

Methane reduction by quercetin, tannic and salicylic acids:

influence of molecular structures on methane formation and

fermentation in vitro

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SUPPLEMENTARY MATERIALS

Table S1. Compound-dependent parameters optimized by syringe infusion of pure standards. Declustering Potential (DP), Collision Energy (CE), and Cell Exit Potential (CEP).

Compounds	Q1 mass (m/z)	DP (V)	Q3 mass (m/z) ^a	CE (eV)	CXP (V)
Catechin/ Epicatechin	289.2	-125	108.8 /123.0	-29/-39	-8/-15
Catechin-2,3,4- ¹³ C ₃	292.0	-135	108.8	-30	-8
Rutin	609.2	-203	300.0 /270.9	-50/-69	-17/-22
Quercetin	300.9	-133	150.9 /106.9	-27/-38	-14/-11
Enterolactone-2,3,5- ¹³ C ₃	300.0	-128	191.9	-30	-14
Enterodiol-1,2,4- ¹³ C ₃	304.1	-140	255.1	-32	-17
Salicylic acid	136.9	-36	93.0	-20	-7

Salicylic acid-D ₄	140.9	-35	97.0	-21	-10
Gallic acid	168.9	-40	124.9	-20	-18

^a quantifier ions are given in bold.

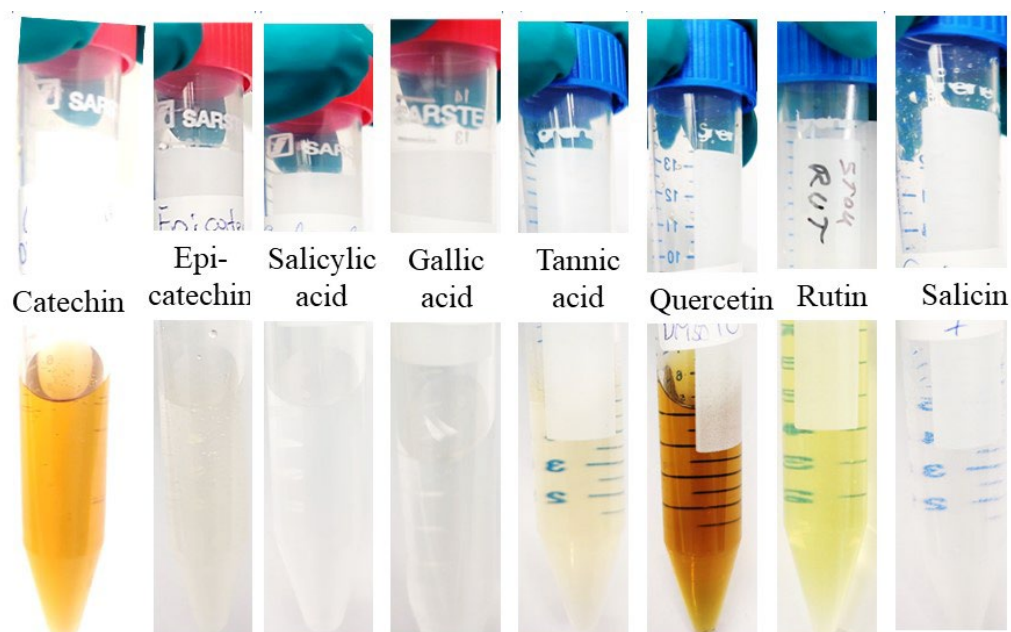


Figure S1. Plant Secondary Metabolites (PSMs) dissolved in either water or dimethyl sulfoxide (DMSO), 2 mL of each PSM was added to in vitro fermentation system.

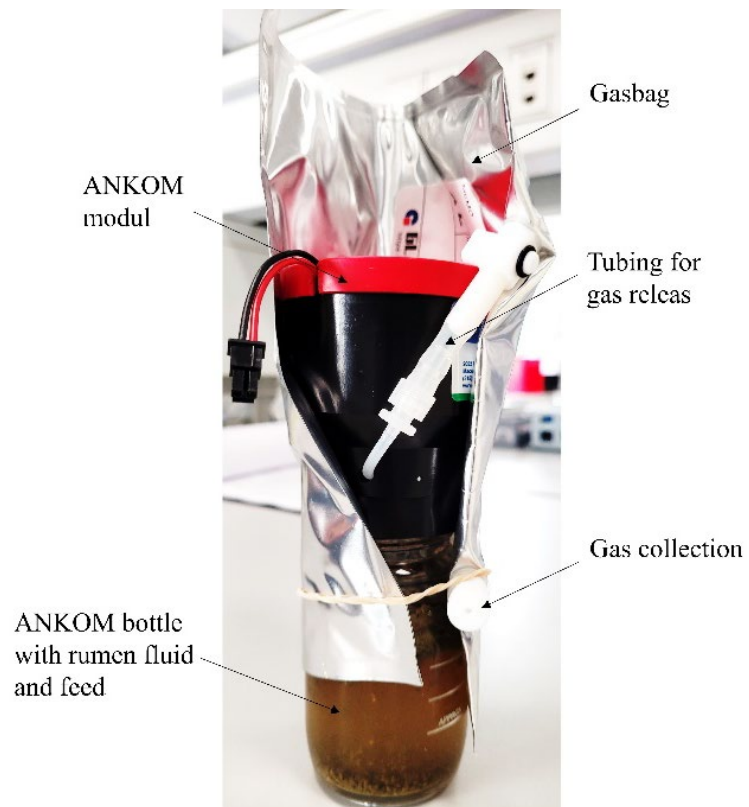


Figure S2. In vitro system simulating rumen fermentation (ANKOM®).