g, 0.778 mmol) was added to an acetonitrile solution of isonicotinamide (0.190 g, 1.56 mmol) and stirred for 30 min. The resulting clear solution was used to grow crystals by vapor diffusion with diethyl ether at 278 K.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 3.

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full crystallographic data

IUCrData (2021). 6, x211073 [https://doi.org/10.1107/S2414314621010737]

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Crystal data

[Ag(C₆H₆N₂O)₂](CF₃O₃S)·2C₂H₃N

$M_r = 583.30$

Triclinic, *P*1

$a = 9.4566$ (2) Å

$b = 11.0330$ (3) Å

$c = 11.9848$ (3) Å

$\alpha = 114.000$ (2)°

$\beta = 103.9287$ (19)°

$\gamma = 95.129$ (2)°

$V = 1083.72$ (5) Å³

$Z = 2$

$F(000) = 584$

$D_x = 1.788$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 12499 reflections

$\theta = 4.2\text{--}76.2^\circ$

$\mu = 9.00$ mm⁻¹

$T = 100$ K

Plate, clear colourless

$0.27 \times 0.10 \times 0.07$ mm

Data collection

XtaLAB Synergy, Dualflex, HyPix
diffractometer

Radiation source: micro-focus sealed X-ray
tube, PhotonJet (Cu) X-ray Source

Mirror monochromator

Detector resolution: 10.0000 pixels mm⁻¹

ω scans

Absorption correction: gaussian
(CrysAlisPro; Rigaku OD, 2019)

$T_{\min} = 0.544$, $T_{\max} = 1.000$

24470 measured reflections

4387 independent reflections

4211 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.044$

$\theta_{\max} = 76.5^\circ$, $\theta_{\min} = 4.2^\circ$

$h = -10 \rightarrow 11$

$k = -13 \rightarrow 13$

$l = -15 \rightarrow 14$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.027$

$wR(F^2) = 0.072$

$S = 1.08$

4387 reflections

300 parameters

0 restraints

Primary atom site location: dual

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0394P)^2 + 0.7P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.002$

$\Delta\rho_{\max} = 0.77$ e Å⁻³

$\Delta\rho_{\min} = -0.78$ e Å⁻³

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Ag1	0.33026 (2)	0.42666 (2)	0.38452 (2)	0.01913 (7)
S1	0.80813 (6)	0.92406 (6)	0.27008 (6)	0.01835 (13)
F1	0.59457 (16)	1.01941 (15)	0.16879 (15)	0.0272 (3)
F3	0.77762 (19)	1.00737 (17)	0.09034 (15)	0.0314 (4)
F2	0.80186 (19)	1.16528 (15)	0.27750 (16)	0.0353 (4)
O2	-0.13731 (19)	0.48669 (17)	0.77676 (16)	0.0195 (3)
O1	0.68870 (19)	0.28645 (17)	-0.09989 (16)	0.0209 (4)
O5	0.96843 (19)	0.96695 (17)	0.31236 (17)	0.0238 (4)
O3	0.7525 (2)	0.79175 (17)	0.16192 (18)	0.0263 (4)
N3	0.1996 (2)	0.40757 (19)	0.50408 (18)	0.0151 (4)
N1	0.4587 (2)	0.4184 (2)	0.25573 (18)	0.0156 (4)
O4	0.7400 (2)	0.9518 (2)	0.37019 (19)	0.0350 (5)
N2	0.7732 (2)	0.5144 (2)	-0.0006 (2)	0.0191 (4)
H2A	0.822879	0.511950	-0.054633	0.023*
H2B	0.774705	0.592520	0.062122	0.023*
N4	-0.1246 (2)	0.2695 (2)	0.73398 (19)	0.0184 (4)
H4A	-0.184009	0.264893	0.779082	0.022*
H4B	-0.088239	0.198602	0.695256	0.022*
N6	0.1601 (2)	0.1602 (2)	0.1866 (2)	0.0262 (5)
N5	0.4275 (3)	0.7251 (2)	0.5268 (2)	0.0325 (5)
C9	0.0129 (2)	0.3874 (2)	0.6467 (2)	0.0150 (4)
C3	0.6141 (2)	0.4105 (2)	0.0827 (2)	0.0149 (4)
C2	0.5905 (3)	0.5331 (2)	0.1669 (2)	0.0164 (5)
H2	0.627509	0.615873	0.166627	0.020*
C10	0.0835 (2)	0.5137 (2)	0.6652 (2)	0.0150 (4)
H10	0.068538	0.594775	0.726905	0.018*
C8	0.0393 (3)	0.2713 (2)	0.5556 (2)	0.0167 (5)
H8	-0.005429	0.183595	0.541580	0.020*
C11	0.1752 (3)	0.5196 (2)	0.5931 (2)	0.0157 (4)
H11	0.223022	0.606106	0.606713	0.019*
C12	-0.0895 (3)	0.3837 (2)	0.7243 (2)	0.0158 (4)
C5	0.4811 (3)	0.3005 (2)	0.1740 (2)	0.0174 (5)
H5	0.443286	0.219160	0.176380	0.021*
C6	0.6954 (3)	0.3993 (2)	-0.0140 (2)	0.0165 (5)
C1	0.5128 (3)	0.5324 (2)	0.2506 (2)	0.0171 (5)
H1	0.496800	0.616240	0.307082	0.021*
C4	0.5567 (3)	0.2924 (2)	0.0869 (2)	0.0165 (5)
H4	0.569408	0.206974	0.030278	0.020*
C7	0.1313 (3)	0.2857 (2)	0.4862 (2)	0.0166 (5)
H7	0.147119	0.206188	0.423132	0.020*
C16	0.0818 (3)	0.1753 (2)	0.1065 (2)	0.0220 (5)
C17	-0.0166 (3)	0.1959 (3)	0.0045 (3)	0.0261 (5)
H17A	-0.119652	0.152936	-0.012235	0.039*
H17B	0.012716	0.155320	-0.073582	0.039*
H17C	-0.009032	0.293417	0.030980	0.039*

C14	0.5027 (3)	0.8301 (3)	0.5709 (3)	0.0259 (6)
C13	0.7428 (3)	1.0349 (2)	0.1983 (2)	0.0213 (5)
C15	0.6018 (3)	0.9632 (3)	0.6278 (3)	0.0302 (6)
H15A	0.642520	0.973096	0.563323	0.036*
H15B	0.683648	0.971724	0.700319	0.036*
H15C	0.545821	1.034243	0.657809	0.036*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ag1	0.02101 (11)	0.02377 (11)	0.01987 (11)	0.00738 (7)	0.01347 (7)	0.01206 (8)
S1	0.0220 (3)	0.0175 (3)	0.0209 (3)	0.0087 (2)	0.0108 (2)	0.0103 (2)
F1	0.0224 (7)	0.0293 (8)	0.0322 (8)	0.0106 (6)	0.0075 (6)	0.0153 (7)
F3	0.0405 (9)	0.0372 (9)	0.0328 (9)	0.0161 (7)	0.0204 (7)	0.0243 (7)
F2	0.0379 (9)	0.0142 (7)	0.0423 (10)	0.0057 (6)	0.0000 (8)	0.0083 (7)
O2	0.0237 (9)	0.0203 (8)	0.0220 (8)	0.0116 (7)	0.0139 (7)	0.0113 (7)
O1	0.0266 (9)	0.0191 (8)	0.0220 (9)	0.0069 (7)	0.0151 (7)	0.0093 (7)
O5	0.0220 (9)	0.0198 (8)	0.0302 (10)	0.0066 (7)	0.0069 (7)	0.0117 (7)
O3	0.0290 (10)	0.0154 (8)	0.0335 (10)	0.0036 (7)	0.0094 (8)	0.0100 (8)
N3	0.0163 (9)	0.0182 (9)	0.0141 (9)	0.0071 (7)	0.0076 (7)	0.0080 (8)
N1	0.0157 (9)	0.0194 (9)	0.0153 (9)	0.0062 (7)	0.0073 (8)	0.0092 (8)
O4	0.0402 (12)	0.0530 (13)	0.0336 (11)	0.0270 (10)	0.0243 (9)	0.0294 (10)
N2	0.0229 (10)	0.0198 (10)	0.0190 (10)	0.0041 (8)	0.0137 (8)	0.0089 (8)
N4	0.0220 (10)	0.0187 (10)	0.0229 (10)	0.0086 (8)	0.0154 (8)	0.0116 (8)
N6	0.0259 (11)	0.0295 (11)	0.0251 (11)	0.0045 (9)	0.0108 (9)	0.0125 (10)
N5	0.0456 (15)	0.0285 (13)	0.0290 (12)	0.0141 (11)	0.0177 (11)	0.0133 (10)
C9	0.0143 (11)	0.0192 (11)	0.0139 (11)	0.0049 (9)	0.0044 (9)	0.0093 (9)
C3	0.0132 (10)	0.0173 (11)	0.0165 (11)	0.0050 (8)	0.0047 (9)	0.0092 (9)
C2	0.0174 (11)	0.0175 (11)	0.0178 (11)	0.0058 (9)	0.0069 (9)	0.0098 (9)
C10	0.0163 (11)	0.0156 (10)	0.0135 (10)	0.0061 (8)	0.0054 (9)	0.0058 (9)
C8	0.0178 (11)	0.0167 (11)	0.0184 (11)	0.0047 (9)	0.0075 (9)	0.0091 (9)
C11	0.0161 (11)	0.0164 (11)	0.0164 (11)	0.0052 (8)	0.0053 (9)	0.0085 (9)
C12	0.0151 (11)	0.0198 (11)	0.0147 (11)	0.0057 (9)	0.0055 (9)	0.0088 (9)
C5	0.0175 (11)	0.0171 (11)	0.0200 (12)	0.0040 (9)	0.0074 (9)	0.0094 (9)
C6	0.0143 (10)	0.0207 (11)	0.0180 (11)	0.0068 (9)	0.0057 (9)	0.0107 (9)
C1	0.0184 (11)	0.0183 (11)	0.0174 (11)	0.0060 (9)	0.0072 (9)	0.0091 (9)
C4	0.0177 (11)	0.0169 (11)	0.0162 (11)	0.0054 (9)	0.0074 (9)	0.0070 (9)
C7	0.0201 (11)	0.0161 (11)	0.0158 (11)	0.0059 (9)	0.0082 (9)	0.0072 (9)
C16	0.0253 (13)	0.0204 (11)	0.0227 (13)	0.0038 (10)	0.0143 (11)	0.0081 (10)
C17	0.0293 (14)	0.0265 (13)	0.0261 (13)	0.0082 (11)	0.0116 (11)	0.0131 (11)
C14	0.0328 (14)	0.0335 (15)	0.0226 (13)	0.0188 (12)	0.0154 (11)	0.0170 (12)
C13	0.0243 (12)	0.0169 (11)	0.0247 (13)	0.0077 (9)	0.0091 (10)	0.0095 (10)
C15	0.0310 (15)	0.0315 (14)	0.0327 (15)	0.0113 (12)	0.0126 (12)	0.0162 (12)

Geometric parameters (Å, °)

Ag1—N3	2.162 (2)	C9—C12	1.505 (3)
Ag1—N1	2.162 (2)	C3—C2	1.396 (3)

S1—O5	1.4442 (18)	C3—C6	1.510 (3)
S1—O3	1.4433 (18)	C3—C4	1.390 (3)
S1—O4	1.434 (2)	C2—H2	0.9500
S1—C13	1.831 (2)	C2—C1	1.380 (3)
F1—C13	1.337 (3)	C10—H10	0.9500
F3—C13	1.333 (3)	C10—C11	1.378 (3)
F2—C13	1.331 (3)	C8—H8	0.9500
O2—C12	1.240 (3)	C8—C7	1.380 (3)
O1—C6	1.235 (3)	C11—H11	0.9500
N3—C11	1.348 (3)	C5—H5	0.9500
N3—C7	1.347 (3)	C5—C4	1.380 (3)
N1—C5	1.346 (3)	C1—H1	0.9500
N1—C1	1.345 (3)	C4—H4	0.9500
N2—H2A	0.8800	C7—H7	0.9500
N2—H2B	0.8800	C16—C17	1.459 (4)
N2—C6	1.338 (3)	C17—H17A	0.9800
N4—H4A	0.8800	C17—H17B	0.9800
N4—H4B	0.8800	C17—H17C	0.9800
N4—C12	1.331 (3)	C14—C15	1.460 (4)
N6—C16	1.144 (3)	C15—H15A	0.9800
N5—C14	1.141 (4)	C15—H15B	0.9800
C9—C10	1.395 (3)	C15—H15C	0.9800
C9—C8	1.395 (3)		
N1—Ag1—N3	172.78 (7)	O2—C12—C9	118.8 (2)
O5—S1—C13	103.79 (11)	N4—C12—C9	118.3 (2)
O3—S1—O5	113.68 (11)	N1—C5—H5	118.5
O3—S1—C13	101.70 (11)	N1—C5—C4	122.9 (2)
O4—S1—O5	115.49 (12)	C4—C5—H5	118.5
O4—S1—O3	116.01 (13)	O1—C6—N2	123.2 (2)
O4—S1—C13	103.62 (11)	O1—C6—C3	119.5 (2)
C11—N3—Ag1	119.93 (16)	N2—C6—C3	117.3 (2)
C7—N3—Ag1	121.89 (15)	N1—C1—C2	123.0 (2)
C7—N3—C11	118.0 (2)	N1—C1—H1	118.5
C5—N1—Ag1	122.04 (16)	C2—C1—H1	118.5
C1—N1—Ag1	120.28 (16)	C3—C4—H4	120.3
C1—N1—C5	117.6 (2)	C5—C4—C3	119.4 (2)
H2A—N2—H2B	120.0	C5—C4—H4	120.3
C6—N2—H2A	120.0	N3—C7—C8	122.9 (2)
C6—N2—H2B	120.0	N3—C7—H7	118.5
H4A—N4—H4B	120.0	C8—C7—H7	118.5
C12—N4—H4A	120.0	N6—C16—C17	179.3 (3)
C12—N4—H4B	120.0	C16—C17—H17A	109.5
C10—C9—C12	118.3 (2)	C16—C17—H17B	109.5
C8—C9—C10	118.2 (2)	C16—C17—H17C	109.5
C8—C9—C12	123.5 (2)	H17A—C17—H17B	109.5
C2—C3—C6	123.5 (2)	H17A—C17—H17C	109.5
C4—C3—C2	117.9 (2)	H17B—C17—H17C	109.5

C4—C3—C6	118.5 (2)	N5—C14—C15	178.8 (3)
C3—C2—H2	120.5	F1—C13—S1	110.99 (17)
C1—C2—C3	119.1 (2)	F3—C13—S1	111.58 (16)
C1—C2—H2	120.5	F3—C13—F1	107.2 (2)
C9—C10—H10	120.3	F2—C13—S1	111.75 (17)
C11—C10—C9	119.4 (2)	F2—C13—F1	107.49 (19)
C11—C10—H10	120.3	F2—C13—F3	107.6 (2)
C9—C8—H8	120.5	C14—C15—H15A	109.5
C7—C8—C9	118.9 (2)	C14—C15—H15B	109.5
C7—C8—H8	120.5	C14—C15—H15C	109.5
N3—C11—C10	122.6 (2)	H15A—C15—H15B	109.5
N3—C11—H11	118.7	H15A—C15—H15C	109.5
C10—C11—H11	118.7	H15B—C15—H15C	109.5
O2—C12—N4	122.8 (2)		
Ag1—N3—C11—C10	174.85 (16)	C2—C3—C4—C5	-0.8 (3)
Ag1—N3—C7—C8	-175.54 (17)	C10—C9—C8—C7	-1.2 (3)
Ag1—N1—C5—C4	176.85 (17)	C10—C9—C12—O2	17.8 (3)
Ag1—N1—C1—C2	-177.40 (17)	C10—C9—C12—N4	-162.0 (2)
O5—S1—C13—F1	-172.06 (16)	C8—C9—C10—C11	0.4 (3)
O5—S1—C13—F3	68.43 (19)	C8—C9—C12—O2	-161.1 (2)
O5—S1—C13—F2	-52.1 (2)	C8—C9—C12—N4	19.1 (3)
O3—S1—C13—F1	69.70 (18)	C11—N3—C7—C8	-0.6 (3)
O3—S1—C13—F3	-49.8 (2)	C12—C9—C10—C11	-178.5 (2)
O3—S1—C13—F2	-170.33 (18)	C12—C9—C8—C7	177.7 (2)
N1—C5—C4—C3	0.6 (3)	C5—N1—C1—C2	-0.7 (3)
O4—S1—C13—F1	-51.0 (2)	C6—C3—C2—C1	179.0 (2)
O4—S1—C13—F3	-170.54 (18)	C6—C3—C4—C5	-179.5 (2)
O4—S1—C13—F2	68.9 (2)	C1—N1—C5—C4	0.2 (3)
C9—C10—C11—N3	0.3 (3)	C4—C3—C2—C1	0.4 (3)
C9—C8—C7—N3	1.3 (3)	C4—C3—C6—O1	11.6 (3)
C3—C2—C1—N1	0.4 (3)	C4—C3—C6—N2	-167.8 (2)
C2—C3—C6—O1	-167.0 (2)	C7—N3—C11—C10	-0.2 (3)
C2—C3—C6—N2	13.6 (3)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N2—H2 <i>A</i> ...O2 ⁱ	0.88	2.06	2.898 (3)	160
N2—H2 <i>B</i> ...O3	0.88	2.09	2.939 (3)	162
N4—H4 <i>A</i> ...O1 ⁱⁱ	0.88	2.05	2.927 (3)	171
N4—H4 <i>B</i> ...O5 ⁱⁱⁱ	0.88	2.22	3.033 (3)	154

Symmetry codes: (i) $x+1, y, z-1$; (ii) $x-1, y, z+1$; (iii) $-x+1, -y+1, -z+1$.