


Salicylate Intoxication in an Infant: A Case Report

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Abstract In children, the most common cause of an elevated anion gap (AG) with ketonemia, ketonuria, hyperglycemia, and glycosuria is diabetic ketoacidosis. However, when the clinical history is not clear, other causes must be considered. A 9-month-old girl was transferred to our pediatric intensive care unit (PICU) because of severe metabolic acidosis. On admission, she presented with *Kussmaul* breathing, tachycardia, irritability, and fever. Blood gasses revealed metabolic acidosis with superimposed respiratory alkalosis and elevated AG. Fluid replacement and bicarbonate for urine alkalinization were started. Ketonemia, acidic urine with glycosuria, ketonuria, and high blood glucose prompted an insulin infusion. Measurement of plasma salicylate confirmed toxic levels. When confronted, the parents admitted to accidentally preparing the child's bottle with water containing salicylic acid 1000 mg. Although the incidence of salicylate intoxication has declined, it remains an important cause of pediatric morbidity and mortality.

Key Points

As salicylate intoxication can manifest with hyperpnea, tachypnea, tachycardia, fever, coagulopathy, ketoacidosis, hyperglycemia, glycosuria, and ketonuria, it can be mistaken for other critical conditions that are more common in childhood, such as septicemia or diabetic ketoacidosis.

When an infant or child presents with increased anion gap (AG) metabolic acidosis and respiratory alkalosis, salicylate intoxication must be suspected and considered, even if caregivers deny any possibility of accidental poisoning.

Severe complications can appear when salicylate levels are decreasing or near therapeutic.

If criteria for severe intoxication (altered mental status, non-cardiogenic pulmonary edema, or non-responsive AG acidosis) are present, extracorporeal removal of salicylate should be performed.

Introduction

Metabolic acidosis is characterized by a decrease in serum pH that results from either a primary decrease in plasma bicarbonate concentration or an increase in hydrogen ion concentration [1]. It is divided into processes that are associated with a normal or elevated anion gap (AG) [2]. A high AG occurs when extra-unmeasured anions are added to the blood, and early recognition is important to allow the

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clinician to formulate a differential diagnosis and understand the underlying cause [1, 3, 4].

In children, the most common causes of increased AG include diabetic ketoacidosis, lactic acidosis, poisoning (alcohol, methanol, salicylates, etc.), and kidney failure. However, infants are more likely to develop conditions that occur with normal AG metabolic acidosis [5].

Between 1950 and 1970, salicylate intoxication was the most common cause of death by poisoning in children. Safety measures were undertaken, which contributed to near eradication of aspirin-related deaths after the 1990s [6]. In spite of this decrease, salicylate intoxication remains a problem, mostly because it is not common, which causes a delay in diagnosis and, consequently, higher mortality rates [7]. Early recognition and prompt effective treatment can successfully prevent morbidity and mortality [8].

We present a case report of salicylate intoxication, its clinical manifestations, metabolic effects, and potentially life-threatening complications. Despite its now lower incidence in the pediatric emergency department (ED), it is important that physicians be alert to this condition to ensure early identification, as presentation can be misleading and therefore associated with severe consequences.

Case Presentation

A healthy 9-month-old girl with Romanian mother and Portuguese father was brought to her local hospital ED because of a history of cough, stridor, and breathing difficulty. She had experienced multiple episodes of vomiting in the preceding 24 h and an isolated temperature elevation of 37.5 °C that responded to ibuprofen given by the mother. In the ED, the infant was treated with oral dexamethasone and nebulized adrenaline without improvement in symptoms. Laboratory evaluation was normal, and venous blood gases showed pH 7.36, pCO₂ 18.4 mmHg, bicarbonate 11.1 mmol/l, chlorine 125 mmol/l. Severe metabolic acidosis prompted transfer to our pediatric intensive care unit (PICU).

On PICU admission, the girl had *Kussmaul* breathing, tachycardia (205 pm), irritability, and fever (38.5 °C). The parents denied any possibility of accidental or intentional toxic ingestion. We proceeded to provide fluids and bicarbonate replacement.

Blood gases measured in the PICU were pH 7.37, pCO₂ 13.6 mmHg, bicarbonate 11.1 mmol/l, chlorine 129 mmol/l, base excess (BE) –17.5 mmol/l, AG 27 mmol/l, and lactate 36 mg/dl. Given the elevated lactate levels, along with the persistent hyperthermia (39 °C), lack of response to antipyretics, and sinus tachycardia (210 bpm), a fluid bolus was given and antibiotic therapy (ceftriaxone) started. Laboratory results showed no signs of infection.

Urinalysis revealed acidic urine (pH 5), glycosuria (67 mg/dl; normal range <15), and ketonuria (150 mg/dl). At first, capillary glycemia was normal (106 mg/dl), but she had ketonemia (4.5 mg/dl). After some hours, glycemia increased to 225 mg/dl, and therapy with fluid infusion plus intravenous insulin was started.

Investigations

Despite these therapeutic measures, compensated metabolic acidosis with low bicarbonate, elevated AG, and decreased BE persisted. The possibility of ingestion of an anion-forming substance was considered, and plasma salicylate levels revealed toxic values (76 mg/dl; normal range <20 mg/dl). The parents were confronted and finally admitted to accidentally preparing the child's formula with water containing Aspegic 1000 mg[®] (DL-lysine acetylsalicylate 1000 mg).

Differential Diagnosis

The differential diagnosis was sepsis, diabetic ketoacidosis, and other toxicities.

Treatment

After 23 h, salicylates decreased to 39.1 mg/dl, but the patient's clinical condition was worsening. She became oligo-anuric, with bilateral crackles on auscultation and low peripheral oxygen saturation. She had a self-limiting seizure, requiring no therapeutic intervention.

Treatment with continuous venovenous hemodiafiltration (CVVHDF) was started. Salicylate levels decreased to 10.3 mg/dl after 7 h of CVVHDF.

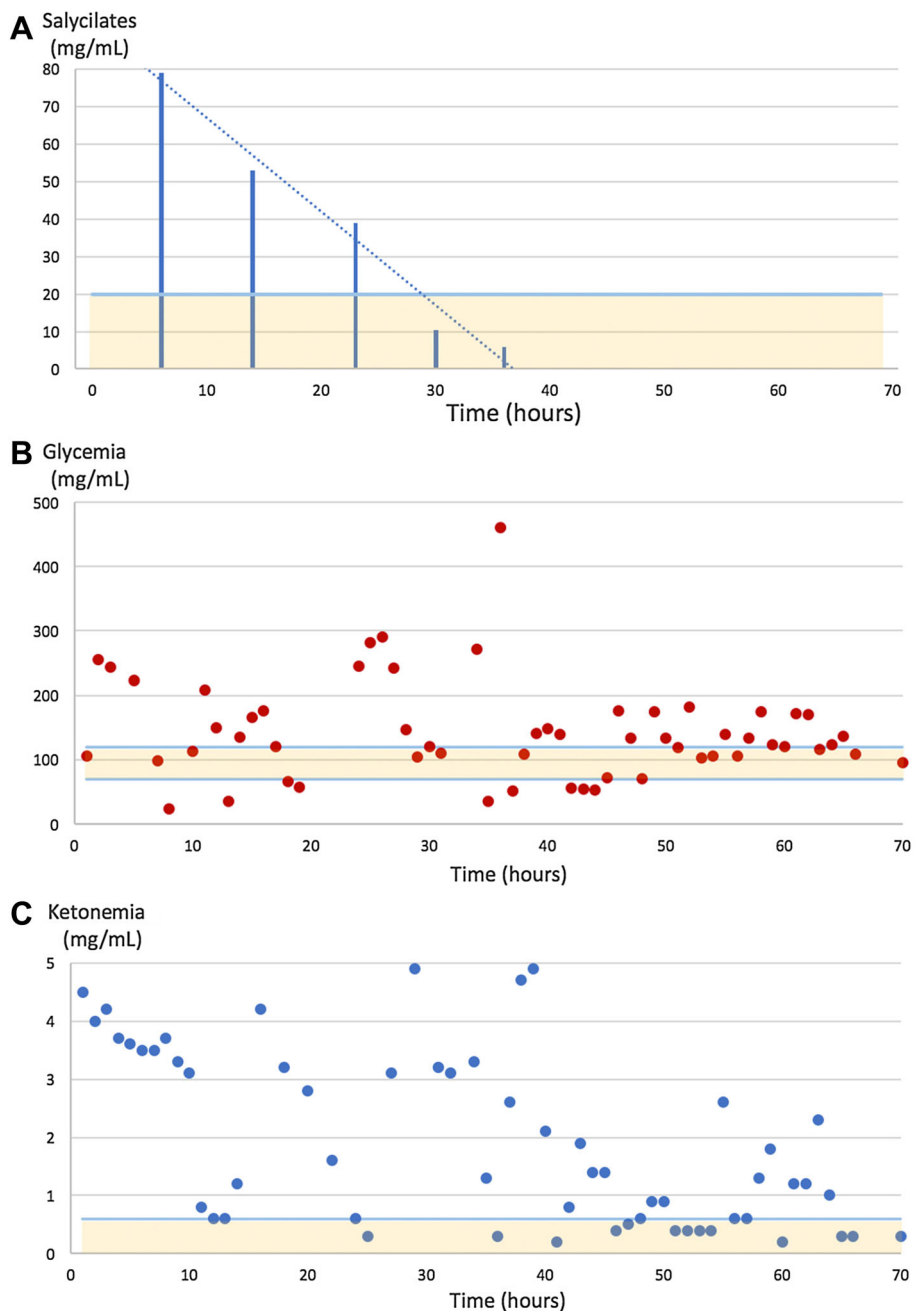
Therapy with bicarbonate replacement was maintained until 48 h of hospitalization, when urine pH was 9 and plasma salicylate level was 0.7 mg/dl.

Although the blood was negative for salicylates, unstable glycemia and ketonemia values persisted. Blood glucose levels oscillated between 36 and 460 mg/dl and blood ketone bodies between 0.6 and 4.9 mg/dl (Fig. 1). Insulin and fluid therapy were maintained for 72 h.

Outcomes and Follow-Up

On day 4 of the PICU stay, clinical improvement was noted, with apyrexia, normal glycemia, and negative ketonemia. She was transferred back to the referring

Fig. 1 Evolution over time of salicylate levels (a), glycemia (b), and ketonemia (c)



hospital where the local social services were contacted; she was discharged after 48 h of medical surveillance.

Discussion

Salicylates are pharmacological agents that can cause fatal poisoning in children. They are weak acids that, when unchanged in a metabolic acidosis environment, can move easily across cellular barriers, including the blood–brain barrier, and are responsible for tissue toxicity [9]. Manifestations are nonspecific and diverse, demanding a high

index of suspicion to avoid delayed diagnosis and treatment [10]. Salicylates stimulate the medullary respiratory center of the medulla, causing tachypnea and hyperventilation and therefore respiratory alkalosis. Consequently, they can cause nausea and vomiting via direct stimulation of the chemoreceptor trigger zone in the medulla. Respiratory alkalosis secondary to hyperventilation is critical to survival as it prevents central nervous system toxicity. On the other hand, it plays a role in the development of metabolic acidosis via uncoupling of oxidative phosphorylation in the mitochondria, which leads to accumulation of organic acids, including lactic acid and keto acids [10].

The hallmarks of acute salicylate overdose are hyperventilation leading to respiratory alkalosis, hyperthermia, and increased AG metabolic acidosis, which were all present in this clinical case.

Salicylate intoxication can be very misleading, especially when there is occult exposure, as in this case. The initial findings of tachypnea, high fever, high lactate levels, and metabolic acidosis mimicked sepsis [11]. The negative infection parameters in laboratory evaluations made this diagnosis less likely. As such, diabetic ketoacidosis was suspected, as it can have the same clinical presentation: hyperglycemia, AG metabolic acidosis, ketonuria, and ketonemia [12]. The degree of acidosis is less pronounced in salicylate intoxication because of simultaneous respiratory alkalosis. Salicylate intoxication induces complex alterations in carbohydrate metabolism, leading to hyperglycemia or hypoglycemia, particularly in infants. Hyperglycemia can result from glycogenolysis, stimulation of gluconeogenesis, and decreased peripheral utilization of glucose [12]. The alterations in glucose homeostasis can persist even when salicylate levels are undetectable, as in this case.

The Extracorporeal Treatments in Poisoning (EXTRIP) workgroup recommends bicarbonate as the first-line therapy [13]. Bicarbonate produces alkalemia and alkaluria. Alkalemia minimizes the passage of salicylates to the central nervous system, whereas alkaluria promotes renal excretion of salicylate [14]. It also involves correction of fluid and electrolyte imbalance. Urinalysis to monitor urine pH is important, as it is a good predictor of alkalinization success [15].

The EXTRIP workgroup recommendations for initiating extracorporeal treatment are salicylate levels > 100 mg/dl, salicylate levels > 90 mg/dl with impaired renal function, altered mental status, or new hypoxemia requiring supplemental oxygen [13].

Our patient's salicylate levels were below the reference level required to start extracorporeal treatment (76 mg/dl). However, in spite of decreasing salicylate levels, she was clinically deteriorating and progressed to severe salicylate intoxication (characterized by the presence of altered mental status and acute respiratory distress syndrome requiring supplemental oxygen). The development of any of these situations is an indication for extracorporeal removal of salicylates, which can be achieved with hemodialysis, CVVHDF, peritoneal dialysis, hemoperfusion, and transfusion exchange in newborns [14]. Extracorporeal treatment offers advantages over bicarbonate administration, including more rapid clearance of salicylates and a more predictable correction of acidosis [13]. After 7 h of treatment, the infant showed clinical improvement without further signs or symptoms.

Life-threatening complications of salicylate intoxication may occur when plasma concentrations are decreasing or near therapeutic, as happened in this case, where clinical deterioration occurred after 23 h of treatment [15]. Therefore, frequent plasma salicylate monitoring is not recommended, and clinical decisions should not be made solely based on this level [13].

Conclusion

Salicylate poisoning is a severe pediatric emergency and should be considered in children who present with unexplained metabolic acidosis and respiratory alkalosis. Prognosis in this disorder depends on early recognition and management.

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Compliance with Ethical Standards

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Conflict of interest Rita Espírito Santo, Sara Vaz, Filipa Jalles, Leonor Boto and Francisco Abecasis have no conflicts of interest that are directly relevant to the content of this report.

Consent Written informed consent was obtained from the parents of the patient for publication of this case report. A copy of the written consent may be requested for review from the corresponding author.

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