



Structures of Cyclic Organosulfur Compounds From Garlic (*Allium sativum* L.) Leaves

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Five new cyclic organosulfur compounds, foliogarlic disulfanes A_1 (1), A_2 (2), and A_3 (3) and foliogarlic trisulfane A_1 (4) and A_2 (5), were isolated from the leaves of *Allium* sativum (garlic). The chemical structures of these compounds were elucidated on the basis of physicochemical evidence including Nuclear Magnetic Resonance (NMR) and Mass Spectrometry (MS). Compounds 1–5 were obtained as complex compounds with disulfane or trisulfane and tetrahydro-2*H*-difuro[3,2-*b*:2',3'-*c*]furan-5(5a*H*)-one. In addition, the hypothetical biosynthetic pathways of these compounds were suggested.

Keywords: Allium sativum L., garlic, organosulfur compound, foliogarlic disulfane, foliogarlic trisulfane

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Allium plants (Allieae), such as garlic, onion, and chives, have been cultivated as not only foodstuffs but also medicinal plants in the worldwide from ancient. For example, the extract of Allium plants, such as garlic, has shown anticancer, antidiabetic, and antibacterial effects. In addition, the National Cancer Institute in the United States had focused on Allium species as expecting cancer prevention (Theisen, 2001). Allium plants are well-known to have various cysteine sulfoxide derivatives such as alliin, methiin, and propiin (Rose et al., 2005). The type and contents of cysteine sulfoxides were also known to be different among Allium species (Fritsch and Keusgen, 2006). The cysteine sulfoxides change to thiosulfinates, such as allicin (Cavallito and Bailey, 1944; Cavallito et al., 1944), by the reaction with enzyme called alliinase (Ellmore and Feldberg, 1994) when the tissues of Allium plants are broken. Allicin has been reported to have several biological effects (Gebhardt et al., 1994; Briggs et al., 2000; Cañizares et al., 2004; Oommen et al., 2004; Arditti et al., 2005). However, unstable thiosulfinates including allicin are changed to organosulfur compounds, such as ajoene (Block et al., 1984), methyl 1-(methylthio)ethyl disulfane, and 5,7-diethyl-1,2,3,4,6-pentathiepane (Kuo et al., 1990). These compounds are also comparatively unstable and volatile. Although ajoene was known to have significantly anticancer effect, the application as medicines is difficult. On the other hand, several cyclic organosulfur compounds with anticancer effects isolated from the bulbs of the Allium sativum (garlic) have been reported by Nohara et al. (2012, 2013, 2014). Thus, cyclic organosulfur compounds are important on the development of medicines including anticancer effect. On the basis of this background, we have isolated several comparatively stable organosulfur compounds from Allium fistulosum (green onion and welsh onion) (Fukaya et al., 2018, 2019a) and Allium schoenoprasum var. foliosum (Japanese chive) (Fukaya et al., 2019b). In the course of our ongoing research program for discovery of bioactive organosulfur compounds, the constituents from the leaves of Allium sativum were examined. In this article, we discuss the isolation and the structure elucidation of cyclic organosulfur compounds, foliogarlic disulfanes $A_1(1)$, $A_2(2)$, and $A_3(3)$ and foliogarlic trisulfane $A_1(4)$ and $A_2(5)$, from the leaves of A. sativum and the biosynthetic pathways.

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RESULTS AND DISCUSSIONS

The fresh leaves of *A. sativum* (15.0 kg) were mixed with water. Then, acetone was added into the mixture to be 80% acetone solution. The solution was concentrated after standing for 4 days (96 h) at room temperature. The acetone extract was portioned between ethyl acetate (EtOAc) and water. The organic fraction was evaporated *in vacuo* and obtained EtOAc fraction as syrup (41.98 g, 0.27% from the plant). The EtOAc fraction was also subjected with the normal and reversed-phase column chromatography and high-performance liquid chromatography (HPLC) to give foliogarlic disulfanes A₁ (1, 0.00013%), A₂ (2, 0.00021%), and A₃ (3, 0.00009%) and foliogarlic trisulfanes A₁ (4, 0.00015%) and A₂ (5, 0.00008%) (Figure 1).

Foliogarlic disulfanes A₁ (1) was obtained as yellow oil and showed positive optical rotation (+160.9). In the Electrospray Ionization MS (ESIMS) measurement of 1, a pseudomolecular ion peak [M + Na]⁺ was observed at m/z 343.0282, and the molecular formula was determined as C₁₂H₁₆O₆S₂ on the basis of the High Resolution ESIMS (HRESIMS) peak and the ¹³C NMR data. The ¹³C NMR spectra of 1 showed signals corresponding to a secondary methyl group at δ_C 9.1 (6-CH₃), a methine at δ_C 48.1 (C-6), a diastereotopic oxygen-bearing methylene at δ_C 74.98 (C-3); an oxygen-bearing methine at δ_C 75.03 (C-2); two methines neighboring the electron-withdrawing atom at δ_C

90.4 (C-3a) and $\delta_{\rm C}$ 103.2 (C-7); an oxygen-bearing quaternary carbon at δ_C 79.6 (C-5a); a two-oxygen-bearing quaternary carbon at $\delta_{\rm C}$ 119.1 (C-8a); and a lactone carbonyl carbon at $\delta_{\rm C}$ 175.4 (C-5) (Table 1, Figure 2, and Supplementary Material). The correlations of COSY double-quantum filter (DQF COSY) NMR spectroscopy were observed between 6-CH₃, H-6, and H-7 and between H-2, H-3, and H-3a (Figure 2). The heteronuclear multiple-bond correlation (HMBC) spectrum of 1 is shown in Figure 2. Namely, the correlation of H-2 to C-8a, H-7 to C-5a and H-3a to C-5a indicates acetal structure, the correlation of H-3a to C-5 and C-5a indicates a lactone, and the correlations of H-6 to C-5a and C-7, H-7 to C-5 and 6-CH₃, 6-CH₃ to C-5a, C-6, and C-7 indicate a secondary methyl moiety. These evidences indicate that compound 1 had a tetrahydro-2Hdifuro[3,2-b:2',3'-c]furan-5(5aH)-one skeleton. In addition, 1-propenyl disulfane structure at the side chain was confirmed by High Resolution MS (HRMS) and NMR (Nuclear Magnetic Resonance) data. Next, the NOESY spectrum of 1 showed key correlations between H-3a and H-6; and H-6 and H-7 (Figure 2). The results prove that the relative configurations among H-3, H-6, and H-7 were of the same orientation, respectively. Furthermore, the 1 H and 13 C NMR signals of 1 assigned to tetrahydro-2*H*-difuro[3,2-b:2',3'-c]furan-5(5a*H*)-one skeleton were superimposable on those of known compound, kujounin A₃, except for 1-propenyl disulfane moiety (Fukaya et al., 2019a). All the evidences support that the chemical structure of 1



Position	1		2		2	
	δ _H (J, Hz) ^a	δ_{c}^{a}	δ_H (J, Hz) ^a	δ_{c}^{a}	δ _H (J, Hz) ^b	δc ^b
2	4.05 (m)	75.03	4.00 (dd, J = 5.5, 7.5)	75.0	α 4.11 (dd, <i>J</i> = 4.8, 10.3)	74.2
			4.01 (dd, J = 5.5, 9.5)		β 4.23 (dd, J = 3.4, 10.3)	
3	4.30 (m)	74.98	4.30 (m)	75.5	4.44 (m)	74.4
За	4.62 (d-like)	90.4	4.60 (d, J = 2.5)	90.4	4.64 (d, <i>J</i> = 1.3)	88.6
5		175.4		175.0		171.4
5a		79.6		82.3		81.6
6	2.91 (m)	48.1	2.63 (m)	49.8	2.75 (m)	49.7
7	5.55 (d, 7.0)	103.2	4.78 (d, J = 7.0)	94.5	5.00 (d, J = 3.5)	97.5
8a		119.1		117.0		117.0
3′	3.45 (m)	43.5	3.47 (d, J = 7.5)	43.8	3.48 (m)	42.7
4′	5.88 (m)	134.6	5.86 (m)	134.4	5.88 (m)	133.0
5′	5.09 (d like, 10.0)	118.9	5.12 (d-like, <i>J</i> = 9.5)	119.1	5.15 (d-like, <i>J</i> = 11.6)	120.0
	5.19 (d like, 17.0)		5.18 (d-like, <i>J</i> = 16.0)		5.20 (d-like, <i>J</i> = 16.4)	
6-CH ₃	1.18 (d, 7.0)	9.1	1.13 (d, <i>J</i> = 7.5)	12.6	1.21 (d, <i>J</i> = 7.6)	14.0

^{a1}H NMR, ¹³C NMR (CD₃OD, 500 MHz).

^{b1}H NMR, ¹³C NMR (CDCl₃, 600 NMR).

was $(3S^*, 3aR^*, 5aS^*, 6R^*, 7R^*, 8aR^*)$ -3,5a-dihydroxy-6-methyl-7-(allyldisulfanyl)tetrahydro-2*H*-difuro[3,2-*b*:2',3'-*c*]furan-5(5a*H*)-one.

Foliogarlic disulfanes $A_2(2)$ and $A_3(3)$ were isolated as yellow oil with positive specific rotations (2: $[\alpha]_D^{25} + 139.0^\circ$ in MeOH) and negative specific rotations (3: $[\alpha]_D^{25} - 213.6^\circ$ in MeOH). In the ESIMS spectra of 2 and 3, the same quasi-molecular ion peaks (2 and 3: $[M+Na]^+$) were observed at m/z 343. The molecular formulas (2 and 3: C12H16O6S2) were determined on the basis of HRESIMS peaks at [2: m/z 343.0277, 3: m/z 343.0282 (calcd. 343.0281)] and the ¹³C NMR data. The ¹H and ¹³C NMR spectrum of 2 and 3 showed signals corresponding to a secondary methyl group, a methine, a diastereotopic oxygen-bearing methylene, and an oxygen-bearing methine (Tables 1, 2 and Figure 2). On the basis of this evidence and detailed examination of DQF COSY and HMBC experiments, the planner structures of 2 and 3 were found to be the same as that of 1. Next, the relative configurations of 2 and 3 were characterized by the detailed NOESY experiments. The NOESY spectrum of 2 showed key correlations between H-3a and 6-CH₃; and H-7 and 6-CH₃ (Figure 2). The NOESY spectrum of 3 showed key correlations between H-3a and H-6; and H-7 and 6-CH₃ (Figure 2). In addition, the ¹H and ¹³C NMR signals of 2 and 3 were superimposable on those of known compounds, kujounin A1 and A2, respectively, except for 1-propenyl disulfane structure (Fukaya et al., 2018). Consequently, the chemical structures of foliogarlic disulfanes A_2 (2) and A_3 (3) were determined (3S*,3aR*,5aS*,6S*,7R*,8aR*)-3,5a-dihydroxy-6-methylas 7-(allyldisulfanyl)tetrahydro-2*H*-difuro[3,2-*b*:2',3'-*c*]furan-5(5aH)-one and (3S*,3aR*,5aS*,6R*,7S*,8aR*)-3,5a-dihydroxy-6-methyl-7-(allyldisulfanyl)tetrahydro-2H-difuro[3,2-b:2',3'*c*]furan-5(5a*H*)-one, respectively.

Foliogarlic trisulfanes A_1 (4) and A_2 (5) were isolated as yellow oil with positive specific rotations (4: $[\alpha]_D^{25} + 124.6^\circ$

in MeOH) and negative specific rotations (5: $[\alpha]_{D}^{25} - 119.8^{\circ}$ in MeOH). In the ESIMS spectra of 4 and 5, the same quasimolecular ion peaks (4 and 5: [M+Na]⁺) were observed at m/z 375. The molecular formulas (4 and 5: $C_{12}H_{16}O_6S_3$) were determined on the basis of HRESIMS peaks at [4: m/z 374.9998, 3: m/z 374.0003 (calcd. 374.0001)], and the ¹³C NMR data. On the basis of the detailed analysis of the ¹H and ¹³C NMR, 2D-NMR (DQF COSY, HMBC, NOESY) spectrum of 4 and 5, the relative structures of tetrahydro-2H-difuro[3,2-b:2',3'c]furan-5(5aH)-one skeleton on 4 and 5 were found to be the same as those of 1 and 3, respectively (Tables 3, 4 and Figure 3). Next, the ¹H and ¹³C NMR spectrum at the side chain showed signals corresponding to an allyl group, as well as those of compounds 1-3. The determination of the sulfur linkage was confirmed by the HRMS spectrum. Namely, the pseudomolecular formula was established as C12H16O6S3Na. Therefore, compounds 4 and 5 were found to have a trisulfane bridge. Finally, the relative configurations of 4 and 5 were characterized by the comparison of ¹³C NMR data with 1 and 3 and the NOESY experiments. The ¹³C NMR signals of 4 and 5 were superimposable on those of 1 and 3. All the evidences supported that the chemical structures of 4 and 5 were (3*S**,3*aR**,5*aS**,6*R**,7*R**,8*aR**)-3,5*a*-dihydroxy-6-methyl-7-(allyltrisulfanyl)tetrahydro-2H-difuro[3,2-b:2',3'-c]furan-5(5aH)-one and (3S*,3aR*,5aS*,6R*,7S*,8aR*)-3,5a-dihydroxy-6-methyl-7-(allyltrisulfanyl)tetrahydro-2H-difuro[3,2-b:2',3'*c*]furan-5(5a*H*)-one, respectively.

The biological synthetic pathways for compounds 1– 5 are presumed. At first, allicin is generated from alliin by alliinase when plant tissues of *A. sativum* are broken. Next, allicin is decomposed into intermediates (a), (b), and (c) by hydrolysis and is reconstructed to disulfane (d) and trisulfane (e) (Jacob, 2006). Finally, the structure of tetrahydro-2*H*-difuro[3,2-*b*:2',3'-*c*]furan-5(5a*H*)-one



skeleton is formed from semidehydroascorbate by cyclization and sulfane formation with the intermediates d and e. Consequently, compounds 1-5 were presumed to be obtained (Figure 4).

rare compound derived from medicinal plants. The biological effects of these cyclic organosulfur compounds should be studied further.

CONCLUSION

Five new organosulfur compounds, foliogarlic disulfanes 1-3 and foliogarlic trisulfanes 4 and 5, were isolated from the leaves of *A. sativum*. These compounds 1-5 have a tetrahydro-2*H*-difuro[3,2-*b*:2',3'-*c*]furan-5(5a*H*)-one skeleton with methyl group at 6-position and 2-propenyl disulfane or 2-propenyl trisulfane group at 7-position. Particularly, foliogarlic trisulfanes 4 and 5 with a trisulfane moiety are a

EXPERIMENTAL

General

The following instruments were used to obtain physical data: specific rotations, a Horiba (Kyoto, Japan) SEPA-300 digital polarimeter (l = 5 cm); IR spectra, JASCO (Tokyo, Japan) FT/IR-4600 Fourier Transform Infrared Spectrometer; ESIMS, Agilent Technologies (CA, US) Quadrupole LC/MS 6130; HRESIMS, SHIMADZU LCMS-IT-TOF; ¹H NMR spectra, JEOL (Tokyo, Japan) JNM-LA 500 (500 MHz) spectrometer; ¹³C-NMR spectra,

TABLE 2 | ¹H NMR and ¹³C NMR data of 3.

Position	3					
	δ_H (J, Hz) ^a	δc ^a	δ_H (J, Hz) ^b	δc ^b		
2	4.04 (dd, J = 4.5, 10.0)	75.9	α 4.11 (dd, J = 4.1, 10.3)	75.4		
	4.07 (dd, J = 3.0, 10.0)		β 4.27 (dd, J = 1.4, 10.3)			
3	4.29 (m)	74.7	4.48 (m)	73.9		
За	4.75 (s-like)	89.9	4.85 (s-like)	87.5		
5		174.9		172.2		
5a		80.0		78.4		
6	2.69 (m)	47.9	2.71 (m)	46.5		
7	5.11 (d, J = 10.0)	96.3	5.16 (d, J = 9.6)	95.5		
8a		119.1		117.2		
3′	3.46 (d, <i>J</i> = 7.5)	44.1	3.44 (d, J = 7.6)	43.2		
4′	5.85 (m)	134.3	5.85 (m)	132.5		
5′	5.14 (d like, J = 10.0)	119.4	5.19 (d like, <i>J</i> = 10.3)	119.5		
	5.20 (d like, J = 16.5)		5.22 (d like, <i>J</i> = 15.8)			
6-CH ₃	1.09 (d, J = 6.5)	8.4	1.19 (d, <i>J</i> = 6.8)	7.9		

^{a1}H NMR, ¹³C NMR (CD₃OD, 500 MHz).

^{b1}H NMR, ¹³C NMR (CDCI₃, 600 NMR).

TABLE 3 | ¹H NMR and ¹³C NMR data of 4.

Position	4						
	δ_{H} (J, Hz) ^a	δc^a	δ _H (J, Hz) ^b	δc ^b			
2	4.07 (m)	75.7	4.28 (m)	75.0			
3	4.26 (m)	75.0	4.42 (m)	73.9			
За	4.66 (d, J = 1.5)	90.6	4.68 (s-like)	88.5			
5		175.1		170.2			
5a		79.5		78.0			
6	2.97 (m)	48.0	2.83 (m)	47.0			
7	5.68 (d, J = 7.0)	102.3	5.71 (d, <i>J</i> = 6.9)	99.5			
8a		119.1		119.0			
3′	3.59 (dd, J = 6.5, 13.0)	42.3	3.49 (m)	42.6			
	3.63 (dd, J = 7.5, 13.0)						
4′	5.86 (m)	134.3	5.86 (m)	132.5			
5′	5.17 (d-like, J = 10.0)	119.5	5.16 (d-like, <i>J</i> = 9.6)	119.6			
	5.22 (d-like, <i>J</i> = 17.0)		5.21 (<i>J</i> = 16.5)				
6-CH ₃	1.16 (d, <i>J</i> = 7.0)	9.2	1.28 (d, J = 6.9)	8.6			

^{a1}H NMR, ¹³C NMR (CD₃OD, 500 MHz).

^{b1}H NMR, ¹³C NMR (CDCl₃, 600 NMR).

JEOL JNM-LA 500 (125 MHz) spectrometer; NOESY spectra, JNM-ECA 600 (600 MHz) spectrometer; HPLC, a Shimadzu (Kyoto, Japan) SPD-20AVP UV-VIS detector. YMC-triart C18 (250 \times 4.6 mm i.d. and 250 \times 10 mm i.d.) and YMC-triart PFP (250 \times 4.6 mm i.d. and 250 \times 10 mm i.d.) columns were used for analytical and preparative purposes. The following experimental materials were used for chromatography: normal-phase silica gel column chromatography, silica gel BW-200 (Fuji Silysia Chemical, Ltd. (Aichi, Japan), 150–350 mesh); reversed-phase silica gel column chromatography, Cosmosil 140C₁₈-OPN

TABLE 4 | ¹H NMR and ¹³C NMR data of 5.

Position	5		
	δ_H (J, Hz) ^a	δc^a	
2	4.02 (m)	75.6	
3	4.25 (m)	75.1	
3a	4.71 (d, <i>J</i> = 1.5)	90.2	
5		175.0	
5a		80.1	
6	2.63 (m)	48.2	
7	5.23 (d, <i>J</i> = 9.5)	96.6	
8a		118.8	
3′	3.56 (m)	42.7	
4′	5.82 (m)	134.0	
5′	5.15 (d like, <i>J</i> = 10.0)	119.9	
	5.21 (d like, $J = 16.5$)		
6-CH ₃	1.11 (d, $J = 6.5$)	8.7	

^{a1}H NMR, ¹³C NMR (CD₃OD, 500 MHz).

[Nacalai Tesque (Kyoto, Japan)], TLC, precoated TLC plates with silica gel $60F_{254}$ [Merck (NJ, US), 0.25 mm] (ordinary phase), and silica gel RP-18 F_{254S} (Merck, 0.25 mm) (reversed phase); reversed-phase HPTLC, precoated TLC plates with silica gel RP-18 WF_{254S}. Detection was achieved by spraying with 1% Ce (SO₄) 2–10% aqueous H₂SO₄ followed by heating.

Plant Material

Fresh leaves of *A. sativum* cultivated in Kochi prefecture, Japan, were obtained as commercial products purchased from Japan Agricultural Cooperatives (JA) farmers' market (Kochi, Japan) in April 2017. The plants were identified by the authors (H.M. and S.N.).

Extraction and Isolation

The fresh leaves of A. sativum (15.2 kg) were chopped and mixed with water, and then acetone was added to the mixture to be 80% acetone solution. The mixture was soaked for 4 days (96 h) at room temperature. Evaporation of the filtrate under reduced pressure provided acetone extract (1,500.37 g, 9.87%). The extract was partitioned between EtOAc and H₂O (1:1, vol/vol) to obtain EtOAc fraction (41.98 g, 0.27%) and aqueous phase. The EtOAc-soluble fraction (41.98 g) was subjected to normal phase silica gel column chromatography [1,260 g, CHCl₃-MeOH (1:0 \rightarrow $100:1 \rightarrow 50:1 \rightarrow 30:1 \rightarrow 10:1 \rightarrow 0:1$, vol/vol)] to give nine fractions {Fr.1 (1,471.6 mg), Fr.2 (715.5 mg), Fr.3 (7,193.2 mg), Fr.4 (8,339.2 mg), Fr.5 (4,085.8 mg), Fr.6 (1,334.9 mg), Fr.7 (4,367.0 mg), Fr.8 (617.7 mg), Fr.9 (5,841.2 mg)}. r. 5 (4,085.8 mg) was further separated by reversed-phase silica gel column chromatography [200 g, MeOH-H₂O (2:8 \rightarrow 4:6 \rightarrow $6:4 \rightarrow$ $8:2 \rightarrow 1:0, \text{ vol/vol})$ to give 13 fractions {Fr.5-1 (52.3 mg), Fr.5-2 (21.0 mg), Fr.5-3 (31.0 mg), Fr.5-4 (19.1 mg), Fr.5-5 (84.4 mg), Fr.5-6 (281.7 mg), Fr.5-7 (43.5 mg), Fr.5-8 (26.3 mg), Fr.5-9 (67.9 mg), Fr.5-10 (482.9 mg), Fr.5-11 (2,637.7 mg), Fr.5-12 (178.7 mg), Fr.5-13 (16.9 mg)}. Fr.5-5 (84.4 mg) was



purified by HPLC {mobile phase: MeOH-H₂O (35:65, vol/vol) [YMC-triart PFP (250 \times 10 mm i.d.)]} to give **1** (6.0 mg) and **2** (16.3 mg). Fr.5-6 (281.7 mg) was purified by HPLC {mobile phase: MeOH-H₂O (50:50, vol/vol) [YMC-triart C18 (250 \times 10 mm i.d.)]} to give **1** (14.6 mg), **2** (16.0 mg), **4** (24.2 mg), and **5** (12.3 mg). Fr.5-6-5 (37.1 mg) was purified by HPLC {mobile phase: MeOH-H₂O (45:55, vol/vol) [YMC-triart C18 (250 \times 10 mm i.d.)]} to give **3** (14.7 mg) and **4** (4.5 mg).

Foliogarlic Disulfane A₁ (1)

Yellow oil; $[\alpha]_D^{25}$ +160.9 (MeOH); HRESIMS: calcd for $C_{12}H_{16}O_6S_2Na$ (M+Na)⁺: 343.0281, found: 343.0282; IR(ATR): 3,400, 2,975, 1,782 cm⁻¹; ¹H NMR (CD₃OD), ¹³C NMR (CD₃OD, 500 MHz): given in **Table 1**.

Foliogarlic Disulfane A₂ (2)

Yellow oil; $[\alpha]_D^{25}$ +139.0 (MeOH); HRESIMS: calcd for $C_{12}H_{16}O_6S_2Na$ (M+Na)⁺: 343.0281, found: 343.0277; IR(ATR): 3,400, 2,970, 1,785 cm⁻¹; ¹H NMR (CD₃OD, CDCl₃), ¹³C NMR (CD₃OD, CDCl₃); given in **Table 1**.

Foliogarlic Disulfane A₃ (3)

Yellow oil; $[\alpha]_D^{25}$ -213.6 (MeOH); HRESIMS: calcd for $C_{12}H_{16}O_6S_2Na$ (M+Na)⁺: 343.0281, found: 343.0282; IR(ATR): 3,400, 2,981, 1,785 cm⁻¹; ¹H NMR (CD₃OD, CDCl₃), ¹³C NMR (CD₃OD, CDCl₃): given in **Table 2**.

Foliogarlic Trisulfane A₁ (4)

Yellow oil; $[\alpha]_D^{25}$ +124.6 (MeOH); HRESIMS: calcd for $C_{12}H_{16}O_6S_3Na$ (M+Na)⁺: 375.0001, found: 374.9998; IR(ATR): 3,400, 2,975, 1,780 cm⁻¹; ¹H NMR (CD₃OD, CDCl₃), ¹³C NMR (CD₃OD, CDCl₃): given in **Table 3**.

Foliogarlic Trisulfane A₂ (5)

Yellow oil; $[\alpha]_D^{25}$ -119.8 (MeOH); HRESIMS: calcd for $C_{12}H_{16}O_6S_3Na~(M+Na)^+$: 375.0001, found: 375.0003; IR(ATR): 3,400, 2,931, 1,789 cm⁻¹; ¹H NMR (CD₃OD), ¹³C NMR (CD₃OD): given in **Table 4**.



DATA AVAILABILITY STATEMENT

All datasets generated for this study are included in the article/**Supplementary Material**.

AUTHOR CONTRIBUTIONS

MF: isolation of constituents and Structure elucidation of new compounds. SeiN: Structure elucidation of new compounds and overall supervision in this study. HH and DN: isolation of constituents. SouN: Preparation of plant extracts and overall supervision in this study. TY: Structure elucidation of new compounds. HM: overall supervision in this study.

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SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fchem. 2020.00282/full#supplementary-material

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Conflict of Interest: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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