## PAPER

# Synthesis and bioevaluation of thienopyrimidines bearing a pyrazoline unit as selective PI3Kα inhibitors†

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A series of thienopyrimidines containing a pyrazoline unit (**4a–d**, **7a–d** and **13a–l**) were designed and synthesized. The target compounds were evaluated for antiproliferative activity against A549, HepG2 and MCF-7 cancer cell lines. Among the twenty target compounds, most of them exhibited excellent antiproliferative activity against one or several cancer cell lines. Compound **13f** showed the best activity against A549, MCF-7 and HepG2 cancer cell lines, with $IC_{50}$ values of 2.84 ± 0.09 μM, 2.88 ± 0.43 μM and 2.08 ± 0.36 μM, respectively. Four selected compounds (**13c**, **13f**, **13g** and **13h**) were further evaluated for their inhibitory activity against the PI3Kα/mTOR protein kinase. Moreover, time-dependent and dose-dependent experiments, AO fluorescence staining, Annexin V-FITC/PI staining and docking studies were carried out in this study. The results indicated that compound **13f** may be a potential selective PI3Kα inhibitor.

## 1. Introduction

Phosphatidylinositol 3-kinases (PI3Ks) are a family of enzymes involved in cellular functions such as cell growth, proliferation, differentiation, migration, mobility and apoptosis, which in turn are involved in cancer.[1,2] PI3Ks are currently divided into four different classes: Class I, Class II, Class III, and Class IV.[3] The Class I family consists of three isoforms, α, β, and δ.[4] Among the different PI3K subfamily proteins, PI3Kα is the most important isoform in cell proliferation in response to growth factor-tyrosine kinase pathway activation.[5,6] There are currently more than ten PI3Kα inhibitors in clinical trials, including GDC-0941 and PI-103.[7,8] Many research groups are attempting to develop some more PI3Kα inhibitors.

In our previous research, a series of thianopyrimidine derivatives (compound I, Fig. 1) were designed and synthesized as PI3K inhibitors. Some of them showed moderate to excellent antiproliferative activity. The structure-activity relationships (SARs) and docking studies exhibited that chiral carbon atoms may affect antiproliferative activity *in vitro*.[9]

To further confirm the effect of the chiral carbon atoms to the activity of the target compounds, we replaced the chiral carbon atoms of the pyrazolines by C=C double bond, and at the same time, inspired by GDC-0941, we replaced the 7,8-dihydro-5*H*-thiopyrano[4,3-*d*]pyrimidine with the thieno[3,2-*d*]pyrimidine structure, resulting in target compounds **4a–d**. Similarly, compounds **7a–d** were designed and synthesized *via* bioisosterism from compounds **4a–d**. To our disappointment, compounds **4a–d** and **7a–d** didn't exhibit excellent antiproliferative activity as expected against the tested cancer cell lines. To our delight, compounds **7a–d** showed better activity than compounds **4a–d**. Therefore, we continued to structurally modify the compounds based on the series of compounds **7a–d**. Inspired by anti-NSCLC inhibitor AZD-9291, we replaced the morpholine by *N,N,N*-trimethylethylenediamine and reintroduced the chiral carbon and aryl group to the target compounds. As a result, compounds **13a–l** were designed and synthesized. The design strategy for all target compounds is shown in Fig. 2.

PI3K–Akt–mTOR pathway plays a key role in the growth and proliferation of many kinds of cancer such as lung cancer, breast cancer, prostate cancer, lymphoma, ovarian cancer and liver cancer.[10,11] It was reported that the A549 (human lung cancer), HepG2 (human liver cancer), and MCF-7 (human breast cancer) cancer cell lines are highly sensitive to PI3K/Akt/mTOR signaling pathway inhibitors.[12] Therefore, we disclosed the synthesis and antiproliferative activity of all the target compounds against the A549, HepG2, and MCF-7 cancer cell lines, and the inhibitory activity against PI3Kα and mTOR kinases. Moreover, time- and dose-dependent experiments, AO

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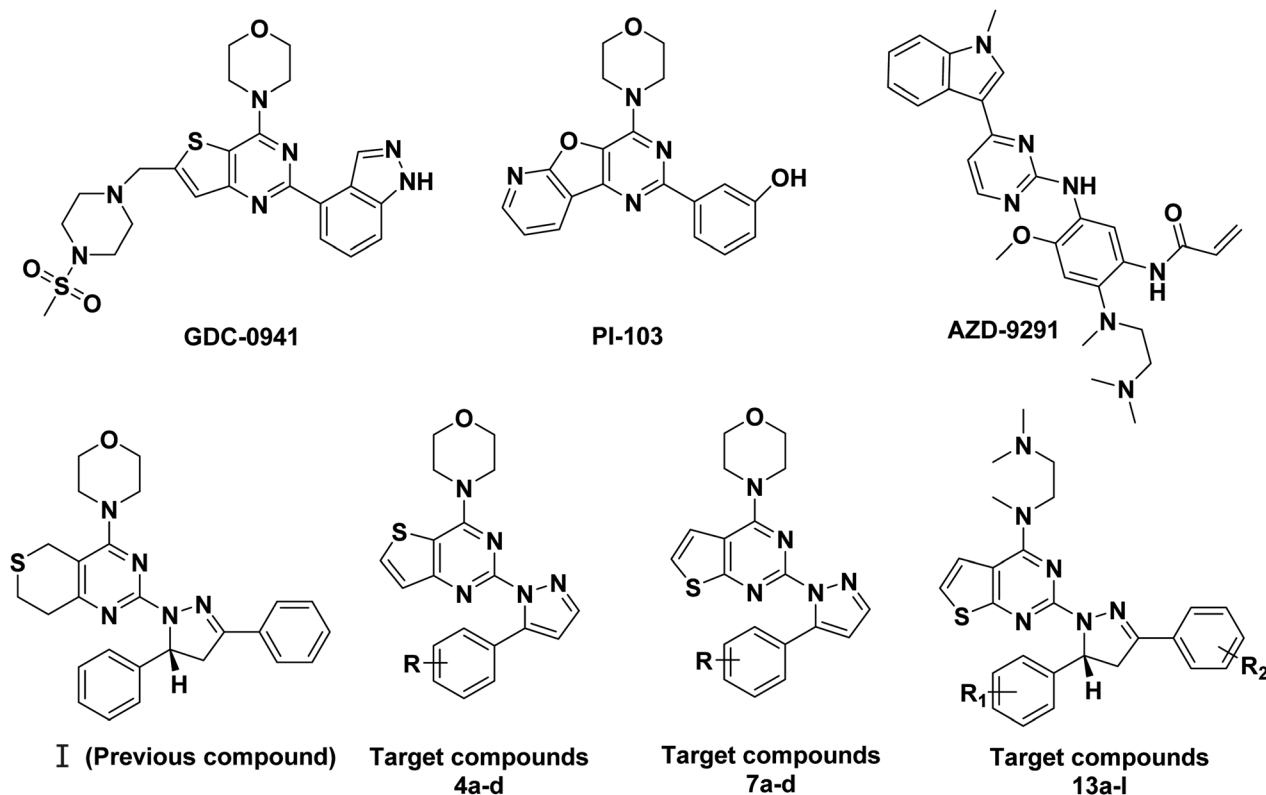‡ These authors contribute equally to this work.

Fig. 1 Structures of representative compounds and target compounds.

and Annexin V-FITC/PI staining and docking studies were also carried out.

## 2. Chemistry

The synthetic routes of target compounds **4a–d**, **7a–d** and **13a–l** are described in Schemes 1–3.

The intermediates (**1**, **5** and **11**) were synthesized from commercially available methyl 2-aminothiophene-3-carboxylate or ethyl 2-amino-4,5,6,7-tetrahydrobenzo[*b*]thiophene-3-carboxylate through five steps or six steps, respectively, which were reported in our previously research.[10] Then, intermediates **1**, **5** and **11** were each treated with 80% hydrazine monohydrate in refluxing ethanol, generating **2**, **6** and **12**, respectively.[13,14] Nextly, **2** and **6** were each reacted with phenylpropenone (**3a–d**), and the reaction was carried out for 1 hour to obtain the compounds **4a–d** and **7a–d**. The key intermediate **12** condensed with the corresponding chalcone (**10a–l**) to afford the target compounds **13a–l**.

## 3. Results and discussion

### 3.1. Biological evaluation

Taking GDC-0941 as the reference compound, the target compounds (**4a–d**, **7a–d** and **13a–l**) were evaluated for the antiproliferative activity against three cancer cell lines A549, MCF-7, and HepG2 by the 3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2-*H*-tetrazolium bromide (MTT) method. In addition,

the activity against PI3Kα/mTOR kinase of the most promising compound **13f** was further evaluated. The results expressed as $IC_{50}$ values were summarized in Tables 1–3, where the values were the average of at least two independent experiments.

From the results in Table 1, most of the compounds of **4a–d** and **7a–d** were inactive against the three cancer cells. Among them, compounds **7a** and **7b** showed better activity than the others against the three tested cancer cell lines, and the antiproliferative activity of these two compounds against MCF-7 and HepG2 cancer cell lines was better than that of A549 cancer cell lines. In the series of compound **7a–d**, the substituent *R* on the benzene ring has an effect on the antiproliferative activity of the compound, and when substituent *R* is hydrogen and the methyl group, it's contributes an increase to the antiproliferative activity of the compound. Taken compounds **7a–d** and **4a–d** as comparison, we found that compounds with thieno[2,3-*d*]pyrimidine as the backbone structure showed better antiproliferative activity than the compounds with thieno[3,2-*d*]pyrimidine as the backbone structure.

From the results in Table 2, most of the compounds showed moderate to excellent antiproliferative activity against three cancer cells lines (A549, HepG2 and MCF-7). Among them, three compounds (**13f**, **13g** and **13h**) showed better activity against A549 cells lines compared with the positive compound GDC-0941. The most promising compound **13f** exhibited the best activity against A549, HepG2 and MCF-7 cancer cell lines with the $IC_{50}$ values of 2.84 ± 0.09 μM, 2.08 ± 0.36 μM and 2.88 ± 0.43 μM, respectively. The antiproliferative activity of
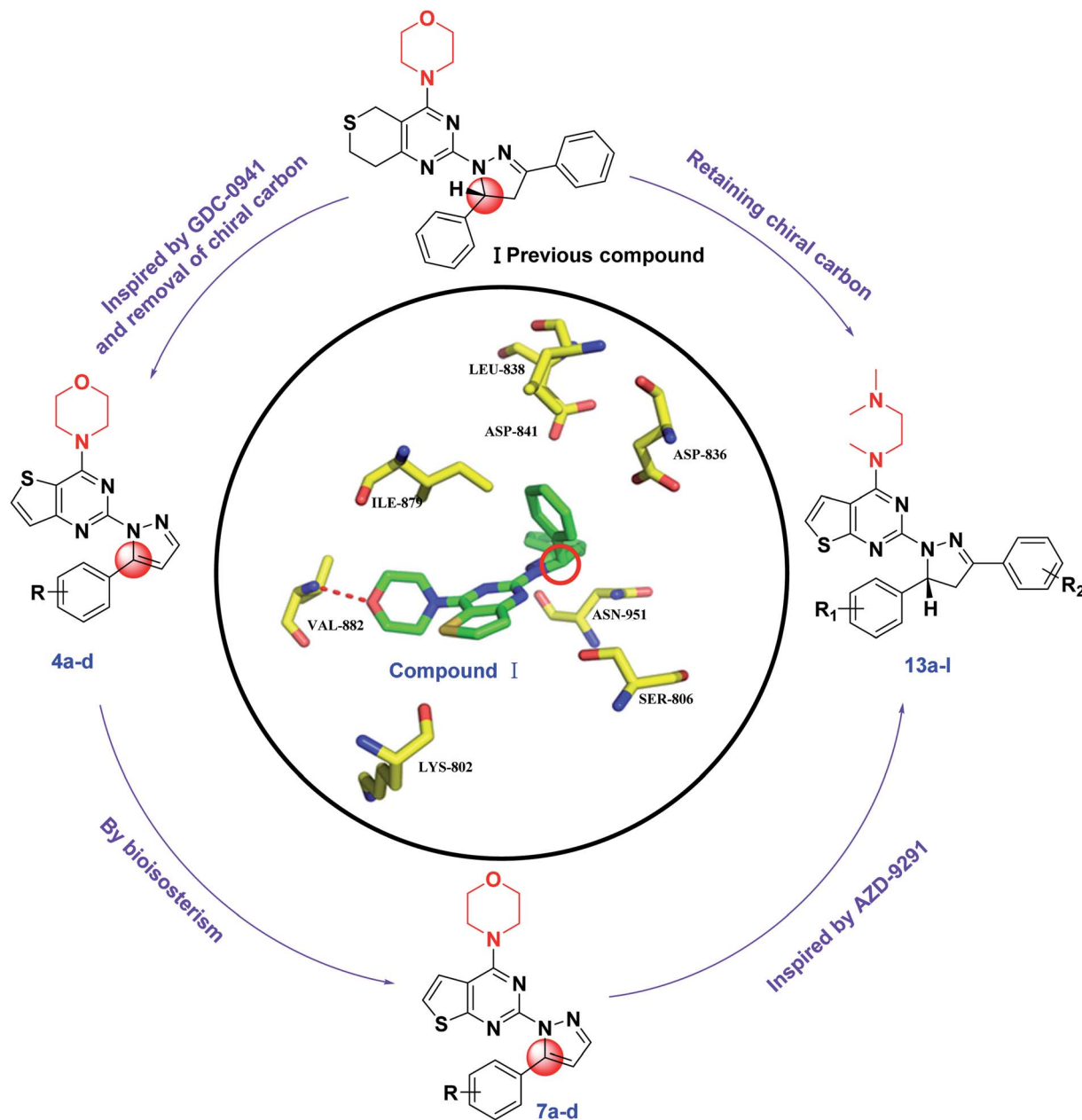
Fig. 2   Structures and design strategy for target compounds **4a–d**, **7a–d** and **13a–l**.

compound **13f** against A549 cancer cell lines was comparable to the positive control GDC-0941 (6.99 ± 0.21 μM). On the whole, the morpholine is replaced by *N,N,N*-trimethylethylenediamine, and the chiral carbon and aryl groups are reintroduced into the target compound, which is advantageous for the anti-proliferative activity of the compound. The introduction of substituents on $R_1$ and $R_2$ has different effects on anti-proliferative activity, and the introduction of electron donating groups is more conducive to improve antiproliferative activity. The effect on the activity about the introduction of substituents is probably related to a change in the electronic configuration of the benzopyrazole structure. What's more, compounds substituted with electron donating groups (CH₃, –OCH₃) in $R_1$

group showed better activity than those with none substituent. In general, it's seemed to be that target compounds with electron donating groups (EDGs) have a significant impact on the *in vitro* activity, such as compounds **13c**, **13f**, **13g** and **13h**.

Compared **4a–d**, **7a–d** and **13a–l**, it was found that the introduction of *N,N,N*-trimethylethylenediamine and aryl group to the thieno[2,3-*d*]pyrimidine part and reintroduced the chiral carbon had a significant impact to the antiproliferative activity. The electron cloud density distribution in the molecule of the compound is adapted to the specific site of the protein receptor, which facilitates the binding of the compound to the receptor and enhances the activity of the compounds. The 5-position benzene ring structure and *N,N,N*-trimethylethylenediamine
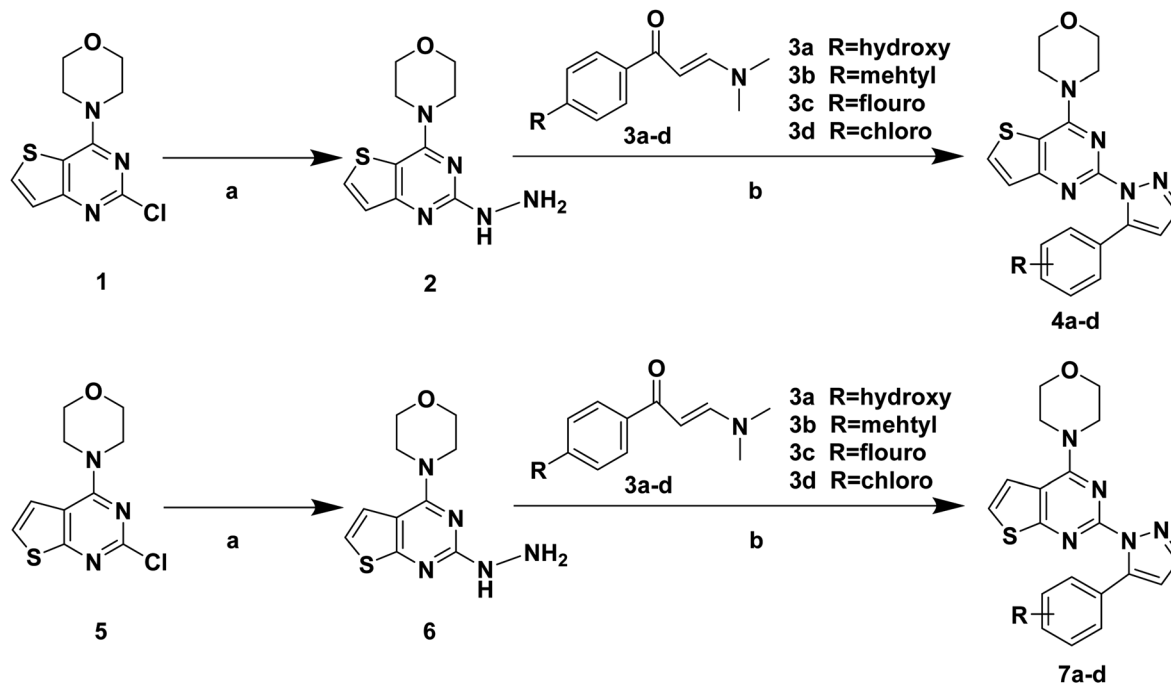
**Scheme 1** Synthetic routes of target compounds. Reagents and conditions: (a) 80% $NH_2NH_2 \cdot H_2O$, EtOH, 78 °C, 1 h; (b) 1 eq. HCl, EtOH, 70 °C, 6 h.

structure may promote the tight binding of the compounds to the protein receptor.

Activity against PI3Kα and mTOR kinase of four selected compounds **13c**, **13f**, **13g** and **13h** was further carried out in this paper. As shown in Table 3, these four selected compounds have better inhibitory activity with $IC_{50}$ values ranging from 0.92 μM to 7.16 μM against PI3Kα kinase than that against mTOR kinase with $IC_{50}$ higher than 10 μM. Overall, the inhibitory activity of the above four compounds against PI3Kα kinase is much better than that of mTOR kinase and compound **13f** is the most potential selective PI3Kα inhibitor.

### 3.2. Dose-dependent and time-dependent *in vitro* effects

Comparing the three series of compounds, the compound **13f** showed the best antiproliferative activity and enzyme

inhibitory activity. To further confirm whether the concentration of this compound has an effect on the inhibition of cell viability, the compound **13f** was formulated into seven different concentrations, and the inhibition rate of the seven concentrations on the three cancer cells was measured by the MTT method. As shown in Fig. 3a, the higher the concentration of the compound was, the better the inhibition activity was. The inhibition rate was positively correlated with the compound concentration, and the inhibition to cancer cells was dose-dependent.

In order to investigate the effect of time on inhibition of cell viability, compound **13f** was formulated into five different concentration gradients, and the inhibition rate of the compound against the same concentration of A549 cancer cell lines was determined by MTT assay. As shown in Fig. 3b, at the
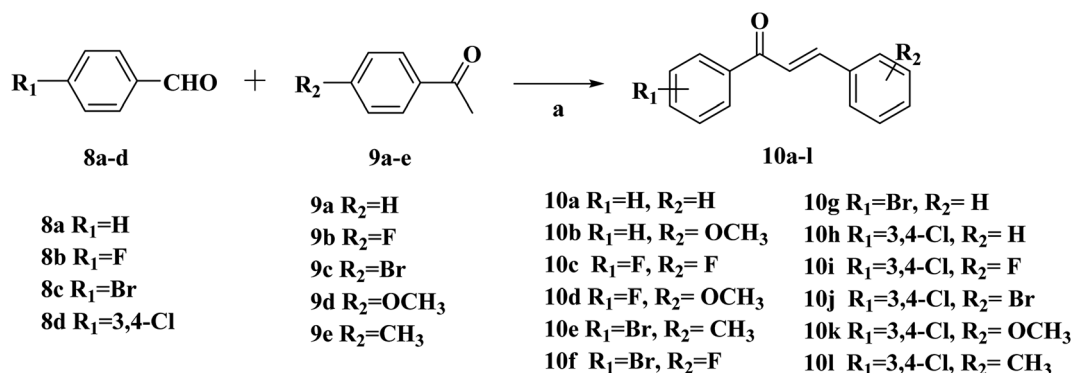


**Scheme 2** Synthetic routes of chalcone **10a–l**. Reagents and conditions: (a) 10% NaOH, EtOH, r.t., 24 h.
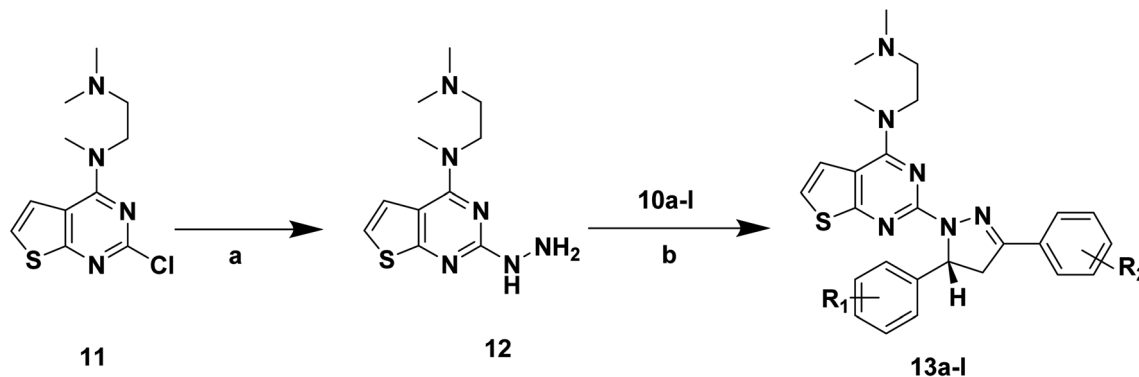
**Scheme 3** Synthetic routes of target compounds. Reagents and conditions: (a) 80% $NH_2NH_2 \cdot H_2O$, EtOH, 78 °C, 1 h; (b) glacial acetic acid, $H_2SO_4$; 100 °C.

same concentration, the inhibition rate of compound **13f** on A549 cancer cell lines was positively correlated with time, the longer the compound **13f** was exposed to A549 cancer cell lines, the higher the inhibition rate was.

Summarizing the relationship between the dose, time and the inhibition rate of compound **13f** on A549 cancer cell lines, it was found that compound **13f** inhibited the proliferation of A549 cancer cell lines in a dose-dependent and time-dependent manner.

**Table 1** *In vitro* cell viability of target compound **4a–d**, **7a–d**



| Compd | R | IC$_{50}$$^a$ (μM) | | |
|---|---|---|---|---|
| | | A549 | MCF-7 | HePG2 |
| **4a** | 4-H | >100 | >100 | >100 |
| **4b** | 4-CH$_3$ | >100 | >100 | >100 |
| **4c** | 4-F | >100 | >100 | >100 |
| **4d** | 4-Cl | >100 | >100 | >100 |
| **7a** | 4-H | 37.62 ± 1.88 | 25.30 ± 0.93 | 20.16 ± 1.15 |
| **7b** | 4-CH$_3$ | 35.04 ± 0.97 | 30.95 ± 1.05 | 33.94 ± 1.13 |
| **7c** | 4-F | >100 | >100 | >100 |
| **7d** | 4-Cl | >100 | >100 | >100 |
| GDC-0941$^b$ | — | 6.99 ± 0.21 | 0.07 ± 0.03 | 0.20 ± 0.08 |

$^a$ The values are an average of two separate determinations. $^b$ Used as a positive controls.

### 3.3. Morphology of cells by fluorescence microscopy

In order to investigate whether the target compound **13f** could induce the apoptosis of A549 cancer cell lines, acridine orange (AO) staining assay was carried out. As show in the Fig. 4, the control group cell (Fig. 4a) was treated with nothing and showed normal A549 cell shape. But in the test group (Fig. 4b), the cell shape was abnormal with cell shrinkage, chromatin condensation after compound **13f** acted on the A549 cells. The results indicated that the compounds **13f** could induce apoptosis of A549 cancer cell lines.
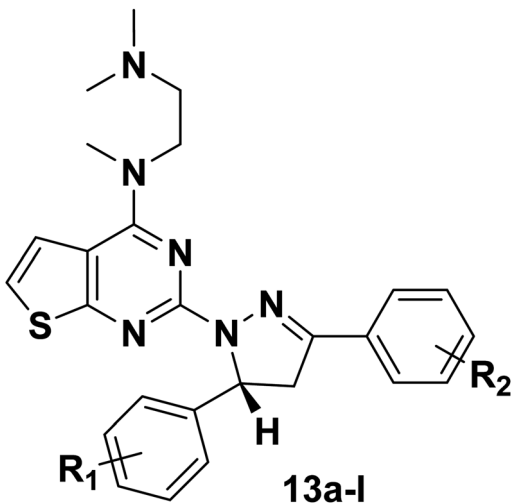
### 3.4. Apoptosis result analyzing

In order to further reveal the mechanism of apoptosis of A549 induced by the compound **13f**, the experiment of Annexin V-FITC and propidium iodide (PI) double staining was carried out. The results were shown in Fig. 5. Compared with the control group, compound **13f** with concentration of 1.6 μM could induce the late apoptotic significantly, with 7.68% late apoptotic and 9.14% early apoptotic. And the same trend was also observed in other concentration. In addition, the number of apoptotic and dead cells increased with increasing concentration of the compound. From this, it was found that the target compound **13f** can induce apoptosis in a concentration-dependent manner.

Through in-depth research, it was found that the synthesized target compounds can inhibit the proliferation of A549 cells in a time-dependent and dose-dependent manner. AO staining and Annexin V-FITC/PI flow cytometry showed the compound **13f** could induce apoptosis of human lung cancer A549 cells.

### 3.5. Molecular docking study

To explore the binding modes of target compounds with the active site of PI3Kα/mTOR, molecular docking simulation studies were carried out by the AutoDock4.2 software and the docking results were processed and modified in PyMOL 1.8.x software. Based on the *in vitro* inhibition results, we selected compound **13f** as ligand example, the structures of PI3Kα (PDB ID code: 3BDS) and mTOR (PDB ID code: 4JSV) were selected as the docking models. The results were shown in Fig. 6.

**Table 2** *In vitro* cell viability of target compound **13a–l**



| Compd | $R_1$ | $R_2$ | IC$_{50}$[a] (μM) | | |
| --- | --- | --- | --- | --- | --- |
| | | | A549 | MCF-7 | HepG2 |
| **13a** | 4-OCH$_3$ | 3,4-Di Cl | NA | NA | NA |
| **13b** | 4-Br | 3,4-Di Cl | 17.43 ± 0.13 | 12.82 ± 0.23 | 12.90 ± 0.64 |
| **13c** | 4-F | 4-F | 9.23 ± 0.02 | 6.51 ± 0.31 | 6.66 ± 0.13 |
| **13d** | 4-F | 3,4-Di Cl | 9.17 ± 0.32 | 6.82 ± 0.17 | 7.27 ± 0.68 |
| **13e** | 4-H | 3,4-Di Cl | 13.41 ± 0.29 | 10.48 ± 0.54 | 8.05 ± 0.05 |
| **13f** | 4-OCH$_3$ | 4-H | 2.84 ± 0.09 | 2.88 ± 0.43 | 2.08 ± 0.36 |
| **13g** | 4-CH$_3$ | 3,4-Di Cl | 8.75 ± 0.24 | 4.98 ± 0.04 | 6.96 ± 0.44 |
| **13h** | 4-CH$_3$ | 4-Br | 5.43 ± 0.17 | NA | 4.45 ± 0.25 |
| **13i** | 4-H | 4-H | 9.43 ± 0.10 | 5.04 ± 0.06 | 6.70 ± 0.14 |
| **13j** | 4-OCH$_3$ | 4-F | NA | NA | 83.86 ± 0.17 |
| **13k** | 4-F | 4-Br | 22.65 ± 0.35 | 18.39 ± 0.18 | 10.89 ± 0.20 |
| **13l** | 4-H | 4-Br | 22.95 ± 0.15 | 17.17 ± 0.48 | 15.79 ± 1.23 |
| GDC-0941[b] | — | — | 6.99 ± 0.21 | 0.20 ± 0.08 | 0.07 ± 0.03 |

[a] The values are an average of two separate determinations. [b] Used as a positive controls.

In the docking model of compound **13f** and PI3Kα protein kinase (Fig. 6a), the 5-position benzene ring structure is tightly packed by the hydrophobic group and is in a hydrophobic cavity, allowing the compound to bind tightly to the protein receptor. The *N,N,N*-trimethylethylenediamine structure forms a hydrogen bond with the hydrophilic amino acid residue HIS-658, which moderately enhances the solubility of the compound. Pyrazoline and pyrimidine ring structure are close to the hydrophobic amino acid residue ARG-849. As shown in Fig. 6b, it can be observed that compound **13f** has no binding to any amino acid residues in the periphery. This further explains the reason why compound **13f** is less active against mTOR kinase. These results of the molecular docking study showed that compound **13f** may be a potential PI3Kα inhibitor.

**Table 3** PI3Kα/mTOR kinase activity and cytotoxicity of selected compounds and positive controls

| Compd | IC$_{50}$[a] (μM) | |
| --- | --- | --- |
| | PI3Kα[c] | mTOR[d] |
| **13c** | 4.62 ± 0.54 | >10 |
| **13f** | 0.92 ± 0.14 | >10 |
| **13g** | 3.31 ± 0.12 | >10 |
| **13h** | 7.16 ± 0.27 | >10 |
| GDC-0941[b] | 0.003 | 0.58 |

[a] The values are an average of two separate determinations. [b] Used as a positive controls. [c] PI3Kα: phosphatidylinositol-3-kinase alpha subunit. [d] mTOR: mammalian target of rapamycin.

## 4. Conclusions

In summary, a series of thienopyrimidines containing pyrazoline unit (**4a–d**, **7a–d** and **13a–l**) were designed and synthesized. The pharmacological results indicated that most of the compounds showed moderate antiproliferative activity against three cancer cell lines. In particular, the compound **13f** showed the best activity against A549, MCF-7 and HepG2 cancer cell lines, with the IC$_{50}$ values of 2.84 ± 0.09 μM, 2.88 ± 0.43 μM and 2.08 ± 0.36 μM, respectively. Four selected compounds (**13c**,
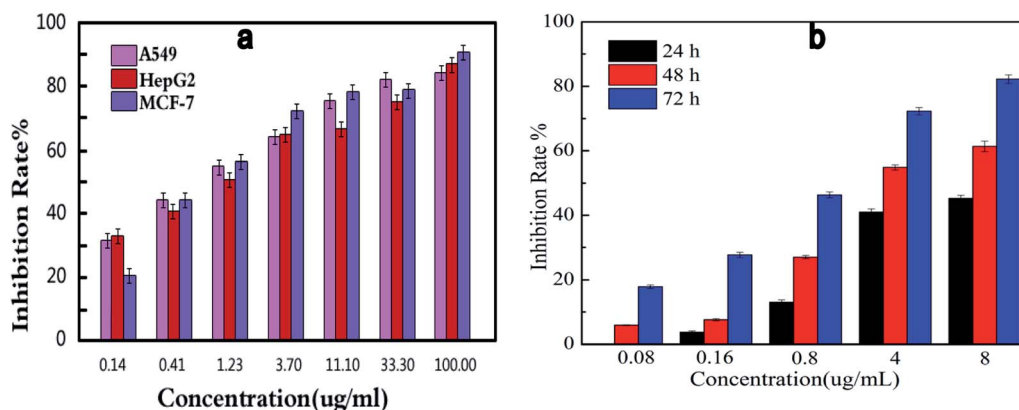
Fig. 3 Dose-dependent (a) and time-dependent (b) *in vitro* effects of compound **13f** on inhibitory activity against A549.

**13f, 13g** and **13h**) were further evaluated for the inhibitory activity against PI3Kα/mTOR protein kinase. In addition, based on further studies on the compound **13f**, it was found that **13f** can inhibit the proliferation of A549 cancer cell lines in a time-dependent and dose-dependent manner. AO staining and Annexin V-FITC/PI flow cytometry showed that the compound **13f** induced apoptosis of A549 cancer cell lines. The results prompted us that compound **13f** may be a potential selective PI3Kα inhibitor.

## 5.   Experimental section

### 5.1.   Chemistry

All melting points were obtained on a Büchi Melting Point B-540 apparatus (Büchi Labortechnik, Flawil, Switzerland) and were uncorrected. NMR spectra were performed using Bruker 400 MHz spectrometers (Bruker Bioscience, Billerica, MA, USA) with TMS as an internal standard. Mass spectra (MS) were taken in ESI mode on Agilent 1100LCMS (Agilent, Palo Alto, CA, USA). TLC analysis was carried out on silica gel plates GF254 (Qingdao Haiyang Chemical, China). All the materials were obtained from commercial suppliers and used without purification, unless otherwise specified. Yields were not optimized.

### 5.2.   General procedure for the preparation of compounds 1, 5 and 11

Compounds **1**, **5** and **11** were synthesized according to the reported procedures by our research group.[13,14]

### 5.3.   General procedure for the preparation of compounds 2, 6 and 12

Dissolve 2.5 g of **1** (4-(2-chlorothieno[2,3-*d*]pyrimidin-4-yl) morpholine) or **5** (4-(2-chlorothieno[3,2-*d*] pyrimidin-4-yl) morpholine) in 100 mL of hydrazine hydrate. Reaction at 80 °C for 5 hours, cool down to room temperature and precipitate a pale yellow solid. The solid was filtered off and the filter cake was dried to give the product.[15]

### 5.4.   General procedure for the preparation of target compounds 4a–d and 7a–d

**2** (4-(2-Mercaptothieno[2,3-*d*]pyrimidin-4-yl)morpholine) or **6** (4-(2-mercaptothieno[3,2-*d*]pyrimidin-4-yl)morpholine) (0.001 mol) and **3a–d** (phenyl propyl ketone) (0.001 mol) were dissolved in ethanol (12 mL), 1 eq. of HCl was used as a catalyst, and reacted at 70 °C for 6 hours. The reaction was stopped, cooled to room temperature, and the reaction solution was adjusted to a pH of 6–7. Solid precipitated, suction filtered, and
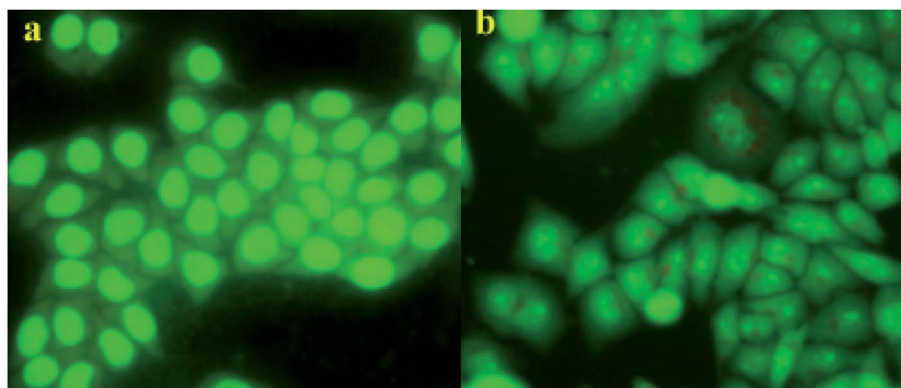


Fig. 4 Effect of compound **13f** on A549 cell by acridine orange (AO) single staining. ((a) is the control group, (b) is the test group).
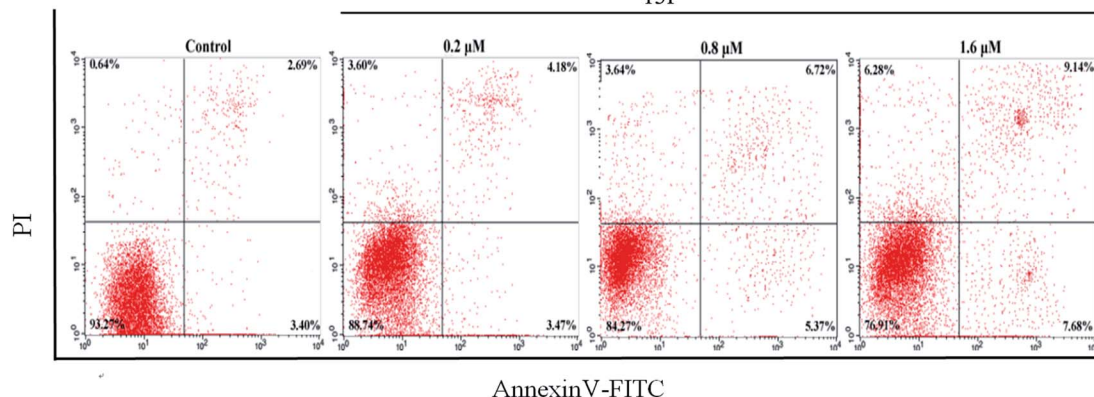
Fig. 5  Effect of compound 13f induced apoptosis on A549 cells using Annexin V-FITC and propidium iodide (PI) double staining by flow cytometry.

the filter cake dried. Due to the formation of by-products in the reaction, the dried filter cake was separated by column chromatography with $CH_2Cl_2 : AcOH = 4 : 1$, and the solvent was removed under reduced pressure to obtain pure target product 4a–d or 7a–d.

**4-(2-(5-Phenyl-1H-pyrazol-1-yl)thieno[3,2-d]pyrimidin-4-yl) morpholine (4a).** A yellow solid, mp 243.6–244.8 °C; ESI-MS m/z: $[M + H]^+$: 363.4; $^1$H-NMR (400 MHz, DMSO) δ 8.73 (d, J = 2.5 Hz, 1H), 7.95 (d, J = 7.4 Hz, 2H), 7.69 (d, J = 6.2 Hz, 1H), 7.60 (s, 1H), 7.46 (s, 2H), 7.37 (t, J = 7.3 Hz, 1H), 7.06 (s, 1H), 4.04–3.95 (m, 4H), 3.81–3.73 (m, 4H).

**4-(2-(5-(p-Tolyl)-1H-pyrazol-1-yl)thieno[3,2-d]pyrimidin-4-yl) morpholine (4b).** A yellow solid, mp 166.8–171.5 °C; ESI-MS m/z: $[M + H]^+$: 377.1; $^1$H-NMR (400 MHz, DMSO) δ 8.71 (s, 1H), 7.84 (d, J = 7.8 Hz, 2H), 7.68 (d, J = 6.1 Hz, 1H), 7.59 (d, J = 6.1 Hz, 1H), 7.27 (d, J = 7.7 Hz, 2H), 7.00 (s, 1H), 3.98 (d, J = 4.2 Hz, 4H), 3.77 (s, 4H), 2.34 (s, 3H).

**4-(2-(5-(4-Fluorophenyl)-1H-pyrazol-1-yl)thieno[3,2-d] pyrimidin-4-yl)morpholine (4c).** A yellow solid, mp 228–231 °C; ESI-MS m/z: $[M + H]^+$: 381.4; $^1$H-NMR (400 MHz, DMSO) δ 8.73 (d, J = 2.5 Hz, 1H), 8.00 (dd, J = 8.4, 5.6 Hz, 2H), 7.69 (d, J = 6.1 Hz, 1H), 7.59 (d, J = 6.1 Hz, 1H), 7.29 (t, J = 8.8 Hz, 2H), 7.05

(d, J = 2.5 Hz, 1H), 3.99 (d, J = 4.4 Hz, 4H), 3.78 (d, J = 4.3 Hz, 4H).

**4-(2-(5-(4-Chlorophenyl)-1H-pyrazol-1-yl)thieno[3,2-d] pyrimidin-4-yl)morpholine (4d).** A yellow solid, mp 250–251 °C; ESI-MS m/z: $[M + H]^+$: 397.8; $^1$H-NMR (400 MHz, DMSO) δ 8.74 (d, J = 2.5 Hz, 1H), 7.98 (d, J = 8.4 Hz, 2H), 7.69 (d, J = 6.1 Hz, 1H), 7.60 (d, J = 6.1 Hz, 1H), 7.52 (d, J = 8.4 Hz, 2H), 7.08 (d, J = 2.5 Hz, 1H), 3.99 (s, 4H), 3.78 (d, J = 4.3 Hz, 4H).

**4-(2-(5-Phenyl-1H-pyrazol-1-yl)thieno[2,3-d]pyrimidin-4-yl) morpholine (7a).** An orange yellow solid, mp 223–225 °C; ESI-MS m/z: $[M + H]^+$: 363.4; $^1$H-NMR (400 MHz, DMSO) δ 8.75 (s, 1H), 8.31 (dd, J = 5.5, 1.9 Hz, 1H), 7.96 (d, J = 7.2 Hz, 2H), 7.54 (dd, J = 5.5, 1.9 Hz, 1H), 7.47 (q, J = 7.4 Hz, 2H), 7.37 (t, J = 7.8 Hz, 1H), 7.08–7.00 (m, 1H), 4.03 (s, 4H), 3.83–3.77 (m, 4H).

**4-(2-(5-(p-Tolyl)-1H-pyrazol-1-yl)thieno[2,3-d]pyrimidin-4-yl) morpholine (7b).** A light yellow solid, mp 286–289 °C; ESI-MS m/z: $[M + H]^+$: 377.4; $^1$H-NMR (400 MHz, DMSO) δ 8.82 (d, J = 2.1 Hz, 1H), 8.41 (d, J = 5.5 Hz, 1H), 7.95 (d, J = 7.9 Hz, 2H), 7.64 (d, J = 5.4 Hz, 1H), 7.37 (d, J = 7.9 Hz, 2H), 7.10 (d, J = 2.2 Hz, 1H), 4.13 (s, 4H), 3.91 (d, J = 4.2 Hz, 4H), 2.45 (s, 3H).

**4-(2-(5-(4-Fluorophenyl)-1H-pyrazol-1-yl)thieno[2,3-d] pyrimidin-4-yl)morpholine (7c).** A light yellow solid, mp 202–205 °C; ESI-MS m/z: $[M + H]^+$: 381.4; $^1$H-NMR (400 MHz, DMSO)
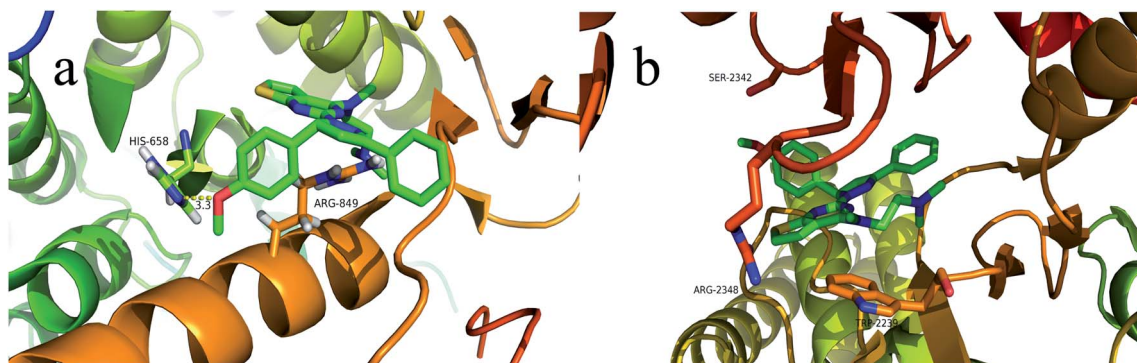


Fig. 6  Binding modes of compound 13f with PI3Kα/mTOR (a show the binding pattern of 13f to the PI3Kα protein kinase; b shows the pattern binding of 13f to the mTOR protein kinase).

δ 8.85 (d, $J$ = 2.6 Hz, 1H), 8.46–8.39 (m, 1H), 8.15–8.08 (m, 2H), 7.66–7.62 (m, 1H), 7.40 (t, $J$ = 8.9 Hz, 2H), 7.14 (d, $J$ = 2.6 Hz, 1H), 4.13 (s, 4H), 3.91 (d, $J$ = 4.5 Hz, 4H).

**4-(2-(5-(4-Chlorophenyl)-1*H*-pyrazol-1-yl)thieno[2,3-*d*] pyrimidin-4-yl)morpholine (7d).** A yellow solid, mp 280–282 °C; ESI-MS *m/z*: [M + H]⁺: 397.8; ¹H-NMR (400 MHz, DMSO) δ 8.63 (d, $J$ = 2.6 Hz, 1H), 8.19 (d, $J$ = 5.5 Hz, 1H), 7.86 (d, $J$ = 8.5 Hz, 2H), 7.41 (dd, $J$ = 6.9, 4.8 Hz, 3H), 6.95 (d, $J$ = 2.7 Hz, 1H), 3.95–3.84 (m, 4H), 3.72–3.62 (m, 4H).

### 5.5. General procedure for the preparation of target compounds 10a–l

Compounds **10a–l** were synthesized according to the reported procedures by our research group.[13]

### 5.6. General procedure for the preparation of target compounds 13a–l

**10a–l** (chalcone) (0.001 mol) and **12** ($N^1$-(2-mercaptothieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine) (0.001 mol) dissolved in 20 mL of acetic acid, concentrated $H_2SO_4$ is the catalyst. The reaction solution was heated on a 100–110 °C oil bath, and the reaction was confirmed by TLC, and excess solvent was removed under reduced pressure. The precipitated solid is suction filtered, washed with ethanol, dried, and recrystallize with a suitable solvent mixture. The target compound **13a–1** is obtained.

**$N^1$-(2-(3-(3,4-Dichlorophenyl)-5-(4-methoxyphenyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethyl-ethane-1,2-diamine (13a).** A white solid, mp 232–233 °C; ESI-MS *m/z*: [M + H]⁺: 555.2; ¹H-NMR (400 MHz, DMSO) δ 8.14 (d, $J$ = 5.5 Hz, 1H), 7.76 (d, $J$ = 8.5 Hz, 2H), 7.68–7.54 (m, 2H), 7.33–7.18 (m, 2H), 7.07 (d, $J$ = 8.7 Hz, 2H), 6.02–5.74 (m, 1H), 4.03 (d, $J$ = 4.5 Hz, 2H), 3.89 (s, 1H), 3.86 (s, 3H), 3.62 (s, 1H), 3.46 (s, 3H), 3.28–3.17 (m, 2H), 2.84 (s, 6H).

**$N^1$-(2-(5-(4-Bromophenyl)-3-(3,4-dichlorophenyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethyl-ethane-1,2-diamine (13b).** A white solid, mp 291–294 °C; ESI-MS *m/z*: [M + H]⁺: 604.3; ¹H-NMR (400 MHz, DMSO) δ 8.05 (d, $J$ = 5.4 Hz, 1H), 7.69 (d, $J$ = 8.6 Hz, 2H), 7.64 (d, $J$ = 8.5 Hz, 2H), 7.55 (d, $J$ = 8.2 Hz, 1H), 7.51 (s, 1H), 7.25–7.08 (m, 2H), 5.79 (dd, $J$ = 12.0, 5.0 Hz, 1H), 3.86 (dd, $J$ = 17.7, 12.3 Hz, 2H), 3.63 (s, 2H), 3.48 (s, 1H), 3.15 (dd, $J$ = 17.5, 5.4 Hz, 2H), 2.57 (d, $J$ = 2.6 Hz, 2H), 2.37 (s, 6H).

**$N^1$-(2-(3,5-Bis(4-fluorophenyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-tri-methylethane-1,2-diamine (13c).** A white solid, mp 256–257 °C; ESI-MS *m/z*: [M + H]⁺: 492.5; ¹H-NMR (400 MHz, DMSO) δ 8.03 (d, $J$ = 5.5 Hz, 1H), 7.84–7.75 (m, 2H), 7.48 (d, $J$ = 8.2 Hz, 2H), 7.28 (t, $J$ = 8.6 Hz, 2H), 7.18 (t, $J$ = 7.7 Hz, 3H), 5.75 (dd, $J$ = 12.3, 4.4 Hz, 1H), 3.86 (dd, $J$ = 17.6, 12.1 Hz, 1H), 3.73 (d, $J$ = 7.0 Hz, 1H), 3.55 (d, $J$ = 13.6 Hz, 1H), 3.41 (s, 2H), 3.17 (s, 1H), 3.08 (dd, $J$ = 17.5, 4.7 Hz, 1H), 2.36 (s, 2H), 2.20 (s, 6H).

**$N^1$-(2-(3-(3,4-Dichlorophenyl)-5-(4-fluorophenyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine (13d).** A white solid, mp 244–247 °C; ESI-MS *m/z*: [M + H]⁺: 543.4; ¹H-NMR (400 MHz, DMSO) δ 8.06 (d, $J$ = 5.5 Hz, 1H), 7.83–7.77 (m, 2H), 7.58–7.51 (m, 2H), 7.29 (t, $J$ = 8.8 Hz, 2H), 7.19

(dd, $J$ = 14.5, 6.9 Hz, 2H), 5.80 (dd, $J$ = 12.4, 5.0 Hz, 1H), 3.88 (dd, $J$ = 17.5, 12.4 Hz, 2H), 3.46 (s, 2H), 3.18 (dd, $J$ = 17.7, 5.1 Hz, 3H), 2.83 (s, 2H), 2.56 (s, 6H).

**$N^1$-(2-(3-(3,4-Dichlorophenyl)-3-phenyl-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidine-4-yl)-$N^1$, $N^2$,$N^2$-trimethylethane-1,2-diamine (13e).** A white solid, mp 262–263 °C; ESI-MS *m/z*: [M + H]⁺; 525.5; ¹H-NMR (400 MHz, DMSO) δ 8.05 (d, $J$ = 5.4 Hz, 1H), 7.77 (d, $J$ = 7.6 Hz, 2H), 7.62–7.50 (m, 2H), 7.48–7.37 (m, 3H), 7.20 (dd, $J$ = 20.5, 6.8 Hz, 2H), 5.77 (dd, $J$ = 12.2, 5.1 Hz, 1H), 3.88 (dd, $J$ = 17.7, 12.4 Hz, 1H), 3.79–3.70 (m, 1H), 3.53–3.43 (m, 1H), 3.26 (s, 3H), 3.14 (dd, $J$ = 17.7, 5.0 Hz, 1H), 2.20 (s, 2H), 2.11 (s, 6H).

**$N^1$-(2-(5-(4-Methoxyphenyl)-3-phenyl-4,5-dihydro-1*H*-pyrazol-4-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine (13f).** A white solid, mp 287–289 °C; ESI-MS *m/z*: [M + H]⁺; 486.6; ¹H-NMR (400 MHz, DMSO) δ 8.04 (d, $J$ = 5.4 Hz, 1H), 7.70 (d, $J$ = 8.3 Hz, 2H), 7.30 (t, $J$ = 7.7 Hz, 2H), 7.26–7.19 (m, 3H), 7.16 (d, $J$ = 5.6 Hz, 1H), 7.00 (d, $J$ = 8.4 Hz, 2H), 5.82–5.71 (m, 1H), 3.92–3.84 (m, 2H), 3.80 (s, 3H), 3.46 (s, 2H), 3.17 (s, 1H), 3.09 (d, $J$ = 4.1 Hz, 1H), 3.05 (d, $J$ = 4.2 Hz, 1H), 2.85 (s, 2H), 2.56 (s, 6H).

**$N^1$-(2-(3-(3,4-Dichlorophenyl)-5-(*p*-tolyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine (13g).** A white solid, mp 269–270 °C; ESI-MS *m/z*: [M + H]⁺: 539.5; ¹H-NMR (400 MHz, DMSO) δ 8.02 (d, $J$ = 5.4 Hz, 1H), 7.65 (d, $J$ = 8.0 Hz, 2H), 7.54 (d, $J$ = 8.3 Hz, 1H), 7.49 (s, 1H), 7.23 (dd, $J$ = 14.9, 6.7 Hz, 3H), 7.16 (d, $J$ = 8.3 Hz, 1H), 5.73 (dd, $J$ = 12.1, 5.2 Hz, 1H), 3.84 (dd, $J$ = 17.6, 12.2 Hz, 1H), 3.76–3.68 (m, 1H), 3.49–3.39 (m, 2H), 3.24 (s, 3H), 2.33 (s, 3H), 2.17 (s, 2H), 2.08 (s, 6H).

**$N^1$-(2-(3-(4-Bromophenyl)-5-(*p*-tolyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine (13h).** A white solid, mp 272–273 °C; ESI-MS *m/z*: [M + H]⁺:549.5; ¹H-NMR (400 MHz, DMSO) δ 8.08 (d, $J$ = 5.5 Hz, 1H), 7.65 (d, $J$ = 8.0 Hz, 2H), 7.50 (d, $J$ = 8.3 Hz, 2H), 7.24 (dd, $J$ = 17.4, 8.2 Hz, 4H), 7.17 (d, $J$ = 5.4 Hz, 1H), 5.83 (dd, $J$ = 12.0, 4.5 Hz, 1H), 3.86 (dd, $J$ = 17.7, 12.0 Hz, 2H), 3.40 (s, 3H), 3.10 (dd, $J$ = 17.5, 4.4 Hz, 2H), 2.89 (s, 2H), 2.74 (d, $J$ = 7.9 Hz, 6H), 2.34 (s, 3H).

**$N^1$-(2-(3,5-Diphenyl-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*] pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine (13i).** A white solid, mp 225–227 °C; ESI-MS *m/z*: [M + H]⁺: 456.6; ¹H-NMR (400 MHz, DMSO) δ 8.14 (d, $J$ = 5.5 Hz, 1H), 7.86 (d, $J$ = 6.9 Hz, 2H), 7.53 (dd, $J$ = 15.7, 8.1 Hz, 3H), 7.39 (t, $J$ = 7.5 Hz, 2H), 7.30 (dd, $J$ = 14.4, 6.4 Hz, 4H), 5.90 (dd, $J$ = 11.9, 4.4 Hz, 1H), 3.98 (dd, $J$ = 17.6, 12.1 Hz, 2H), 3.41 (s, 3H), 3.18 (dd, $J$ = 17.5, 4.6 Hz, 2H), 2.88 (s, 2H), 2.60 (s, 6H).

**$N^1$-(2-(3-(4-Fluorophenyl)-5-(4-methoxyphenyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine (13j).** A white solid, mp 286–287 °C; ESI-MS *m/z*: [M + H]⁺: 504.2; ¹H-NMR (400 MHz, DMSO) δ 8.08 (d, $J$ = 5.2 Hz, 1H), 7.71 (d, $J$ = 8.9 Hz, 2H), 7.21–7.09 (m, 4H), 7.01 (d, $J$ = 8.8 Hz, 3H), 5.84 (d, $J$ = 7.0 Hz, 1H), 3.91 (d, $J$ = 17.0 Hz, 2H), 3.81 (s, 3H), 3.73 (s, 2H), 3.17 (s, 1H), 3.10 (d, $J$ = 13.5 Hz, 3H), 2.85 (s, 1H), 2.74 (s, 6H).

**$N^1$-(2-(3-(4-Bromophenyl)-5-(4-fluorophenyl)-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-**

**diamine (13k).** A white solid, mp 291–292 °C; ESI-MS *m/z*: [M + H]$^+$: 552.1; $^1$H NMR (400 MHz, DMSO) $\delta$ 8.03 (d, $J$ = 5.5 Hz, 1H), 7.84–7.75 (m, 2H), 7.48 (d, $J$ = 8.2 Hz, 2H), 7.28 (t, $J$ = 8.6 Hz, 2H), 7.18 (t, $J$ = 7.7 Hz, 3H), 5.75 (dd, $J$ = 12.3, 4.4 Hz, 1H), 3.86 (dd, $J$ = 17.6, 12.1 Hz, 1H), 3.73 (d, $J$ = 7.0 Hz, 1H), 3.55 (d, $J$ = 13.6 Hz, 1H), 3.41 (s, 2H), 3.17 (s, 1H), 3.08 (dd, $J$ = 17.5, 4.7 Hz, 1H), 2.36 (s, 2H), 2.20 (s, 6H).

$N^1$-(2-(3-(4-Bromophenyl)-5-phenyl-4,5-dihydro-1*H*-pyrazol-1-yl)thieno[2,3-*d*]pyrimidin-4-yl)-$N^1$,$N^2$,$N^2$-trimethylethane-1,2-diamine (13l). A white solid, mp 261–262 °C; ESI-MS *m/z*: [M + H]$^+$: 534.1; $^1$H-NMR (400 MHz, DMSO) $\delta$ 8.06 (d, $J$ = 5.5 Hz, 1H), 7.75 (d, $J$ = 6.9 Hz, 2H), 7.49 (d, $J$ = 8.5 Hz, 2H), 7.47–7.38 (m, 3H), 7.18 (dd, $J$ = 13.6, 6.9 Hz, 3H), 5.79 (dd, $J$ = 11.9, 4.7 Hz, 1H), 3.88 (dd, $J$ = 17.5, 12.1 Hz, 2H), 3.47 (d, $J$ = 6.8 Hz, 2H), 3.20–3.04 (m, 3H), 2.93 (s, 2H), 2.63 (s, 6H).

### 5.7. Cytotoxicity assay *in vitro*

The target compounds (**4a–d**, **7a–d** and **13a–l**) were evaluated antiproliferative activity against A549 (human lung cancer cell), HepG2 (human hepatocellular carcinoma cell), MCF-7 (human breast cancer) by the standard MTT assay, with GDC-0941 as positive control (all cells were obtained from the Shanghai Institute of Biological Sciences, Chinese Academy of Sciences, Shanghai, China). The experiment was conducted according to the method reported by our research group.[16]

### 5.8. PI3Kα and mTOR kinase assay

Four selected compounds (**13c**, **13f**, **13g** and **13h**) are tested for their activity against PI3Kα and mTOR using a Kinase-Glo® Luminescent Kinase Assay, with GDC-0941 as positive control. The experiment was conducted according to the method reported by our research group.[17]

### 5.9. Acridine orange single staining

The cancer cell apoptotic of target compound **13f** were evaluated with A549 cancer cell lines by acridine orange single staining. The experiment was conducted according to the method reported by our research group.[18]

### 5.10. Cell apoptosis assay by flow cytometry

A549 cells were seeded in a 6-well plate at $1 \times 10^5$ cells per well and incubated for 24 h. Then treated with compound **13f** for 24 h. Cells were harvested and fixed with ice-cold 70% ethanol at 4 °C for 12 h. Ethanol was removed and the cells were washed with cold PBS. Then cells were incubated in 0.5 mL of PBS containing 1 mg mL$^{-1}$ RNase for 30 min at 37 °C. Then cells were stained with propidium iodide (PI) in the dark for 30 min. The DNA contents was then measured by flow cytometer.

### 5.11. Docking studies

Molecular docking simulation studies were carried out by the AutoDock4.2 software (The Scripps Research Institute, USA). The docking tutorial we used and the detailed AutoDock basic operational methods can be found at: http://autodock.scripps.edu/faqs-help/tutorial. The protein preparation process of flexible docking mainly includes fixing the exact residues, adding hydrogen atoms, removing irrelevant water molecules, adding charges, *etc.* The potent compounds were selected as ligand examples, and the structures of PI3Kα (PDB ID code: 3BDS) and mTOR (PDB ID code: 4JSV) were selected as the docking models. Only the best-scoring ligand–protein complexes were used for the binding site analyses. All the docking results were processed and modified in Open-Source PyMOL 1.8.x software (https://pymol.org).

## Conflicts of interest

There are no conflicts to declare.

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