






Post-translational amino acid conversion in photosystem II as a possible origin of photosynthetic oxygen evolution

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Photosynthetic oxygen evolution is performed at the Mn cluster in photosystem II (PSII). The advent of this reaction on ancient Earth changed its environment by generating an oxygenic atmosphere. However, how oxygen evolution originated during the PSII evolution remains unknown. Here, we characterize the site-directed mutants at the carboxylate ligands to the Mn cluster in cyanobacterial PSII. A His residue replaced for D1-D170 is found to be post-translationally converted to the original Asp to recover oxygen evolution. Gln/Asn residues in the mutants at D1-E189/D1-D342 are also converted to Glu/Asp, suggesting that amino-acid conversion is a common phenomenon at the ligand sites of the Mn cluster. We hypothesize that post-translational generation of carboxylate ligands in ancestral PSII could have led to the formation of a primitive form of the Mn cluster capable of partial water oxidation, which could have played a crucial role in the evolutionary process of photosynthetic oxygen evolution.

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In oxygenic photosynthesis, oxygen evolution by water oxidation is carried out in the oxygen-evolving complex (OEC)^{1,2}, which has an inorganic catalytic core of Mn₄CaO₅ (Mn cluster), formed in photosystem II (PSII) through a photoassembly process called photoactivation^{3–5} (Fig. 1). The water oxidation not only provides electrons necessary for CO₂ fixation, but also plays a significant role in sustenance of the environment and life on Earth as a source of oxygen in the atmosphere. In the OEC, two water molecules are oxidized to one oxygen molecule and four protons through a cycle of five intermediates (S₀–S₄ states)^{1,2}, which are advanced by light-induced electron transfer. All extant oxyphototrophs have essentially the identical OEC structure with conserved amino acid residues including ligands to the Mn and Ca ions, i.e., six carboxylate and one histidine ligands from D170, E189, H332, E333, D342, and A344(C-terminus) of the D1 protein and E354 of the CP43 protein (Fig. 1a)⁶.

The advent of photosynthetic oxygen evolution on ancient Earth was a key event in the co-evolution of Earth and life. It produced an oxidative atmosphere and promoted the evolution of aerobic life followed by its extensive diversification^{7,8}. It is unclear when photosynthetic oxygen evolution originated on Earth; the estimates by geochemical and phylogenetic studies span widely from 3.8 to 2.3 billion years ago^{9–16}. Recent evolutionary studies of the PSII proteins by Cardona and coworkers suggested that a functional OEC, nearly identical to that found in extant PSII, existed at a homodimeric photosystem stage before the duplications that led to the evolution of CP43/D1 and CP47/D2^{14,15}. They suggested that this duplication could have occurred at a very early stage during the evolution of life in the early Archean. However, how the ligand environment of the Mn cluster was developed and the water oxidation reaction was optimized in the ancestral PSII remain unknown^{15,17–19}.

We recently characterized a mutant of the cyanobacterium *Synechocystis* sp. PCC 6803, in which one of the ligands to the Mn

cluster, D1-D170, was replaced with His (D1-D170H), using liquid chromatography-mass spectrometry (LC-MS) and Fourier transform infrared (FTIR) spectroscopy²⁰. We found that the His residue at position 170 was converted to the original Asp residue to form normal OEC and restore oxygen evolution during phototrophic growth. Although it was confirmed that this amino-acid conversion occurred at least after transcription²⁰, its exact mechanism has yet to be clarified.

Here, we investigate the mechanism of the amino-acid conversion in mutants at the ligands to the Mn cluster using LC-MS and FTIR analyses. We analyze the PSII from D1-D170H cells incorporated with isotope-labeled histidine to clarify whether the conversion occurs at the protein level after translation. Furthermore, we examine the conversion in mutants at other carboxylate ligands, D1-E189 and D1-D342. The obtained results provide insights into the evolutionary process of photosynthetic water oxidation and the origin of oxygen in the ancient atmosphere.

Results

Post-translational conversion of His to Asp. The mechanism of the His→Asp conversion in the D1-D170H mutant was first examined at the RNA level. The sequence of *psbA2* mRNA obtained from mixotrophically grown D1-D170H cells was determined from its cDNA. The codon for the amino acid at position 170 was CAT for His, and no trace of a codon for Asp (GAT/GAC) was detected at this position (Supplementary Fig. 1). This indicates that no modification occurred in the mRNA of the *psbA2* gene after transcription in D1-D170H cells.

Modification at the protein level was next examined using isotope-labeled histidine. D1-D170H cells were incorporated with ¹³C₆-labeled histidine and isotope labeling of the Asp residue after the conversion was examined. Flash-induced FTIR difference spectra upon the S₁→S₂ transition (S₂/S₁ difference) measured with the PSII complexes from D1-D170H cells incorporated with unlabeled ([¹²C-His]D170H) and ¹³C₆-labeled ([¹³C-His]D170H) histidine were compared (Fig. 2a). Clear ¹³C-induced changes were observed in the prominent bands in 1600–1500 and 1450–1300 cm⁻¹, typical regions of the asymmetric and symmetric COO⁻ stretching vibrations, respectively, of the carboxylate groups around the Mn cluster²¹. The ¹²C-minus-¹³C double difference spectrum more clearly revealed the changes showing bands at 1586/1566/1551/1529 cm⁻¹ and 1399/1376/1362/1320 cm⁻¹, suggesting the ¹³C-induced downshifts of differential signals in the S₂/S₁ difference spectrum. The spectral feature in the symmetric COO⁻ region was well reproduced by the quantum mechanics/molecular mechanics (QM/MM) calculations (Fig. 2b), in which the carbon atoms of D1-D170 were selectively labeled with ¹³C. The calculated normal modes (Fig. 2b, sticks) revealed that the experimental bands at 1399(S₁)/1362(S₂) cm⁻¹ and 1376(S₁)/1320(S₂) cm⁻¹ are attributed to the vibrations of unlabeled and ¹³C-labeled D170, respectively.

The ¹³C labeling of D1-D170 converted from His was further examined by LC-MS analysis (Fig. 2c, d). In both [¹²C-His]D170H and [¹³C-His]D170H cells, the H170 residue in 58–59% and 3–4% of the D1 proteins was converted to Asp and Asn, respectively (Fig. 2d). In the D1 protein from [¹³C-His]D170H cells, the MS spectral peaks of ¹³C₄-labeled D170 (Fig. 2c) and N170 (Supplementary Fig. 2b) in addition to the peak of ¹³C₆-labeled H170 (Supplementary Fig. 2a) were detected. The MS chromatogram bands (Fig. 2c and Supplementary Fig. 2) showed that 66% of the His, Asp, and Asn residues at D1-170, in total, was labeled with ¹³C (Fig. 2d), which is similar to ~61% ¹³C labeling at D1-H332 as a histidine ligand to the Mn cluster and at D1-H190 interacting with Y_Z, indicating that 60–70% of His residues were labeled with ¹³C in [¹³C-His]D170H cells. It is

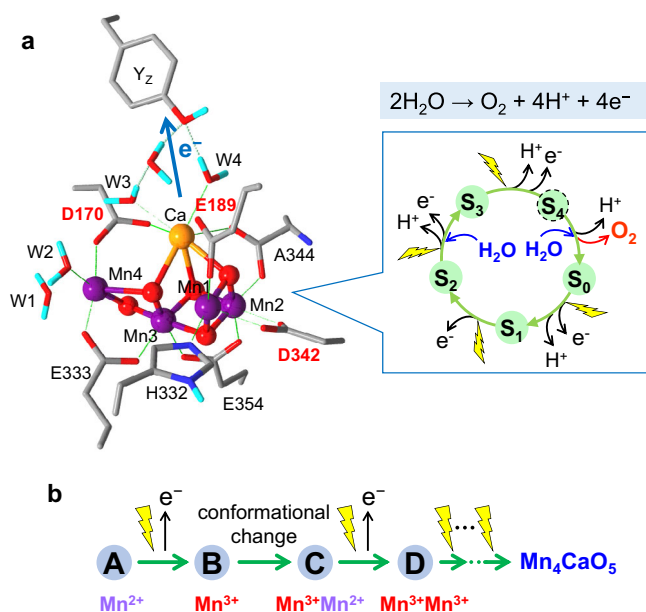


Fig. 1 Structure and reaction of the oxygen-evolving Mn cluster and its photoassembly process. **a** Structure of the Mn cluster and its ligands,

which was obtained by QM/MM calculation based on the X-ray crystallographic structure⁶. Mn, purple; Ca, orange; O, red; N, blue; and H, cyan. The amino acid residues are on the D1 protein except for E354 on the CP43 protein. D1-D170, E189, and D342 are the sites of mutations in this study. The inset shows the S-state cycle of water oxidation. **b** Early process of the photoassembly of the Mn cluster.

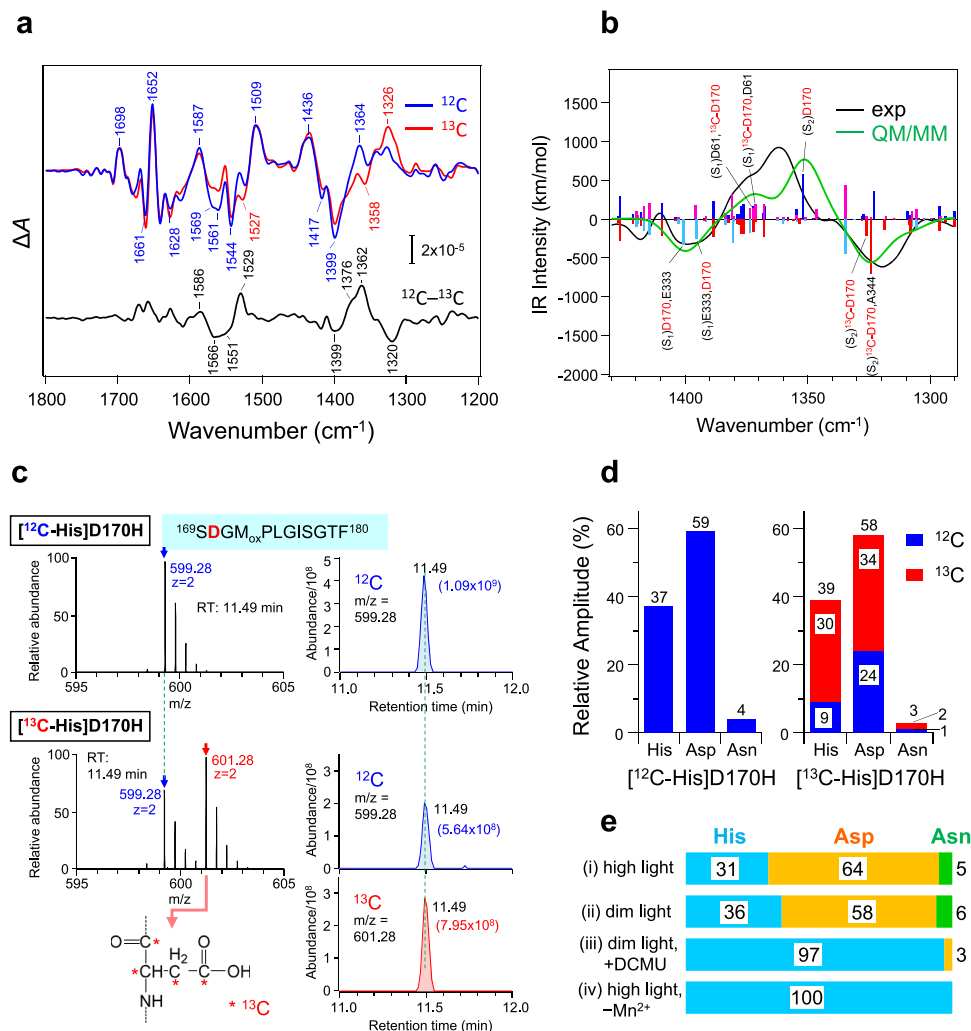


Fig. 2 FTIR and LC-MS analyses of the D1-D170H mutant. **a** Flash-induced S_2/S_1 FTIR difference spectra of the PSII complexes from D1-D170H cells incorporated with unlabeled (^{12}C -His]D170H: blue line) and $^{13}\text{C}_6$ -labeled histidine (^{13}C -His]D170H: red line), together with a ^{12}C -minus- ^{13}C double difference spectrum (black line). **b** The ^{12}C -minus- ^{13}C infrared difference spectrum simulated by QM/MM calculations (green line) in comparison with the experimental spectrum (black line). The normal modes of the unlabeled (S_1 : cyan sticks; S_2 : blue sticks) and ^{13}C -D170-labeled (S_1 : magenta sticks; S_2 : red sticks) OEC are superimposed. **c** MS spectra (left) and MS chromatograms (right) of the polypeptide fragments of the D1 proteins (between S169 and F180 with oxidized M172) in ^{12}C -His]D170H and ^{13}C -His]D170H cells. The data of the D1 fragment that has Asp at position 170 are shown. A peak of the target polypeptide fragment in the MS spectrum is indicated by a blue (^{12}C -D170) or red (^{13}C -D170) arrow. The area intensity of the chromatogram peak is given in parentheses. **d** LC-MS estimation of the relative amounts of the D1 fragments having His, Asp, and Asn residues at position 170 with ^{12}C (blue) and ^{13}C (red) atoms in ^{12}C -His]D170H (left) and ^{13}C -His]D170H (right) cells. **e** LC-MS estimation of the relative amounts of the D1 fragments with His, Asp, and Asn residues at position 170 in D1-D170H cells grown with glucose under (i) high light, (ii) dim light, (iii) dim light in the presence of DCMU, and (iv) high light in the absence of Mn^{2+} . Data of (i) and (iii) are taken from Kitajima-Ihara et al.²⁰. Source data are provided as a Source Data file.

noted that no appreciable conversion of the D1-H332 ligand to Asp was detected.

Thus, from the FTIR, QM/MM, and LS-MS results together with the RNA analysis, it is definitely concluded that D1-H170 in the D1-D170H mutant was post-translationally converted to the original Asp residue (and to Asn to a minor extent) at the protein level.

Involvement of light-induced oxidation of Mn. We previously showed that phototrophic growth is necessary for this His→Asp conversion²⁰. Although under heterotrophic condition with herbicide DCMU and dim light, the His→Asp conversion hardly occurred (~3%; Fig. 2e, iii)²⁰, the same dim-light condition in the absence of DCMU converted ~58% of H170 to Asp (Fig. 2e, ii and Supplementary Fig. 3), indicating the necessity of electron transfer in PSII for the conversion.

The involvement of a Mn ion(s) in the amino-acid conversion was (were) further examined. It was previously reported that *Synechocystis* sp. PCC 6803 cells can grow in Mn^{2+} -depleted medium keeping the PSII level²². We grew D1-D170H mutant cells under mixotrophic condition in Mn^{2+} -depleted medium. The PSII complexes prepared from the thus obtained cells showed no O_2 evolution and LC-MS analysis detected only His at D1-170 without any trace of Asp (Fig. 2e, iv and Supplementary Fig. 4). These observations thus indicate that a Mn^{2+} ion(s) photo-oxidized in the OEC site is (are) directly involved in the mechanism of the His→Asp conversion.

Conversion at other ligand sites and amino acid dependence. We further examined the amino acid conversion at other ligand sites of the Mn cluster and dependence on the amino acid species. D1-E189 and D1-D342 were replaced with corresponding amide

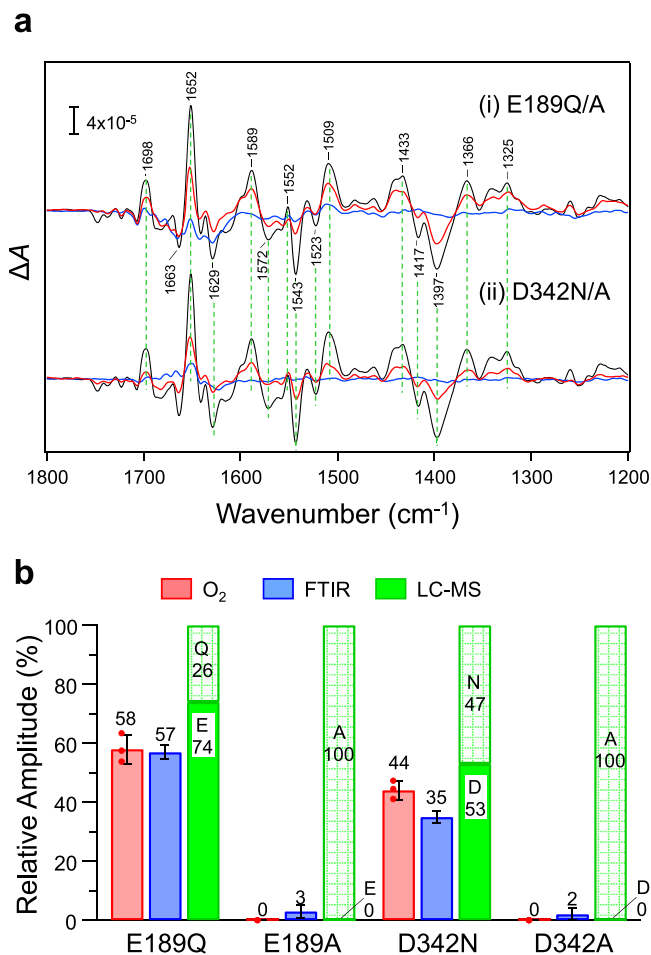


Fig. 3 FTIR and LC-MS analyses of the D1-E189Q/A and D1-D342N/A mutants. **a** S_2/S_1 FTIR difference spectra of the PSII complexes from (i) D1-E189Q (red line) and E189A (blue line) and (ii) D1-D342N (red line) and D342A (blue line) cells in comparison with the spectrum of WT* PSII (black lines). The amplitudes of the spectra were normalized based on the protein amounts estimated from the amide II band. **b** Comparison of the O₂ evolution activity (red bars) and the FTIR amplitude (ΔA between 1433 and 1397 cm^{-1} ; blue bars) relative to those of WT* and the relative contents of amino acid residues at the mutation sites estimated by LC-MS analysis (green bars) in the PSII complexes from D1-E189Q, E189A, D342N, and D342A cells. The O₂ evolution activities were presented as mean values \pm standard deviation ($n = 3$), and the error of the FTIR amplitude was estimated from the root-mean-square noise level in the 1400–1390 cm^{-1} region of the dark-minus-dark spectrum. Source data are provided as a Source Data file.

residues (E189Q and D342N) and a smaller Ala residue (E189A and D342A). The PSII complexes from E189Q and D342N cells grown mixotrophically showed O₂ evolution activities of 58 ± 3 and $44 \pm 5\%$ of the WT* PSII, whereas those from E189A and D342A cells showed no O₂ activity (Fig. 3b). The S_2/S_1 FTIR difference spectra of the E189Q and D342N PSII (Fig. 3a) showed features very similar to those of the WT* PSII, which is consistent with the previous results^{23,24}, but with amplitudes of 57 and 35%, respectively, on the protein basis (Fig. 3b). In contrast, the FTIR spectra of the E189A and D342A PSII showed almost no intensities (Fig. 3a, b).

The LC-MS analysis of the D1 protein from these mutants (Fig. 3b, Supplementary Figs. 5 and 6) further showed that in D1-E189Q and D1-D342N cells, 74 and 53% of the D1 proteins, respectively, underwent conversion from Q189/N342 to the

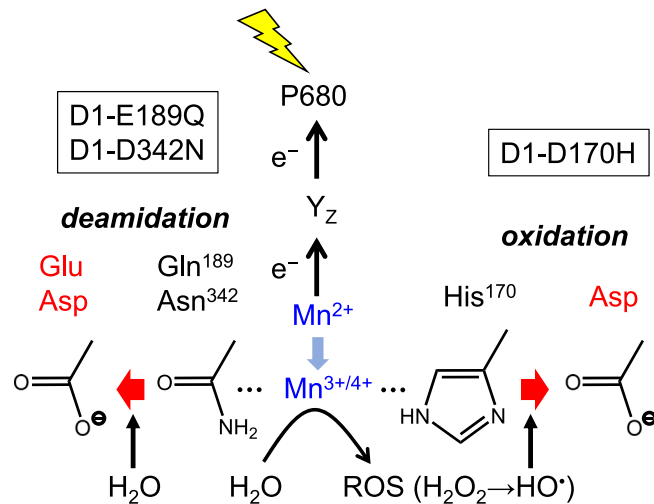


Fig. 4 Possible mechanisms of post-translational amino acid conversion to form carboxylate ligands. His in the D1-D170H mutant is oxidized to Asp by reactive oxygen species (ROS) such as a hydroxyl radical (HO^\bullet) produced from H_2O_2 , which is formed by partial oxidation of water at the $\text{Mn}^{3+}/\text{Mn}^{4+}$ site. Gln and Asn in the D1-E189Q and D1-D342N mutants are converted to Glu and Asp, respectively, by deamidation.

original Glu/Asp residues, whereas such conversion was detected neither in the E189A nor D342A mutants (Fig. 3b). The slightly lower O₂ evolution and FTIR-estimated $S_1 \rightarrow S_2$ activities than the extents of conversion estimated by the LC-MS analysis in E189Q and D342N cells (Fig. 3b) may be ascribed to the presence of some inactive centers in the isolated PSII core complexes with converted residues.

These results indicate that the amino-acid conversion to the original carboxylate residues is a common phenomenon at the ligand sites of the Mn cluster, although the capability of conversion depends on the amino acid species. His/Asn and Gln side chains have enough numbers of carbon atoms for conversion into Asp and Glu, respectively, whereas Ala cannot be changed to an Asp/Glu residue without insertion of a carbon atom(s).

Discussion

In this study, we found that some amino acid residues at the ligand sites of the Mn cluster can be post-translationally converted to the original carboxylate residues. An oxidized Mn ion(s) produced by light-induced electron transfer is (are) most likely involved in the mechanism of this amino-acid conversion. In the photoactivation process, the first Mn^{2+} ion, probably bound to D1-D170 and D1-E189^{5,25}, is oxidized to Mn^{3+} , and then after the conformational change of the D1 C-terminal region, involving D1-E333 and D1-D342⁵, another Mn^{2+} is oxidized to Mn^{3+} to form a relatively stable binuclear complex^{3,4}. Further illumination oxidizes Mn ions to $\text{Mn}^{3+/4+}$ to finally construct the Mn cluster with a Mn_4CaO_5 form (Fig. 1b).

Some cases of post-translational amino-acid conversion have been reported in vivo and in vitro^{26–29}. One typical case is metal-catalyzed oxidation of a His residue by a hydroxyl radical induced by H_2O_2 with an aid of a metal ion, typically Fe^{2+} or Cu^+ , through Fenton-like reactions^{26,27}. His is first converted to 2-oxohistidine and then to Asp or Asn. Mn^{2+} is also effective in producing a hydroxyl radical from H_2O_2 ³⁰, and some Mn oxide complexes supplemented with H_2O_2 have been reported to oxidize various organic compounds^{31,32}. Indeed, H_2O_2 has been detected as a byproduct in water oxidation reactions of some modified OEC³³. It is thus speculated that reactive oxygen species

(ROS) such as H_2O_2 and a hydroxyl radical produced from water during photoactivation oxidizes a His residue at the ligand site to Asp or Asn (Fig. 4). Another case is deamidation of Asn and Gln residues to change into Asp and Glu, respectively^{28,29}. Deamidation is carried out by hydrolysis of amide groups to carboxylate groups, and in proteins the rate of deamidation strongly depends on the primary sequence and the 3D structure around the reacting residue²⁸. A relationship was also suggested between metal binding and deamidation in amyloidogenic proteins²⁹. It is thus possible that Mn binding at amide residues promoted deamidation to generate Glu/Asp ligands (Fig. 4).

The amino acid residues that are converted to carboxylate residues may not be limited to His and amide residues. It has been reported that mutations of D1-D170 to Tyr, Trp, Asn, His, Met, Glu, Arg, Val, Leu, and Ile, all of which have long side chains at least with a γ -carbon, retained partial O_2 evolution activity^{34–36}, suggesting the presence of the intact Mn cluster in some PSII complexes in the mutants (note that Glu replaced for D1-D170 may form a partially functional Mn cluster³⁷). In contrast, mutations of D1-D170 to shorter amino acid residues such as Ala and Ser did not show O_2 evolution^{34,35}. In addition, mutations of D1-E189 to Gln, Arg, Lys, Leu, and Ile, which have an aliphatic δ -carbon, largely restored the O_2 evolution, whereas mutations to residues having shorter aliphatic chains showed no or only a trivial amount of O_2 evolution^{36,38}. Furthermore, mutations of D1-D342 to Asn, His, and Glu having a γ -carbon showed some O_2 evolution^{35,39}, whereas the Ala mutant did not support O_2 evolution³⁹ in agreement with our result (Fig. 3). These observations can be explained by post-translational conversions from residues with side chains long enough to form Asp/Glu and the absence of conversion from those with shorter side chains. In particular, the presence of O_2 activity upon mutations of D170 or E189 to aliphatic residues such as Val, Leu, and Ile^{36,38}, which cannot ligate the Mn cluster, is understandable if these side chains are converted to the Asp/Glu side chains. These amino-acid conversions except for deamidation of Asn/Gln may also be promoted by ROS formed from water through $\text{Mn}^{3+/4+}$ ions, although the exact conversion mechanism for each amino acid residue requires further investigations.

In the D170H, E189Q, and D342N mutants, relatively large amounts (>50%) of the D1 proteins with converted original amino acid ligands were accumulated in mixotrophically grown cells (Figs. 2 and 3). In the steady state of the turnover of the D1 protein in the cell culture of the mutants, where the total amount of the D1 protein is constant and the relative amounts of the converted functional and initial unconverted D1 proteins ($[\text{cD1}]$ and $[\text{iD1}]$, respectively) are unchanged, their ratio ($[\text{cD1}]/[\text{iD1}]$) is determined by the rate constant of conversion (k_c) relative to that of degradation of the converted D1 protein (k_2) ($[\text{cD1}]/[\text{iD1}] = k_c/k_2$; Supplementary Fig. 7). The turnover of the D1 protein in the *Synechocystis* sp. PCC 6803 mutant with *psbA2* as a single gene for the D1 protein (corresponding to our WT*) was previously shown to occur in several hours under the light of $100 \mu\text{mol photons m}^{-2}\text{s}^{-1}$ ⁴⁰. Thus, the presence of comparable amounts of the converted and unconverted D1 proteins indicates that amino-acid conversion in the D170H, E189Q, and D342N mutants through ROS-mediated oxidation or deamidation has time constants of hours under the light of $\sim 50 \mu\text{mol photons m}^{-2}\text{s}^{-1}$ used in our experiments. In the oxidation by ROS, a variety of oxidized products, such as 2-oxo-histidine in the His \rightarrow Asp conversion, would be formed as impaired D1 proteins. However, because non-functional D1 proteins degrade much faster than the functional D1 protein (i.e., large k_2)^{40,41}, they are hardly accumulated in cells. For the same reason, the D1 protein modified by the His \rightarrow Asp conversion at D1-H332, an original His ligand to the Mn cluster, will not be accumulated in cells, because of the formation of a non-functional D1 protein³⁹.

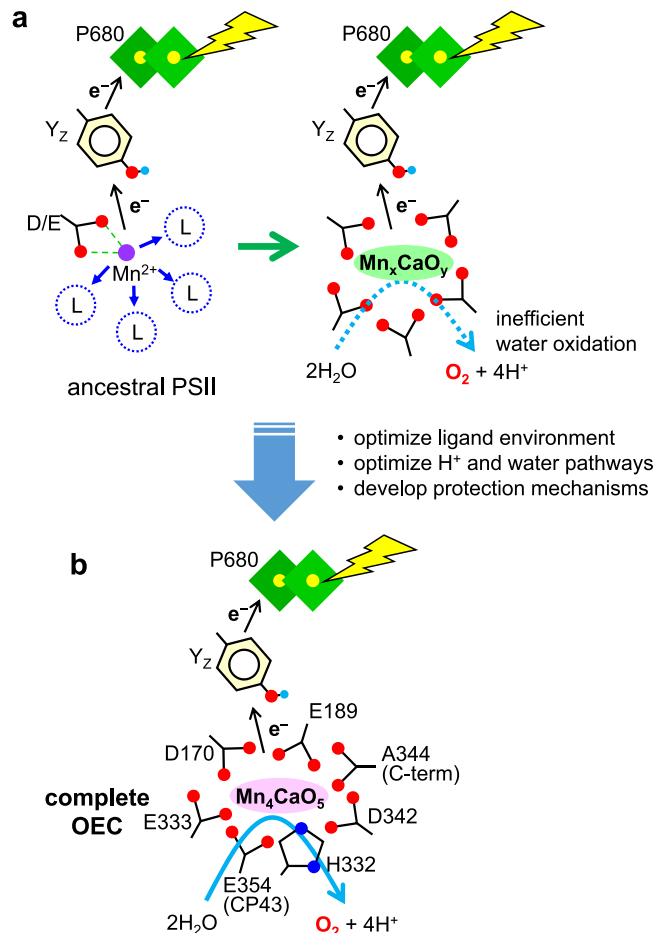


Fig. 5 Hypothetical mechanism of generation of a primitive OEC in ancestral PSII.

a The ancestral PSII had a D/E residue(s) for binding of an initial Mn^{2+} ion, which can be oxidized by light-induced electron transfer. Carboxylate ligands were generated by post-translational amino-acid conversion and some sort of Mn-oxide complex was formed in the OEC site to perform partial water oxidation. During this inefficient O_2 evolution, the ligand environment and proton and water pathways were optimized, and protection mechanisms against ROS were developed in PSII. “L” indicates non-carboxylic amino acid residues at the ligand positions. **b** Completed OEC capable of efficient water oxidation.

Post-translational amino-acid conversions so far reported inactivate proteins in relevance to aging and diseases^{27–29}. The amino-acid conversion found in this study, which generates carboxylate ligands in OEC, is a very unique case that restores an enzymatic function, photosynthetic oxygen evolution, essential for life and its evolution.

One of the biggest questions in the evolution of photosynthesis is how the OEC originated in the ancestral PSII^{15,17–19}. Our data suggest that water oxidation could have originated before a fully complete ligand sphere had time to evolve in the early photosystem. We speculate that in the ancestral PSII, which has at least a single Mn binding site by an Asp/Glu residue(s), a $\text{Mn}^{3+/4+}$ ion formed by light-induced electron transfer in the reaction center promoted post-translational conversion of nearby amino-acid residues to generate Asp/Glu ligands (Fig. 5). Such an initial metal binding site on the electron-donor side could have existed even in reaction centers at a very early stage of photosynthesis evolution. Indeed, evidence for such an ancient site may be found in extant homodimeric type I reaction centers. In the *Heliobacterium modesticaldum* type I reaction center, an exposed Ca-binding site was recently discovered with structural similarities to the OEC in

PSII, including two carboxylate ligands to the Ca atom, an Asp and the C-terminal carboxylic group^{42,43}. The presence of several carboxylate ligands with negative charges is significant for the formation of a functional Mn cluster not only by fixing the Mn/Ca ions but also by tuning its redox potential. Recently, Chernev et al.¹⁹ showed light-induced formation of birnessite-type Mn(III,IV)-oxide nanoparticles in apo-PSII, and proposed that such a Mn-oxide particle was down-sized to develop the Mn cluster. Thus different types of high-valent Mn-oxide complexes in various sizes could be formed in the OEC site by light-induced electron transfer during the evolution of PSII. It has also been reported that many high-valent Mn-oxide complexes function as catalysts of water oxidation⁴⁴. It is thus presumed that some sort of photo-produced Mn-oxide complex was stabilized and functionalized by post-translationally generated carboxylate ligands in the ancestral PSII, and carried out partial oxygen evolution.

This inefficient water oxidation, however, would have been crucial for further development of PSII; in addition to heterodimerization and optimization of the ligand environment, the proton-exit and water-delivery pathways would have been formed around the Mn cluster to optimize the water oxidation reaction, while various protection mechanisms and repair systems would have been developed^{14,45}. It would also have stimulated the evolution of aerobic respiration and oxygen-related enzymes^{46–48}. Thus, the post-translational amino-acid conversion, beyond the central dogma, to generate the carboxylate ligands to the Mn cluster could have played a significant role in the evolution of photosynthetic water oxidation and that of aerobic life.

Methods

Construction of site-directed mutants. Site-directed mutagenesis of the *psbA2* gene encoding the D1 subunit was performed in *Synechocystis* sp. PCC 6803^{49,50}. Plasmid pRN123, which involved the coding region of *psbA2*, was used as a parental vector for site-directed mutagenesis. The host *Synechocystis* strain, which lacks all of the three *psbA* genes ($\Delta psbA1/\Delta psbA2/\Delta psbA3$) and contains a hexahistidine tag attached to the C-terminus of the CP47 protein, was transformed with pRN123 to provide a control strain with the wild-type D1 protein (WT*). Mutation of D1-D170 to His²⁰, D1-E189 to Gln/Ala, and D1-D342 to Asn/Ala were introduced into pRN123 by replacing the GAT (D170), GAG(E189), and GAC(D342) codons with CAT (His), CAG(Gln)/GCG(Ala), and AAC(Asn)/GCC(Ala) codons, respectively, and the obtained plasmids were introduced into the host $\Delta psbA1/\Delta psbA2/\Delta psbA3$ strain. These strains were maintained on BG-11⁵¹ agar plates containing antibiotics (kanamycin, chloramphenicol, erythromycin, and spectinomycin, 5 $\mu\text{g mL}^{-1}$ for each), in the presence of 5 mM glucose and 10 μM 3-(3,4)-dichlorophenyl-1,1-dimethylurea (DCMU) under a continuous low-light condition.

Cell culture and sample preparations. WT* and mutant cells were grown mixotrophically in 40 mL of BG-11 medium, which was supplemented with 5 mM glucose and the above mentioned antibiotics (5 $\mu\text{g mL}^{-1}$ for each) and was bubbled with air containing 3% (v/v) CO₂, at 30 °C under continuous illumination (~20 $\mu\text{mol photons m}^{-2}\text{s}^{-1}$). For preparation of PSII core complexes, cells were grown mixotrophically in an 8 L culture bottle without antibiotics under continuous illumination (~50 $\mu\text{mol photons m}^{-2}\text{s}^{-1}$). For a low-light condition, dim light with an intensity of ~5 $\mu\text{mol photons m}^{-2}\text{s}^{-1}$ was used. Cells cultured in three bottles (total volume of 24 L) were used for preparation of PSII core complexes from each strain. An aliquot of the cell culture at this stage was saved and used for DNA sequencing²⁰, which showed the proper genotypes of individual mutants and confirmed no trace of the wild-type *psbA2* gene.

For cell growth in a Mn²⁺-depleted medium, the BG-11 medium in the absence of MnCl₂·4H₂O was prepared with high-grade reagents. In addition to removing MnCl₂, we replaced ammonium ferric citrate, which was found to contain high Mn²⁺ contamination, in the original BG-11 medium with FeSO₄·7H₂O (special grade) and citric acid.

PSII core complexes were purified using Ni²⁺ affinity column chromatography⁴⁹. Thylakoid membranes suspended in a buffer (pH 6.0) containing 50 mM Mes-NaOH, 5 mM CaCl₂, 10 mM MgCl₂, and 25% (w/v) glycerol (buffer A) were solubilized with 0.8% (w/v) *n*-dodecyl β -D-maltoside (DM) at a Chl concentration of 1.0 mg mL⁻¹ by stirring for 20 min on ice. After centrifugation at 43,000 *g* for 10 min, the resultant supernatant was applied to a Ni²⁺ affinity column, which was then washed with buffer A containing 0.04% DM and 1 mM L-histidine. PSII complexes were eluted with buffer A containing 0.04%

DM and 50 mM L-histidine and then concentrated by ultrafiltration (AmiconUltra-15, NMWL 100,000).

The O₂ evolution activities of PSII complexes were measured using a Clark-type oxygen electrode at 30 °C under saturating light in a buffer containing 50 mM Mes-NaOH (pH 6.0), 1 M sucrose, 10 mM NaCl, 5 mM CaCl₂, 0.04% *n*-dodecyl β -D-maltoside (DM) in the presence of 4 mM potassium ferricyanide and 0.1 mM 2,6-dichloro-1,4-benzoquinone (DCBQ) as exogenous electron acceptors. Values by three measurements on distinct samples were averaged for each mutant.

Incorporation of ¹³C₆-labeled histidine into D1-D170H cells. For incorporation of ¹³C₆-labeled histidine into D1-D170H cells, a histidine-tolerant strain was first isolated⁵² using the D1-D170H mutant constructed previously²⁰. D1-D170H cells were repeatedly cultured on a BG-11 agar plate with the above antibiotics supplemented with L-histidine, the content of which was increased from 30 μM to 240 μM in a stepwise way. This strain was maintained on a BG-11 agar plate in the presence of 120 μM L-histidine. Cells of this strain were grown in liquid BG-11 medium as described above except for the presence of 120 μM unlabeled (¹²C) or ¹³C₆-labeled (Cambridge Isotope Laboratories, Inc., 97–99 at. % ¹³C) L-histidine. Cells cultured in two bottles (total volume of 16 L) were used for preparation of PSII core complexes.

Sequencing of the *psbA2* mRNA from D1-D170H cells. Total RNA was isolated from mixotrophically grown D1-D170H cells using the TRIzol Max Bacterial RNA Isolation Kit (Thermo Fisher Scientific). Genomic DNA was digested by DNase using the ReliaPrep RNA Cell Miniprep System (Promega). The cDNA of the *psbA2* mRNA was obtained by reverse transcription-PCR using ReverTra Plus (Toyobo) with a primer 5'-GTCAAAGCCACAGGAGCTTGCTCCCC-3'. From the obtained cDNA, a DNA fragment (1.0 kbp) in the internal region of *psbA2* was amplified by PCR using a primer set of 5'-AGCGCGAAAGCGCTTCCTGTGG G-3' and 5'-GTCAAAGCCACAGGAGCTTGCTCCCC-3'. The DNA sequence including the amino acid residue at position 170 was determined with a primer 5'-CGGATTTTCTGCTACATGGG-3' using an ABI 3100 DNA sequencer.

Measurement of flash-induced FTIR difference spectra. For FTIR measurements, PSII suspension (~2 mg of Chl mL⁻¹, 10 μL) in a buffer (10 mM Mes-NaOH, 5 mM NaCl, 5 mM CaCl₂, 40 mM sucrose, and 0.06% DM, pH 6.0) was added with 1 μL of 100 mM potassium ferricyanide and was dried on a CaF₂ plate under N₂ gas. For the PSII samples of D1-E189Q/A and D1-D342N/A as well as their control WT* sample, 1 μL of 20 mM NaHCO₃ was further added to the suspension to prevent pre-oxidation of the non-heme iron, whose signals often contaminate the spectra. The dried film was sealed in a cell formed by another CaF₂ plate with a greased Teflon spacer (0.5 mm in thickness), enclosing 2 μL of 40% (v/v) glycerol solution without touching the sample to moderately hydrate the film⁵³. The sample temperature was kept at 10 °C by circulating cold water through a copper holder.

FTIR difference spectra were measured using a Bruker Vertex 80 spectrophotometer (with the OPUS 7.8 software) with an MCT detector at 4 cm⁻¹ resolution²⁰. A Ge filter (Andover, 4.50ILP-25) was placed in the infrared path to cut light at >2200 cm⁻¹. FTIR spectra with 20 scans were recorded twice before and once after single-flash illumination from a Nd:YAG laser (Quanta-Ray INDI-40-10; 532 nm, ~7 ns fwhm, ~7 mJ pulse⁻¹ cm⁻²). The measurement was repeated 100 times for D1-D170H and 160 times for D1-E189Q/A, D1-D342N/A, and WT* with a dark interval of 2 min, and spectra were averaged. A difference spectrum of after-minus-before illumination provided the changes upon the S₁ → S₂ transition, while that of the two spectra before illumination represented the noise level.

Mass spectrometry analysis. Amino acid sequences of the D1 proteins were analyzed using mass spectrometry²⁰. PSII core complexes solubilized with 3% lithium lauryl sulfate in the presence of 75 mM dithiothreitol were subjected to SDS-PAGE. The isolated D1 protein was applied to nano-liquid chromatography-tandem mass spectrometry (nLC-MS/MS) after digestion with chymotrypsin. The digestion mixture was separated using a nano-electrospray column (NTCC analytical column, C18, ϕ 75 $\mu\text{m} \times$ 100 mm, 3 μm ; Nikkoy Technos, Tokyo, Japan) on a nanoflow LC (Easy nLC 1000; Thermo Fisher Scientific, Inc., Waltham, MA, USA) with a linear gradient of 0–60% of 0.1% formic acid/acetonitrile in 0.1% formic acid solution at a flow rate of 300 nL/min over 30 min. The LC was coupled on-line to a Q-Exactive mass spectrometer (Thermo Fisher Scientific, Inc.) equipped with a nanospray ion source. A data-dependent TOP 10 method in positive ion mode was used to operate the mass spectrometer. Obtained MS/MS spectra were searched using the MASCOT 2.7 program (Matrix Science, London, UK) against an in-house database containing the D1 protein and its mutants. FreeStyle 1.3 SP2 and Xcalibur 4.1.50, Qual Browser (Thermo Fisher Scientific, Inc.) were used to draw the MS spectra and MS chromatograms.

Quantum mechanics/molecular mechanics calculations. The infrared spectra of the carboxylate groups in the OEC were simulated using quantum mechanics/molecular mechanics (QM/MM) calculations^{21,54}. The coordinates of heavy atoms in the Mn cluster, surrounding amino acid residues, water molecules, and two Cl⁻

ions within 20 Å from the Mn cluster were extracted from the X-ray structure (PDB ID: 4UB6)⁶ of PSII in the S₁ state. Hydrogen atoms were optimized using the AMBER force field⁵⁵ fixing all heavy atoms. QM/MM calculations were carried out by the ONIOM method⁵⁶ with the electronic embedding scheme in the Gaussian 16 program package⁵⁷. The QM region (Supplementary Fig. 8) consists of the Mn cluster, Cl-1, seven amino acid ligands (D1-D170, D1-E189, D1-H332, D1-E333, D1-D342, D1-A344, CP43-E354) and nearby amino acid residues (Y_Z, D1-H190, D1-D61, D1-H337, CP43-R357, D1-N181, D2-K317), and 17 surrounding water molecules including four water ligands (W1–W4), while other atoms were assigned to the MM region. D1-H337 was assumed to have a protonated cation form⁵⁸ and W2 was fully protonated H₂O⁵⁴. Geometry optimization and normal mode analysis of the QM region were performed using an unrestricted DFT method with the B3LYP functional and the basis sets of LANL2DZ and 6-31 G(d) for metal atoms and other atoms, respectively. In geometry optimization, the coordinates of the QM region were fully relaxed, while those of the MM region were fixed. The oxidation states of the Mn ions were assumed to be III₂IV₂ and III₁IV₃ in high spin states (15et and 14et) in the S₁ and S₂ states, respectively. The S₂ state was assumed to have an open cubane conformation with oxidized Mn₄(IV). In calculation of OEC with ¹³C-labeled D1-D170, two carbon atoms of the D170 side chains in the QM region were substituted with ¹³C. To generate infrared spectra in the symmetric COO⁻ stretching region, a Gaussian band with a 16 cm⁻¹ width (FWHM) was assumed for each normal mode and all bands in this region were co-added^{21,54}. Calculated vibrational frequencies were scaled with a scaling factor of 0.9628 to match the major negative peak of the simulated S₂/S₁ spectrum of unlabeled OEC to the experimental peak at 1399 cm⁻¹.

Reporting summary. Further information on research design is available in the Nature Research Reporting Summary linked to this article.

Data availability

The data supporting the findings in this study are available within the manuscript and the Supplementary Information file. The sequence of cDNA of the *psbA2* mRNA of D1-D170H was deposited in the DNA Data Bank of Japan under accession number LC717798. The coordinates of the PSII complex used in QM/MM calculations were obtained from PDB ID: 4UB6. Source data are provided with this paper.

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References

- Shen, J.-R. The structure of photosystem II and the mechanism of water oxidation in photosynthesis. *Annu. Rev. Plant Biol.* **66**, 23–48 (2015).
- Cox, N., Pantazis, D. A. & Lubitz, W. Current understanding of the mechanism of water oxidation in photosystem II and its relation to XFEL data. *Annu. Rev. Biochem.* **89**, 795–820 (2020).
- Dasgupta, J., Ananyev, G. M. & Dismukes, G. C. Photoassembly of the water-oxidizing complex in photosystem II. *Coord. Chem. Rev.* **252**, 347–360 (2008).
- Bao, H. & Burnap, R. L. Photoactivation: The light-driven assembly of the water oxidation complex of photosystem II. *Front. Plant Sci.* **7**, 578 (2016).
- Sato, A., Nakano, Y., Nakamura, S. & Noguchi, T. Rapid-scan time-resolved ATR-FTIR study on the photoassembly of the water-oxidizing Mn₄CaO₅ cluster in photosystem II. *J. Phys. Chem. B* **125**, 4031–4045 (2021).
- Suga, M. et al. Native structure of photosystem II at 1.95 Å resolution viewed by femtosecond X-ray pulses. *Nature* **517**, 99–103 (2015).
- Holland, H. D. The oxygenation of the atmosphere and oceans. *Philos. Trans. R. Soc. B* **361**, 903–915 (2006).
- Lyons, T. W., Diamond, C. W., Planavsky, N. J., Reinhard, C. T. & Li, C. Oxygenation, life, and the planetary system during Earth's middle history: An overview. *Astrobiology* **21**, 906–923 (2021).
- Crowe, S. A. et al. Atmospheric oxygenation three billion years ago. *Nature* **501**, 535–538 (2013).
- Lyons, T. W., Reinhard, C. T. & Planavsky, N. J. The rise of oxygen in Earth's early ocean and atmosphere. *Nature* **506**, 307–315 (2014).
- Schirmer, B. E., Guggen, M. & Donoghue, P. C. J. Cyanobacteria and the Great Oxidation Event: evidence from genes and fossils. *Palaeontology* **58**, 769–785 (2015).
- Fischer, W. W., Hemp, J. & Johnson, J. E. Evolution of oxygenic photosynthesis. *Annu. Rev. Earth Planet Sci.* **44**, 647–683 (2016).
- Soo, R. M., Hemp, J., Parks, D. H., Fischer, W. W. & Hugenholtz, P. On the origins of oxygenic photosynthesis and aerobic respiration in Cyanobacteria. *Science* **355**, 1436–1439 (2017).
- Cardona, T., Sanchez-Baracaldo, P., Rutherford, A. W. & Larkum, A. W. Early Archean origin of Photosystem II. *Geobiology* **17**, 127–150 (2019).
- Oliver, T., Sanchez-Baracaldo, P., Larkum, A. W., Rutherford, A. W. & Cardona, T. Time-resolved comparative molecular evolution of oxygenic photosynthesis. *Biochim. Biophys. Acta Bioenerg.* **1862**, 148400 (2021).
- Fournier, G. P. et al. The Archean origin of oxygenic photosynthesis and extant cyanobacterial lineages. *Proc. R. Soc. B: Biol. Sci.* **288**, 20210675 (2021).
- Dismukes, G. C. et al. The origin of atmospheric oxygen on Earth: The innovation of oxygenic photosynthesis. *Proc. Natl Acad. Sci. USA* **98**, 2170–2175 (2001).
- Raymond, J. & Blankenship, R. E. The origin of the oxygen-evolving complex. *Coord. Chem. Rev.* **252**, 377–383 (2008).
- Chernev, P. et al. Light-driven formation of manganese oxide by today's photosystem II supports evolutionarily ancient manganese-oxidizing photosynthesis. *Nat. Commun.* **11**, 6110 (2020).
- Kitajima-Ihara, T. et al. Fourier transform infrared and mass spectrometry analyses of a site-directed mutant of D1-Asp170 as a ligand to the water-oxidizing Mn₄CaO₅ cluster in photosystem II. *Biochim. Biophys. Acta Bioenerg.* **1861**, 148086 (2020).
- Nakamura, S. & Noguchi, T. Quantum mechanics/molecular mechanics simulation of the ligand vibrations of the water-oxidizing Mn₄CaO₅ cluster in photosystem II. *Proc. Natl Acad. Sci. USA* **113**, 12727–12732 (2016).
- Salomon, E. & Keren, N. Manganese limitation induces changes in the activity and in the organization of photosynthetic complexes in the cyanobacterium *Synechocystis* sp. strain PCC 6803. *Plant Physiol.* **155**, 571–579 (2011).
- Strickler, M. A., Hillier, W. & Debus, R. J. No evidence from FTIR difference spectroscopy that glutamate-189 of the D1 polypeptide ligates a Mn ion that undergoes oxidation during the S₀ to S₁, S₁ to S₂, or S₂ to S₃ transitions in photosystem II. *Biochemistry* **45**, 8801–8811 (2006).
- Strickler, M. A., Walker, L. M., Hillier, W., Britt, R. D. & Debus, R. J. No evidence from FTIR difference spectroscopy that aspartate-342 of the D1 polypeptide ligates a Mn ion that undergoes oxidation during the S₀ to S₁, S₁ to S₂, or S₂ to S₃ transitions in photosystem II. *Biochemistry* **46**, 3151–3160 (2007).
- Zabret, J. et al. Structural insights into photosystem II assembly. *Nat. Plants* **7**, 524–538 (2021).
- Uchida, K. & Kawakishi, S. Ascorbate-mediated specific oxidation of the imidazole ring in a histidine derivative. *Bioorg. Chem.* **17**, 330–343 (1989).
- Stadtman, E. R. Metal ion-catalyzed oxidation of proteins: Biochemical mechanism and biological consequences. *Free Radic. Biol. Med.* **9**, 315–325 (1990).
- Robinson, N. E. Protein deamidation. *Proc. Natl Acad. Sci. USA* **99**, 5283–5288 (2002).
- Sadakane, Y. & Kawahara, M. Implications of metal binding and asparagine deamidation for amyloid formation. *Int. J. Mol. Sci.* **19**, 2449 (2018).
- Yim, M. B., Berlett, B. S., Chock, P. B. & Stadtman, E. R. Manganese(II)-bicarbonate-mediated catalytic activity for hydrogen peroxide dismutation and amino acid oxidation: Detection of free radical intermediates. *Proc. Natl Acad. Sci. USA* **87**, 394–398 (1990).
- Bokare, A. D. & Choi, W. Review of iron-free Fenton-like systems for activating H₂O₂ in advanced oxidation processes. *J. Hazard. Mater.* **275**, 121–135 (2014).
- Shul'pin, G. B., Suss-Fink, G. & Shul'pina, L. S. Oxidations by the system “hydrogen peroxide-manganese(IV) complex-carboxylic acid” Part 3. Oxygenation of ethane, higher alkanes, alcohols, olefins and sulfides. *J. Mol. Catal. A Chem.* **170**, 17–34 (2001).
- Pospisil, P. Production of reactive oxygen species by photosystem II. *Biochim. Biophys. Acta Bioenerg.* **1787**, 1151–1160 (2009).
- Nixon, P. J. & Diner, B. A. Aspartate 170 of the photosystem II reaction center polypeptide D1 is involved in the assembly of the oxygen-evolving manganese cluster. *Biochemistry* **31**, 942–948 (1992).
- Chu, H. A., Nguyen, A. P. & Debus, R. J. Site-directed photosystem II mutants with perturbed oxygen-evolving properties. 1. Instability or inefficient assembly of the manganese cluster in vivo. *Biochemistry* **33**, 6137–6149 (1994).
- Chu, H. A., Nguyen, A. P. & Debus, R. J. Amino-acid residues that influence the binding of manganese or calcium to photosystem II. 1. The luminal interhelical domains of the D1 polypeptide. *Biochemistry* **34**, 5839–5858 (1995).
- Debus, R. J. Alteration of the O₂-producing Mn₄Ca cluster in photosystem II by the mutation of a metal ligand. *Biochemistry* **60**, 3841–3855 (2021).
- Debus, R. J., Campbell, K. A., Pham, D. P., Hays, A. M. A. & Britt, R. D. Glutamate 189 of the D1 polypeptide modulates the magnetic and redox properties of the manganese cluster and tyrosine Y_Z in photosystem II. *Biochemistry* **39**, 6275–6287 (2000).
- Chu, H. A., Nguyen, A. P. & Debus, R. J. Amino acid residues that influence the binding of manganese or calcium to photosystem II. 2. The carboxy-terminal domain of the D1 polypeptide. *Biochemistry* **34**, 5859–5882 (1995).
- Komenda, J. et al. Degradation of the Photosystem II D1 and D2 proteins in different strains of the cyanobacterium *Synechocystis* PCC 6803 varying with respect to the type and level of *psbA* transcript. *Plant Mol. Biol.* **42**, 635–645 (2000).

41. Vermaas, W. et al. Turnover of the D1 protein and of Photosystem II In a *Synechocystis* 6803 mutant lacking Tyr_Z. *Photosynth. Res.* **45**, 99–104 (1995).
42. Cardona, T. & Rutherford, A. W. Evolution of Photochemical Reaction Centres: More Twists? *Trends Plant Sci.* **24**, 1008–1021 (2019).
43. Gisriel, C. J., Azai, C. & Cardona, T. Recent advances in the structural diversity of reaction centers. *Photosynth. Res.* **149**, 329–343 (2021).
44. Robinson, D. M. et al. Photochemical water oxidation by crystalline polymorphs of manganese oxides: Structural requirements for catalysis. *J. Am. Chem. Soc.* **135**, 3494–3501 (2013).
45. Shao, S., Cardona, T. & Nixon, P. J. Early emergence of the FtsH proteases involved in photosystem II repair. *Photosynthetica* **56**, 163–177 (2018).
46. Brochier-Armanet, C., Talla, E. & Gribaldo, S. The multiple evolutionary histories of dioxygen reductases: Implications for the origin and evolution of aerobic respiration. *Mol. Biol. Evol.* **26**, 285–297 (2009).
47. Jabłońska, J. & Tawfik, D. S. The evolution of oxygen-utilizing enzymes suggests early biosphere oxygenation. *Nat. Ecol. Evol.* **5**, 442–448 (2021).
48. Boden, J. S., Konhauser, K. O., Robbins, L. J. & Sanchez-Baracaldo, P. Timing the evolution of antioxidant enzymes in cyanobacteria. *Nat. Commun.* **12**, 4742 (2021).
49. Nagao, R., Ueoka-Nakanishi, H. & Noguchi, T. D1-Asn-298 in photosystem II is involved in a hydrogen-bond network near the redox-active tyrosine Y_Z for proton exit during water oxidation. *J. Biol. Chem.* **292**, 20046–20057 (2017).
50. Nagao, R., Yamaguchi, M., Nakamura, S., Ueoka-Nakanishi, H. & Noguchi, T. Genetically introduced hydrogen bond interactions reveal an asymmetric charge distribution on the radical cation of the special-pair chlorophyll P680. *J. Biol. Chem.* **292**, 7474–7486 (2017).
51. Stanier, R. Y., Kunisawa, R., Mandel, M. & Cohen-Bazire, G. Purification and properties of unicellular blue-green algae (order Chroococcales). *Bacteriol. Rev.* **35**, 171–205 (1971).
52. Tang, X. S. et al. Identification of histidine at the catalytic site of the photosynthetic oxygen-evolving complex. *Proc. Natl Acad. Sci. USA* **91**, 704–708 (1994).
53. Noguchi, T. & Sugiura, M. Flash-induced FTIR difference spectra of the water oxidizing complex in moderately hydrated photosystem II core films: Effect of hydration extent on S-state transitions. *Biochemistry* **41**, 2322–2330 (2002).
54. Yamamoto, M., Nakamura, S. & Noguchi, T. Protonation structure of the photosynthetic water oxidizing complex in the S₀ state as revealed by normal mode analysis using quantum mechanics/molecular mechanics calculations. *Phys. Chem. Chem. Phys.* **22**, 24213–24225 (2020).
55. Case, D. A. et al. *AMBER 2016*, University of California, San Francisco. (2016).
56. Vreven, T. et al. Combining quantum mechanics methods with molecular mechanics methods in ONIOM. *J. Chem. Theory Comput.* **2**, 815–826 (2006).
57. Frisch, M. J. et al. *Gaussian 16*, Revision C. 01, Gaussian Inc.: Wallingford, CT (2016).
58. Nakamura, S. & Noguchi, T. Infrared determination of the protonation state of a key histidine residue in the photosynthetic water oxidizing center. *J. Am. Chem. Soc.* **139**, 9364–9375 (2017).

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Author contributions

T.N. designated the study. R.N. constructed the site-directed mutants and Y.S. isolated a histidine-tolerant strain. Y.S., T.M., and T.K. prepared samples and performed O₂ evolution and FTIR analyses. T.S. and N.D. performed LC-MS analysis. T.N. performed QM/MM calculations. T.N. wrote the first draft, and Y.S., T.S., T.M., T.K., R.N., N.D., and T.N. completed the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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