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# Quantitative analysis of fragrance allergens in various matrixes of cosmetics by liquid–liquid extraction and GC–MS

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## Abstract

Fragrances are the most common chemicals in cosmetics to which people expose every day. However, the unwanted allergic reactions such as contact dermatitis caused by direct contact with fragrances may happen. In Directive 2003/15/EC of the EU, cosmetic product containing one or more of 26 fragrance allergens must be declared on the package label. In addition, commission regulation (EU) 2017/1410 amending Annexes II and III of cosmetic regulation 1223/2009 restricted fragrance chemical of methyl eugenol, and prohibited Lyrall, atranol, chloroatranol to be used in cosmetic. In this study, an efficient and sensitive GC–MS method for 3 banned fragrances, 26 fragrance allergens along with restricted methyl eugenol in cosmetics was established. Sample preparation by liquid–liquid extraction was developed by testing various solvent systems to simplify traditional complex extraction methodologies. Validation of the proposed method showed good linearities in a wide concentration ranges of 0.1–10 µg/mL. The intra-day and inter-day recoveries were between 84.4 and 119% with coefficient of variation (CV) below 13.5%. The limit of quantifications (LOQs) of 27 fragrance allergens were in the range of 2–20 µg/g. A surveillance study consisted with 82 cosmetics was conducted, among which 31 products claimed fragrance-free. The results showed some fragrance-free claims were false. In the other hand, there were seven cosmetics labeled containing Lyrall, but only four were detected. The top fragrance allergens detected in the samples were linalool, limonene, and geraniol. The analysis of fragrance allergens in cosmetics indicated that potential contact allergy related to these products should be considered, even though some fragrance allergens were from natural extracts, such as oak moss absolute.

**Keywords:** Allergen, Cosmetics, Fragrances, GC–MS, Liquid–liquid extraction

## 1. Introduction

Fragrance substances are derived from natural sources or chemical syntheses. They are organic compounds with pleasant smell, which are enormously used in perfumes and perfumed consumer goods such as cosmetics, detergents and other household products for the purpose of masking unpleasant odors from chemical ingredients [1]. Reports have demonstrated that fragrances in cosmetics are the most common allergens in human daily life [2,3], and may cause allergic contact

dermatitis, irritant contact dermatitis, photosensitivity dermatitis, urticaria, and asthma [4,5]. According to Directive 2003/15/EC of EU all cosmetics shall declare any of 26 fragrance allergens contained within the product if occurrences above 0.01% in leave-on and rinse-off products. In addition, the regulation (EU) No. 2017/1410 amending Annexes II (prohibited substances) and III (restricted substances) of Cosmetics Regulation 1223/2009 prohibits the use of Lyrall, atranol, chloroatranol, and restricts methyl eugenol [6,7]. Methyl eugenol is recognized as a human carcinogen, and may occur

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in natural herbal extracts [8]. The limitations of methyl eugenol derived from natural sources in cosmetics are described as following: not exceed 0.01% in fine fragrance, 0.004% in eau de toilette, 0.002% in a fragrance cream, 0.0002% in other leave-on products and in oral hygiene products, and 0.001% in rinse-off products.

Sample preparation for cosmetic analysis is crucial because complex matrixes such as high fat, emulsifier, and high solvent may seriously interfere in the determination of fragrances. Various approaches based on the different partition techniques such as liquid–liquid [10], liquid–solid [9], or liquid–gas [10] have been established for the extraction and cleaning. This study adapted liquid–liquid extraction method [11] and investigated various extraction solvents. An effective and sensitive method was developed and validated for the simultaneous determination of 23 restricted, 3 banned, and one restricted fragrance allergens in various types of cosmetic matrix. Surveillance consisted with 31 claimed fragrance-free and 51 perfumed cosmetics in various matrixes such as cream, lotion, shampoo, soaps, deodorants, shower gel, and perfumes purchased from commercial markets was analyzed and discussed.

## 2. Materials and method

### 2.1. Chemicals and samples

Reference standards amyl cinnamyl alcohol, benzyl alcohol, benzyl benzoate, and eugenol were purchased from USP (Rockville, MD, USA). Limonene, methyl-2-octynoate, cinnamyl alcohol, citronellol, citral,  $\alpha$ -isomethyl ionone, anisyl alcohol, hydroxy citronellol, geraniol, farnesol, linalool, Lilial®, Lylal®, benzyl salicylate, amyl cinnamyl, atranol, hexyl cinnamal, and benzyl cinnamate were purchased from Sigma–Aldrich (St. Louis, MO, USA). Isoeugenol was from AccuStandard (New Haven, USA). Cinnamal and coumarin were from Chem Service (West Chester, PA, USA). Chloroatranol was from Carbo-synth (Compton, UK). 4,4'-dibromobiphenyl was from Supelco (Bellefonte, PA, USA). 1,4-Dibromobenzene was from Chem Service (West Chester, PA, USA). A total of 82 cosmetic products including leave on and rinse-of products such as cream, lotion, shampoo, soaps, deodorants, shower gel, and perfumes, were collected from various commercial shops in Taiwan. Samples were stored at room temperature until use.

### 2.2. Standard solutions preparation

Stock solution of individual compounds was prepared by dissolving standard compound each 10 mg in 10 mL methyl tert-butyl ether, and further diluted into 10–100  $\mu\text{g}/\text{mL}$  with methyl tert-butyl ether. The calibration solutions were prepared by diluting standard solutions including internal standard solutions in either matrix solution or methyl tert-butyl ether to the final concentrations of 0.1–10  $\mu\text{g}/\text{mL}$ . Internal standards, 4,4'-dibromobiphenyl and 1,4-dibromobenzene based on EN16274 and a GC–MS method by IFRA [11], were prepared at concentration levels of 1  $\mu\text{g}/\text{mL}$  each. Two set of standard solutions of each fragrance compounds were utilized as calibration curves.

### 2.3. Sample extraction

Each sample 0.5 g was weighted into a 50 mL amber centrifuge tube. Deionized water 5 mL and 5 mL methyl tert-butyl ether was added. The tube was mixed by a Hulamixer® sample mixer (Thermo Fisher inc, Waltham, MA, US) for 30 min, and then water was removed by adding of 5 g anhydrous sodium sulfate before centrifuging for 30 min at  $3000 \times g$ . The supernatant was collected and filtered with a syringe filter. The filtrate 0.5 mL was added of 10  $\mu\text{L}$  internal standard solution and then dilute to 1 mL with methyl tert-butyl ether prior to analysis.

### 2.4. GC–MS separation

A GC–MS system consisted with a G188A auto-sampler, 7890A gas chromatograph, and G7080B single quadrupole mass selective detector (Agilent Technologies, Palo Alto, USA) was utilized. Separation was carried out on a vf-5ms capillary column ( $30\text{m} \times 0.25\text{ mm i.d.}, 0.25\text{ }\mu\text{m}$  film thickness, Agilent). Helium as carrier gas was set at a constant flow of 1.0 mL/min. Sample solution 2.0  $\mu\text{L}$  was injected in pulsed splitless mode. GC oven was ramped from 60 to 125  $^{\circ}\text{C}$  at 3  $^{\circ}\text{C}/\text{min}$ , 125–230  $^{\circ}\text{C}$  at 7  $^{\circ}\text{C}/\text{min}$ , and 230–300  $^{\circ}\text{C}$  at 20  $^{\circ}\text{C}/\text{min}$  respectively, and with initial and final hold of 2 and 5 min, respectively. Mass spectrometer was operated in selective ion monitor (SIM) mode. Table 1 presented the specific  $m/z$  of the target fragrance allergens and internal standards.

### 2.5. Method validation

Neat standard calibration curves were obtained by diluting standard solutions with methyl tert-

Table 1. Purities and selected fragments of fragrance allergens and internal standard compounds (IS).

Compounds	Purity (%)	CAS No.	*Quantifier and Qualifiers
Amylcinnamic aldehyde	98	122-40-7	*129, 117, 202
Anise alcohol	99.5	105-13-5	*138, 137, 109
Atranol	98.2	526-37-4	*151, 152, 106
Benzyl alcohol	100	100-51-6	*79, 107, 108
Benzyl benzoate	100	120-51-4	*105, 91, 212
Benzyl cinnamate	98.5	103-41-3	*131, 192, 193
Benzyl salicylate	99.1	118-58-1	*91, 92, 228
Chloroatranol	98.9	57074-21-2	*185, 186, 187
Cinnamic alcohol	98.7	104-54-1	*92, 134, 115, 105
Cinnamic aldehyde	97.6	104-55-2	*131, 132, 103
Citral: neral	98	5392-40-5	*69, 109, 119
Citral: geraniol	98	5392-40-5	*69, 94, 84
Citronellol	99	106-22-9	*69, 67, 81
Coumarin	99.5	91-64-5	*118, 146, 89
Eugenol	100	97-53-0	*164, 149, 131
Farnesol	98.3	4602-84-0	*69, 81, 93
Geraniol	99	106-24-1	*69, 93, 123
Hexylcinnamic aldehyde	97.6	101-86-0	*129, 145, 216
Hydroxycitronellal	98	107-75-5	*59, 71, 43
Isoeugenol	99.3	97-54-1	*164, 149, 131
Lilial	97.5	80-54-6	*189, 147, 204
Limonene	97	5989-27-5	*68, 93, 67
Linalool	99	78-70-6	*93, 121, 136
Lyrals 1	95	31906-04-4	*105, 136, 163
Lyrals 2	95	31906-04-4	*136, 105, 192
Methyl-2-octynoate	99.9	111-12-6	*95, 123, 79
Methyl eugenol	98	93-15-2	*178, 147, 163
$\alpha$ -Amylcinnamyl alcohol	100	101-85-9	*133, 115, 205, 204
$\alpha$ -Isomethyl ionone	91.8	127-51-5	*95, 123, 79
1,4-Dibromobenzene (IS)	—	106-37-6	*236, 238, 234
4,4'-Dibromobiphenyl (IS)	—	92-86-4	*312, 310, 314

butyl ether to final concentrations ranged between 0.1 and 10  $\mu\text{g/g}$ . Matrix-matched standard calibration curves were prepared in body lotion extract to final concentration between 0.1 and 10  $\mu\text{g/g}$  in accordance with the sample preparation procedure described in section 2.3. Body lotion contained complex fats and ingredients with low volatilities such as glycerin, caprylic/carlic tryglyceride, ethylhexyl stearate, cetearyl alcohol would be a suitable represented matrix for this study. Limits of quantitation (LOQs) of 27 fragrance allergens were assessed by adding standard compounds into blank matrixes at concentration ranges between 0.1 and 10.0  $\mu\text{g/g}$ . The LOQ was estimated as the lowest concentration of analyte that can be quantified with the suitable precision and accuracy using a criteria of S/N ratio over 10. The intra/inter-day accuracy (recovery in %) and precision (RSD in %) were assessed by spiking two concentration levels of analytes in 5 replicates. Matrix effects were estimated by comparing the responding area of the

analytes between in neat solvent and in the matrix [12] which was calculated by the following formula:

$$\text{Matrix effect (ME)} = \frac{\text{area of (analyte in solvent)} - \text{analyte in matrix}}{\text{analyte in solvent}} \times 100\%$$

### 3. Results and discussion

#### 3.1. Liquid–liquid extraction

Preliminary tests of extracting fragrance allergens in spiked blank cosmetic by acetone, methanol, and acetonitrile showed interferences and low recoveries. Further, tests of liquid–liquid extraction (LLE) were applied and the recoveries of the fragrance allergens were determined in test solutions consisting of pre-spiked fragrance allergens in blank cosmetic. After clean up, recoveries were determined by the formula described as following. Recovery (%) = (peak area of analyte in pre-spiked extract/peak area of analyte in post-spiked extract)\*100%. The results showed methyl tert-butyl ether/

water partition offered better extraction results over hexane/water, methanol/hexane and acetonitrile/hexane (Table S1). The results of extremely nonpolar/polar solvent system such as hexane/

water were not satisfied due to anise alcohol, benzyl alcohol, benzyl salicylate, and hydroxycitronellal were relatively polar compounds. The replacement of water to methanol and acetonitrile (lower

Table 2. Validation parameters of the method.

Compounds	Linear range ( $\mu\text{g/mL}$ )	$r^2$	LOQ ( $\mu\text{g/g}$ )	Spiked level ( $\mu\text{g/g}$ )	Recovery		CV	
					Intra-day (%)	Inter-day (%)	Intra-day (%)	Inter-day (%)
Amylcinnamic aldehyde	0.1–1	0.998	2	2	99.7	93.7	1.9	7.3
				4	98.1	97.0	0.4	4.3
Anise alcohol	0.1–1	0.997	2	2	119	116	3.9	3.4
				4	109	103	7.3	7.7
Atranol	0.1–1	0.999	2	2	107	107	6.8	4.2
				4	102	107	2.3	3.9
Benzyl alcohol	0.1–1	0.998	2	2	100	93.9	5.2	9.5
				4	101	96.9	5.0	5.8
Benzyl benzoate	0.1–1	0.998	2	2	84.5	87.2	0.7	4.4
				4	95.6	96.1	1.2	2.9
Benzyl cinnamate	0.1–1	0.998	2	2	85.8	85.1	6.9	9.0
				4	94.6	94.3	0.3	3.1
Benzyl salicylate	0.1–1	0.995	2	2	106	96.0	4.2	12
				4	104	103	8.5	6.4
Chloroatranol	0.1–1	0.998	2	2	102	104	5.1	4.9
				4	102	100	2.7	3.2
Cinnamic alcohol	0.1–1	0.998	2	2	115	109	3.7	5.1
				4	113	109	4.0	5.9
Cinnamic aldehyde	0.1–1	0.999	2	2	100	96.4	0.9	7.5
				4	100	99.5	0.8	4.4
Citral	0.1–1	0.998	2	2	86.1	92.3	8.1	7.2
				4	90.9	102	5.7	8.9
Citronellol	0.1–1	0.996	2	2	95.1	99.5	12	14
				4	93.8	102	6.3	11
Coumarin	0.1–1	0.997	2	2	89.8	88.1	0.7	4.1
				4	96.8	96.4	0.4	3.3
Eugenol	0.1–1	0.998	2	2	105	109	2.9	6.0
				4	104	109	1.9	5.9
Farnesol	0.5–5	0.995	10	10	103	99.2	11	7.2
				20	95.2	98.4	5.4	11
Geraniol	1–10	0.995	20	20	114	110	1.0	4.3
				40	90.6	92.7	4.0	3.4
Hexylcinnamic aldehyde	0.1–1	0.998	2	2	98.9	96.7	1.5	5.2
				4	99.9	99.4	0.6	3.6
Hydroxycitronellal	0.1–1	0.999	2	2	105	97.6	2.9	12
				4	99.2	96.0	2.1	6.1
Isoeugenol	0.1–1	0.998	2	2	107	105	2.4	5.1
				4	102	102	2.1	5.5
Lilial	0.1–1	0.998	2	2	90.1	87.1	0.8	4.9
				4	95.9	95.0	0.5	3.3
Limonene	0.1–1	0.998	2	2	90.1	87.7	0.8	5.5
				4	92.4	95.5	4.9	5.5
Linalool	0.1–1	0.999	2	2	97.8	93.7	0.9	7.6
				4	92.6	97.5	0.9	4.1
Lylal	0.1–1	0.999	2	2	101	93.1	3.6	7.4
				4	93.9	92.6	1.8	7.5
Methyl-2-octynoate	0.1–1	0.997	2	2	104	98.5	1.6	10
				4	102	99.2	7.8	6.9
Methyl eugneol	0.1–1	0.998	2	2	96.5	94.2	0.9	5.9
				4	99.8	99.3	0.6	3.5
$\alpha$ -Amylcinnamyl alcohol	0.1–1	0.999	2	2	98.5	97.8	1.5	4.4
				4	98.4	98.8	2.0	3.6
$\alpha$ -Isomethyl ionone	0.1–1	0.998	2	2	99.6	96.8	1.0	5.9
				4	99.9	99.1	0.7	3.8

polarities to water) in the liquid–liquid system did improve the recoveries of polar compounds, but some compounds such as cinnamic alcohol, methyl-2-octynoate, limonene, and  $\alpha$ -isomethyl ionone showed decreased recoveries, due to these compound were immiscible in methanol and acetonitrile. Hence, in the LLE partition system, water was remained and hexane was replaced to methyl tert-butyl ether (relatively higher polarity to hexane). This methyl tert-butyl ether/water system showed excellent recovery rates over hexane methanol, acetonitrile, acetone, water, and their mixtures. The water in the extraction system was further removed by sodium sulfate anhydrous. Chromatogram (Fig. S1) showed there was no significant interfere of 27 analytes in a lotion matrix. LLE offered fast sample preparation and removed most of fats by relatively high polar solvent system. The recoveries of analytes obtained were acceptable. Therefore, LLE would be a suitable methodology for preparing sample for GC/MS in complex cosmetic matrix.

### 3.2. Method validation

Fragrance allergens in total 27 compounds (24 of 26 EU fragrances, in which 2 are natural extracts; 2 of 3 banned fragrances, in which one is overlapped with 26 EU fragrances; one restricted fragrance, methyl eugenol) were evaluated at the

concentration ranges of 0.1–10.0  $\mu\text{g/mL}$  (0.1, 0.2, 0.5, 0.7, 1, 2, 5, 7, and 10  $\mu\text{g/mL}$ ) with the internal standards at 1  $\mu\text{g/mL}$  in duplicates. Good linearity was achieved at the concentrations of 1–10  $\mu\text{g/mL}$  for geraniol, 0.5–5  $\mu\text{g/mL}$  for farnesol, and 0.1–1  $\mu\text{g/mL}$  for other 25 fragrance allergens in this study. The coefficient of determination ( $r^2$ ) were all higher than 0.995. The LOQs determined for farnesol was 10  $\mu\text{g/g}$ ; for geraniol was 20  $\mu\text{g/g}$ ; for other 25 fragrance allergens were 2  $\mu\text{g/g}$ . In recovery studies, the intra-day accuracies of fragrance allergens were between 84.5 and 119%, while the precision (RSD) located in the range between 0.4 and 12%. The inter-day accuracies were obtained between 85.1 and 116% with precision between 2.9 and 13% (see Table 2).

### 3.3. Matrix effect in GC analysis

Matrix effects of analytes were shown in Table 3. Significant signal enhanced or suppressed results were observed in most fragrance allergens in this study, suggesting there were either matrix enhancement or suppression for analytes in GCMS analysis of cosmetics. It could be the fatty matrix and some polar ingredients such as glycerin co-eluted with analytes. Therefore, matrix-matched calibration curves were suggested for the analysis of various types cosmetics.

Table 3. Matrix effects of fragrances in cosmetics by GCMS analysis.

Compounds	Equation of the solvent-only calibration curve	Equation of the matrix-matched calibration curves	Matrix effect (%)
Amylcinnamic aldehyde	$y = 0.7802 x + 0.0201$	$y = 0.5789 x + 0.0149$	–25.8
Anise alcohol	$y = 0.6286 x - 0.0608$	$y = 0.8491 x - 0.1057$	35.1
Atranol	$y = 0.1994 x - 0.0367$	$y = 0.5844 x - 0.0261$	193.0
Benzyl alcohol	$y = 0.6550 x - 0.0151$	$y = 0.7803 x - 0.0112$	19.1
Benzyl benzoate	$y = 1.5441 x - 0.0517$	$y = 1.2172 x + 0.0570$	–21.2
Benzyl cinnamate	$y = 0.3606 x - 0.0660$	$y = 0.5876 x + 0.0232$	63.0
Benzyl salicylate	$y = 1.9756 x - 0.3642$	$y = 0.8054 x + 0.0796$	–59.2
Chloroatranol	$y = 0.1714 x - 0.0333$	$y = 0.3927 x - 0.0073$	129.0
Cinnamic alcohol	$y = 0.0606 x - 0.0111$	$y = 0.1591 x - 0.0320$	162.5
Cinnamic aldehyde	$y = 0.9004 x - 0.0497$	$y = 1.0459 x - 0.0068$	16.2
Citral	$y = 0.0012 x - 0.0454$	$y = 0.0015 x + 0.0055$	25.0
Citronellol	$y = 0.6078 x - 0.0003$	$y = 0.1701 x - 0.0332$	–72.0
Coumarin	$y = 0.7988 x + 0.0474$	$y = 0.5863 x + 0.0280$	–26.6
Eugenol	$y = 0.6557 x - 0.0373$	$y = 0.5913 x - 0.0144$	–9.8
Farnesol	$y = 0.7961 x - 0.0599$	$y = 0.3247 x - 0.1443$	–59.2
Geraniol	$y = 2.0415 x - 0.1315$	$y = 1.9945 x - 0.0355$	–2.3
Hexylcinnamic aldehyde	$y = 0.6486 x - 0.3410$	$y = 0.4884 x - 0.0005$	–24.7
Hydroxycitronellal	$y = 0.7330 x - 0.0490$	$y = 0.8613 x - 0.0440$	17.4
Isoeugenol	$y = 0.1946 x - 0.0257$	$y = 0.2172 x - 0.0121$	11.6
Lilial	$y = 0.6405 x + 0.0089$	$y = 0.4649 x + 0.0192$	–27.4
Limonene	$y = 0.65815 x + 0.0219$	$y = 0.6985 x + 0.0107$	6.1
Linalool	$y = 0.52964 x - 0.0115$	$y = 0.4322 x + 0.0003$	–18.4
Lylal	$y = 0.0130 x - 0.0007$	$y = 0.0113 x - 0.0055$	–13.0
Methyl-2-octynoate	$y = 0.2520 x - 0.0156$	$y = 0.3088 x - 0.0038$	22.5
Methyl eugenol	$y = 1.2359 x + 0.0923$	$y = 0.8935 x + 0.0341$	–27.7
$\alpha$ -Amylcinnamyl alcohol	$y = 0.4150 x - 0.0602$	$y = 0.5491 x - 0.0226$	32.3
$\alpha$ -Isomethyl ionone	$y = 0.9301 x + 0.0455$	$y = 0.6829 x + 0.0037$	–26.6

Table 4. Contents of fragrances in the products.

Compounds	Leave-on Products (n = 21, lotion, cream)			Leave-on Products (n = 14, perfume, deodorant)			Rinse-off products (n = 16, shampoo, shower gel)		
	N	Conc.(µg/g)	Mean (µg/g)	N	Conc.(µg/g)	Mean (µg/g)	N	Conc.(µg/g)	Mean (µg/g)
Amylcinnamic aldehyde	1	47	47	2	81–752	416	1	48	48
Anisyl alcohol	0			0			0		
Atranol	0			0			0		
Benzyl alcohol	7	4–6044	1976	7	9–294	128	5	5–3510	74
Benzyl benzoate	7	17–3594	676	7	58–4699	856	3	33–51	40
Benzyl cinnamate	0			2	53–108	80	0		
Benzyl salicylate	6	9–2390	494	6	20–13973	3346	5	16–276	135
Chloroatranol	0			0			0		
Cinnamic alcohol	1	97	97	4	102–1789	775	0		
Cinnamic aldehyde	0			1	21	21	0		
Citral	6	2–301	145	11	7–196	94	5	3–117	59
Citronellol	12	4–321	70	11	4–8100	2200	6	7–34	21
Coumarin	8	14–249	128	10	14–4535	559	5	40–246	113
Eugenol	2	15–220	117	4	40–200	106	5	9–1017	425
Farnesol	3	- <sup>a</sup>	- <sup>a</sup>	4	- <sup>a</sup>	- <sup>a</sup>	0		
Geraniol	13	69–445	269	12	43–3688	914	9	68–1827	425
Hexylcinnamic aldehyde	3	75–1038	417	5	75–17868	4664	6	93–533	237
Hydroxycitronellal	5	24–196	86	9	83–4040	926	7	12–498	263
Isoeugenol	1	15	15	5	36–130	79	1	43	43
Lilial®	4	45–3927	1048	2	7650–15305	11477	4	134–3958	1331
Limonene	19	5–19092	1687	13	84–5603	1748	13	4–14798	2492
Linalool	20	12–862	252	14	7–6574	1557	13	36–1784	436
Lyrall®	2	1–2	2	1	204	204	1	1787	1787
Methyl-2-octynoate	2	2–3	3	0			0		
Methyl eugenol	0			1	- <sup>a</sup>	- <sup>a</sup>	0		
α -Amylcinnamyl alcohol	0			0			0		
α-Isomethyl ionone	4	11–116	59	9	105–4124	1822	5	41–493	163

<sup>a</sup> Detected but concentration below LOQ.

#### 3.4. Cosmetic labeling review and contents of fragrances of 51 non fragrance-free products

Fifty one cosmetics labeled containing fragrance ingredients were examined including 35 leave-on (21 lotion, cream, 12 perfume and 2 deodorants) and 16 rinse-off products. The most frequently identified fragrances were linalool (91.4%), limonene (85.7%), geraniol (80.0%), citronellol (71.4%) and coumarin (51.4%) in leave-on products. Limonene (68.8%), linalool (62.5%), coumarin (43.8%), citronellol (37.5%), geraniol (31.3%) and benzyl salicylate (31.3%) were the most frequently identified fragrances in rinse-off products. The labeling check results were similar to a previous study which investigated 283 cosmetic labels in Italy [13].

Contents of the target fragrance allergens in the cosmetics were quantitated by matrix-matched calibration and described in Table 4. Fragrances were present in 74% (20/27) of the lotions and creams, mostly linalool (95%), limonene (90%), geraniol (60%), citronellol (55%) and benzyl benzoate (45%). Perfums and deodorants showed a rate of 82% (22/27) and most frequently identified were linalool (100%), limonene (86.67%), geraniol (80%), coumarin

(73.33%) and citral (73.33%). Fragrance allergens were detected more often in perfumes and deodorants. The most abundant fragrances was limonene (19092 µg/g) in leave-on products, followed by hexyl cinnamic aldehyde (17868 µg/g) and Lilial (15303 µg/g). According to IFRA standards, hexyl cinnamic aldehyde and Lilial were limited in perfumes as 10.7% and 1.86%, respectively. Most products were clearly and properly labeled, except one cosmetic and one perfume in which limonene (51 µg/g), benzyl alcohol (6044 µg/g), linalool (34 µg/g, 12 µg/g) and benzyl benzoate (59 µg/g) were found exceed the 0.001% (10 µg/g) limitation, and linalool (211 µg/g) and hydroxycitronellal (485 µg/g) were exceed the 0.01% (100 µg/g) limitation. Methyl eugenol, a restricted fragrance, was found in one perfume sample, but the concentration was below LOQ.

#### 3.5. Contents of fragrances in claimed fragrance-free cosmetics

In 31 fragrance-free commercial cosmetics 5 samples were found fragrances. Limonene (1500 µg/g), linalool (29 µg/g), geraniol (71 µg/g) and citronellol (28 µg/g) were found in a hair conditioner sample. A





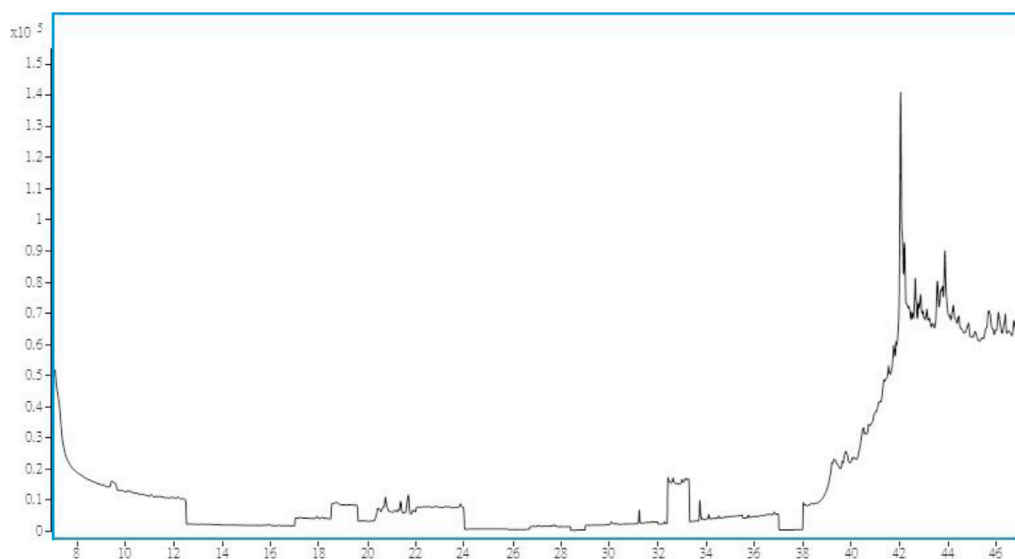


Fig. S2. GC–MS chromatogram (SIM) of the lotion matrix.

Table. S1. Evaluation the extraction efficiency using different organic solvents.

Compounds	MTBE/ water (%)	hexane/ water (%)	methanol/ hexane (%)	acetonitrile/ hexane (%)
Amylcinnamic aldehyde	97.31	96.16	89.56	73.44
Anise alcohol	86.39	12.80	78.79	77.08
Atranol	103.44	92.98	83.30	88.73
Benzyl alcohol	103.45	35.82	97.44	76.58
Benzyl benzoate	89.14	93.09	67.00	81.76
Benzyl cinnamate	91.38	103.10	90.18	84.89
Benzyl salicylate	87.95	64.40	97.03	86.78
Chloroatranol	85.37	92.34	65.32	76.79
Cinnamic alcohol	105.07	65.93	38.05	34.54
Cinnamic aldehyde	96.38	89.43	90.57	81.96
Citral	89.57	94.12	67.01	77.92
Citronellol	88.78	46.15	96.99	37.00
Coumarin	95.45	70.87	72.38	83.06
Eugenol	98.73	78.73	65.87	83.19
Farnesol	108.57	87.80	83.30	72.74
Geraniol	91.74	85.46	74.80	71.44
Hexylcinnamic aldehyde	93.19	95.87	73.75	91.18
Hydroxycitronellal	97.00	52.31	85.75	79.83
Isoeugenol	103.17	61.28	105.30	81.19
Lilial	98.71	93.99	76.05	72.52
Limonene	96.05	80.66	19.28	48.94
Linalool	100.48	58.14	82.56	56.11
Lylal	98.20	97.48	85.83	70.29
Methyl-2-octynoate	98.99	91.87	62.01	75.86
Methyl eugenol	93.43	91.44	66.25	81.87
$\alpha$ -Amylcinnamyl alcohol	97.84	99.17	89.56	83.65
$\alpha$ -Isomethyl ionone	91.73	96.87	62.27	35.38

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