

OPEN

Self-assembling behavior and interface structure in vertically aligned nanocomposite $(\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3)_{1-x}:(\text{CeO}_2)_x$ films on (001) $(\text{La},\text{Sr})(\text{Al},\text{Ta})\text{O}_3$ substrates

Shao-Dong Cheng^{1,2}, Lu Lu², Sheng Cheng², Lv Kang Shen², Ming Liu², Yan Zhu Dai^{1,2}, Sheng-Qiang Wu¹ & Shao-Bo Mi^{1*}

Heteroepitaxial oxide-based nanocomposite films possessing a variety of functional properties have attracted tremendous research interest. Here, self-assembled vertically aligned nanocomposite $(\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3)_{1-x}:(\text{CeO}_2)_x$ ($x = 0.2$ and 0.5) films have been successfully grown on single-crystalline (001) $(\text{La},\text{Sr})(\text{Al},\text{Ta})\text{O}_3$ substrates by the pulsed laser deposition technique. Self-assembling behavior of the nanocomposite films and atomic-scale interface structure between $\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3$ matrix and CeO_2 nanopillars have been investigated by advanced electron microscopy techniques. Two different orientation relationships, $(001)[100]_{\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3} // (001)[1-10]_{\text{CeO}_2}$ and $(001)[100]_{\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3} // (110)[1-10]_{\text{CeO}_2}$, form between $\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3$ and CeO_2 in the $(\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3)_{0.8}:(\text{CeO}_2)_{0.2}$ film along the film growth direction, which is essentially different from vertically aligned nanocomposite $(\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3)_{0.5}:(\text{CeO}_2)_{0.5}$ films having only $(001)[100]_{\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3} // (001)[1-10]_{\text{CeO}_2}$ orientation relationship. Both coherent and semi-coherent $\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3/\text{CeO}_2$ interface appear in the films. In contrast to semi-coherent interface with regular distribution of interfacial dislocations, interface reconstruction occurs at the coherent $\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3/\text{CeO}_2$ interface. Our findings indicate that epitaxial strain imposed by the concentration of CeO_2 in the nanocomposite films affects the self-assembling behavior of the vertically aligned nanocomposite $(\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3)_{1-x}:(\text{CeO}_2)_x$ films.

Complex oxide-based nanocomposite films have attracted considerable research interest due to a variety of functional properties, such as multiferroicity¹, superconductivity²⁻⁴, ferromagnetism⁵, and ferroelectricity⁶. The nanocomposite films can be achieved in laminated structure⁶⁻⁸, vertical nanocomposite architecture^{1-3,5,9}, and three-dimensional framework constructed by interlayering the both^{4,10}. Simultaneous phase separation and strain-driven self-assembly processes were first shown to result in vertically aligned nanocomposite (VAN) films related to multiferroic applications and high-temperature superconductor applications¹⁻⁴. Importantly, novel and unprecedented properties could occur in the resultant VAN films, which are not present in the individual phases of the VAN films, e.g., multiferroic in the $\text{BaTiO}_3\text{-CoFe}_2\text{O}_4$ nanostructures¹ and enhanced flux-pinning in $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$ films incorporating self-aligned BaZrO_3 nanodots and nanorods²⁻⁴. Moreover, it was found that the self-assembling behavior and the physical properties of the VAN films can be influenced by the phase composition^{9,11} and the film growth parameters (e.g., growth temperature^{12,13}, deposition frequency¹⁴, and substrate termination¹⁵). In fact, the molar ratio (x) of two immiscible phases influences the epitaxial strain of the nanocomposite films on the substrates. For example, the molar ratio of MgO in the nanocomposite $(\text{LiFe}_5\text{O}_8)_{1-x}:(\text{MgO})_x$ films can tune the structure of LiFe_5O_8 nanopillar arrays prepared on fluorophlogopite substrates. As a result, the $(\text{LiFe}_5\text{O}_8)_{1-x}:(\text{MgO})_x$ VAN films exhibit a higher saturation magnetization (M_s), small nonzero coercivity and nonzero remanence compared with the pure LiFe_5O_8 film¹¹. In addition, the changing of the molar ratio of MgO in the $(\text{La}_{0.7}\text{Ca}_{0.3}\text{MnO}_3)_{1-x}:(\text{MgO})_x$ nanocomposite films on MgO (001) substrates can cause phase transition of $\text{La}_{0.7}\text{Ca}_{0.3}\text{MnO}_3$ from an orthorhombic ($0 < x \leq 0.1$) to a rhombohedral structure ($0.33 \leq x \leq 0.8$)⁹.

¹State Key Laboratory for Mechanical Behavior of Materials, Xi'an Jiaotong University, Xi'an, 710049, China. ²School of Microelectronics, Xi'an Jiaotong University, Xi'an, 710049, China. *email: shaobo.mi@xjtu.edu.cn

Additionally, it is believed that two-phase boundaries in the VAN film are of great importance, which provide large vertical interfacial areas and thus induce the coupling effect between the two immiscible phases^{16,17}. From this aspect, the VAN films possess superior properties over the single-phase films, e.g., low dielectric loss in VAN BiFeO₃:Sm₂O₃ films¹⁶. To better understand the performance of the VAN films, it is necessary to explore the self-assembling behavior of the nanocomposite films and interface structure between two immiscible phases at the atomic scale.

The perovskite-type manganites (e.g., Ln_{1-x}Ba_xMnO₃ (Ln = La and Pr)) exhibit a wealth of fascinating physical properties and potential practical applications^{18,19}. In particular, half-doped Pr_{0.5}Ba_{0.5}MnO₃ (PBMO) shows fantastic magnetic behaviors and excellent mixed ionic/electronic conductivity, which enable its potential applications in spintronic devices and solid oxide fuel cell^{20,21}. Recently, the (PBMO)_{1-x}:(CeO₂)_x nanocomposite films were successfully fabricated, and enhanced magnetic properties (e.g., magnetoresistance and magnetization) were obtained in VAN (PBMO)_{1-x}:(CeO₂)_x films compared with the pure PBMO films^{22,23}. Furthermore, the microstructure (e.g., lattice mismatch and crystallographic orientation relationship (OR) between PBMO and CeO₂) of semi-coherent PBMO/CeO₂ interface in the (PBMO)_{0.65}:(CeO₂)_{0.35} film was presented²³. Nevertheless, atomic-scale structure and chemical composition of the PBMO/CeO₂ interface, and strain relaxation behavior of the (PBMO)_{1-x}:(CeO₂)_x films on (001)-oriented (La,Sr)(Al,Ta)O₃ (LSAT) substrates have not been investigated. In addition, the effect of epitaxial strain on the self-assembling growth of the nanocomposite films remains unclear in the (PBMO)_{1-x}:(CeO₂)_x/LSAT heterosystem.

It is known that the self-assembling growth of VAN films can be accomplished by tuning the epitaxial strain imposed by changing the molar ratio (*x*) of two immiscible phases in the VAN films^{9,11}. In the present contribution, to deeply understand the self-assembling behavior and the related structure-property in the (PBMO)_{1-x}:(CeO₂)_x nanocomposite films, the (PBMO)_{1-x}:(CeO₂)_x (*x* = 0.2 and 0.5) films have been prepared on LSAT (001) substrates. We focus our research interest on film-growth behaviors and heterointerface structure investigated by advanced electron microscopy techniques.

Results and Discussion

A low-magnification bright-field (BF) TEM image of (PBMO)_{0.5}:(CeO₂)_{0.5} film and (PBMO)_{0.8}:(CeO₂)_{0.2} film on LSAT substrates is displayed in Figs. 1a,b, respectively, viewed along the [100] LSAT zone axis. The thickness of the films is about 130 nm and the film-substrate interface is sharp, as indicated by a horizontal white arrow. No misfit dislocations are observed at the interface of the nanocomposite films on the LSAT substrates. Moreover, CeO₂ nanopillars embedded in PBMO matrix with a width of about 10–15 nm can be recognized, as indicated by a vertical yellow arrow in Figs. 1a,b. It is noted that narrow CeO₂ nanopillars with straight sidewalls only exist in the (PBMO)_{0.8}:(CeO₂)_{0.2} film, as demonstrated by a vertical red arrow in Fig. 1b. In most cases, the CeO₂ nanopillars penetrate the whole film.

In Figs. 1a,b, the inserted SAED pattern taken from the nanocomposite film show the intense and sharp diffraction spots, indicating high-quality epitaxy of the film. The diffraction spots from PBMO matrix can be indexed, as shown by a green open square. The diffraction spots from CeO₂ nanopillars are visible, as indicated by a vertical yellow arrow. In comparison, one additional set of diffraction spots of CeO₂ nanopillars appears in the SAED pattern of the (PBMO)_{0.8}:(CeO₂)_{0.2} film, as indicated by a horizontal red arrow in the insert in Fig. 1b. On the basis of the SAED patterns, both (PBMO)_{0.5}:(CeO₂)_{0.5} and (PBMO)_{0.8}:(CeO₂)_{0.2} film have an OR of (001)[100]_{PBMO}//(001)[110]_{CeO2} (OR-I) between CeO₂ and PBMO. Apart from the OR-I, the OR of (001)[100]_{PBMO}//(110)[110]_{CeO2} (OR-II) between CeO₂ and PBMO exists in the (PBMO)_{0.8}:(CeO₂)_{0.2} film. In fact, there is a rotation of 90° between OR-I and OR-II (See Fig. S1 of the Supplemental Material).

Fig. 1c displays a typical high-resolution HAADF-STEM image viewed along the [100] PBMO zone axis, which shows the existence of two types of OR between CeO₂ and PBMO in the nanocomposite films. It is known that under the HAADF imaging conditions, the atomic columns appear dots in a dark background, and the intensity (*I*) of bright dots is roughly proportional to the square of the atomic number (*Z*) of the atom column²⁴. The CeO₂ nanopillars have a bright contrast in the PBMO matrix. It is found that CeO₂/PBMO interface can be either semi-coherent or coherent along the film-growth direction, as shown by yellow dashed lines and by red dashed lines, respectively. Interfacial dislocations are visible at the semi-coherent interface, as demonstrated by a horizontal yellow arrow.

Fig. 2a shows a high-resolution HAADF-STEM image of the (PBMO)_{0.5}:(CeO₂)_{0.5} nanocomposite film on the LSAT substrate, viewed along the [100] LSAT zone axis. A horizontal white arrow denotes the film-substrate interface. It is found that the coherent growth of PBMO film on LSAT substrate occurs, and the relatively small lattice mismatch (0.7%) between PBMO (*a*_{PBMO} = 0.3895 nm²⁵) and LSAT (*a*_{LSAT} = 0.3868 nm²⁶) is accommodated by the lattice elastic energy. In addition, the CeO₂ nanopillar epitaxially grows directly on the LSAT substrate with (001)[110]_{CeO2}//(001)[100]_{LSAT}, which may be due to the small lattice mismatch (−1.1%) calculated by $\Delta f = \frac{\sqrt{2}a_{\text{CeO}_2} - 2a_{\text{LSAT}}}{2a_{\text{LSAT}}}$, where *a*_{CeO₂} and *a*_{LSAT} are the lattice parameter of CeO₂ (*a*_{CeO₂} = 0.5411 nm²⁷) and LSAT, respectively. In contrast, the PBMO/CeO₂ interface is semi-coherent and interfacial dislocations are observed, as demonstrated by horizontal red arrows.

The atom-scale structure of the semi-coherent PBMO/CeO₂ interface has been investigated by EDS element mapping²⁸. Fig. 2b is a typical high-resolution HAADF-STEM image of the PBMO/CeO₂ interface. The corresponding EDS maps of Mn, Ba, Ce and Pr are shown in Figs. 2c–f, respectively. In the PBMO matrix, Pr and Ba cations site at the same atomic columns, indicating that A-site disordered PBMO is obtained. According to the EDS measurements, no elemental segregation at the PBMO/CeO₂ interface. In the CeO₂ nanopillar, Pr and Ce site at the same atomic columns, implying that Pr³⁺ ions dope into CeO₂ and partially replace Ce⁴⁺ ions. The substitution of Pr³⁺ in Ce⁴⁺ can result in the formation of (Ce,Pr)O_{2-δ} and oxygen vacancies generated in the (Ce,Pr)O_{2-δ}.

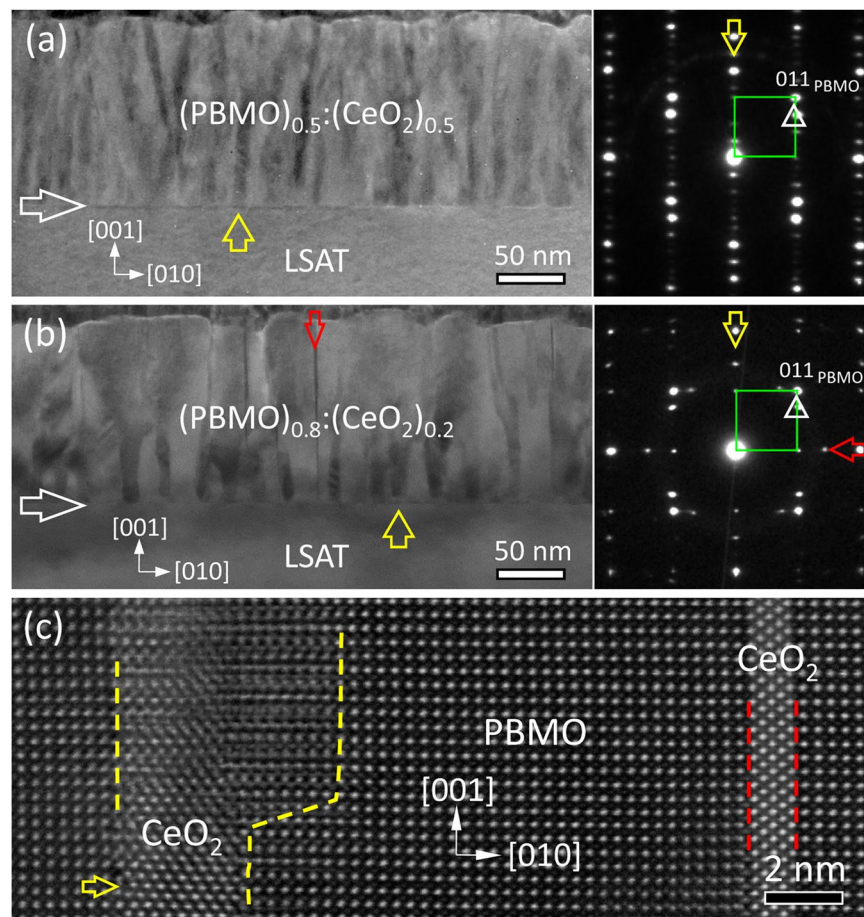


Figure 1. A low magnification BF-TEM image of $(\text{PBMO})_{0.5}:(\text{CeO}_2)_{0.5}$ film (a) and $(\text{PBMO})_{0.8}:(\text{CeO}_2)_{0.2}$ film (b) on LSAT substrate, recorded along the $[100]$ LSAT zone axis. The corresponding SAED pattern of the nanocomposite film is inserted. A horizontal white arrow in (a,b) indicates the film-substrate interface. CeO_2 nanopillars embedded in PBMO matrix with different dimensions are indicated by a vertical yellow arrow and a vertical red arrow, respectively. In the inserted SAED pattern, the diffraction spots from PBMO matrix are indicated by a green open square. Two sets of diffraction spots from CeO_2 nanopillars are denoted by a vertical yellow arrow and a horizontal red arrow, respectively. (c) A high-resolution HAADF-STEM image of $(\text{PBMO})_{0.8}:(\text{CeO}_2)_{0.2}$ film showing two types of ORs between CeO_2 and PBMO, viewed along the $[100]$ PBMO zone axis. The interfaces is indicated by yellow dashed lines and red dashed lines.

phase retain the charge balance. The reduced ratio of Pr/Ba in PBMO and oxygen vacancies in the $(\text{Ce,Pr})\text{O}_{2-\delta}$ can influence the transport and magnetic properties of the nanocomposite films^{29,30}.

Apart from the OR-I between CeO_2 and PBMO, CeO_2 nanopillars with the OR-II in PBMO matrix exist in the $(\text{PBMO})_{0.8}:(\text{CeO}_2)_{0.2}$ nanocomposite film. Fig. 3a shows a typical high-resolution HAADF-STEM image of the nanocomposite $(\text{PBMO})_{0.8}:(\text{CeO}_2)_{0.2}$ film on the LSAT substrate with the OR-II, viewed along the $[100]$ LSAT zone axis. The PBMO/ CeO_2 interface is denoted by a red curved dashed line. In contrast to the coherent PBMO/ CeO_2 interface along the film-growth direction, the lateral PBMO/ CeO_2 interface is semi-coherent. Interfacial dislocations are observed, as indicated by vertical yellow arrows. It should be noted that CeO_2 nanopillars do not grow directly on LSAT substrate. A large lattice mismatch between CeO_2 and LSAT may result in the difficulty in nucleating CeO_2 nanopillars on LSAT substrate with $(110)[1\bar{1}0]_{\text{CeO}_2} // (001)[100]_{\text{LSAT}}$.

Atomic-resolution HAADF- and ABF-STEM image of the coherent PBMO/ CeO_2 interface is displayed in Figs. 3b,c, respectively, recorded in the same region simultaneously and viewed along the $[100]$ PBMO zone axis. Under the HAADF- and ABF-STEM imaging conditions, different atomic columns including cations and oxygen can be identified at the interface^{24,31}. The interfacial CeO_2 layer is indicated by a vertical yellow arrow and the terminated layer of the PMBO film is indicated by a vertical red arrow. The contrast of oxygen atoms is visible in the terminated layer in Fig. 3c, as denoted by a horizontal purple arrow. Based on the HAADF- and ABF-STEM observations, the PBMO film terminates at the MnO_2 layer at the PBMO/ CeO_2 interface.

The structure of the coherent PBMO/ CeO_2 interface has been further examined by atomic-resolved EDS and EELS mapping^{32,33}. Figs. 3d–h show a typical coherent PBMO/ CeO_2 interface and the corresponding EDS map of element Mn, Ba, Ce and Pr, respectively. The EDS measurement in Figs. 3g,h indicates that Pr and Ce ions occupy the same site in CeO_2 , which is further confirmed by EELS measurement, as shown in the EELS maps

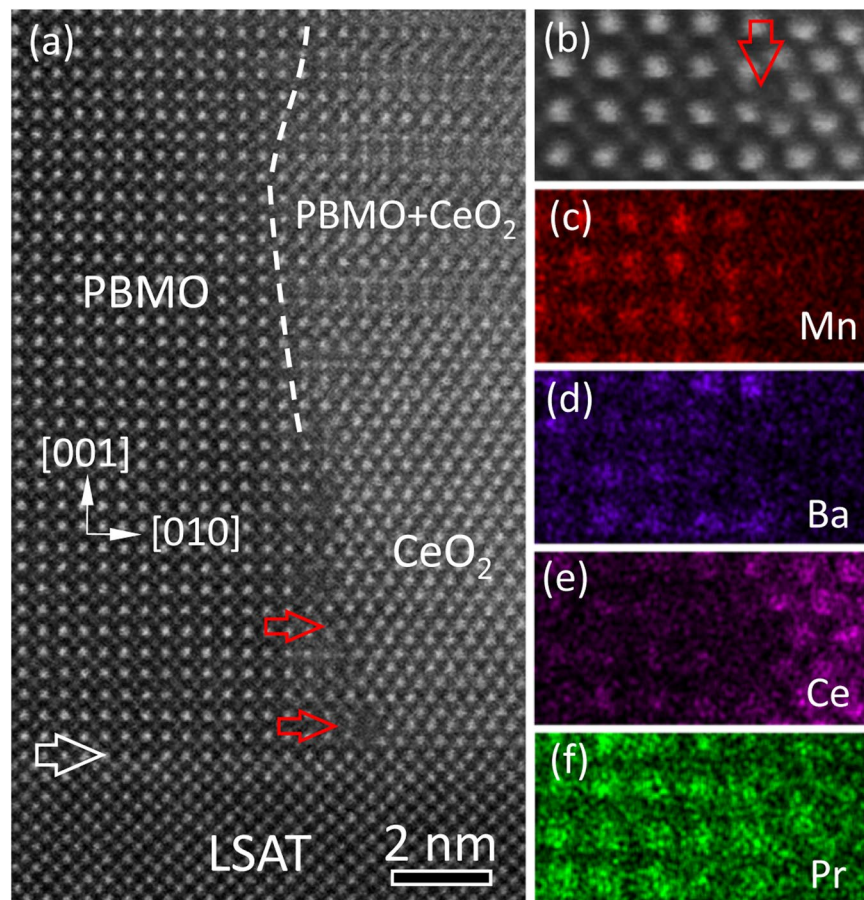


Figure 2. (a) An atomic-resolution HAADF-STEM image of the nanocomposite film on LSAT substrate, viewed along [100] LSAT zone axis. A horizontal white arrow indicates the film-substrate interface. A white dashed curved line shows the PBMO/CeO₂ interface. Interfacial dislocations at the interface are demonstrated by horizontal red arrows. A typical high-resolution HAADF-STEM image of the semi-coherent PBMO/CeO₂ interface (b) and the corresponding EDS maps of Mn-K α 1 (c), Ba-L α 1 (d), Ce-L α 1 (e) and Pr-L α 1 (f), respectively. A vertical red arrow in (b) denotes the interface.

of Ce-M_{4,5} and Pr-M_{4,5} in Figs. 3i,j. At the interface, CeO₂ atomic plane faces MnO₂ atomic plane of PBMO. The atomic plane indicated by vertical purple arrows and vertical green arrows in Fig. 3d is BaO and PrO, respectively, which have different intensities from other (Pr_{0.5}Ba_{0.5})O planes in PBMO as shown in Figs. 3f,h,j. In other words, interface reconstruction occurs at the PBMO/CeO₂ interface, resulting in the formation of a single unit-cell thickness of A-site ordered PBMO structure. It is worth mentioning that the distortion of MnO₆ octahedra is different between A-site ordered and disordered PBMO³⁴. In addition, the A-site ordered PBMO occurs a ferromagnetic-paramagnetic transition at about 320 K, while A-site disordered PBMO has $T_C \approx 140$ K²⁵.

It is worth noting that in our work the VAN (PBMO)_{1-x}:(CeO₂)_x ($x = 0.2$ and 0.5) films coherently grow on the LSAT substrates. For the CeO₂ nanopillars embedded in the PBMO matrix with the OR-I, with the change of the molar ratio (x) of CeO₂ to PBMO, the strain of the VAN (PBMO)_{1-x}:(CeO₂)_x films can be estimated by $\Delta f = \frac{(1-x)a_{\text{PBMO}} + x\frac{a_{\text{CeO}_2}}{\sqrt{2}}}{a_{\text{LSAT}}} - 1$, as shown by a red line in Fig. 4. It can be seen that the strain between the VAN (PBMO)_{1-x}:(CeO₂)_x films and the LAST substrates is close to zero while the molar ratio of CeO₂ is about 0.39. In addition, the epitaxial strain of VAN (PBMO)_{0.5}:(CeO₂)_{0.5} film and VAN (PBMO)_{0.8}:(CeO₂)_{0.2} has an opposite sign. With the reduction of the molar ratio of CeO₂ to PBMO, the compressive strain of the nanocomposite films increases, as indicated by a horizontal green arrow.

In the VAN (PBMO)_{0.8}:(CeO₂)_{0.2} film, CeO₂ nanopillars do appear in the PBMO matrix with either OR-I or OR-II. For CeO₂ embedded in PBMO matrix with the OR-II, the CeO₂ nanopillars have few CeO₂ unit cells in width. Compared with the A-site disordered PBMO, the A-site ordered PBMO at the PBMO/CeO₂ interface leads to the reduction of lattice parameter (1.8%)²⁵, which can partially release the epitaxial strain of the VAN (PBMO)_{0.8}:(CeO₂)_{0.2} film³⁵. Additionally, semi-coherent PBMO/CeO₂ interface with interfacial dislocations appears along the in-plane direction, as demonstrated in Fig. 3a, which can also relax the compressive strain of the VAN (PBMO)_{0.8}:(CeO₂)_{0.2} film on the LSAT substrate. In contrast, the VAN (PBMO)_{0.5}:(CeO₂)_{0.5} film undertakes the tensile strain on the LSAT substrate, as shown in Fig. 4. In the case of the appearance of CeO₂ nanopillars with the OR-II in PBMO matrix, the tensile strain of the nanocomposite film would be further increased, which

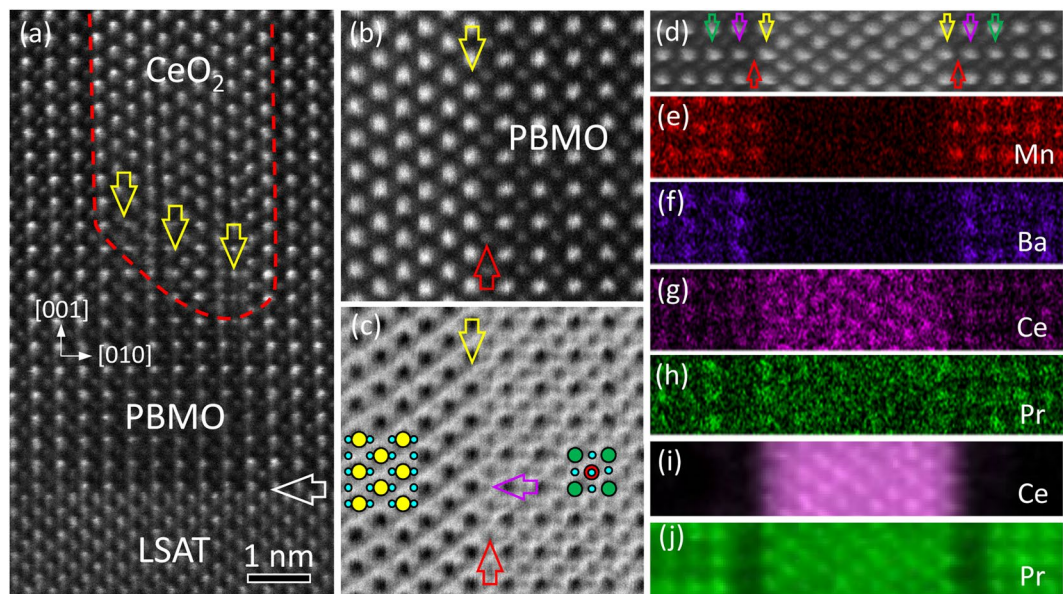


Figure 3. (a) A typical HAADF-STEM image of CeO₂ nanopillar embedded in PBMO matrix with the OR-II on LSAT substrate, viewed along [100] LSAT zone axis. A horizontal white arrow indicates the film-substrate interface. A red dashed curved line denotes the PBMO/CeO₂ interface. Interfacial dislocations are indicated by vertical yellow arrows. An atomic-resolution HAADF-STEM (b) and the corresponding ABF (c) image of the coherent PBMO/CeO₂ interface. Interfacial CeO₂ and MnO₂ layer is indicated by a vertical yellow and a vertical red arrow, respectively. The oxygen column in MnO₂ layer is shown by a horizontal purple arrow in (c). A typical HAADF-STEM image of coherent PBMO/CeO₂ interface (d) and the corresponding EDS maps of Mn-K α 1 (e), Ba-L α 1 (f), Ce-L α 1 (g) and Pr-L α 1 (h), respectively. (i,j) The corresponding EELS maps of Ce-M_{4,5} edge and Pr-M_{4,5} edge, respectively. A vertical green, purple, red and yellow arrow indicates the PrO, BaO, MnO₂ and Ce/Pr atom column at the interface, respectively.

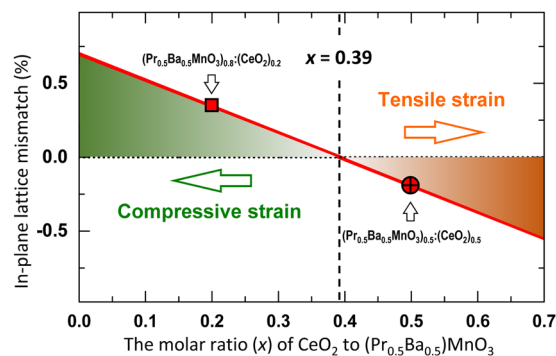


Figure 4. In-plane lattice mismatch between nanocomposite (PBMO)_{1-x}:(CeO₂)_x film and LSAT substrate as a function of the molar ratio (*x*) of CeO₂ to PBMO in the film for the existence of only OR-I between CeO₂ and PBMO in the nanocomposite film. The gradual increase of compressive (or tensile) strain is indicated by a horizontal green (or orange) arrow.

is in agreement with the experimental observations that no OR-II occurs between CeO₂ and PBMO in the VAN (PBMO)_{0.5}:(CeO₂)_{0.5} film.

Compare with A-site disordered PBMO, A-site ordered PBMO has a relative low *M_s* and high magnetoresistance at low temperatures²⁵. Nevertheless, considering a very small volume fraction (~20%) of the A-site ordered PBMO in the (PBMO)_{0.8}:(CeO₂)_{0.2} film, the magnetic properties (e.g., *M_s*) of the (PBMO)_{1-x}:(CeO₂)_x films on the LSAT substrates would be mainly affected by the epitaxial strain imposed by the CeO₂ nanopillars within the films²³. In other words, the volume fraction of CeO₂ and the crystallographic OR between CeO₂ and PBMO in the VAN films change the strain state and the magnetic properties of the PBMO film^{22,23}. In addition, it was found that the electrical resistivity of the VAN (PBMO)_{0.5}:(CeO₂)_{0.5} film is over 4 times larger than that of the pure PBMO film in our previous work²². It is believed that the vertical semi-coherent phase boundary can increase the difficulty of charge carriers transport, and result in an increase of resistivity of the film system^{9,36}. The appearance of A-site ordered PBMO at the coherent PBMO/CeO₂ interface could lead to a decrease of electrical

resistivity since A-site ordered PBMO has two orders lower electrical resistivity than A-site disordered PBMO²⁵. But, the A-site ordered PBMO in the (PBMO)_{0.8}:(CeO₂)_{0.2} film has a very small volume fraction, which could not strongly affect the resistivity of the VAN (PBMO)_{0.8}:(CeO₂)_{0.2} film²³. Importantly, our work demonstrates that the epitaxial strain can lead to the formation of A-site ordered PBMO at the heterointerface, which provides a strategy to fabricate A-site ordered PBMO thin films on the substrates (e.g., CeO₂/YSZ buffered Si substrates^{37,38}) by using strain engineering in the heterosystems.

In summary, the VAN (PBMO)_{1-x}:(CeO₂)_x films prepared on (001)-oriented LSAT substrates have been systematically studied by advanced electron microscopy. While the VAN film under tensile strain, an OR of (001)[100]_{PBMO}//(001)[110]_{CeO2} occurs between CeO₂ and PBMO in the film. In contrast, the VAN film under compressive strain contains two types of OR, (001)[100]_{PBMO}//(001)[110]_{CeO2} and (001)[100]_{PBMO}//(110)[110]_{CeO2} between CeO₂ and PBMO. In addition, interface reconstruction occurs at the coherent PBMO/CeO₂ interface, resulting in the formation of a single unit-cell-thick layer of A-site ordered PBMO at the interface. Our results demonstrate that self-assembling behavior of the nanocomposite (PBMO)_{1-x}:(CeO₂)_x films can be modulated by epitaxial strain.

Material and Methods

Thin film preparation. The composite targets of (PBMO)_{1-x}:(CeO₂)_x ($x = 0.2$ and 0.5) were sintered by a standard ceramic sintering method. The (PBMO)_{1-x}:(CeO₂)_x films were fabricated on (001) LSAT single-crystalline substrates by a KrF (wavelength $\lambda = 248$ nm) excimer pulsed laser deposition system with laser energy density of 2.0 J cm^{-2} at 3 Hz. During the film deposition, oxygen pressure is 250 mTorr and substrate temperature is 800 °C.

Thin film characterization. Cross-sectional transmission and scanning transmission electron microscopy (TEM/STEM) specimens were prepared by focused ion beam (FIB) milling (FEI Helios NanoLab 600i)³⁹. Diffraction contrast imaging, selected-area electron diffraction (SAED), high-angle annular dark-field (HAADF) and annular bright-field (ABF) imaging, energy dispersive X-ray spectroscopy (EDS) mapping and electron energy-loss spectroscopy (EELS) mapping were carried out on a probe aberration-corrected JEOL JEM-ARM200F equipped with an Oxford X-Max^N 100TLE spectrometer and a Gatan Enfina spectrometer, operated at 200 kV. In STEM mode, a probe size of 0.1 nm at semi-convergence angle of 22 mrad was used. HAADF and ABF detectors covered angular ranges of 90–176 and 11–22 mrad, respectively. EELS collection angle was 72 mrad and energy resolution was 1.2 eV at the dispersion of 0.3 eV/pixel.

Data availability

All data needed to evaluate the conclusions in the paper are present in the paper and/or the Supplementary Materials. Additional data related to this paper may be requested from the authors.

Received: 24 September 2019; Accepted: 23 January 2020;

Published online: 11 February 2020

References

- Zheng, H. *et al.* Multiferroic BaTiO₃-CoFe₂O₄ nanostructures. *Science* **303**, 661–663 (2004).
- Goyal, A. *et al.* Irradiation-free, columnar defects comprised of self-assembled nanodots and nanorods resulting in strongly enhanced flux-pinning in YBa₂Cu₃O_{7- δ} films. *Supercond. Sci. Technol.* **18**, 1533–1538 (2005).
- Kang, S. *et al.* High-performance high-Tc superconducting wires. *Science* **311**, 1911–1914 (2006).
- Wee, S. *et al.* Self-assembly of nanostructured, complex, multication films via spontaneous phase separation and strain-driven ordering. *Adv. Funct. Mater.* **23**, 1912–1918 (2013).
- Mohaddes-Ardabili, L. *et al.* Self-assembled single-crystal ferromagnetic iron nanowires formed by decomposition. *Nat. Mater.* **3**, 533–538 (2004).
- Schlom, D. G. *et al.* Oxide nano-engineering using MBE. *Mater. Sci. Eng., B* **87**, 282–291 (2001).
- Yadav, A. K. *et al.* Observation of polar vortices in oxide superlattices. *Nature* **530**, 198–201 (2016).
- Lu, L. *et al.* Topological defects with distinct dipole configurations in PbTiO₃/SrTiO₃ multilayer films. *Phys. Rev. Lett.* **120**, 177601 (2018).
- Moshnyaga, V. *et al.* Structural phase transition at the percolation threshold in epitaxial (La_{0.7}Ca_{0.3}MnO₃)_{1-x}:(MgO)_x nanocomposite films. *Nat. Mater.* **2**, 247–252 (2003).
- Chen, A. *et al.* Strain enhanced functionality in a bottom-up approach enabled 3D super-nanocomposites. *Adv. Funct. Mater.* **29**, 1900442 (2019).
- Lan, G. *et al.* Flexible Lithium Ferrite Nanopillar Arrays for Bending Stable Microwave Magnetism. *ACS Appl. Mater. Interfaces* **10**, 39422–39427 (2018).
- Jing, H.-M. *et al.* Microstructure and electrical conductivity of (Y,Sr)CoO_{3- δ} thin films tuned by the film-growth temperature. *J. Alloy. Compd.* **714**, 181–185 (2017).
- Zheng, H., Chu, Y.-H., Ramesh, R. & Salamanca-Riba, L. Heteroepitaxially enhanced magnetic anisotropy in BaTiO₃-CoFe₂O₄ nanostructures. *Appl. Phys. Lett.* **90**, 113113 (2007).
- Chen, A. *et al.* Microstructure, magnetic, and low-field magnetotransport properties of self-assembled (La_{0.7}Sr_{0.3}MnO₃)_{0.5}:(CeO₂)_{0.5} vertically aligned nanocomposite thin films. *Nanotechnology* **22**, 315712 (2011).
- Fan, M. *et al.* Self-organized epitaxial vertically aligned nanocomposites with long-range ordering enabled by substrate nanotemplating. *Adv. Mater.* **29**, 1606861 (2017).
- Yang, H. *et al.* Vertical interface effect on the physical properties of self-assembled nanocomposite epitaxial films. *Adv. Mater.* **21**, 3794–3798 (2009).
- MacManus-Driscoll, J. L. Self-assembled heteroepitaxial oxide nanocomposite thin film structures: designing interface-induced functionality in electronic materials. *Adv. Funct. Mater.* **20**, 2035–2045 (2010).
- Barnabé, A., Millange, F., Maignan, A., Hervieu, M. & Raveau, B. Barium-based manganites Ln_{1-x}Ba_xMnO₃ with Ln = {Pr, La}: phase transitions and magnetoresistance properties. *Chem. Mater.* **10**, 252–259 (1998).
- Ueda, Y. & Nakajima, T. Novel structures and electromagnetic properties of the A-site-ordered/disordered manganites RBaMn₂O₆/R_{0.5}Ba_{0.5}MnO₃ (R = Y and rare earth elements). *J. Phys.-Condens. Matter* **16**, S573–S583 (2004).
- Autret, C. *et al.* Magnetization steps in a noncharge-ordered manganite, Pr_{0.5}Ba_{0.5}MnO₃. *Appl. Phys. Lett.* **82**, 4746–4748 (2003).

21. Sengodan, S. *et al.* Layered oxygen-deficient double perovskite as an efficient and stable anode for direct hydrocarbon solid oxide fuel cells. *Nat. Mater.* **14**, 205–209 (2015).
22. Shen, L. *et al.* Enhanced magnetic properties in epitaxial self-assembled vertically aligned nanocomposite $(\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3)_{0.5}(\text{CeO}_2)_{0.5}$ thin films. *J. Mater. Chem. C* **4**, 10955–10961 (2016).
23. Cheng, S. *et al.* Structural transition induced enhancement of magnetization and magnetoresistance in epitaxial $(\text{Pr}_{0.5}\text{Ba}_{0.5}\text{MnO}_3)_{1-x}(\text{CeO}_2)_x$ vertical aligned thin films. *Cryst. Eng. Comm.* **20**, 5017–5024 (2018).
24. Mi, S. B., Jia, C. L., Vrejoiu, I., Alexe, M. & Hesse, D. Atomic-scale structure and properties of epitaxial $\text{PbZr}_{0.2}\text{Ti}_{0.8}\text{O}_3/\text{SrRuO}_3$ heterointerfaces. *Adv. Mater. Interfaces* **2**, 1500087 (2015).
25. Trukhanov, S. V. *et al.* Study of A-site ordered $\text{PrBaMn}_2\text{O}_{6-\delta}$ manganite properties depending on the treatment conditions. *J. Phys.-Condens. Matter* **17**, 6495–6506 (2005).
26. Pawlak, D. A. *et al.* Structure and spectroscopic properties of $(\text{AA}')(\text{BB}')\text{O}_3$ mixed-perovskite crystals. *J. Mater. Res.* **20**, 3329–3337 (2005).
27. Montini, T., Melchionna, M., Monai, M. & Fornasiero, P. Fundamentals and catalytic applications of CeO_2 -based materials. *Chem. Rev.* **116**, 5987–6041 (2016).
28. Wu, S. Q. *et al.* B-site ordering and strain-induced phase transition in double-perovskite $\text{La}_2\text{NiMnO}_6$ films. *Sci. Rep.* **8**, 2516 (2018).
29. Bharathi, R. N. & Sankar, S. Structural, optical and magnetic properties of Pr doped CeO_2 nanoparticles synthesized by citrate–nitrate auto combustion method. *J. Mater. Sci.-Mater. Electron.* **29**, 6679–6691 (2018).
30. Jing, H. M. *et al.* Formation of Ruddlesden–Popper faults and their effect on magnetic properties in $\text{Pr}_{0.5}\text{Sr}_{0.5}\text{CoO}_3$ thin films. *ACS Appl. Mater. Interfaces* **10**, 1428–1433 (2018).
31. Mi, S.-B. *et al.* Atomic structures of planar defects in $0.95(\text{Na}_{0.5}\text{Bi}_{0.5})\text{TiO}_3-0.05\text{BaTiO}_3$ lead-free piezoelectric thin films. *J. Alloy. Compd.* **636**, 183–186 (2015).
32. Muller, D. A. *et al.* Atomic-scale chemical imaging of composition and bonding by aberration-corrected microscopy. *Science* **319**, 1073–1076 (2008).
33. Watanabe, M., Okunishi, E. & Aoki, T. Atomic-level chemical analysis by EELS and XEDS in aberration-corrected scanning transmission electron microscopy. *Microsc. Microanal.* **16**, 66–67 (2010).
34. Rondinelli, J. M. & Fennie, C. J. Octahedral rotation-induced ferroelectricity in cation ordered perovskites. *Adv. Mater.* **24**, 1961–1968 (2012).
35. People, R. & Bean, J. C. Calculation of critical layer thickness versus lattice mismatch for $\text{Ge}_x\text{Si}_{1-x}/\text{Si}$ strained-layer heterostructures. *Appl. Phys. Lett.* **47**, 322–324 (1985).
36. Chen, A. *et al.* Tunable low-field magnetoresistance in $(\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3)_{0.5}(\text{ZnO})_{0.5}$ self-assembled vertically aligned nanocomposite thin films. *Adv. Funct. Mater.* **21**, 2423–2429 (2011).
37. Yamada, T., Wakiya, N., Shinozaki, K. & Mizutani, N. Epitaxial growth of SrTiO_3 films on $\text{CeO}_2/\text{yttria-stabilized zirconia}/\text{Si}(001)$ with TiO_2 atomic layer by pulsed-laser deposition. *Appl. Phys. Lett.* **83**, 4815–4817 (2003).
38. Kim, H. *et al.* Self-assembled single-phase perovskite nanocomposite thin films. *Nano Lett.* **10**, 597–602 (2010).
39. Jing, H.-M. *et al.* Atomic-scale imaging of heterointerface and planar faults in epitaxial $(\text{Pr}, \text{Sr})_2\text{CoO}_4$ films on $\text{SrTiO}_3(001)$ substrates. *J. Cryst. Growth* **511**, 93–98 (2019).

Acknowledgements

The work was supported by the National Basic Research Program of China (No. 2015CB654903) and the National Natural Science Foundation of China (51471169 and 51390472).

Author contributions

S.C., L.S. and M.L. prepared the thin film materials, L.L. and Y.D. prepared the TEM samples, S.D.C., S.Q.W. and S.B.M. conducted the electron microscopy experiments and analysis, S.D.C., L.L. and S.B.M. wrote the manuscript. All authors reviewed the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

Supplementary information is available for this paper at <https://doi.org/10.1038/s41598-020-59166-1>.

Correspondence and requests for materials should be addressed to S.-B.M.

Reprints and permissions information is available at www.nature.com/reprints.

Publisher's note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.



Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons license, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons license, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons license and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this license, visit <http://creativecommons.org/licenses/by/4.0/>.

© The Author(s) 2020