ORIGINAL ARTICLE



Profiling of secondary metabolites and DNA typing of three different *Annona* cultivars grown in Egypt

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Abstract

Objectives Natural products are often efficacious and safe alternatives to synthetic drugs. This study explored secondary leaves and bark metabolites profiles in extracts of a new Egyptian hybrid, *Annona cherimola* × *Annona squamosa*, known as Abdel Razek. This hybrid exhibited 100% similarity with *A. cherimola* as evidenced by random amplified polymorphic DNA (RAPD) and inter-simple sequence repeat (ISSR) analyses.

Methods Primary constituents in methanol extracts of different plant organs were identified. Extracts richest in alkaloids and polyphenolics were assessed for in vitro antioxidant activity and the most potent were further studied in vivo for treating gastric ulcer in rats. The latter activity was assessed histopathologically.

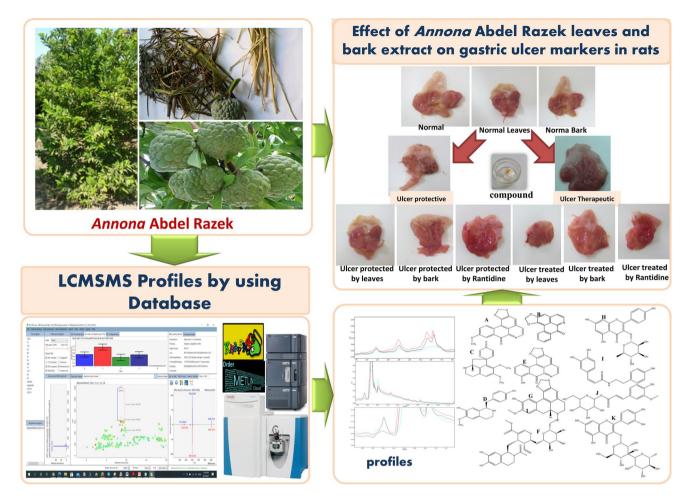
Results Structural analysis with HPLC/ESI-MSⁿ, and UPLC/HESI-MS/MS identified 63 metabolites, including seven amino acids, 20 alkaloids, 16 flavonoids, eight phenolics and other compounds. Severe stomach alteration was observed after ethanol induction in rats. Ulcer score, oxidative stress biomarkers, cell organelles biomarker enzymes, and gastrointestinal histological features improved to variable degrees after treatment with *Annona* Abdel Razek hybrid leaves and bark methanol extracts. **Conclusion** Extracts of *Annona* Abdel Razek had showed in vitro antioxidant effect and may be promising for the treatment of gastric ulcers.

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Graphical abstract



Keywords Annona Abdel Razek · Gastric ulcer, antioxidants · Enzymes · Polyphenolics · Alkaloids

Abbreviations

UPLC	Ultra performance liquid chromatography
LCMSMS	Liquid chromatography-mass spectrometry
HPLC	High performance liquid chromatography
HESI	Heated Electrospray Ionization
ESI	Electrospray ionization
CID	Collision-induced dissociation
EI	Electron ionization
ABTS ⁺	2,2'-azino-bis(3-ethylbenzothiazoline-6-sul-
	fonic acid)
DPPH	2,2-diphenyl-1-picrylhydrazyl

1 Introduction

Gastric ulcers affect a large percentage of world's population and are caused by a variety of conditions, including stress, smoking, alcohol intake, nutritional deficiencies, *H*. *pylori* infection, and NSAIDs (Martel et al., 2020). A metabolic imbalance in stomach mucosal defenses is involved in the pathogenesis of ulceration. Acid, pepsin, and *H. pylori* are common stressors, and mucus, prostaglandins, bicarbonate, nitric oxide, and growth factors are protective elements (Graham, 2014). Damage to the gastrointestinal mucosa caused by alcohol involves microvascular injury, rupture of the vascular endothelium, increased vascular permeability, edema, and epithelial elevation (Penninger et al., 2020). Treatment of peptic ulcers first focuses on lowering stomach acid production, and then on restoring gastric mucosal protection. Proton pump inhibitors, histamine receptor blockers, mucosal barrier medicines, and prostaglandin analogues are all used in modern treatment (Graham, 2014).

Several species of *Annona* (Family: Annonaceae) are used in traditional medicine, reflecting their anti-anxiety, antidepressant, anticonvulsant, and tranquilizing properties (Martínez-Vázquez et al., 2012). Annona is the second largest genus in the family, containing approximately 166 species (Nagy & Wardowski, 1990). A. squamosa and A. cherimola and their hybrid (Abdel Razek) rich produce substantial amounts of secondary metabolites, including phenolics, flavonoids, alkaloid isoquinolines, peptides, acetogenins, lectins, and volatile oils (Dahiya & Dahiya, 2021). Acetogenins are most commonly isolated from A. muricata, A. squamosa, A. cherimola and A. classiflora (Prabhakaran et al., 2016). This situation prompted scientists to look for novel anti-ulcer medications in natural materials due to the fact that the side-effects of the natural compounds are comparatively lower than those of the synthetic ones. Therefore, many Annona species exhibit anticancer, antioxidant, antibacterial, and antifungal properties (Leiteet al., 2020). Such properties are the basis for substantial research on natural products (Huang et al., 2019), and the present study explored secondary metabolites in leaves and bark from the new hybrid Annona Abdel Razek for their antioxidant and anti-ulcerative effects in rats.

2 Material and methods

2.1 Phytochemical study

2.1.1 Chemicals

Analytical grade solvents and chemicals used in this study were purchased from Fisher Scientific Co. (Bishop Meadow Road, Loughborough, UK).

2.1.2 Plant products and DNA fingerprinting

Tissues from three Annona cultivars, A. squamosa, A. cherimola, and Abdel Razek, were collected from a private farm in Mansoriya district, Giza, Egypt. Plants were identified by Prof. Dr Mona Marzouk, Plant and Systematic Department at the National Research Centre, Egypt and voucher specimens numbers 521, 522, and 523 were deposited in the herbarium. Presence/absence of RAPD and ISSR markers were scored using DNA fragments of gels (Tables 1, S1–S3). Only five primers were successful in producing repeatable polymorphic DNA products. A UV transilluminator was used to record RAPD and ISSR patterns (Pharmawati et al., 2004).

2.1.3 Successive extraction and evaluation of total polyphenolics, flavonoids, and alkaloids A. Abdel **Razek plant tissues**

Five hundred grams from different parts (leaves, barks, fruits, and seeds) were extracted by soaking in 10 L MeOH (100, 80, and 50%) for 48 h at room temperature. Using a rotary evaporator, the mixed alcoholic extracts were concentrated under decreased pressure at 40 °C. The complete extract of each part was suspended in 0.5 L water for 24 h, then partitioned with 0.5 L of different solvent (Me₂Cl₂, EtOAc), and BuOH before being filtered through Whatman No. 54. Total flavonoids and polyphenols were assessed according to (Hamed et al., 2019), whereas total alkaloids were determined according to (El-Gengaihi et al., 2020).

2.1.4 Profiling of secondary metabolites using LC-MS/MS

Two hundred mg of dried leaves and bark were separately sonicated with two ml methanol for 5 min at 37 °C. The filtered extract was injected in negative and positive ion mode into an ion-trap Esquire 3000 mass spectrometer with an ESI ion source. Analysis conditions for nebulizer, dry gas flow, and dry gas temperature were 30 psi, 9.0 L/min, and 310 °C, respectively. Skimmer 1 was set to a negative voltage of - 10.0 V. Scan resolution was 13,000 m/z/sec with a range of 150-3000 m/z. An Acquity system (Waters, Milford, USA) was combined with a Q-Exactive hybrid MS/MS quadrupole-Orbitrap mass spectrometer (Thermo, Bremen, Germany). A BEH shield C18 column (1502.1 mm, 1.7 m) was used for chromatographic separation with deionized water acidified with 0.1 percent formic acid (solvent A) and acetonitrile (solvent B). The mobile phase gradient was 0-14 min from 5% B to 50% B, then 14-20 min to 98% B, and sustained for 5 min. The system was reset to initial settings and re-equilibrated for 3 min between runs. Q-Exactive MS conditions were HESI ion source voltage of -3 kV or

Table 1 RAPD and ISSR primers and nucleotide sequences		Nucleotide s Analysis	sequences for RAPD-PCR	Nucleotid	e sequences for ISSR procedure
1	No	Name	Sequence	Name	Sequence
	1	OP- A01	5' GAA AGG GGT G 3'	14A	5' CTC TCT CTC TCT CTC TTG3'
	2	OP-B02	5' TCG GGG ATA G 3'	44B	5' CTC TCT CTC TCT CTC TTG3'
	3	OP-B07	5' GAA AGG GGT G 3'	HB-08	5' GAG AGA GAG AGA GG3'
	4	OP-B11	5' GTA GAC CCG T 3'	HB-11	5' GTG TGT GTG TGT TGT CC 3'
	5	OP-C04	5' CCG CAT CTA C 3'	HB-15	5' GTG GTG GTG GC 3'

3 kV, gas (N2) flow rate of 48 L/min, and auxiliary gas flow rate of 13 L/min. Ion source capillary and auxiliary gas heater temperatures were 250 °C and are 380 °C, respectively. Collisions power for CID MS/MS studies was 15 eV (Mohammed et al., 2021).

2.1.5 Data processing

MSMS fragmentation patterns were generated for 63 phenolics, alkaloids, terpenes, and other metabolite using ion mobility tandem mass spectrometry to allow identification of extract composition. These four fractions from raw MS data were output to analysis base file (abf) format and packaged in MS-DIAL 4.60 (http://prime.psc.riken.jp/). MSP format libraries were used with enhanced standardized untargeted lipidomic and metabolomic patterns to filter noisy spectra via classical spectral similarity calculations. Selected metabolites were identified by comparing fragmentation patterns and retention indices with similar information from KNApSAcK, Metlin, Golm, MassBank, Fiehn Bin Base, LipidBlast, ReSpect, and RIKEN databases (Ammar et al., 2021; Mohammed et al., 2021).

2.2 Pharmacological Investigation

2.2.1 In vitro antioxidant analysis

The ABTS⁺ was determined in various extracts using the methods provided by (Dinkova-Kostova et al., 2007)). The DPPH (1,1-diphenyl-2-picrylhydrazyl) free radicals scavenging test was carried out according to Ammar et al. (2021). The Percent of inhibition of DPPH and ABTS⁺ radicals was estimated using the following formula:

Percent of inhibition = $[(A_{control} - A_{sample})/A_{control}] \times 100$,

where A is the absorbance of DPPH⁻ and ABTS⁺ at 517 and 734 nm, respectively.

2.2.2 Acute toxicity

Forty-eight rats, weighing 100–120 g were separated into two groups of 24 animals. Each group was subdivided into three subgroups of eight. Each animal received an oral dose of 250, 500, or 1000 mg/kg body weight of Abdel Razek leaves or bark alcoholic extract. Numbers of dead animals were counted over 15 days. The mortality rate and LC50 were also recorded. Extracts were safe at all tested concentrations, and a dose of 500 mg/kg body weight was chosen for further investigation (El-Gengaihi et al., 2013).

2.2.3 Doses and administration routes

Rats were administered 0.5 ml/100 g body weight absolute ethanol orally after a 24-h fast (Fahmi et al., 2019). Abdel Razek extracts were administered orally for 1 week after ulcer development in therapeutic model and 1 week before ulcer development as protective model (Mohammeda et al., 2020). Ranitidine given orally (100 mg/kg bw/day for one week) as a reference drugs for antiulcer effects (Fahmi et al., 2019).

2.2.4 Experimental groups

Seventy-two male albino Wistar rats were divided equally into nine groups.

Group 1 rats served as controls.

Groups 2and 3 animals received alcoholic extracts of bark or leaves daily for 1 week.

Group 4 rats received ethanol dose on an empty stomach for 24 h. Animals were euthanized one hour later to provide a baseline for ulcer production.

Groups (protective model) 5 and 6 rats received plant extracts or ranitidine daily for 7 days before receiving one oral dose of absolute ethanol on an empty stomach for 24 h and sacrificed 1 h later.

Group 7 animals received ethanol and then euthanized after seven days.

Groups (therapeutic model) 8 and 9 rats received one oral dose of absolute ethanol and then plant extracts or ranitidine daily for 7 days and sacrificed 24 h after the last administration.

2.2.5 Samples preparations

Stomach tissue was homogenized in a normal physiological saline solution (0.9% NaCl) (1:5 w/v). The homogenate was centrifuged for 15 min at 300 g at 4 °C, and the supernatant kept at -80 °C until further analysis (Khalaf-Allah et al., 2016).

2.2.6 Ulcer index

2.2.6.1 Stomach lesion counts Stomachs were removed, opened from the long curvature, cleaned with saline solution, expanded, and placed on the dissection plate. Numbers of lesions were counted using a magnifying glass (Guedes et al., 2008; Mohammeda et al., 2020).

2.2.6.2 Stomach total acidity Gastric contents were collected and centrifuged for 15 min at 300 g. Supernatant vol-

umes (ml) were measured and total acidity determined by titration with 0.1 N NaOH using 2% phenolphthalein as an indicator. The results are expressed as milliequivalents per liter (mEq/L), where:

mEq = Vol of NaOH × normality of NaOH × equivalent weight of HCl × 1000/Sample volume

2.2.7 Oxidative stress and cell organelles markers

Glutathione (GSH) (Bhardwaj et al., 2000), malondialdehyde (MDA) (Buege & Aust, 1978), superoxide dismutase (SOD) (Nishikimi et al., 1972), nitric oxide (NO) (Montgomery & Dymock 1961), catalase (CAT) (Shelton & Rice 1957), succinate dehydrogenase (SDH) (Swanson, 1955), lactate dehydrogenase (LDH) (Wattiaux & De Duve, 1956), glucose-6 phosphatase (G-6-Pase) (Bodansky & Schwartz, 1963), acid phosphatase (AP) (Bodansky & Schwartz, 1963), alkaline phosphatase (ALP), 5'-nucleotidase (5'NT) and total protein (Bradford, 1976) were measured in stomach tissue.

2.2.8 Histopathological analysis

Slices of stomach tissues were embedded in paraffin blocks after fixation in 10% formalin. Masson's trichrome and hematoxylin and eosin (H&E) were used to stain 5 μ m thick sections. Sections were examined under a light microscope for pathological abnormalities (Bancroft, 1996).

2.2.9 Statistics

All biochemical data are presented as means \pm SD of eight rats per group. Statistical analysis used one-way analysis of

variance (ANOVA), with Costat Software, followed by least significance difference (LSD) tests between groups; p < 0.05 was used as a threshold for significance.

3 Results

3.1 DNA Fingerprinting

Primer OP-A01 showed eight bands of 270–1200 bp showing four monomorphic and four polymorphic bands indicating 50% polymorphism (Tables S3 and figure S1 A (supplementary data)). OP-B02 showed 11 bands of 260–1390 bp. Two monomorphic and nine polymorphic bands were noted, suggesting 81.82% polymorphism. Similar results for OP-B07, OP-B11, and OP-C04 were 11 bands of 155 to 1350 bp, 11 bands of 225 to 1455 bp, and seven bands of 180 to 680 bp, respectively. These primers displayed nine monomorphic and two polymorphic bands indicating 18.18% polymorphism, three monomorphic and eight polymorphic bands suggesting 72.73% polymorphism, and three monomorphic and four polymorphic bands indicating 57.14% polymorphism, respectively.

Likewise, primers A, 44B, HB-08, HB-11, and HB-15 produced six bands of 295–820 bp, eight bands of 280–640 bp, nine bands of 145–825 bp, 12 bands of 295–1040 bp, and eight bands of 340–1070 bp, respectively. These primers exhibited four monomorphic and two polymorphic bands suggesting 33.33% polymorphism, six monomorphic and two polymorphic bands indicating 25% polymorphism, no monomorphic and nine polymorphic bands suggesting 100% polymorphism, four monomorphic and eight polymorphic bands indicating 66.67% polymorphism, and five monomorphic and three polymorphic bands suggesting 37.5% polymorphism, respectively (figure *S1 B*).

Table 2Total polyphenolic,flavonoid and alkaloid contents	Annona Abdel Razek parts	Chloroform		Ethyl Acetate	Butanol	Total alcohol
	Total phenolics	(mg/g Dw)				
	Leaves	1.3 ± 0.02	15.65 ± 0.36		18.36 ± 0.21	20.36 ± 0.32
	Bark	0.9 ± 0.10	13.36 ± 0.42		15.36 ± 0.25	12.87 ± 0.46
	Fruit	0.2 ± 0.01	8.96 ± 0.20		6.32 ± 0.42	5.65 ± 0.85
	Seed	0.12 ± 0.01	6.65 ± 0.21		8.63 ± 0.52	7.89 ± 0.45
	Total flavonoids	(mg/g Dw)				
	Leaves	0.07 ± 0.01	1.02 ± 0.02		3.2 ± 0.03	4.2 ± 0.01
	Bark	0.04 ± 0.00	0.98 ± 0.21		1.86 ± 0.21	0.98 ± 0.02
	Fruit	0.02 ± 0.00	0.78 ± 0.10		2.32 ± 0.02	1.9 ± 0.01
	Seed	0.02 ± 0.00	0.45 ± 0.02		0.87 ± 0.14	0.6 ± 0.02
	Total alkaloids (mg/g Dw)				
	Leaves	0.01 ± 0.002	0.001 ± 0.005		0.009 ± 0.002	0.05 ± 0.005
	Bark	0.02 ± 0.001	0.001 ± 0.004		0.02 ± 0.001	0.08 ± 0.004
	Fruit	0.01 ± 0.006	0.001 ± 0.001		0.01 ± 0.001	0.05 ± 0.003

3.2 Phytochemical analysis

3.2.1 Total polyphenolic, flavonoid and alkaloid contents

Bark, fruit, and seed extracts showed lower total phenol and flavonoid content than leaves extracts (Table 2). The highest total polyphenolic and flavonoid concentrations were found in butanol and total alcohol extracts of Abdel Razek (18.36, 20.36, 3.2, and 4.2 mg/g, respectively). Further, total alkaloids extracted by different solvents were limited, and no significant differences were observed among plant parts extracts. The total alkaloid content was greater in bark than in seeds as 0.08 vs. 0.06 mg/g DW, respectively.

3.3 Metabolomics analysis

Metabolomic studies established profiles of 63 compounds, consisting of phenols, flavonoids, coumarins, amino acids, and alkaloids, from leaves and bark. Compounds were identified by comparing their molecular mass (<5 ppm), mass spectroscopy, and retention times to standard compounds, data from the ChEBI, PubChem, Metlin, and KNApSAck databases, and results from earlier research. Several compounds co-eluted from the LC column, and the mass spectrometer was unable to fragment these compounds independently. This limitation was addressed by: (i) extending the separation period of HPLC-MSⁿ runs to 1 h and (ii) using a C18 reversed-phase UPLC-HESI-MSMS to analyze fractions derived from polyamide preparative LC runs. Secondary metabolites included alkaloids, phenolic acids, flavonoids and polyphenols, such as coumaric acid, coumaroyl quinic acid, kaempferol, feruloyl hexoside, feruloyl-quinic acid, caffeoyl quinic acid, caffeic acid, caffeic acid hexoside, epicatechin, catechin, alkaloids (boldine, caranine, oxoanolobine, reticuline, ipecoside, coclaurine), and amino acid derivatives (n-fructosyl isoleucine, n-fructosyl pyroglutamate, and tryptophan) (Table 3 and Figs. 1, 2, & S3).

3.4 Biochemical investigations

3.4.1 In vitro antioxidant effect

Evaluation of the in vitro antioxidant activity of the alcoholic extracts from leaves, bark, fruits and seeds of Abdel Razek hybrid, as well as the total alkaloids, flavonoids and polyphenolics fractions are presented in (Table 4). The leaves, fruits and bark of this species were examined against vitamin C and trolox as standards. The results revealed that both leaves and barks of hybrid Abdel Razek extract induced 16.66 and 13.33, 22.85 and 9.22% of inhibition by using DPPH⁻ and ABTS⁺ as free radicals scavenging agents, respectively (Table 4).

3.4.2 In vivo protective effect of Annona Abdel Razek

3.4.2.1 Ulcer index Regarding the pH level in control rats protected with the plant extracts, the results revealed insignificant changes (Table S4 and Fig. 3a). Gastroulcerative rats recorded significant decrease by 40.90% compared with the control group. Gastroulcerative rats protected with the leaves and bark of Annona Abdel Razek, and ranitidine drug showed significant increase in the pH level by 61.53, 61.53, and 72.92%, respectively, compared with the ulcer group. The gastric volume in control rats protected with the plant extracts and ranitidine drug recorded insignificant changes indicating extract safety (Table S4 and Fig. 3a). The ulcerative stomach showed significant increase in its volume content by 2260.08% compared with the control group. Protected rats with leaves and bark of Annona Abdel Razek and ranitidine drug recorded significant decrease in gastric volume by 64.40, 57.40 and 34.30%, respectively in comparison with the ulcer group. Regarding to the gastric total acidity, normal rats protected with the plant extracts showed insignificant changes as compared with the control group (Table S4 and Fig. 3a). The gastric ulcerative rats showed significant decrease in the total acidity by 76.65% versus the control group. Protective rats with leaves and bark of the plant extract and ranitidine drug proved significant increase in the total acidity level by 194.00, 229.00 and 171.64%, respectively as compared with the ulcerative group. By examining the lesions counts in the stomach of ulcerative group, it showed eleven ulcer lesions per stomach (Table S4 and Fig. 3a). In protected rats with leaves and bark of the plant extract and ranitidine drug, the gastric lesions count decreased by 73.33, 84.44 and 70.84%, respectively as compared with the ulcerative group.

3.4.2.2 Oxidative stress The oxidative stress markers, protective rats with extracts of Abdel Razek and ranitidine drug produced did not significantly alter catalase, NO, SOD, glutathione, malondialdehyde and total protein levels as compared with control group (Table S5 and Fig. 3b). Significant increases in these indices were recorded in gastroulcerative rats amounted to 141.22, 280.00, 64.10 and 55.36%, respectively as compared with control group. Protection of gastro-ulcerative rats with leaves and bark Abdel Razek, and ranitidine drug recorded significant decrease in catalase activity by 50.11, 68.93, and 71.70%, respectively, while, nitric oxide significantly decreased by 56.84, 75.78, and 63.15%, respectively. Similarly, superoxide dismutase significantly decreased by 86.45, 82.06, and 80.72%. In the same manner, glutathione level decreased by 70.63, 70.47, and 71.56%, respectively. Malondialdehyde also decreased by 51.56, 46.87, and 39.84%, while the total protein content decreased by 22.56, 16.99, and 30.71%, respectively (Table S5 and Fig. 3b).

No	Rt [Min]	Tentative metabolite identification	Classes	Chemical formula	HRMS of exa	ct mass measure	HRMS of exact mass measured, calculated and fragmentation ions	entation ions	Δ ppm	Fragentation in Low Resolution	n in Low	λ max [nm]	λ max [nm] Referencess
					[H-H] ⁻	Fragmenta- tion ions	[M+H] ⁺	Fragmenta- tion ions		Negative ion mode ESI (-)	Positive ion mode ESI (+)		
mino	acids gro	Amino acids groups and organic acids											
	1.72	2,6-dihydroxy- 4-Methoxytoluene ^{a,b}	Organic com- pound	$C_8H_{10}O_3$	153.0546, 153.0546	123.0438, 109.0282	155.0703, 155.0703	113.9643, 109.1015	0.1376	ļ	I	237, 280	XCMS, MS- Dial
6	2.25	N-Fructosyl isoleucine ^a	Amino acid	$C_{12}H_{23}NO_7$	292.1405, 292	150.0040, 130.0860	294.2052, 294.2100	276.1441, 230.1387, 132.1021	4.866/5.2520	I	I	208, 225	XCMS, MS- Dial
ŝ	2.48	N-Fructosyl pyroglutamate ^{a,b}	Amino acid	$C_{11}H_{17}NO_8$	290.0872, 290.0870	230.0672, 200.0559, 170.0447, 128.0340	I	1	0.7132	1	1	213, 222, 270	MS-Dial
	3.29	Pantothenic acid ^a	vitamin	$C_9H_{17}NO_5$	218.1027, 218.1023	146.0809, 88.0388	I	I	1.9399	ļ	I	I	MS-Dial
9	3.88	Tryptophan ^{a,b}	Amino Acid	$C_{11}H_{12}O_2N_2$	203.0818, 203.0815	142.0652, 116.0489, 97.0283	I	I	1.2432	I	I	215, 239, 258	MS-Dial
27	5.61	Acetylleucine ^{a,b}	Amino Acid	C ₈ H ₁₅ NO ₃	172.0969, 172.0968	130.0860	I	I	0.2497	ļ	I	I	MS-Dial
	5.72	Methoxytyrosine ^{a,b}	Amino Acid	$C_{10}H_{13}NO_4$	210.0792, 210.0743	124.0391, 94.0281	212.0918, 212.0917	$\begin{array}{c} 194.0801,\\ 145.0498,\\ 127.0394,\\ 97.0290\end{array}$	0.7146/0.1506	1	1	203, 218	MS-Dial
35	6.31	AcetylPhenylalanine ^{a,b}	Amino Acid	$C_{11}H_{13}NO_3$	206.0218, 206.0812	164.0706, 147.0439, 91.0538, 72.0076	208.0970, 208.0968	190.1308, 151.0758, 104.0471	0.7931	I	I	237, 280, 306	MS-Dial
kalo	Alkaloids groups	r-											
	3.10	Boldine ^b	Aporphine Akaloids	C ₁₉ H ₂₁ NO ₄	326.1380, 326.1387	178.9966, 164.0706, 147.0441, 101.0230	328.1541, 238.1543	297.1125, 285.1139, 265.0860, 283.060, 237.0909, 178.0865, 121.0651, 58.0660	2.0040/-0.5801	1	328, 283, 249, 175, 121	213, 222, 258	(Wu et al., 1993)
-	3.64	Ipecoside ^b	Alkaloid terpene glycosides	$C_{27}H_{35}O_{12}N$	564.2158, 564.2161	504.3117, 207.1137, 162.0549	566.2220, 566.2245	434.1812, 272.1281, 161.0598, 107.0.98	2.1690	564, 504, 207	566, 434, 273, 175, 272, 161, 107	221	(Harborne, 1999) KNApSAcK
-	3.79	Caranine ^{a,b}	Indolizidine Alkaloids	$C_{16}H_{18}O_{3}N$	270.1140, 270.1144	162.0550	272.1281, 272.1281	255.1016, 161.0598, 123.0445,	0.1384	I	272, 255, 107, 161, 123	250, 350	(Harborne, 1999) KNApSAcK

No	Rt [Min]	Tentative metabolite identification	Classes	Chemical formula	HRMS of exa	ct mass measure	HRMS of exact mass measured, calculated and fragmentation ions	ntation ions	∆ ppm	Fragentation in Low Resolution	ı in Low	λ max [nm]	λ max [nm] Referencess
					-[H-H]-	Fragmenta- tion ions	[M+H] ⁺	Fragmenta- tion ions		Negative ion mode ESI (-)	Positive ion mode ESI (+)		
12	4.20	Fenfangjine G ^b	Isoquinolines, Alkaloid	C ₂₂ H ₂₈ NO ₈	432.1667, 432.1647	270.1138, 162.0549	434.1810, 434.1809	416.1701, 387.2764, 309.1223, 255.1021, 174.0759, 107.0495	0.1302	1	434, 272, 161, 255, 107	206, 234, 250, 350	(Ogino et al., 1998) KNApSAcK
13	4.50	Oxoanolobine ^{a.b}	Oxoapor- phine Alka- loids	$C_{17}H_{20}O_4N$	I	I	302.1387, 302.1392	285.1118, 270.0853, 253.0851, 191.0703, 107.0496	- 1.8151	I	302, 285, 153, 107, 253, 191, 107	234, 310	(Simeon et al., 1989) KNApSAcK
14	4.53	(-)-Phanostenine ^{a.b}	A porphine alkaloid	$C_{19}H_{20}O_4N$	I	I	326.1386, 326.1387	295.0967, 265.0859	- 3.9663	I	326, 297, 218, 151, 158	I	(Rabêlo et al., 2013) KNApSAcK
16	4.82	Tembetarine ^{a.b}	Isoquinolines, Alkaloid	$C_{20} H_{26} O_4 N$	I	I	344.1852, 344.1856	299.1281, 165.0548, 147.0441	- 1.1421	I	344, 299,165, 279, 192, 177	230, 280	(Nishiyama et al., 2004) KNApSAcK
18	5.23	Corytuberine ^{a,b}	A porphine Alkaloid	$C_{19}H_{21}NO_4$	1	I	328.1553, 328.1557	297.1129, 265.0859, 237.0912, 178.064	-1.0290	I	328, 297, 178, 265, 237, 191	203, 277, 300	(Harborne, 1999) KNApSAcK
19	5.28	Laurifoline ^{a.b}	Aporphine Alkaloid	$C_{20}H_{24}NO_4^+$	340.1558, 340.1557	325.1324, 310.1091, 179.0340, 135.0441	342.1697, 342.1700	297.1123, 265.0859, 237.0908, 178.0867, 143.9961	-0.7188	I	342, 297, 256, 245	270	(Saxena, 1979) KNApSAcK
22	5.45	Reticuline ^{a,b}	Benzyliso- quinoline Alkaloids	$C_{19}H_{23}O_4N$	328.0965, 328.0968	191.0553	330.1693, 330.1686	299.1274, 192.1020, 175.0755, 137.0597	2.0276	1	330, 299, 192, 175	220, 280	(Simeon, Rios, Villar 1989) KNApSAcK
59	5.80	Coclaurine ^b	Isoquinoline Alkaloids	C ₁₇ H ₁₈ NO ₃	284.1258, 284.1300	245.0671, 176.0707, 107.0488	286.1440, 286.1438	269.1173, 237.0916, 175.0967, 143.0494, 107.0496	1.9256	I	286, 269, 107, 237, 175, 137, 107	280, 320	(Nishiyama et al., 2006)
30	5.84	Anomoline ^b	Isoquinoline alkaloids	$C_{18}H_{22}O_4N$	314.1398, 314.1400	180.0652, 121.0280	316.1543, 316.1543	299.1279, 267.1009, 121.0651	3.6460/0.2160	I	316, 299, 267, 205, 107		(Harborne, 1999)
33	6.15	Thalifoline ^b	Quinolone Alkaloids	$C_{11}H_{12}O_{3}N$	206.0819, 206.0812	164.0707, 147.0439	208.0970, 208.0968	190.1308, 151.0758, 104.0471	3.6620	206, 164, 147		211, 278	(Chen et al., 1999)
39	7.18	Dehydronuciferine	Isoquinoline Alkaloids	C ₁₉ H ₁₉ NO ₂	I	I	294.1483, 294.1489	265.2290, 249.0909, 147.0441	I	1	I	I	(Liu et al., 2014) metlin

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Tab	Table 3 (continued)	100 million (1997)											
°N N	Rt [Min]	Tentative metabolite identification	Classes	Chemical formula	HRMS of exac	et mass measure	HRMS of exact mass measured, calculated and fragmentation ions	ntation ions	Δ ppm	Fragentation in Low Resolution	in Low	λ max [nm] Referencess	Referencess
					[H-H] ⁻	Fragmenta- tion ions	[M+H] ⁺	Fragmenta- tion ions		Negative ion mode ESI (-)	Positive ion mode ESI (+)		
52	9.80	Liriodenine ^b	Aporphine Alkaloids	$C_{17}H_9\;O_3N$	1	. 1	276.0656, 276.0655	188.0970	0.0668		276,188, 117,75	247, 269, 302	(Chen & Wu, 2001)
53	06.6	Oxostephanosine ^b	Aporphine Alkaloids	$C_{17}H_9O_4N$	I	I	292.0604, 292.0604	237.0477, 177.0917	- 0.1370	I	292, 263, 234, 205, 177, 150	215, 245, 275, 320, 364,448	(Pharadai et al., 1985)
54	10.30	Lanuginosine ^{a,b}	Aporphine Alkaloids	$C_{18} H_{11} O_4 N$			306.0761, 306.0761	291.0513, 263.0513, 222.0036, 179.0766	0.0865		306, 275, 263, 221, 205	246, 271, 315	(Chen et al., 1998)
99	12.47	(+)-Stepharine ^b	Proaporphine Alkaloids	$C_{18}H_{19}O_{3}N$	296.1287, 296.1281	I	298.1440, 298.1438	177.0547, 145.0284	- 1.8519/0.6085	296, 251, 183	298, 177, 145, 117	279	(Chen et al., 1997)
19	12.58	N-Methylcrotsparine ^b	Proaporphine Alkaloids	$C_{18} H_{20} O_3 N$	296.1294, 296.1300	281.1060, 190.0509	298.1436, 298.1438	280.2633, 250.2538, 177.0546, 163.0390, 105.0705	- 2.0297/- 0.7222	1	298, 269, 192, 161, 107	290, 350	(Simeon et al., 1989) KNApSAcK
62	12.70	Crabbine ^b	Aporphine Alkaloids	$\mathrm{C}_{20}\mathrm{H}_{24}\mathrm{O}_{5}\mathrm{N}$	ł	I	358.1654, 358.1642	313.1433, 281.1169, 135.0807	- 3.6119	I	358, 313, 281, 190, 135	212, 280	(Yang et al., 1993)
Poly	phenol and	Polyphenol and Flavonoid Groups											
9	3.38	Catechols ^{a,b}	Organic com- pound	$C_6H_6O_2$	109.0280, 109.0284	81.0331, 67.0173	1	I	- 3.5845	I	1	I	MS-Dial
r	3.58	N-cis-caffeoyltyramine ^{a,b}	Phenolic acid	$C_{17}H_{17}O_4N$	I	I	300.1233, 300.1230	251.1083, 175.0755, 121.0652, 107.0498	0.7529		300, 107	215, 234, 280	(Chen et al., 1998)
п	4.17	b-D-glucoside of 1-methyl iso lariciresinol ^b	Lignan Glycosides	$C_{27}H_{35}O_{12}$	548.1784, 548.1790	254.0822	550.1905, 550.1898	256.0295	1.3099	548, 254, 238	550, 256, 184	280, 330	(Martins & Nunez, 2015)
15	4.60	Benzoic acid+10, 2MeO, 0-Hex ^b	Hydrolyzable tannins	$C_{15}H_{20}O_{10}$	359.0964, 359.0973	197.0450, 182.0212, 153.0547	I	I	- 2.4176	I	I	201, 222, 270	MS-Dial
17	5.20	Coumaroyl quinic acid	Phenolic acid	C ₁₆ H ₁₈ O ₈	337.0782, 337.0765	191.0555, 163.0385, 119.0486	1	I	5.0711	I	I	234, 275	MS-Dial
20	5.36	(Afzelin) Kaempferol 3-O-alpha-thamnoside ^a	Flavonoid	C ₂₁ H ₂₀ O ₁₀	431.0973, 431.0978	282.0511, 89.0226	1	I	- 1.2721	431, 281, 235, 178, 131, 89	433, 476, 415, 361, 272, 361, 272, 184, 123, 363, 294, 251, 174, 209, 110	270	(Chang et al., 1998a) KNApSAcK

No	Rt [Min]	Tentati ve metabolite identification	Classes	Chemical formula	HRMS of exa	ot mass measure	HRMS of exact mass measured, calculated and fragmentation ions	entation ions	Δ ppm	Fragentation in Low Resolution	in Low	λ max [nm]	λ max [nm] Referencess
					-[H-H]	Fragmenta- tion ions	[M+H] ⁺	Fragmenta- tion ions		Negative ion mode ESI (-)	Positive ion mode ESI (+)		
21	5.42	6,7-Dihydroxycoumarin ^{a,b}	Coumarins	$C_9H_6O_4$	177.0184, 177.0182	133.0282, 105.0328	179.0349, 179.0339	151.0388, 123.0444, 109.0652	0.8569 /0.7925	1	1	206, 237, 280	MS-Dial
23	5.52	Feruloyl Hexoside ^{a.b}	Phenolic acid $C_{16}H_{20}O_9$	$C_{16}H_{20}O_{9}$	355.1042, 355.1024	193.0499, 149.0595, 103.0022, 85.0279	I	I	5.2350	355, 191, 129, 85	I	210, 239	MS-Dial
24	5.55	Caffeoyl quinic acid ^{ab}	Phenolic acid $C_{I_0}H_{I_8}O_9$	C ₁₆ H ₁₈ O ₉	353.0883, 353.0881	191.0554, 179.0340, 161.0236, 135.0437, 85.0281	355.1023, 355.1025	163.0390, 145.0288, 117.0288	0.1650	353, 323, 191,135	I	213, 232, 280	(Simeon et al., 1989)
25	5.57	Caffeic acid hexoside ^a	Phenolic acid	$C_{15}H_{18}O_9$	341.0793	191.0552, 179.0341, 135.0439	343.0818, 343.0812	191.0338, 147.0441	I	I	I	213, 234, 277	MS-Dial
26	5.68	Coumaric acid	Phenolic acid	C ₉ H ₈ O ₃	163.0390, 163.0390	119.0489	165.0547, 165.0546	147.0442, 119.0493	0.0957	239, 277	I	I	MS-Dial
31	5.86	Feruloyl quinic acid ^{a,b}	Phenolic acid	$C_{17}H_{20}O_9$	367.1038, 367.1024	193.0499, 173.045, 149.0233, 134.0360	I	I	3.9892	1	I	201, 234, 277	MS-Dial
32	5.88	Caffeic acid ^{a,b}	Phenolic acid	$C_9H_8O_4$	179.0341, 179.0339	135.0438	181.0493, 181.0495	171.0653, 163.0390, 145.0283, 121.0652, 107.0860	0.9630/-1.4617	I	I	208, 237, 277	MS-Dial
34	6.16	Catechin ^a	Flavanols	$C_{15}H_{14}O_{6}$	289.0717, 289.0707	245.0814, 203.0708, 151.0389, 125.0230	291.0889, 291.0863	207.0562, 147.0440, 139.0392, 123.0444	3.5327	I	I	237, 289	MS-Dial
36	6.34	(-)-Epicatechin ^a	Flavanols	C ₁₅ H ₁₄ O ₆	289.0718, 289.0707	245.0816, 205.0500, 179.0340, 151.0390, 125.0229	I	207.0566, 179.0704, 147.0439, 139.0392, 123.0443	3.9550	1	I	237, 289	MS-Dial
37	7.00	Kaempferol 3-rutinoside- 7-glucoside ^{a,b}	Flavonoid	C ₃₃ H ₃₉ O ₂₀	755.2029, 755.2035	285.0403,	I	I	- 0.7262	755, 593, 285, 230, 211		218, 282, 321	(Beckmann & Geiger, 1963) KNApSAcK
38	7.10	Isorhamnetin 3-glucoside ^{a,b}	Flavonoid	$C_{22}H_{21}O_{12}$	477.1028, 477.1033	314.0437, 271.0280	I	I	- 1.1494	477, 314, 135	479, 460, 306	212, 321	(Santos & Sala- tino, 2000)

No	Rt [Min]	Tentative metabolite identification	Classes	Chemical formula	HRMS of exa	ct mass measure	HRMS of exact mass measured, calculated and fragmentation ions	intation ions	Δ ppm	Fragentation in Low Resolution	in Low	λ max [nm]	λ max [nm] Referencess
					[M-H] ⁻	Fragmenta- tion ions	[M+H] ⁺	Fragmenta- tion ions		Negative ion mode ESI (-)	Positive ion mode ESI (+)		
9	7.69	Quercetin 3-rutinoside- 7-glucoside ^{atb}	Flavonoid	$C_{33}H_{40}O_{21}$	771.1998, 771.1984	609.1608, 300.0278, 271.0238, 178.9975	1	. 1	1.8754	771, 609, 300, 271, 179	1	206, 284	(Beckmann & Geiger, 1963) KNApSAcK
41	7.76	Quercetin 3,7-O-beta- diglucopyranoside ^{a,b}	Flavonoid	$C_{27}H_{29}O_{17}$	625.1399, 625.1405	301.0358	627.1555,627.1563,	303.0500, 314.1380, 177.0548	- 0.8772/1.1548	625, 579, 372, 301	627, 314, 177	222, 331	(Santos & Sala- tino, 2000) KNApSAcK
42	8.09	Kaempferol 3-rhanno- syl-(1->2)-galactoside ^{a,b}	Flavonoid	$C_{27} H_{30} O_{15}$	593.1523, 593.1501	285.0403	595.1642, 595.1663	475.4135, 287.0548	3.7887/-3.4433	593, 285, 255	595, 558, 474, 335, 287, 236	226, 278	(Yasukawa & Takido, 1987)
43	8.20	Kaempferol 3-rhanno- syl-(1->2)-galactoside- 7-glucoside ^{a,b}	Flavonoid	$C_{33}H_{40}O_{20}$	755.2029, 755.2091	285.0403, 253.0823	757.2211, 757.2219	527.0624, 353.0652, 287.0549	1.0564	755, 594, 498, 425, 359, 299, 285	I	I	(Harborne & Baxter, 1999)
4	8.49	Rutin (Quercetin 3-O-rutino- side) ^a	Flavonoid	C ₂₇ H ₃₀ O ₁₆	609.1584, 609.1814	300.0276, 178.9971	611.1592, 611.1607	303.0500	- 2.3241	609, 301, 178, 254	611, 465, 367, 303, 249, 272, 202, 153, 110	256, 352	(Mohammed, Hamed et al., 2021)
45	8.50	Hirsutrin (Quercetin 3-0-beta- D-glucopyranoside)	Favonoid	$C_{21}H_{21}O_{12}$	463.0898, 463.0871	300.0277, 151.0025	465.1022, 465.1028	303.0497,	5.721/-1.2461	463, 301, 255, 151	465, 303, 165, 229	223, 278	(Mohammed, Hamed et al., 2021)
46	9.06	Wogonin ^a	Methoxyfla- vone	C ₁₆ H ₁₂ O ₅	283.0632, 283.0596	239.0709, 145.0281, 117.0330, 93.0329, 65.0233	1	I	I	I	I	I	(Santos & Sala- tino, 2000)
47	9.30	Trifolin (Kaempferol 3-O-beta- D-galactopyranoside) ^a	Flavonoid	C ₂₁ H ₂₁ O ₁₁	I	I	449.1078, 449.1084	287.0551, 127.0394	- 1.2211		449, 287, 172	280	(Santos & Sala- tino, 2000) KNApSAcK
8	9.50	Plantaginin ^b	Glycosyloxy- flavone	C ₂₁ H ₂₀ O ₁₁	447.0942, 447.0922	284.0329, 255.0297, 161.0701	449.1081, 449.1090	303.0500, 287.0553, 216.1577	4.4498/ 0.5840	447, 284,, 383, 297, 221, 163, 123	449,303, 287, 279, 472	241, 277	(Santos & Sala- tino, 2000) KNApSAcK
49	9.52	Isorhannetin 3-O-glu- coside ^a	Flavonoid	$C_{22} H_{22} O_{12}$	477.1028, 4771028	314.0438, 269.1769, 163.0378	479.1184, 479.119	317.0656, 293.0294	0.0645/-1.1446	I	479, 295, 241, 166, 259, 163, 115, 97		(Santos & Sala- tino, 2000) KNApSAcK
50	9.67	Quercetin-3-O-vicianoside	Flavonoid	$C_{26}H_{28}O_{16}$	595.1369, 595.1372	300.0276	597.1461 597.1460	303.0500, 97.0921	0.5040	595, 300, 135, 177		239, 270, 308	(Santos & Sala- tino, 2000)
51	9.79	Quercetin 3-sophoro- trioside	Flavonoid	C ₃₃ H ₃₉ O ₂₂	787.1927, 787.1927	478.0904, 385.0577, 352.0215	I	I	0.0000	787, 727, 661, 353, 385, 283, 216	789, 481, 303, 165	278, 280	(Harborne, 1999)

		Rt [Min] Tentative metabolite identification	Classes	Chemical formula	HRMS of exa	ct mass measure	HRMS of exact mass measured, calculated and fragmentation ions	intation ions	Δ ppm	Fragentation in Low Resolution	in Low	λ max [nm]	λ max [nm] Referencess
					-[H-M]	Fragmenta- tion ions	[M+H] ⁺	Fragmenta- tion ions		Negative ion mode ESI (-)	Positive ion mode ESI (+)		
55	10.42	Cis N-Feruloyltyramine ^{ab}	Phenolic Acid	$C_{18}H_{19}NO_4$	312.12415, 312.1225	190.0501, 178.0500, 148.0517, 135.0441	314.1385, 314.1387	206.0819, 177.0546, 145.0284, 121.0652	- 0.6234/3.3380	312, 177, 135	314, 206, 177, 255	280, 360	(Chang et al., 1998b)
56	11.65	Quercetin ^{a,b}	Flavonoid	$C_{15}H_9O_7$	301.0348, 301.0354	178.9983, 151.0026, 121.0281	303.2312, 303.2319	257.2276	1.8144/-2.1619	301, 153, 149, 165, 137	303, 179, 121, 151, 273, 257, 229	215, 278, 241, 280	(Mohammeda et al., 2020)
57	11.73	Kaempferol ^{a,b}	Flavonoid	C ₁₅ H ₁₀ O ₆	285.0399, 285.0405	151.0033, 133.0291	ı	I	- 1.9120		287, 153, 133, 165, 121		(Mohammed et al., 2020)
58	12.03	2'-Hydroxyflavanone ^{a.b}	Flavanone	C ₁₅ H ₁₂ O ₃	239.0713, 239.0703	145.0283, 119.0489, 93.0328	I	I	4.3872	239, 145, 119	241, 223, 129	241, 270	Ms-dial
59	12.15	Patrinoside ^b	Iridoid Glu- cosides	$C_{21}H_{34}O_{11}$	461.2014, 461.2017	321.2769, 210.0682	I	I	3.2736	461, 323, 143		212, 306	(Demirezer et al., 1999)
63	15.10	Apigenin ^a	Flavonoid	$C_{15}H_{10}O_5$	269.0458, 269.0444	ı	I	I	5.0438	269, 162	I	I	Ms-dial

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 Table 4 In vitro antioxidant effect of Annona Abdel Razek different plant parts

Sample Con. µg/ml	ABTS	DPPH
Vit. C	46.88 ± 0.15	52.67 ± 0.32
Trolox	31.72 ± 0.05	33.34 ± 0.06
Leaves extract	22.85 ± 0.10	16.66 ± 0.19
Barks extract	9.22 ± 0.25	13.33 ± 0.37
Fruits extract	117.50 ± 0.86	76.65 ± 0.48
Seeds extract	151.80 ± 0.85	101.4 ± 0.14
Total alkaloids extract	227.90 ± 0.45	94.11±0.16
BuOH extract (flavonoids)	101.30 ± 0.1	47.5 ± 0.58
EtOAc extract (polyphenolics)	49.44 ± 0.07	50.82 ± 0.04

Data represented as % of inhibition of triplicate readings

3.4.2.3 Cell organelles marker enzymes The present study revealed insignificant changes in SDH, LDH, G-6-Pase, AP and 5'-Nu in normal rats protected with the plant extracts and ranitidine drug as compared with the control group (Table S6 and Fig. 3c) in concerning cell organelles marker enzymes. Gastroulcerative rats showed significant increase in SDH, LDH, G-6-Pase, AP and 5'-Nu activities by 209.27, 390.29, 102.03, 172.04 and 116.56%, respectively as comped with the control group. Ulcerative rats protected with plant extracts and ranitidine drug recorded significant decrease in the selected enzyme markers with variable degrees as compared with the ulcerative group. SDH showed significant decrease by 78.86, 71.87and 71.60% after protection with leaves and bark of hybrid Abdel Razek, and ranitidine drug, respectively. LDH recorded inhibition activity by 76.64, 74.53, and 79.23%, while G-6-Pase showed inhibition by 40.50, 35.13, 37.56, 32.84 and 37.16%, respectively. In addition, AP enzyme decreased by 51.56, 59.22,

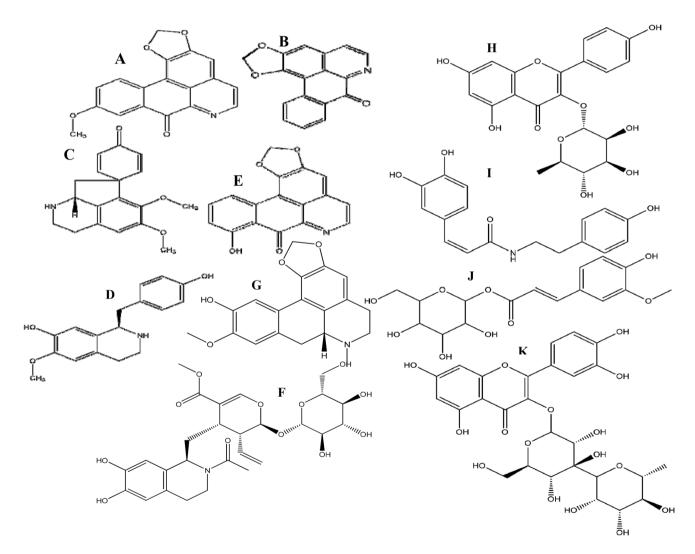


Fig. 1 Structure of compounds A Lanuginosine, B Liriodenine, C Stepharine, D Coclaurine, E Oxostephanosine, F Ipecoside, G (–)-Pphanostenine, H Kaempferol-*O*-3-alpha-rhamnopyranoside, I N-cis-caffeoyltyramine, J Caffeic acid hexoside and K Rutin

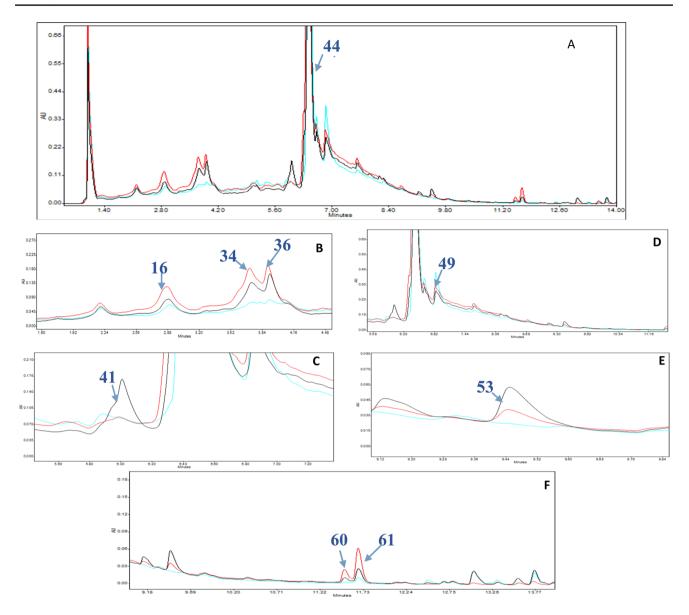


Fig. 2 A Major metabolites by comparison profiles between UPLC-UV chromatograms of compounds recorded at 330 nm of *Annona* species, where *A. cherimola* (black line), *A. squamosa* (green line) and *A.* Abdel Razek (red line).

and 52.85%, while 5'-Nu decreased by 47.12, 44.68, and 50.27%, respectively.

3.4.3 In vivo therapeutic effect of Annona Abdel Razek

3.4.3.1 Ulcer markers In the therapeutic effect of the plant extract under investigation, Stomach pH in control rats treated with plant extracts did not change significantly, suggesting extract safety. Gastro-ulcerative rats showed significant decrease by 38.46% as compared with the control group. After treated with leaves and barks Abel Razek extracts as well as ranitidine drug, it showed significant increase in the pH level by 53.50, 57.50, and 55.00%, respectively when compared with the ulcer group (Table *S7* and Fig. 4a). Gas-

tric volume in control rats treated with Abdel Razek extracts and the selected drug, recorded insignificant changes. Ulcerative stomach showed significant increase in its gastric volume content by 185.18% when compared with the control group. Treated rats with leaves and bark of hybrid Abdel Razek, and ranitidine drug recorded significant decrease in gastric volume by 64.90, 60.92, and 53.31%, respectively when compared with the ulcer group (Table *S7* and Fig. 4a). Normal rats treated with plant extracts recorded insignificant changes in gastric total acidity when compared with the control group (Table S7). The gastric ulcerative rats showed significant decrease in the total acidity values by 43.90% when compared with the control group. Treated rats with leaves and barks of Abdel Razek, and rantidine

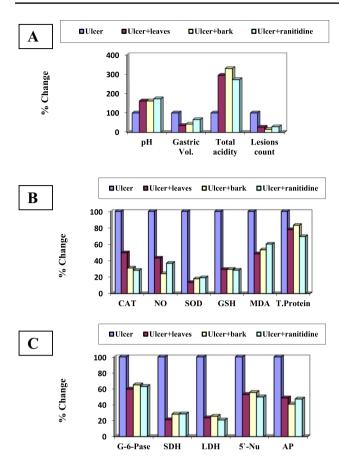


Fig.3 Percentage changes over ulcer group of the protective effect of *A*. Abdel Razek extracts and ranitidine on **a** gastric ulcer index, **b** antioxidant and protein levels and **c** cell organelles marker enzymes

drug recorded significant increase in the total acidity level by 71.42, 74.69, and 28.57%, respectively comparing these values with those of the ulcer group (Table *S7* and Fig. 4a). The stomach of ulcerative group showed nearly twenty one ulcer lesions per stomach. Combined treatment with ethanol and two extracts or ranitidine reduced the number of lesions/organ by 71.92%, 58.55%, and 68.41%, respectively compared with the ulcerative group. (Table *S7* and Fig. 4a).

3.4.3.2 Antioxidant and total protein levels Rats treated with Abdel Razek extracts or ranitidine drug showed not significant changes in catalase, NO, SOD, glutathione, malondialdehyde and total protein levels comparison with the control group. Significant increase in catalase, NO, SOD, glutathione, malondialdehyde and total protein levels comparison with the control group. Significant increase in catalase, NO, SOD, glutathione, malondialdehyde and total protein levels in gastroulcerative rats by 130.23, 732.00, and 558.97 and 61.41%, were observed as compared with the control group, respectively. Similarly, decreased impacts of gastroulcerative rats with leaves and barks Abdel Razek and ranitidine drug were observed for nitric oxide 84.80%, 83.65%, and 82.21%; superoxide dismutase, 88.16%, 88.77%, and 86.75%; glutathione level, 73.36%, 75.81%, and 71.76%;

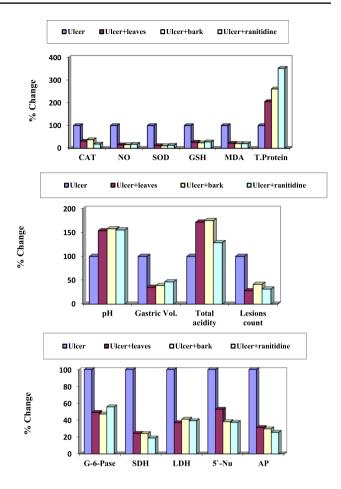


Fig. 4 Percentage changes over ulcer group of the therapeutic effect of *A*. Abdel Razek extracts and ranitidine on **a** gastric ulcer markers, **b** antioxidant and protein levels and **c** cell organelles marker enzymes

malondialdehyde, 77.82, 79.18, and 79.37%, respectively. In contrast, total protein content significantly increased -105.26%, 160.52%, and 251.26%, respectively (Table S8 and Fig. 4b).

3.4.3.3 Stomach marker enzymes Insignificant changes were observed in SDH, LDH, G-6-Pase, AP and 5'-Nu enzymes in normal rats treated with leaves and barks of the plant extract under investigation as well as the reference drug in comparison with the control group. Gastroulcerative rats showed significant increase in SDH, LDH, G-6-Pase, AP and 5'-Nu activities by 287.16, 199.50, and 183.55%, respectively versus to the control group. Ulcerative rats treated with Abdel Razek leaves and barks extracts and ranitidine recorded significant decrease in the selected enzymes by variable degrees as compared with the ulcer group. SDH showed significant decrease by 75.61, 75.90, and 81.20%, respectively. Similarly, LDH, G-6-Pase, and AP enzymes decreased by 62.98%, 58.92, and 60.71%; 50.90%, 52.79%, and 44.27%; and 68.81%, 70.37%, and 74.49%, respectively

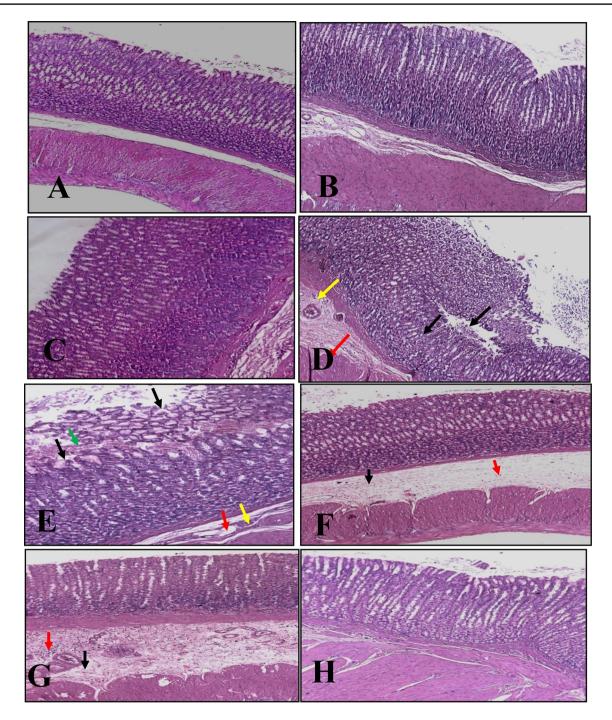


Fig. 5 Photograph of rat stomach mucosal layers. A Normal rat stomach showing no damage to the mucosa and normal mucosal and submucosal layers (H&Ex100). **B** & **C** Normal rats treated with Abdel Razek's leaves and barks, respectively, showing minimal damage of the surface epithelium and no leucocyte infiltration of the submucosal layer (H&E stain $10\times$). **D** Stomach mucosa after one hour of ethanol ulceration, demonstrating erosion of the surface epithelium (black arrow), moderate edoema (red arrow), and moderate leucocyte infiltration of the submucosal layer (yellow arrow) (H&E stain $10\times$). **E** Stomach mucosa after one hour of ethanol ulceration, revealing erosion of the surface epithelium (black arrow), mild edema (red arrow), mild leucocyte infiltration of the submucosal layer (yellow arrow), and hemorrhage (green arrow) (H&E stain 20×). F Ulcerative gastric mucosa treated with Abdel Razek leaves demonstrating minimal rupture of the surface epithelium (healed), considerable edema (black arrow), and minor leucocyte infiltration of the submucosal layer (red arrow) (H&E stain 10×). G Ulcerative stomach mucosa treated with Abdel Razek's barks, demonstrating no damage of the surface epithelium (healed), considerable edema (black arrow), and minor leucocyte infiltration of the submucosal layer (red arrow) (H&E stain 10×). H Ulcerative gastric mucosa treated with ranitidine, revealing no disruption of the submucosal layer (H&E stain 10×) while 5'-Nu decreased by 47.12, 61.80, and 62.63%, respectively (Tables *S*9 and Fig. 4c).

3.5 Stomach histopathological features

3.5.1 Protective effect of Annona Abdel Razek

No lesions were seen in gastric mucosa, including in normal mucosal and submucosal layers in control rats. Treatment with leaves and bark extracts in normal rats produced no damage to surface epithelia and did not induce leukocyte infiltration into submucosal layers (Fig. 5A, B and C). Surface epithelium was eroded with moderate edema and moderate leucocyte infiltration of the submucosal layer with bleeding in a section of stomach mucosa exposed to ethanol (Fig. 5D). Gastric mucosa was protected by administration of extracts where no disruption of the surface epithelium (healed) with significant edema and minor leucocyte infiltration of the submucosal layer was observed on histological examination (Fig. 5E, F and G).

3.5.2 Therapeutic effect of Annona Abdel Razek

A histological section of stomach mucosa one week after ethanol administration (H&E stain 10x) shows a deep ulcer with surface epithelium (black arrow), considerable edema (red arrow), and leucocyte infiltration into the submucosal layer (yellow arrow) (Fig. 6A). Surface epithelium was eroded with mild to moderate edema and minor leucocyte infiltration into the submucosal layer in ulcerative gastric mucosa treated with Abdel Razek leaves (Fig. 6B). After administration of bark extract, surface epithelium of ulcerative gastric mucosa was intact with slight edema and minor leucocyte infiltration of the submucosal layer (Fig. 6C). The ulcerative stomach mucosa treated with ranitidine appeared normal on histological examination (Fig. 6D).

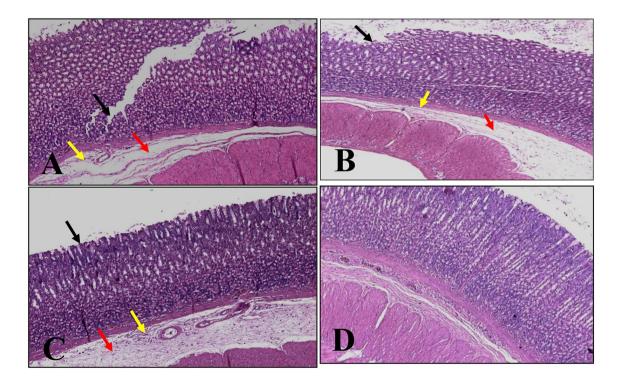


Fig. 6 Photograph of rat stomach mucosal layers. **A** Gastric mucosa of ethanol induced ulcer for one week showing deep ulcer with surface epithelium (black arrow) with moderate edema (red arrow) and leucocytes infiltration of the submucosal layer (yellow arrow) (H&E stain $10\times$). **B** Ulcerative gastric epithelium pretreated with Abdel Razek leaves revealing surface epithelial degradation (black arrow), mild-moderate edema (red arrow), and minor leucocyte infiltration of the submucosal layer (yellow arrow) (H&E stain $10\times$). **C** Ulcerative

gastric mucosa treated with Abdel Razek barks shows intact (healed) surface epithelium (black arrow), modest edema (red arrow), and mild leucocyte infiltration of the submucosal layer (yellow arrow) (H&E stain $10 \times$). **D** Ulcerative gastric mucosa treated with ranitidine, revealing no rupture of the surface epithelium (healed), no edema, and no leukocyte infiltration of the submucosal layer (H&E stain $10 \times$).

4 Discussion

A combination of RAPD and ISSR data indicates 100% similarity between A. cherimola and Annona Abdel Razek and 80% between A. squamosa and Annona Abdel Razek. Sixtythree secondary metabolites were detected in leaves and bark of the new cultivar, Annona Abdel Razek, using UPLC highresolution MS for simultaneous chromatographic separation and mass spectrometric detection. Phytochemical studies revealed the presence of alkaloids identified as liriodenine, lanuginosine, oxostephanosine, stepharine, coclaurine, corytuberine, fenfangjine G, oxoanolobine and phanostenine. These compounds are belonging to the oxoaporphine, aporphine, isoquinolines, and benzylisoquinoline groups. Alkaloids include a basic nitrogen atom and exhibit antiulcer properties. The antiulcer activity of 2-phenylquinoline is demonstrated in several pharmacological investigations that used EtOH to induce (Leite et al., 2020). Malonyl-CoA and p-coumaroyl-CoA are two fundamental metabolites that give flavonoids their fifteen-carbon skeletons (C6-C3-C6). The condensation of three molecules of malonyl-CoA with one molecule of p-coumaroyl-CoA produce a chalcone intermediate, which is key to the biosynthetic process. Chalcones are the building blocks for the wide variety of flavonoid derivatives found in plants (de Lira Mota et al., 2009). Several flavonol glycosides, including kaempferol, quercetin, and isorhamnetin, were found in the leaves and bark of Annona Abdel Razek. Phenolic acids (p-coumaric, ferulic, and caffeic) and their quinic acid derivatives were also identified as chemicals with antioxidant properties.

Ethanol destabilizes the mucus-bicarbonate-phospholipid layer, resulting in retro-diffusion of H+ions and damage to epithelial cells. This damage leads to altered levels of mucosal enzymes, such as SDH, LDH, G-6-Pase, AP and 5'-Nu. Mast cell activation causes the release of inflammatory mediators that encourage neutrophils to migrate to injured areas. Subsequently, generation of ROS increases cellular damage and causes tissue necrosis (Fahmi et al., 2019). Stomach ulcers are also affected by stomach acidity, and controlling acidity is a substantial challenge (Aboul Naser et al., 2020). Antiacids, H₂ receptor blockers, such as ranitidine and its analogues, anticholinergics (pirenzepine), and proton pump blockers, such as omeprazole, were all employed (Wallace & Sharkey, 2011). However, most currently available medications have limited efficacy and cause adverse side effects, making stomach ulcer therapy a major concern.

Natural products have a significant role in controlling gastric ulcer disease (Newman & Cragg, 2020), reflecting their notable structural diversity and selective biological activity (Cragg & Newman, 2013). Flavonoids play an essential role in preventing gastric lesions caused by various ulcerogenic agents. They also aid in healing such ulcers and may be novel agents for suppressing peptic ulcers (de Lira Mota et al., 2009). Several reports on *Annona* species (Ma & Liu, 2014) support our current results that indicate that extracts of *A*nnona Abdel Razek leaves and bark may protect gastromucosal integrity via synergistic actions of flavonoids, condensed tannins, and saponins, in addition to alkaloids. Further, isocorydine, liriodenine, lanuginosine, oxostephanosine, stepharine, and coclaurine have all been identified as aporphine benzylisoquinolinic alkaloids. These alkaloids have been isolated from a variety of plant parts and species, including *Annona squamosa* twigs (Yadav et al., 2011), *Isopyrum thalictroides* (Ranunculaceae) roots and rhizomes (Istatkova & Philipov, 2004), and *Dactylicapnos scandens* (Papaveraceae) roots (Wang et al., 2018).

Qualitative and quantitative differences in phenolic, flavonoid and alkaloid constituents exist between *A. cherimolia* and *A. squamosa* and constituents identified in the present work. Current results are most similar to data reported by Yadav et al., 2011, who indicated that antisecretory mechanisms of isolated aporphine alkaloids from twigs *A. squamosa* such as N-methylcorydaldine, lanuginosine, (+)-anomuricine, and N-methyl-6,7-dimethoxyisoquinolone, involve inhibition of gastric H⁺/ K⁺-ATPase activity.

Gastro-duodenal ulcers are associated with marked alterations in multiple biochemical parameters (Esmaeilnejad et al., 2012). A substantial increase in MDA, SOD, CAT, NO, and GSH was observed in the current study. Gastric mucosal MDA levels in patients with peptic ulcer and gastritis are likely to indicate free radical-mediated gastric mucosal damage (Ren et al., 2021). Hydroxyl radicals oxidize essential cellular structural and functional proteins, as well as membrane lipids. The loss of membrane fluidity, ion transport, integrity, and loss of various cellular functions, are all caused by lipid peroxidation (Aboul Naser et al., 2020). Glutathione levels are dependent on the activity of several enzymes (Rahman et al., 2020). For example, an increase in GSH concentration can be caused by the upregulation of γ -glutamyl-cysteine synthetase, an enzyme involved in GSH synthesis. Increased activity of GSH peroxidase and GSH transferase, however, reduces GSH concentration. Thus, a decrease in glutathione peroxidase and glutathione transferase is reported in indomethacin-induced stomach ulcers (Koc et al., 2008), ethanol-induced mucosal injury (Dejban et al., 2020), and stress ulcers (Liu et al., 2011).

The efficacy of antisecretory and antiulcerogenic actions of selected extracts is supported by an observed decrease in gastric volume, lesion count, and pH (Mohammed et al., 2020). Also, total protein levels can be used as an indicator of the severity of cellular dysfunction in a variety of diseases. Protein synthesis stimulation has been identified as a contributory self-healing mechanism that encourages regeneration (Bryndin & Bryndina, 2020).

The histological sections showed altered gastric mucosa after ethanol ulceration, with deep ulcers reaching the basement membrane of the lamina propria. Some polymorphous lymphocyte fibrin was found at the swollen ulcer base (Okabe & Pfeiffer, 1972). The ulcer is bordered by hyperplastic gastric glands. Few lymphocytes and polymorphonuclear leucocytes with a high degree of fibrosis are found in the lamina propria. Our histological observations of the ulcerative mucosa are consistent with these findings. We also noticed a well-developed epithelium at the ulcer margin in rats protected or treated with plant extracts. Interestingly, ulcer healing occurs either by a slow process involving cell migration to the luminal surface and deposition on the area stripped by an ulcerogenic agent or through a rapid process involving cell migration to the mucosal surface and deposition on areas reduced by the ulcerogenic agent (Joshi et al., 2011).

5 Conclusion

The metabolomics profile of crude extracts of *Annona* Abdel Razek contains 63 compounds, including flavonoids, phenolic compounds, amino acids, and alkaloids. *Annona* Abdel Razek leaves and bark extracts exhibit anti-ulcerogenic properties. The bark extract shows better protective and healing activity for gastric ulcers than leaves extract. Further, the bark extract is more efficacious for therapy rather than prophylaxis.

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Author contributions MAM, SEG, EMH, PK: Designed the plant experiment, conducted the identification tests, and analyzed the data of this section. MAH, MAM, AMAE Participated in biological tests and analysis of data of this part. MAH: Analyzed histopathological images and revised the final form of the paper. All authors reviewed the final form of the paper and agreed on submittal for publication.

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Data availability Data availability statements provide a statement about where data supporting the results reported in a published article can be found, including datasets analyzed or generated during the study.

Declarations

Conflict of interest The authors have no conflicts of interest to declare.

Ethical approval Experiments on animals were conducted after the approval of the Institutional Ethical Committee, no. (12/106).

Consent to participate Not applicable.

Consent for publication Not applicable.

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