Isolation and Identification of Two Phenolic Compounds from a Moderately Cytotoxic Fraction of *Cousinia verbascifolia* Bunge

Abstract

Background: Little information is available about chemical components of the *Cousinia* genus. A primary cytotoxicity screening on *Cousinia verbascifolia* showed moderate cytotoxic activity against OVCAR-3 ovarian and HT-29 colon cancer cells. Therefore, the aim of this study is a phytochemical investigation to identify the compounds responsible for this bioactivity. **Materials and Methods:** Extraction was done through percolation and fractionations by reverse phase column chromatography and normal column chromatography. Using standard 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide assay Fr.b8 with moderate cytotoxicity was selected for identification of major components. Fr.b8 was subjected to polyamide column chromatography. More purification was done using a new modified recycle high-performance liquid chromatography (HPLC) with flow splitter. **Results:** Two known compounds: Apigenin (flavone) and caffeic acid (phenolic acid) were obtained from phenolic bioactive fraction for the first time from this plant. **Conclusions:** Apigenin and caffeic acid with known antitumor and matrix metalloproteinase inhibitory effects seem to be the bioactive components responsible for moderate cytotoxicity of phenolic fraction. Recycle HPLC following with flow splitting is a new method useful for isolation of closely eluted compounds in HPLC chromatogram.

Keywords: Apigenin, Asteraceae, caffeic acid, colon cancer, Cousinia verbascifolia, ovary cancer, recycle high-performance liquid chromatography

Introduction

Plant kingdom provides a rich and wonderful source of biologically active compounds. There are about 250,000–500,000 plant species on earth, out of which only about 20% have been submitted to biological and phytochemical screening.^[1-3] Hence, a huge number of plant species still remain to be studied and described in terms of biological and phytochemical profiles.

Cousinia, one of the largest genera of *Asteraceae* with 600–700 species, is one of the most diverse genera in central and South-West Asia. About 250 species of the genus are distributed throughout Iran, comprising about 200 endemic species.^[4-7] Based on phytochemical studies on a few *Cousinia* species chemical constituents including acetylenes, triterpenes, steroids, sesquiterpene lactones, and flavonoids were reported.^[8-17] However, little information is available about the phytochemical and biological properties of the species grow in Iran. In a phytochemical study on the aerial parts of three Iranian species of the genus

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Cousinia (Cousinia picheriana, Cousinia piptocephala, and Cousinia canescens), all three species yielded guianolide class sesquiterpens including 10 (14), 11 (13)-guiadien, 12, 6-olide, 4 (15), 10 (14)-guiadien, 12, 6-olide, and 4 (15), 10 (14), 11 (13)-guiatrien, 12, 6-olide derivatives mainly differed in functions on C-3, C-4, C-15, and ester located on C-8. In addition, C. canescens contained oxygenated bisabolene derivatives.^[9] In another study on cytotoxicity of total ethanol extract of seven Iranian Cousinia species against fibrosarcoma WEHI 164 cancer cells, they showed medium cytotoxicity with inhibitory concentration ranged between $18.4~\pm~0.59$ and $87.9~\pm~0.58~\mu\text{g/mL}.$ In the same study on fibrosarcoma WEHI 164 cancer cells, most of the Cousinia species had considerable inhibitory effect on matrix metalloproteinase (MMP) activity.^[18]

Cousinia verbascifolia Bunge is endemic to Kopetdagh (NE Iran and S Turkmenistan), and could be seen

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Seyed Ebrahim Sajjadi, Mustafa Ghanadian, Mehrangiz Haghighi

From the Department of Pharmacognosy, School of Pharmacy and Pharmaceutical Sciences, Isfahan University of Medical Sciences, Isfahan, Iran

Address for correspondence: Dr. Mustafa Ghanadian, Department of Pharmacognosy, School of Pharmacy and Pharmaceutical Sciences, Isfahan University of Medical Sciences, Isfahan, Iran. E-mail: ghannadian@gmail.com



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in open areas, scrublands, and stony slopes.^[6] It is a monocarpic biennial plant, up to 45 cm high, stems simple or branched from the base, leaves lyrate with spiny-dentate margin, flowers 50–150; corolla pink, rose or purple, 20–35 mm long. The flowering period of the plant is May to July.^[6] In a primary study by authors, different fractions of *C. verbascifolia* showed medium cytotoxicity against OVCAR-3 and HT-29 cells.^[20] In this present study, due to the observed inhibitory properties of semi-polar fractions of *C. verbascifolia*'s extract against cancer cells, phytochemical investigation on these semi-polar fractions were carried out using a new modified recycle high-performance liquid chromatography (HPLC) technique which was used in the purification process.

Materials and Methods

General

Nuclear magnetic resonance (NMR) spectra were recorded on a Bruker Avance AV 400, using pyridine-d6 as solvent. Electrospray ionization mass spectrometry (ESI-MS) spectra were measured in negative mode on Shimadzu 2010EV LC-MS system (Shimadzu, Japan).

Chromatographic conditions

Column chromatography runs were performed using silica gel, 63–200 μ m (Merck), polyamide DC6 (MACHEREY-NAGEL, Duren, Germany), and reverse phase chromatography was done by LiChroprep RP-18, 40–60 μ m (Merck). High performance thin layer chromatography was performed on silica gel GF-254 plates (Merck KGaA, Darmstadt, Germany). Plates were developed by natural product reagent (1% methanolic diphenyl-boric acid-ethanolamine) and visualized by ultraviolet (UV) fluorescent colors at 254/366 nm UV lamps. Recycle-high pressure liquid chromatography was done on a Waters HPLC apparatus (Waters Assoc., Milford, MA, USA), at 250 nm using silica gel column (YMC-Pack SIL, 250 mm \times 20 mm, YMC Co., Kyoto, Japan).

Plant material

Flowering aerial parts of *C. verbascifolia* (*Asteraceae*) were collected from Salehabad, Torbat-e Jam in Razavi Khorasan Province, Iran, in May 2013. It was identified by Dr. Iraj Mehregan. A voucher specimen (No. 2838) was deposited at the Herbarium of the Pharmacognosy Department, School of Pharmacy, Isfahan University of Medical Sciences, Isfahan, I. R. Iran.

Extraction, isolation, and purification

Air dried aerial parts of *C. verbascifolia* (3.5 kg) were extracted via percolation method using methanol as solvent. Filtration and in vacuum concentration of methanol extract resulted in a green gum which was subjected to reverse phase chromatography for removing fats and chlorophylls using methanol:water (80:20) as solvent. Defatted

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fraction (41.3 g) was then subjected to a silica gel column chromatography using hexane/acetone as mobile phase with increasing polarity $(5 \rightarrow 50\%)$ that afforded 11 fractions (Fr.b1-8). These fractions were studied for cytotoxic activity using 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide standard method against ovarian and colon cancer cells.^[19,20] Detailed description of this study is published by authors elsewhere.^[20] The cytotoxicity screening revealed that Fr.b2, Fr.b4, and Fr.b8 were the most cytotoxic fractions.^[20] Fr.b2 contained mainly fatty acids and common steroids which were discarded. Inspired from ¹H-NMR spectra, Fr.b4 contained sesquiterpene lactones but with inadequate amounts for further analysis that were excluded from the study. Finally, bioactive fraction, Fr.b8, was selected for identification of major components responsible for the observed cytotoxic effects. Fr.b8 was then subjected to polyamide DC6 column (40 cm \times 2 cm) using chloroform: methanol $(2\rightarrow 20\%)$ with increasing polarity to yield fractions Fr.b8a-d. After thin layer chromatography visualization, Fr.b8d eluted by chloroform:methanol (88:12) with positive reaction to natural product reagent was selected and injected into the recycle HPLC system using YMC-SIL column (25 cm \times 2.5 cm), particle size: 10 μ m, mobile phase: chloroform:methanol (85:15), injection volume: 2 ml, flow rate: 3 ml/min, chart rate: 0.1 cm/min, and detector wavelength: 250 nm.

In water, 501 HPLC pump a recycle valve and a flow splitter were incorporated in HPLC system between detector and HPLC pump, so that the chromatographic profile could reenter the column after passing through the detector. This process resembles using a longer column and thus provides more theoretical plates leading to increasing the column efficiency and resolution. In the case of observing tailing in the peaks, flow is splitted before reentering the column. Finally, during any cycle in which the desired resolution is achieved, samples are collected. In this process, after three recycles following the flow splitting (80% to pump and 20% to waste), the peaks were sufficiently resolved to compounds A, B, and C [Figure 1]. Compound B was excluded from the study because of insufficient purity, and compounds A and C were subjected to mass and NMR spectroscopy which led to the identification of compounds 1 (30 mg) and 2 (10 mg) [Figure 2].

Results

Two known compounds were obtained from phenolic bioactive fraction for the first time from this plant. Their physical properties are given bellow:

Compound 1

Pale yellow powder, ¹H-NMR in pyridine-d6 (400 MHz) ppm: 7.76 (d, 2H, J = 8.9 Hz), 6.84 (d, 2H, J = 8.8 Hz), 6.50 (s, ¹H), 6.37 (d, 1H, J = 2.1 Hz), 6.11 (d, 1H, J = 2.1 Hz); ¹³C-NMR (100 MHz, pyridine-d6) δ_{c} : 163.8 (C2), 104.4 (C3), 183.2 (C4), 106.5 (C4a),



Figure 1: Isolation of apigenin (A) and caffeic acid (C) from *Cousinia verbascifolia* by implementing recycle high-performance liquid chromatography technique in the purification process. Fraction Fr.b8d was injected into the recycle high-performance liquid chromatography system using YMC-SIL column (10 µm, 25 cm × 2.5 cm) using chloroform:methanol (85:15) at 250 nm. After three recycles, the peaks were sufficiently resolved to compounds A, B, and C

163.2 (C5), 100.5 (C6), 166.4 (C7), 95.3 (C8), 159.0 (C8a), 122.8 (C1'), 129.4 (C2'), 117.3 (C3'),163.3 (C4'), 117.3 (C5'), 122.8 (C6'). ESI negative mass m/z 269 (M–1).

Compound 2

White powder, ¹H-NMR in pyridine-d6 (400 MHz) ppm: 9.63 (d, 1H, J = 15.9 Hz, H-7), 9.16 (d, 1H, J = 1.4 Hz, H-2), 8.73 (d, 1H, J = 8.1 Hz, H-5), 8.71 (dd, 1H, J = 1.6 Hz, J = 8.2 Hz, H-6), 8.33 (d, ¹H, J = 15.8 Hz, H-8); ¹³C-NMR (100 MHz, pyridine-d6) δ_{c} : 171.7 (C1), 116.7 (C2), 148.7 (C3), 147.9 (C4), 117.8 (C5), 122.7 (C6), 146.2 (C-7), 117.8 (C-8), 170.6 (C-9). ESI negative mass m/z 179 (M-1).

Discussion

Compound 1 was isolated as a pale yellowish solid with positive reaction to FeCl₃ and natural product reagents. Molecular formula was determined by NMR data and negative ESI-MS m/z 269 (M-1) as C₁₅H₁₀O₅. Eleven degrees of unsaturation and NMR data suggested the presence of one carbonyl carbon, seven olefin bands, and therefore, three rings in the skeleton. The UV spectrum showed absorption maxima at 265 and 340 nm that are characteristic of flavones.[8] The 1H-NMR spectrum displayed meta-coupled doublets at δ 6.11 (1H, d, J = 2 Hz) and 6.37 (1H, d, J = 2 Hz) described to H-6 and H-8. In addition, ortho-coupled proton signals at δ 6.84 (²H, d, J = 8. 5 Hz) and 7.76 (2H, d, J = 8.5 Hz), corresponding to H-5',3' and H-6',2' as well as δ 6.50 (¹H, s) corresponding to H-3 proton, indicated that compound 1 is a flavone derivative. Comparison of the spectral data with those published before allowed us to establish the structure of compound 1 as 4',5,7-trihydroxyflavon or apigenin.^[21]



Figure 2: Apigenin (a) and caffeic acid (b) isolated from *Cousinia* verbascifolia Bunge

Compound 2 was yielded as a white solid with positive reaction to FeCl₂. The six degrees of unsaturation derived from positive HR-ESI-MS m/z 181.0495 (calcd. for $C_{9}H_{8}O_{4} + H^{+}$, 181.0495) and NMR data suggested the presence of one carbonyl carbon, four olefin bands, and therefore, one ring in the skeleton. The ¹H-NMR spectra displayed meta-coupled doublets at δ 9.16 (¹H, d, J = 1.4 Hz) as well as ortho-coupled proton signals at $\delta_{\rm H}$ 8.73 (1H, d, J = 8.1 Hz) and 8.71 (1H, d, J = 1.6, 8.2 Hz), described to H-2, H-6 and H-8. The above ABX spin coupling system resembled 3,4-disubstituted benzyl derivatives.^[22] In addition, two trans-olefin protons at δ_{H} 9.63 (1H, d, J = 15.9 Hz) and 9.63 (1H, d, J = 15.9 Hz) with relatively large coupling constant J = 15.8 Hz and acidic proton at down field aria at $\delta_{_{\rm H}}$ 10.1 ppm were observed. Comparison of the spectral data with those published before allowed us to establish the structure of compound 2 as 3-(3,4-dihydroxyphenyl)-2-propenoic acid or caffeic acid.[23,24]

Numerous studies have reported anticancer properties of both apigenin and caffeic acid through various mechanisms and pathways.^[25,26] In a report by Wang et al., they investigated the cell growth inhibition and cell-cycle arrest caused by apigenin against HT-29 colon cancer cell line. The results showed that apigenin induces a cell cycle arrest at G2/M phase, but this inhibitory effect on cancer cells appeared to be cytostatic, not cytotoxic.^[27] Antitumor effects of caffeic acid against colon and hepatocarcinoma cell lines are also reported through MMP-9 inhibitory effects.^[28-30] There are many evidence supporting those selective MMP inhibitors are effective in prevention and treatment of cancer.[28-32] Caffeic acid and its derivatives selectively inhibited MMP-2 and -9.^[29] In a similar study, hepatocarcinoma cells with phorbol ester (PMA)-induced MMP-9 expression were also suppressed followed by caffeic acid derivatives treatment. Their oral administrations significantly reduced the hepatic metastasis.^[28,29] Apigenin has also been reported to inhibit MMP-9 secretion.[33] The IC₅₀ of apigenin and caffeic acid against HT-29 colon cancer cell line was reported as 50 and 60 µM, respectively.^[27,34] In a previous study, Shahverdi et al. demonstrated cytotoxicity and MMP inhibitory effects of C. verbascifolia.^[18] These results together with aforementioned properties of apigenin and caffeic acid may explain the moderate cytotoxicity of C. verbascifolia phenolic fraction.

Conclusion

Two phenolic compounds: Apigenin (flavone) and caffeic acid (phenolic acid) were isolated for the first time from *C. verbascifolia*. Both of these compounds possess antitumor and MMP inhibitory effects. Thus, the bioactivities observed from *C. verbascifolia* in the previous studies might be due to the presence of these two phenolic compounds. Moreover, recycle HPLC following with flow splitting is a new method useful for isolation of closely eluted compounds in HPLC chromatogram.

Limitation of the study

The limitation of this study was inadequate amounts of sesquiterpene lactones in fraction Fr.b4 for further purification and elucidation. One another limitation of this study was the sample size due to the low yield of the extract. Although apigenin and caffeic acid seems to be responsible for observed cytotoxic effects but more phytochemical investigation with larger amounts of the plant material is suggested for identification of sesquiterpene lactones as another possible compounds responsible for cytotoxic activities in this plant.

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Conflicts of interest

There are no conflicts of interest.

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