

RESEARCH ARTICLE

REVISED Proximate and antioxidant activities of bio-preserved *ogi* flour with garlic and ginger [version 2; peer review: 2 approved]

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Abstract

Background: *Ogi* from locally available cereals remains a relatively affordable complementary food in West Africa, but has a tendency to spoil due it high moisture content. This study explored effects of garlic and ginger as biopreservatives in *ogi* flour.

Methods: *Ogi* flour was prepared from sorghum and quality protein maize grains with different concentrations of garlic and ginger powder (2 and 4% w/w) by fermentation technique. These samples were stored for 16 weeks during which the total titratable acidity, pH, proximate composition, mineral content and total antioxidant activities were determined.

Results: The proximate compositions of bio-preserved *ogi* samples were relatively stable throughout storage. The addition of garlic and ginger slightly increased the ash (0.04%), crude protein and mineral contents (mg/ 100g) of the samples. Magnesium (10.85-13.13 and 5.17-9.72); zinc (1.37-1.78 and 7.01-8.50), manganese (1.30-1.71 and 0.45-0.86) and iron (1.53-1.77 and 0.68-2.77) contents increased on addition (of garlic and ginger) to maize *ogi* and sorghum *ogi* flours respectively. The free radical scavenging activity; total phenolic and flavonoid contents increased correspondingly with the antioxidants activity.

Conclusion: Although not well known to *ogi* consumer, the bio-preserved ogi flours showed better nutritional values and have potential as a health food.

Keywords

Garlic, Ginger, Antioxidants, Quality, Ogi flour, Biopreservation

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REVISED Amendments from Version 1

In this version we corrected few typographical errors in the manuscript present continuous tense were used to describe the process of production under methods for *ogi* preparation co-fermented with garlic and ginger.

See referee reports

Introduction

Cereal grains have become the most important plant group in terms of the human diet¹. Sorghum is crucial in developing countries food security because it's one of the most important staple foods for millions of poor rural people in the semiarid tropics of Asia and Africa². Nigeria is the largest sorghum producer in 2016 in West and Central Africa region which accounts for approximately 23% of its production in Africa with forecast indicating it will become the largest sorghum grain producer in the world by 2020^{3,4}. Sorghum is rich in fats, protein, fiber, and minerals such as potassium, phosphorus, iron and calcium². Sorghum is gluten-free, and therefore a suitable alternative grain for people with gluten intolerance⁵. Maize and sorghum are important cereal crops in Africa and Asia, and are consumed in various ways (porridges, snacks etc.)⁶. They are the main staples of the majority of the Nigerian populace. Maize contains natural bioactive chemical compounds such as carotenoids and phenolic compounds⁷. Quality protein maize (QPM) is nutritionally superior over the normal maize due to balanced amounts of essential amino acids, with a high content of lysine and tryptophan, and low content of leucine and isoleucine⁸. Quality protein maize contains the optimal amount of amino acids in protein intake when compared with the amino acid composition to egg protein⁹. Replacement of normal maize with quality protein maize (QPM) will impart better nutritional value to the consumers, due to its higher tryptophan (55%) and lysine (30%) content compared to normal maize. This will also contribute to food and nutrition security of the poor communities, and improve linear growth in weaning children by 19.3% where maize is consumed as staple food¹. Ogi is a thin gruel commonly used as a breakfast cereal and for infant weaning food in West Africa because it is readily available, cheap, and can be produced at household level. It can be prepared by fermentation of maize, sorghum and millet^{10,11}. Ogi in paste form has the tendency to spoil because of its high moisture content¹². Garlic and ginger can be used biopreservative due to their antibacterial and antifungal properties to extend the shelf life of food¹³. Garlic is found almost all over the world, and is an important herb which is now an integral part of human diet and has also been linked to health benefit such as anticancer, antioxidant, therapeutic effect, stimulation of digestion, and absorption of food^{14,15}. Ginger is also widely used around the world in food as a spice. Both are generally regarded as safe (GRAS) for consumption in food¹⁶. There is need to preserve ogi with naturally available spices, such as garlic and ginger, that are widely used and available. The study focused on the effect of garlic and ginger on the nutritional quality of ogi flour based on the proximate and mineral composition, in addition to its antioxidant activities during storage,

with the aim of preserving and enhancing the nutritional level.

Methods

Materials

Chemicals and materials used in this study were of analytical grade namely; 1,1-diphenyl-2-picrylhydrazyl radical (Sigma Aldrich, Germany, D9132), Gallic acid (Sigma Aldrich, Germany, G7384), Quercetin (Sigma Aldrich, Germany), Folin Ciocalteau phenol reagent (Sigma Aldrich, Germany), Folin Ciocalteau phenol reagent (Sigma Aldrich, Germany, F9252), Aluminum chloride (BDH, England,101084), potassium acetate (Sigma Aldrich, Germany, 791733-500G), sulfuric acid (38308-1EA), Ethanol (BDH, England, BDH1156-4LP), phenolphthalein (Sigma Aldrich, 74760-100ML), sodium hydroxide (Sigma Aldrich, 38227 1EA), hexane (Sigma Aldrich, Germany, C100307-2.5L) Boric acid (Sigma Aldrich, Germany, 382801EA), Whatman No.1 filter paper (28413923) supplied by Finlab Nigeria Limited and Equilab Business solution Limited Nigeria.

Powdered garlic and ginger preparation

Two hundred and fifty (250) grams of garlic bulbs and ginger rhizomes that were freshly harvested were washed, drained, peeled, diced into cubes, and dried at 65 °C for 12 h using hot air oven (Gallenkamp, UK). They were then ground using a grinder (Marlex Appliances PVT, Mumbai, India). Powdered garlic and ginger passed through a sieve (60 μ m) (BS mesh sieves, Dual manufacturing Co. Chicago, USA) for removal of residues¹³.

Ogi preparation co-fermented with garlic and ginger

Quality protein maize (ART/98/SW06/OB/W) was obtained from the Institute of Agricultural Research and Training (I.A.R.T.), Ibadan, Nigeria, while sorghum was procured from a local market in Ile - Ife, Osun State, Nigeria. 15 kilogram of grains were examined, winnowed, and steeped separately for 3 days. Millings into smooth paste after the grains were drained and this was done with an attrition mill (No 1 Premier mill, England). Powdered garlic and ginger were added for co-fermentation to smooth paste of maize/ sorghum at 2 and 4% (w/w) garlic or ginger singly and in combination (ginger-garlic), which resulted into 7 conditions. The samples were labeled A-H as follows: A: control samples (without garlic/ginger); B: *Ogi* + 2% Garlic; C: *Ogi* + 4% Garlic; D: Ogi + 2% Ginger; E: Ogi + 4% Ginger; F: Ogi + 2% Garlic + 2% Ginger; G: Ogi + 2% Garlic + 4% Ginger; H: Ogi + 4% Garlic + 2% Ginger. The slurry was evenly homogenized, then allowed to ferment spontaneously (naturally) at ambient temperature (27± 2°C) for 24 h. After fermentation the water was pressed in muslin cloth to form an ogi cake¹⁷.

Preparation of biopreserved ogi flour

Ogi cakes were dried for 48 h at $42\pm 2^{\circ}$ C with a cabinet dryer (Gallenkamp, UK), which was then grounded into flour, cooled for 5 min at room temperature, then packaged in a pouch and sealed. The packaged samples were stored at room temperature for 16 weeks during which samples were obtained for analysis at 4 week intervals (monthly). *Ogi* flour samples were then placed on a shelf for further analysis¹⁸.

Determination of titratable acidity and pH

The total titratable acidity (TTA) of ogi flour was determined for all samples to quantify the acid produced during sample storage. 1g ogi flour was reconstituted in 10 ml of distilled water. Three drops of phenolphthalein was added as indicator; then titrated against 0.1M NaOH while gently swirling the content in the conical flask until pink colour appeared. Each ml of 0.1N NaOH used was equivalent to 90.08 mg of lactic acid. Titration readings were taken in triplicate and mean values of the readings were calculated. Total titratable acidity of lactic acid (g/ml) was calculated. A pH meter (Corning Scholar 425, UL Laboratories, Shenzhen, China) was used to determine pH values of the 5g of reconstituted flour in 50 ml of distilled water. Buffer at pH 4.0 and 7.0 were used to calibrate the pH meter. The pH of all the samples were read after stabilization of the value on the apparatus screen, the pH values were recorded in triplicate and mean values of the reading was calculated¹⁹.

Determination of proximate composition and mineral content

Moisture content determination. 5 g of each sample was weighed in triplicate into pre-weighed moisture content cans. The samples were dried for 3 h at 105 °C in the Gallenkamp hot-air oven (Gallenkamp, UK) and the weight was taken. The drying continued until their weights were constant. The samples were cooled to room temperature in a desiccator and weighed. The final weight of each sample was determined¹⁹. The moisture content was calculated from weight loss equation below

Moisture content = $\frac{w2 - w3}{w1} \times 100$ (%)

 w_1 = Weight of sample before drying (g)

 w_2 = Weight of sample after drying (g)

Crude protein determination. 2 g of ogi sample was weighed into a digestion flask. Kjeltec catalyst 31835-2501AE (0.8 g) and 15 ml of concentrated sulphuric acid was added to each flask. Each flask was heated on pre heated digester set (K12, Behr LaborTechnik, Germany) at 420 °C in a fume cupboard, and digested until a clear homogenous mixture was obtained. After digestion, the flask was removed from the heater, cooled, and the content was diluted with 50 ml of distilled water. The flask was then placed in micro-kjedahl analyser (Kjelmaster K-375, Buchi, Switzerland) where it received 50 ml of NaOH automatically. The mixture was subsequently heated up to release ammonia which was distilled into a conical flask containing 25 ml of 2% (w/v) boric acid as an indicator for 4 min, the ammonia reacted with boric acid to form ammonium borate which was titrated against 0.1M hydrochloric (HCl) acid until the purplish - grey end point was attained. The percentage nitrogen content of the samples was calculated using the equation below:

Nitrogen =
$$\frac{A \times M \times 0.014}{weight of sample(g)} \times 100 \ (\% g)$$

where A = 0.1 HCl (ml)

Crude protein content was estimated by multiplying with the factor 6.25 (The protein content in food is estimated by multiplying the determined nitrogen content by a nitrogen-to-protein conversion factor 6.25 as the standard. AOAC, 2010). The experiment was carried out in triplicate and the means for each sample were recorded

Crude fat content determination. Fat content of all the ogi samples was determined by a continuous extraction liquid - solid method using soxhlet extractor with a reflux condenser and a distillation flask (E914, Buchi, Switzerland). Each sample (2 g) was weighed into a fat free thimble plugged with cotton wool and placed in the appropriate chamber of the extractor. The distillation flask was filled to two third capacities with n-hexane (60–80 boiling points); the flask was boiled on a heating mantle; the distillate was collected. Thereafter, n-hexane was recovered into a clean container until almost all had been distilled. The remaining solvent in the mixture was evaporated in a Gallenkamp hot-air oven (Gallenkamp, UK) set at 70 °C. The flask was allowed to cool subsequently in a desiccator (PYREX, Corning, Inc USA after which the final weight of the flask was determined. The difference in the final and initial weight of the distillation flask represented the oil extracted from the sample¹⁹.

The percentage of crude fat was obtained using the equation below:

$$Fat = \frac{Final weight of flask - initial weight of flask}{weight of sample(g)} \times 100 (\%)$$

The experiment was carried out in triplicate for each sample.

Crude fibre determination. The crude fibre was determined using the weighed samples resulting from fat extraction. Each sample was transferred into conical flask and 100 ml boiling 1.25% H₂SO₄ added. Each beaker was heated for 30 min with periodical rotation to prevent adherence of solids to the sides of the beakers. The solution was filtered using Whatman No.1 filter paper (28413923) and rinsed with 50 ml portions boiling water; repeated trice then dried. Boiling 1.25% (w/v) NaOH solution (200 ml) was added and the mixture was boiled for 30 min after which the contents of each beaker was removed and filtered; washed with 25 ml boiling 1% sulphuric acid, three portions of 50 ml boiling water and 25 ml ethanol. The residue was dried at 100 °C to a constant weight followed by cooling in a desiccator at room temperature and weighed. The weighed residue was ignited at 600 °C in a Gallenkamp muffle furnace (Gallenkamp, UK) for 30 min, cooled in a desiccator and reweighed¹⁹.

The percentage crude fibre in each sample was calculated as:

Crude fibre =
$$\frac{w2 - w3}{w1} \times 100$$
 (%)
W₁ = Weight of sample (g)
W₂ = Weight of crucible + sample (g)
W₃ = Weight of crucible + Ash (g)

The experiment was carried out in triplicate for each sample and the average calculated for each sample.

Total ash determination. The total ash (inorganic residue from the incineration of organic matter) was determined by dry ashing procedure. The samples (2 g) were weighed into a preweighed dry porcelain crucible. The samples were incinerated in a Gallenkamp muffle furnace (Gallenkamp, UK) at 550 °C for 6 h. After ashing, the remains were removed from the furnace, cooled to room temperature in a desiccator and weighed¹⁹. The porcelain crucible was weighed and the % total ash weight was obtained by using the equation below:

Total ash =
$$\frac{weight of ash(g)}{weight of sample} \times 100 (\%)$$

Carbohydrate determination. The carbohydrate content was determined by difference. The sum of the moisture, ash, crude fiber, fat and protein of the respective samples was subtracted from 100 to obtain percentage carbohydrate¹⁹.

Determination of mineral content

The amount of minerals present in the sample was determined as described by $AOAC^{19}$. The ash of the sample obtained was dissolved in 10ml of 2M HNO_3 and boiled for 5 min, filtered through Whatman No.1 filter paper into volumetric flask. The filtrate was made up with distilled water to 50 ml and used for determination of minerals content. Zinc (Zn), Manganese (Mn), Magnesium (Mg) and Iron (Fe) were determined by using Atomic Absorption Spectrophotometer (AAS 220GF, Buck). The standard curve for each mineral was prepared from known concentrations of mineral and the mineral content of the samples was estimated from the standard curve, while sodium and potassium content were determined using Jenway flame photometer PFP7 (Cole-palmer, UK).

Estimation of total phenolic content

Total phenolic content was determined using the modified procedure from Olaniran *et al.*¹⁶. Extracts (0.1 ml) of *ogi* flour samples were pippeted into 5.9 ml distilled water; afterward, 1.0 ml Folin Ciocalteau reagent was added to 1.0 ml of the diluted extract in test tubes. The mixture was left for 5 min before addition of 2 ml of 20% (w/v) Na₂CO₃. After 30 min of rigorous mixing was done with a vortex mixer, absorbance was taken at 725 nm using a spectrophotometer (Model SP9, PyeUnican UK). The results were expressed as Gallic acid equivalent (GAE) using a calibration curve with Gallic acid as standard (100 mg/ ml) y = 0.0022x - 0.0292, R² = 0.9962.

1, 1-diphenyl-2-picrylhyrazyl (DPPH) radical scavenging activity of *ogi* flour extract

The free radical scavenging ability of *ogi* flour extracts using α , diphenyl- β -picrylhydrazyl (DPPH) were carried out following Pownall *et al.*²⁰. 1 mL of 0.3mM DPPH dissolved in ethanol in different test tubes was added to different concentrations of 1 mL of the aqueous extracts of each of the samples. The tubes were then shaken vigorously and allowed to stand for 30 min at room temperature in the dark. A control was also prepared as

mention previously without the addition of the sample. Absorbance of the samples was measured at 517 nm using a UV-VIS spectrophotometer (Model SP9, PyeUnican UK) to record the changes. Free radical scavenging ability was expressed as 50% maximal radical inhibition concentration (DPPH IC₅₀).

Determination of total flavonoid content

The total flavonoid content (TFC) of *ogi* flour extract was determined following Lamien *et al.*²¹. *Ogi* flour extracts (0.5 ml) had ethanol (0.5 ml), 50 µl of aluminum chloride (10%), potassium acetate (50 µl), and water (1.4 ml) added to them, and then were incubated for 30 min at room temperature. The absorbance of the reaction mixture was read at 415 nm using spectrophotometer (Model SP9, PyeUnican UK). 0.01 g quercetin dissolved in 20 ml of ethanol was used to prepared standard quercetin solutions (y = 0.001x + 0.0018, $R^2 = 0.992$). The quantity of flavonoids present in the extracts was expressed as quercetin equivalent (QE). The quercetin solution without sample solution was used as positive control due to it's a polyphenol content. All determinations were carried out in triplicate.

Statistical analysis

The means were calculated and separated using MS Excel 2010 and SAS 9.4 version (2014) respectively. Means were separated with Duncan Multiple Range Test (DMRT) at 5% level of probability.

Results and discussions

The total titratable acidity and pH values of all biopreserved ogi flour samples with only 2% garlic, 4% garlic, 4% ginger and samples with blends of 2% garlic-2% ginger, 2% garlic-4% ginger and 4% garlic-2% ginger were stable throughout the 16 weeks of storage. Addition of powdered garlic and ginger improved the stability of ogi flour in terms of pH and total titratable acidity (TTA) values throughout the study of 16 weeks when compared with the control. pH and total titratable acidity of the control were stable for 8 weeks during storage (Figure 1 and Figure 2). With the exception of ogi flour (sorghum) containing 2% ginger, the pH was stable for 12 weeks followed by a slight increase from 3.75-3.88 till the end of storage. The addition of garlic and ginger slightly increased the ash content (0.04%), similar trends were observed in the protein content. However, in all biopreserved ogi samples containing garlic-ginger, a decrease in moisture content was recorded, with the lowest in ogi (sorghum) containing 2% garlic-4% ginger (7.70 %), when compared to the control sample (8.17 %). The moisture content of all biopreserved samples as presented in Table 1 and Table 2 ranged between 7.72-8.17%, and is less than the 10% recommendation for a floury product as reported by Ikujenlola et al.²². The proximate composition of samples was comparable to findings reported during the production of ogi flour from cereal^{17,23}. The addition also increased most of the mineral content of ogi samples. Ogi (maize) containing blends of 4% garlic and 2% ginger had the highest amount of sodium, iron, and manganese. Ogi (sorghum) containing blends of 4% garlic and 2% ginger has the highest amount of magnesium and zinc (Table 3 and Table 4). The addition of garlic and ginger also



Figure 1. pH of *Ogi* flour made with (a) maize and (b) sorghum with Garlic and Ginger. Sample codes: A: *Ogi* Flour, B: *Ogi* Flour + 2% Garlic, C: *Ogi* Flour + 4% Garlic, D: *Ogi* Flour + 2% Ginger, E: *Ogi* Flour + 4% Ginger, F: *Ogi* Flour + 2% Garlic + 2% Ginger, G; *Ogi* Flour+2% Garlic + 4% Ginger, H: *Ogi* Flour + 4% Garlic + 2% Ginger.

enhanced the obtainable minerals in *ogi*. Garlic has been reported as rich source of minerals²⁴. The total phenolic content (TPC) of the *ogi* flour without biopreservative, increased throughout the 16 weeks of storage from 144.50, to 152.63 GAE mg/g, and 171.50-185.75 GAE mg/g, for maize and sorghum respectively. Stable TPCs for the first 8 weeks of storage

were observed in biopreserved ogi flour (maize) samples, which then increased up to the end of the storage period (Figure 3a and b). The total antioxidant radical scavenging activities of ogi flour (maize) without biopreservatives decreased throughout the period of storage from 1.54 to 1.85 mg/ml. However, the total antioxidant radical scavenging activities of ogi flour (sorghum)



Figure 2. Titratable acidity *Ogi* flour made with (**a**) maize and (**b**) sorghum with Garlic and Ginger. Sample codes: A: *Ogi* Flour, B: *Ogi* Flour + 2% Garlic, C: *Ogi* Flour + 4% Garlic, D: *Ogi* Flour + 2% Ginger, E: *Ogi* Flour + 4% Ginger, F: *Ogi* Flour + 2% Garlic + 2% Ginger, G; *Ogi* Flour+2% Garlic + 4% Ginger, H: *Ogi* Flour + 4% Garlic + 2% Ginger.

without biopreservative was stable in the first 8 weeks, and then decreased until the end of the storage period. Relatively stable total antioxidant radical scavenging activities were observed in all biopreserved *ogi* flour from maize throughout the period of storage (Figure 4a), while in sorghum samples a gradual increase throughout the 16 weeks of storage was recorded (Figure 4b). The total flavonoid content (TFC) of all *ogi* flour samples

during the 16 weeks of storage ranged between 114.64-168.11 mg QUE/ 100g, and 150.70-198.83 mg QUE/ 100g in samples from maize and sorghum respectively. The total flavonoid content of all *ogi* flour (maize) without biopreservatives decreased throughout the period of storage from 132.01 to 114.04 mg QUE/ 100g. However, the TFC of all biopreserved samples were relatively stable throughout the 16 weeks of storage

Sample/weeks	%								
	Protein	Fat	Fiber	Ash	Moisture	Carbohydrate			
A _o	10.32±0.03 ^{ab}	3.41±0.01ª	2.15 ± 0.02^{bc}	1.25 ± 0.03^{ab}	8.15±0.01ª	74.72±0.01ª			
16	10.47 ± 0.03^{ab}	3.41±0.01ª	2.18±0.02 ^b	1.40±0.03ª	8.03±0.02ª	74.50 ± 0.02^{a}			
Во	10.45±0.01 ^{ab}	3.44 ± 0.03^{a}	2.21±0.02 ^b	1.33 ± 0.02^{ab}	8.05±0.03ª	74.52±0.01ª			
16	10.47 ± 0.03^{ab}	3.41 ± 0.01^{a}	2.18±0.02 ^b	1.40±0.03ª	8.03±0.02ª	74.50 ± 0.02^{a}			
C _o	10.53±0.02 ^{ab}	3.52 ± 0.01^{a}	2.27±0.03 ^b	1.37±0.01ª	8.01±0.01ª	74.30±0.01ª			
16	10.55±0.01 ^{ab}	3.50 ± 0.02^{a}	2.25±0.03b	1.44±0.02ª	7.98 ± 0.01^{a}	74.28 ± 0.02^{a}			
D _o	10.38±0.02 ^{ab}	3.39 ± 0.01^{a}	2.17±0.01 ^{bc}	1.27 ± 0.03^{ab}	8.13±0.01ª	74.66±0.03ª			
16	10.43±0.01 ^{ab}	3.42±0.03 ^a	2.15±0.01°	1.31±0.02 ^{ab}	8.15±0.01ª	74.54 ± 0.02^{a}			
Eo	10.41±0.03 ^{ab}	3.42±0.01ª	2.19±0.01 ^{bc}	1.30 ± 0.02^{ab}	8.11±0.01ª	74.57 ± 0.02^{a}			
16	10.40 ± 0.01^{ab}	3.41 ± 0.01^{a}	2.17±0.02°	1.37 ± 0.02^{ab}	8.14±0.03ª	74.52 ± 0.02^{a}			
Fo	10.71±0.02 ^{ab}	3.59 ± 0.03^{a}	2.33±0.01b	1.44±0.01ª	7.99 ± 0.02^{a}	73.94±0.01ª			
16	10.73±0.03 ^{ab}	3.58 ± 0.01^{a}	2.35±0.03b	1.47±0.02ª	7.98±0.01ª	73.91±0.03ª			
G _o	10.78±0.01 ^{ab}	3.65 ± 0.02^{a}	2.35±0.01 ^b	1.46±0.03ª	7.94 ± 0.01^{a}	73.82 ± 0.02^{a}			
16	10.82±0.01 ^{ab}	3.63 ± 0.02^{a}	2.33±0.02b	1.52±0.03ª	7.91±0.01ª	73.79±0.01ª			
H _o	10.78±0.01 ^{ab}	3.65 ± 0.02^{a}	2.35±0.01b	1.46±0.03ª	7.94±0.01ª	73.82 ± 0.02^{a}			
16	10.93±0.02 ^{ab}	3.73±0.02ª	2.37±0.03b	1.58±0.02ª	7.81±0.02 ^{ab}	73.63±0.02ª			

Table ⁻	1. Proximate	Composition	of Ogi	(maize)) flour with	Garlic and	Ginger.
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Sample codes: A: *Ogi* Flour, B: *Ogi* Flour + 2% Garlic, C: *Ogi* Flour + 4% Garlic, D: *Ogi* Flour + 2% Ginger, E: *Ogi* Flour + 4% Ginger, F: *Ogi* Flour + 2% Garlic + 2% Ginger, G; *Ogi* Flour+2% Garlic + 4% Ginger, H: *Ogi* Flour + 4% Garlic + 2% Ginger.

Sample/weeks	%								
	Protein	Fat	Fiber	Ash	Moisture	Carbohydrate			
A _o	11.03±0.01ª	2.53±0.01b	3.05 ± 0.03^{ab}	1.20 ± 0.01^{ab}	8.10 ± 0.02^{a}	74.03±0.03ª			
16	11.05±0.01ª	2.50±0.03b	3.11 ± 0.02^{ab}	1.17 ± 0.01^{bc}	8.17 ± 0.03^{a}	73.94 ± 0.03^{a}			
Во	11.15±0.01ª	2.62±0.02 ^b	3.18 ± 0.01^{ab}	1.28 ± 0.01^{ab}	8.04 ± 0.02^{a}	73.73±0.02ª			
16	11.19±0.02 ^a	2.58±0.03 ^b	3.19 ± 0.01^{ab}	1.31 ± 0.02^{ab}	8.02±0.03ª	73.71±0.01ª			
C _o	11.18±0.01ª	2.66±0.02 ^b	3.27 ± 0.01^{ab}	1.30 ± 0.02^{ab}	8.01 ± 0.03^{a}	73.58 ± 0.01^{a}			
16	11.19±0.03ª	2.58±0.01 ^b	3.19 ± 0.02^{ab}	1.31 ± 0.01^{ab}	8.02±0.02ª	73.71±0.01ª			
D _o	11.07±0.01ª	2.55±0.03b	3.09 ± 0.02^{ab}	1.22±0.01 ^{ab}	8.08 ± 0.02^{a}	73.99±0.01ª			
16	11.15±0.03ª	2.51±0.02 ^b	3.12 ± 0.03^{ab}	1.25±0.01b	8.05 ± 0.02^{a}	73.95±0.01ª			
Eo	11.10±0.01ª	2.57±0.02 ^b	3.14 ± 0.03^{ab}	1.30 ± 0.02^{ab}	8.05±0.01ª	73.84±0.01ª			
16	11.15±0.02ª	2.55±0.02b	3.17 ± 0.01^{ab}	1.33±0.03 ^{ab}	8.01 ± 0.02^{a}	73.79±0.01ª			
Fo	11.21±0.01ª	2.69±0.01b	3.52±0.02ª	1.35 ± 0.01^{ab}	7.99±0.01ª	73.24±0.03ª			
16	11.25±0.02ª	2.67±0.03 ^b	3.50±0.02 ^b	1.38 ± 0.01^{ab}	7.91 ± 0.02^{a}	73.17±0.01ª			
G _o	11.24±0.03ª	2.78 ± 0.01^{ab}	3.60 ± 0.02^{a}	1.40±0.01ª	7.75 ± 0.01^{a}	73.23±0.03ª			
16	11.27±0.03ª	2.77±0.01b	3.57±0.01ª	1.47±0.02ª	7.73±0.01 ^{ab}	73.19±0.03ª			
H _o	11.28±0.02ª	2.85 ± 0.01^{ab}	3.65±0.03ª	1.45±0.01ª	7.72±0.01 ^{ab}	73.05±0.01ª			
16	11.30±0.03ª	2.82±0.01 ^b	3.62 ± 0.02^{a}	1.50±0.01ª	7.70±0.03 ^{ab}	73.02±0.01ª			

Table 2. Proximate Composition of Ogi (sorghum) flour with Garlic and Ging	ger.
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Sample codes: A: *Ogi* Flour, B: *Ogi* Flour + 2% Garlic, C: *Ogi* Flour + 4% Garlic, D: *Ogi* Flour + 2% Ginger, E: *Ogi* Flour + 2% Ginger, F: *Ogi* Flour + 2% Garlic + 2% Ginger, G; *Ogi* Flour + 2% Garlic + 4% Ginger, H: *Ogi* Flour + 4% Garlic + 2% Ginger.

Sample code	Sodium	Magnesium	Zinc	Iron	Manganese
A	4.69±0.01 ^d	10.85±0.01 ^b	1.37±0.03°	1.53 ± 0.01^{bc}	1.30 ± 0.04^{ab}
В	5.29 ± 0.02^{d}	11.32 ± 0.01^{ab}	1.48±0.01°	1.69±0.01 ^b	1.45 ± 0.01^{ab}
С	5.35±0.01 ^d	11.48 ± 0.01^{ab}	1.51±0.01°	1.73±0.01 ^b	1.61 ± 0.03^{a}
D	5.11 ± 0.06^{d}	12.99±0.01ª	1.40±0.02°	1.55 ± 0.01^{bc}	1.58 ± 0.05^{a}
E	5.28±0.01 ^d	12.05 ± 0.01^{ab}	1.44±0.05°	1.59±0.01 ^{bc}	1.60 ± 0.05^{a}
F	5.69±0.03°	12.58 ± 0.01^{ab}	1.57±0.01°	1.71±0.01 ^b	1.65 ± 0.03^{a}
G	6.19±0.04°	12.61 ± 0.01^{ab}	1.66±0.03°	1.75±0.02 ^b	1.69 ± 0.02^{a}
Н	6.55±0.01°	13.13±0.01ª	1.78±0.02°	1.78±0.03 ^b	1.71 ± 0.01^{a}

Table 3. Mineral Content (mg/100g) of Ogi (maize) flour with Garlic and Ginger.

Values are means (n = 3) \pm standard deviation. Means followed by different superscripts are significantly different (p < 0.05) along column according to Duncan multiple range test. Sample codes: A: *Ogi*, B: *Ogi* + 2% Garlic, F: *Ogi* + 2% Garlic + 2% Ginger, G; *Ogi* + 2% Garlic + 4% Ginger, H: *Ogi* + 4% Garlic + 2% Ginger.

Table 4	Mineral	Content	(ma/100a)	of Oai	(sorahum)	flour	with (Garlic an	d Ginger
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Sample code	Sodium	Magnesium	Zinc	Iron	Manganese
А	12.51±0.01 ^b	5.17 ± 0.01^{d}	7.01 ± 0.01^{b}	0.68 ± 0.03^{de}	0.45±0.01°
В	15.00 ± 0.03^{ab}	7.50±0.01°	7.18±0.01 ^b	1.04±0.01 ^d	0.56±0.01°
С	15.44 ± 0.01^{ab}	7.80±0.01°	7.64 ± 0.01^{ab}	1.37±0.01°	0.58±0.01°
D	15.12 ± 0.01^{ab}	7.10±0.01°	7.15±0.01 ^b	0.90 ± 0.01^{d}	0.74 ± 0.01^{b}
E	15.18 ± 0.01^{ab}	7.40±0.01°	7.52 ± 0.01^{ab}	1.29±0.01°	0.77 ± 0.01^{b}
F	16.02±0.04ª	8.05±0.01°	7.63 ± 0.01^{ab}	1.40 ± 0.01^{ab}	0.79 ± 0.01^{b}
G	16.13±0.01ª	9.59 ± 0.01^{b}	7.99 ± 0.01^{a}	1.69 ± 0.01^{a}	0.88 ± 0.01^{b}
Н	16.15±0.01ª	9.72 ± 0.01^{b}	8.50 ± 0.01^{a}	1.77±0.01ª	0.86 ± 0.01^{b}

Values are means (n = 3) \pm standard deviation. Means followed by different superscripts are significantly different (p < 0.05) along column according to Duncan multiple range test. Sample codes: A: *Ogi*, B: *Ogi* + 2% Garlic, F: *Ogi* + 2% Garlic + 2% Ginger, G; *Ogi* + 2% Garlic+ 4% Ginger, H: *Ogi* + 4% Garlic+ 2% Ginger.

(Figures 5a and 5b). Comparing biopreserved samples the highest total flavonoid content was observed in samples containing blends of 4% garlic-2% ginger (168.11 and 198.83 mg QUE/ 100g for maize and sorghum respectively) and the lowest in samples containing only 2% ginger (140.01 and 170.44mg QUE/ 100g for maize and sorghum respectively). Relatively stable total antioxidant radical scavenging activities were observed in all biopreserved ogi flour from maize throughout the period of storage. It was observed that antioxidant radical scavenging activities of all biopreserved ogi flour samples from sorghum gradually increased throughout the 16 weeks of storage (Figure 5) However, the TFC of all biopreserved samples were relatively stable throughout the 16 weeks of storage. Garlic antimicrobial active such as allicin contains ajoene, methyl allyl trisulfide and diallyl disulfide which are organosulfur compounds^{25,26}. Likewise borneol, α -pinene, linalool and camphene in ginger are responsible for its antimicrobial activities while 1, 8- cineole have been reported as key component responsible for antimycotic activity which are crucial in the preservation of food²⁷. The results of this study showed that garlic and ginger can be considered good sources of natural compounds with significant antioxidant activity. A combination of garlic and ginger exerted a synergistic effect on the radical scavenging activities of *ogi* samples with the highest effect observed in samples containing 4% garlic and 2% ginger. It also increased the total flavonoid content, and enhanced its stability in flour samples during storage.

Conclusion

This study has shown that the proximate compositions of biopreserved *ogi* flour samples were relatively stable, and that the addition of garlic and ginger slightly increased mineral



Figure 3. Total Phenolic Content *Ogi* flour made with (a) maize and (b) sorghum with Garlic and Ginger. Sample codes: A: *Ogi* Flour, B: *Ogi* Flour + 2% Garlic, C: *Ogi* Flour + 4% Garlic, D: *Ogi* Flour + 2% Ginger, E: *Ogi* Flour + 4% Ginger, F: *Ogi* Flour + 2% Garlic + 2% Ginger, G; *Ogi* Flour+2% Garlic + 4% Ginger, H: *Ogi* Flour + 4% Garlic + 2% Ginger.

content during storage. The free radical scavenging activity; total phenolic and flavonoid content also increased correspondingly with the antioxidant activity of the *ogi*. Therefore, the addition of garlic and ginger as biopreservative into *ogi* flour both singly or

as blends at 2 or 4 % w/w can be used, and will not negatively affect its quality. Although not well known to *ogi* consumers, the biopreserved *ogi* flours showed better nutritional values and may be administered as a health food.



Figure 4. DPPH Radical Scavenging Activity *Ogi* flour made with (**a**) maize and (**b**) sorghum with Garlic or Ginger. Sample codes: A: *Ogi* Flour, B: *Ogi* Flour + 2% Garlic, C: *Ogi* Flour + 4% Garlic, D: *Ogi* Flour + 2% Ginger, E: *Ogi* Flour + 4% Ginger, F: *Ogi* Flour + 2% Garlic + 2% Ginger, G; *Ogi* Flour+2% Garlic + 4% Ginger, H: *Ogi* Flour + 4% Garlic + 2% Ginger.



Figure 5. Total Flavonoid Content *Ogi* flour made with (a) maize and (b) sorghum with Garlic or Ginger. Sample codes: A: *Ogi* Flour, B: *Ogi* Flour, + 2% Garlic, C: *Ogi* Flour + 4% Garlic, D: *Ogi* Flour + 2% Ginger, E: *Ogi* Flour + 4% Ginger, F: *Ogi* Flour + 2% Garlic + 2% Ginger, G; *Ogi* Flour+2% Garlic + 4% Ginger, H: *Ogi* Flour + 4% Garlic + 2% Ginger.

Data availability

Underlying data is available from Figshare.

Figshare: Dataset 1. Data for Proximate and antioxidant activities of bio-preserved ogi flour with garlic and ginger. https://doi.org/10.6084/m9.figshare.7415795.v1²⁸

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Grant information

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Abimbola K. Arise

University of Ilorin, Llorin, Nigeria

Abstract: This is well written and provides an exact overview of the work that was done.

Introduction: This section is comprehensive and relevant. Although the references cited are relevant

Methodology: The choice of experimental design and method are adequate and there is a clear thread between the sections.

Results: The results are clearly presented and adequately interpreted.

Discussion: The data analysis methods are adequate and result sufficiently discussed and compared to existing literature. This study is relevant and important particularly in the light of combating malnutrition in Africa especially among children because ogi is the major weaning food in Africa. The paper integrates well with the current research. The paper is generally well laid out and written.

However, there is need to pay attention to tenses. Some other editorial correction have been indicated in the reviewed manuscript attached.

Is the work clearly and accurately presented and does it cite the current literature? Yes

Is the study design appropriate and is the work technically sound? Yes

Are sufficient details of methods and analysis provided to allow replication by others? Yes

If applicable, is the statistical analysis and its interpretation appropriate? $\ensuremath{\mathsf{Yes}}$

Are all the source data underlying the results available to ensure full reproducibility? Yes

Are the conclusions drawn adequately supported by the results? Yes

Competing Interests: No competing interests were disclosed.

Reviewer Expertise: Food chemistry, protein chemistry, food product development, cereals and legume processing

I confirm that I have read this submission and believe that I have an appropriate level of expertise to confirm that it is of an acceptable scientific standard.

Reviewer Report 20 March 2019

https://doi.org/10.5256/f1000research.18649.r45441

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Dolapo Oladiran 🔟

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The research aims to investigate the effect of the use of garlic and ginger as bio-preservatives on the proximate and antioxidant properties of *ogi* flour made from cereal grains (sorghum or maize). The experiment has been well designed and overall, this study contributes to the knowledge about the potential use of local spices as natural preservatives in food. Still, authors should consider the following important comments.

There are a few typographical errors throughout the manuscript e.g. in the last paragraph under the introduction, authors should consider 'nutritional' instead of 'nutritionals' quality of *ogi*.... Under methods specifically where '*ogi* preparation co-fermented with garlic and ginger' is described, authors should decide whether to use past tense (as in winnowed, steeped) or present continuous tense (as in draining, milling).

The authors should consider giving further scientific depth. This can be done by expounding on the compounds present in ginger and garlic that make them good preservatives and the scientific rationale behind how these compounds exert their bio-preservative properties. If these are added, the technicality of the study especially the discussion bit would be greatly improved.

Is the work clearly and accurately presented and does it cite the current literature? $\ensuremath{\mathsf{Yes}}$

Is the study design appropriate and is the work technically sound? Partly Are sufficient details of methods and analysis provided to allow replication by others? $\gamma_{\mbox{es}}$

If applicable, is the statistical analysis and its interpretation appropriate? $\ensuremath{\mathsf{Yes}}$

Are all the source data underlying the results available to ensure full reproducibility? Yes

Are the conclusions drawn adequately supported by the results? Yes

Competing Interests: No competing interests were disclosed.

Reviewer Expertise: Food chemistry, Sensory Science, Oral processing and ingestive behaviours, Nutrition

I confirm that I have read this submission and believe that I have an appropriate level of expertise to confirm that it is of an acceptable scientific standard.

Comments on this article

Version 1

Author Response 05 Jul 2019

Abiola Olaniran, Landmark University, Omuaran, Nigeria

1. Typographical errors in the manuscript as suggested by reviewers 1 and 2 have been corrected. 2. As suggested reviewer 1, present continuous tense have been used to describe the process of production under methods for *ogi* preparation co-fermented with garlic and ginger.

Competing Interests: No competing interests were disclosed.

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