



Synthesis of six-membered silacycles by borane-catalyzed double sila-Friedel–Crafts reaction

Yafang Dong¹, Masahiko Sakai¹, Kazuto Fujii¹, Kohei Sekine^{1,2} and Yoichiro Kuninobu^{*1,2}

Letter

Open Access

Address:

¹Interdisciplinary Graduate School of Engineering Sciences, Kyushu University, 6-1 Kasugakoen, Kasuga-shi, Fukuoka 816-8580, Japan and ²Institute for Materials Chemistry and Engineering, Kyushu University, 6-1 Kasugakoen, Kasuga-shi, Fukuoka 816-8580, Japan

Email:

Yoichiro Kuninobu* - kuninobu@cm.kyushu-u.ac.jp

* Corresponding author

Keywords:

borane; cyclic compound; organosilane; sila-Friedel–Crafts; silylation

Beilstein J. Org. Chem. **2020**, *16*, 409–414.

doi:10.3762/bjoc.16.39

Received: 01 February 2020

Accepted: 11 March 2020

Published: 17 March 2020

This article is part of the thematic issue "C–H functionalization for materials science".

Guest Editor: K. Itami

© 2020 Dong et al.; licensee Beilstein-Institut.

License and terms: see end of document.

Abstract

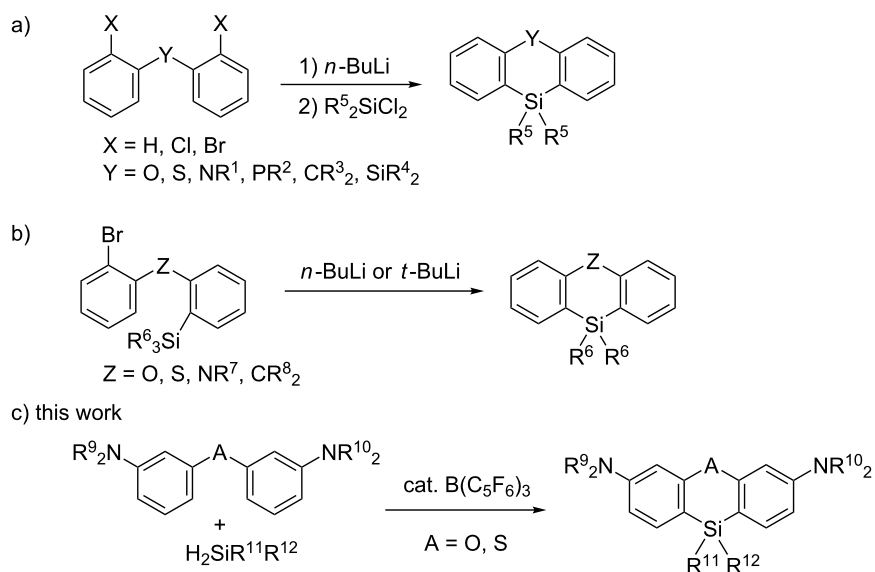
We have developed a catalytic synthetic method to prepare phenoxasilins. A borane-catalyzed double sila-Friedel–Crafts reaction between amino group-containing diaryl ethers and dihydrosilanes can be used to prepare a variety of phenoxasilin derivatives in good to excellent yields. The optimized reaction conditions were also applicable for diaryl thioethers to afford their corresponding six-membered silacyclic products. The gram-scale synthesis of a representative bis(dimethylamino)phenoxasilin and the transformation of its amino groups have also been demonstrated.

Introduction

Six-membered silacyclic compounds, such as phenoxasilin and phenothiasilin derivatives, are attractive compounds for applications as organic electronic materials [1–4], ligands [5–10], and reagents [11–14]. Therefore, the development of new methods to construct silacyclic skeletons is highly desirable. These compounds are commonly synthesized upon the reaction of heteroatom-bridged dilithiated diaryl compounds, such as dilithiated diaryl ethers and dilithiated diaryl thioethers with a range of dichlorosilane derivatives (Scheme 1a) [15–24]. An intramolecular silylation via Si–C bond cleavage can also be used to prepare a variety of six-membered silacyclic derivatives (Scheme 1b) [25]. However, some problems still remain in terms of the functional group tolerance and versatility of these

previously reported synthetic methods due to the use of a stoichiometric amount of the organolithium reagents. In addition, despite these contributions, catalytic reaction systems have not been developed as much [26,27].

The sila-Friedel–Crafts reaction is emerging as a powerful tool for C–H silylation [28,29]. In addition, intra- and intermolecular sila-Friedel–Crafts reactions have been recently developed [30–39], which have great potential as efficient synthetic strategies to construct silacycles. For example, the intramolecular C–H silylation of biphenylhydrosilanes can be used to prepare various silafluorene derivatives [30–34] and the ruthenium-catalyzed intermolecular Friedel–Crafts-type reaction of



Scheme 1: Synthetic methods of six-membered silacyclic compounds.

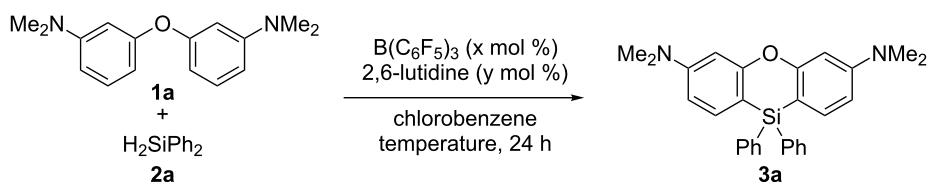
2-phenylindole with a variety of dihydrosilanes affords indole-fused benzosiloles [39]. We have also contributed to the synthesis of silafluorenes from biphenyls and dihydrosilanes using a borane-catalyzed double sila-Friedel–Crafts reaction [40,41]. Subsequently, we envisaged that the catalytic reaction between diaryl ethers and dihydrosilanes may be a useful protocol to prepare phenoxasilin derivatives (Scheme 1c). Herein, we report a borane-catalyzed double sila-Friedel–Crafts reaction used for the synthesis of six-membered silacyclic compounds, such as phenoxasilin and phenothiasilin derivatives.

Results and Discussion

A double sila-Friedel–Crafts reaction was initially investigated using diaryl ether **1a** and dihydrodiphenylsilane (**2a**) as model

substrates (Table 1). Under the optimized reaction conditions used for the synthesis of silafluorenes in our previous report [40] ($B(C_6F_5)_3$ (5.0 mol %) and 2,6-lutidine (7.5 mol %) in chlorobenzene at 100 °C), the desired reaction between **1a** with **2a** proceeded to give phenoxasilin **3a** in 60% yield (Table 1, entry 1). The structure of phenoxasilin **3a** was confirmed using single-crystal X-ray crystallography (see Supporting Information File 1 for details) [42]. Upon increasing the reaction temperature to 140 °C, the yield of **3a** was improved to 88% (Table 1, entry 2). Although the reaction in the presence of 3.0 mol % of the catalyst also proceeded efficiently (Table 1, entry 3, conditions A), the yield of **3a** decreased when compared to that obtained using 1.5 mol % of the catalyst (Table 1, entry 4). The best result was obtained in the absence of 2,6-luti-

Table 1: Optimization of the reaction conditions for the synthesis of phenoxalin **3a**.



entry ^a	x (mol%)	y (mol %)	temp (°C)	yield (%)
1	5.0	7.5	100	60
2	5.0	7.5	140	88
3	3.0	7.5	140	97
4	1.5	7.5	140	87
5	3.0	0	140	99

^a**1a** (0.250 mmol), **2a** (0.750 mmol), chlorobenzene (0.4 mL).

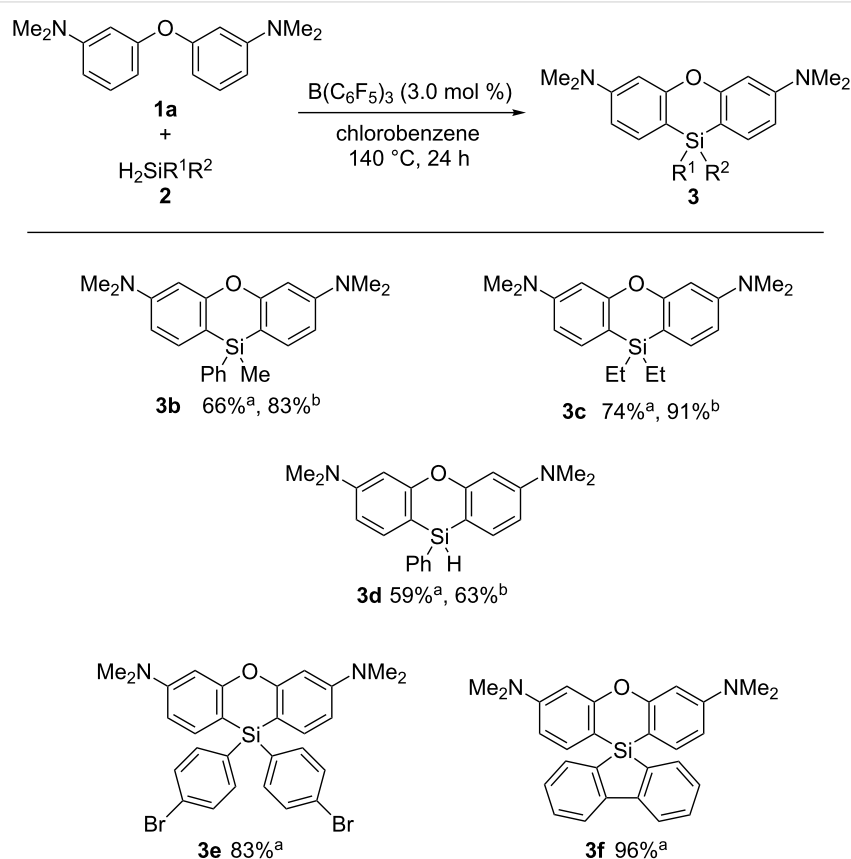
dine by which phenoxasilin **3a** formed in 99% yield (Table 1, entry 5, conditions B).

Next, the scope of the dihydrosilane starting materials used in the reaction was investigated (Scheme 2). The reactions of phenylmethylsilane (**2b**) and diethyldihydrosilane (**2c**) afforded their corresponding phenoxasilin derivatives **3b** and **3c** in 66 and 74% yield, respectively. The yields of **3b** and **3c** were improved to 83 and 91% in the presence of a catalytic amount of 2,6-lutidine, probably due to the acceleration of the deprotonation step by 2,6-lutidine [33]. In the case of phenylsilane (**2d**), the phenoxasilin product **3d** was formed in 59% yield using conditions B and in 63% yield under conditions A. Di(4-bromophenyl)dihydrosilane (**2e**) was transformed successfully into phenoxasilin **3e** in 83% yield without loss of the bromine substituent. The reaction system was also applicable for 9,9-dihydro-5-silafluorene (**2f**), which gave the spiro-type phenoxasilin **3f** in 96% yield.

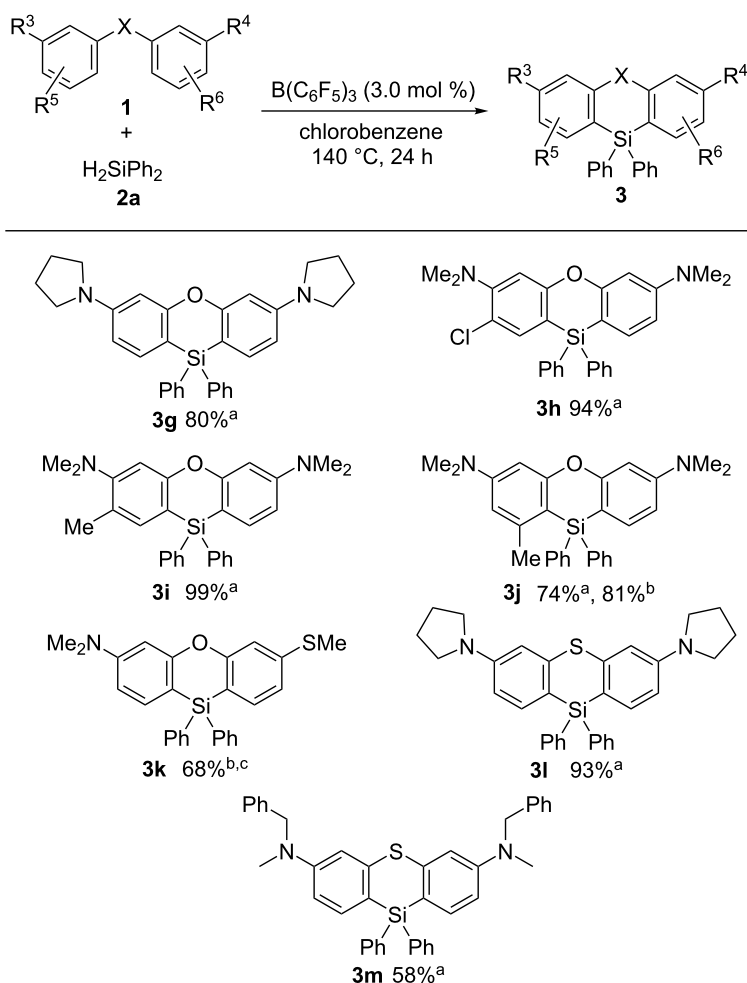
We then investigated the scope of the starting biaryl ethers used in the reaction as well as related derivatives thereof using dihydrodiphenylsilane (**2a**, Scheme 3). Pyrrolidine-substituted diaryl ether **1b** was transformed into phenoxasilin **3g** in 80% yield.

Also, the chloro-substituted diaryl ether gave its corresponding phenoxasilin **3h** in 94% yield without affecting the chlorine substituent. The methyl-substituted phenoxasilin derivatives **3i** and **3j** were formed in good yield despite of the steric hindrance of the methyl group in **3j**. When one of the NMe₂ groups was replaced with a SMe group, a mixture of the corresponding phenoxasilin product (**3k**) and the hydrosilane compound (**3k'**) was obtained via a single sila-Friedel–Crafts reaction in 35% yield in the presence of 2,6-lutidine (**3k:3k'** = 63:37). This result was probably due to the weaker electron-donating ability of the SMe group compared to that of NMe₂. The double C–H silylation reaction proceeds efficiently upon increasing the temperature to 180 °C that afforded the mixture (**3k:3k'** = 92:8) in 68% yield. The reaction system can also be applied to the synthesis of phenothiasilin **3l** that was obtained in 93% yield starting from diaryl thioether **1g**. *N*-(Benzyl)methylamine-substituted diaryl thioether **1h** was also transformed into phenothiasilin **3m** in 58% yield. The corresponding six-membered silacycles were not formed using *N*-aryl-bridged biaryls as substrates.

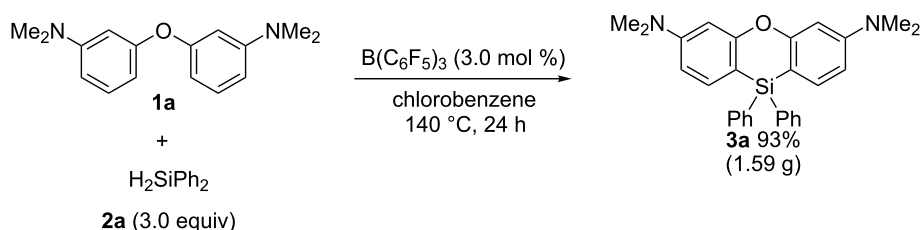
To test the applicability of the method, a gram-scale synthesis of phenoxasilin **3a** was carried out (Scheme 4). The reaction of



Scheme 2: Scope of dihydrosilanes. Conditions: a: conditions B (Table 1, entry 5); b: conditions A (Table 1, entry 3).



Scheme 3: Scope of diaryl ether and diaryl thioether derivatives. Conditions: a: conditions B (Table 1, entry 5); b: conditions A (Table 1, entry 3). c: temperature 180°C .



Scheme 4: Gram-scale Synthesis of **3a**.

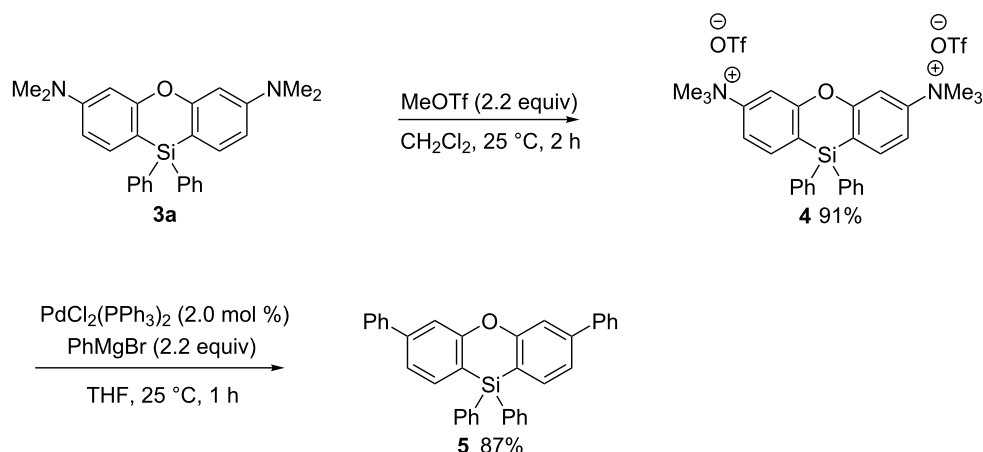
diaryl ether **1a** (1.00 g) with dihydrosilane (**2a**, 2.16 g) in the presence of a catalytic amount of $\text{B}(\text{C}_6\text{F}_5)_3$ afforded phenoxasilin **3a** in 93% yield (1.59 g).

Finally, the transformation of the amino groups in phenoxasilin **3a** into phenyl groups was carried out (Scheme 5). First, the ammonium salt **4** was prepared by treating **3a** with MeOTf fol-

lowed by a palladium-catalyzed cross-coupling reaction with the Grignard reagent (PhMgBr) that afforded the desired diphenylated phenoxasilin **5** in 87% yield [43].

Conclusion

In summary, we have developed a new catalytic synthetic method to prepare six-membered silacyclic compounds, such as

Scheme 5: Transformation of the amino groups in **3a**.

phenoxasilin and phenothiasilin derivatives, using a double silyl-Friedel–Crafts reaction. The reaction system is applicable to diaryl ethers with halogen substituents or sterical hindrance. A gram-scale synthesis of phenoxasilins and transformation of the amino groups in the phenoxasilin product were also achieved. We hope that the developed protocol will prove to be a useful and efficient method to synthesize six-membered silacyclic compounds.

Supporting Information

Supporting Information File 1

Experimental procedures, compounds characterization data, and copies of ^1H and ^{13}C NMR spectra.

[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-39-S1.pdf>]

Supporting Information File 2

CIF file for **3a**.

[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-39-S2.cif>]

Funding

This work was supported in part by JSPS KAKENHI Grant Numbers JP 17H03016 and 18H04656, The Sumitomo Foundation, and A-STEP (VP30218088652) from JST. Y.D. is grateful to the CSC (China Scholarship Council) for the PhD fellowship.

ORCID® iDs

Kohei Sekine - <https://orcid.org/0000-0001-7588-3176>

Yoichiro Kuninobu - <https://orcid.org/0000-0002-8679-9487>

References

- Li, J.; Ding, D.; Wei, Y.; Zhang, J.; Xu, H. *Adv. Opt. Mater.* **2016**, *4*, 522–528. doi:10.1002/adom.201500673
- Hayashi, H.; Nakao, H.; Miyabayashi, T.; Murase, M. *Jpn. J. Appl. Phys.* **2013**, *52*, 05DA13. doi:10.7567/jjap.52.05da13
- Sun, J. W.; Baek, J. Y.; Kim, K.-H.; Moon, C.-K.; Lee, J.-H.; Kwon, S.-K.; Kim, Y.-H.; Kim, J.-J. *Chem. Mater.* **2015**, *27*, 6675–6681. doi:10.1021/acs.chemmater.5b02515
- Matsuo, K.; Yasuda, T. *Chem. Sci.* **2019**, *10*, 10687–10697. doi:10.1039/c9sc04492b
- Kranenburg, M.; van der Burgt, Y. E. M.; Kamer, P. C. J.; van Leeuwen, P. W. N. M.; Goubitz, K.; Fraanje, J. *Organometallics* **1995**, *14*, 3081–3089. doi:10.1021/om00006a057
- Kranenburg, M.; Kamer, P. C. J.; van Leeuwen, P. W. N. M. *Eur. J. Inorg. Chem.* **1998**, 25–27. doi:10.1002/(sici)1099-0682(199801)1998:1<25::aid-ejic25>3.0.co;2-k
- van der Veen, L. A.; Keeven, P. H.; Schoemaker, G. C.; Reek, J. N. H.; Kamer, P. C. J.; van Leeuwen, P. W. N. M.; Lutz, M.; Spek, A. L. *Organometallics* **2000**, *19*, 872–883. doi:10.1021/om990734o
- Bronger, R. P. J.; Kamer, P. C. J.; van Leeuwen, P. W. N. M. *Organometallics* **2003**, *22*, 5358–5369. doi:10.1021/om034012f
- Clayden, J.; Fletcher, S. P.; Senior, J.; Worrall, C. P. *Tetrahedron: Asymmetry* **2010**, *21*, 1355–1360. doi:10.1016/j.tetasy.2010.06.017
- Rajesh, K.; Dudle, B.; Blacque, O.; Berke, H. *Adv. Synth. Catal.* **2011**, *353*, 1479–1484. doi:10.1002/adsc.201000867
- Corey, J. Y.; Corey, E. R.; Chang, V. H. T.; Hauser, M. A.; Leiber, M. A.; Reinsel, T. E.; Riva, M. E. *Organometallics* **1984**, *3*, 1051–1060. doi:10.1021/om00085a015
- Betson, M. S.; Clayden, J.; Worrall, C. P.; Peace, S. *Angew. Chem., Int. Ed.* **2006**, *45*, 5803–5807. doi:10.1002/anie.200601866
- Betson, M. S.; Clayden, J. *Synlett* **2006**, 745–746. doi:10.1055/s-2006-933111
- Braddock-Wilking, J.; Corey, J. Y.; French, L. M.; Choi, E.; Speedie, V. J.; Rutherford, M. F.; Yao, S.; Xu, H.; Rath, N. P. *Organometallics* **2006**, *25*, 3974–3988. doi:10.1021/om060391b
- Oita, K.; Gilman, H. J. *Am. Chem. Soc.* **1957**, *79*, 339–342. doi:10.1021/ja01559a026

16. Hitchcock, C. H. S.; Mann, F. G.; Vanterpool, A. *J. Chem. Soc.* **1957**, 4537–4546. doi:10.1039/jr9570004537
17. Gilman, H.; Miles, D. *J. Org. Chem.* **1958**, *23*, 1363–1365. doi:10.1021/jo01103a036
18. Gilman, H.; Trepka, W. *J. Org. Chem.* **1961**, *26*, 5202–5203. doi:10.1021/jo01070a512
19. Gilman, H.; Trepka, W. *J. Org. Chem.* **1962**, *27*, 1418–1422. doi:10.1021/jo01051a071
20. Belsky, V. K.; Saratov, I. E.; Reikhsfeld, V. O.; Simonenko, A. A. *J. Organomet. Chem.* **1983**, *258*, 283–289. doi:10.1016/s0022-328x(00)99273-8
21. Corey, J. Y.; Trankler, K. A.; Braddock-Wilking, J.; Rath, N. P. *Organometallics* **2010**, *29*, 5708–5713. doi:10.1021/om100544f
22. Wittenberg, D.; McNinch, H. A.; Gilman, H. *J. Am. Chem. Soc.* **1958**, *80*, 5418–5422. doi:10.1021/ja01553a025
23. McCarthy, W. Z.; Corey, J. Y.; Corey, E. R. *Organometallics* **1984**, *3*, 255–263. doi:10.1021/om00080a016
24. van der Boon, L. J. P.; Hendriks, J. H.; Roolvink, D.; O'Kennedy, S. J.; Lutz, M.; Slootweg, J. C.; Ehlers, A. W.; Lammertsma, K. *Eur. J. Inorg. Chem.* **2019**, 3318–3328. doi:10.1002/ejic.201900641
25. Onoe, M.; Morioka, T.; Tobisu, M.; Chatani, N. *Chem. Lett.* **2013**, *42*, 238–240. doi:10.1246/cl.2013.238
26. Li, H.; Wang, Y.; Yuan, K.; Tao, Y.; Chen, R.; Zheng, C.; Zhou, X.; Li, J.; Huang, W. *Chem. Commun.* **2014**, *50*, 15760–15763. doi:10.1039/c4cc06636g
27. Sato, Y.; Takagi, C.; Shintani, R.; Nozaki, K. *Angew. Chem., Int. Ed.* **2017**, *56*, 9211–9216. doi:10.1002/anie.201705500
28. Bhr, S.; Oestreich, M. *Angew. Chem., Int. Ed.* **2017**, *56*, 52–59. doi:10.1002/anie.201608470
29. Richter, S. C.; Oestreich, M. *Trends Chem.* **2020**, *2*, 13–27. doi:10.1016/j.trechm.2019.07.003
30. Furukawa, S.; Kobayashi, J.; Kawashima, T. *J. Am. Chem. Soc.* **2009**, *131*, 14192–14193. doi:10.1021/ja906566r
31. Furukawa, S.; Kobayashi, J.; Kawashima, T. *Dalton Trans.* **2010**, *39*, 9329–9336. doi:10.1039/c0dt00136h
32. Arai, H.; Nakabayashi, K.; Mochida, K.; Kawashima, T. *Molecules* **2016**, *21*, 999. doi:10.3390/molecules21080999
33. Curless, L. D.; Ingleson, M. J. *Organometallics* **2014**, *33*, 7241–7246. doi:10.1021/om501033p
34. Omann, L.; Oestreich, M. *Angew. Chem., Int. Ed.* **2015**, *54*, 10276–10279. doi:10.1002/anie.201504066
35. Chen, Q.-A.; Klare, H. F. T.; Oestreich, M. *J. Am. Chem. Soc.* **2016**, *138*, 7868–7871. doi:10.1021/jacs.6b04878
36. Yin, Q.; Klare, H. F. T.; Oestreich, M. *Angew. Chem., Int. Ed.* **2016**, *55*, 3204–3207. doi:10.1002/anie.201510469
37. Ma, Y.; Wang, B.; Zhang, L.; Hou, Z. *J. Am. Chem. Soc.* **2016**, *138*, 3663–3666. doi:10.1021/jacs.6b01349
38. Han, Y.; Zhang, S.; He, J.; Zhang, Y. *J. Am. Chem. Soc.* **2017**, *139*, 7399–7407. doi:10.1021/jacs.7b03534
39. Omann, L.; Oestreich, M. *Organometallics* **2017**, *36*, 767–776. doi:10.1021/acs.organomet.6b00801
40. Dong, Y.; Takata, Y.; Yoshigoe, Y.; Sekine, K.; Kuninobu, Y. *Chem. Commun.* **2019**, *55*, 13303–13306. doi:10.1039/c9cc07692a
41. Ureshino, T.; Yoshida, T.; Kuninobu, Y.; Takai, K. *J. Am. Chem. Soc.* **2010**, *132*, 14324–14326. doi:10.1021/ja107698p
See for Rhodium-catalyzed intramolecular C–H silylation for the synthesis of silafluorenes.
42. CCDC 1979913 (**3a**) contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <https://www.ccdc.cam.ac.uk/structures/>.
43. Reeves, J. T.; Fandrick, D. R.; Tan, Z.; Song, J. J.; Lee, H.; Yee, N. K.; Senanayake, C. H. *Org. Lett.* **2010**, *12*, 4388–4391. doi:10.1021/ol1018739

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (<http://creativecommons.org/licenses/by/4.0>). Please note that the reuse, redistribution and reproduction in particular requires that the authors and source are credited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (<https://www.beilstein-journals.org/bjoc>)

The definitive version of this article is the electronic one which can be found at: [doi:10.3762/bjoc.16.39](https://doi.org/10.3762/bjoc.16.39)