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Article

Three New Germacrane-Type Sesquiterpenes with NGF-Potentiating Activity from *Valeriana officinalis* var. *latiofolia*

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Abstract: Three new germacrane-type sesquiterpenoids, volvalerenal F (1), volvalerenal G (2) and volvalerenic acid D (3), along with five known compounds 4-8, were isolated from the CHCl₃ soluble partition of the ethanol extract of *Valeriana officinalis* var. *latiofolia*. The structures of the new compounds were determined on the basis of spectroscopic evidence, including their 1D- and 2D-NMR spectra, as well as mass spectrometry. The eight germacrane-type sesquiterpenoids showed nerve growth factor (NGF) potentiating activity, which mediates the neurite outgrowth in PC 12D cells. This study intends to reveal the chemical basis of the use of *V. officinalis* var. *latiofolia* as a dietary supplement.

Keywords: *Valeriana officinalis* var. *latiofolia*; germacrane-type sesquiterpenes; volvalerenal; volvalerenic acid; PC 12D; NGF-potentiating activity

1. Introduction

The genus *Valeriana*, from the family *Valerianaceae*, consists of about 250 species widely distributed all over the World. *Valeriana* is a perennial herb native to Europe, Asia and North

America [1,2]. Eleven out of 28 *Valeriana* genus species (including one variant) found in China are traditionally used as medicines. Their dried underground parts (roots and rhizomes) exhibit anodyne, antiphlogistic, expectorant and antiasthmatic activities [3–5].

Valerians have been used clinically as tranquillizers for the treatment of nervousness, agitation and as a mild sedative to improve sleep [6–13]. Previous studies reported that *Valeriana* contains numerous chemical constituents, including volatile oil, iridoids, flavones, alkaloids, amino acids, and lignans, *etc.* [14–20]. Valerian (*Valeriana officinalis Linn.*) has been included in the pharmacopoeias in Europe and the United States [21,22], its extracts are sold as dietary supplements and were listed among the top 10 best-selling herbal supplements in the United States in 2002 [23]. Valerian is also widely used as a precious perfume added into food, drink, cosmetics, tobacco, and so on. With significant medicinal and commercial value, it will surely deserve future development.

V. officinalis var. *latiofolia*, a variant of *V. officinalis Linn*, is produced mainly in the Guizhou, Sichuan and Yunnan provinces in Southwest China. Particularly in Guizhou, *V. officinalis* var. *latiofolia* is widely cultivated and has become one of the local industry pillars [1,24]. This variant shares some common pharmaceutical actions and chemical constituents with *V. officinalis Linn* [25]. Abundant research has been done on *V. officinalis* var. *latiofolia* so far, and it is assumed that the sesquiterpenoids from the volatile oil and iridoids are the main contributors to its antidepressant and antinervousness activities [24,25]. Pairs of active compounds have been isolated and identified from *Valeriana*, including germacrane-type sesquiterpenoids, volvalerenals A–E, volvalerenic acids A–C, valerianin A–B and heishuixiecaoline A–C [13,16–19]. However, neither the active component(s) responsible for the therapeutic properties of *Valeriana* nor the related molecular mechanisms are clearly understood, which severely hinders the wider application of *Valeriana* products. Therefore, in this paper the chemical constituents of *V. officinalis* var. *latiofolia* have been systematically investigated, and eight germacrane-type sesquiterpenoids (including three new compounds, volvalerenal F (1), volvalerenal G (2) and volvalerenic acid D (3)) were isolated and identified from chloroform extracts of this herb's syrup. Additionally, the NGF-potentiating activities of the obtained products were evaluated.

2. Results and Discussion

Volvalerenal F (1) was isolated as a colorless oil. High-resolution mass spectrometry (HR-ESI-MS) $(m/z \ 271.1678 \ [M+Na]^+$, calcd. for C₁₆H₂₄O₂Na, 271.1669) of 1 indicated that its molecular formula is C₁₆H₂₄O₂, and together with the NMR data (Table 1), implied five unsaturated degrees in the molecule. The IR spectrum indicated the presence of carbonyl (1730 cm⁻¹) and carbon-carbon double-bond (1627 cm⁻¹) absorptions. The ¹H-NMR spectrum of 1 (Table 1) displayed three olefinic protons at $\delta_{\rm H} \ 5.27$ (1H, dd, J = 5.4, 10.2 Hz, H-1), 5.34 (1H, ddd, J = 9.0, 9.0, 3.2 Hz, H-4), and 5.21 (1H, dd, J = 9.0, 9.0, Hz, H-5) indicating the presence of two double bonds, methylene proton peaks at $\delta_{\rm H} \ 4.42$ (1H, d, $J = 12.0 \ {\rm Hz}$, H-14a) and $\delta_{\rm H} \ 4.21$ (1H, d, $J = 12.0 \ {\rm Hz}$, H-14b), and three methyls at $\delta_{\rm H} \ 0.98$ (3H, s, 13-CH₃), 1.04 (3H, s, 12-CH₃) and 2.01 (3H, s, 16-CH₃). Considering the DEPT spectra, the ¹³C-NMR spectrum of 1 (Table 1) suggested the existence of an acetate carbonyl carbon at $\delta_{\rm C} \ 171.3$ (C-15), four olefinic carbons at $\delta_{\rm C} \ 128.8$ (C-1), 133.1 (C-10), 128.3 (C-4) and 129.6 (C-5), as well as five methylenes at $\delta_{\rm C} \ 28.2$ (C-2), 26.8 (C-3), 23.0 (C-8), 35.3 (C-9) and 61.7 (C-14), two methines at $\delta_{\rm C} \ 26.4$ (C-6) and 31.9 (C-7) and three methyls at $\delta_{\rm C} \ 15.6$ (C-13), 21.1 (C-16) and 28.9 (C-12).

NO.	1		2		3	
	δ _C	$\delta_{\rm H}$	δ _C	$\delta_{\rm H}$	δ _C	δ _H
1	128.8	5.27 dd (5.4, 10.2)	128.2	5.22 dd (5.4, 11.4)	130.7	5.33 dd (5.4, 11.4)
2a	28.2	2.23 <i>m</i> H-α	28.3	2.22 <i>m</i> (H-α)	29.2	2.38 <i>m</i> (H-β)
2b		2.11 <i>m</i> H-β		2.15 <i>m</i> (H-β)		2.26 <i>m</i> (H-α)
3a	26.8	2.14 <i>m</i> H-β	24.0	2.71 <i>dt</i> (12.0, 4.2, H-α)	27.1	2.71 <i>dt</i> (12.6, 4.0, H-α)
3b		2.08 <i>m</i> H-α		2.05 <i>td</i> (12.6, 4.2, H-β)		2.17 <i>td</i> (12.6, 4.0, H-β)
4	128.3	5.34 ddd (9.0, 9.0, 3.2)	143.9		132.1	
5	129.6	5.21 dd (9.0, 9.0)	158.9	6.48 d (9.0)	144.8	6.72 <i>d</i> (9.6)
6	26.4	1.14 <i>t</i> (9.0, H-α)	31.2	1.52 <i>t</i> (13.2, 9.0, H-α)	30.0	1.27 <i>dd</i> (9.6, 4.0, H-α)
7	31.9	0.47 <i>m</i> (H-α)	38.8	1.02 <i>m</i> (H-α)	36.7	0.82 <i>m</i> (H-α)
8a	23.0	1.73 <i>m</i> (H-β)	24.6	1.86 <i>m</i> (H-β)	24.7	1.81 <i>m</i> (H-β)
8b		0.76 <i>m</i> (H-α)		0.90 <i>m</i> (H-α)		0.89 <i>m</i> (H-α)
9a	35.3	2.39 <i>m</i> (H-β)	35.5	2.60 <i>m</i> (H-β)	36.2	2.36 <i>m</i> (H-β)
9b		1.87 <i>t</i> (12.6, α)		1.88 <i>m</i> (H-α)		2.25 <i>m</i> (H-α)
10	133.1		139.1		134.3	
11	17.2		22.4		21.5	
12	28.9	1.04 <i>s</i>	28.7	1.16 s	28.9	1.12 <i>s</i>
13	15.6	0.98 s	16.0	1.20 s	16.1	1.14 <i>s</i>
14a	61.7	4.42 d (12.0)	196.4	9.20 s	172.9	
14b		4.21 <i>d</i> (12.0)				
15	171.3		59.1	3.69 d (12.0)	62.8	4.32 d (12.0)
				3.43 <i>d</i> (12.0)		4.15 d (12.0)
16	21.1	2.01 s			172.0	
17					20.8	1.97 s

Table 1. ¹H and ¹³C-NMR Spectroscopic Data of Compounds 1–3.

Compound 1 recorded in $CDCl_3$. Compounds 2 and 3 recorded in CD_3OD . ¹H-NMR recorded at 600 MHz. ¹³C-NMR recorded at 150 MHz.

To confirm the structure of **1**, ¹H-¹H COSY and HMBC experiments were conducted (Figure 1), which showed the key correlations such as H-1/H-2, H-2/H-3, H-3 (a,b)/H-4, H-4/H-5, H-5/H-6, H-6/H-7, and H-8/H-9 in its COSY, H-14/C-1, H-14/C-9, H-14/C-10, H-14/C-15, H-3/C-4, H-5/C-4 and H-12/C-6, H-12/C-7, H-12/C-11 and H-12/C-13 in its HMBC. The coupling constant of 9.0 Hz between H-4 and H-5 indicated the *Z*-configuration of the double bond [26,27]. The coupling constant of 9.0 Hz and the NOE correlations between H-6 and H-7 suggested the *syn* and α -oriented of the cyclopropane moiety, and the α -orientation of H-6 and H-7 were assigned by the correlations of H-7/CH₃-12 and H-6/CH₃-12 [28]. The correlations of H-2(b)/H-15(a, b) indicated $\Delta^{1,10}$ to be *Z* configured. From the above data, the structure of **1** was identified as 14-acetoxy-11,11-dimethylbicyclo[8.1.0]undeca-4*Z* (5),10*Z* (1)-diene, and the compound was named volvalerenal F.





Compound **2** was isolated as a colorless oil. The molecular formula was assigned as $C_{15}H_{22}O_2$ on the basis of HR-ESI-MS from the $[M+H]^+$ signal at *m/z* 235.1696 (calcd. for $C_{15}H_{23}O_2$, 235.1693), with five degrees of unsaturation. The IR spectra displayed the presence of carbonyl (1726 cm⁻¹), α , β -unsaturated aldehyde (1678 cm⁻¹), and carbon-carbon double-bond (1626 cm⁻¹) absorptions.

The ¹³C-NMR and DEPT spectra of **2** (Table 1) showed an aldehyde carbon at $\delta_{\rm C}$ 196.4 (C-14), four double bond carbons at $\delta_{\rm C}$ 128.2 (C-1), 143.9 (C-4), 158.9 (C-5) and 139.1 (C-10), as well as five methylenes (one oxygenated) at $\delta_{\rm C}$ 24.0 (C-3), 24.6 (C-8), 28.3 (C-2), 35.5 (C-9) and 59.1 (C-15), two methines at $\delta_{\rm C}$ 31.2 (C-6) and 38.8 (C-7), and two methyls at $\delta_{\rm C}$ 16.0 (C-13) and 28.7 (C-12). These data suggested that compound **2** was also a germacrane-type sesquiterpenoid. Its ¹H-NMR spectrum displayed an aldehydic proton obviously at $\delta_{\rm H}$ 9.20 (1H, s, H-14), an oxygenated methine proton at $\delta_{\rm H}$ 3.69 (1H, d, J = 12.0 Hz, H-15a) and 3.43 (1H, d, J = 12.0 Hz, H-15b). Above data indicated **2** was structurally similar to heishuixiecaoline B reported in the literature [13].

The proposed structure was further confirmed by HMBC correlations (Figure 2). Key long-range correlations were observed between H-12/C-6, H-12/C-7, H-12/C-11, H-12/C-13, H-15/C-1, H-15/C-10, H-15/C-9, and H-14/C4, H-14/C-5, which suggested the aldehyde group and hydroxyl group were located to C-4 and C-15, respectively. The α -orientation of H-6 and H-7 were assigned as being the same as those of **1** by the correlations of H-6/H-7, H-7/CH₃-12 and H-6/CH₃-12 in the NOESY experiment (Figure 4), and the *E*- and *Z*- configurations of $\Delta^{4,5}$ and $\Delta^{1,10}$ were determined to be the same as in compound **1** by the correlations of H-5/H-14, H-2 (a, b)/H-15 and H-5 with H-1.

Figure 2. The correlations of structures, ¹H-¹H COSY and Key HMBC of **2**.



On the basis of the above data, the structure of **2** was identified as 4-formyl-10-hydroxymethyl-11, 11-dimethylbicyclogermacren-4E (5), 10Z (1)-diene, and the product was named volvalerenal G.

Volvalerenic acid D (**3**) was also isolated as a colorless oil. The HR-ESI-MS of **3** indicated that its molecular formula is $C_{17}H_{24}O_2$ (*m/z* 293.1744 [M+H]⁺, calcd. for 293.1747). The IR data was completely similar to that of compound **1**, which suggested that **3** was also a germacrane-type sesquiterpenoid.

The NMR spectrum of compound **3** (Table 1) showed the following signals: an acetate carbonyl carbon, four olefinic carbons, five methylenes (one oxygenated), two methines and three methyls, which were quite similar to those of compound **1**. In addition, it is noteworthy that the NMR data displayed an obvious carboxyl carbon signal at δ_C 172.0 (C-16). The ¹H-¹H COSY spectrum showed key correlations such as H-6/H-7 and H-8/H-9, and key long-range correlations were observed in the HMBC experiments between H-3/C-14 and H-5/C-14 (Figure 3), which suggested the carboxyl group was located to C-4.

Figure 3. The correlations of structures, ¹H-¹H COSY and Key HMBC of **3**.



The NOESY correlations (Figure 4), the coupling constant of 4.0 Hz and the NOE correlations between H-6 and H-7 suggested a *trans* geometry around the cyclopropane ring, and the correlations of H-6/H-3a, H-2 (a, b)/H-15 suggested the *E*- and *Z*- configuration of $\Delta^{4,5}$ and $\Delta^{1,10}$.





From the above data, the structure of **3** was identified as 15-acetoxy-4-carboxy-11, 11-dimethyl-bicyclogermacren-4E(5), 10Z(1)-diene, and it compound was named Volvalerenic acid D.

The five known compounds were identified as madolin A (4) and B (7) [26], vovalerenal A (6) and B (5) [18] and heishuixiecaoline B (8) [13] by comparing their NMR spectroscopic data with the literature values. The structures of compounds 1-8 are shown in Figure 5.



The propensity of compounds **1–8** to enhance the activity on nerve growth factor (NGF)-mediated neurite outgrowth in PC 12D cells was assessed as described previously [25]. The neurite-bearing cells accounted for 22.74% and 100% in the control experiments incubated with 2 and 50 ng/mL NGF after 48 h, respectively.

Under 2 ng/mL NGF conditions, all eight tested compounds (at 10, 30, 100 μ mol) showed NGF-potentiating activities in various levels. Compound **3** (at 100 μ mol) reached 50.15%, in particular (Table 2).

Commonia	NGF(ng/mL)	Cell viability (%)			
Compound		10 µmol	30 µmol	100 µmol	
1	2	24.85 ± 0.98	33.97 ± 1.77 ^b	43.61 ± 2.11 ^c	
2	2	24.42 ± 1.12	35.34 ± 1.48 ^b	44.30 ± 1.85 ^c	
3	2	$26.48 \pm 0.89 \; ^{a}$	37.51 ± 1.66 ^b	50.15 ± 2.23 ^c	
4	2	23.50 ± 1.26	29.33 ± 0.88 ^b	33.87 ± 1.63 ^c	
5	2	24.37 ± 1.01	30.12 ± 1.97 ^b	40.79 ± 1.17 ^c	
6	2	24.74 ± 1.47	34.84 ± 2.35 ^b	44.15 ± 2.19 °	
7	2	24.45 ± 1.01	33.96 ± 1.13 ^b	46.69 ± 2.14 ^c	
8	2	24.26 ± 0.73	33.79 ± 1.17 ^b	49.25 ± 1.25 °	
	2		$23.08\%\pm1.28$		
	50		100%		
	0		$3.12\%\pm0.88$		

Table 2. Effects of compounds 1–8 on the proportion of neurite-bearing PC 12D cells in the presence of NGF.

Results expressed as mean \pm SD (n = 6) of eight independent experiments. The 100% value was obtained from 50 ng/mL NGF in the absence of compounds. A statistically significant difference (a, b and c, p < 0.01) from the control (2 ng/mL NGF) in the absence of compounds was apparent.

3. Experimental

3.1. General

Optical rotations were measured with a Perkin-Elmer 343 polarimeter (Perkin-Elmer, Waltham, MA, USA). IR spectra were recorded on the Bio-Rad FTS-65A spectrometer (Bio-Rad, Richmond, VA, USA). UV spectra were recorded using the UV-2501PC spectromter (Shimadzu, Japan). ¹H and ¹³C-NMR spectra were obtained on a JNM-ECS400 MHz spectrometer (JEOL, Tokyo, Japan) and a Varian UNITY INOVA 600 spectrometer (Varian, Palo Alto, CA, USA), and the chemical shifts were given on δ (ppm) scale with TMS as an internal standard. The HR-ESI-MS were recorded on a 9.4-TQ-FT-MS Apex Qe (Bruker Co., Billerica, MA, USA). Silica gel (60-120 mesh, 200-300 mesh, Qingdao Marine Chemical Group Co., Qingdao, China), and Sephadex LH-20 (Pharmacia, Uppsala, Sweden) were employed for column chromatography. HPLC was carried out using Waters 600E system (Waters, Milford, MA, USA): an analytical column, ODS (5 µm, 4.6 × 250 mm, Hanbon Science & Technology Co., Ltd, Huaian, China), preparative column, a YMC C18 (5 μ m, 20.0 \times 250 mm, YMC, Kyoto, Japan), detector, Alltech ELSD (evaporative lightscattering detector, Alltech, Los Angeles, CA, USA) 2000ES. Flash chromatography was carried out on Teledyne ISCO Combi Flash Rf with Prepacked 80 g silica gel (200-300 mesh) columns (Teledyne Isco, Lincoln, NE, USA). TLC was carried out using silica gel 60 (>230 mesh, Qingdao Marine Chemical Group Co.) and GF254 plates precoated with silica gel 60. Spots on TLC were visually observed under UV light and/or by spraying with anisaldehyde-H₂SO₄ reagent followed by heating.

3.2. Plant Material

The dry roots of *V. officinalis* var. *latifolia* were collected from the Jiangkou region of Guizhou Province, China, in April 2012. The plant was identified by Prof. Bin Li (Beijing Institute of Radiation Medicine), and a voucher specimen (KYXC-20120313) is deposited in the herbarium of the Beijing Institute of Radiation Medicine, Beijing, China.

3.3. Extraction and Isolation

The air-dried roots of *V. officinalis* var. *latifolia* (50 kg) were exhaustively refluxed three times with 60% EtOH (400 L) to give a residue (11 kg) after removal of solvent under reduced pressure. The EtOH extract was suspended in H₂O and then partitioned successively with CHCl₃ (3×10 L). The CHCl₃ extract (152.2 g) was subjected to silica gel (200–300 mesh) column chromatography, eluted with petroleum ether-acetone (from 100:1, 75:1, 50:1, 30:1, 25:1, 20:1, 15:1, 10:1, 7:1, 5:1, 3:1, 2:1 and 1:1, v/v) to afford thirteen fractions (A–M). Fraction E (4.711 g) was subjected to flash silica gel chromatography column (80 gram flash column, 60 m L/min) with CHCl₃/CH₃OH (from 50:1 to 10:1) to yield eight fractions, E1-E8. Fractions E2-E3 (0.225g) was chromatographed by Sephadex LH-20 (2×120 cm, CHCl₃/CH₃OH, 1:1) to obtain compound **1** (33 mg). Fractions E4-E7 (0.818 g) was subjected to flash silica gel chromatography (40 gram flash column, petroleum ether-acetone, 5:1, 30 mL/min) and purified by Sephadex LH-20 (2×120 cm, CHCl₃/CH₃OH, 1:1) to afford compound **6** (86 mg). Compound **4** (168 mg) was isolated from fraction E8 by a series of repeated Sephadex LH-20 (2×150 cm, CHCl₃-CH₃OH, 1:1) column chromatography fractionations. Fraction I (3.678 g) was separated

chromatographically on flash silica gel column (80 gram flash column, 60 mL/min) with CHCl₃/CH₃OH (40:1 to 2:1), and a total of 50 tubes (15 mL each) were collected. Tubes 19–40 (2.245 g) were chromatographed by flash silica gel chromatography (40 gram flash column, petroleum ether-acetone, 4:1, 30 mL/min) to obtain compound **8** (tubes 10-13, 78 mg). Fraction J (6.366 g) was subjected to a series of purification steps using flash silica-gel column chromatography (80 gram flash column, CHCl₃/CH₃OH, 20:1 to 1:1, 60 mL/min) to give ten fractions (J1-J8). Fraction J2-J4 (0.167 g) was chromatographed by Sephadex LH-20 column chromatography (2 × 150 cm, CH₃OH), and purified by preparative HPLC (CH₃OH/H₂O, 75:25, flow rate: 2.0 mL·min⁻¹) to afford compounds **2** (68 mg) and **5** (35 mg). Compounds **3** (31 mg) and **7** (78 mg) were obtained from fraction J6-J8 (0.167 g) by preparative HPLC (CH₃OH-H₂O, 50:50, flow rate: 2.0 mL·min⁻¹).

3.4. Compound Characterization

Volvalerenal F (1): colorless oil. $[\alpha]_{D}^{20}$ +20.0 (c 0.8, CHCl₃); UV (CHCl₃) λ_{max} 237 nm; IR (film) v_{max} 3445, 3171, 2960, 2924, 2852, 1730, 1627, 1261, 1095, 1024 cm⁻¹; ¹H-NMR (CHCl₃, 600 MHz) data, see Table 1; ¹³C-NMR (CHCl₃, 150 MHz) data, see Table 1; HR-ESI-MS *m/z* 271.1678 [M+Na]⁺ (calcd. for C₁₆H₂₄O₂Na, 271.1669).

Volvalerenal G (**2**): colorless oil. $[\alpha]_{D}^{20}$ +53.9 (c 0.1, MeOH); UV (CHCl₃) λ_{max} 264 nm; IR (film) v_{max} 3382, 2933, 2864, 1618, 1298, 1190, 1072, 921, 793 cm⁻¹; ¹H-NMR (CHCl₃, 600 MHz) data, see Table 1; ¹³C-NMR (CHCl₃, 150 MHz) data, see Table 1; HR-ESI-MS *m/z* 235.1696 [M+H]⁺ (calcd. for C₁₅H₂₃O₂, 235.1693).

Volvalerenic Acid D (**3**): colorless oil. $[\alpha]_{D}^{20}$ +8.5 (c 0.47, MeOH); UV (CHCl₃) λ_{max} 240 and 271 nm; IR (film) v_{max} 3384, 3245, 2931, 2862, 1238, 1118, 1027, 862, 768 cm⁻¹; ¹H-NMR (CHCl₃, 600 MHz) data, see Table 1; ¹³C-NMR (CHCl₃, 150 MHz) data, see Table 1; HR-ESI-MS *m/z* 293.1744 [M+H]⁺ (calcd. for C₁₇H₂₅O₂, 293.1747).

3.5. Activity Screening in Vitro

PC 12D cell line was obtained from Insitute of Biochemistry and Cell Biology, CAS. It was cultured in Dulbecco's modified Eagles Medium (DMEM, Gibco, New York, NY, USA) with 10% fetal calf serum (Gibco, New York, NY, USA), and 5% equine serum (Gibco), and then the cells were maintained at 37.0 °C in a humidified atmosphere which contained 6% CO₂ [25]. The test cell line was seeded in 24-well culture paltes (2×10^4 cells/well) coated with poly-L-lysine (Gibco). After 24 h, the medium was changed to test medium containing 1% fetal calf serum, 2% equine serum and varying concentrations of NGF (50 ng/mL for positive control, 2 ng/mL for test samples and significant difference control, Sigma, St. Louis, MO, USA) and test compounds 1–8 (10, 30, 100 µmol). After incubating for 48 h, the cells were fixed with 1% glutaraldehyde (Sigma) in phosphate buffer, and the cells with neurites outgrowth were counted (with at least 100 cells examined/viewing area, three viewing areas/well, six wells/sample). The ratio of the neurite-bearing cells to total cells was determined and expressed as a percentage.

4. Conclusions

Three new germacrane-type sesquiterpenoids, volvalerenal F (1), volvalerenal G (2) and volvalerenic acid D (3), along with five known compounds 4-8, were isolated from the CHCl₃ soluble partition of the

ethanol extract of *Valeriana officinalis* var. *latiofolia*. NGF plays a key role in the functions of the central and peripheral nervous system [29] and all the sesquiterpenoids obtained displayed certain NGF-potentiating activities. From the current investigation it can be predicted that sesquiterpenoids will be promising candidates for dietary supplements and medicines, although further studies are needed to determine the pharmacological activities and the mechanism of these eight compounds in animals.

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Conflicts of Interest

The authors declare no conflict of interest.

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Sample Availability: Samples of the compounds 1–8 are available from the authors.

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