



# Article Daphnauranins C–E, Three New Antifeedants from Daphne aurantiaca Roots

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Received: 11 September 2018; Accepted: 20 September 2018; Published: 21 September 2018

**Abstract:** Daphnauranins C–E (compounds 1–3), two sesquiterpenoids and one monoterpenoid were isolated from the roots of *Daphne aurantiaca* Diels. Daphnauranin C is a 9-O-13 etherified and hydroperoxy-substituted guaiane sesquiterpenoid, daphnauranin D is a guaiane sesquiterpenoid ketal, and daphnauranin E is a monoterpenoid lactone. Their structures were elucidated by comprehensive analyses of MS, 1D NMR, and 2D NMR spectroscopic data. In an anti-feeding activities test, daphnauranins C–E showed activity against male fruit fly with anti-feeding indexes (AI) up to 39.1, 39.2, and 27.8% respectively, at 1 mM.

Keywords: Daphne aurantiaca; Thymelaeceae; sesquiterpenoid; antifeedant

# 1. Introduction

With the increase of global population and the changes in food-consumption habits, the food production industry is under unprecedented pressure. One of the biggest threats in agriculture are insects, which cause enormous production losses. To solve this problem there is a need for continuous effective prophylaxis and treatment. Therefore, finding and developing new effective and ecologically safe insecticides and insect repellents is one of the most important science topics for chemical and biological researchers in agriculture [1]. With multifarious natural products, traditional herbs have a long history of use as insecticides and insect repellents [2,3]. Natural products have become a treasury to explore for new insecticides or lead compounds with incomparable excellent structural diversity, ecological safety, biodegradability, and sustainability properties [4–7]. Daphne aurantiaca Diels., a traditional herb, was used as the raw material to make insect repellent paper by Tibetan Buddhists [8]. In previous studies, this plant showed some different natural products from other species. A few diterpenoids and several sesquiterpenoids with special skeletons and insect repellent activities were isolated from the stems of this plant [9–12]. To seek new insecticide and insect repellent from natural products, herein we studied the D. aurantiaca roots and found three new compounds, a 9-O-13 etherifie and hydroperoxy-substituted guaiane sesquiterpenoid [daphnauranin C (1)], a guaiane sesquiterpenoid ketal [daphnauranin D (2)], a monoterpenoid lactone [daphnauranin E (3)], and two known compounds: auranticanol C (4) and auranticanol F (5) (Figure 1). The isolation process and structural elucidation details of these new compounds, as well as their anti-insect assay results, are described in this paper.



Figure 1. The structures of compounds 1–5.

### 2. Results and Discussion

Daphnauranin C (1) was isolated as a colorless amorphous powder. Its molecular formula of  $C_{15}H_{22}O_4$  was determined by the HRESIMS ion at m/z 289.1408 [M + Na]<sup>+</sup> (calcd for  $C_{15}H_{22}O_4Na$ , 289.1415) (Supplementary Materials), indicative of five degrees of unsaturation. The IR spectrum revealed the presence of hydroxyl (3441 cm<sup>-1</sup>) and double bond (1629 and 1661 cm<sup>-1</sup>) absorptions. The <sup>1</sup>H-NMR spectrum (Table 1) of compound 1 exhibited signals of two methyls [ $\delta_{\rm H}$  1.36 (3H, s, H-14) and 0.92 (3H, d, J = 6.9 Hz, H-15)], two oxygenated methylene protons ( $\delta_{\rm H}$  4.54 (1H, d, J = 13.2 Hz, Ha-13) and 4.49 (1H, d, J = 13.2 Hz, Hb-13), and two olefinic protons ( $\delta_{\rm H}$  5.32 (1H, d, J = 1.3 Hz, Ha-12) and 5.11 (1H, d, J = 1.3 Hz, Hb-12). The <sup>13</sup>C-NMR and DEPT spectroscopic data (Table 1) showed 15 carbon resonances, including two methyls, six methylenes (one olefinic and one oxygenated), two methines, and five quaternary carbons (three olefinic, two oxygenated). According to a comparison of the corresponding NMR data, compound 1 was similar to auranticanol C (4) [11], a rare peroxyhydroxyl-substituted guaiane sesquiterpenoid, except for the markedly different shifts at  $\delta_{\rm C}$ 75.8 (s, C-1), 43.4 (t, C-8), 112.4 (s, C-9), and 96.3 (s, C-10) instead of  $\delta_{\rm C}$  106.7 (s, C-1), 199.1 (s, C-8), 126.1 (s, C-9), and 166.8 (s, C-10) in auranticanol C, indicating compound 1 was generated from auranticanol C by reductive deoxygenation of C-8 and etherification of C-9 to C-13. The determined  $C_{15}H_{22}O_4$ molecular formula and the key HMBC correlations from H-14 to C-1 and C-10, from H-8 to quaternary carbon C-10, and from H-8 and H-13 to quaternary carbon C-9 verified this hypothesis. The other <sup>1</sup>H-<sup>1</sup>HCOSY and HMBC correlations (Figure 2) also confirmed this atom connectivity. The relative configuration of compound 1 was determined to be the same as that of auranticanol C (Figure 2) by ROESY cross-peaks H-3 $\alpha$  [ $\delta_{\rm H}$  1.32 (1H, m)] /H-5 [ $\delta_{\rm H}$  2.46 (1H, m)], H-15/H-3 $\beta$  [ $\delta_{\rm H}$  1.60 (1H, m)], H-15/H-6 $\beta$  ( $\delta_{\rm H}$  1.69 (1H, dd, J = 5.2, 13.2 Hz), and H-6 $\beta$ /H-12. Thus, the structure of compound 1 was assigned as shown in Figure 1 and it was named daphnauranin C.

	Compound					
	1		2		3	
No.	$\delta_{ m H}$ mult. (J in Hz)	δ <sub>C</sub>	$\delta_{ m H}$ mult. (J in Hz)	$\delta_{\rm C}$	$\delta_{ m H}$ mult. (J in Hz)	$\delta_{\rm C}$
1	-	75.8s	-	141.0s		178.2s
2	1.68 m 1.63 m	31.7t	1.65 m 1.40 m	32.5t	5.68 s	112.8d
3	1.60 m 1.32 m	31.5t	2.33 m 2.19 m	31.7t		86.8s
4	1.96 m	37.6d	1.88 m	38.4d	1.96 dd (2.4, 15.6) 1.52 dd (3.2, 15.6)	47.2t
5	2.46 m	36.8d	2.18 m	39.7d	4.31 dddd (2.2, 2.4, 3.2, 3.5)	66.7d
6	1.69 dd (5.2, 13.2) 1.54 dd (11.6, 13.2)	32.4t	1.81 dd (1.3, 14.4) 1.66 m	29.2d	2.48 dd (2.2, 15.4) 1.77 dd (3.5, 15.4)	45.6t
7	-	78.9s	-	78.9s		35.9s
8	2.49 d (14.0) 1.86 d (14.0)	43.4t	-	107.5s		182.6s
9	-	112.4s	2.95 d (17.2) 2.26 d (17.2)	33.6t	1.77 s	26.9q
10	-	93.6s	-	120.2s	1.46 s	26.4q
11	-	154.3s	-	152.4s	1.26 s	30.6q
12	5.32 d (1.3) 5.11 d (1.3)	107.6t	5.17 d (1.3) 5.00 d (1.3)	104.6t		
13	4.54 d (13.2) 4.49 d (13.2)	70.0t	4.50 d (13.1) 4.23 d (13.1)	67.1t		
14	1.36 s	25.9q	1.76 s	23.9q		
15	0.92 d (6.9)	14.4q	0.88 d (7.1)	15.6q		
-OMe			3.17 s	48.7q		

Table 1. <sup>1</sup>H (400 MHz) and <sup>13</sup>C NMR (100 MHz) Data of Compounds 1–3 (in CDCl<sub>3</sub>).



**Figure 2.** Key <sup>1</sup>H-<sup>1</sup>H COSY (–), HMBC (H $\rightarrow$ C), and ROESY ( $\leftrightarrow$ ) correlations of 1–3.

Daphnauranin D (2) was obtained as a colorless oil, and possessed the molecular formula  $C_{16}H_{24}O_3$  based on HRESIMS (m/z 287.1620 [M + Na]<sup>+</sup>, calcd for  $C_{16}H_{24}O_3$ Na, 287.1623) with five degrees of unsaturation (Supplementary Materials). The <sup>13</sup>C-NMR and DEPT spectroscopic data (Table 1) showed 16 carbon resonances classified into two methyls, five methylenes (one olefinic and one oxygenated), two methines, and five quaternary carbons (three olefinic and two oxygenated). Detailed inspection of spectral data of **2** suggested that it was similar to auranticanol F (5) (also isolated from stems of *D. aurantiaca*) [11], except for an additional methoxy group ( $\delta_C$  48.7 q) in compound **2**. Compound **2** should thus be generated from auranticanol F (5) via C-8 methoxylation as shown, which was further supported by the HMBC correlations (Figure 1) from H-OMe [ $\delta_H$  3.17 (3 H, s)] to C-8 [ $\delta_C$  107.5 (s)] with the aid of its molecular formula  $C_{16}H_{24}O_3$ . The configuration 7- $\alpha$ OH and 8- $\beta$ OMe in **2** was deduced from its ROESY cross-peaks (Figure 2) H-6 $\beta$  [ $\delta_H$  1.81 (1H, dd, J = 1.3, 14.4 Hz)] to H-8-OMe and H-12 [ $\delta_H$  5.17 (1H, d, J = 1.3 Hz)]. The configuration 4- $\alpha$ H, 5- $\beta$ H were determined by ROESY cross-peaks (Figure 2) H-15 [ $\delta_H$  0.88 (3H, d, J = 7.1 Hz)] to H-6 $\beta$  and H-5 [ $\delta_H$  2.18 (1H, m)] with molecular model and comparison of 1D and 2D NMR data to those of auranticanol F. Thus, the structure of compound **2** was assigned as shown and named daphnauranin D.

The molecular formula of daphnauranin E (3), was determined as  $C_{11}H_{16}O_3$ , with four degrees of unsaturation, from HREIMS (m/z 196.1102 [M]<sup>+</sup>, calcd for C<sub>11</sub>H<sub>16</sub>O<sub>3</sub>, 196.1099 (Supplementary Materials). The <sup>13</sup>C-NMR and DEPT spectroscopic data (Table 1) showed 11 carbon resonances, including three methyls, two methylenes, two methines (one olefinic and one oxygenated), and four quaternary carbons (one olefinic, one oxygenated, and one carbonyl). By comparing the molecular formula and NMR data (Table 1) with  $\Delta^2$ -8-*m*-menthenecarboxylic acid [13], compound **3** was proposed to be derived from this compound via the oxygenation of C-3 and C-5 to a hydroxyl or ester group. The key HMBC correlations (Figure 2) from H-5 [ $\delta_{\rm H}$  4.31 (1H, dddd, J = 2.2, 2.4, 3.2, 3.5 Hz)] to C-1 [ $\delta_{\rm C}$ 178.2 (s)], from H-4 [ $\delta_{\rm H}$  3.70 (1H, dd, J = 2.4, 15.6 Hz) and 3.26 (1H, dd, J = 3.2, 15.6 Hz)] to C-3 [ $\delta_{\rm C}$  86.8 (s)], and from H-2 [ $\delta_{\rm H}$  5.68 (1H, m)] to C-3, and <sup>1</sup>H <sup>1</sup>H COSY correlations between H-4/H-5 [ $\delta_{\rm H}$  4.31 (1H, dddd, J = 2.2, 2.4, 3.2, 3.5 Hz) and H-5/H-6 [ $\delta_{H}$  2.48 (1H, dd, J = 2.2, 15.4 Hz) and 1.77 (1H, dd, J = 3.5, 15.4 Hz] in compound **3** (Figure 1) supported the assignment. The linkage of 3-O-8 lactone in compound **3** was determined by the <sup>13</sup>C NMR data of C-3 and C-11 [ $\delta_{\rm C}$  182.6 (s)] and the determined molecular formula  $C_{11}H_{16}O_3$ . The relative configuration of compound **3** was determined as  $\beta$ -Me-11 and  $\alpha$ -6-OH based on its NOESY NMR spectrum (Figure 2) revealing the key NOE of H-5/H-11 [ $\delta_{\rm H}$ 1.26 (3H, s)]. Thus, the structure of compound daphnauranin E (3) was assigned as shown.

Compounds 1–3 were all tested for their anti-feeding activities against male fruit fly (*Drosophila melanogaster*) as described before [9,14]. As a result, the anti-feeding index (AI) of 1–3 (at the concentration of 1 mM) were  $39.1 \pm 5.8$ ,  $39.2 \pm 3.9$  and  $27.8 \pm 6.2\%$ , respectively. Meanwhile AI of the negative control and the positive control (nicotine at 1 mM) were  $17.9 \pm 2.4$  and  $28.5 \pm 3.9\%$ , respectively (Table 2).

Compounds	AI (%)	
1	$39.1\pm5.8$	
2	$39.2\pm3.9$	
3	$27.8\pm6.2$	
Blank Control	$17.9\pm2.4$	
Nicotine (Positive Control)	$28.5\pm3.9$	

Table 2. Anti-feeding index (AI) of compounds 1–3 against to male fruit fly (*D. melanogaster*).

# 3. Experimental

#### 3.1. General Information

Optical rotations were acquired on a SEAP-300 polarimeter (Horiba, Kyoto, Japan) ECD were obtained on a Chirascan instrument (Applied Photophysics Ltd., Surrey, UK), UV spectra were

measured on a UV 210A spectrophotometer (Hitachi, Hong Kong, China), IR spectra were measured on a FTS-135 spectrometer (Bio-Rad, Hercules, CA, USA) with KBr pellets. 1D and 2D NMR spectra were obtained using an AV400 or DRX-500 instrument (Bruker, Billerica, MA, USA) with TMS as an internal standard, and ESIMS, HRESIMS, and HREIMS were recorded with a Bruker HCT/Esquire (Billerica, MA, USA), VG Auto Spec-3000 mass spectrometer (Manchester, UK) or Autospec Premier spectrometer (Waters, Milford, MA, USA). Column chromatography (CC) was performed on silica gel (200–300 mesh, Qingdao Marine Chemical Inc., Qingdao, China), ODS (40–70 µm, Fuji Silysia Chemical Ltd., Nagoya, Japan), and Sephadex LH-20 (GE Healthcare Bio-Sciences AB, Uppsala, Sweden). Fractions were monitored by TLC and heating after spraying with 7% H<sub>2</sub>SO<sub>4</sub> in EtOH.

# 3.2. Plant Material

The roots of *Daphne aurantiaca* Diels. were obtained in Shangri-La, Yunnan Province, People's Republic of China in July 2014,. The voucher specimen (HUANG0008) identified by Prof. Dr. Y. Niu (Kunming Institute of Botany, Chinese Academy of Sciences) was deposited at the Hainan Key Laboratory for Research and Development of Natural Products from Li Folk Medicine, Institute of Tropical Bioscience and Biotechnology, Chinese Academy of Tropical Agriculture Sciences, Haikou, People's Republic of China.

#### 3.3. Extraction and Isolation

The air-dried roots of *D. aurantiaca* (2.5 kg) were powdered and extracted with 95% EtOH by refluxing for three hours (3  $\times$  13 L). The combined EtOH solution was concentrated with a rotary evaporator followed by suspension in water (2 L) and successive extraction with EtOAc (3  $\times$  5 L). The EtOAc extract (143 g) was first subjected to silica gel (200–300 mesh,  $\phi$  16 × 150 cm) CC eluted with CHCl<sub>3</sub>/MeOH (from 50:1 to 1:1, v/v) to obtain fractions A–C. Fraction B (3.8) was chromatographed repeatedly over a ODS (40–70  $\mu$ m,  $\phi 4 \times 18$  cm) with MeOH/H<sub>2</sub>O (gradient elution with 20, 30, 40, 50, 60, 70, 80, and 90%, each 500 mL) and Sephadex LH-20 CC (MeOH as solvent) to yield auranticanol F (5, 23.2 mg), respectively. Fraction C (43 g) was then subjected to CC over silica gel (200–300 mesh,  $\phi 6 \times 45$  cm) eluted with petroleum ether/acetone (from 3:1 to 0.5:1, v/v) to give four fractions C1–C4. Fraction C2 (3.8) was chromatographed repeatedly over a ODS (40–70 μm,  $\phi 4 \times 18$  cm) with MeOH/H<sub>2</sub>O (gradient elution with 20, 30, 40, 50, 60, 70, 80, and 90%, each 500 mL) to obtain fractions C2a-C2d. Fractions C2a-C2d was chromatographed repeatedly over Sephadex LH-20 CC, using MeOH as solvent to yield 1 (5.9 mg), 2 (7.9 mg), and 3 (4.3 mg), respectively. Fractions C3 (1.6 g) was chromatographed repeatedly over a ODS (40–70  $\mu$ m,  $\phi$  4  $\times$  18 cm) with MeOH/H<sub>2</sub>O (gradient elution with 20, 30, 40, 50, 60, 70, 80, and 90%, each 500 mL) and Sephadex LH-20 CC (MeOH as solvent) to yield auranticanol C (4, 5.2 mg).

# 3.3.1. Daphnauranin C (1)

Colorless amorphous powder;  $[\alpha]_D^{18.2}$ + 36.84 (*c* 0.230, MeOH); UV (MeOH)  $\lambda_{max}$  (log $\varepsilon$ ) 202 (3.46); IR (KBr)  $\nu_{max}$  3441, 2961, 2933, 2872, 1629, 1661, 1439, 1377, 1341, 1305, 1284, 1160, 1048, 1034, 975, 931, 897; <sup>1</sup>H- and <sup>13</sup>C-NMR data see Table 1; ESIMS positive m/z [M + Na]<sup>+</sup> 289 (80); HRESIMS m/z [M + Na]<sup>+</sup> 289.1408 (calcd for C<sub>15</sub>H<sub>22</sub>O<sub>4</sub>Na, 289.1415).

# 3.3.2. Daphnauranin D (2)

Colorless oil;  $[\alpha]_D^{17.0}$ +2.05 (*c* 0.114, MeOH); ECD (MeOH)  $\Delta \epsilon$  273 (+1.15), 356 (-0.094); UV (MeOH)  $\lambda_{max}$  (log $\epsilon$ ) 203 (3.70); IR (KBr)  $\nu_{max}$  3426, 2955, 2932, 2871, 1712, 1631, 1455, 1434, 1378, 1113, 1035; <sup>1</sup>H- and <sup>13</sup>C-NMR data see Table 1; ESIMS positive m/z [M + Na]<sup>+</sup> 287 (40); HRESIMS m/z [M + Na]<sup>+</sup> 287.1620 (calcd for C<sub>16</sub>H<sub>24</sub>O<sub>3</sub>Na, 287.1623).

Colorless amorphous powder;  $[\alpha]_D^{16.4}$ -23.43 (*c* 0.327, MeOH); UV (MeOH)  $\lambda_{max}$  (log $\varepsilon$ ) 361 (1.87), 212 (3.57); IR (KBr)  $\nu_{max}$  3433, 2962, 2931, 2875, 1734, 1627, 1454, 1379, 1264, 1232, 1165, 1051, 1030, 965; <sup>1</sup>H- and <sup>13</sup>C-NMR data see Table 1; ESIMS positive m/z [M + H]<sup>+</sup> 197 (45); HREIMS m/z [M]<sup>+</sup> 196.1102 (calcd for C<sub>11</sub>H<sub>16</sub>O<sub>3</sub>, 196.1099).

# 3.4. Anti-Feeding Activity Bioassay

The anti-feeding activity was tested on male fruit fly (*Drosophila melanogaster* supplied by JoeKai Biotech LLC, Bejing, China) by the feeding counting method as reported in the literature [9,14]. Test compounds, positive control nicotine (98%, Sigma-Aldrich Corporation, St. Louis, MO, USA), and negative control DMSO (dimethylsulfoxide, Sinopharm Chemical Reagent Co., Ltd., Shanghai, China) were dissolved in DMSO to 100 mM, and then diluted to 1 mM with 4% red sugar water. Fifty starved (17 h) male fruit flies were put into one tube for one treatment. Each test was carried out with nine replicates. After 7 min feeding, the number of rubescent abdomen and no-rubescent abdomen fruit fly was counted. The feeding index (FI) was the percentage of rubescent abdomen fruit fly. Thus, the anti-feeding index (AI) was calculated by the following equation AI  $\% = [(1 - FI) \times 100]$ .

# 4. Conclusions

Previously, a series of sesquiterpenoids with skeletal diversity revealed the chemical diversity of the *Daphne* genus. The special substituted compounds isolated from *Daphne aurantiaca* also showed the accessibility of biosynthesis. The bioactivity evaluation assay showed that daphnauranins C-E have prominent anti-feeding activities against male fruit fly. Therefore, these three new compounds may be used for potential agricultural chemical development.

**Supplementary Materials:** The following are available online at http://www.mdpi.com/1420-3049/23/10/2429/s1.

Author Contributions: S.Z.H.: Write and revise the manuscript. Q.Y.M.: Activity test. Q.W.: Compound isolation. H.F.D.: Method adviser. Y.Q.L.: Method adviser. J.Z.: theory and orientation adviser. Y.X.Z.: modification and theory adviser.

Funding: This research received no external funding.

Acknowledgments: This work was supported by China Agriculture Research System (CARS-21), Financial Fund of the Ministry of Agriculture and Rural Affairs, P.R. of China (NFZX2018), National Natural Science Foundation of China (No. 31300294), and Central Public-interest Scientific Institution Basal Research Fund for Chinese Academy of Tropical Agricultural Sciences (17CXTD-15, 1630052016008). The authors thank Y.L. Huang (dept. of clinical development. Pfizer (China). Shanghai) for initial proofreading of this paper.

Conflicts of Interest: The authors declare no conflict of interest.

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**Sample Availability:** Samples of the compounds Daphnauranins B and E, auranticanol F are available from the authors.



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