



Aptamer-based cocaine assay using a nanohybrid composed of ZnS/Ag₂Se quantum dots, graphene oxide and gold nanoparticles as a fluorescent probe

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Abstract

Authors report on a new fluoro-graphene-plasmonic nanohybrid aptamer-based fluorescent nanoprobe for cocaine. To construct the nanoprobe, newly synthesized glutathione-capped ZnS/Ag₂Se quantum dots (QDs) were first conjugated to graphene oxide (GO) to form a QD-GO nanocomposite. The binding interaction resulted in a fluorescence turn-ON. Thereafter, cetyltrimethylammonium bromide (CTAB)-gold nanoparticles (AuNPs) were directly adsorbed on the QD-GO nanocomposite to form a novel QD-GO-CTAB-AuNP nanohybrid assembly that resulted in a fluorescence turn-OFF. Streptavidin (strep) was then adsorbed on the QDs-GO-CTAB-AuNP nanohybrid assembly which allowed binding to a biotinylated MNS 4.1 anticocaine DNA aptamer (B) receptor. The addition of cocaine into the strep-B-QDs-GO-CTAB-AuNP aptamer nanoprobe system aided affinity to the aptamer receptor and in turn turned on the fluorescence of the nanoprobe in a concentration-dependent manner. Under optimum experimental conditions, we found the strep-B-QD-GO-CTAB-AuNP to be far superior in its sensitivity to cocaine than the tested strep-B-QDs (no GO and CTAB-AuNPs), strep-B-QD-CTAB-AuNP (no GO) and strep-B-QD-GO (no CTAB-AuNP). In addition, the investigation of localized surface plasmon resonance (LSPR) amplified signal from tested plasmonic NPs shows that CTAB-AuNPs was far superior in amplifying the fluorescence signal of the nanoprobe. A detection limit of 4.6 nM (1.56 ng.mL⁻¹), rapid response time (~2 min) and excellent selectivity against other drugs, substances and cocaine metabolites was achieved. The strep-B-QD-GO-CTAB-AuNP aptamer-based fluorescent nanoprobe was successfully applied for the determination of cocaine in seized adulterated cocaine samples.

Keywords Nanohybrid · Graphene oxide · Quantum dots · Gold nanoparticle · Fluorescence

Introduction

Abuse of cocaine remains a critical public health challenge due to the associated health risks connected with its illicit use. Organ damage, cardiovascular problems, insomnia, loss

of appetite, violent behaviour, anxiety and paranoia are all documented examples of health-related issues associated with cocaine use [1]. Rapid, simple, selective, sensitive and inexpensive methods for cocaine detection are currently needed to meet the detection requirements (both presumptive and confirmatory) for efficient analysis. Traditional confirmatory chemical analysis for cocaine identification relies on complex separation techniques such as high-performance liquid chromatography (HPLC) and gas chromatography-mass spectrometry (GC-MS) [2]. Despite these techniques being able to achieve detection limit levels required for confirmatory identification of cocaine, practical drawbacks for real-life implementation such as reproducibility, throughput, mobility and cost, often leads to challenging decisions on the trade-offs associated with various aspects of the analytical device output if further miniaturization are desired for real-time, in-field and wearable applications [3]. To overcome the drawbacks of chromatographic techniques, aptamer-based

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nanoprobes have been investigated as alternative detection methods for cocaine identification [4].

Aptamers are single-stranded synthetic DNA or RNA oligonucleotide sequences that selectively bind to a target analyte via shape recognition [5]. Aptamers used in biosensing applications are nearly exclusively artificial and are evolved from a large random pool of oligonucleotide sequences through the process known as Systematic Evolution of Ligands by Exponential Enrichment (SELEX) [6]. To date, a variety of aptameric nanoprobes based on fluorescence transduction assays have been developed for cocaine [7–9]. However, there is still a continual need to develop more efficient probes to meet the growing demand of rapid and accurate testing of illicit drugs.

With the advent of new nano-based materials and components, more sophisticated soft-matter nanofabrication and targeted assembly techniques can lead to the development of prospective nanoprobes devices for illicit drugs by utilizing highly selective and sensitive biocompatible sensing elements. There is little doubt that nanoprobes based on the direct combination of nanoreporter elements and biorecognition units, that are capable of molecular binding events and on-spot transduction, are a far more efficient method for biorecognition unit utilization than traditional chemical-based assays. It is foreseen that the utilization of hybrid nanostructures in nanoprobe development [10] may pave the way to overcome the challenges associated with achieving global commercial success for nanobiosensing of illicit drugs.

In this work, we report on the development of a new fluorescent nanohybrid assembly nanoprobe platform for cocaine identification based on the use of an anticocaine aptamer as a bioreceptor and fluorescence technique as the transduction signal. ZnS/Ag₂Se quantum dots (QDs), capped with L-glutathione (GSH), and a novel component for the nanoprobe platform was synthesized and used as a fluorescent-emitting nanoreporter element.

Graphene oxide (GO), one of the components of the nanohybrid assembly, is a 2D lattice of graphitic carbon atoms densely packed in a honeycomb like hexagonal pattern [11]. The large 2D aromatic structural surface of GO makes it an ideal platform for the adsorption of biomolecules, while its electrical properties and high surface-to-volume ratio are known to influence sensing signals [12]. Metallic gold nanoparticle (AuNP), another component of the nanohybrid assembly, is characterized by the unique absorption and scattering properties dominated by the localized surface plasmon resonance (LSPR) and is considered the most stable of the noble metal NP class [13]. Its resistance to surface oxidation and chemical inertness have attracted extensive use in nanotechnology applications [14]. Particularly, the LSPR signal of AuNPs has been known to amplify fluorescence signal of QDs, thereby leading to lower detection limits [15].

To construct the nanoprobe, GO was first conjugated to L-glutathione (GSH)-ZnS/Ag₂Se core/shell QDs to form a QDs-GO nanocomposite which resulted in a fluorescence enhancement (“ON state”) process. Thereafter, plasmonic cetyltrimethylammonium bromide (CTAB)-AuNPs were adsorbed onto the QDs-GO nanocomposite to form a QDs-GO-CTAB-AuNP nanohybrid assembly which triggered a fluorescence quenching (“OFF state”) process. Streptavidin (Strep) protein was then adsorbed onto the QDs-GO-CTAB-AuNP nanohybrid to form a strep-QDs-GO-CTAB-AuNP nanohybrid assembly. The incorporation of a biotinylated anticocaine DNA aptamer (B) into the strep-QDs-GO-CTAB-AuNP nanohybrid system, ensured the binding affinity to the adsorbed streptavidin. Cocaine being added to the nanoprobe system and its subsequent binding to the aptamer receptor, triggered fluorescence enhancement of the strep-B-QDs-GO-CTAB-AuNP nanohybrid nanoprobe (“ON state”). The plasmonic CTAB-AuNP was found to amplify the fluorescence signal in the presence of detected cocaine concentration via localized surface plasmon resonance (LSPR)-enhanced fluorescence intensity. To the best of our knowledge, this is the first reported QDs-GO-CTAB-AuNP nanohybrid aptamer-based fluorescence nanoprobe for cocaine.

Experimental

Materials

Silver nitrate (AgNO₃) (≥ 99%), sodium chloride (≥ 99%), potassium chloride (99.0–100.5%), acetaminophen (≥ 99.0%), d-amphetamine, (+) methamphetamine hydrochloride, sodium phosphate dibasic (≥ 99.0%), potassium phosphate dibasic (≥ 99.0%), benzocaine (≥ 99%), lidocaine hydrochloride (≥ 99%), cocaine hydrochloride (≥ 97.5%), (+) methamphetamine hydrochloride, phenacetin (≥ 98%), benzoylecgonine solution, ecgonine methyl ester solution, octadecene (90%), trioctylphosphine oxide (99%), trioctylphosphine (97%), hexadecylamine (90%), sulphur (99.999%), selenium (Se) (99.5 + %), oleic acid (99%), and H₂O₂ (30% w/w) in solution with stabilizer were purchased from Sigma Aldrich (<https://www.sigmaaldrich.com/united-kingdom.html>). Tris(hydroxymethyl)aminomethane (≥99.8%) was purchased from Formedium (<https://www.formedium.com/>). Levamisole HCl (99 + %), diltiazem (98%), CTAB (~98%), MES (99 + %), gold (III) chloride trihydrate (HAuCl₄·3H₂O) (≥99.9%), potassium acetate (KAc) (≥99.0%), Thermo Scientific™ Pierce™ EDC Crosslinker, N-hydroxysuccinimide, sodium acetate (NaAc), graphite powder (99.9995%), diethylzinc, GSH (98 + %) and potassium permanganate (KMnO₄) (99 + %) were purchased from Thermo Fisher (<https://www.fishersci.co.uk/gb/en/home.html>). The biotinylated MNS

4.1 anticocaine DNA aptamer oligonucleotide with the nucleic acid sequence:

BIOTEG-5'- GGGAGACAAGGAAAATCCTTCAATGA AGTGGGTCGACA was synthesized and purified by Eurofins (<https://www.eurofins.co.uk/>). All other chemicals were used as received. A Milli-Q water system was used as the water source.

Characterization

Fluorescence and UV/vis absorption measurements were carried out on a Cary Eclipse (Varian) spectrophotometer. Scanning electron microscopy (SEM) and energy dispersive X-ray (EDX) analysis were carried out using a JEOL JSM 7400F field emission SEM integrated with an Oxford Instruments Inca EDX spectrometer. Transmission electron microscopy (TEM) measurements were carried out using a JEOL JEM-1200EX operated at 80 kV. Zeta potential (ZP) and dynamic light scattering (DLS) analysis were carried out using Zetasizer Nano ZS series (ZEN3600, Malvern). The Raman spectra were collected using an in-house built microprobe system equipped with a continuous wave laser sources emitting at 633 nm, the Oriol MS257 monochromator fitted with the Andor Newton EMCCD detector TE-cooled to -70°C . The backscattering configuration was used for the signal collection. The incident power on the samples was 7 mW. The spectra were recorded using a 40x objective, a 1 s accumulation time with a total of 10 accumulations. The synthesis of GSH-ZnS/Ag₂Se core/shell QDs [16], GO nanosheets [17, 18] and CTAB-AuNPs [19] are described in the [Electronic Supplementary Material](#).

Conjugation of GO nanosheets to GSH-ZnS/Ag₂Se core/shell QDs

Scheme 1 shows the process for the conjugation of GO to GSH-ZnS/Ag₂Se core/shell QDs. To activate the carboxylic groups on GO, 20 mg GO was dissolved in 5 mL of MilliQ H₂O and 10 mL 0.1 M EDC was added and stirred for ~10 min. Thereafter, 20 mg GSH-ZnS/Ag₂Se QDs in 25 mL MilliQ H₂O and 10 mL 0.1 M NHS were added and the solution was stirred for ~3 h. The QDs-GO fluorescent nanocomposite was purified by acetone via centrifugation and later dried.

Preparation of strep-B-QDs-GO-CTAB-AuNP nanohybrid assembly

To form the QDs-GO-CTAB-AuNP nanohybrid assembly, 0.4 mg.mL⁻¹ QDs-GO nanocomposite solution was mixed with a solution of CTAB-AuNPs (0.1 nM) in a 1:2 v/v ratio, yielding a final concentration of 0.06 nM QDs-GO-CTAB-

AuNP. Thereafter, the QDs-GO-CTAB-AuNP solution (0.06 nM) was mixed with 0.5 mg.mL⁻¹ streptavidin solution (in phosphate buffered saline (PBS), pH 7.4) in a 6:1 v/v ratio to aid adsorption of the protein onto the nanohybrid assembly. Subsequently, 10 μM biotinylated MNS 4.1 anticocaine DNA aptamer was added into the strep-QDs-GO-CTAB-AuNP solution and stirred to aid binding of the biotin end of the DNA aptamer to the streptavidin portion of the QDs-GO-CTAB-AuNP nanohybrid assembly. The formed strep-B-QDs-GO-CTAB-AuNP nanohybrid nanoprobe solution was stored at 4°C prior to use.

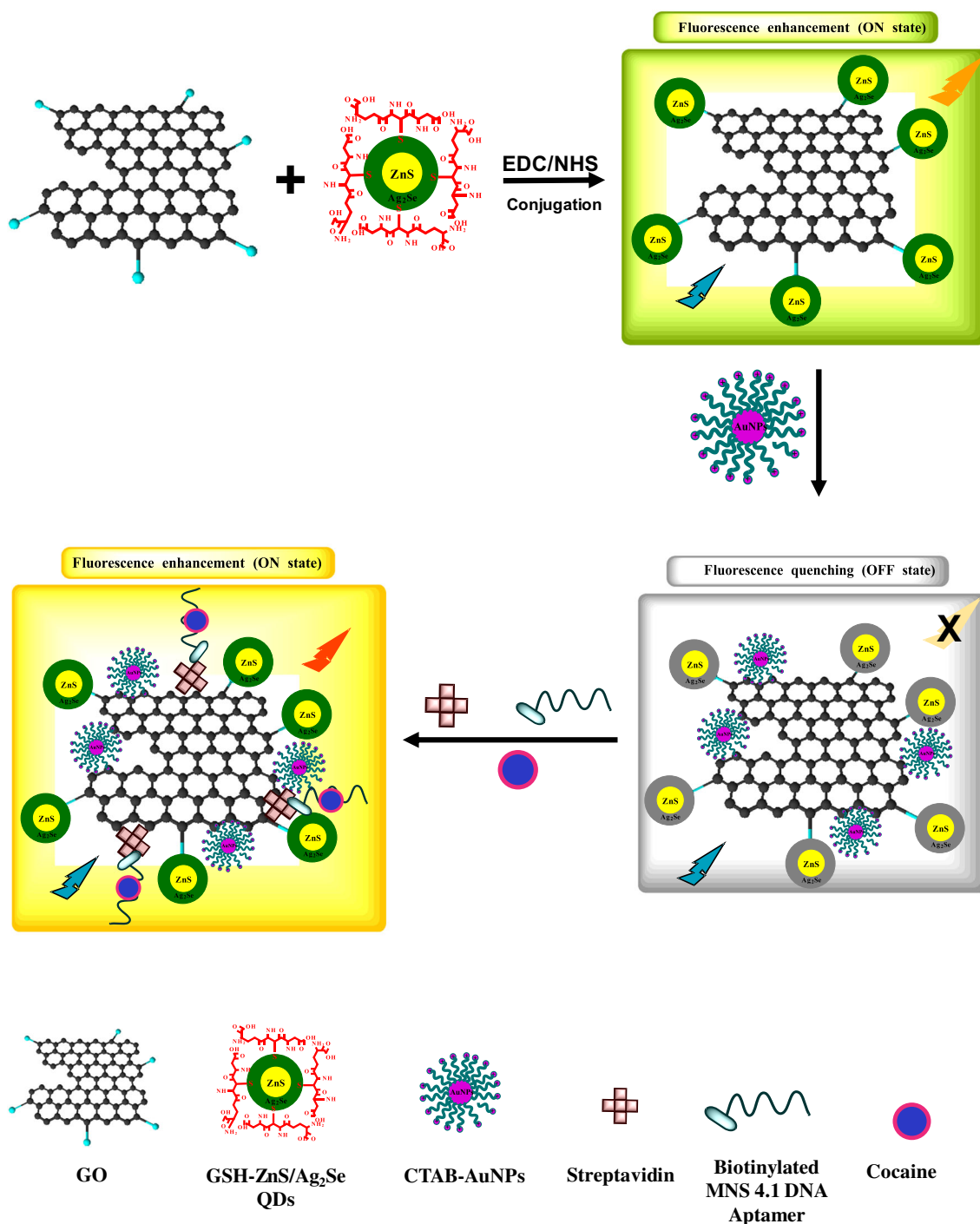
Fluorescence bioassay

A novel buffer for cocaine determination was prepared by mixing 0.38 g MES, 0.25 g NaAc, 0.25 g KAc, 0.37 g KCl in 50 mL of MilliQ H₂O and 6.7 mL 0.1 M HCl was added and made up to 100 mL with MilliQ H₂O. Thereafter, the pH was adjusted to 2.2 using a solution of HCl and NaOH. For the fluorescence detection, appropriate cocaine concentration was dissolved in MES-NaAc-KAc-KCl-HCl buffer, pH 2.2 and 175 μl of the cocaine solution was mixed with 175 μl strep-B-QDs-GO-CTAB-AuNP nanohybrid nanoprobe solution. The fluorescence measurement was taken immediately in a quartz cuvette (~ 2 min) at an excitation wavelength of 290 nm within the photoluminescence (PL) wavelength range of 300 nm to 650 nm.

Results and discussion

Structural properties

GO is known to be characterized by automatically thin and continuous two-dimensional (2D) carbon atom arrays that is embedded with carboxy groups on the edges and hydroxy and epoxy groups on the carbon basal plane [20]. Due to a significant portion of the basal plane being comprised of carbon domains, it is believed these regions of the graphene sheet are relatively hydrophobic [21]. However, GO is also thought to be composed of an amphiphilic structure which contributes to its aggregation and adsorption properties [21]. The SEM image of GO nanosheets obtained from chemical exfoliation of graphite oxide is shown in Fig. 1a. From the SEM image, GO nanosheets is observed to be characterized by crumpled and randomly aggregated thin sheets that folds like wrinkled silk waves into individual nanosheets. The corresponding SEM image of the QDs-GO nanocomposite (Fig. 1b) shows the presence of the QDs particles well anchored on the GO nanosheet surface. Similar features of embedded particles on the GO nanosheet surface was also observed for the QDs-GO-CTAB-AuNP nanohybrid assembly (Fig. 1c). From the



Scheme 1 Descriptive representation of the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe for cocaine. GO is first conjugated to the QDs to form a QDs-GO nanocomposite which then results in fluorescence emission enhancement (ON state). Plasmonic CTAB-AuNP is then adsorbed to the QDs-GO nanocomposite to form a QDs-GO-CTAB-AuNP nanohybrid assembly and the fluorescence emission is quenched (OFF state). Streptavidin protein is then adsorbed

onto the QDs-GO-CTAB-AuNP nanohybrid assembly and a biotinylated DNA anticocaine aptamer receptor is bonded to the adsorbed streptavidin. Cocaine being added to the system, binds to the aptamer receptor and this interaction switches on the fluorescence intensity of the strep-B-QDs-GO-CTAB-AuNP nanohybrid assembly in a concentration-dependent manner

observed morphological feature, it is noteworthy to affirm that both the QDs and plasmonic CTAB-AuNPs were well anchored on the GO nanosheet surface.

For the TEM analysis, Fig. 2a shows that the morphology of GO is characterized by exfoliated folded individual nanosheets with no trace of bulk aggregates. The TEM image of

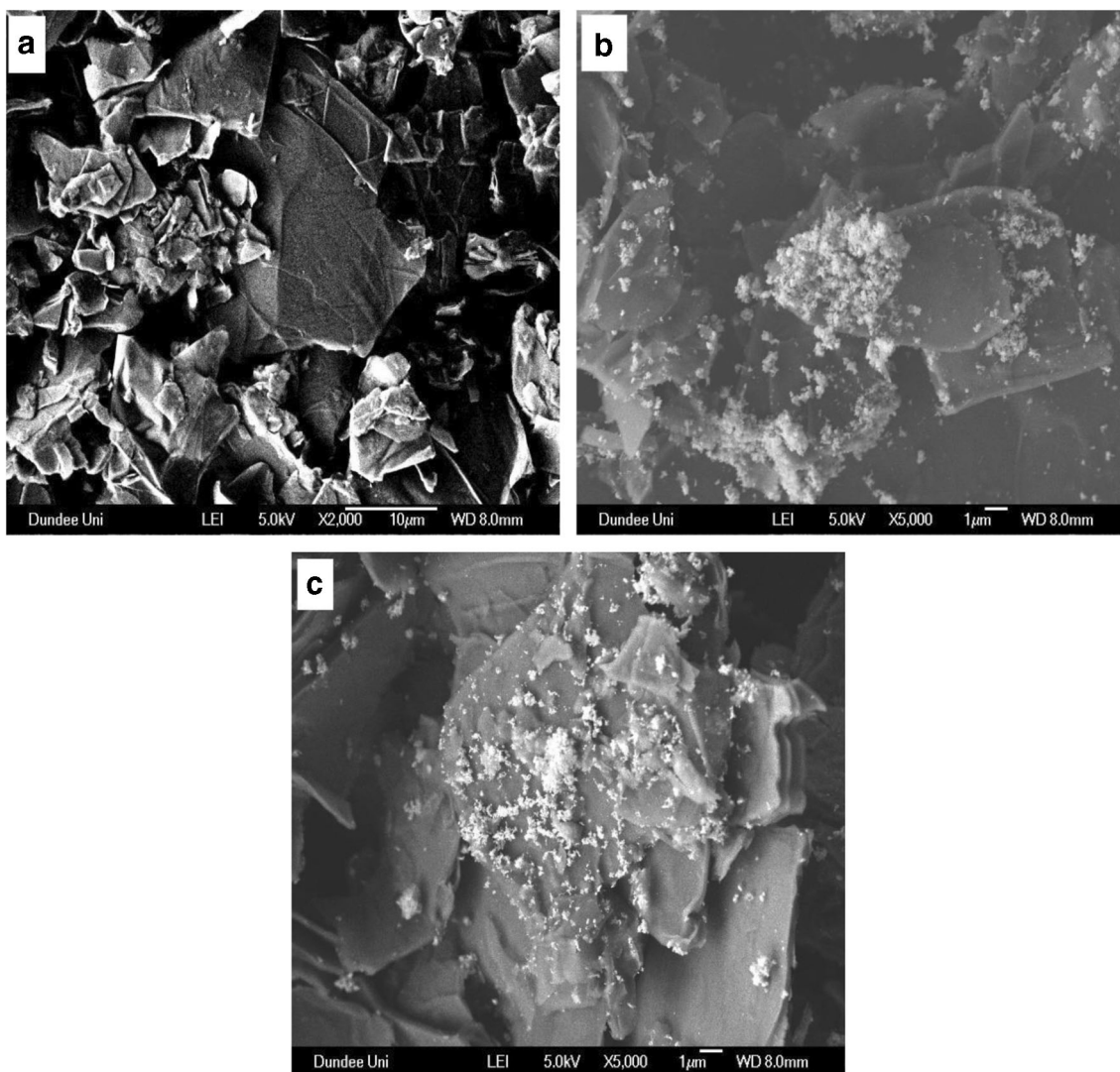
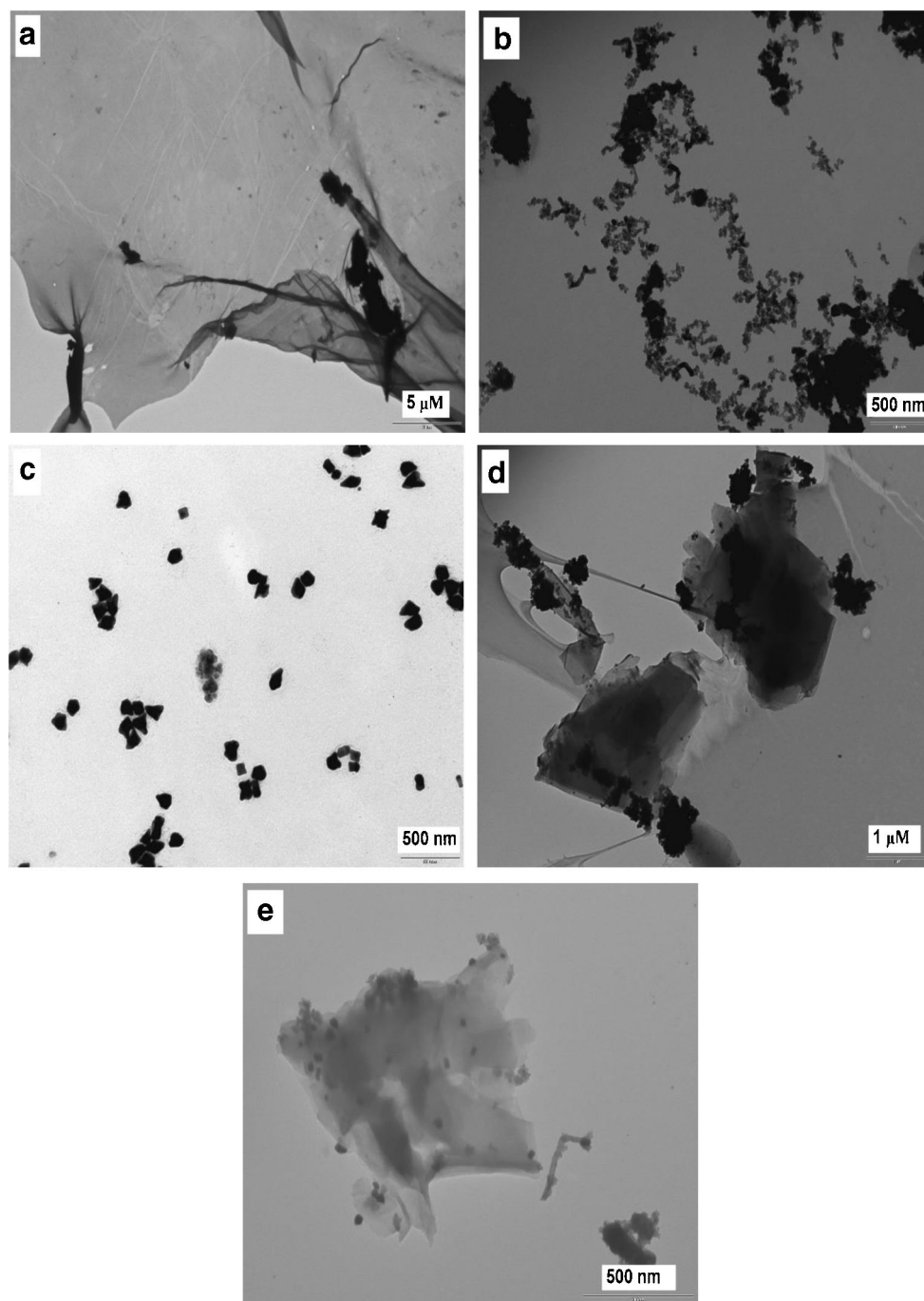


Fig. 1 a SEM images of (a) GO, b QDs-GO nanocomposite and c QDs-GO-CTAB-AuNP nano hybrid

the newly synthesized GSH-ZnS/Ag₂Se QDs was characterized by heterogenous QDs growth morphology as shown in Fig. 2b. Studies have shown that crystalline overcoating of a semiconductor layer may be formed on QDs nanocrystals either by ion displacement reactions or heterogeneous nucleation [22]. In some cases, it has been suggested that heterogeneous colloidal semiconductor particles exhibit better efficiency than single ensemble particles [23]. In general, shell coating on a core particle (in this case, the coating of Ag₂Se shell on ZnS core) normally occurs via heterogeneous nucleation. It is postulated that the direct deposition of the embryos of shell materials on the core surface, triggers the continuous formation of nuclei and QDs growth on the surface itself, instead of the formation of new nuclei in the bulk phase [23]. Therefore, the heterogenous formation of GSH-ZnS/Ag₂Se QDs can be explained in terms of this phenomenon. Figure 2c shows the TEM image

of CTAB-AuNPs. From the morphological display, the presence of cubic and prism-shaped particles was observed while most of the particles were irregularly-shaped. The average particle size was determined to be ~43 nm from 57 measured particles. Figure 2d and e shows the TEM images of the QDs-GO nanocomposite and the QDs-GO-CTAB-AuNP nano hybrid assembly. From the TEM images, it is apparent that GO sheets characterized by wrinkled surfaces were visibly seen for both images. Particularly, the binding of the QDs on the GO surface appears to be in lumps along the sheet edges. A close observation of the QDs-GO-CTAB-AuNP morphology reveal both the presence of particles spread across the sheet surface and the presence of lumps of particles along the sheet edge. Text and figures on the EDX analysis (Fig. S1), XRD pattern (Fig. S2), DLS and ZP analysis (Fig. S3) of the nanomaterials are provided in the [Electronic Supporting Material](#).

Fig. 2 a TEM images of (a) GO, (b) GSH-ZnS/Ag₂Se QDs, (c) CTAB-AuNPs, (d) QDs-GO nanocomposite and (e) QDs-GO-CTAB-AuNP nano hybrid

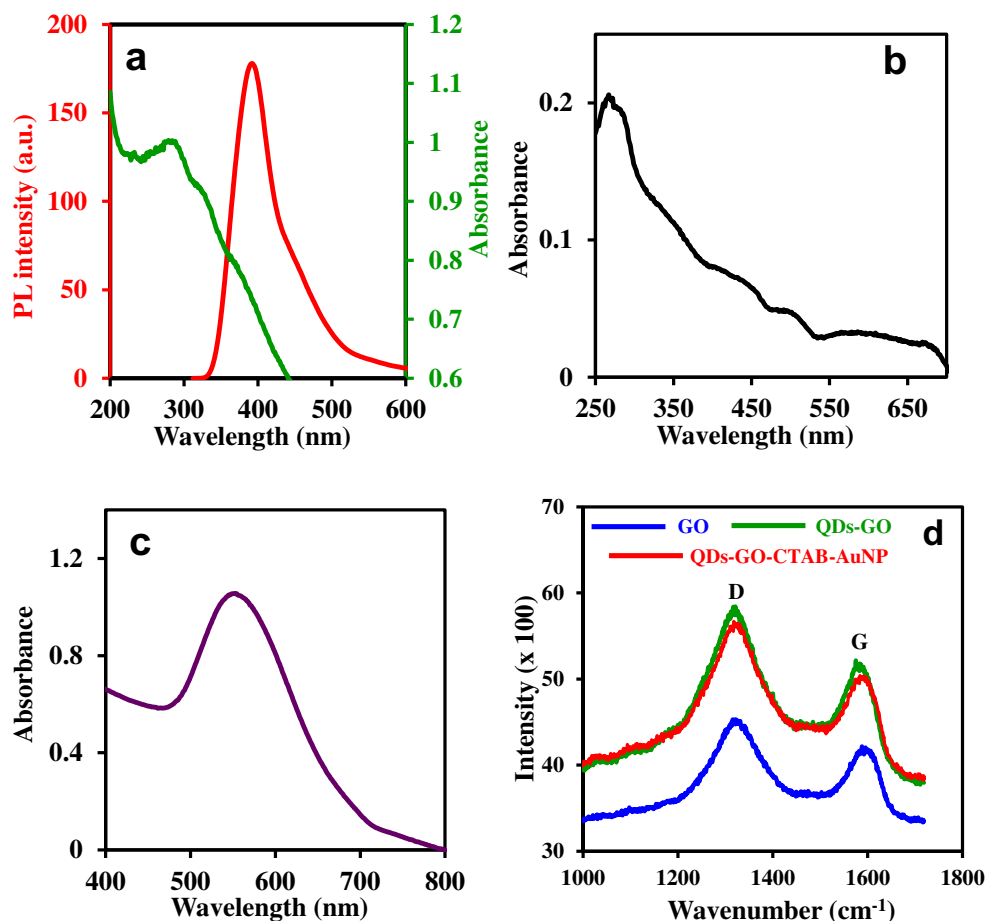


Optical properties

Figure 3a shows the UV/vis absorption and PL emission spectra of the GSH-ZnS/Ag₂Se QDs. The UV/vis absorption spectrum of the QDs is characterized by an absorption peak maximum at a wavelength of 288 nm and the PL emission spectrum is characterized by an emission wavelength at 395 nm. The UV/vis absorption spectrum of GO, shown in Fig. 3b is consistent with the absorption spectrum for the exfoliation of graphite oxide to GO and is distinct for a quasi-2D material.

Consistent with published UV/vis spectrum for GO, the observed absorption maximum around 267 nm is attributed to the $\pi \rightarrow \pi^*$ transition of the aromatic C-C bond [24], while the relatively broad absorption around 331–371 nm is attributed to the $n \rightarrow \pi^*$ transition of the C=O bonds [25]. UV/vis absorption spectrum of the positively-charged plasmonic CTAB-AuNPs (Fig. 3c) is characterized by a distinct LSPR absorption band at 554 nm and is consistent with the LSPR absorption wavelength range for non-spherical AuNPs [26].

Fig. 3 a UV/vis and PL emission spectra of GSH-ZnS/Ag₂Se QDs. UV/vis absorption spectra of (b) GO and (c) CTAB-AuNPs. (d) Raman spectra of GO, QDs-GO nanocomposite and the QDs-GO-CTAB-AuNP nano hybrid. $\lambda_{exc} = 290$ nm



The characteristic D and G bands of carbon-based materials, projected by Raman analysis, have been used extensively to analyse the effects of carbon grain size, disorder and defect states on the carbon graphitic domain [27]. The Raman spectra of GO, QDs-GO nanocomposite and QDs-GO-CTAB-AuNP are shown in Fig. 3d. From the displayed spectra, GO is characterized by the D and G bands which is also visibly projected in the spectra of the QDs-GO nanocomposite and the QDs-GO-CTAB-AuNP nano hybrid. The D band corresponds to defects associated with amorphous carbon species, grain boundaries and vacancies, while the G band is associated with the E_{2g} mode of graphite which relates to sp²-bonded vibrational carbon atoms in a 2D hexagonal lattice [27]. From Fig. 3d, the D bands of GO, QDs-GO nanocomposite and QDs-GO-CTAB-AuNP are projected at 1336 cm⁻¹, 1321 cm⁻¹ and 1322 cm⁻¹ while the G bands are projected at 1595 cm⁻¹, 1575 cm⁻¹ and 1584 cm⁻¹ respectively. From the data, the binding of GO to the QDs and the subsequent adsorption of CTAB-AuNPs to the QD-GO nanocomposite induced both the D and G bands to shift to lower wavenumber relative to GO. The ratio of intensity (I_D/I_G) of the D band relative to the G band is usually used to determine the average size of the sp² graphitic domain and the extent of defect/disorder degree [28]. The obtained I_D/I_G value of 1.12 for

the QD-GO nanocomposite and QDs-GO-CTAB-AuNP nano hybrid was higher than the value obtained for GO ($I_D/I_G = 1.06$). Thus, it implies that the nano hybrids did not increase the ordered degree of graphitic carbon layers. Text and figures (Fig. S4A-C) on the FT-IR analysis of the QDs, GO, CTAB-AuNPs, QDs-GO nanocomposite and the QDs-GO-CTAB-AuNPs nano hybrid assembly are provided in the [Electronic Supporting Material](#).

Binding effects on fluorescence emission and LSPR absorption

The effects of binding interactions on the fluorescence emission of the QDs and the LSPR absorption of the plasmonic CTAB-AuNPs was studied. Figure 4a shows the fluorescence emission changes induced by the binding of GO to the QDs, binding of CTAB-AuNPs to the QDs-GO nanocomposite and binding of the streptavidin-biotinylated DNA aptamer to the QDs-GO-CTAB-AuNP nano hybrid. From the data, the binding of GO to the QDs resulted in marked fluorescence enhancement (ON state) and a red-shift in emission wavelength (17 nm) relative to the fluorescence of the unconjugated QDs. The red-shift may be attributed to agglomeration of the QDs upon binding to GO. Upon adsorption of the plasmonic

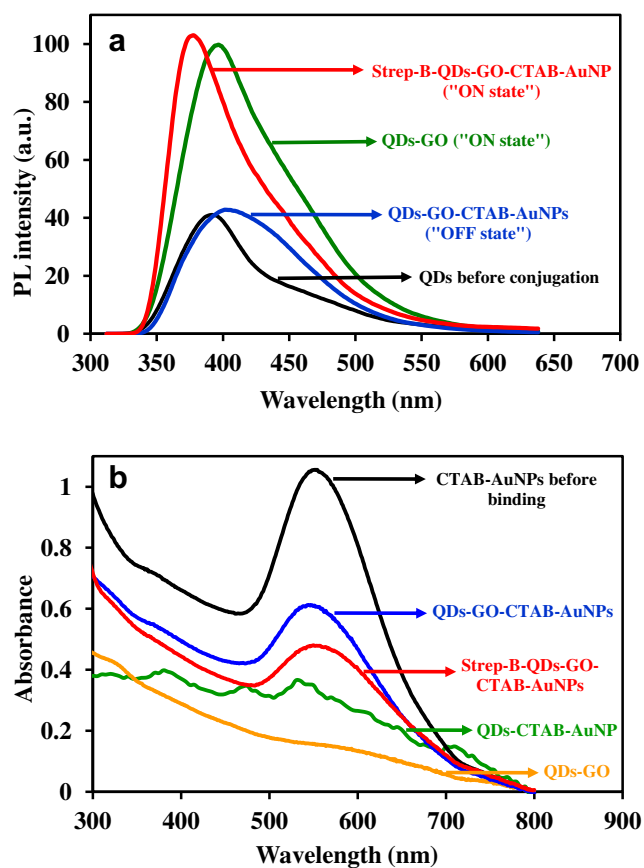


Fig. 4 **a** PL emission spectra of the QDs before conjugation, QDs-GO nanocomposite, QDs-GO-CTAB-AuNP and strep-B-QDs-GO-CTAB-AuNP nanohybrid assembly. **b** UV/vis absorption spectra of CTAB-AuNPs before binding, QDs-GO nanocomposite, QDs-CTAB-AuNPs, QDs-GO-CTAB-AuNP and the strep-B-QDs-GO-CTAB-AuNP nanohybrid assembly

CTAB-AuNPs to the QDs-GO nanocomposite, the fluorescence was quenched, leading to the fluorescence of the conjugated QDs being switched off. The subsequent binding interaction between streptavidin and the biotinylated DNA aptamer on the QDs-GO-CTAB-AuNP nanohybrid assembly, led to the fluorescence being switched on and a blue-shift of the emission wavelength. The blue shift in emission wavelength can be attributed to disaggregation of the QDs upon binding of the protein and DNA aptamer. Thus, the strep-B-QDs-GO-CTAB-AuNP nanoprobe can function as an “ON-OFF-ON” aptamer-based fluorescence nanoprobe for cocaine.

Figure 4b shows the comparative binding changes on the LSPR absorption feature of CTAB-AuNPs. From the data, the binding of the QDs and GO individually to CTAB-AuNPs led to disappearance of the LSPR absorption. However, when CTAB-AuNPs was directly bonded to the QDs-GO nanocomposite, the LSPR absorption peak was well projected but relatively quenched in comparison to CTAB-AuNPs. The subsequent binding of the streptavidin-biotinylated DNA aptamer to the QDs-GO-CTAB-AuNP nanohybrid did not induce disappearance of the LSPR absorption peak but rather led to

further reduction of the LSPR absorption. Based on the observed result, it is noteworthy to suggest that the presence of GO in the nanohybrid system, created a platform for retaining the LSPR absorption characteristic of the plasmonic CTAB-AuNPs, which is essential as a signal amplifier for ultrasensitive fluorescence determination of the target cocaine.

Working principle of the nanoprobe (Scheme 1)

GO was first conjugated to the QDs to form the QDs-GO nanocomposite via EDC/NHS carbodiimide conjugation chemistry involving the formation of an amide bond linkage between the carboxylic groups on GO and the amino group on the QDs. The binding interaction triggered a fluorescence enhancement process, which is a direct contrast to the quenching effects of GO on QDs fluorescence as reported in the literature [29–31]. It is also noteworthy to emphasize that the fluorescence enhancement effect of GO on QDs [17] and on a molecular fluorophore [32] have been reported in the literature. Förster resonance energy transfer (FRET) from the donor QDs to the acceptor GO has been the most popular mode of quenching interaction reported between fluorescent-emitting QDs and GO [33]. Even though we observed a spectral overlap between the QDs fluorescence and GO absorption spectrum as shown in Fig. S5A, the absence of fluorescence quenching indicates that a different mode of chemical interaction occurred between the QDs and GO. Based on reported interpretation on the fluorescence enhancement effect of GO on QDs fluorescence [32], we believe that when the band gap of the QDs is less than or equal to the phonon energies, photoexcited electrons from the valence to the conduction band can induce equal amounts of holes in the valence band [34]. Judging by the morphology of the QDs shown via the TEM image, it is likely possible that interfacial surface defect created non-radiative electron-hole dynamics on the QDs surface. Therefore, we believe that conjugation of GO to the QDs induced conformability between the defect state of the QDs and the adsorptive properties of GO which may have induced fluorescence enhancement of the QDs [32].

CTAB-AuNPs were subsequently bound to the QD-GO nanocomposite and can bind to the conjugated-GO via electrostatic interaction between the CTAB headgroup and free OH groups on GO and via hydrophobic interaction involving the hydrophobic alkyl chains of CTAB [35]. The binding of CTAB-AuNPs then triggered fluorescence quenching of the QDs-GO-CTAB-AuNP nanohybrid assembly. To investigate the fluorescence quenching effect of CTAB on the QDs-GO nanocomposite, we have displayed the spectral overlap between the UV/vis absorption spectrum of CTAB and the fluorescence emission spectrum of the QDs-GO nanocomposite in Fig. S5B. The displayed spectral overlap suggests that FRET

must have led to the fluorescence quenching process. However, from Fig. 5b, the absorption spectrum of the QD-GO-CTAB-AuNPs was quenched relative to the unbonded CTAB-AuNPs, suggesting that FRET was not the mechanism for the quenching process between the QDs-GO nanocomposite and CTAB-AuNPs. Therefore, we believe that the direct suppression of the radiative decay rate of the QDs-GO emission by CTAB-AuNPs must have led to the quenching process [36]. To aid binding of the biotinylated DNA aptamer, streptavidin was first adsorbed on the QDs-GO-CTAB-AuNP nano hybrid resulting in a streptavidin-biotin DNA-aptamer interaction on the surface of the nano hybrid. The incorporation of cocaine into the strep-B-QDs-GO-CTAB-AuNP system, induced marked fluorescence enhancement process in a concentration dependent manner. The MNS-4.1 anticocaine aptamer is known to have a strong affinity for cocaine. A target-induced conformation change in the form of a noncanonical three-way junction is formed upon binding of the unfolded aptamer with cocaine (Fig. S6) [1]. This binding interaction then induced fluorescence transduction changes in the nanoprobe, resulting in a fluorescence switched on process for cocaine determination.

Optimization of method

The following parameters were optimized: (a) Sample pH, (b) nanoprobe efficiency and (c) LSPR effect from different plasmonic NPs. Respective text and figures on optimization (Fig. S7A-C) are given in the [Electronic Supporting Material](#). In general, the following experimental conditions were found to give best results: (a) Optimal pH value = pH 2.2, (b) optimal probe efficiency = strep-B-QDs-GO-CTAB-AuNP and (c) optimal LSPR effect = CTAB-AuNPs.

Selectivity of the nanoprobe

The selectivity of the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe for cocaine determination was studied in the presence of other drugs and substances and cocaine metabolites. Figure 5 shows the selectivity data for cocaine determination using the aptamer-based fluorescence nanoprobe. From the data, a significant and superior fluorescence intensity signal was obtained for cocaine (100 μ M) in comparison to the other tested drugs and substances (100 μ M). Particularly, the nanoprobe was insensitive to the tested cocaine metabolites (benzoylecgonine and ecgonine methyl ester (100 μ M)), thus affirming the unique selectivity of the nanoprobe to cocaine alone. Based on the obtained results, we can say with confidence that the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe can successfully be used for the selective determination of cocaine and confirm its presence in a sample.

Quantitative cocaine determination

Quantitative determination of cocaine was carried in the concentration range of 0.1–100 μ M by the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe. Figure 6a shows the fluorescence emission spectral changes for each concentration of cocaine detected. From the data, the fluorescence of the aptamer-based strep-B-QDs-GO-CTAB-AuNP nanoprobe was switched on in a concentration-dependent manner. A slight red shift and broadening of the fluorescence emission spectrum was observed at low concentrations of cocaine. This can be attributed to the aptamer folding interaction with low concentration of cocaine and its subsequent effect on the optical property of the fluorophore nanoprobe. The corresponding fluorescence linear plot, shown in Fig. 6b was used to determine the limit of detection

Fig. 5 Selectivity of the strep-B-QD-GO-CTAB-AuNP aptamer nanoprobe to cocaine in comparison to other drugs and cocaine metabolites. Concentration of cocaine and other tested drugs and substances = 100 μ M. Control = strep-B-QDs-GO-CTAB-AuNP solution without cocaine. Error bars are standard deviation of 3 replicate analysis

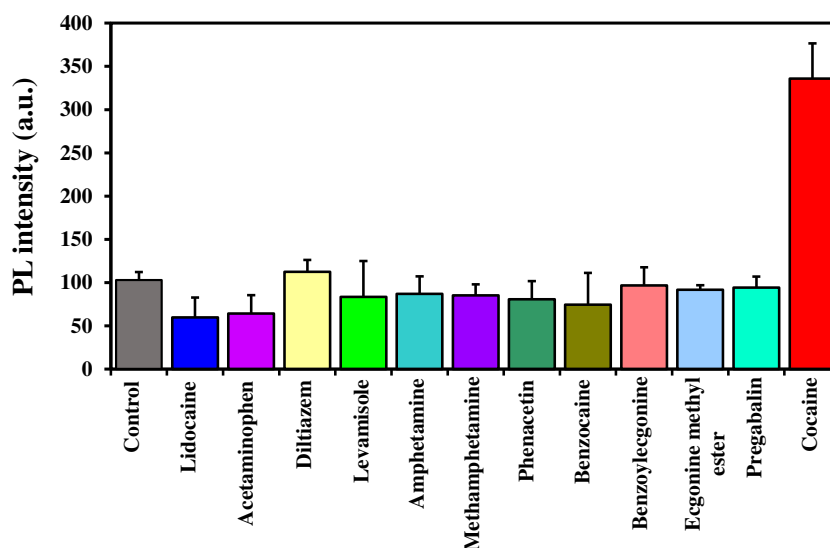
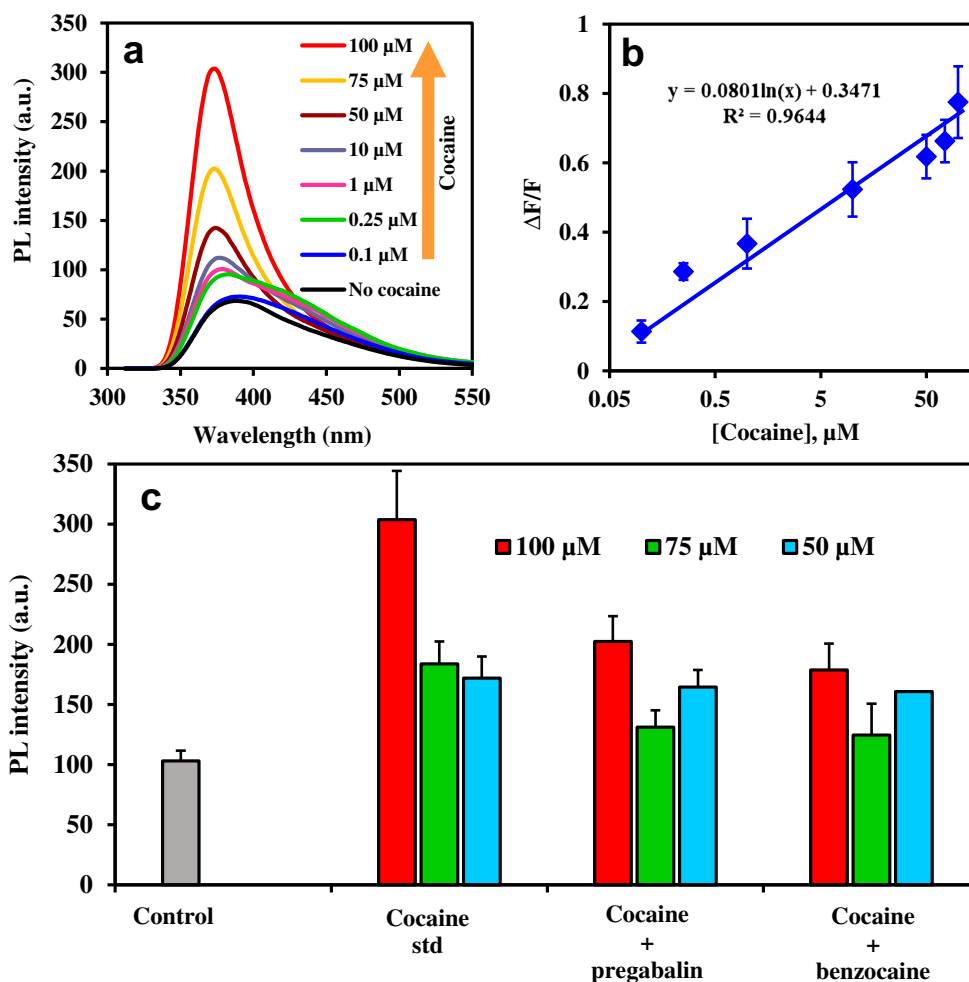


Fig. 6 **a** Fluorescence “turn ON” spectra of the strep-B-QDs-GO-CTAB-AuNP aptamer nanoprobe to cocaine quantitative determination, **b** Photoluminescence (PL) intensity calibration plot for cocaine quantitative determination, **c** PL intensity signal for the determination of 100, 75 and 50 μM pure cocaine standard in comparison to adulterated cocaine samples containing benzocaine and pregabalin. Error bars are standard deviation of 3 replicate analysis. $\lambda_{\text{exc}} = 290 \text{ nm}$. Maximum PL intensity signal taken at $\sim 370 \text{ nm}$



(LOD) of the nanoprobe by multiplying the standard deviation of blank measurement ($n = 9$) by 3 and dividing by the slope of the linear plot. The LOD obtained for cocaine determination using the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe was 4.6 nM ($1.56 \text{ ng}\cdot\text{mL}^{-1}$) while the

linear range was from $0.1\text{--}100 \mu\text{M}$. Comparison of the analytical performance of the developed nanoprobe with other published probes for cocaine (Table 1) shows that our detection system exhibited a more favourable balance between sensitivity and response time than most probes.

Table 1 Comparison of the analytical performance of the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe with other published aptamer-based fluorescence nanoprobe for cocaine

Strategy	Method	LOD	Detection time	Ref.
FRET aptamer assay	Fluorescence turn-OFF	$10 \mu\text{M}$	–	[37]
Structure-switching aptamers-SYBR Gold-exonuclease I	Fluorescence turn-ON	$5 \mu\text{M}$	25 min	[7]
2-amino-5,6,7-trimethyl-1,8-naphthyridine aptamer assay	Fluorescence turn-ON	200 nM	20 s	[1]
QDs-FRET aptamer assay	Fluorescence turn-OFF and turn-ON	$0.5 \mu\text{M}$	1 min 40 s	[8]
Strand displacement amplification aptamer assay	Fluorescence turn-ON	1 nM	2 h	[38]
G-Quadruplex-Iridium (III) complex aptamer assay	Fluorescence turn-ON	30 nM	30 min	[4]
2-Amino-5,6,7-trimethyl-1,8-naphthyridine-SYBR Green I aptamer assay	Fluorescence turn-ON	56 nM	20 s	[39]
GO and exonuclease III-assisted signal amplification	Fluorescence turn-ON	0.1 nM	17 min	[9]
Strep-B-QDs-GO-CTAB-AuNP aptamer assay	Fluorescence turn-ON	4.6 nM	$\sim 2 \text{ min}$	This work

Determination of cocaine in seized adulterated samples

The applicability of the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe to detect seized adulterated cocaine samples was investigated. Seized cocaine samples containing benzocaine and pregabalin adulterants were detected at 100, 75 and 50 μM and the obtained fluorescent signals were compared to the signal obtained for a pure cocaine standard sample. Figure 6c shows that the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe was efficient in detecting adulterated cocaine within the sample. As expected, the fluorescent intensity was lower in comparison to the pure cocaine sample due to the adulterated nature of the drug sample. Thus, it is imperative to conclude that the constructed strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe is suitable for the real-life determination of both pure and adulterated cocaine samples.

An attempt to detect cocaine using the strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe in spiked urine sample was unsuccessful. Despite this limitation, most illicit cocaine samples are illegally produced and transported in white or off-white crystalline powder and their physical appearance is only slightly changed in adulterated samples which themselves are known to appear in fine dry white powders. As we have demonstrated in our work, our nanoprobe is sensitive and selective enough to detect cocaine and its adulterated sample in their crystalline powdered form.

Conclusions

A novel strep-B-QDs-GO-CTAB-AuNP aptamer-based fluorescence nanoprobe platform has been developed for cocaine. Under optimum experimental conditions, the constructed QDs-GO-CTAB-AuNP nanohybrid was found to exhibit far superior sensitivity to cocaine than the tested strep-B-QDs (no GO and CTAB-AuNPs), strep-B-QDs-CTAB-AuNP (no GO) and strep-B-QDs-GO (no CTAB-AuNP). Also, LSPR signal from CTAB-AuNPs was found to amplify the fluorescence intensity in superior fashion than other tested plasmonic NPs. In general, the new aptamer-based fluorescent nanoprobe platform exhibited low detection limit, excellent selectivity and rapid response time. The application of the nanoprobe to detect seized adulterated cocaine was successfully achieved.

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References

- Roncancio D, Yu H, Xu X, Wu S, Liu R, Debord J, Lou X, Xiao Y (2014) A label-free aptamer-fluorophore assembly for rapid and specific detection of cocaine in biofluids. *Anal Chem* 86:11100–11106
- Barroso M, Dias M, Vieira DN, Queiroz JA, Lopez-Rivadulla M (2008) Development and validation of an analytical method for the simultaneous determination of cocaine and its main metabolite, benzoylecgonine, in human hair by gas chromatography/mass spectrometry. *Rapid Commun Mass Spectrom* 22:3320–3326
- Bozokalfa G, Akbulut H, Demir B, Guler E, Gumus ZP, Demirkol AO, Aldemir E, Yamada S, Endo T, Coskunol H, Timur S, Yagci Y (2016) Polypeptide functional surface for the aptamer immobilization: electrochemical cocaine biosensing. *Anal Chem* 88:4161–4167
- Ma D-L, Wang M, He B, Yang C, Wang W, Leung C-H (2015) A luminescent cocaine detection platform using a split g-quadruplex-selective iridium (III) complex and a three-way DNA junction architecture. *ACS Appl Mater Interfaces* 7:19060–19067
- Jayasena SD (1999) An emerging class of molecules that rival antibodies in diagnostics. *Clin Chem* 45:1628–1650
- Ellington AD, Szostak JW (1990) In vitro selection of RNA molecules that bind specific ligands. *Nature* 346:818–822
- Zheng D, Zou R, Lou X (2012) Label-free fluorescent detection of ions, proteins, and small molecules using structure-switching aptamers, SYBR gold, and exonuclease I. *Anal Chem* 84:3554–3560
- Zhang C-Y, Johnson L (2009) Single quantum-dot-based aptameric nanosensor for cocaine. *Anal Chem* 81:3051–3055
- Zhang Y, Sun Z, Tang L, Zhang H, Zhang GJ (2016) Aptamer based fluorescent cocaine assay based on the use of graphene oxide and exonuclease III-assisted signal amplification. *Microchim Acta* 183: 2791–2797
- Zhang S, Geryak R, Geldmeier J, Kim S, Tsukruk VV (2017) Synthesis, assembly, and applications of hybrid nanostructures for biosensing. *Chem Rev* 117:12942–13038
- Geim AK, Novoselov KS (2007) The rise of graphene. *Nat Mater* 6: 183–191
- Kuila T, Bose S, Khanra P, Mishra AK, Kim NH, Lee JH (2011) Recent advances in graphene-based biosensors. *Biosens Bioelectron* 26:4637–4648
- Oliveira E, Núñez Santos HM, Fernández-Lodeiro J, Fernández-Lodeiro A, Capelo JL, Lodeiro C (2015) Graphene and graphene oxide: biofunctionalization and applications in biotechnology. *Sens Actuatur B-Chem* 212:297–328
- Chen MS, Goodman DW (2004) The structure of catalytically active gold on titania. *Science* 306:252–255

15. Adegoke O, Morita M, Kato T, Ito M, Suzuki T, Park EY (2017) Localized surface plasmon resonance-mediated fluorescence signals in plasmonic nanoparticle-quantum dot hybrids for ultrasensitive Zika virus RNA detection via hairpin hybridization assays. *Biosens Bioelectron* 94:513–522
16. Adegoke O, McKenzie C, Nic Daeid N (2019) Multi-shaped cationic gold nanoparticle-L-cysteine-ZnSeS quantum dots hybrid nanozyme as an intrinsic peroxidase mimic for the rapid colorimetric detection of cocaine. *Sensor Actuat B-Chem* 287:416–427
17. Adegoke O, Forbes PBC (2016) L-cysteine-capped core/shell/shell quantum dot-graphene oxide nanocomposite fluorescence probe for polycyclic aromatic hydrocarbon detection. *Talanta* 146:780–788
18. Hummers WS Jr, Offeman RE (1958) Preparation of graphitic oxide. *J Am Chem Soc* 80:1339–1339
19. Kundu S, Peng L, Liang H (2008) A new route to obtain high-yield multiple-shaped gold nanoparticles in aqueous solution using microwave irradiation. *Inorg Chem* 47:6344–6352
20. Dreyer DR, Park S, Bielawski C, Ruoff RS (2010) The chemistry of graphene oxide. *Chem Soc Rev* 39:228–240
21. Kim F, Cote LJ, Huang J (2010) Graphene oxide: surface activity and two-dimensional assembly. *Adv Mater* 22:1954–1958
22. Danek M, Jensen KF, Murray CB, Bawendi MG (1996) Synthesis of luminescent thin-film CdSe/ZnSe quantum dot composites using CdSe quantum dots passivated with an overlayer of ZnSe. *Chem Mater* 8:173–180
23. Chaudhuri RG, Paria S (2012) Core/shell nanoparticles: classes, properties, synthesis mechanisms, characterization, and applications. *Chem Rev* 112:2373–2433
24. Chua CK, Pumera M (2014) Chemical reduction of graphene oxide: a synthetic chemistry viewpoint. *Chem Soc Rev* 43:291–312
25. Eda G, Chhowalla M (2010) Chemically derived graphene oxide: towards large-area thin film electronics and optoelectronics. *Adv Mater* 22:2392–2415
26. Priece P, Salami HA, Padilla RH, Zhong Z, Lopez-Sanchez JA (2016) Anisotropic gold nanoparticles: preparation and applications in catalysis. *Chinese J Catal* 37:1619–1650
27. Liu L, An M, Yang P, Zhang J (2015) Superior cycle performance and high reversible capacity of SnO₂/graphene composite as an anode material for lithium-ion batteries. *J Sci Rep-UK* 5:9055
28. Ferrari AC, Roberson J (2000) Interpretation of Raman spectra of disordered and amorphous carbon. *J Phys Rev B* 61:14095–14107
29. Liu M, Zhao H, Quan X, Chen S, Fan X (2010) Distance-independent quenching of quantum dots by nanoscale-graphene in self-assembled sandwich immunoassay. *Chem Commun* 42:7909–7911
30. Chen Z, Berciaud S, Nuckolls C, Heinz TF, Brus LE (2010) Energy transfer from individual semiconductor nanocrystals to graphene. *ACS Nano* 5:2964–2968
31. Marales-Narváez E, Pérez-López B, Pires LB, Merkoçi A (2012) Simple Förster resonance energy transfer evidence for the ultrahigh quantum dot quenching efficiency by graphene oxide compared to other carbon structures. *Carbon* 50:2987–2993
32. Iliut M, Gabudean A-M, Leordean C, Simon T, Teodorescu C-M, Astilean S (2013) Riboflavin enhanced fluorescence of highly reduced graphene oxide. *Chem Phys Lett* 586:127–131
33. Li M, Zhou X, Guo S, Wu N (2013) Detection of lead (II) with a “turn-on” fluorescent biosensor based on energy transfer from CdSe/ZnS quantum dots to graphene oxide. *Biosens Bioelectron* 43:69–74
34. Wang Q, Geng B, Wang S (2009) ZnO/au hybrid nanoarchitectures: wet-chemical synthesis and structurally enhanced photocatalytic performance environ. *Sci Technol* 43:8968–8973
35. Gomez CV, Tene T, Guevara M, Usca GT, Colcha D, Brito H, Molina R, Bellucci S, Tavoraro A (2019) Preparation of few-layer graphene dispersions from hydrothermally expanded graphite. *Appl Sci* 9:2539
36. Samanta A, Zhou Y, Zou S, Yan H, Liu Y (2014) Fluorescence quenching of quantum dots by gold nanoparticles: a potential long range spectroscopic ruler. *Nano Lett* 14:5052–5057
37. Stojanovic MN, de Prada P, Landry DW (2001) Aptamer-based folding fluorescent sensor for cocaine. *J Am Chem Soc* 123:4928–4931
38. He J-L, Wu Z-S, Zhou H, Wang H-Q, Jiang J-H, Shen G-L, Yu R-Q (2010) Fluorescence aptameric sensor for strand displacement amplification detection of cocaine. *Anal Chem* 82:1358–1364
39. Wang J, Song J, Wang X, Wu S, Zhao Y, Luo P, Meng C (2016) An ATMND/SGI based label-free and fluorescence ratiometric aptasensor for rapid and highly sensitive detection of cocaine in biofluids. *Talanta* 161:437–442

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