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Catalyst- and Additive-Free C(sp³)—H Functionalization of (Thio)barbituric Acids *via* C-5 Dehydrogenative Aza-Coupling Under Ambient Conditions

Goutam Brahmachari,* Anindita Bhowmick, and Indrajit Karmakar



tolerance for various functional groups, no need for chromatographic purification, good to excellent yields of products within reasonable reaction times in minutes, clean reaction profile, and gram-scale synthetic applicability make this procedure attractive, green, and cost-effective.

1. INTRODUCTION

Exploration of carbon-heteroatom bond-forming strategies always lies at the forefront of synthetic organic chemistry because this drives open new avenues for synthesizing functional organic molecules.¹ Amid various techniques for generating carbon-heteroatom bonds, C-H bond functionalization has recently emerged as promising in modern organic synthesis due to its inherent multifaceted advantages.² Still, many of these reported methodologies are associated with noticeable shortcomings simultaneously. On many occasions, they could not become devoid of the use of oxidants, costly and sensitive metal catalysts and additives, harmful organic solvents, tedious purification steps, and harsh reaction conditions.³ Eventually, green chemistry-directed C-H functionalization for useful chemical transformations is of current choice among synthetic chemists. Among the C-H bond functionalizations, although $C(sp^3)$ -H has recently been reported to emerge,⁴ is still quite a challenging task, mainly due to the lack of activation and selectivity.⁵

As part of our ongoing endeavors in green chemistrydirected C–H functionalizations of useful organic scaffolds,⁶ we thus felt pertinent in designing an efficient and practical $C(sp^3)$ –H functionalization protocol to synthesize a series of diversely substituted barbituric/2-thiobarbituric hydrazones *via* C-5 dehydrogenative aza-coupling.⁷ Substituted pyrimidines, categorically barbituric/2-thiobarbituric acids, act as building blocks for a handful of useful drugs with hypnotic, sedative, anticonvulsant, anaesthetic, antioxidant, antifungal, and central nervous system depressant activity.⁸ The recent literature thus shows a considerable number of synthetic endeavors with barbituric/2-thiobarbituric acids yielding diverse series of functionalized barbituric/2-thiobarbituric acid derivatives with promising medicinal, pharmaceutical, and material properties.⁹ Similarly, hydrazone derivatives also demonstrated their useful applicabilities as medicinal and pharmaceutical agents,¹⁰ chemosensors,¹¹ linkers in preparing bifunctional molecules,¹² and ligands, directing groups, or auxiliaries in organic synthesis.¹³ The new series of these molecules is thus anticipated to exhibit somewhat enhanced bioactivity and material properties.

We have recently developed a straightforward and practical synthetic method for a new series of diversely functionalized 5-(2-arylhydrazono)pyrimidine-2,4,6(1H,3H,5H)-triones (4)/5-(2-arylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-diones (4') from a one-pot three-component reaction between

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Scheme 1. Catalyst-Free C(sp³)-H Functionalization of Barbituric/2-Thiobarbituric Acids via C-5 Dehydrogenative Aza-Coupling under Ambient Conditions



Table 1. Optimization of Reaction Conditions⁴

2,5-di-CIC₆H₃, 3,5-di-CF₃C₆H₃, 4'-CIC₆H₄-4-O-C₆H₄



^aReaction conditions: a mixture of barbituric acid (1a; 0.2 mmol) and aniline (2a; 0.2 mmol) was reacted either with TBN (in different proportions starting from 0.5 to 1.5 equiv) or sodium nitrite (1.0 equiv) in the absence or presence of various solvents (2 mL) without the aid of any catalysts and additives upon stirring at room temperature (28–30 °C); rt = room temperature. ^bIsolated yields.

barbituric acids (1)/2-thiobarbituric acids (1'), primary aromatic amines (2), and *tert*-butyl nitrite (TBN) (3) in an acetonitrile solvent under ambient conditions without the aid of any catalysts/additives (Scheme 1). The transformation involves $C(sp^3)$ —H functionalization of barbituric/2-thiobarbituric acids to yield the hydrazone molecular hybrids *via* C-5 dehydrogenative aza-coupling. The present method is associated with several benefits such as operational simplicity, catalyst-free ambient reaction conditions, a broad substrate scope, avoidance of chromatographic purification, good to excellent yields, short reaction times, and gram-scale synthetic applicability. Recently, the development of catalyst-free and room-temperature-based synthetic methodologies is demanding due to their manifold advantages.¹⁴

2. RESULTS AND DISCUSSION

To arrive at the optimum reaction conditions for this dehydrogenative aza-coupling through $C(sp^3)$ -H functionalization, we performed several trial reactions (Table 1, entries 1–11) with our model entry between unsubstituted barbituric acid (1a; 0.2 mmol), aniline (0.2 mmol), and TBN (varying amounts) targeting the desired product 5-(2-

phenylhydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4a) under ambient environments. As we perceived that the transformation might be implemented self-catalytically, we did not use any catalysts/additives. First, we checked the reaction under neat conditions and were delighted to isolate product 4a in 63% yield (Table 1, entry 1) within 25 min. We then conducted a set of 10 more reactions using different solvent systems such as H₂O, EtOH, EtOH-H₂O (1:1), dimethyl sulfoxide (DMSO), dimethylformamide (DMF), 1,4dioxane, CH₃CN, and CH₃CN-H₂O (1:1) and also by varying the proportion of TBN starting from 0.5 to 1.5 equiv as the nitrating agent (Table 1, entries 2-11). We replaced TBN with sodium nitrite to check its relative nitrating performance (Table 1, entry 12). From these experimental outcomes, the best-suited reaction conditions for the model entry came out to be stirring the mixture of barbituric acid (1a; 0.2 mmol), aniline (0.2 mmol), and TBN (0.2 mmol; 1.0 equiv) dissolved in 2 mL of acetonitrile (CH₃CN) at room temperature (28-30)°C) to obtain 4a in an excellent yield of 93% within 15 min (Table 1, entry 8). No tedious column chromatography was needed to purify compound 4a, which was then characterized by the conventional spectroscopic [¹H NMR, ¹³C NMR,

Table 2. Synthesis of Diversely Functionalized 5-(2-Arylhydrazono)pyrimidine-2,4,6(1H,3H,5H)-triones (4)^{*a*} and 5-(2-Arylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-diones (4')^{*a*}



Table 2. continued



^{*a*}Reaction conditions: a mixture of barbituric acid (1; 0.2 mmol)/2-thiobarbituric acids (1'; 0.2 mmol) and aromatic amines (2; 0.2 mmol) was reacted with TBN (0.2 mmol; 1.0 equiv) dissolved in 2 mL of acetonitrile upon stirring at room temperature (28–30 °C) in the absence of any catalysts and additives. Isolated yields; NR = no reaction.



Figure 1. (a) ORTEP view of molecule 4d, showing the atom-labeling scheme (displacement ellipsoids are drawn at the 50% probability level, and H atoms are shown as small spheres of arbitrary radii) and (b) packing arrangement of 4d viewed along the *a*-axis.

distortionless enhancement by polarization transfer (DEPT-135), and high-resolution mass spectrometry (HRMS)] studies. Table 1 summarizes the overall results.

We then approached to validate the applicability of this newly developed protocol under the optimized reaction conditions. We performed a set of 14 different reactions from the mixture of barbituric acid (1a; 0.2 mmol), TBN (0.2 mmol), and diversely substituted primary aromatic amines (2b-2o; 0.2 mmol each) containing both electron-releasing and electron-withdrawing groups such as methyl, methoxyl, bromo, di-chloro, nitro, bis-trifluoromethyl, trifluoromethoxy, thiomethyl (-SCH₃), thiotrifluoromethyl (-SCF₃), 4-chlorophenoxy, and so forth. All these diverse substituents at the aromatic ring of anilines were found to be well-tolerated while implementing the transformations undergone in a facile manner, resulting in 5-(2-arylhydrazono)pyrimidine-2,4,6-(1H,3H,5H)-triones 4b-4o in good to excellent yields ranging from 69 to 97% within 8-50 min (Table 2, compounds 4b**4o**). Another set of 11 more reactions were designed from the mixture of *N*,*N*-dimethylbarbituric acid (1b), TBN, and varying anilines under the optimized reaction conditions; all the reactions proceeded smoothly, giving rise to the desired products 4p-4z in 62–95% with 15–75 min.

Encouraged by the experimental outcomes, we planned to extend the substrate scope by replacing barbituric acids with 2-thiobarbituric acids. Accordingly, we first checked the model entry by reacting 2-thiobarbituric acid (1'a; 0.2 mmol) with TBN (0.2 mmol) and aniline (2a; 0.2 mmol) in acetonitrile (2 mL) upon stirring at ambient conditions. As anticipated, we obtained the desired product, 5-(2-phenylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'a) in 93% yield at 10 min (Table 2, compound 4'a). We then performed a diverse range of 20 more reactions between thiobarbituric acid (1'a)/N,N-diethyl-2-thiobarbituric acid (1'b), TBN, and varying anilines under the optimized reaction conditions. To our delight, all these reactions proceeded smoothly, thereby

Scheme 2. Proposed Mechanism for the Catalyst-Free $C(sp^3)$ -H Functionalization of Barbituric/2-Thiobarbituric Acids *via* C-5 Dehydrogenative Aza-Coupling under Ambient Conditions



Scheme 3. Representative Gram-Scale Experiments for Compounds 4a and 4'a



affording the desired 5-(2-arylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-diones 4'b-4'u with similar yields ranging from 72 to 95% within a reasonable time scale of 10-55 min (Table 2, products 4'b-4'u). However, primary aliphatic amines (*viz. n*-butylamine, benzylamine, and *c*hexylamine) did not give the desired product(s) as expected. Table 2 summarizes the overall results.

The synthesized products 4/4' (4a-4z and 4'a-4'u) were purified upon washing the crude products with distilled water (5 mL), followed by drying in the open air (see the Experimental Section). All the compounds were fully characterized based on their elemental analyses and detailed spectral studies, including ¹H NMR, ¹³C NMR, DEPT-135, ¹⁹F NMR (for fluorine atom-containing molecules such as 4g, 4h, 4n, 4v, 4'g, 4'q, and 4'r), and HRMS (for representative compounds 4a, 4p, 4'a, and 4'k). We obtained the 2D-NMR [correlation spectroscopy (COSY-45), heteronuclear multiple quantum coherence (HMQC), and heteronuclear multiple bond coherence (HMBC)] spectra for one representative compound (4t) to assign respective the ¹H and ¹³C NMR data (see the Supporting Information). Further, we were also successful in developing suitable crystals for 5-(2-(2methoxyphenyl)hydrazineylidene)pyrimidine-2,4,6-(1H,3H,5H)-trione (4d; Table 2), and its single-crystal X-ray analysis [CCDC 2170064; unit cell parameters: a = 25.717(6)Å, b = 7.9954(13) Å, c = 12.908(3) Å, $\alpha = 90.00^{\circ}$, $\beta = 90.00^{\circ}$,

 $\gamma = 90.00^{\circ}$, and *Pbcn*] documented in the present communication (see the Supporting Information) is in full structural agreement. The ORTEP and packing diagram of the molecule are represented in Figure 1.

TBN, a versatile chemical reagent, is well-regarded for diazotization reactions.¹⁵ For our model reaction, we measured the pH to be 1.49 of the reaction mixture at the very beginning of the reaction. Hence, the transformation occurs under a highly acidic medium (self-catalysis). We thus herein propose a plausible mechanism (Scheme 2) for the catalyst-free C-5 dehydrogenative aza-coupling of barbituric/2-thiobarbituric acids (1/1'), leading to the synthesis of substituted 5-(2arylhydrazono)pyrimidine-2,4,6(1H,3H,5H)-triones (4)/5-(2arylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-diones (4'). The transformation begins with the formation of a diazonium salt 5 from the reaction between anilines 2 and TBN 3 under the reaction conditions. The electrophilic terminal azo-nitrogen of diazo-compound 5 is then attacked by the tautomeric form of barbituric/2-thiobarbituric acids (1/1')through the nucleophilic C-5 position, thereby giving rise to adduct 6. Intermediate 6 rapidly tautomerizes, following the path to yield 4/4', not 7. We performed geometry optimization of the two possible structures, calculated at the DFT/B3LYP level along with the 6-31++G(d,p) basis set using Gaussion09 software.¹⁶ The respective optimized energies were calculated to be -830.6117 and -825.9333 a.u.,

respectively, thereby indicating more excellent thermodynamic stability of structure 4/4'. The relatively greater stability of 4/4' compared to that of 7 is attributed to intramolecular hydrogen bonding of type N–H···O present only within the structural skeleton 4/4'. Further, the transformation proceeds through an ionic pathway that received experimental support when we observed that the conversion is not influenced by any radical scavengers (*viz.* (tetramethylpiperidin-1-yl)oxyl, buty-lated hydroxytoluene, and 1,4-benzoquinone). The proposed mechanism shows that water and *tert*-butanol are the green wastes in this transformation (Scheme 2).

Further, we checked the effectiveness of this catalyst- and additive-free protocol for larger-scale reactions using the model entries (Table 2, compounds 4a and 4'a) in 1.0, 5.0, and 10.0 millimolar scales (Scheme 3). The reactions proceeded satisfactorily in all the three larger scales, affording both the desired products 4a and 4'a with similar yields (90–94%) within almost the same time frame (10–15 min) compared to that in the lower sub-millimolar scale (see the Experimental Section).

3. CONCLUSIONS

In conclusion, we have developed a simple, straightforward, and efficient synthetic protocol for a new series of diversely functionalized 5-(2-arylhydrazono)pyrimidine-2,4,6-(1H,3H,5H)-triones (4)/5-(2-arylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-diones (4') from a one-pot threecomponent reaction between barbituric acids (1)/2-thiobarbituric acids (1'), primary aromatic amines (2), and TBN (3) in an acetonitrile solvent, without the aid of any catalysts/ additives, just under ambient conditions that are sufficient enough to implement the facile $C(sp^3)$ -H functionalization of barbituric/2-thiobarbituric acids to furnish the hydrazone molecular hybrids via C-5 dehydrogenative aza-coupling. The process is devoid of any column chromatographic purification. The key advantages of this newly developed protocol are the operational simplicity, use of commercially available low-cost starting materials with a broader substrate scope, mild reaction conditions at room temperature, avoidance of catalysts/ additives, clean reaction profile, excellent regioselectivity, good to excellent yields with short reaction times, no need for chromatographic purification, energy efficiency, and gramscale synthetic applicability.

4. EXPERIMENTAL SECTION

4.1. General. All chemicals (analytical grade) were purchased from reputed companies and used without further purification. ¹H, ¹³C, and ¹⁹F NMR spectra were collected at 400 MHz, 100 MHz, and 376 MHz respectively, on a Bruker DRX spectrometer using DMSO- d_6 and CDCl₃ as the solvent. Chemical shifts were reported in δ (ppm), relative to the internal standard, tetramethylsilane. The signals observed are described as s (singlet), d (doublet), t (triplet), and m (multiplet). Coupling constants are reported as the J value in hertz. Structural assignments were determined with additional information from gCOSY, gHSQC, and gHMBC experiments. Elemental analyses were performed using a PerkinElmer 2400 Series II elemental analyzer instrument. MS was performed using a Bruker maXis Impact (Q-TOF) and a Microtek Q-TOF Micro YA 263 Waters high-resolution mass spectrometer. X-ray single crystallographic data were collected on an X'Calibur CCD area detector diffractometer (Agilent SuperNova SC-XRD). Melting point was recorded on a Chemiline CL-725 melting point apparatus and is uncorrected. thin-layer chromatography (TLC) was performed using silica gel 60 F_{254} (Merck) plates.

4.2. General Procedure for the Synthesis of Substituted 5-(2-Arylhydrazono)pyrimidine-2,4,6-(1H,3H,5H)-triones (4) and 5-(2-Arylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-diones (4'). A magnetic stir bar, barbituric acids (1; 0.2 mmol)/2-thiobarbituric acids (1'; 0.2 mmol), primary aromatic amines (2; 0.2 mmol), TBN (3; 0.20 mmol), and 2 mL of acetonitrile were transferred to an oven-dried reaction vessel in a sequential manner. The reaction mixture was then stirred at ambient temperature (28-30 °C) for a stipulated time frame (10–75 min, monitored by TLC). On completion of the reaction, 2 mL of distilled water was added to the resulting mixture and shaken well for a while when the product started to precipitate. The precipitate was allowed to settle down and then filtered off using ordinary filter paper. The crude solid mass was thoroughly washed with 5 mL of distilled water, followed by drying in the open air to obtain pure products, 5-(2-arylhydrazono)pyrimidine-2,4,6-(1H,3H,5H)-triones 4 (4a-4z)/5-(2-arylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-diones 4' (4'a-4'u). All the compounds were fully characterized based on their elemental analyses and detailed spectral studies, including ¹H NMR, ¹³C NMR, DEPT-135, ¹⁹F NMR (for 4g, 4h, 4n, 4v, 4'g, 4'q, and 4'r), and HRMS (for 4a, 4p, 4'a, and 4'k). We obtained the 2D NMR (COSY-45, HMQC, and HMBC) spectra for one representative compound (4t) to assign the respective ¹H and ¹³C NMR data (see the Supporting Information). Single X-ray crystallographic studies for a representative entry, 5-(2-(2methoxyphenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4d; CCDC 2170064; see the Supporting Information), supported the structural skeleton deduced from detailed spectroscopic studies.

4.3. Gram-Scale Syntheses of Two Representative Compounds 4a and 4'a. A magnetic stir bar, barbituric acid (1a; 1.0 mmol, 0.128 g)/2-thiobarbituric acid (1'a; 1.0 mmol, 0.144 g), aniline (2a; 1.0 mmol, 0.093 g), TBN (3; 1.0 mmol, 0.115 g), and 10 mL of acetonitrile were transferred to an oven-dried round-bottom flask in a sequential manner. The reaction mixtures were then stirred at ambient temperature (28-30 °C) for 15 and 10 min (monitored by TLC), respectively. On completion of the reaction, 10 mL of distilled water was added to the resulting mixture and shaken well for a while when the product started to precipitate. The precipitate was allowed to settle down and filtered off using ordinary filter paper. The crude solid mass was thoroughly washed with 15 mL of distilled water, followed by drying in the open air to obtain the pure product. Upon drying, pure 4a and 4'a were isolated in 94% (0.218 g) and 92% (0.228 g) yields, respectively, and the physical and spectral properties (see the Supporting Information) of which were found to be similar to those obtained by the 0.2 mmol scale reaction. Further, two more gram-scale preparations (viz. 5 mmol and 10 mmol scale) were performed for both 4a and 4'a similarly. The mixtures of barbituric acid (1a; 5.0 mmol, 0.640 g/10.0 mmol, 1.281 g)/2thiobarbituric acid (1'a; 5.0 mmol, 0.721 g/10.0 mmol, 1.441 g), aniline (2a; 5.0 mmol, 0.465 g/10.0 mmol, 0.931 g), TBN (3; 5.0 mmol, 0.575 g/10.0 mmol, 1.150 g), and 20 mL/30 mL of acetonitrile were reacted by stirring at ambient conditions, and the corresponding products were isolated upon standard workup in 91% (4a: 1.055 g in the 5 mmol scale and 2.117 g in

the 10 mmol scale) and 90% (4'a: 1.117 g in the 5 mmol scale and 2.234 g in the 10 mmol scale) yields within the same time scale of 15 and 10 min, respectively.

4.4. Physical and Spectral Data of all the Synthesized Compounds 4 (4a-4z and 4'a-4'u). Physical and spectral data of all the synthesized compounds 4 (4a-4z and 4'a-4'u)are given below.

4.4.1. 5-(2-Phenylhydrazono)pyrimidine-2,4,6(1H,3H,5H)trione (4a). Straw yellow powder, yield: 93% (43 mg; 0.2 mmol scale), mp 298–299 °C. ¹H NMR (400 MHz, DMSO d_6): δ 14.14 (br s, 1H, -NH), 11.51 (br s, 1H, -NH), 11.29 (br s, 1H, -NH), 7.57 (d, 2H, J = 7.6 Hz, Ar-H), 7.47–7.43 (m, 2H, Ar-H), 7.26–7.22 (m, 1H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 162.30 (-NHCO-), 160.15 (-NHCO-), 149.99 (-NHCONH-), 141.42 (C), 129.88 (2× CH), 126.31 (CH), 117.83 (C), 116.81 (2× CH) ppm. HRMS (TOF MS ES⁺): m/z 233.0675 [M + H]⁺ calcd for C₁₀H₈N₄O₃H; found, 233.0676.

4.4.2. 5-(2-(o-Tolyl)hydrazono)pyrimidine-2,4,6-(1H,3H,5H)-trione (**4b**). Orange yellow powder, yield: 69% (34 mg; 0.2 mmol scale), mp 290–291 °C. ¹H NMR (400 MHz, DMSO-d₆): δ 14.49 (br s, 1H, -NH), 11.59 (br s, 1H, -NH), 11.33 (br s, 1H, -NH), 7.67 (d, 1H, J = 8.0 Hz, Ar-H), 7.37–7.31 (m, 2H, Ar-H), 7.17 (t, 1H, J = 7.6 Hz, Ar-H), 2.35 (s, 3H, Ar-CH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-d₆): δ 162.84 (-NHCO-), 159.85 (-NHCO-), 149.80 (-NHCONH-), 139.17 (C), 131.19 (CH), 127.63 (CH), 125.91 (CH), 125.61 (C), 118.51 (C), 114.70 (CH), 16.33 (Ar-CH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₃: C, 53.66; H, 4.09; N, 22.75. Found: C, 53.52; H, 4.13; N, 22.84.

4.4.3. 5-(2-(p-Tolyl)hydrazono)pyrimidine-2,4,6-(1H,3H,5H)-trione (4c). Yellow powder, yield: 87% (43 mg; 0.2 mmol scale), mp 297–298 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.19 (br s, 1H, -NH), 11.45 (br s, 1H, -NH), 11.23 (br s, 1H, -NH), 7.47 (d, 2H, J = 8.4 Hz, Ar-H), 7.26 (d, 2H, J = 8.4 Hz, Ar-H), 2.31 (s, 3H, Ar-CH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 162.19 (-NHCO-), 159.88 (-NHCO-), 149.79 (-NHCONH-), 139.03 (C), 135.68 (C), 130.10 (2× CH), 117.14 (C), 116.62 (2× CH), 20.53 (Ar-CH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₃: C, 53.66; H, 4.09; N, 22.75. Found: C, 53.79; H, 4.14; N, 22.63.

4.4.4. 5-(2-(2-Methoxyphenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (**4d**). Yellowish orange powder, yield: 90% (47 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO-*d*₆): δ 14.42 (s, 1H, -NH), 11.52 (s, 1H, -NH), 11.29 (s, 1H, -NH), 7.64 (d, 1H, *J* = 7.6 Hz, Ar-H), 7.25–7.18 (m, 2H, Ar-H), 7.10–7.07 (m, 1H, Ar-H), 3.94 (s, 3H, Ar-OCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆): δ 162.51 (-NHCO-), 159.89 (-NHCO-), 149.85 (-NHCONH-), 148.12 (*C*), 129.85 (*C*), 126.70 (*C*H), 121.67 (*C*H), 118.49 (*C*), 114.76 (*C*H), 112.15 (*C*H), 56.28 (Ar-OCH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₄: *C*, 50.38; H, 3.84; N, 21.37. Found: C, 50.47; H, 3.89; N, 21.49.

4.4.5. 5-(2-(4-Methoxyphenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (**4e**). Greenish yellow powder, yield: 94% (49 mg; 0.2 mmol scale), mp 294–295 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.29 (s, 1H, –NH), 11.43 (s, 1H, –NH), 11.22 (s, 1H, –NH), 7.53 (d, 2H, J = 9.2 Hz, Ar-H), 7.02 (d, 2H, J = 8.8 Hz, Ar-H), 3.76 (s, 3H, Ar-OCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 162.50 (–NHCO–), 160.46 (-NHCO-), 158.20 (*C*), 150.13 (-NHCONH-), 134.93 (*C*), 118.61 (2× *C*H), 116.66 (*C*), 115.23 (2× *C*H), 55.74 (Ar-OCH₃) ppm. Elemental Anal. Calcd (%) for $C_{11}H_{10}N_4O_4$: C, 50.38; H, 3.84; N, 21.37. Found: C, 50.52; H, 3.79; N, 21.49.

4.4.6. 5-(2-(4-(Methylthio)phenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (**4f**). Brownish yellow powder, yield: 97% (54 mg; 0.2 mmol scale), mp 299–301 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.18 (s, 1H, –NH), 11.49 (s, 1H, –NH), 11.27 (s, 1H, –NH), 7.53 (d, 2H, J = 8.4 Hz, Ar-H), 7.33 (d, 2H, J = 8.8 Hz, Ar-H), 2.48 (s, 3H, Ar-SCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 162.33 (–NHCO–), 160.19 (–NHCO–), 150.02 (–NHCONH–), 138.77 (C), 136.27 (C), 127.22 (2× CH), 117.56 (2× CH), 117.49 (C), 15.04 (Ar-SCH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₃S: C, 47.48; H, 3.62; N, 20.13. Found: C, 47.62; H, 3.57; N, 20.22.

4.4.7. 5-(2-(4-((*Trifluoromethyl*)*thio*)*phenyl*)*hydrazono*)*pyrimidine-2,4,6*(1*H,3H,5H*)-*trione* (*4g*). Deep yellow powder, yield: 80% (53 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO-*d*₆): δ 13.98 (br s, 1H, -NH), 11.59 (br s, 1H, -NH), 11.37 (br s, 1H, -NH), 7.78 (d, 2H, *J* = 8.8 Hz, Ar-H), 7.71 (d, 2H, *J* = 8.8 Hz, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆): δ 161.76 (-NHCO-), 159.66 (-NHCO-), 149.79 (-NHCONH-), 144.17 (C), 137.88 (2× CH), 129.57 ($-SCF_3$, J_{CF}^{-1} = 307 Hz), 119.42 (C), 118.68 (C), 117.68 (2× CH) ppm. ¹⁹F NMR (376 MHz, DMSO-*d*₆): δ –42.53 ppm. Elemental Anal. Calcd (%) for C₁₁H₇F₃N₄O₃S: C, 39.76; H, 2.12; N, 16.86. Found: C, 39.59; H, 2.07; N, 16.91.

4.4.8. 5-(2-(4-(Trifluoromethoxy)phenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4h). Greenish yellow powder, yield: 85% (54 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.04 (br s, 1H, -NH), 11.52 (br s, 1H, -NH), 11.33 (br s, 1H, -NH), 7.71 (d, 2H, J = 8.8 Hz, Ar-H), 7.46 (d, 2H, J = 8.4 Hz, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 161.87 (-NHCO-), 159.81 (-NHCO-), 149.80 (-NHCONH-), 145.58 (C), 140.63 (C), 122.47 (2× CH), 120.08 ($-OCF_3$, J_{CF}^{-1} = 254 Hz), 118.48 (C), 118.19 (2× CH) ppm. ¹⁹F NMR (376 MHz, DMSO- d_6): δ – 57.02 ppm. Elemental Anal. Calcd (%) for C₁₁H₇F₃N₄O₄: C, 41.78; H, 2.23; N, 17.72. Found: C, 41.81; H, 2.18; N, 17.87.

4.4.9. 5-(2-(2-Nitrophenyl))hydrazono)pyrimidine-2,4,6-(1H,3H,5H)-trione (4i). Yellow powder, yield: 85% (47 mg; 0.2 mmol scale), mp 278–279 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 15.19 (s, 1H, –NH), 11.74 (s, 1H, –NH), 11.51 (s, 1H, –NH), 8.26 (d, 1H, J = 8.4 Hz, Ar-H), 8.09 (d, 1H, J = 8.4 Hz, Ar-H), 7.89–7.86 (m, 1H, Ar-H), 7.40–7.36 (m, 1H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 161.57 (–NHCO–), 159.71 (–NHCO–), 149.93 (–NHCONH–), 137.67 (C), 136.78 (CH), 135.62 (C), 126.21 (CH), 124.98 (CH), 122.31 (C), 117.33 (CH) ppm. Elemental Anal. Calcd (%) for C₁₀H₇N₅O₅: C, 43.33; H, 2.55; N, 25.27. Found: C, 43.46; H, 2.61; N, 25.41.

4.4.10. 5-(2-(3-Nitrophenyl)hydrazono)pyrimidine-2,4,6-(1H,3H,5H)-trione (**4**j). Mustard yellow powder, yield: 90% (50 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 13.99 (br s, 1H, -NH), 11.59 (br s, 1H, -NH), 11.39 (s, 1H, -NH), 8.44 (s, 1H, Ar-H), 8.03 (d, 2H, *J* = 8.0 Hz, Ar-H), 7.73-7.69 (m, 1H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 161.57 (-NHCO-), 159.67 (-NHCO-), 149.78 (-NHCONH-), 148.66 (*C*), 142.99 4.4.11. 5-(2-(4-Nitrophenyl)hydrazono)pyrimidine-2,4,6-(1H,3H,5H)-trione (4k). Yellow powder, yield: 94% (52 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO d_6): δ 11.46 (s, 1H, -NH), 8.29 (d, 2H, J = 9.2 Hz, Ar-H), 7.76 (d, 2H, J = 9.2 Hz, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 161.75 (-NHCO-), 159.74 (-NHCO-), 149.92 (-NHCONH-), 147.06(C), 143.99 (C), 125.77 (2× CH), 120.88 (C), 116.87 (2× CH) ppm. Elemental Anal. Calcd (%) for C₁₀H₇N₅O₅: C, 43.33; H, 2.55; N, 25.27. Found: C, 43.48; H, 2.49; N, 25.39.

4.4.12. 5-(2-(4-Bromophenyl))hydrazono)pyrimidine-2,4,6-(1H,3H,5H)-trione (4I). Pale yellow powder, yield: 69% (43 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.03 (br s, 1H, -NH), 11.54 (br s, 1H, -NH), 11.33 (br s, 1H, -NH), 7.63 (d, 2H, J = 8.8 Hz, Ar-H), 7.54 (d, 2H, J = 9.2 Hz, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 162.12 (-NHCO-), 160.06 (-NHCO-), 149.99 (-NHCONH-), 140.95 (C), 132.64 (2× CH), 118.81 (2× CH), 118.42 (C), 118.18 (C) ppm. Elemental Anal. Calcd (%) for C₁₀H₇BrN₄O₃C, 38.61; H, 2.27; N, 18.01. Found: C, 38.49; H, 2.33; N, 18.19.

4.4.13. 5-(2-(2,5-Dichlorophenyl))hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4m). Golden yellow powder, yield: 71% (43 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO-d₆): δ 14.36 (br s, 1H, -NH), 11.77 (br s, 1H, -NH), 11.51 (br s, 1H, -NH), 7.68 (d, 1H, J = 2.4 Hz, Ar-H), 7.64 (d, 1H, J = 8.8 Hz, Ar-H), 7.30 (dd, 1H, J = 8.4 and 2.4 Hz, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-d₆): δ 162.48 (-NHCO-), 159.35 (-NHCO-), 149.67 (-NHCONH-), 138.83 (C), 133.39 (C), 131.56 (CH), 125.69 (CH), 121.32 (C), 119.13 (C), 115.33 (CH) ppm. Elemental Anal. Calcd (%) for C₁₀H₆Cl₂N₄O₃. C, 39.89; H, 2.01; N, 18.61. Found: C, 39.71; H, 2.07; N, 18.49.

4.4.14. 5-(2-(3,5-Bis(trifluoromethyl)phenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4n). Bright yellow powder, yield: 82% (60 mg; 0.2 mmol scale), mp 289–290 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 13.96 (br s, 1H, –NH), 11.65 (br s, 1H, –NH), 11.46 (s, 1H, –NH), 8.26 (s, 2H, Ar-H), 7.86 (s, 1H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO d_6): δ 161.62 (–NHCO–), 160.08 (–NHCO–), 150.02 (–NHCONH–), 143.97 (C), 132.25–131.25 (m, 2C), 123.32 (2× CF₃, $J_{CF}^{-1} = 271$ Hz), 120.32 (C), 118.15 (CH), 117.20 (2× CH) ppm. ¹⁹F NMR (376 MHz, DMSO- d_6): δ –61.64. Elemental Anal. Calcd (%) for C₁₂H₆F₆N₄O₃. C, 39.14; H, 1.64; N, 15.22. Found: C, 39.27; H, 1.51; N, 15.34.

4.4.15. 5-(2-(4-(4-Chlorophenoxy)phenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (40). Greenish yellow powder, yield: 93% (67 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.18 (br s, 1H, -NH), 11.48 (br s, 1H, -NH), 11.28 (br s, 1H, -NH), 7.62–7.44 (m, 4H, Ar-H), 7.09 (br s, 4H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 162.08 (-NHCO–), 159.88 (-NHCO–), 155.57 (C), 154.27 (C), 149.82 (-NHCONH–), 137.49 (C), 129.97 (2× CH), 127.39 (C), 120.27 (2× CH), 119.96 (2× CH), 118.53 (2× CH), 117.42 (C) ppm. Elemental Anal. Calcd (%) for C₁₆H₁₁ClN₄O₄: C, 53.57; H, 3.09; N, 15.62. Found: C, 53.71; H, 3.14; N, 15.51. 4.4.16. 1,3-Dimethyl-5-(2-phenylhydrazono)pyrimidine-

4.4.16. 1,3-Dimethyl-5-(2-phenylhydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (**4p**). Yellow powder, yield: 87% (45 Article

mg; 0.2 mmol scale), mp 266–267 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.64 (br s, 1H, –NH), 7.57–7.55 (m, 2H, Ar-H), 7.46–7.42 (m, 2H, Ar-H), 7.29–7.26 (m, 1H, Ar-H), 3.43 (s, 3H, –NCH₃), 3.39 (s, 3H, –NCH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.31 (–N(CH₃)CO–), 159.26 (–N(CH₃)CO–), 150.85 (CO(N–CH₃)₂), 140.88 (C), 129.88 (2× CH), 127.39 (CH), 117.40 (2× CH), 116.46 (C), 28.71 (–NCH₃), 27.62 (–NCH₃) ppm. HRMS (TOF MS ES⁺): *m/z* 261.0982 [M + H]⁺ calcd for C₁₂H₁₂N₄O₃H; found, 261.0976; *m/z* 283.0802 [M + Na]⁺ calcd for C₁₂H₁₂N₄O₃Na; found, 283.0794.

4.4.17. 1,3-Dimethyl-5-(2-(o-tolyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4q). Yellow powder, yield: 95% (52 mg; 0.2 mmol scale), mp 251–252 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.83 (br s, 1H, -NH), 7.94 (d, 1H, J = 8.0 Hz, Ar-H), 7.31 (t, 1H, J = 7.6 Hz, Ar-H), 7.23–7.16 (m, 2H, Ar-H), 3.43 (s, 3H, -NCH₃), 3.39 (s, 3H, -NCH₃), 2.45 (s, 3H, Ar-CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.43 (-N(CH₃)CO–), 159.32 (-N(CH₃)CO–), 150.84 (CO(N– CH₃)₂), 139.14 (C), 131.15 (CH), 127.89 (CH), 127.15 (CH), 126.38 (C), 117.00 (C), 116.49 (CH), 28.70 (-NCH₃), 27.63 (-NCH₃), 17.05 (Ar-CH₃) ppm. Elemental Anal. Calcd (%) for C₁₃H₁₄N₄O₃: C, 56.93; H, 5.14; N, 20.43. Found: C, 56.77; H, 5.19; N, 20.34.

4.4.18. 1,3-Dimethyl-5-(2-(p-tolyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (**4r**). Yellow powder, yield: 87% (45 mg; 0.2 mmol scale), mp 244–245 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.67 (br s, 1H, -NH), 7.45 (d, 2H, *J* = 8.4 Hz, Ar-H), 7.23 (d, 2H, *J* = 8.4 Hz, Ar-H), 3.41 (s, 3H, -NCH₃), 3.37 (s, 3H, -NCH₃), 2.37 (s, 3H, Ar-CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.32 (-N(CH₃)CO-), 159.35 (-N(CH₃)CO-), 150.88 (CO(N-CH₃)₂), 138.59 (C), 137.70 (C), 130.42 (2× CH), 117.34 (2× CH), 115.97 (C), 28.64 (-NCH₃), 27.55 (-NCH₃), 21.25 (Ar-CH₃) ppm. Elemental Anal. Calcd (%) for C₁₃H₁₄N₄O₃: C, 56.93; H, 5.14; N, 20.43. Found: C, 56.78; H, 5.09; N, 20.36.

4.4.19. 5-(2-(2-Methoxyphenyl)hydrazono)-1,3-dimethylpyrimidine-2,4,6(1H,3H,5H)-trione (4s). Orange yellow powder, yield: 95% (52 mg; 0.2 mmol scale), mp 220–222 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.73 (br s, 1H, -NH), 7.93 (dd, 1H, J = 8.0 and 1.2 Hz, Ar-H), 7.24–7.19 (m, 1H, Ar-H), 7.04 (t, 1H, J = 7.6 Hz, Ar-H), 6.96 (d, 1H, J = 8.4 Hz, Ar-H), 3.98 (s, 3H, Ar-OCH₃) 3.41 (s, 3H, $-NCH_3$), 3.38 (s, 3H, $-NCH_3$) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 160.97 (2C), 159.51 ($-N(CH_3)CO-$), 150.92 ($CO(N-CH_3)_2$), 149.10 (C), 130.08 (C), 127.77 (CH), 121.86 (CH), 116.76 (CH), 111.20 (CH), 56.20 (Ar-OCH₃), 28.63 ($-NCH_3$), 27.59 ($-NCH_3$) ppm. Elemental Anal. Calcd (%) for C₁₃H₁₄N₄O₄: C, 53.79; H, 4.86; N, 19.30. Found: C, 53.84; H, 4.79; N, 19.18.

4.4.20. 5-(2-(4-Methoxyphenyl)hydrazono)-1,3-dimethylpyrimidine-2,4,6(1H,3H,5H)-trione (4t). Yellow powder, yield: 62% (36 mg; 0.2 mmol scale), mp 211–212 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.77 (br s, 1H, –NH), 7.49 (d, 2H, J = 9.2 Hz, Ar-H), 6.94 (d, 2H, J = 9.2 Hz, Ar-H), 3.82 (s, 3H, Ar-OCH₃), 3.39 (s, 3H, –NCH₃), 3.36 (s, 3H, –NCH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.37 (–N(CH₃) CO–), 159.44 (–N(CH₃)CO–), 159.20 (C), 150.89 (CO(N–CH₃)₂), 134.40 (C), 118.86 (2× CH), 115.58 (C), 115.10 (2× CH), 55.74 (Ar-OCH₃), 28.58 (–NCH₃), 27.49 (–NCH₃) ppm. Elemental Anal. Calcd (%) for C₁₃H₁₄N₄O₄: C, 53.79; H, 4.86; N, 19.30. Found: C, 53.68; H, 4.77; N, 19.21.

4.4.21. 1,3-Dimethyl-5-(2-(4-(methylthio)phenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4u). Orange powder, yield: 82% (50 mg; 0.2 mmol scale), mp 220–222 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.68 (br s, 1H, -NH), 7.48 (d, 2H, J = 8.4 Hz, Ar-H), 7.28 (d, 2H, J = 8.8 Hz, Ar-H), 3.41 (s, 3H, -NCH₃), 3.37 (s, 3H, -NCH₃), 2.50 (s, 3H, Ar-SCH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.33 (-N(CH₃)CO-), 159.25 (-N(CH₃)CO-), 150.81 (CO(N-CH₃)₂), 138.35 (C), 138.19 (C), 127.51 (2× CH), 117.84 (2× CH), 116.20 (C), 28.68 (-NCH₃), 27.59 (-NCH₃), 15.95 (Ar-SCH₃) ppm. Elemental Anal. Calcd (%) for C₁₃H₁₄N₄O₃S: C, 50.97; H, 4.61; N, 18.29. Found: C, 50.89; H, 4.68; N, 18.37.

4.4.22. 1,3-Dimethyl-5-(2-(4-(trifluoromethoxy)phenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4v). Bright yellow powder, yield: 76% (52 mg; 0.2 mmol scale), mp 174–175 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.61 (br s, 1H, -NH), 7.58 (d, 2H, J = 8.8 Hz, Ar-H), 7.29 (d, 2H, J = 8.8 Hz, Ar-H), 3.42 (s, 3H, -NCH₃), 3.39 (s, 3H, -NCH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.25 (-N(CH₃) CO-), 159.04 (-N(CH₃)CO-), 150.69 (CO(N-CH₃)₂), 147.70 (C), 139.39 (C), 122.59 (2× CH), 120.52 (-OCF₃, J_{CF}¹ = 257 Hz), 118.47 (2× CH), 117.01 (C), 28.78 (-NCH₃), 27.70 (-NCH₃) ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ -58.03 ppm. Elemental Anal. Calcd (%) for C₁₃H₁₁F₃N₄O₄: C, 45.36; H, 3.22; N, 16.28. Found: C, 45.23; H, 3.19; N, 16.39.

4.4.23. 1,3-Dimethyl-5-(2-(2-nitrophenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (**4w**). Yellow powder, yield: 84% (51 mg; 0.2 mmol scale), mp 260–262 °C. ¹H NMR (400 MHz, DMSO-*d*₆): δ 15.24 (s, 1H, –NH), 8.29 (d, 1H, *J* = 7.6 Hz, Ar-H), 8.11 (d, 1H, *J* = 8.0 Hz, Ar-H), 7.92 (t, 1H, *J* = 7.6 Hz, Ar-H), 7.42 (t, 1H, *J* = 7.6 Hz, Ar-H), 3.24 (s, 6H, 2× –NCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆): δ 159.86 (–N(CH₃)CO–), 158.31 (–N(CH₃)CO–), 150.48 (CO(N–CH₃)₂), 137.38 (C), 136.65 (CH), 135.63 (C), 126.08 (CH), 125.00 (CH), 121.45 (C), 117.20 (CH), 28.31 (–NCH₃), 27.49 (–NCH₃) ppm. Elemental Anal. Calcd (%) for C₁₂H₁₁N₅O₅: C, 47.22; H, 3.63; N, 22.94. Found: C, 47.08; H, 3.69; N, 22.82.

4.4.24. 1,3-Dimethyl-5-(2-(3-nitrophenyl)hydrazono)pyrimidine-2,4,6(1H,3H,5H)-trione (4x). Yellow powder, yield: 66% (40 mg; 0.2 mmol scale), mp 228–230 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.03 (br s, 1H, -NH), 8.51– 8.50 (m, 1H, Ar-H), 8.09–8.04 (m, 2H, Ar-H), 7.76–7.72 (m, 1H, Ar-H), 3.23 (s, 3H, –NCH₃), 3.22 (s, 3H, –NCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 159.89 (–N(CH₃) CO–), 158.56 (–N(CH₃)CO–), 150.60 (CO(N–CH₃)₂), 148.67 (C), 143.02 (C), 131.06 (CH), 123.03 (CH), 119.85 (CH), 118.88 (C), 111.21 (CH), 28.21 (–NCH₃), 27.30 (–NCH₃) ppm. Elemental Anal. Calcd (%) for C₁₂H₁₁N₅O₅: C, 47.22; H, 3.63; N, 22.94. Found: C, 47.38; H, 3.70; N, 22.85.

4.4.25. 5-(2-(4-Bromophenyl)hydrazono)-1,3-dimethylpyrimidine-2,4,6(1H,3H,5H)-trione (**4y**). Yellow powder, yield: 81% (55 mg; 0.2 mmol scale), mp 278–280 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.57 (br s, 1H, –NH), 7.55 (d, 2H, J = 8.8 Hz, Ar-H), 7.43 (d, 2H, J = 8.8 Hz, Ar-H), 3.42 (s, 3H, –NCH₃), 3.38 (s, 3H, –NCH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.26 (–N(CH₃)CO–), 159.04 (–N(CH₃)CO–), 150.70 (CO(N–CH₃)₂), 139.93 (C), 132.94 (2× CH), 120.46 (C), 118.72 (2× CH), 116.87 (C), 28.76 (-NCH₃), 27.69 (-NCH₃) ppm. Elemental Anal. Calcd (%) for C₁₂H₁₁BrN₄O₃: C, 42.50; H, 3.27; N, 16.52. Found: C, 42.63; H, 3.24; N, 16.43.

4.4.26. 5-(2-(4-(4-Chlorophenoxy)phenyl)hydrazono)-1,3dimethylpyrimidine-2,4,6(1H,3H,5H)-trione (**4z**). Yellow powder, yield: 80% (62 mg; 0.2 mmol scale), mp 202–203 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.73 (br s, 1H, –NH), 7.54 (d, 2H, *J* = 8.4 Hz, Ar-H), 7.32 (d, 2H, *J* = 8.4 Hz, Ar-H), 7.05 (d, 2H, *J* = 8.4 Hz, Ar-H), 6.96 (d, 2H, *J* = 8.4 Hz, Ar-H), 3.42 (s, 3H, –NCH₃), 3.38 (s, 3H, –NCH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 161.38 (–N(CH₃)CO–), 159.29 (–N(CH₃)CO–), 156.37 (C), 155.43 (C), 150.84 (CO(N– CH₃)₂), 136.64 (C), 130.09 (2× CH), 129.12 (C), 120.48 (2× CH), 119.90 (2× CH), 118.98 (2× CH), 116.27 (C), 28.71 (–NCH₃), 27.62 (–NCH₃) ppm. Elemental Anal. Calcd (%) for C₁₈H₁₅ClN₄O₄: C, 55.89; H, 3.91; N, 14.49. Found: C, 55.78; H, 3.97; N, 14.58.

4.4.27. 5-(2-Phenylhydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'a). Brownish yellow powder, yield: 93% (46 mg; 0.2 mmol scale), mp 298–299 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.24 (br s, 1H, -NH), 12.61 (br s, 1H, -NH), 12.45 (br s, 1H, -NH), 7.62 (d, 2H, J = 7.6 Hz, Ar-H), 7.49–7.46 (m, 2H, Ar-H), 7.29–7.26 (m, 1H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.59 (-NHCSNH-), 160.07 (-NHCO-), 158.41 (-NHCO-), 141.25 (C), 129.84 (2× CH), 126.63 (CH), 118.76 (C), 117.09 (2× CH) ppm. HRMS (TOF MS ES⁺): m/z 271.0260 [M + Na]⁺ calcd for C₁₀H₈N₄O₂SNa; found, 271.0264.

4.4.28. 2-Thioxo-5-(2-(o-tolyl)hydrazono)dihydropyrimidine-4,6(1H,5H)-dione (4'b). Pale brown powder, yield: 95% (50 mg; 0.2 mmol scale), mp 285–286 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.60 (br s, 1H, –NH), 12.68 (br s, 1H, –NH), 12.49 (br s, 1H, –NH), 7.68 (d, 1H, J = 8.4 Hz, Ar-H), 7.38–7.32 (m, 2H, Ar-H), 7.19 (t, 1H, J = 7.2 Hz, Ar-H), 2.37 (s, 3H, Ar-CH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.47 (–NHCSNH–), 160.77 (–NHCO–), 158.29 (–NHCO–), 139.03 (C), 131.30 (CH), 127.75 (CH), 126.48 (CH), 126.18 (C), 119.47 (C), 115.02 (CH), 16.36 (Ar-CH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₂S: C, 50.37; H, 3.84; N, 21.36. Found: C, 50.24; H, 3.89; N, 21.48.

4.4.29. 2-Thioxo-5-(2-(p-tolyl)hydrazono)dihydropyrimidine-4,6(1H,5H)-dione (4'c). Orange powder, yield: 94% (49 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.29 (br s, 1H, -NH), 12.57 (br s, 1H, -NH), 12.40 (br s, 1H, -NH), 7.51 (d, 2H, J = 8.4 Hz, Ar-H), 7.28 (d, 2H, J = 8.4 Hz, Ar-H), 2.32 (s, 3H, Ar-CH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.43 (-NHCSNH-), 160.06 (-NHCO-), 158.36 (-NHCO-), 138.92 (C), 136.34 (C), 130.22 (2× CH), 118.18 (C), 117.05 (2× CH), 20.64 (Ar-CH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₂S: C, 50.37; H, 3.84; N, 21.36. Found: C, 50.22; H, 3.79; N, 21.27.

4.4.30. 5-(2-(2-Methoxyphenyl)hydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'd). Red powder, yield: 95% (50 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO-d₆): δ 14.55 (br s, 1H, -NH), 12.59 (br s, 1H, -NH), 12.43 (br s, 1H, -NH), 7.66 (d, 1H, J = 8.0 Hz, Ar-H), 7.28–7.25 (m, 1H, Ar-H), 7.19 (d, 1H, J = 8.0 Hz, Ar-H), 7.09 (t, 1H, J = 7.6 Hz, Ar-H), 3.95 (s, 3H, Ar-OCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-d₆): δ 177.44 (-NHCSNH-), 160.41 (-NHCO-), 158.32 (-NHCO-), 148.45 (C), 129.66 (C), 127.39 (CH), 121.78 (CH), 119.43 (C), 115.13 (CH), 112.27 (CH), 56.38 (Ar-OCH₃) ppm. Elemental Anal. Calcd (%) for $C_{11}H_{10}N_4O_3S$: C, 47.48; H, 3.62; N, 20.13. Found: C, 47.45; H, 3.66; N, 20.17.

4.4.31. 5-(2-(4-Methoxyphenyl)hydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'e). Red powder, yield: 77% (43 mg; 0.2 mmol scale), mp 280–282 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.41 (s, 1H, –NH), 12.52 (s, 1H, –NH), 12.36 (s, 1H, –NH), 7.59 (d, 2H, J = 8.8 Hz, Ar-H), 7.05 (d, 2H, J = 8.8 Hz, Ar-H), 3.79 (s, 3H, Ar-OCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.29 (–NHCSNH–), 160.07 (–NHCO–), 158.42 (–NHCO–), 158.29 (C), 134.64 (C), 118.76 (2× CH), 117.59 (C), 115.03 (2× CH), 55.52 (Ar-OCH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₃S: C, 47.48; H, 3.62; N, 20.13. Found: C, 47.33; H, 3.69; N, 20.27.

4.4.32. 5-(2-(4-(Methylthio)phenyl)hydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'f). Brownish red powder, yield: 85% (50 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.28 (br s, 1H, -NH), 12.57 (br s, 1H, -NH), 12.41 (br s, 1H, -NH), 7.58 (d, 2H, J = 8.8 Hz, Ar-H), 7.35 (d, 2H, J = 8.8 Hz, Ar-H), 3.35 (s, 3H, Ar-SCH₃) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.39 (-NHCSNH-), 159.97 (-NHCO-), 158.30 (-NHCO-), 138.46 (C), 136.77 (C), 126.93 (2× CH), 118.33 (C), 117.72 (2× CH), 14.78 (Ar-SCH₃) ppm. Elemental Anal. Calcd (%) for C₁₁H₁₀N₄O₂S₂: C, 44.88; H, 3.42; N, 19.03. Found: C, 44.69; H, 3.37; N, 19.17.

4.4.33. 2-Thioxo-5-(2-(4-(trifluoromethoxy)phenyl)hydrazono)dihydropyrimidine-4,6(1H,5H)-dione (4'g). Yellow powder, yield: 77% (51 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 12.55 (br s, 2H, 2× -NH), 7.73 (d, 2H, J = 8.8 Hz, Ar-H), 7.48 (d, 2H, J = 8.8 Hz, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.73 (-NHCSNH-), 159.17 (-NHCO-), 146.16 (-NHCO-), 140.58 (C), 122.63 (2× CH), 121.49 (C), 119.36 (C), 118.94 (C), 118.83 (2× CH) ppm. ¹⁹F NMR (376 MHz, DMSO- d_6): δ -56.99 ppm. Elemental Anal. Calcd (%) for C₁₁H₇F₃N₄O₃S: C, 39.76; H, 2.12; N, 16.86. Found: C, 39.61; H, 2.17; N, 16.92.

4.4.34. 5-(2-(2-Nitrophenyl)hydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'h). Orange yellow powder, yield: 75% (44 mg; 0.2 mmol scale), mp 277–279 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 15.24 (s, 1H, –NH), 12.77 (s, 1H, –NH), 12.61 (s, 1H, –NH), 8.28 (d, 1H, *J* = 8.0 Hz, Ar-H), 8.09 (d, 1H, *J* = 8.4 Hz, Ar-H), 7.92–7.88 (m, 1H, Ar-H), 7.43–7.39 (m, 1H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.89 (–NHCSNH–), 159.40 (–NHCO–), 157.97 (–NHCO–), 137.32 (C), 136.68 (CH), 135.72 (C), 126.14 (CH), 125.20 (CH), 122.80 (C), 117.39 (CH) ppm. Elemental Anal. Calcd (%) for C₁₀H₇N₅O₄S: C, 40.96; H, 2.41; N, 23.88. Found: C, 40.78; H, 2.36; N, 23.75.

4.4.35. 5-(2-(4-Bromophenyl)hydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (**4**'*i*). Yellow powder, yield: 78% (51 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO-d₆): δ 14.10 (br s, 1H, -NH), 12.60 (br s, 1H, -NH), 12.45 (br s, 1H, -NH), 7.65 (d, 2H, J = 8.8 Hz, Ar-H), 7.59 (d, 2H, J = 8.8 Hz, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO-d₆): δ 177.59 (-NHCSNH-), 159.76 (-NHCO-), 158.25 (-NHCO-), 140.73 (C), 132.53 (2× CH), 119.20 (C), 118.98 (2× CH), 118.48 (C) ppm. Elemental Anal. Calcd (%) for C₁₀H₇BrN₄O₂S: C, 36.71; H, 2.16; N, 17.13. Found: C, 36.59; H, 2.21; N, 17.24. 4.4.36. 5-(2-(4-(4-Chlorophenoxy)phenyl)hydrazono)-2thioxodihydropyrimidine-4,6(1H,5H)-dione (4'j). Deep orange powder, yield: 92% (69 mg; 0.2 mmol scale), mp >300 °C. ¹H NMR (400 MHz, DMSO- d_6): δ 14.30 (br s, 1H, -NH), 12.58 (br s, 1H, -NH), 12.42 (br s, 1H, -NH), 7.66– 7.45 (m, 4H, Ar-H), 7.13–7.08 (m, 4H, Ar-H) ppm. ¹³C{¹H} NMR (100 MHz, DMSO- d_6): δ 177.49 (-NHCSNH–), 159.97 (-NHCO–), 158.42 (-NHCO–), 155.44 (C), 154.82 (C), 137.32 (C), 130.07 (2× CH), 127.58 (C), 120.48 (2× CH), 119.96 (2× CH), 119.05 (2× CH), 118.44 (C) ppm. Elemental Anal. Calcd (%) for C₁₆H₁₁ClN₄O₃S: C, 51.27; H, 2.96; N, 14.95. Found: C, 51.14; H, 2.91; N, 14.82.

4.4.37. 1,3-Diethyl-5-(2-phenylhydrazineylidene)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'k). Brownish yellow powder, yield: 85% (52 mg; 0.2 mmol scale), mp 192–193 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.88 (br s, 1H, –NH), 7.58 (d, 2H, J = 8.0 Hz, Ar-H), 7.46–7.42 (m, 2H, Ar-H), 7.31–7.27 (m, 1H, Ar-H), 4.62–4.52 (m, 4H, 2× –NCH₂CH₃), 1.34–1.31 (m, 6H, 2× –NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.70 (S= C(–NC₂H₅)₂), 159.45 (–N(C₂H₅)CO–), 157.67 (–N-(C₂H₅)CO–), 140.80 (2C), 129.92 (2× CH), 127.76 (CH), 117.69 (2× CH), 43.89 (–NCH₂CH₃), 42.72 (–NCH₂CH₃), 12.49 (–NCH₂CH₃), 12.29 (–NCH₂CH₃) ppm. HRMS (TOF MS ES⁺): m/z 305.1067 [M + H]⁺ calcd for C₁₄H₁₆N₄O₂SN₄; found, 3027.0888.

4.4.38. 1,3-Diethyl-2-thioxo-5-(2-(o-tolyl)hydrazineylidene)dihydropyrimidine-4,6(1H,5H)-dione (4'I). Yellow powder, yield: 94% (60 mg; 0.2 mmol scale), mp 175– 176 °C. ¹H NMR (400 MHz, CDCl₃): δ 15.09 (br s, 1H, -NH), 7.97 (d, 1H, J = 8.0 Hz, Ar-H), 7.34–7.30 (m, 1H, Ar-H), 7.24–7.18 (m, 2H, Ar-H), 4.62–4.54 (m, 4H, 2x -NCH₂CH₃), 2.48 (s, 3H, Ar-CH₃), 1.35–1.31 (m, 6H, 2x -NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.69 (S=C(-NC₂H₅)₂), 159.60 (-N(C₂H₅)CO-), 157.82 (-N(C₂H₅)CO-), 139.15 (C), 131.24 (CH), 127.99 (CH), 127.59 (CH), 126.82 (C), 118.30 (C), 116.83 (CH), 43.91 (-NCH₂CH₃), 42.74 (-NCH₂CH₃), 17.13 (Ar-CH₃), 12.54 (-NCH₂CH₃), 12.30 (-NCH₂CH₃) ppm. Elemental Anal. Calcd (%) for C₁₅H₁₈N₄O₂S: C, 56.58; H, 5.70; N, 17.60. Found: C, 56.71; H, 5.77; N, 17.53.

4.4.39. 1,3-Diethyl-2-thioxo-5-(2-(p-tolyl)hydrazono)dihydropyrimidine-4,6(1H,5H)-dione (4'm). Orange yellow powder, yield: 87% (52 mg; 0.2 mmol scale), mp 224–225 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.94 (br s, 1H, –NH), 7.47 (d, 2H, J = 8.0 Hz, Ar-H), 7.23 (d, 2H, J = 8.4 Hz, Ar-H), 4.60–4.51 (m, 4H, 2× –NCH₂CH₃), 2.36 (s, 3H, Ar-CH₃), 1.33–1.29 (m, 6H, 2× –NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.66 (S=C(–NC₂H₅)₂), 159.47 (–N(C₂H₅)CO–), 157.75 (–N(C₂H₅)CO–), 138.52 (C), 138.21 (C), 130.49 (2× CH), 117.65 (2× CH), 117.27 (C), 43.83 (–NCH₂CH₃), 42.65 (–NCH₂CH₃), 21.29 (Ar-CH₃), 12.48 (–NCH₂CH₃), 12.27 (–NCH₂CH₃) ppm. Elemental Anal. Calcd (%) for C₁₅H₁₈N₄O₂S: C, 56.58; H, 5.70; N, 17.60. Found: C, 56.69; H, 5.74; N, 17.69.

4.4.40. 1,3-Diethyl-5-(2-(2-methoxyphenyl))hydrazono)-2thioxodihydropyrimidine-4,6(1H,5H)-dione (4'n). Orange yellow powder, yield: 91% (61 mg; 0.2 mmol scale), mp 248–249 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.97 (br s, 1H, -NH), 7.97 (dd, 1H, J = 8.0 and 0.8 Hz, Ar-H), 7.27–7.23 (m, 1H, Ar-H), 7.08–7.04 (m, 1H, Ar-H), 6.98 (d, 1H, J = 8.4 Hz, Ar-H), 4.63–4.55 (m, 4H, 2× -NCH₂CH₃), 4.01 (s, 3H, Ar-OCH₃), 1.35–1.31 (m, 6H, 2× –NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.79 (S= C(-NC₂H₅)₂), 159.13 (–N(C₂H₅)CO–), 158.01 (–N-(C₂H₅)CO-), 149.40 (C), 130.10 (C), 128.31 (CH), 121.98 (CH), 118.13 (C), 117.16 (CH), 111.33 (CH), 56.29 (Ar-OCH₃), 43.88 (–NCH₂CH₃), 42.68 (–NCH₂CH₃), 12.57 (–NCH₂CH₃), 12.33 (–NCH₂CH₃) ppm. Elemental Anal. Calcd (%) for C₁₅H₁₈N₄O₃S: C, 53.88; H, 5.43; N, 16.75. Found: C, 53.75; H, 5.49; N, 16.87.

4.4.41. 1,3-Diethyl-5-(2-(4-methoxyphenyl)hydrazono)-2thioxodihydropyrimidine-4,6(1H,5H)-dione (**4'o**). Red powder, yield: 88% (59 mg; 0.2 mmol scale), mp 205–206 °C. ¹H NMR (400 MHz, CDCl₃): δ 15.08 (br s, 1H, –NH), 7.54 (d, 2H, J = 8.8 Hz, Ar-H), 6.95 (d, 2H, J = 8.8 Hz, Ar-H), 4.61– 4.51 (m, 4H, 2× –NCH₂CH₃), 3.83 (s, 3H, Ar-OCH₃), 1.33– 1.29 (m, 6H, 2× –NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.61 (S=C(–NC₂H₅)₂), 159.58 (–N-(C₂H₅)CO–), 159.56 (C), 157.86 (–N(C₂H₅)CO–), 134.34 (C), 119.26 (2× CH), 116.94 (C), 115.20 (2× CH), 55.76 (Ar-OCH₃), 43.79 (–NCH₂CH₃), 42.62 (–NCH₂CH₃), 12.48 (–NCH₂CH₃), 12.27 (–NCH₂CH₃) ppm. Elemental Anal. Calcd (%) for C₁₅H₁₈N₄O₃S: C, 53.88; H, 5.43; N, 16.75. Found: C, 53.74; H, 5.50; N, 16.66.

4.4.42. 1,3-Diethyl-5-(2-(4-(methylthio)phenyl)hydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'p). Red powder, yield: 81% (57 mg; 0.2 mmol scale), mp 185–187 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.97 (br s, 1H, -NH), 7.51 (d, 2H, J = 8.8 Hz, Ar-H), 7.28 (d, 2H, J = 8.4 Hz, Ar-H), 4.59–4.53 (m, 4H, 2× -NCH₂CH₃), 2.51 (s, 3H, Ar-SCH₃), 1.33–1.29 (m, 6H, 2× -NCH₂CH₃), ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.58 (S=C(-NC₂H₅)₂), 159.51 (-N(C₂H₅)CO-), 157.72 (-N(C₂H₅)CO-), 139.07 (C), 138.05 (C), 127.38 (2× CH), 118.14 (2× CH), 117.47 (C), 43.89 (-NCH₂CH₃), 42.72 (-NCH₂CH₃), 15.82 (Ar-SCH₃), 12.50 (-NCH₂CH₃), 12.29 (-NCH₂CH₃) ppm. Elemental Anal. Calcd (%) for C₁₅H₁₈N₄O₂S₂: C, 51.41; H, 5.18; N, 15.99. Found: C, 51.56; H, 5.13; N, 15.87.

4.4.43. 1,3-Diethyl-2-thioxo-5-(2-(4-((trifluoromethyl)-thio)phenyl)hydrazono)dihydropyrimidine-4,6(1H,5H)-dione (4'q). Bright yellow powder, yield: 83% (67 mg; 0.2 mmol scale), mp 231–232 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.78 (br s, 1H, -NH), 7.72 (d, 2H, J = 8.4 Hz, Ar-H), 7.62 (d, 2H, J = 8.8 Hz, Ar-H), 4.61–4.52 (m, 4H, 2× -NCH₂CH₃), 1.34–1.30 (m, 6H, 2× -NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.52 (S=C(-NC₂H₅)₂), 159.34 (-N-(C₂H₅)CO–), 157.31 (-N(C₂H₅)CO–), 142.88 (C), 138.06 (2× CH), 129.44 (-SCF₃, J_{CF}^{-1} = 306 Hz), 122.71 (C), 118.76 (C), 118.20 (2× CH), 44.04 (-NCH₂CH₃), 42.90 (-NCH₂CH₃), 12.48 (-NCH₂CH₃), 12.27 (-NCH₂CH₃) ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ -42.79 ppm. Elemental Anal. Calcd (%) for: C₁₅H₁₅F₃N₄O₂S₂: C, 44.55; H, 3.74; N, 13.85. Found: C, 44.69; H, 3.69; N, 13.77.

4.4.44. 1,3-Diethyl-2-thioxo-5-(2-(4-(trifluoromethoxy)phenyl)hydrazono)dihydropyrimidine-4,6(1H,5H)-dione (4'r). Yellow powder, yield: 85% (65 mg; 0.2 mmol scale), mp 176–178 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.86 (br s, 1H, –NH), 7.61 (d, 2H, *J* = 8.8 Hz, Ar-H), 7.30 (d, 2H, *J* = 8.8 Hz, Ar-H), 4.62–4.53 (m, 4H, 2× –NCH₂CH₃), 1.34–1.31 (m, 6H, 2× –NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.61 (S=C(-NC₂H₅)₂), 159.45 (–N(C₂H₅) CO–), 157.50 (–N(C₂H₅)CO–), 147.97 (C), 139.33 (C), 122.60 (2× CH), 118.79 (2× CH), 118.40 (C), 118.23 (C), 43.98 (–NCH₂CH₃), 42.83 (–NCH₂CH₃), 12.49 Article

 $(-NCH_2CH_3)$, 12.28 $(-NCH_2CH_3)$ ppm. ¹⁹F NMR (376 MHz, CDCl₃): δ -58.01 ppm. Elemental Anal. Calcd (%) for $C_{15}H_{15}F_3N_4O_3S$: C, 46.39; H, 3.89; N, 14.43. Found: C, 46.51; H, 3.93; N, 14.51.

4.4.45. 1,3-Diethyl-5-(2-(2-nitrophenyl)hydrazono)-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (**4**'s). Yellow powder, yield: 78% (59 mg; 0.2 mmol scale), mp 155–157 °C. ¹H NMR (400 MHz, CDCl₃): δ 15.79 (br s, 1H, –NH), 8.35 (d, 1H, J = 8.4 Hz, Ar-H), 8.29 (d, 1H, J = 8.0 Hz, Ar-H), 7.77– 7.74 (m, 1H, Ar-H), 7.38–7.34 (m, 1H, Ar-H), 4.61–4.56 (m, 4H, 2× –NCH₂CH₃), 1.35–1.31 (m, 6H, 2× –NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.55 (S= C(-NC₂H₅)₂), 158.01 (–N(C₂H₅)CO–), 157.22 (–N(C₂H₅) CO–), 137.36 (C), 136.46 (C), 136.23 (CH), 126.13 (CH), 125.95 (CH), 120.96 (C), 118.92 (CH), 44.11 (–NCH₂CH₃), 43.11 (–NCH₂CH₃), 12.53 (–NCH₂CH₃), 12.31 (–NCH₂CH₃) ppm. Elemental Anal. Calcd (%) for C₁₄H₁₅N₅O₄S: C, 48.13; H, 4.33; N, 20.05. Found: C, 48.02; H, 4.38; N, 20.14.

4.4.46. 5-(2-(4-Bromophenyl)hydrazono)-1,3-diethyl-2thioxodihydropyrimidine-4,6(1H,5H)-dione (4't). Brownish yellow powder, yield: 72% (55 mg; 0.2 mmol scale), mp 255–257 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.79 (br s, 1H, -NH), 7.54 (d, 2H, *J* = 8.8 Hz, Ar-H), 7.44 (d, 2H, *J* = 8.8 Hz, Ar-H), 4.59–4.49 (m, 4H, 2× -NCH₂CH₃), 1.32–1.29 (m, 6H, 2× -NCH₂CH₃) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.56 (S=C(-NC₂H₅)₂), 159.42 (-N(C₂H₅) CO-), 157.49 (-N(C₂H₅)CO-), 139.88 (C), 133.02 (2× CH), 120.92 (C), 118.99 (2× CH), 118.09 (C), 43.95 (-NCH₂CH₃), 42.80 (-NCH₂CH₃), 12.49 (-NCH₂CH₃), 12.28 (-NCH₂CH₃) ppm. Elemental Anal. Calcd (%) for: C₁₄H₁₅BrN₄O₂S: C, 43.87; H, 3.94; N, 14.62. Found: C, 43.73; H, 3.99; N, 14.69.

4.4.47. 5-(2-(2,5-Dichlorophenyl)hydrazono)-1,3-diethyl-2-thioxodihydropyrimidine-4,6(1H,5H)-dione (4'u). Yellow powder, yield: 86% (64 mg; 0.2 mmol scale), mp 264–266 °C. ¹H NMR (400 MHz, CDCl₃): δ 14.91 (br s, 1H, -NH), 8.04 (d, 1H, J = 2.0 Hz, Ar-H), 7.36 (d, 1H, J = 8.4 Hz, Ar-H), 7.17 (dd, 1H, J = 8.8 and 2.4 Hz, Ar-H), 4.62–4.54 (m, 4H, 2× $-NCH_2CH_3$), 1.35–1.31 (m, 6H, 2× $-NCH_2CH_3$) ppm. ¹³C{¹H} NMR (100 MHz, CDCl₃): δ 177.54 (S= C($-NC_2H_5$)₂), 159.08 ($-N(C_2H_5)CO-$), 157.24 ($-N-(C_2H_5)CO-$), 138.56 (C), 134.91 (C), 130.85 (CH), 127.47 (CH), 121.09 (C), 119.74 (C), 117.91 (CH), 44.09 ($-NCH_2CH_3$), 42.99 ($-NCH_2CH_3$), 12.53 ($-NCH_2CH_3$), 12.29 ($-NCH_2CH_3$) ppm. Elemental Anal. Calcd (%) for: C₁₄H₁₄Cl₂N₄O₂S: C, 45.05; H, 3.78; N, 15.01. Found: C, 45.17; H, 3.82; N, 15.12.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsomega.2c03073.

Scanned copies of respective ¹H NMR, ¹³C NMR, and DEPT-135 spectra for all the synthesized compounds 4a-4z and 4'a-4'u; ¹⁹F NMR (for fluorine atomcontaining molecules such as 4g, 4h, 4n, 4v, 4'g, 4'q, and 4'r), 2D NMR (for one representative compound, 4t), and HRMS (for representative compounds 4a, 4p, 4'a, and 4'k) spectra, and X-ray single crystallographic analysis for a representative entry, 5-(2-(2-4)) methoxyphenyl)hydrazono)pyrimidine-2,4,6-(1*H*,3*H*,5*H*)-trione (4d) (PDF)

CCDC 2170064 contains the supplementary crystallographic data for 4d that can be obtained free of charge from *The Cambridge Crystallographic Data Centre via* www.ccdc.cam.ac.uk/data_request/cif. Single X-ray crystallographic data for 4d (CIF)

NMR FID FAIR data, including the primary NMR FID files, for compounds 4a-4z and 4'a-4'u (ZIP)

AUTHOR INFORMATION

Corresponding Author

Goutam Brahmachari – Laboratory of Natural Products & Organic Synthesis, Department of Chemistry, Visva-Bharati (a Central University), Santiniketan, West Bengal 731 235, India; © orcid.org/0000-0001-9925-6281;

Email: brahmg2001@yahoo.co.in, goutam.brahmachari@ visva-bharati.ac.in

Authors

- Anindita Bhowmick Laboratory of Natural Products & Organic Synthesis, Department of Chemistry, Visva-Bharati (a Central University), Santiniketan, West Bengal 731 235, India
- Indrajit Karmakar Laboratory of Natural Products & Organic Synthesis, Department of Chemistry, Visva-Bharati (a Central University), Santiniketan, West Bengal 731 235, India

Complete contact information is available at: https://pubs.acs.org/10.1021/acsomega.2c03073

Notes

The authors declare no competing financial interest.

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REFERENCES

(1) (a) Holman, K. R.; Stanko, A. M.; Reisman, S. E. Palladiumcatalyzed cascade cyclizations involving C-C and C-X bond formation: strategic applications in natural product synthesis. Chem. Soc. Rev. 2021, 50, 7891-7908. (b) Zhang, Q.; Shi, B.-F. 2-(Pyridin-2yl)isopropyl (PIP) Amine: An Enabling Directing Group for Divergent and Asymmetric Functionalization of Unactivated Methylene C(sp3)-H Bonds. Acc. Chem. Res. 2021, 54, 2750-2763. (c) Cheng, L.-J.; Mankad, N. P. C-C and C-X coupling reactions of unactivated alkyl electrophiles using copper catalysis. Chem. Soc. Rev. 2020, 49, 8036-8064. (d) Cavedon, C.; Seeberger, P. H.; Pieber, B. Photochemical Strategies for Carbon-Heteroatom Bond Formation. Eur. J. Org. Chem. 2020, 2020, 1379-1392. (e) Li, M.; Hong, J.; Xiao, W.; Yang, Y.; Qiu, D.; Mo, F. Electrocatalytic Oxidative Transformation of Organic Acids for Carbon-Heteroatom and Sulfur-Heteroatom Bond Formation. ChemSusChem 2020, 13, 1661-1687. (f) Corma, A.; Leyva-Pérez, A.; Sabater, M. J. Gold-Catalyzed Carbon-Heteroatom Bond-Forming Reactions. Chem. Rev. 2011, 111, 1657-1712. (g) Hartwig, J. F. Carbon-heteroatom bond formation catalysed by organometallic complexes. Nature 2008, 455, 314.

(2) (a) Bosque, I.; Chinchilla, R.; Gonzalez-Gomez, J. C.; Guijarro, D.; Alonso, F. Cross-dehydrogenative coupling involving benzylic and allylic C-H bonds. Org. Chem. Front. 2020, 7, 1717-1742.
(b) Gandeepan, P.; Müller, T.; Zell, D.; Cera, G.; Warratz, S.; Ackermann, L. 3d Transition Metals for C-H Activation. Chem. Rev. 2019, 119, 2192-2452. (c) Wang, D.; Weinstein, A. B.; White, P. B.; Stahl, S. S. Ligand-promoted palladium-catalyzed aerobic oxidation reactions. Chem. Rev. 2018, 118, 2636-2679.

(3) (a) Ma, B.; Wu, P.; Wang, X.; Wang, Z.; Lin, H.-X.; Dai, H.-X. Efficient Synthesis of Spirooxindole Pyrrolones by a Rhodium(III)-Catalyzed C-H Activation/Carbene Insertion/Lossen Rearrangement Sequence. Angew. Chem., Int. Ed. 2019, 131, 13469-13473. (b) Zhang, G.; Fan, Q.; Zhao, Y.; Ding, C. Copper-Promoted Oxidative Intramolecular C-H Amination of Hydrazones to Synthesize 1H-Indazoles and 1H-Pyrazoles Using a Cleavable Directing Group. Eur. J. Org. Chem. 2019, 2019, 5801-5806. (c) Loup, J.; Dhawa, U.; Pesciaioli, F.; Wencel-Delord, J.; Ackermann, L. Enantioselective C-H Activation with Earth-Abundant 3d Transition Metals. Angew. Chem., Int. Ed. 2019, 58, 12803-12818. (d) Huang, X.; Chen, Y.; Zhen, S.; Song, L.; Gao, M.; Zhang, P.; Li, H.; Yuan, B.; Yang, G. Cobalt-Catalyzed Aerobic Cross-Dehydrogenative Coupling of C-H and Thiols in Water for C-S Formation. J. Org. Chem. 2018, 83, 7331-7340. (e) Yu, Q.; Yang, Y.; Wan, J.-P.; Liu, Y. Copper-Catalyzed C5-H Sulfenylation of Unprotected 8-Aminoquinolines Using Sulfonyl Hydrazides. J. Org. Chem. 2018, 83, 11385-11391.

(4) (a) Kim, Y. B.; Won, J.; Lee, J.; Kim, J.; Zhou, B.; Park, J.-W.; Baik, M.-H.; Chang, S. Ni-catalyzed intermolecular $C(sp^3)$ -H amidation tuned by bidentate directing groups. ACS Catal. 2021, 11, 3067–3072. (b) Wang, P.; Yang, Z.; Wu, T.; Xu, C.; Wang, Z.; Lei, A. Electrochemical oxidative $C(sp^3)$ -H/N-H cross-coupling for N-Mannich bases with hydrogen evolution. ChemSusChem 2019, 12, 3073–3077. (c) Džambaski, Z.; Bondžić, B. P. Dehydrogenative $C(sp^3)$ -H bond functionalization of tetrahydroisoquinolines mediated by organic oxidants under mild conditions. Org. Biomol. Chem. 2019, 17, 6420–6425. (d) Lu, Y.-S.; Yu, W.-Y. Cp*Rh(III)-Catalyzed Cross-Coupling of Alkyltrifluoroborate with α -Diazomalonates for C(sp3)-C(sp3) Bond Formation. Org. Lett. 2016, 18, 1350–1353.

(5) (a) Yang, Y.; Gao, W.; Wang, Y.; Wang, X.; Cao, F.; Shi, T.; Wang, Z. Recent advances in copper promoted inert C(sp3)-H functionalization. ACS Catal. 2021, 11, 967–984. (b) Jazzar, R.; Hitce, J.; Renaudat, A.; Sofack-Kreutzer, J.; Baudoin, O. Functionalization of Organic Molecules by Transition-Metal-Catalyzed C(sp3)– H Activation. Chem.—Eur. J. 2010, 16, 2654–2672.

(6) (a) Karmakar, I.; Brahmachari, G. Electrochemical and mechanochemical synthesis of dihydrofuro[3,2-c]chromenones via intramolecular Csp3-H cross-dehydrogenative oxygenation within warfarin frameworks: an efficient and straightforward dual approach. *Green Chem.* 2022, 24, 2825–2838. (b) Nayek, N.; Karmakar, P.; Mandal, M.; Karmakar, I.; Brahmachari, G. Photochemical and electrochemical regioselective cross-dehydrogenative C(sp2)-H sulfenylation and selenylation of substituted benzo[a]phenazin-5-ols. *New J. Chem.* 2022, 46, 13483–13497. (c) Brahmachari, G.; Bhowmick, A.; Karmakar, I. Visible light-driven and singlet oxygen-mediated photochemical cross-dehydrogenative C₃–H sulfenylation of 4-hydroxycoumarins with thiols using rose bengal as a photosensitizer. *J. Org. Chem.* 2021, 86, 9658–9669.

(7) (a) Brahmachari, G.; Karmakar, I.; Karmakar, P. Catalyst- and solvent-free Csp2-H functionalization of 4-hydroxycoumarins via C-3 dehydrogenative aza-coupling under ball-milling. *Green Chem.* 2021, 23, 4762–4770. (b) Matano, Y. Synthesis of aza-, oxa-, and thiaporphyrins and related compounds. *Chem. Rev.* 2017, 117, 3138–3191. (c) Yazdanbakhsh, M. R.; Ghanadzadeh, A.; Moradi, E. Synthesis of some new azo dyes derived from 4-hydroxy coumarin and spectrometric determination of their acidic dissociation constants. *J. Mol. Liq.* 2007, 136, 165–168. (d) Chohan, Z. H.; Shaikh, A. U.; Rauf, A.; Supuran, C. T. Antibacterial, antifungal and cytotoxic properties of novel N-substituted sulfonamides from 4-hydroxycoumarin. *J. Enzyme Inhib. Med. Chem.* 2006, 21, 741–748. (e) Palacios,

F.; Alonso, C.; Amezua, P.; Rubiales, G. Synthesis of Aza Polycyclic Compounds Derived from Pyrrolidine, Indolizidine, and Indole via Intramolecular Diels–Alder Cycloadditions of Neutral 2-Azadienes. *J. Org. Chem.* **2002**, *67*, 1941–1946.

(8) (a) Rathee, P.; Tonk, R. K.; Dalal, A.; Ruhil, M. K.; Kumar, A. Synthesis and application of thiobarbituric acid derivatives as antifungal agents. *Cell Mol. Biol.* **2016**, *62*, 1000141. (b) Sachar, A.; Gupta, P.; Gupta, S.; Sharma, R. L. Synthesis of some novel barbituric acid and 1,3-cyclohexanedione based condensed heterocycles. *Indian J. Chem., Sect. B* **2009**, *48*, 1187–1194. (c) Habib, N. S.; Soliman, R.; El-Tombary, A. A.; El-Hawash, S. A.; Shaaban, O. G. Synthesis of thiazolo[4,5-d]pyrimidine derivatives as potential antimicrobial agents. *Arch. Pharmacal Res.* **2007**, *30*, 1511–1520. (d) Thakur, R.; Mohan, R.; Kidwai, M. Ecofriendly synthesis of novel antifungal (thio)barbituric acid derivatives. *Acta Chim. Slov.* **2005**, *52*, 88–92. (e) Tozkoparan, B.; Ertan, M.; Kelicen, P.; Demirdamar, R. Synthesis and anti-inflammatory activities of some thiazolo[3,2-*a*]pyrimidine derivatives. *Farmaco* **1999**, *54*, 588–593.

(9) (a) Jana, A.; Bhaumick, P.; Panday, A. K.; Mishra, R.; Choudhury, L. H. I₂/DMSO mediated multicomponent reaction for the synthesis of 2-arylbenzo[d]imidazo[2,1-b] thiazole derivatives. *Org. Biomol. Chem.* **2019**, *17*, 5316–5330. (b) Mobinikhaledi, A.; Kalhor, M. Synthesis and biological activity of some oxo- and thioxopyrimidines. *Int. J. Drug Dev. Res.* **2010**, *2*, 268–272. (c) Mohamed, N. R.; El-Saidi, M. M. T.; Ali, Y. M.; Elnagdi, M. H. Utility of 6-amino-2-thiouracil as a precursor for the synthesis of bioactive pyrimidine derivatives. *Bioorg. Med. Chem.* **2007**, *15*, 6227– 6235. (d) Sharma, P.; Rane, N.; Gurram, V. K. Synthesis and QSAR studies of pyrimido[4,5-d]pyrimidine-2,5-dione derivatives as potential antimicrobial agents. *Bioorg. Med. Chem. Lett.* **2004**, *14*, 4185– 4190.

(10) (a) Ding, Y.; Li, H.; Meng, Y.; Zhang, T.; Li, J.; Chen, Q.-Y.; Zhu, C. Direct synthesis of hydrazones by visible light mediated aerobic oxidative cleavage of the C=C bond. Org. Chem. Front. 2017, 4, 1611–1614. (b) Krátký, M.; Bősze, S.; Baranyai, Z.; Stolaříková, J.; Vinšová, J. Synthesis and biological evolution of hydrazones derived from 4-(trifluoromethyl)benzohydrazide. Bioorg. Med. Chem. Lett. 2017, 27, 5185–5189. (c) Kauthale, S.; Tekale, S.; Damale, M.; Sangshetti, J.; Pawar, R. Synthesis, antioxidant, antifungal, molecular docking and ADMET studies of some thiazolyl hydrazones. Bioorg. Med. Chem. Lett. 2017, 27, 3891–3896.

(11) (a) Cvrtila, I.; Fanlo-Virgós, H.; Schaeffer, G.; Monreal Santiago, G. M.; Otto, S. Redox control over acyl hydrazone photoswitches. J. Am. Chem. Soc. 2017, 139, 12459–12465.
(b) Xiao, H.; Zhang, M.; Liu, J.; Han, Z.; Yang, L.; Wu, X. A novel rhodamine B fluorescent probe for Hg²⁺: synthesis and evaluation. Chin. J. Org. Chem. 2016, 36, 2413–2418. (c) Yang, Y.; Gao, C.-Y.; Liu, J.; Dong, D. Recent developments in rhodamine salicylidene hydrazone chemosensors. Anal. Methods 2016, 8, 2863–2871.
(d) Khattab, T. A.; Gaffer, H. E. Synthesis and application of novel tricyanofuran hydrazone dyes as sensors for detection of microbes. Color. Technol. 2016, 132, 460–465. (e) Su, X.; Aprahamian, I. Hydrazone-based switches, metallo-assemblies and sensors. Chem. Soc. Rev. 2014, 43, 1963–1981.

(12) Dyniewicz, J.; Lipiński, P. F. J.; Kosson, P.; Leśniak, A.; Bochyńska-Czyż, M.; Muchowska, A.; Tourwé, D.; Ballet, S.; Misicka, A.; Lipkowski, A. W. Hydrazone linker as a useful tool for preparing chimeric peptide/nonpeptide bifunctional compounds. *ACS Med. Chem. Lett.* **2017**, *8*, 73–77.

(13) (a) Huang, Z.; Wang, C.; Dong, G. A Hydrazone-Based exo -Directing-Group Strategy for β C–H Oxidation of Aliphatic Amines. Angew. Chem., Int. Ed. **2016**, 55, 5299–5303. (b) Chourasiya, S. S.; Kathuria, D.; Nikam, S. S.; Ramakrishnan, A.; Khullar, S.; Mandal, S. K.; Chakraborti, A. K.; Bharatam, P. V. Azine-hydrazone tautomerism of guanylhydrazones: evidence for the preference toward the azine tautomer. J. Org. Chem. **2016**, 81, 7574–7583. (c) Ros, A.; López-Rodríguez, R.; Estepa, B.; Álvarez, E.; Fernandez, R.; Lassaletta, J. M. Hydrazone as the directing group for Ir-catalyzed arene diborylations and sequential functionalizations. J. Am. Chem. Soc. **2012**, 134, 4573– 4576. (d) Job, A.; Janeck, C. F.; Bettray, W.; Peters, R.; Enders, D. The SAMP-/RAMP-hydrazone methodology in asymmetric synthesis. *Tetrahedron* **2002**, *58*, 2253–2329.

(14) Brahmachari, G. *Catalyst-Free Organic Synthesis*, 1st ed.; The Royal Society of Chemistry: Cambridge, London, 2017, (ISBN: 978-1-78262-412-7).

(15) (a) Dahiya, A.; Sahoo, A. K.; Alam, T.; Patel, B. K. tert-Butyl Nitrite (TBN), a Multitasking Reagent in Organic Synthesis. *Chem.*– *Asian J.* **2019**, *14*, 4454–4492. (b) Yedage, S. L.; Bhanage, B. M. tert-Butyl nitrite-mediated synthesis of *N*-nitrosoamides, carboxylic acids, benzocoumarins, and isocoumarins from amides. *J. Org. Chem.* **2017**, *82*, 5769–5781. (c) Maity, S.; Naveen, T.; Sharma, U.; Maiti, D. Stereoselective nitration of olefins with 'BuONO and TEMPO: direct access to nitroolefins under metal-free conditions. *Org. Lett.* **2013**, *15*, 3384–3387.

(16) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A. et al. *Gaussian 09*, Revision E.01; GAUSSIAN 09, Inc.: Wallingford CT, 2013.