

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

Ethyl 2-(1,3-benzodioxol-5-yl)-1-[3-(1*H*-imidazol-1-yl)propyl]-1*H*-benzimidazole-5-carboxylateYeong Keng Yoon,^a Mohamed Ashraf Ali,^a Tan Soo Choon,^a Safra Izuani Jama Asik^b and Ibrahim Abdul Razak^{b*‡}^aInstitute for Research in Molecular Medicine, Universiti Sains Malaysia, Minden 11800, Penang, Malaysia, and ^bSchool of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

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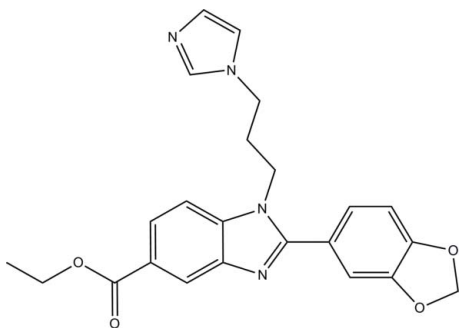
Received 14 December 2011; accepted 19 December 2011

Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.051; wR factor = 0.126; data-to-parameter ratio = 20.6.

In the title compound, $\text{C}_{23}\text{H}_{22}\text{N}_4\text{O}_4$, the essentially planar [maximum deviation = 0.022 (1) Å] benzimidazole ring system forms dihedral angles of 86.16 (7) and 37.38 (6)°, respectively, with the imidazole and benzene rings. The dioxolane ring adopts an envelope conformation with the methylene C atom at the flap. In the crystal, $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\text{N}$ interactions link the molecules into a ribbon along the a axis. The crystal packing is further stabilized by weak $\pi-\pi$ stacking interactions [centroid-centroid distances = 3.5954 (8) and 3.7134 (8) Å] and $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For the biological activity of benzimidazole derivatives, see: Grassmann *et al.* (2002); Demirayak *et al.* (2002). For puckering parameters, see: Cremer & Pople (1975). For stability of the temperature controller used in the data collection, see: Cosier & Glazer (1986). For a related structure, see: Yoon *et al.* (2011).



‡ Thomson Reuters ResearcherID: A-5599-2009.

Experimental

Crystal data

$\text{C}_{23}\text{H}_{22}\text{N}_4\text{O}_4$
 $M_r = 418.45$
 Orthorhombic, $Pbca$
 $a = 15.8554$ (2) Å
 $b = 15.3988$ (2) Å
 $c = 16.2292$ (2) Å
 $V = 3962.43$ (9) Å³
 $Z = 8$
 Mo $K\alpha$ radiation
 $\mu = 0.10$ mm⁻¹
 $T = 100$ K
 $0.42 \times 0.28 \times 0.20$ mm

Data collection

Bruker SMART APEXII CCD area-detector diffractometer
 Absorption correction: multi-scan (*SADABS*; Bruker, 2009)
 $T_{\min} = 0.960$, $T_{\max} = 0.981$
 39067 measured reflections
 5782 independent reflections
 4429 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.054$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.051$
 $wR(F^2) = 0.126$
 $S = 1.06$
 5782 reflections
 280 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.42$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

C_{g1} and C_{g4} are the centroids of $C11/C12/O3/C23/O4$ and $C1-C6$ rings, respectively.

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|------------------------------|-------|-------------|-------------|---------------|
| $C2-H2A\cdots O3^i$ | 0.95 | 2.48 | 3.3922 (17) | 162 |
| $C15-H15B\cdots O4^{ii}$ | 0.99 | 2.49 | 3.213 (2) | 130 |
| $C23-H23A\cdots N4^{iii}$ | 0.99 | 2.43 | 3.379 (2) | 159 |
| $C10-H10A\cdots C_{g4}^{iv}$ | 0.95 | 2.65 | 3.3005 (16) | 126 |
| $C16-H16C\cdots C_{g1}^v$ | 0.98 | 2.91 | 3.7282 (19) | 142 |

Symmetry codes: (i) $x - \frac{1}{2}, y, -z + \frac{3}{2}$; (ii) $x - 1, y, z$; (iii) $-x + 2, -y + 2, -z + 1$; (iv) $-x, y + \frac{3}{2}, -z + \frac{3}{2}$; (v) $-x + 1, -y + 2, -z + 1$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINTE* (Bruker, 2009); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

The authors thank the Malaysian Government and Universiti Sains Malaysia for the Research University Grants Nos. 1001/PFIZIK/811151 and (1001/PSK/8620012). The authors also wish to express their thanks to the Pharmacogenetic and Novel Therapeutic Research, Institute for Research in Molecular Medicine, Universiti of Sains Malaysia, Penang.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IS5032).

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supplementary materials

Acta Cryst. (2012). E68, o247-o248 [doi:10.1107/S1600536811054572]

Ethyl 2-(1,3-benzodioxol-5-yl)-1-[3-(1*H*-imidazol-1-yl)propyl]-1*H*-benzimidazole-5-carboxylate

Y. K. Yoon, M. A. Ali, T. S. Choon, S. I. J. Asik and I. A. Razak

Comment

Benzimidazole derivatives are of wide interest because of their diverse biological activities and various clinical applications. In particular, 2-substituted benzimidazoles have been proven as effective drug leads, thus generating pharmacological interests (Grassmann *et al.*, 2002; Demirayak *et al.*, 2002). As part of our ongoing structural studies of benzimidazole derivatives (Yoon *et al.*, 2011), we now report the structure of the title compound.

Fig. 1 shows the molecular structure of the title compound. The benzimidazole (N1–N2/C1–C7) ring is approximately planar with a maximum deviation of 0.022 (1) Å for atoms C6 and C7. The mean plane through this ring forms dihedral angles of 86.16 (7) and 37.38 (6)° with the mean plane through the imidazole (N3/N4/C20–C22) and benzene (C8–C13) rings, respectively. The dioxolane (O3/O4/C11/C12/C23) ring adopts an envelope conformation with puckering parameters $Q = 0.1209$ (14) Å and $\phi = 138.2$ (7)° with atom C23 at the flap (Cremer & Pople, 1975).

In the crystal packing of (Fig. 2), intermolecular C2—H2A···O3($x - 1/2, y, -z + 3/2$), C15—H15B···O4($-1 + x, y, z$) and C23—H23A···N4($2 - x, 2 - y, 1 - z$) interactions form the molecules into ribbon stacked along the *a*-axis. π – π interactions are observed within the benzimidazole ring system between the imidazole (N1/N2/C1/C6–C7; centroid *Cg*2) and benzene, (C1–C6; centroid *Cg*4) rings with a *Cg*2···*Cg*4($1 - x, 2 - y, 1 - z$) distance of 3.5954 (8) Å and between the benzene, (C1–C6; centroid *Cg*4) rings with a *Cg*4···*Cg*4($1 - x, 2 - y, 1 - z$) distance of 3.7134 (8) Å. The crystal packing are further stabilized by weak C—H··· π interactions (Table 1) involving the benzene ring of the benzimidazole moiety and the dioxolane ring with the distances of 3.3005 (16) and 3.7282 (19) Å, respectively.

Experimental

Ethyl-4-[3-(1*H*-imidazol-1-yl)propylamino]-3-aminobenzoate (0.84 mmol) and sodium metabisulfite adduct of piperonal (1.68 mmol) were dissolved in DMF. The reaction mixture was reflux at 130 °C for 2 hrs. After completion, the reaction mixture was diluted in ethyl acetate (20 mL) and washed with water (20 mL). The organic layer was collected, dried over Na₂SO₄ and the evaporated in vacuo to yield the product. The product was recrystallised from ethyl acetate.

Refinement

All the H atoms positioned geometrically and refined using a riding model with C—H = 0.95–0.99 Å. The U_{iso} values were constrained to be 1.5 U_{eq} (methyl-H atom) and 1.2 U_{eq} (other H atoms). The rotating model group was applied for the methyl group.

Figures

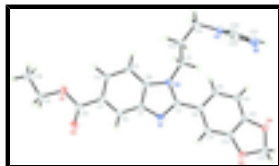


Fig. 1. The structure of the title compound, showing 50% probability displacement ellipsoids. Hydrogen atoms are shown as spheres of arbitrary radius.

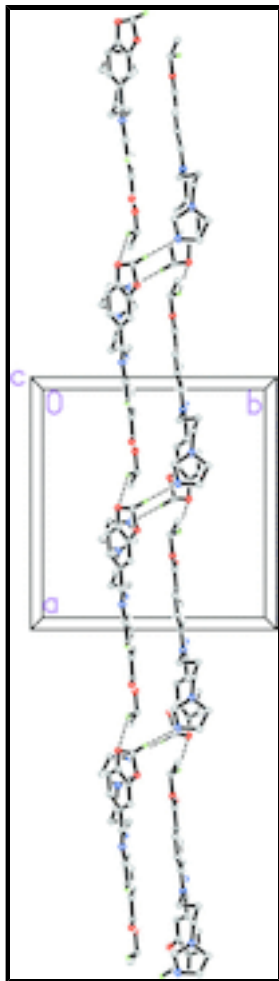


Fig. 2. The crystal packing, viewed along the *c*-axis, showing the molecules are connected into ribbon along *a* axis. Hydrogen bonds are shown as dashed lines.

Ethyl 2-(1,3-benzodioxol-5-yl)-1-[3-(1*H*-imidazol-1-yl)propyl]- 1*H*-benzimidazole-5-carboxylate

Crystal data

$C_{23}H_{22}N_4O_4$

$M_r = 418.45$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 15.8554 (2) \text{ \AA}$

$b = 15.3988 (2) \text{ \AA}$

$c = 16.2292 (2) \text{ \AA}$

$F(000) = 1760$

$D_x = 1.403 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 7948 reflections

$\theta = 2.2\text{--}29.9^\circ$

$\mu = 0.10 \text{ mm}^{-1}$

$T = 100 \text{ K}$

$V = 3962.43 (9) \text{ \AA}^3$
 $Z = 8$

Block, yellow
 $0.42 \times 0.28 \times 0.20 \text{ mm}$

Data collection

| | |
|--|--|
| Bruker SMART APEXII CCD area-detector diffractometer | 5782 independent reflections |
| Radiation source: fine-focus sealed tube graphite | 4429 reflections with $I > 2\sigma(I)$ |
| φ and ω scans | $R_{\text{int}} = 0.054$ |
| Absorption correction: multi-scan (SADABS; Bruker, 2009) | $\theta_{\text{max}} = 30.0^\circ$, $\theta_{\text{min}} = 2.2^\circ$ |
| $T_{\text{min}} = 0.960$, $T_{\text{max}} = 0.981$ | $h = -22 \rightarrow 22$ |
| 39067 measured reflections | $k = -17 \rightarrow 21$ |
| | $l = -22 \rightarrow 22$ |

Refinement

| | |
|---------------------------------|--|
| Refinement on F^2 | Primary atom site location: structure-invariant direct methods |
| Least-squares matrix: full | Secondary atom site location: difference Fourier map |
| $R[F^2 > 2\sigma(F^2)] = 0.051$ | Hydrogen site location: inferred from neighbouring sites |
| $wR(F^2) = 0.126$ | H-atom parameters constrained |
| $S = 1.06$ | $w = 1/[\sigma^2(F_o^2) + (0.0539P)^2 + 1.7654P]$ |
| 5782 reflections | where $P = (F_o^2 + 2F_c^2)/3$ |
| 280 parameters | $(\Delta/\sigma)_{\text{max}} = 0.001$ |
| 0 restraints | $\Delta\rho_{\text{max}} = 0.42 \text{ e \AA}^{-3}$ |
| | $\Delta\rho_{\text{min}} = -0.26 \text{ e \AA}^{-3}$ |

Special details

Experimental. The crystal was placed in the cold stream of an Oxford Cryosystems Cobra open-flow nitrogen cryostat (Cosier & Glazer, 1986) operating at 100.0 (1) K.

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|-------------|-------------|-------------|----------------------------------|
| O1 | 0.21824 (6) | 0.91984 (8) | 0.49482 (7) | 0.0282 (3) |
| O2 | 0.25425 (7) | 0.90679 (7) | 0.62843 (7) | 0.0271 (2) |

supplementary materials

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|------|--------------|--------------|--------------|------------|
| O3 | 0.89060 (6) | 0.92843 (7) | 0.71394 (6) | 0.0239 (2) |
| O4 | 0.98149 (6) | 0.84912 (7) | 0.63028 (7) | 0.0263 (2) |
| N1 | 0.58208 (7) | 0.86631 (8) | 0.59836 (7) | 0.0194 (2) |
| N2 | 0.61041 (7) | 0.87485 (8) | 0.46195 (7) | 0.0187 (2) |
| N3 | 0.79424 (8) | 0.84083 (8) | 0.26915 (7) | 0.0225 (3) |
| N4 | 0.91848 (9) | 0.89721 (9) | 0.30370 (9) | 0.0296 (3) |
| C1 | 0.50764 (8) | 0.87928 (9) | 0.55479 (8) | 0.0182 (3) |
| C2 | 0.42516 (8) | 0.88928 (9) | 0.58316 (9) | 0.0190 (3) |
| H2A | 0.4125 | 0.8865 | 0.6403 | 0.023* |
| C3 | 0.36188 (8) | 0.90350 (9) | 0.52518 (9) | 0.0191 (3) |
| C4 | 0.38044 (9) | 0.90927 (9) | 0.44011 (9) | 0.0202 (3) |
| H4A | 0.3360 | 0.9195 | 0.4021 | 0.024* |
| C5 | 0.46187 (9) | 0.90041 (9) | 0.41113 (8) | 0.0200 (3) |
| H5A | 0.4747 | 0.9048 | 0.3541 | 0.024* |
| C6 | 0.52433 (8) | 0.88464 (9) | 0.46988 (8) | 0.0182 (3) |
| C7 | 0.64171 (9) | 0.86485 (9) | 0.54132 (9) | 0.0184 (3) |
| C8 | 0.73210 (8) | 0.85749 (9) | 0.56059 (8) | 0.0182 (3) |
| C9 | 0.78806 (9) | 0.81167 (10) | 0.51027 (9) | 0.0212 (3) |
| H9A | 0.7671 | 0.7838 | 0.4622 | 0.025* |
| C10 | 0.87401 (9) | 0.80575 (10) | 0.52865 (9) | 0.0221 (3) |
| H10A | 0.9118 | 0.7742 | 0.4944 | 0.027* |
| C11 | 0.90130 (8) | 0.84735 (9) | 0.59813 (9) | 0.0196 (3) |
| C12 | 0.84659 (9) | 0.89351 (9) | 0.64855 (8) | 0.0185 (3) |
| C13 | 0.76188 (9) | 0.89913 (9) | 0.63259 (8) | 0.0195 (3) |
| H13A | 0.7248 | 0.9297 | 0.6683 | 0.023* |
| C14 | 0.27411 (9) | 0.91028 (9) | 0.55607 (9) | 0.0203 (3) |
| C15 | 0.13033 (9) | 0.92811 (14) | 0.51877 (11) | 0.0346 (4) |
| H15A | 0.1172 | 0.9892 | 0.5326 | 0.042* |
| H15B | 0.1185 | 0.8917 | 0.5677 | 0.042* |
| C16 | 0.07810 (10) | 0.89887 (12) | 0.44744 (11) | 0.0332 (4) |
| H16A | 0.0182 | 0.9039 | 0.4615 | 0.050* |
| H16B | 0.0914 | 0.8382 | 0.4346 | 0.050* |
| H16C | 0.0904 | 0.9353 | 0.3994 | 0.050* |
| C17 | 0.65550 (9) | 0.89081 (10) | 0.38489 (8) | 0.0205 (3) |
| H17A | 0.7154 | 0.9030 | 0.3976 | 0.025* |
| H17B | 0.6316 | 0.9431 | 0.3582 | 0.025* |
| C18 | 0.65105 (9) | 0.81489 (10) | 0.32436 (9) | 0.0239 (3) |
| H18A | 0.6717 | 0.7613 | 0.3515 | 0.029* |
| H18B | 0.5918 | 0.8053 | 0.3075 | 0.029* |
| C19 | 0.70487 (10) | 0.83430 (11) | 0.24854 (9) | 0.0270 (3) |
| H19A | 0.6968 | 0.7876 | 0.2074 | 0.032* |
| H19B | 0.6859 | 0.8896 | 0.2235 | 0.032* |
| C20 | 0.84682 (9) | 0.77241 (10) | 0.28605 (9) | 0.0241 (3) |
| H20A | 0.8332 | 0.7124 | 0.2839 | 0.029* |
| C21 | 0.92254 (10) | 0.80816 (10) | 0.30661 (9) | 0.0252 (3) |
| H21A | 0.9715 | 0.7759 | 0.3210 | 0.030* |
| C22 | 0.84051 (10) | 0.91387 (10) | 0.28135 (9) | 0.0273 (3) |
| H22A | 0.8188 | 0.9709 | 0.2745 | 0.033* |
| C23 | 0.97877 (9) | 0.91055 (11) | 0.69681 (9) | 0.0248 (3) |

| | | | | |
|------|--------|--------|--------|--------|
| H23A | 1.0084 | 0.9647 | 0.6811 | 0.030* |
| H23B | 1.0066 | 0.8861 | 0.7463 | 0.030* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|------------|-------------|------------|-------------|-------------|-------------|
| O1 | 0.0160 (5) | 0.0441 (7) | 0.0245 (5) | 0.0033 (4) | 0.0008 (4) | 0.0050 (5) |
| O2 | 0.0243 (5) | 0.0349 (6) | 0.0220 (5) | 0.0040 (5) | 0.0030 (4) | 0.0006 (5) |
| O3 | 0.0217 (5) | 0.0320 (6) | 0.0181 (5) | -0.0026 (4) | -0.0009 (4) | -0.0034 (4) |
| O4 | 0.0179 (5) | 0.0356 (6) | 0.0253 (5) | -0.0013 (4) | -0.0006 (4) | -0.0055 (5) |
| N1 | 0.0199 (5) | 0.0202 (6) | 0.0182 (6) | 0.0012 (4) | -0.0008 (4) | -0.0003 (4) |
| N2 | 0.0174 (5) | 0.0210 (6) | 0.0178 (5) | 0.0014 (4) | 0.0011 (4) | -0.0003 (4) |
| N3 | 0.0263 (6) | 0.0261 (7) | 0.0151 (5) | 0.0031 (5) | 0.0024 (4) | -0.0009 (5) |
| N4 | 0.0328 (7) | 0.0283 (7) | 0.0276 (7) | -0.0050 (6) | 0.0045 (5) | 0.0027 (6) |
| C1 | 0.0212 (6) | 0.0171 (7) | 0.0163 (6) | 0.0011 (5) | -0.0008 (5) | 0.0002 (5) |
| C2 | 0.0213 (6) | 0.0182 (7) | 0.0174 (6) | 0.0007 (5) | 0.0011 (5) | -0.0007 (5) |
| C3 | 0.0193 (6) | 0.0179 (7) | 0.0199 (7) | 0.0012 (5) | 0.0011 (5) | -0.0007 (5) |
| C4 | 0.0199 (6) | 0.0210 (7) | 0.0196 (7) | 0.0010 (5) | -0.0018 (5) | 0.0015 (5) |
| C5 | 0.0223 (7) | 0.0218 (7) | 0.0160 (6) | 0.0009 (5) | 0.0003 (5) | 0.0005 (5) |
| C6 | 0.0183 (6) | 0.0173 (7) | 0.0188 (6) | 0.0002 (5) | 0.0014 (5) | -0.0009 (5) |
| C7 | 0.0209 (6) | 0.0160 (7) | 0.0184 (6) | 0.0010 (5) | -0.0009 (5) | -0.0008 (5) |
| C8 | 0.0201 (6) | 0.0170 (7) | 0.0176 (6) | 0.0002 (5) | -0.0010 (5) | 0.0020 (5) |
| C9 | 0.0220 (6) | 0.0197 (7) | 0.0220 (7) | 0.0011 (5) | -0.0014 (5) | -0.0037 (5) |
| C10 | 0.0211 (6) | 0.0217 (7) | 0.0236 (7) | 0.0020 (5) | 0.0025 (5) | -0.0027 (6) |
| C11 | 0.0172 (6) | 0.0205 (7) | 0.0212 (7) | -0.0015 (5) | 0.0010 (5) | 0.0034 (5) |
| C12 | 0.0229 (6) | 0.0180 (7) | 0.0146 (6) | -0.0022 (5) | 0.0000 (5) | 0.0015 (5) |
| C13 | 0.0217 (6) | 0.0198 (7) | 0.0171 (6) | 0.0019 (5) | 0.0010 (5) | -0.0002 (5) |
| C14 | 0.0206 (6) | 0.0172 (7) | 0.0230 (7) | 0.0010 (5) | 0.0000 (5) | 0.0002 (5) |
| C15 | 0.0174 (7) | 0.0565 (12) | 0.0300 (9) | 0.0047 (7) | 0.0029 (6) | 0.0029 (8) |
| C16 | 0.0223 (7) | 0.0453 (10) | 0.0320 (9) | -0.0005 (7) | -0.0004 (6) | 0.0054 (7) |
| C17 | 0.0214 (6) | 0.0228 (7) | 0.0173 (6) | 0.0009 (5) | 0.0023 (5) | 0.0007 (5) |
| C18 | 0.0228 (7) | 0.0275 (8) | 0.0215 (7) | 0.0005 (6) | -0.0025 (5) | -0.0044 (6) |
| C19 | 0.0279 (7) | 0.0372 (9) | 0.0159 (6) | 0.0055 (6) | -0.0021 (5) | -0.0030 (6) |
| C20 | 0.0296 (7) | 0.0215 (7) | 0.0213 (7) | 0.0032 (6) | 0.0015 (6) | -0.0030 (6) |
| C21 | 0.0272 (7) | 0.0266 (8) | 0.0219 (7) | 0.0011 (6) | 0.0032 (6) | -0.0003 (6) |
| C22 | 0.0381 (8) | 0.0224 (8) | 0.0215 (7) | 0.0003 (6) | 0.0041 (6) | 0.0029 (6) |
| C23 | 0.0207 (6) | 0.0318 (8) | 0.0219 (7) | -0.0028 (6) | -0.0013 (5) | -0.0010 (6) |

Geometric parameters (Å, °)

| | | | |
|--------|-------------|----------|-------------|
| O1—C14 | 1.3396 (17) | C8—C13 | 1.4140 (19) |
| O1—C15 | 1.4526 (18) | C9—C10 | 1.3981 (19) |
| O2—C14 | 1.2170 (18) | C9—H9A | 0.9500 |
| O3—C12 | 1.3792 (16) | C10—C11 | 1.367 (2) |
| O3—C23 | 1.4516 (18) | C10—H10A | 0.9500 |
| O4—C11 | 1.3745 (17) | C11—C12 | 1.3883 (19) |
| O4—C23 | 1.4361 (19) | C12—C13 | 1.3707 (19) |
| N1—C7 | 1.3233 (18) | C13—H13A | 0.9500 |
| N1—C1 | 1.3902 (17) | C15—C16 | 1.493 (2) |

supplementary materials

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|------------|-------------|---------------|-------------|
| N2—C6 | 1.3793 (17) | C15—H15A | 0.9900 |
| N2—C7 | 1.3889 (18) | C15—H15B | 0.9900 |
| N2—C17 | 1.4613 (17) | C16—H16A | 0.9800 |
| N3—C22 | 1.357 (2) | C16—H16B | 0.9800 |
| N3—C20 | 1.3713 (19) | C16—H16C | 0.9800 |
| N3—C19 | 1.460 (2) | C17—C18 | 1.529 (2) |
| N4—C22 | 1.314 (2) | C17—H17A | 0.9900 |
| N4—C21 | 1.374 (2) | C17—H17B | 0.9900 |
| C1—C2 | 1.3950 (19) | C18—C19 | 1.527 (2) |
| C1—C6 | 1.4055 (19) | C18—H18A | 0.9900 |
| C2—C3 | 1.3930 (19) | C18—H18B | 0.9900 |
| C2—H2A | 0.9500 | C19—H19A | 0.9900 |
| C3—C4 | 1.414 (2) | C19—H19B | 0.9900 |
| C3—C14 | 1.4827 (19) | C20—C21 | 1.362 (2) |
| C4—C5 | 1.3808 (19) | C20—H20A | 0.9500 |
| C4—H4A | 0.9500 | C21—H21A | 0.9500 |
| C5—C6 | 1.3960 (19) | C22—H22A | 0.9500 |
| C5—H5A | 0.9500 | C23—H23A | 0.9900 |
| C7—C8 | 1.4714 (18) | C23—H23B | 0.9900 |
| C8—C9 | 1.3971 (19) | | |
| C14—O1—C15 | 116.47 (12) | C8—C13—H13A | 121.5 |
| C12—O3—C23 | 105.42 (11) | O2—C14—O1 | 123.34 (13) |
| C11—O4—C23 | 105.71 (11) | O2—C14—C3 | 124.47 (13) |
| C7—N1—C1 | 104.66 (11) | O1—C14—C3 | 112.18 (12) |
| C6—N2—C7 | 106.21 (11) | O1—C15—C16 | 107.36 (14) |
| C6—N2—C17 | 123.07 (12) | O1—C15—H15A | 110.2 |
| C7—N2—C17 | 129.61 (11) | C16—C15—H15A | 110.2 |
| C22—N3—C20 | 106.20 (13) | O1—C15—H15B | 110.2 |
| C22—N3—C19 | 127.97 (13) | C16—C15—H15B | 110.2 |
| C20—N3—C19 | 125.67 (13) | H15A—C15—H15B | 108.5 |
| C22—N4—C21 | 104.39 (13) | C15—C16—H16A | 109.5 |
| N1—C1—C2 | 130.07 (13) | C15—C16—H16B | 109.5 |
| N1—C1—C6 | 110.32 (12) | H16A—C16—H16B | 109.5 |
| C2—C1—C6 | 119.58 (12) | C15—C16—H16C | 109.5 |
| C3—C2—C1 | 118.01 (13) | H16A—C16—H16C | 109.5 |
| C3—C2—H2A | 121.0 | H16B—C16—H16C | 109.5 |
| C1—C2—H2A | 121.0 | N2—C17—C18 | 113.51 (12) |
| C2—C3—C4 | 121.29 (13) | N2—C17—H17A | 108.9 |
| C2—C3—C14 | 117.30 (12) | C18—C17—H17A | 108.9 |
| C4—C3—C14 | 121.40 (12) | N2—C17—H17B | 108.9 |
| C5—C4—C3 | 121.39 (13) | C18—C17—H17B | 108.9 |
| C5—C4—H4A | 119.3 | H17A—C17—H17B | 107.7 |
| C3—C4—H4A | 119.3 | C19—C18—C17 | 110.03 (13) |
| C4—C5—C6 | 116.60 (13) | C19—C18—H18A | 109.7 |
| C4—C5—H5A | 121.7 | C17—C18—H18A | 109.7 |
| C6—C5—H5A | 121.7 | C19—C18—H18B | 109.7 |
| N2—C6—C5 | 131.09 (13) | C17—C18—H18B | 109.7 |
| N2—C6—C1 | 105.74 (12) | H18A—C18—H18B | 108.2 |
| C5—C6—C1 | 123.12 (13) | N3—C19—C18 | 111.80 (12) |

| | | | |
|--------------|--------------|-----------------|--------------|
| N1—C7—N2 | 113.06 (12) | N3—C19—H19A | 109.3 |
| N1—C7—C8 | 123.27 (12) | C18—C19—H19A | 109.3 |
| N2—C7—C8 | 123.62 (12) | N3—C19—H19B | 109.3 |
| C9—C8—C13 | 120.01 (13) | C18—C19—H19B | 109.3 |
| C9—C8—C7 | 122.22 (12) | H19A—C19—H19B | 107.9 |
| C13—C8—C7 | 117.77 (12) | C21—C20—N3 | 105.91 (14) |
| C8—C9—C10 | 121.81 (13) | C21—C20—H20A | 127.0 |
| C8—C9—H9A | 119.1 | N3—C20—H20A | 127.0 |
| C10—C9—H9A | 119.1 | C20—C21—N4 | 110.72 (14) |
| C11—C10—C9 | 117.00 (13) | C20—C21—H21A | 124.6 |
| C11—C10—H10A | 121.5 | N4—C21—H21A | 124.6 |
| C9—C10—H10A | 121.5 | N4—C22—N3 | 112.77 (14) |
| C10—C11—O4 | 127.96 (13) | N4—C22—H22A | 123.6 |
| C10—C11—C12 | 121.89 (13) | N3—C22—H22A | 123.6 |
| O4—C11—C12 | 110.12 (12) | O4—C23—O3 | 107.33 (11) |
| C13—C12—O3 | 128.07 (13) | O4—C23—H23A | 110.2 |
| C13—C12—C11 | 122.22 (13) | O3—C23—H23A | 110.2 |
| O3—C12—C11 | 109.69 (12) | O4—C23—H23B | 110.2 |
| C12—C13—C8 | 117.05 (13) | O3—C23—H23B | 110.2 |
| C12—C13—H13A | 121.5 | H23A—C23—H23B | 108.5 |
| C7—N1—C1—C2 | -177.49 (15) | C23—O4—C11—C10 | 172.62 (15) |
| C7—N1—C1—C6 | 0.61 (16) | C23—O4—C11—C12 | -9.19 (16) |
| N1—C1—C2—C3 | 178.57 (14) | C23—O3—C12—C13 | -174.54 (15) |
| C6—C1—C2—C3 | 0.6 (2) | C23—O3—C12—C11 | 6.74 (15) |
| C1—C2—C3—C4 | -1.1 (2) | C10—C11—C12—C13 | 1.0 (2) |
| C1—C2—C3—C14 | 177.48 (13) | O4—C11—C12—C13 | -177.28 (13) |
| C2—C3—C4—C5 | 0.5 (2) | C10—C11—C12—O3 | 179.84 (13) |
| C14—C3—C4—C5 | -178.07 (13) | O4—C11—C12—O3 | 1.53 (16) |
| C3—C4—C5—C6 | 0.7 (2) | O3—C12—C13—C8 | 179.77 (13) |
| C7—N2—C6—C5 | 176.99 (15) | C11—C12—C13—C8 | -1.7 (2) |
| C17—N2—C6—C5 | 8.0 (2) | C9—C8—C13—C12 | 1.3 (2) |
| C7—N2—C6—C1 | -0.47 (15) | C7—C8—C13—C12 | -178.39 (12) |
| C17—N2—C6—C1 | -169.43 (12) | C15—O1—C14—O2 | 1.4 (2) |
| C4—C5—C6—N2 | -178.25 (14) | C15—O1—C14—C3 | -179.31 (13) |
| C4—C5—C6—C1 | -1.2 (2) | C2—C3—C14—O2 | 2.2 (2) |
| N1—C1—C6—N2 | -0.08 (16) | C4—C3—C14—O2 | -179.15 (14) |
| C2—C1—C6—N2 | 178.25 (12) | C2—C3—C14—O1 | -177.08 (13) |
| N1—C1—C6—C5 | -177.79 (13) | C4—C3—C14—O1 | 1.5 (2) |
| C2—C1—C6—C5 | 0.5 (2) | C14—O1—C15—C16 | -155.28 (14) |
| C1—N1—C7—N2 | -0.93 (16) | C6—N2—C17—C18 | -81.03 (16) |
| C1—N1—C7—C8 | 176.36 (13) | C7—N2—C17—C18 | 112.77 (16) |
| C6—N2—C7—N1 | 0.91 (16) | N2—C17—C18—C19 | -175.90 (12) |
| C17—N2—C7—N1 | 168.90 (13) | C22—N3—C19—C18 | -99.29 (17) |
| C6—N2—C7—C8 | -176.38 (13) | C20—N3—C19—C18 | 75.53 (18) |
| C17—N2—C7—C8 | -8.4 (2) | C17—C18—C19—N3 | 64.76 (16) |
| N1—C7—C8—C9 | 144.90 (15) | C22—N3—C20—C21 | -0.74 (16) |
| N2—C7—C8—C9 | -38.1 (2) | C19—N3—C20—C21 | -176.49 (13) |
| N1—C7—C8—C13 | -35.4 (2) | N3—C20—C21—N4 | 0.52 (17) |
| N2—C7—C8—C13 | 141.59 (14) | C22—N4—C21—C20 | -0.09 (18) |

supplementary materials

| | | | |
|----------------|-------------|---------------|-------------|
| C13—C8—C9—C10 | -0.3 (2) | C21—N4—C22—N3 | -0.41 (17) |
| C7—C8—C9—C10 | 179.35 (13) | C20—N3—C22—N4 | 0.74 (17) |
| C8—C9—C10—C11 | -0.4 (2) | C19—N3—C22—N4 | 176.37 (13) |
| C9—C10—C11—O4 | 178.02 (14) | C11—O4—C23—O3 | 13.14 (15) |
| C9—C10—C11—C12 | 0.0 (2) | C12—O3—C23—O4 | -12.24 (15) |

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 and Cg4 are the centroids of C11/C12/O3/C23/O4 and C1—C6 rings, respectively.

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-------------------------------------|-------|-------------|-------------|---------------|
| C2—H2A \cdots O3 ⁱ | 0.95 | 2.48 | 3.3922 (17) | 162 |
| C15—H15B \cdots O4 ⁱⁱ | 0.99 | 2.49 | 3.213 (2) | 130 |
| C23—H23A \cdots N4 ⁱⁱⁱ | 0.99 | 2.43 | 3.379 (2) | 159 |
| C10—H10A \cdots Cg4 ^{iv} | 0.95 | 2.65 | 3.3005 (16) | 126 |
| C16—H16C \cdots Cg1 ^v | 0.98 | 2.91 | 3.7282 (19) | 142 |

Symmetry codes: (i) $x-1/2, y, -z+3/2$; (ii) $x-1, y, z$; (iii) $-x+2, -y+2, -z+1$; (iv) $-x, y+3/2, -z+3/2$; (v) $-x+1, -y+2, -z+1$.

Fig. 1

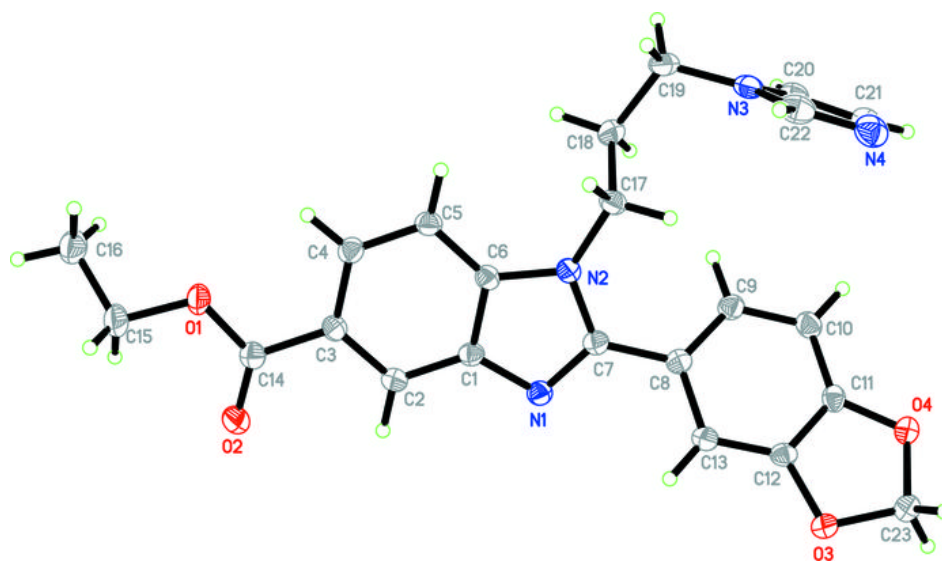


Fig. 2

