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Curve-ftting techniques improve OPENthe mid-infrared analysis of soil organic carbon: a case study for Brookston clay loam particle-size fractions

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Few studies have evaluated structural features of soil organic carbon (SOC) in diferent soil particle fractions, especially SOC changes induced by tillage, using Fourier transformed mid-infrared (MIR) spectroscopy. To make a contribution in this context, soil samples of a Brookston clay loam (mesic Typic Argiaquoll) with averaged pH and organic matter concentration at 7.28 and 43.9g kg[−]1, respectively, were collected from short-term no-tillage (NT97) and mouldboard plow (CT97) treatments initiated in 1997 and long-term no-tillage (NT83) and mouldboard plow (CT83) treatments initiated in 1983 under a corn-soybean rotation, and were separated into sand, silt, and clay fractions using sonication. Structural features of SOC in these soil fractions were investigated using curve-ftting analysis of midinfrared (MIR) spectra. Aromatic C content was found to be greater in clay- than in sand-sized fractions, while aliphatic C content was higher in sand- than in silt- and clay-sized particles. With decrease in tillage intensity, the aliphatic C gradually increased in sand- and clay-sized fractions but not in the silt-sized fraction. The aliphatic C content in sand fraction was signifcantly greater in NT83 than CT83 (*P***<0.05). The aromatic C in silt- and clay-sized fractions was greater in NT83 than in both CT soils, whereas aromatic C contents were higher in both CT soils than in NT83 soil. Signifcantly higher aromatic/aliphatic C ratio in CT83 than NT83 was found in sand-sized fractions, while the opposite trend was found in the silt-sized fraction. These fndings were not apparent until the curve-ftting technique was employed, which has the capacity to quantify many overlapped bands in the spectra. This study demonstrates that the curve-ftting of MIR spectra advances the analysis of organic matter in soil samples.**

Soil organic carbon (SOC) content and quality were found to vary with tillage systems and soil depth^{[1](#page-7-0)-3}, with crop rotation^{[4](#page-7-2)}, as well as with particle size fractions^{5[,6](#page-7-4)}. Characterization of SOC chemical composition in different soil size fractions (e.g., sand, silt and clay) has shown that diferent-sized minerals potentially protect diferent types of SOC, thus physical fractionation of primary organo-mineral complexes provides a useful approach to understand the dynamics and functional attributes of $\text{SOC}^{7,8}$ $\text{SOC}^{7,8}$ $\text{SOC}^{7,8}$ $\text{SOC}^{7,8}$. Moreover, the SOC in physical size fractions responds differently to soil cultivation and management changes⁹. The content of SOC in sand sized fractions is enriched in plant residues, while the SOC associated with silt particles is enriched in microbially-transformed plant residues^{[10](#page-7-8)}. Soil organic C has been found to be significantly correlated to clay content in numerous studies^{[11](#page-7-9),[12](#page-7-10)}.

The ¹³C NMR spectroscopy analysis is a powerful means for studying the SOC characteristics associated with soil particle-size fractions^{10[,13](#page-7-11)}. Despite its efficacy, NMR technique needs significant economic resources, material, equipment and tim[e14.](#page-7-12) Other techniques have also been used to study SOC properties, such as pyrolysis-feld ionization mass spectrometry. Using this technique, the abundance of carbohydrates, phenols and lignin monomers, alkylaromatics and N-containing compounds was found be vary distinctly among the physical fractions¹⁵. It was found using laser induced fuorescence analysis that the SOC was less humifed under NT than under

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other tillage practices and the SOC associated with clay fraction in particular was less humifed than associated with other particle size fractions of an Oxisol¹⁶. Infrared (IR) spectroscopy is used to identify the chemical compounds and their specifc functional groups for many organic materials including soils[17](#page-7-15). In particular, mid-IR (MIR) is becoming increasingly common due to the specifcity of the absorption bands. Diferent intensities of carboxylic, aromatic, CH and NH groups in a variety of bulk soils and silt fractions with diferent fertilizer amendments were reported, and these findings were based on relative absorbance of distinct peaks^{[18](#page-7-16),[19](#page-7-17)}. Calderón *et al*. [20](#page-7-18) investigated the changes of SOC chemistry in diferent SOC fractions during incubation using MIR spectroscopy. The variation of soil organic matter quality among differently managed sites was detected using MIR spectroscopy²¹. However, limited work has been done on differentiating the characteristics of SOC in soil particle size fractions using MIR spectroscopy, particularly with soils under diferent tillage systems. Furthermore, most studies were completed using the original MIR spectra where important absorption information may be missing in the overlapping bands due to the unavoidable interferences with minerals. Overlapping bands make it difficult to directly evaluate parameters such as position, width and area²². To resolve this problem, a curve-fitting technique, mainly based on Levenberg-Marquardt algorithm[23,](#page-7-21)[24,](#page-7-22) has been proposed as a useful and robust method to estimate these band parameters due to its potential in resolving overlapped bands into distinct peaks²⁵⁻²⁷. It has been reported that the curve ftting method provided more feasible results for the band shifs than the second derivative method in the analysis of overlapping overtones²⁵. The principle behind the curve-fitting technique is to represent peaks using analytical functions with some undetermined parameters and optimize these parameters to approximate the actual curve²⁸. Lin *et al.*²⁹ successfully discriminated the compositions in loratadine/hydroxypropyl-ß-cyclodextrin inclusion complex using curve-ftting analysis of Fourier-transform infrared (FTIR) spectroscopy. Other studies have also validated the curve-fitting method in medical science³⁰ and optical engineering applications³¹. O'Shaughnessy *et al.³²* revealed properties of caesium silicate glasses at different temperature by identifying curve ftted bands based upon Raman spectroscopy. Katsumi *et al*. [33](#page-7-30) revealed properties of humic substances from Japanese Andisols, Inceptisols and Entisols by identifying curve ftted bands based upon X-ray difraction spectroscopy. Nevertheless, there is a lack of information using the curve-ftting techniques for SOC analysis, especially for tillage induced SOC changes. We hypothesize that curve-ftting of MIR spectrum of soils will reveal significant more SOC information than the MIR spectrum itself. The objectives of this study were to (i) use curve-ftting to identify functional groups from the MIR spectra of soil samples which have distinct overlapping bands and (ii) assess whether curve-ftting can reveal the impacts of tillage practices on structural features of SOC in light of our current understanding of SOC associated with physical-size fractions in soil.

Materials and Methods

Site description and soil sampling. Soil samples were collected from the 0–10 cm layer of a tillage trial conducted on Brookston clay loam soil (poorly drained; lacustrine; mixed, mesic Typic Argiaquoll) at the Eugene Whelan Research Farm, Agriculture and Agri-Food Canada, Woodslee, Ontario (42°13'N, 82°44'W). The mean annual air temperature at the study site is 8.9°C and the mean annual precipitation is 832 mm. The average texture in the top 10 cm was 258 g kg⁻¹ sand, 336 g kg⁻¹ silt, and 406 g kg⁻¹ clay. The averaged soil pH and soil organic matter content was 7.28 and 43.9g kg⁻¹, respectively. Five treatments were involved in this study, including (1) a long-term no-tillage established in 1983 (NT83), (2) a long-term conventional tillage established in 1983 (CT83), (3) a short-term no-tillage established in 1997 (NT97) afer 13 years of CT, (4) a short-term conventional tillage established in 1997 (CT97) afer 13 years of NT, and (5) a long-term Kentucky bluegrass SOD established in 1983 (SOD83). All tillage treatments were under a corn-soybean rotation and soil samples were collected from the corn phase. All treatments were arranged in complete randomized block design with two feld replicates.

Soil particle fractionation. Soil samples were separated into particle size fractions, including sand- (2000-53 μm), silt- (53-2 μm), and clay- ($<$ 2 μm) size fractions (organic-mineral complexes) using a sonication pro-cedure as described in a previous study^{[34](#page-8-0)} in 4 replications. Briefly, 20 g of bulk soil ($\langle 2 \text{ mm} \rangle$ were placed in a 250 mL breaker with 80 mL of distilled water. The soil-water suspensions were treated with ultrasound using an energy level of 750 J mL⁻¹ and were then washed through a 53 μ m sieve with distilled water into a 2 L cylinder. The sand-size fraction was washed and transferred into an aluminium container, and then oven-dried at 55 °C. The clay was separated from the silt by repeated gravity sedimentation 35 , and then the clay and silt fractions were concentrated using centrifugation. The clay and silt pellets were treated by the method of Gregorich et al.^{[36](#page-8-2)}. The clay, silt and sand fractions refer to clay-, silt- and sand-sized organic-mineral complexes in this study. The SOC concentrations of the bulk soil and particle size fractions were determined using a Leco CNS 2000 analyzer (Leco Cor., St. Joseph, MI) (Table [1](#page-2-0)). The surface soils were free of carbonates, so total C was equivalent to organic carbon as reported by Zhang et al.³⁴. The samples of sand size fractions were ground with a mortar and pestle and stored for MIR spectra collection.

Fourier-transform difuse refectance MIR spectroscopic analysis. About 100 mg of each sample, including clay-, silt- and sand-sized fraction, were scooped into a stainless-steel sample cup, levelled with a spatula and placed on a Pike EasyDiff assembly (Easidiff, Pike Technologies, Madison, WI) on a Bruker-TENSOR 37 spectrometer (Bruker Optik GmbH, Ettlingen, Germany) for difuse refectance Fourier Transform MIR spectral collection. Tree separate MIR spectra for each sample were obtained from three subsamples in the range of 4000 to 500 cm⁻¹ with 64 scans and 2 cm⁻¹ resolution. The background spectrum was determined using potassium bro-mide (KBr)^{[37](#page-8-3)}. Spectra data were pre-processed and evaluated with the OPUS-6.5 software (Bruker Optik GmbH, Germany) using spectra averaging, baseline correction and peak picking techniques.

Curve-ftting was performed using the OPUS-6.5 Sofware to handle the complex spectra of overlapped bands on the basis of the second derivative and Levenberg-Marquardt algorithm and the Lorentz method for the peak shape. The curve-fitting is also known as band-fitting or non-linear regression. It is a mathematical tool for

Table 1. The concentrations of soil organic carbon in whole soil and soil particle size fractions (0–10 cm). Values are means of four replicates. † SOD83, long-term Kentucky bluegrass established in 1983; CT83, long-term conventional tillage established in 1983; CT97, short-term conventional tillage established in 1997 following 13 years of no-tillage (NT); NT83, long-term NT established in 1983; NT97, short-term NT established in 1997 following 13 years of CT.

Figure 1. An example of curve-ftting analysis of mid-infrared spectrum (SOD soil). (**a**) Deconvoluted spectra resolved from curve-ftting, (**b**) ftted spectrum, and (**c**) the original spectrum. Insert: curve-ftting process applied to the range of 1280–1917 cm[−]¹ .

modeling the experimental data. The model adjustable parameters, contained in the vector α of length M, are varied until the best agreement between data and model is obtained. The least square solution is considered to be the best under most conditions obtained by minimizing the objective function $\sum_{i=1}^{N} (y(vi) - f(vi; \alpha))^2$, where the experimental data as a function of the variable v is given by *y*(*vi*) and the number of experimental data by *N*. The fitted data is given by $f(v_i; \alpha)$. The number of bands, their positions, intensities and widths were determined by fitting a highly deconvoluted spectrum. The deconvolution of spectra would allow the partial resolution of otherwise overlapped bands. The procedure of curve-fitting in this study was as follows²²: (i) the frequency range of a spectrum were selected afer baseline correction, (ii) overlapped peaks were identifed and selected for analysis, (iii) curve-ftting of the deconvoluted spectrum to the original spectrum was performed using the curve ft module of the OPUS-6.5 Software. The standard error was controlled between 0.02-0.04 which indicates the accuracy and efficacy of the fitting process during calculation of single component of overlapped bands with the module. For all particle fractions under short- and long-term tillage managements as well as SOD83, the entire frequency range of 4000-500 cm⁻¹ was deconvoluted and curve-fitted to evaluate the impacts of tillage practices on the compositions of SOC. The curve-fitting process in the frequency range from 1280–1917 cm⁻¹ of the SOD83 soil was given as an example to illustrate the feasibility of this method for qualitative and quantitative study of SOC (Fig. [1\)](#page-2-1). Infrared absorption bands of functional groups in soil samples were assigned following a literature review and the functional groups belonging to aliphatic and aromatic types were identifed (Table [2](#page-3-0)). At identifed peaks, upper and lower boundaries were established, afer which a baseline was drawn between the boundaries and an integration was performed to calculate the corrected peak area. Peak area integration on the corrected spectra was then performed using the spectral processing software OPUS 6.5 (Bruker Optik GmbH)³⁸. The intensity and integral of either aliphatic or aromatic group were the sum of the relative absorption intensity and peak area of single peaks in either group for a given sample and were used as relative quantity of diferent functional groups. Four replicates were conducted for spectra collection and analysis.

Table 2. Absorption bands in MIR spectra and assignments. [†]Indicates aliphatic carbon; [‡]indicates aromatic carbon.

Statistical analysis. Analysis of variance (ANOVA) was conducted on the relative intensity and integral of diferent functional groups in particle size samples of soils under the four tillage managements as well as SOD83 before and after curve-fitting using SPSS 13.0 software package. The ANOVA was performed on the distribution of aliphatic and aromatic C in soil particle size fractions for each tillage treatments. When ANOVA was signifcant, the comparisons of aliphatic or aromatic C content among particle fractions and among tillage systems were examined using least signifcant diference test (LSD) at the 5% probability level.

Results and Discussion

Absorption bands of MIR spectrum in different particle size samples. There were noticeable differences in the MIR spectra between the particle size fractions (Fig. [2](#page-4-0)). Most of the major absorption bands found in this study could be interpreted using published results $39-43$ $39-43$. The spectrum of the clay samples generally showed less distinct absorption bands compared with other particle fractions. The sharp peak at around 3625 cm⁻¹ could be due to the stretching of -OH and Fe oxides in soil minerals, and the broad band at around 3380 cm⁻¹ is due to chemically combined water in the clay, the stretching of phenolic OH groups, or the hydrogen-bonded OH. The intensity of these two bands was the strongest in clay fraction and the weakest in sand fraction as expected. The peaks between 1990 cm⁻¹ and 1790 cm⁻¹ were evident in the sand and silt fractions but not in the clay fraction, which is indicative of the presence of quartz in these soil fractions. The peaks between 1620 cm⁻¹ and 1600 cm⁻¹ are indicative of aromatic C=C stretching and/or asymmetric –COO stretching and these were most pronounced in the sand fraction. Similarly, the peak at 1350 cm⁻¹ (symmetric -COO stretching and/or bending of aliphatics) was most intense in sand and silt fractions, but was not that intense in clay fraction. The peak at 1270 cm^{-1} (phenolic –OH) was noticeable in the spectrum clay sample, but missing in the spectra of the other particle size fractions. The peak at 1170 cm⁻¹ (aliphatic -OH or polysaccharides stretching) was weak in sand fraction but intense in clay fraction, which indicates that the clay could have higher absorption capacity towards polysaccharides and other aliphatic –OH rich substance rich relative to the coarser size fractions.

Curve-fitting analysis: SOC under long-term Kentucky Bluegrass management. The absorption bands in an original spectrum only indicate the existence of certain functional groups, such as aromatic rings, phenols, alcohols, etc., in the corresponding sample, but do not reveal meaningful quantitative information. Tis is mainly due to the many overlapped and unidentifed absorption bands in the spectrum, as was evident in Fig. [1.](#page-2-1) To resolve this problem, curve-ftting techniques identifed multiple peaks with relative peak intensity and peak area as was depicted between the 1280 cm⁻¹ and 1917 cm⁻¹ range in Fig. [1.](#page-2-1) The components under the overlapped bands include carboxylic acid, phenolic compounds, aliphatic -CH, aromatic C, etc. (Table [2\)](#page-3-0). The relative intensities of the resolved peaks were considerably diferent from the original spectrum with overlapped bands (Fig. [1\)](#page-2-1), indicating that curve-ftting technique enables quantitative assessment of functional groups of SOC. Using this technique, Calemma *et al*. [44](#page-8-7) resolved two bands from an FTIR spectrum of an oxidized coal sample into nine bands, and Rennert *et al*. [45](#page-8-8) quantitatively defned the relative quantity of functional groups in clay and clay plus silt samples in the frequency range of 1800 - 1200 cm⁻¹.

Overlapped individual bands were resolved into distinct peaks by applying curve-ftting analysis to the entire MIR range (4000 to 500 cm[−]¹) of the SOD83 sample spectra (Table [3\)](#page-5-0), afer which the relative intensity and

area for each individual band were calculated. The integral (the area of absorption band) was calculated using both band intensity and width parameters and could thus provide additional information on the quantity of the components. The intensity and integral of aromatic C were highest in clay-sized fraction, with 1.3 and 2.4 times greater integral than in silt- and sand-sized fractions, respectively. The relatively low aromatic C content in coarse-sized fractions of sand was consistent with the fndings that this fraction was generally enriched in light materials containing partly decomposed plant remnants with low recalcitrance⁴⁶. The intensity of aliphatic C (normally classifed as readily decomposable C) was highest in sand fraction. It is worth noting that the intensity and integral of aliphatic C were noticeably lower in silt than in other fractions and the integral was signifcantly lower than sand fraction (*P*<0.05), indicating less readily decomposable C in silt-sized relative to in sand- even clay-sized fractions of Brookston clay loam soil. These results agree with Bayer *et al.^{[47](#page-8-13)}* who found a higher concentration of semiquinone free radicals in fne-silt (20-2 μm) fraction using electron spin resonance spectroscopy which is indicative of a higher C recalcitrance in this fraction than in other fractions. The aliphatic C content in clay fraction was 34.7% higher than that in silt fraction, indicating that a portion of C associated with clay is not recalcitrant. Tis demonstrates that the relatively labile C is physically associated with mineral-particles in the clay fraction⁴⁸, in addition to the recalcitrant C. This result was consistent with Kadono *et al.*⁴⁹ who noted

Table 3. Relative intensity and integral of diferent functional groups in particle size samples of SOD soil before and afer curve-ftting. Values are means of four replicates. ‡ Diferent uppercase or lowercase letters in the same column indicate significant difference at 0.05 probability level. [§]There was no integral calculation before curveftting.

Figure 3. MIR spectra of soil particle size fractions from soils (0–10 cm) under diferent tillage practices. CT83=long-term conventional tillage established in 1983; CT97=short-term conventional tillage established in 1997 following 13 years of NT; NT83 = long-term no-tillage established in 1983; NT97 = short-term notillage established in 1997 following 13 years of CT.

the readily decomposable nature of clay-associated SOC. Christensen⁵⁰ and Zhang *et al.*³⁴ also found a shorter half-life for the readily decomposable C in clay- relative to in silt-sized fraction, indicating that the readily decomposable C was more easily mineralized in clay- than in silt-sized fraction. The intensity of C connected with oxygen atoms, which is refective of the oxidization of SOC, was 1.21 and 1.94 times higher in the clay- than in

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Table 4. Impacts of tillage practices on distributions of aliphatic and aromatic C in soil particle size fractions. Values are means of four replicates. † CT83, long-term conventional tillage established in 1983; CT97, shortterm conventional tillage established in 1997 following 13 years of no-tillage (NT); NT83, long-term no-tillage established in 1983; NT97, short-term tillage established in 1997 following 13 years of CT. [§]Different letters in the same column within each fraction indicate signifcant diference at 0.05 probability level.

the silt- and the sand-sized fractions, respectively ($P < 0.05$; Table [3](#page-5-0)). This was consistent with the ranking of the aromatic C distribution among the fractions.

These results demonstrate that the curve-fitting technique for MIR spectra of soil particle size fractions can readily identify important spectral information which is not evident in the original MIR spectra. The structural characteristics of SOC revealed through the curve- ftting technique generally agree with the results obtained using other methods such as pyrolysis-field ionization mass spectrometry, ¹³C NMR spectroscopy, and electron spin resonance spectroscopy⁴⁷⁻⁵².

Tillage-induced changes in carbon functional groups in particle size fractions. Tillage treatments resulted in large diferences in the MIR spectra of sand-sized fractions whereas only small diferences were observed for the silt- and especially the clay-sized fractions (Fig. [3](#page-5-1)). The similar responses of sand MIR spectra to tillage changes was also observed in other North American soils^{[9](#page-7-7),[53](#page-8-18)}. This is consistent with the fact that the organic matter in the sand-sized fraction is the frst pool that organic materials (typically plant residue and manure) enter into soil organic matter pool⁵⁴. The coarse fraction of organic matter is composed of plant and fungal fragments and is considered to be the most sensitive pool which can refect changes in soil organic matter due to changes in soil management practices^{[55](#page-8-20)[,56](#page-8-21)}; while organic matter in fine-sized particles undergoes relatively slow changes during cultivation and contributes little to the short- and medium-term fertility of soils^{[57](#page-8-22),[58](#page-8-23)}.

Curve-ftting revealed that the aliphatic C in sand-sized fraction was also signifcantly higher under NT83 than under CT83 (Table [4\)](#page-6-0). Aliphatic C content in the sand fraction was generally lower in the CT97 and CT83 than in NT97 and NT83 soils, indicating that the aliphatic C tends to be mineralized in conventional managed soils more so than in conservation managed soils. The integral of the aromatic C in the sand-sized fraction was 7.0%, 14.3% and 15.5% greater under CT83 than under CT97, NT97 and NT83, respectively. As a result, aromatic to aliphatic C ratio in sand-sized fraction, which in some degree refects SOC recalcitrance, was signifcantly greater under CT83 than under NT83 (P<0.05). This might be due to the more oxidative environment under CT as compared to NT management^{[59](#page-8-24)}. These results showed that C accumulation in coarse particles under NT was primarily in an easily decomposable form, a phenomenon which was observed for both long- and short-term NT. The current results were consistent with Bayer *et al.*⁵⁹ who reported less humification of SOC under NT compared to CT in a soil where they examined the semi-quinone free radicals using electron spin resonance technique. Similar results were also reported by Milori *et al*. [60](#page-8-25), González-Pérez *et al*. [61](#page-8-26) and Martins *et al*. [16](#page-7-14) using the laser-induced fuorescence spectroscopy technique. Results from 13C NMR spectroscopy further showed that cultivation could signifcantly increase SOC mineralization and reduce the concentration of mineral-associated organic carbon^{[62](#page-8-27)}. Using solid-state ¹³C NMR technique, Ding *et al.*^{[51](#page-8-28)} found less aliphatic and more aromatic carbon from soils under conventional compared to under conservation management.

In this study, the aromatic C in the silt-sized fraction was surprisingly higher under NT83 than under other treatments, indicating that long-term NT contributed to recalcitrant C storage in this size fraction. Aromatic/aliphatic C ratio was highest in the silt- as compared with other-sized fractions for all treatments. Tis is consistent with García-Oliva et al.⁶³ who found that the silt-associated SOC had a slower depletion rate than other fractions when forest was converted to pasture. Saab and Martin-Neto⁶⁴ also found that C was more stable in the silt- than in other-sized fractions of a Gleysol employing the electron paramagnetic resonance technique. The aromatic C was found dramati-cally higher in the clay- than in other-sized fractions for all treatments (Table [4](#page-6-0)). This agrees with the findings that the clay-sized fraction may preserve SOC from microbial attack by adsorbing SOC onto clay surfaces, by encapsulation of clay particles or entrapment of SOC in pores⁶⁵. There were no significant differences ($P > 0.05$) of aliphatic and aromatic C in clay- and silt-sized fractions of NT, compared with CT soil (Table [4\)](#page-6-0), which indicates that C in these fractions is not as sensitive to changes in land use management practices compared to the sand-size fractions⁶⁶.

Conclusions

The curve-fitting technique, used in combination with Fourier-transform mid-infrared spectra, can be a robust tool for determining the functional groups and structural features of soil organic carbon in soil particle size samples. Numerous overlapping absorption bands in MIR spectra of soil samples were enhanced and identifed using this technique which advances our ability to qualitatively and semi-quantitatively analyze the spectrum and organic components of the samples. The curve-fitting analysis for mid-infrared spectra further indicates that tillage-induced diferences in SOC characteristics were most obvious in sand-sized fraction with signifcant (i) greater aliphatic C content and (ii) smaller aromatic/aliphatic C ratio in long term NT compared to long-term CT.

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Author Contributions

R.Q.F. and X.M.Y. wrote the main manuscript text; C.F.D. designed the experiments and Z.H.Z. prepared fgures. All authors reviewed the manuscript.

Additional Information

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