

Synthesis, Molecular Docking and Cytotoxic Activity Evaluation of Organometallic Thiolated Gold(I) Complexes

Zeinab Faghih^a, Alireza Yazdani Kachoei^{b, c}, Hossein Alizadeh^{b, c}, Suphia Emamdoost^c, Shima Shirkhan^c and Masood Fereidoonzhad^{b, d*}

^aPharmaceutical Sciences Research Center, Shiraz University of Medical Sciences, Shiraz, Iran. ^bToxicology Research Center, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, Iran. ^cStudent Research Committee, Ahvaz Jundishapur University of Medical Sciences, Ahvaz, Iran. ^dDepartment of Medicinal Chemistry, School of Pharmacy Ahvaz Jundishapur University of Medical Sciences, Ahvaz, Iran.

Abstract

The complex [(PhCH₂NC)AuCl], **1**, was prepared by the reaction of [(Me₂S)AuCl], **A**, with an equimolar amount of benzyl isocyanide (PhCH₂NC) ligand. Through a salt metathesis reaction, the chloride ligand in **1** was replaced by potassium benzothiazole-2-thiolate (Kbt) and potassium benzoimidazole-2-thiolate (Kbi) to afford complexes (PhCH₂NC)Au(κ¹-S-bt), **2a** and (PhCH₂NC)Au(κ¹-S-bi), **2b**, respectively, which were characterized by NMR spectroscopy. The cytotoxic activities of **2a** and **2b** were evaluated against three human cancer cell lines, including A549 (lung), SKOV3 (ovary), and MCF-7 (breast). Our results indicated that **2a** exhibited comparable cytotoxicity on investigated cell lines with cisplatin. It showed a good anti-proliferative activity with IC₅₀ of 19.46, 11.76 and 13.27 μM against A549, SKOV3 and MCF-7 cell lines, respectively. The effects of these complexes on the proliferation of the non-tumorigenic epithelial breast cell line (MCF-10A) showed their good selectivity between the tumorigenic and non-tumorigenic cell lines. Molecular docking simulation studies were also conducted to determine the specific binding site and binding mode of the synthesized gold complexes to DNA and thioredoxin reductase (TrxR) as their proposed targets.

Keywords: Synthesis; Gold(I) complexes; Thiolate ligands; Molecular docking; Cytotoxic activity.

Introduction

The commercially available platinum-based drugs such as cisplatin, carboplatin, and oxaliplatin suffer from severe side effects and resistance in a wide range of cancers (1, 2). Recently, gold-based complexes, such as auronofin, an anti-rheumatic drug, have received great attentions in the treatment of cancers due to their potent anti-proliferative

activity against wide variety of cancer cell lines with different mechanisms of action compared to Pt-based drugs (3).

Two significant oxidation states of gold is +1 and +3 (4). Au(I) as the most stable oxidation state of gold forms usual linear complexes by coordination of two ligands (5). The nature of the coordinating ligands (donor atoms) has an important effect in the stability of the gold(I) complexes (6-9). These complexes usually contain soft donor atoms such as sulfur (10-13), carbon (14-20), and

*Corresponding author:
E-mail: fereidoonzhad-m@ajums.ac.ir

phosphorus (10, 21-23).

Thiol group (RSH) and their corresponding anion forms, called thiolate (RS⁻), as a chemically unique ligands, can be used as a source of sulfur donor ligands in a variety of organometallic complexes (24-27). To date, different thiolate gold(I) complexes especially the type of L–Au^I–SR have been synthesized, so that ligand(L) could be different neutral ligands such as isocyanides (28-31), phosphine (13, 29, 32 and 33), and N-heterocyclic carbenes (NHCs) (34-36). Generally, the presence of thiolate ligands and various L donor ligands in their structure greatly affects their biological activity (13, 32, 34 and 36).

Recent studies demonstrated that the thioredoxin reductase (TrxR) which catalyzed the NADPH-dependent reduction of oxidized thioredoxins, is an effective target for the development of novel antitumor agents (37, 38). Several cytotoxic gold compounds, both gold(I) or gold(III), are potent TrxR inhibitors (39). These enzymes were introduced as the main targets of anticancer gold agents (37). The interaction of the gold complexes with DNA was also proposed for their anticancer properties (40, 41).

Here, in this study, the synthesis, structural characterization, and cytotoxic activity of two new Au(I) complexes with benzyl isocyanide and thiolate ligands (benzothiazole-2-thiolate (bt) and benzoimidazole-2-thiolate (bi)) were demonstrated. The cytotoxic activity was evaluated on three different cancer cell lines including human lung (A549), ovarian (SKOV3), and breast (MCF-7) cancer cell lines. Molecular docking simulation studies were also conducted to determine the specific binding site and binding mode of the synthesized gold complexes to DNA and thioredoxin reductase (TrxR) as their proposed targets.

Experimental

Materials and methods

General procedures and materials

All reactions were carried out under a nitrogen atmosphere using standard Schlenk techniques. NMR spectra (¹H and ¹³C{¹H}) were recorded on a Bruker Avance DPX 400 MHz instrument and referenced to the residual peak of the solvent, *i.e.* CDCl₃ (¹H and ¹³C).

The chemical shifts (δ) were reported as ppm and coupling constants (J) were expressed in Hz. The melting point values were measured by a Buchi 510. The microanalyses were performed using a vario EL CHNS elemental analyzer. Benzyl isocyanide (PhCH₂NC), benzothiazole-2-thiol (Hbt), and benzoimidazole-2-thiol (Hbi) as well as all the solvents were purchased from Aldrich and used without further purification. Complex [(Me₂S)AuCl], **A**, was prepared according to literature method (40).

Synthesis of potassium benzothiazole-2-thiolate (Kbt)

A solution of benzothiazole-2-thiol (Hbt, 373 mg, 2.23 mmol) in MeOH (10 mL) was added to a solution of KOH (125 mg, 2.23 mmol) in MeOH (5 mL). The resulting yellow solution was stirred at room temperature for 1 h, and then the solvent was completely evaporated. The residue was treated with ⁱPrOH (2 mL) and the resulting yellow solid was filtered and dried. This procedure was also used for preparation of potassium benzoimidazole-2-thiolate (Kbi).

Synthesis of [(PhCH₂NC)AuCl], **1**

To a solution of [(Me₂S)AuCl], **A** (200 mg, 0.68 mmol) in CH₂Cl₂ (20 mL), 1 equivalent of PhCH₂NC (83 μ L, 0.68 mmol) was added. The mixture was stirred at room temperature for 1 h and then concentrated (~1 mL) under vacuum, and *n*-pentane (5 mL) was added to give **1** as a white solid, which was filtered and washed with *n*-pentane (2 \times 3 mL) and dried. Yield: 197 mg, 83%; m.p. = 138 °C. Elem. Anal. Calcd. for C₈H₇AuClN (349.57): C, 27.49; H, 2.02; N, 4.01. Found: C, 27.61; H, 2.05; N, 4.06. IR (KBr, cm⁻¹): 2260 (s, $\nu_{C=N}$). NMR data in CDCl₃: δ (¹H) 4.87 (s, 2H, H^e), 7.35 (dd, ³J_{HH} = 7.7 Hz, ⁴J_{HH} = 1.4 Hz, 2H, H^a), 7.46-7.43 (m, 3H, H^b and H^c); δ (¹³C{H}) 48.2 (t, ¹J_{CN} = 7 Hz, C^e), 127.5 (s, 2C, C^a), 129.4 (s, C^d), 129.6 (s, 2C, C^b), 129.7 (s, C^c), 135.6 (t, ¹J_{CN} = 26 Hz, C^f).

Synthesis of (PhCH₂NC)Au(κ^1 -S-bt)], **2a**

An equimolar amount of Kbt (59.5 mg, 0.29 mmol) was dissolved in mixture of MeOH/acetone (2/8 mL) and added to a solution of **1** (100 mg, 0.29 mmol) in CH₂Cl₂ (15 mL). The reaction mixture was stirred at

room temperature for 15 h. Then, the solvent was removed under reduced pressure and the residue was extracted with CH_2Cl_2 (10 mL). The obtained colorless solution was filtered through celite and the filtrate was concentrated (~1 mL) under vacuum, and *n*-pentane (5 mL) was added to give **2** as a white solid, which was filtered and washed with *n*-pentane (3 × 3 mL) and dried. Yield: 87 mg, 74%; m.p. = 172 °C. Elem. Anal. Calcd. for $\text{C}_{15}\text{H}_{11}\text{AuN}_2\text{S}_2$ (480.35): C, 37.51; H, 2.31; N, 5.83. Found: C, 37.46; H, 2.34; N, 5.85. NMR data in CDCl_3 : δ (^1H) δ 8.04 – 7.93 (m, 2H), 7.39 (dtd, J = 24.0, 7.5, 1.5 Hz, 2H), 7.27 – 7.16 (m, 1H), 7.08 – 6.94 (m, 4H), 4.15 (t, J = 0.9 Hz, 2H). δ ($^{13}\text{C}\{\text{H}\}$) 168.9, 152.0, 143.0, 139.0, 134.6, 128.5, 128.2, 126.2, 124.5, 121.5, 121.1, 41.3.

Synthesis of $(\text{PhCH}_2\text{NC})\text{Au}(\kappa^1\text{-S-bi})$, **2b**

The synthesis procedure was as the same as **2a**. Yield: 91 mg, 75%; m.p. = 186 °C. Elem. Anal. Calcd. for $\text{C}_{15}\text{H}_{12}\text{AuN}_3\text{S}$ (463.04): C, 38.89; H, 2.61; N, 9.07. Found: C, 38.82; H, 2.67; N, 9.11. NMR data in CDCl_3 : δ (^1H) δ 7.98 – 7.87 (m, 2H), 7.47 – 7.40 (m, 1H), 7.25 – 7.12 (m, 3H), 7.08 – 6.94 (m, 4H), 4.15 (t, J = 0.9 Hz, 2H). NMR data in CDCl_3 : δ (^1H) 153.4, 143.0, 140.4, 139.0, 137.5, 128.5, 128.2, 126.1, 124.6, 122.2, 117.4, 111.0, 41.3.

Biological Assay

Cell Lines and Cell Culture

Human cancer cell lines, MCF-7 (breast cancer), SKOV3 (ovarian cancer), and A549 (non-small cell lung cancer) were purchased from National Cell Bank of Iran (NCBI, Pasteur Institute, Tehran, Iran). All the cells were cultured in RPMI 1640 medium (Biosera), supplemented with 10% fetal bovine serum (FBS; Gibco) and 1% penicillin–streptomycin and were incubated at 37 °C in humidified CO_2 incubator. MCF10A cells (human breast epithelial cell line) were cultured in DMEM/Ham's F-12 (GIBCO-Invitrogen, Carlsbad, CA) supplemented with 100 ng/mL cholera toxin, 20 ng/mL epidermal growth factor (EGF), 0.01 mg/mL insulin, 500 ng/mL hydrocortisone, and 5% chelex-treated horse serum.

MTT Assay

Cytotoxic activities of **2a** and **2b** were evaluated using standard

3-(4,5-dimethylthiazol-yl)-2,5-diphenyl-tetrazolium bromide (MTT) assay according to a known protocol (25, 42 and 43). Briefly, the cells were harvested and plated in 96-well microplates at a density of 1×10^4 cells per well in 100 μL of complete culture medium. After 24 h of incubation, the cells were treated with five different concentrations of the gold complexes, ranging from 1 to 100 μM in triplicate manner. Each compound was dissolved in DMSO. To avoid bystander cytotoxic effect, the final concentration of DMSO was maintained at about 0.1%. Following 48 h of incubation at 37 °C in humidified CO_2 incubator, the media were completely removed and replaced with 100 μL of new media containing 0.5 mg/mL MTT solution and the plate were incubated for 3 h at room temperature. The media containing MTT were discarded, and 150 μL of DMSO was added to each well to dissolve the formazan crystals. The plates were then incubated for more 30 min at 37 °C in the dark. The absorbance of individual well was read at 492 nm using a microplate ELISA reader. The data were analyzed using Excel 2013 and CurveExpert 1.4 and the 50% inhibitory concentration of each compound was reported as IC_{50} . Each experiment was tested three times for each complex. Data are presented as mean \pm SD.

Molecular docking procedure

The four different 3D crystal structures of DNA (PDB ID: 1BNA, 1LU5, 3CO3 and 198D) and TrxR (PDB ID: 4CBQ) were retrieved from protein data bank (www.rcsb.org/pdb). Co-crystal ligands were excluded from the structures and the PDBs were checked in terms of missing atom types. Subsequently, MGLtools 1.5.6 was applied to convert these corrected PDB files to PDBQT. The structure of each gold(I) complexes was created by HyperChem Professional (Version 8, Hypercube Inc., Gainesville, FL, USA). Each complex was optimized by molecular mechanic methods (MM^+) using HyperChem 8, followed by energy minimization calculations at Hartree-Fock (HF) level, using Gaussian 09. The output structures were then converted to PDBQT using MGLtools 1.5.6. The ligands, thereafter, were docked in the active site of DNA and TrxR using an *in-house*

batch script (DOCKFACE) of AutoDock 4.2, based on Lamarckian genetic algorithm (44-47).

A grid box of $60 \times 74 \times 120$ and $40 \times 40 \times 40$ points in x, y, and z directions was built and centered on the ligand in the complex with a spacing of 0.375 Å for 1BNA and 4CBQ, respectively. Cartesian coordinate for 1BNA in x, y, and z was 15.81, 21.31, and 9.88, and for 4CBQ was -4.923, -7.115, and -22.251, respectively. Parameters of metal ions such as gold were added to the gpf and dpf files to be used in the docking calculation. Visualization of the docked pose has been performed by means of AutoDock Tools 1.5.6 and PyMOL molecular graphics program (48).

Results and Discussion

Synthesis and Characterization of Complexes

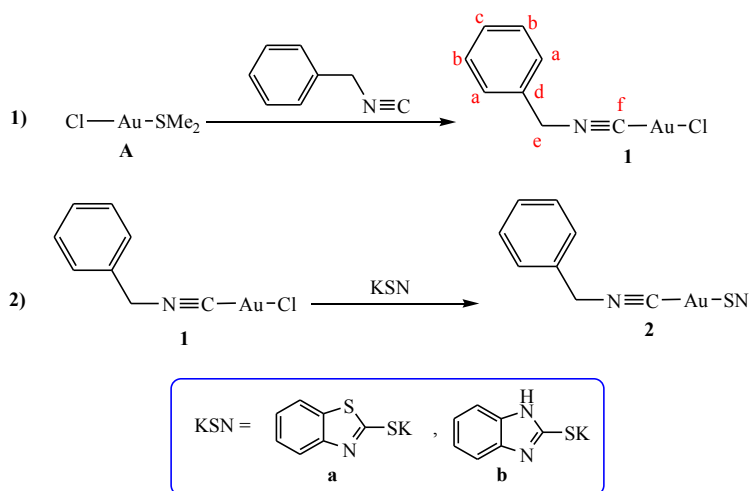
According to the Puddephatt report the precursor complex $[(\text{Me}_2\text{S})\text{AuCl}]$, **A**, was prepared (49). The SMe_2 ligand in **A** is a good leaving group and can be readily substituted by one equivalent of benzyl isocyanide (PhCH_2NC) and afforded the corresponding complex $[(\text{PhCH}_2\text{NC})\text{AuCl}]$, **1**. Complex **1** was also treated with potassium benzothiazole-2-thiolate (Kbt) and potassium benzoimidazole-2-thiolate (Kbi) in a 1:1 molar ratio and yielded complex $(\text{PhCH}_2\text{NC})\text{Au}(\kappa^1\text{-S-bt})$, **2a**,

and $(\text{PhCH}_2\text{NC})\text{Au}(\kappa^1\text{-S-bi})$, **2b**, respectively, through a salt metathesis reaction (Scheme 1). Both complexes were air-stable, colorless solids which were obtained in a good yields and characterized using NMR and elemental analysis.

The ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of **1**, **2a**, and **2b** (in CDCl_3) displayed signal resonances due to the PhCH_2NC ligand in expected regions (similar to the free ligand, with slight shifts) and a simple pattern for the thiolate ligands (50, 51). Furthermore, C^e and C^f of PhCH_2NC ligand indicated a resolving coupling with nitrogen nucleus (^{14}N) in the $^{13}\text{C}\{^1\text{H}\}$ NMR spectra of synthesized complexes which is characteristic feature for numerous isocyanide ligand and their complexes (15, 16 and 52).

Biological Activity studies

The *in-vitro* cytotoxic activity of **2a** and **2b** were evaluated on three cancer cell lines including human ovarian (SKOV3), lung (A549), and breast (MCF-7) carcinoma. As shown in Table 1, **2a**, the exhibited comparable cytotoxicity on the investigated cell lines with cisplatin. It showed a good anti-proliferative activity with IC_{50} of 19.46, 11.76, and 13.27 μM compared with those measured for cisplatin (7.78, 13.27 μM and 11.69 μM , against A549, SKOV3 and MCF-7 cell lines, respectively). **2b** also showed generally a moderate antitumor activity especially



Scheme 1. Synthetic route for preparation of **2a** and **2b**.

against MCF-7 cell line with IC_{50} of 19.14 μ M. Interestingly, **1** showed better antitumor activity against all the studied cancer cell lines compared to **2b**.

To investigate the selectivity between cancer and the normal cell line, the effects of the synthesized gold(I) complexes on the proliferation of the nontumoral cell line (MCF-10A; non-tumorigenic epithelial breast cell line) was also acquired. The results showed good selectivity between the tumorigenic and non-tumorigenic cell lines.

Molecular docking analysis

It was observed that gold acts as an anticancer agent through different mechanisms such as inhibition of the thioredoxinreductase (TrxR), glutathionereductase (GR) enzymes, and intercalation with the DNA (53). Hence, molecular docking simulation studies were conducted to determine the specific binding site and binding mode of the synthesized gold complexes to DNA and TrxR as their proposed targets.

The docking binding energies of the synthesized Au(I) complexes with DNA and TrxR targets are shown in Table 2. The lowest docking binding energies (kcal/mol) in Autodock dlG output file was considered as response in each run. As summarized in Table 2, **2a**, the best compound in the cytotoxic activity, showed also better energy regarding binding to TrxR active site compared to **2a**. These results suggest TrxR as the main target of these gold(I) anticancer agents. The ΔG_{bind} values of the best docked poses of these

compounds are within the range of -7.71 to -13.86 kcal.mol⁻¹ for DNA and -7.80 to -12.26 kcal.mol⁻¹ for TrxR. The validity of the docking procedure was maintained by re-docking of Aurofin, the co-crystal ligand of TrxR, into 3D structure of TrxR. All the docking protocols were done on validated structures with RMSD values below 2 Å.

The docked model suggested that the compound **1** interacted with the minor groove of DNA with -7.71 kcal/mol binding energy through its chloro group with A6, and the benzyl CH₂ group with T7 and T8 base pairs in the minor groove of DNA (PDB ID: 1BNA) (Figure 1a). The main interaction of **2a** was through hydrogen bonding of benzothiazole sulfur group with T7 in the minor groove of DNA (Figure 1b). **2b** interacted with the minor groove of DNA (PDB ID: 3CO3) with -8.55 Kcal/mol binding energy through its sulfur group attached to the gold atom with G9, benzothiazole nitrogen group with G9, gold atom with G7 and the benzyl CH₂ group with T8 base pairs (Figure 1c).

The most important interactions of **2a** in binding to TrxR were the interaction of gold atom with Asp287, isocyanide nitrogen group with Asp284, sulfur group attached to the gold atom with Ser299 and benzothiazole nitrogen group with Cys286 (Figure 2a). Binding mode of compound **2b** with TrxR showed that the nitrogen of isocyanide are involved in the acceptor hydrogen bonding with residue Asp287, and benzothiazole nitrogen groups with Cys286 and Arg288 (Figure 2b).

Table 1. *In-vitro* cytotoxic activity of gold complexes against cancerous and non-cancerous cell lines.

Name	IC_{50} (μ M \pm SD)			
	A549	SKOV3	MCF-7	MCF-10A
1	25.18 \pm 1.63	28.73 \pm 2.29	15.67 \pm 3.87	58.35 \pm 1.39
2a	19.46 \pm 1.15	11.76 \pm 1.49	13.27 \pm 3.37	47.16 \pm 1.28
2b	32.75 \pm 1.47	22.52 \pm 2.23	19.14 \pm 1.28	45.08 \pm 2.61
cisplatin	7.78 \pm 0.54	13.27 \pm 1.23	11.69 \pm 1.57	28.42 \pm 2.45

Table 2. Molecular docking studies of gold complexes on DNA and TrxR targets.

Ligand/Receptor	Docking binding energy (kcal/mol) ^a				
	1BNA ^b	1LU5 ^c	3CO3 ^d	198D ^e	4CBQ ^f
1	-7.71	-6.12	-6.57	-6.68	-7.80
2a	-11.76	-7.47	-8.24	-8.75	-12.26
2b	-13.86	-7.60	-8.55	-8.27	-11.69
Cisplatin	-	-4.44	-4.81	-4.71	-
Auronofin	-	-	-	-	-4.81

^aAll the docking protocols were performed on validated structures with RMSD values below 2 Å.

^bStructure of a B-DNA dodecamer.

^cCrystal structure of the asymmetric platinum complex {Pt(amine)(cyclohexylamine)}²⁺ bound to a dodecamer DNA duplex.

^dCrystal structure of a monofunctional platinum-DNA adduct, cis-{Pt(NH₃)₂(pyridine)}²⁺ bound to deoxyguanosine in a dodecamer duplex.

^eA trigonal form of the idarubicin-D(Cgatec) complex: crystal and molecular structure at 2.0 angstroms resolution.

^fCrystal structure of the thioredoxinreductase from *Entamoebahistoltytica* with auranofin Au(I) bound to Cys286.

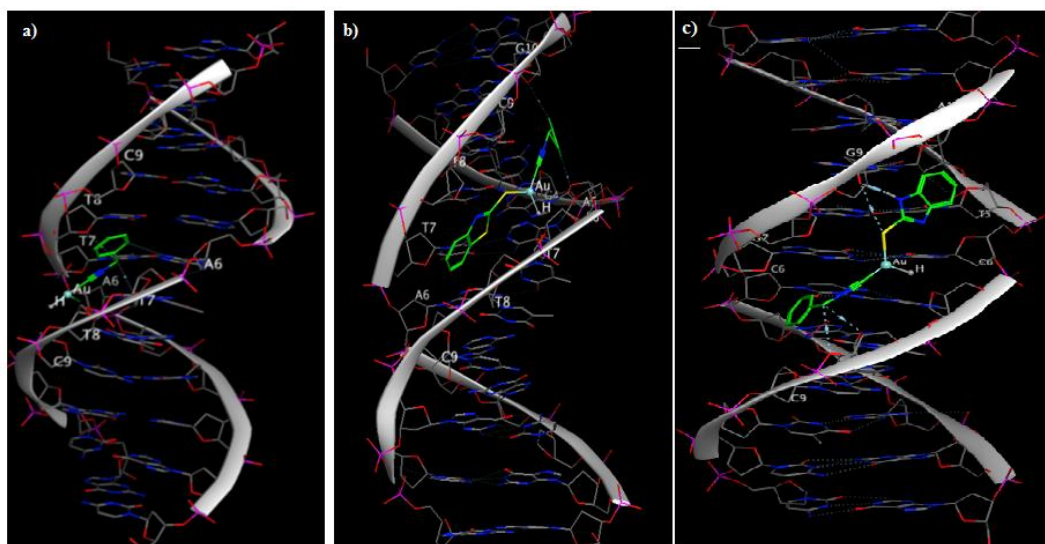


Figure 1. Molecular docking simulation studies of (a) **1**, (b) **2a** with DNA (PDB ID: 1BNA), and (c) **2b** with DNA (PDB ID: 3CO3).

Conclusion

In this study, two novel sulfur-based gold complexes are reported. Complex **1** is readily synthesized by replacement of dimethylsulfide ligand in **A** with benzyl isocyanide. In a salt metathesis reaction, an anion exchange between **1** and potassium benzothiazole-2-thiolate (Kbt) or potassium benzoimidazole-2-thiolate (Kbi), led to formation **2a** or **2b**, respectively. These complexes are fully characterized by NMR and elemental analysis. The cytotoxic activities of **2a** and **2b** against various cancer

cell lines revealed that **2a** has reasonable IC₅₀ with higher potency than **2b**. The evaluation of their cytotoxicity against the non-tumorigenic epithelial breast cell line (MCF-10A), showed good selectivity between the tumorigenic and non-tumorigenic cell lines. Complex **2a** showed a good anti-proliferative effect which is even higher than cisplatin against MCF-7 cell line.

Acknowledgements

The authors would like to thank the research deputy of Ahvaz Jundishapur University of

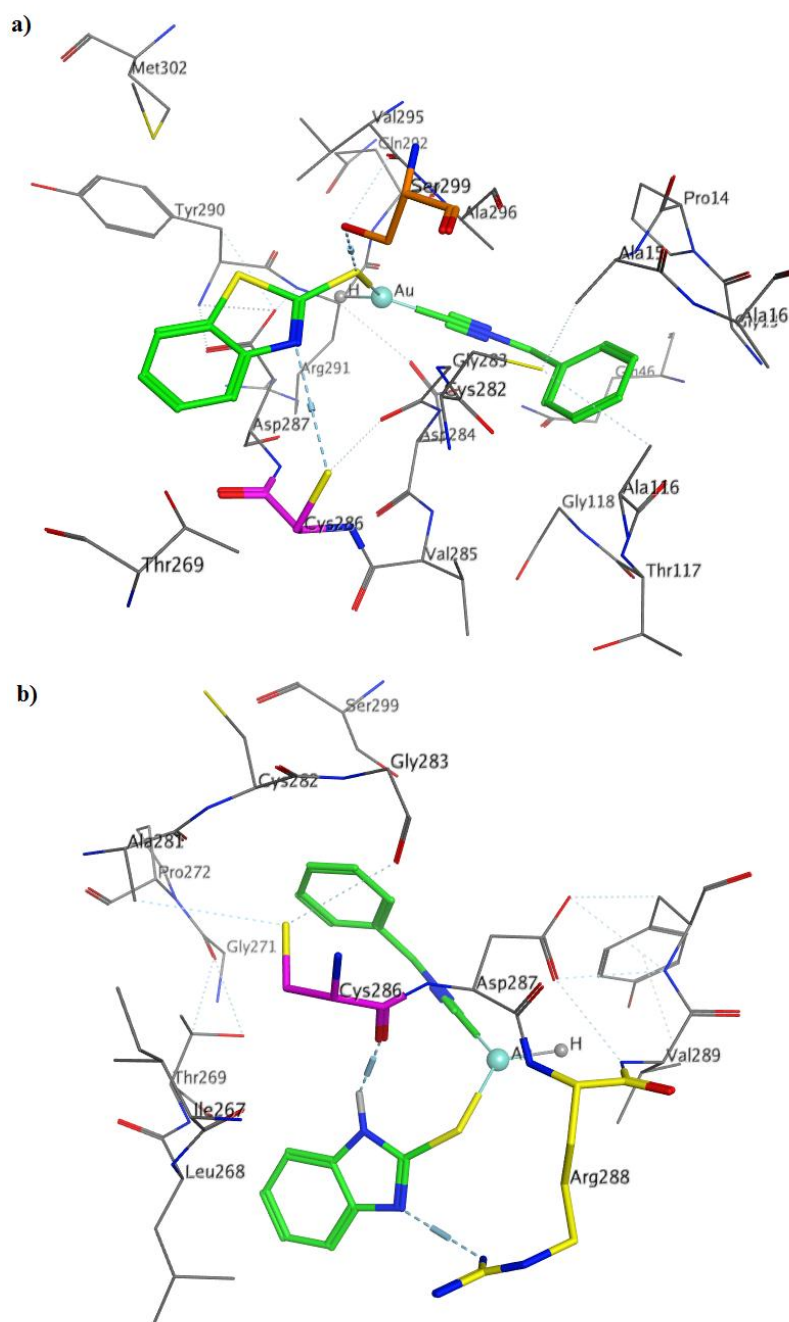


Figure 2. 3D ligand-receptor interactions of (a) **2a**, and (b) **2b** with TrxR (PDB code: 4CBQ).

Medical Sciences and Shiraz Pharmaceutical Sciences Research Center (no. 16266), who supported this work. Collaboration of Medicinal Chemistry, faculty of pharmacy, Ahvaz Jundishapur University of Medical Sciences, in providing the required facilities for this work is greatly acknowledged. This paper is extracted from the thesis of Hossein

Alizadeh (Grant No. U-96077) and Suphia Emamdoost (Grant No. B-97014).

References

- (1) Fong CW. Molecular mechanisms of cellular transport, resistance and cytotoxic side effects of platinum and adjuvant anti-cancer drugs—a molecular orbital study. *J. Comput. Chem.* (2017) 3: 185.

- (2) Santabarbara G, Maione P, Rossi A and Gridelli C. Pharmacotherapeutic options for treating adverse effects of Cisplatin chemotherapy. *Expert Opin. Pharmacother.* (2016) 17: 561-70.
- (3) Nardon C, Boscutti G and Fregona D. Beyond Platins: gold complexes as anticancer agents. *Anticancer Res.* (2014) 34: 487-92.
- (4) Raubenheimer HG and Schmidbaur H. The late start and amazing upswing in gold chemistry. *J. Chem. Educ.* (2014) 91: 2024-36.
- (5) Đurović MD, Bugarčić ŽD and van Eldik R. Stability and reactivity of gold compounds – From fundamental aspects to applications. *Coord. Chem. Rev.* (2017) 338: 186-206.
- (6) Hutchings GJ, Brust M and Schmidbaur H. Gold-an introductory perspective. *Chem. Soc. Rev.* (2008) 37: 1759-65.
- (7) Mohr F. *Gold Chemistry: Applications and Future Directions in the Life Sciences*. 2nd ed. Wiley-VCH Publisher, New Jersey (2009) 47-92.
- (8) Laguna A, *Modern Supramolecular Gold Chemistry: Gold-Metal Interactions and Applications*. Wiley-VCH Publisher, New Jersey (2008) 1-63.
- (9) Schmidbaur H. Ludwig Mond Lecture. High-carat gold compounds. *Chem. Soc. Rev.* (1995) 24: 391-400.
- (10) Hao L, Mansour MA, Lachicotte RJ, Gysling HJ and Eisenberg R. A gold(I) mononuclear complex and its association into binuclear and cluster compounds by hydrogen bonding or metal ion coordination. *Inorg. Chem.* (2000) 39: 5520-9.
- (11) Robson JA, González de Rivera F, Jantan KA, Wenzel MN, White AJP, Rossell O and Wilton-Ely JDET. Bifunctional chalcogen linkers for the stepwise generation of multimetallic assemblies and functionalized nanoparticles. *Inorg. Chem.* (2016) 55: 12982-96.
- (12) Zou T, Lum CT, Lok CN, Zhang JJ and Che CM. Chemical biology of anticancer gold(III) and gold(I) complexes. *Chem. Soc. Rev.* (2015) 44: 8786-801.
- (13) Atrián-Blasco E, Gascón S, Rodríguez-Yoldi MJ, Laguna M and Cerrada E. Novel gold(I) thiolate derivatives synergistic with 5-fluorouracil as potential selective anticancer agents in colon cancer. *Inorg. Chem.* (2017) 56: 8562-79.
- (14) Malwitz MA, Lim SH, White-Morris RL, Pham DM, Olmstead MM and Balch AL. Crystallization and interconversions of vapor-sensitive, luminescent polymorphs of $[(C_6H_{11}NC)_2Au](AsF_6)$ and $[(C_6H_{11}NC)_2Au](PF_6)$. *J. Am. Chem. Soc.* (2012) 134: 10885-93.
- (15) Schneider W, Angermaier K, Sladek A and Schmidbaur H. Ligand influences on the Supramolecular chemistry of simple gold(I) complexes: mononuclear (Isonitrile)gold(I) complexes. *Z. Naturforsch. B* (1996) 51: 790-800.
- (16) Manzano R, Rominger F and Hashmi ASK. Saturated abnormal NHC–Gold(I) complexes: synthesis and catalytic activity. *Organometallics* (2013) 32: 2199-203.
- (17) Collado A, Gomez-Suarez A, Martin AR, Slawin AMZ and Nolan SP. Straightforward synthesis of $[Au(NHC)X]$ (NHC = N-heterocyclic carbene, X = Cl, Br, I) complexes. *Chem. Commun.* (2013) 49: 5541-3.
- (18) Han Z, Bates JI, Strehl D, Patrick BO and Gates DP. Homo- and heteropolynuclear complexes containing bidentate bridging 4-phosphino-N-heterocyclic carbene ligands. *Inorg. Chem.* (2016) 55: 5071-8.
- (19) Seki T, Sakurada K, Muromoto M, Seki S and Ito H. Detailed investigation of the structural, thermal, and electronic properties of gold isocyanide complexes with mechano-triggered single-crystal-to-single-crystal phase transitions. *Chem. Eur. J.* (2016) 22: 1968-78.
- (20) Johnson A and Gimeno MC. An efficient and sustainable synthesis of NHC gold complexes. *Chem. Commun.* (2016) 52: 9664-7.
- (21) Ott I. On the medicinal chemistry of gold complexes as anticancer drugs. *Coord. Chem. Rev.* (2009) 253: 1670-81.
- (22) Joao Carlos L and Laura R. Phosphine-gold(I) compounds as anticancer agents: general description and mechanisms of action. *Anti-Cancer Agents Med. Chem.* (2011) 11: 921-8.
- (23) García-Moreno E, Gascón S, Atrián-Blasco E, Rodríguez-Yoldi MJ, Cerrada E and Laguna M. Gold(I) complexes with alkylated PTA (1,3,5-triazza-7-phosphaadamantane) phosphanes as anticancer metallodrugs. *Eur. J. Med. Chem.* (2014) 79: 164-72.
- (24) Dadkhah Aseman M, Nabavizadeh SM, Shahsavari HR and Rashidi M. C-H reductive elimination during the reaction of cycloplatinated(II) complexes with pyridine-2-thione: Kinetic follow up. *RSC Adv.* (2015) 5: 22692-702.
- (25) Fereidoonzhad M, Niazi M, Ahmadipour Z, Mirzaee T, Faghieh Z, Faghieh Z and Shahsavari HR. Cyclometalated platinum(II) complexes comprising 2-(diphenylphosphino)pyridine and various thiolate ligands: Synthesis, spectroscopic characterization and biological activity. *Eur. J. Inorg. Chem.* (2017) 2017: 2247–54.
- (26) Niazi M, Shahsavari HR, Golbon Haghghi M, Halvagar MR, Hatami S and Notash B. Carbon-sulfur bond reductive coupling from a platinum(II) thiolate complex. *RSC Adv.* (2016) 6: 95073-84.
- (27) Raper ES. Complexes of heterocyclic thionates. Part 1. Complexes of monodentate and chelating ligands.

- Coord. Chem. Rev.* (1996) 153: 199-255.
- (28) Dyadchenko VP, Belov NM, Lemenovskii DA, Antipin MY, Lyssenko KA, Bruce AE and Bruce MRM. Synthesis, crystal and molecular structure of gold(I) thiophenolate with 4'-ferrocenyl[1,1'] biphenylisocyanides. *J. Organomet. Chem.* (2010) 695: 304-9.
- (29) Römbke P, Schier A, Wiesbrock F, and Schmidbaur H. Gold(I) thiosulfonate complexes. *Inorg. Chim. Acta* (2003) 347: 123-8.
- (30) Schneider W, Bauer A and Schmidbaur H. (Isocyanide)gold(I) thiosalicylates: supramolecular assembly based on both aurophilic and hydrogen bonding. *Organometallics* (1996) 15: 5445-6.
- (31) Dyadchenko VP, Belov NM, Dyadchenko MA, Slovokhotov YL, Banaru AM and Lemenovskii DA. A complex of gold(I) benzenethiolate with isocyanide: synthesis and crystal and molecular structures. *Russ. Chem. Bull., Int. Ed.* (2010) 59: 539-43.
- (32) Gutierrez A, Cativiela C, Laguna A and Gimeno MC. Bioactive gold(I) complexes with 4-mercaptoproline derivatives. *Dalton Trans.* (2016) 45: 13483-90.
- (33) Hokai Y, Jurkowicz B, Fernández-Gallardo J, Zakirkhodjaev N, Sanaú M, Muth TR and Contel M. Auranofin and related heterometallic gold(I)-thiolates as potent inhibitors of methicillin-resistant *Staphylococcus aureus* bacterial strains. *J. Inorg. Biochem.* (2014) 138: 81-8.
- (34) Dada O, Curran D, O'Beirne C, Müller-Bunz H, Zhu X and Tacke M. Synthesis and cytotoxicity studies of novel NHC-Gold(I) pseudohalides and thiolates. *J. Organomet. Chem.* (2017) 840: 30-7.
- (35) Chia EY, Naem S, Delaude L, White AJP and Wilton-Ely JDET. Gold(I) complexes bearing mixed-donor ligands derived from N-heterocyclic carbenes. *Dalton Trans.* (2011) 40: 6645-58.
- (36) Mui YF, Fernández-Gallardo J, Elie BT, Gubran A, Maluenda I, Sanaú M, Navarro O and Contel M. Titanocene-Gold Complexes Containing N-heterocyclic carbene ligands inhibit growth of prostate, renal, and colon cancers *in-vitro*. *Organometallics* (2016) 35: 1218-27.
- (37) Bindoli A, Rigobello MP, Scutari G, Gabbiani C, Casini A and Messori L. Thioredoxin reductase: A target for gold compounds acting as potential anticancer drugs. *Coord. Chem. Rev.* (2009) 253: 1692-707.
- (38) Pia Rigobello M, Messori L, Marcon G, Agostina Cinellu M, Bragadin M, Folda A, Scutari G and Bindoli A. Gold complexes inhibit mitochondrial thioredoxin reductase: consequences on mitochondrial functions. *J. Inorg. Biochem.* (2004) 98: 1634-41.
- (39) Serebryanskaya TV, Lyakhov AS, Ivashkevich LS, Schur J, Frias C, Prokop A and Ott I. Gold(I) thiotetrazolates as thioredoxin reductase inhibitors and antiproliferative agents. *Dalton Trans.* (2015) 44: 1161-9.
- (40) Gratteri P, Massai L, Michelucci E, Rigo R, Messori L, Cinellu MA, Musetti C, Sissi C and Bazzicalupi C. Interactions of selected gold(III) complexes with DNA G quadruplexes. *Dalton Trans.* (2015) 44: 3633-9.
- (41) Marcon G, Carotti S, Coronello M, Messori L, Mini E, Orioli P, Mazzei T, Cinellu MA and Minghetti G. Gold(III) complexes with bipyridyl ligands: solution chemistry, cytotoxicity, and DNA binding properties. *J. Med. Chem.* (2002) 45: 1672-7.
- (42) Fereidoonezhad M, Kaboudin B, Mirzaee T, Babadi Aghakhanpour R, Golbon Haghghi M, Faghhih Z, Faghhih Z, Ahmadipour Z, Notash B and Shahsavari HR. Cyclometalated platinum(II) complexes bearing bidentate O,O'-Di(alkyl)dithiophosphate ligands: photoluminescence and cytotoxic properties. *Organometallics* (2017) 36: 1707-17.
- (43) Fereidoonezhad M, Niazi M, Shahmohammadi Beni M, Mohammadi S, Faghhih Z, Faghhih Z and Shahsavari HR. Synthesis, biological evaluation, and molecular docking studies on the DNA binding interactions of platinum(II) rollover complexes containing phosphorus donor ligands. *ChemMedChem* (2017) 12: 456-65.
- (44) Mojaddami A, Sakhteman A, Fereidoonezhad M, Faghhih Z, Najdian A, Khahnadideh S, Sadeghpour H and Rezaei Z. Binding mode of triazole derivatives as aromatase inhibitors based on docking, protein ligand interaction fingerprinting, and molecular dynamics simulation studies. *Res. Pharm. Sci.* (2017) 12: 21-30.
- (45) Zare S, Fereidoonezhad M, Afshar D and Ramezani Z. A comparative QSAR analysis and molecular docking studies of phenyl piperidine derivatives as potent dual NK1R antagonists/serotonin transporter (SERT) inhibitors. *Comput. Biol. Chem.* (2017) 67: 22-37.
- (46) Fereidoonezhad M, Faghhih Z, Mojaddami A, Tabaei SMH and Rezaei Z. Novel approach synthesis, molecular docking and cytotoxic activity evaluation of N-phenyl-2,2-dichloroacetamide derivatives as anticancer agents. *J. Sci. Islam. Repub. Iran* (2016) 27: 39-49.
- (47) Fereidoonezhad M, Faghhih Z, Mojaddami A, Sakhteman A and Rezaei Z. A comparative docking studies of dichloroacetate analogues on four isozymes of pyruvate dehydrogenase kinase in humans. *Indian J. Pharm. Edu. Res.* (2016) 50: S32-S8.
- (48) DeLano WL. Pymol: An open-source molecular

- graphics tool. *CCP4 Newsletter On Protein Crystallography* (2002) 40: 82-92.
- (49) Brandys MC, Jennings MC and Puddephatt RJ. Luminescent gold(I) macrocycles with diphosphine and 4,4'-bipyridyl ligands. *J. Chem. Soc., Dalton Trans.* (2000) 2000: 4601-6.
- (50) Hojoh K, Ohmiya H and Sawamura M. Synthesis of α -Quaternary formimides and aldehydes through umpolung asymmetric copper catalysis with isocyanides. *J. Am. Chem. Soc.* (2017) 139: 2184-7.
- (51) Zakrzewski J, Huras B and Kielczewska A. Synthesis of isoselenocyanates. *Synthesis* (2016) 48: 85-96.
- (52) Fereidoonzehad M, Shahsavari HR, Lotfi E, Babaghasabha M, Fakhri M, Faghih Z, Faghih Z and Hassan Beyzavi M. (Benzyl isocyanide) gold (I) pyrimidine-2-thiolate complex: Synthesis and biological activity. *Appl. Organomet. Chem.* (2018) 32: 1-7.
- (53) Zou T, Lum CT, Lok CN, Zhang JJ and Che CM. Chemical biology of anticancer gold(III) and gold(I) complexes. *Chem. Soc. Rev.* (2015) 44: 8786-801.

This article is available online at <http://www.ijpr.ir>
